

Abstract

# $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$ NASICON-Type Electrolytes with Enhanced Conductivity for Solid-State Lithium-Ion Batteries <sup>†</sup>

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**Abstract:** The use of lithium-ion batteries allows for a reliable and efficient storage of electricity. Commercial batteries use flammable liquid organic electrolytes, which have a low thermal and electrochemical stability. Replacing liquid electrolytes with solid ones would solve these problems. NASICON-structured electrolytes, in particular LATP ( $\text{Li}_{1+y}\text{Ti}_{2-y}\text{Al}_y(\text{PO}_4)_3$ ) and LAGP ( $\text{Li}_{1+y}\text{Ge}_{2-3y}\text{Al}_y(\text{PO}_4)_3$ ), are among the most promising electrolytes for all-solid-state batteries. The partial replacement of titanium ions with germanium ions can lead to materials that combine the high lithium-ion conductivity of LATP with the high chemical stability of LAGP. The aim of this work was to synthesize and study the ionic mobility of  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  ( $x = 0-2$ ,  $y = 0-0.3$ ) with the NASICON structure.  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  ( $x = 0-2$ ,  $y = 0-0.3$ ) electrolytes were synthesized with the solid-state method and investigated using X-ray diffraction and scanning electron microscopy, impedance spectroscopy, and NMR spectroscopy. The processes occurring during the solid-state synthesis of  $\text{Li}_{1+y}\text{Ti}_{2-x-y}\text{Ge}_x\text{Al}_y(\text{PO}_4)_3$  were studied. An increase in conductivity from  $10^{-7}$  S/cm to  $4.6 \cdot 10^{-6}$  S/cm at 25 °C was found when 10% of titanium ions were replaced with germanium. The additional introduction of aluminum resulted in an increase in lithium conductivity of up to  $1.4 \cdot 10^{-4}$  S/cm (25 °C). Since grain boundaries were of decisive importance for the overall ionic conductivity of the NASICON-structured phosphates, the influence of the precursor mechanical treatment on the microstructure and ionic conductivity of the prepared materials was studied. The use of the mechanical treatment led to a significant increase in grain size (reducing the grain boundaries and their resistance) and an increase in ionic conductivity (up to  $6.4 \cdot 10^{-4}$  S/cm at 25 °C). The obtained materials could be considered promising solid electrolytes for all-solid-state lithium batteries with high safety and stability.



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