



Polysaccharide-Based Nanoparticles for Colon-Targeted Drug Delivery Systems

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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). **Abstract:** Polysaccharide biomaterials have gained significant importance in the manufacture of nanoparticles used in colon-targeted drug delivery systems. These systems are a form of non-invasive oral therapy used in the treatment of various diseases. To achieve successful colonic delivery, the chemical, enzymatic and mucoadhesive barriers within the gastrointestinal (GI) tract must be analyzed. This will allow for the nanomaterials to cross these barriers and reach the colon. This review provides information on the development of nanoparticles made from various polysaccharides, which can overcome multiple barriers along the GI tract and affect encapsulation efficiency, drug protection, and release mechanisms upon arrival in the colon. Also, there is information disclosed about the size of the nanoparticles that are usually involved in the mechanisms of diffusion through the barriers in the GI tract, which may influence early drug degradation and release in the digestive tract.

Keywords: nanoparticles; polysaccharide; drug delivery; colon

1. Introduction

Polysaccharides are biopolymers constituted by simple sugar monomers. They are usually isolated from plant material, marine plants, and exogenous metabolites of some bacteria. The polysaccharides chain comprises monosaccharide units linked by glycosidic bonds with hydroxyl, carboxyl, and amino groups. Carbohydrates are stable, non-toxic, hydrophilic, and biodegradable biomolecules in the human body [1,2]. Polysaccharides can be used as matrices for encapsulation, immobilization, and controlled release for many active compounds. This matrix are also used in the food, pharmaceutical, biomedical, and chemical industries. Polysaccharides are the most abundant biomolecules, followed by proteins, lipids, and nucleic acids. Polysaccharides studied as a matrix for encapsulation include chitosan, pectin, alginate, starch, and dextran [3,4].

Encapsulation occurs when liquid and solid particles are trapped in a matrix or coated by a polymer material. The preparation of a proper encapsulating agent is an essential factor in enhancing encapsulation efficiency. Encapsulating agents must have the ability to preserve bioactive compounds under different processes and storage conditions [5]. A controlled release system will increase drug retention time and enhance paracellular and transcellular absorption. The release mechanisms occur by diffusion, biodegradation, and osmosis [6,7]. The application of nanotechnology to encapsulate nutrients and drugs has aroused significant interest and demand due to its great benefits. In addition, nanotechnology has been associated with safety, environmental, ethical, and regulatory problems related to human health and environmental impacts. Information related to the security of nanoparticles is limited. In general, nanoparticles have unique characteristics, such as their small size. This property allows them to cross different biological barriers (e.g., intestinal and mucosal epithelial cells) [8,9].

Natural biopolymers are considered biodegradable and safe for human consumption because they are not toxic and do not affect cell viability, making them ideal for drug delivery systems [10,11]. Other kinds of metallic, solid, and polymeric nanoparticles can be toxic for humans. Researchers have used strategies of coating toxic nanoparticles with biopolymers to reduce their toxicity [12,13]. At the moment, no legislation regulates the use of nanoparticles in the food and pharmaceutical industries. Therefore, most countries have regulations for risk assessment when using nanotechnology. Relevant regulatory agencies include the U.S. Food and Drug Administration (FDA), the European Union, the Australian Government Department of Health, and Health Canada [14].

It is necessary to develop and appropriately evaluate nanomaterials to determine their risk to human health. When manufacturing these nanoparticles, the elements that must be considered are particle and distribution size, shape, state of aggregation/disaggregation, solubility, surface charge, and surface morphology [15,16]. The size and morphology of nanoparticles enhance their functionality, and are dependent on the manufacturing technique for each nanomaterial. These properties allow a high loading capacity, high encapsulation efficiency, stability, sustained release profile, bioavailability of bioactive compounds, and biocompatibility [17,18].

The nanoparticle technology based on polysaccharides plays a vital role in controlling drugs, bioactive agents, and genes for oral administrations. These release systems can occur by diffusion (barrier/matrix), degradation (chemical or physical matrix), or changes in the environment (e.g., pH, ionic strength, and pressure) [19]. Most drugs or bioactive agents administered orally are absorbed in the upper GI tract, but their delivery in the colon is necessary for additional results in specific therapies. Some areas of interest are colonic delivery of peptides and proteins, probiotic bacteria, and microbiota replacement therapies [20,21]. The main release mechanisms in the colon are degradation by colonic microbiota, time and pH-controlled release. Colon release formulations are generally designed to prevent degradation in the stomach and upper GI tract. The colon's main function is the absorption of water, ions, and the storage of feces. The large intestine is colonized by many bacteria—approximately 10¹² per gram of intestinal content. The microorganisms in the colon are beneficial for human health because they are responsible for fermenting indigestible dietary fiber [22].

This review focuses on polysaccharide-based nanoparticles used for oral administration in colon-directed therapies, such as drug administration, peptides, and proteins. This contribution aims to present important information from the literature that shows various polysaccharide-based nanomaterials. Therefore, these nanomaterials can reach the colon through physiological barriers due to their morphological characteristics, such as size, shape, and surface charge.

2. Polysaccharides

Various polymers with complex structures and specific functions have developed naturally. These include amino acids, nucleobases, and mono- and disaccharides. According to the nature of their heteroatom present in the main chain natural polymers can be classified into four types, as shown in Table 1 [23,24]. Polysaccharides are biopolymers isolated from plant, animal, microbial, and algae sources. These are made up of more than ten monosaccharide units linked by O-glycosidic bonds. Due to their great abundance in hydroxyl groups polysaccharides can be modified by carboxylation, esterification, and amination, thereby improving their functional properties [25]. According to their nature, polysaccharides have excellent biocompatibility, biodegradability, non-toxicity, and cell specificity. Therefore, they are considered ideal for various biomedical applications, such as drug, gene delivery, and wound dressing [26,27].

Natural Polymers	Types	Examples	
Hydrocarbon polymers	Natural rubber		
Carbon-oxygen	Carbohydrates	Cellulose; starch; chitin; chitosan; pullulan	
Carbon-oxygen-nitrogen/Sulphur	Proteins	Soya protein; gelatin; casein	
Carbon-oxygen-nitrogen- phosphorus	Nucleic acids	DNA, RNA	

Table 1. Natural polymers and their types.

2.1. Chitosan

Chitosan is a unique linear poly-cationic polysaccharide derived from chitin by deacetylation. It is found primarily in the exoskeleton of crustaceans, such as shrimp and crabs. Its chemical structure consists of a chain of β -(1-4)-linked glucosamine and *N*-acetyl D-glucosamine units. It has functional groups such as polyhydroxy and amino, and these groups can lead to hydrogen bonds. Chitosan is the second most abundant polysaccharide after cellulose. It is used for pharmaceutical applications due to its reactive functional groups, biocompatibility, biodegradability, gel-forming ability, non-toxicity, high charge density, and low pH solubility. This polysaccharide can interact electrostatically with mucus or negatively charged mucosal surfaces, giving it an excellent mucoadhesive property for developing systems for oral administration. The presence of its positively charged amino group (–NH³⁺) can interact with negatively charged proteoglycans on the cell surface, enabling better intestinal absorption of the drug by opening the tight junctions between the epithelial cells [28].

2.2. Hyaluronic Acid

Hyaluronic acid is a non-sulfated, negatively charged glycosaminoglycan composed of D-glucuronic acid and N-acetyl-D-glucosamine linked by β -(1-3) and β -(1-4) bonds. It is a macromolecule produced and secreted by cells as a linear polymer that is not bound to a polypeptide. It is a biocompatible biopolymer of natural origin present in the skin, connective tissues, synovial fluid of the joints, neural tissues, vitreous humor, and has the ability to regulate lubrication. This polysaccharide can be chemically modified due to its groups, such as the carboxylic acid of glucuronic acid, primary and secondary hydroxyl groups, and *N*-acetyl groups, which can alter its properties such as hydrophobicity, biological activity, viscoelasticity, water retention, biocompatibility, cell proliferation, wound regeneration, and specific signal transduction and cell interactions through cell surface receptors [29].

2.3. Pectin

Pectin is a linear heteropolysaccharide that constitutes the cell wall of plants and consists of a linear unbranched chain of α -(1-4) linked D-galacturonic acid units (homogalacturonan) with uronic acids. This polysaccharide contains hydroxyl and carboxyl groups, residues esterified with methyl ether distributed in its linear chain, and a certain quantity of neutral sugars present in the side chains. It is principally composed of galacturonic acid, methyl ester, and sugar units such as arabinose, galactose, and rhamnose. Pectin can form hydrogels in the presence of its ionized carboxyl groups (–COO[–]) that interact with its positively charged anions. For this reason, it is an anionic, biodegradable, biocompatible, and non-toxic polysaccharide used for its mucoadhesive properties in the oral delivery of drugs to the colon. Pectin can remain intact in the upper gastrointestinal tract and be degraded in the colon by pectinases [30].

2.4. Guar Gum

Guar gum is a non-ionic polysaccharide. It is extracted from the seeds of *Cyamopsis* tetragonolobus and consists of a linear chain of (1-4)- β -D-mannopyranosil units with α -D-

galactopyranosyl units attached by (1–6) ramifications. This polysaccharide presents an extraordinarily viscous property due to its intermolecular chain entanglement of galactose side chains. It is a biocompatible and biodegradable polysaccharide with the ability to form gels in aqueous solutions, and is used to formulate hydrophilic matrices for drug delivery due to its enzymatic degradation in the colon [31].

2.5. Dextran

Dextran is a complex branched glucan consisting of α -D-(1-6) glycosidic links and branched at α -(1-3). It is obtained naturally from the lactic acid bacterium. It is a hydrophilic, biodegradable, and biocompatible polysaccharide with abundant hydroxyl groups. This polysaccharide is biocompatible, highly hydrophilic, and shows low protein adsorption. Like other polysaccharides, dextran has many hydroxyl groups, allowing it to be easily conjugated with drugs and proteins to prevent drug absorption in the small intestine [32].

2.6. Alginate

Alginate is an anionic polysaccharide consisting of β -(1-4)-D-mannuronic acid and α -(1-4) L-guluronic acid residues. The α -(1-4) L-guluronic bonds can be cross-linked with divalent ions by sodium ion exchange, forming a gel matrix and the retention of encapsulated charges. Due to their anionic nature, they can interact with cationic compounds. It is extracted mainly from brown marine algae (*Laminaria hyperborean*) and soil bacteria (*Azobacter vinelandii*). This polysaccharide has physicochemical properties such as biodegradability, biocompatibility, low immunogenicity, good mucoadhesion, and non-toxicity. Due to its composition, sequence of arrangement, and molecular weight, the alginate also has functional groups such as polyhydroxy and carboxyl distributed throughout its chain. It is highly reactive and with the possibility of chemical modification (oxidation, amidation, esterification, and sulphation) [33].

2.7. Arabinoxylans

Arabinoxylans (AX) are non-starch polysaccharides found mainly in the cell wall, outer layer, and endosperm of cereals. The AX are composed of a linear chain of β -(1-4) xylose units branched to arabinose units in positions C(O)-3 and C(O)-2. Arabinose can be esterified with monomeric or dimeric ferulic acid (AF). AX are desirable polysaccharides for application in the food, pharmaceutical, and biomedical industries. Similarly, they have become an attractive alternative due to their biodegradability, biocompatibility, non-toxicity, hydrophilic, and gelling properties. Due to their functional properties, AX can act as prebiotics, antioxidants, emulsifiers, and immunomodulators. The gels made from this polysaccharide have been studied for their potential as a drug delivery system directed to the colon because they are biocompatible and have the ability to retain water. These covalent gels can cross the conditions of the upper gastrointestinal tract and be fermented in the colon by the colonic microbiota to release the drug [34,35].

3. Polysaccharide Based-Nanoparticles

Drugs can be loaded into a polysaccharide matrix or be bound to the external surface, improving their aqueous solubility and stability (e.g., proteins). They are used to enhance slow enzymatic degradation, targeting mucosal recognition, receptor binding, adhesion, and transport systems [36,37]. Polysaccharides have presented advantages and benefits when they are used as biomaterials for the manufacture of nanoparticles. Compared to nanoparticles synthesized from other materials, such as metals or synthetic polymers, polysaccharide nanoparticles do not present toxicity problems due to their biocompatibility and biodegradability, thus providing safety, better storage, and physiological stability [38,39].

Nanotechnology-based drug delivery systems aim to target different routes of administration. When designing polysaccharide-based nanoparticles for use in biomedicine and other industries, it is important to consider the route of administration, retention time, and therapy [40]. These factors may be dependent on the polysaccharide used, its size, and its structure. One possible disadvantage when using polysaccharides for manufacturing nanoparticles is their high molecular weights. In addition, in many cases, the low solubility of some polysaccharides limits their potential in chemical modification [27]. Due to the polysaccharide properties, the size of polysaccharide-based nanoparticles has been considered between 10 and 1000 nm [41,42].

These nanoparticles can carry drugs, proteins, peptides, or DNA material to target a specific organ or cell. Depending on the technique, the drug or other active compound can adhere, dissolve, encapsulate, or entrap into the nanoparticle matrix [43]. The significant interest in this technology is due to the fact that the nanoparticles can target specific organs or Diana cells, and reduce side effects. Nanoparticles have a larger surface area, and their smaller size allows them to enter smaller capillaries and cells. Nanoparticle size and surface charge are highly important because many secondary properties, such as surface area, toxicity, degradation, targeting, and absorption mechanism, are related [18]. Polysaccharide-based nanoparticles have shown significant potential in biological and pharmaceutical applications. Natural polysaccharides used to fabricate nanoparticles include chitosan, alginate, pectin, guar gum, dextran, hyaluronic acid, and arabinoxylan (Table 2).

Previously studied polysaccharides used to fabricate nanoparticles include the manufacture of acrylate chitosan-based nanoparticles cross-linked with tripolyphosphate anions (TPPs). The authors obtained chitosan nanoparticles with an average diameter of 356 nm and a Z potential of 13.7 mV [44]. In another study, nanoparticles were synthesized by βlactoglobulin-pectin to encapsulate an anticancer drug (bipyridine ethyl dithiocarbamate Pt (II) nitrate), and they obtained nanoparticles with sizes of 200–250 nm [45]. The fabricated nanoparticles based on bovine serum albumin (BSA)-dextran conjugate to encapsulate curcumin showed nanoparticles with a size of 115 nm [46]. Similarly, arabinoxylan cross-linked with laccase was developed and showed a hydrodynamic diameter of 328 nm [47].

Polysaccharide	Chain Characteristics	Properties	Crosslinker	Degradation Mechanism	Comments	Reference
Chitosan	β-(1-4)-linked N-acetyl D-glucosamine	-Positive Z potential -pKa 6.5 -Protonated at acidic and neutral pH -Mucoadhesive -Cationic	Calcium TPP	Enzymatic	Degradation by enzymes (β-glucosidase)	[48]
Alginate	β-(1-4)-D- mannuronic acid and α-(1-4) L-guluronic acid residues	-Water-soluble -Anionic -Hydrophobicity	Divalent cations Ca ²⁺ , Cu ²⁺ , Zn ²⁺ or Mn ²⁺	pH-responsive	Degradation by enzymes (glu- curonidases etc.)	[33,49]
Pectin	(1-4)-linked α -D-galacturonic acid residues	-Water-soluble -High methoxy -Low methoxy	Ca ²⁺ ions Lacasse	Enzymatic	Degradation by Bacteroides species	[50,51]
Guar Gum	(1-4)- β-D- mannopyranose units with α-D- galactopyranosyl units attached by (1-6) linkages	-Water-soluble	Epichlorohydrin	pH-responsive and enzymatic	Degradation by bacteria (Bacteroides, ruminococci, bifidobacteria)	[19,31]
Dextran	α-(1-6) glycosidic links and branched at α-(1-3) position	-Mucoadhesive -Hydrophilic	Diamine	Enzymatic	Degradation by esterases and endodex- tranases	[52]

Table 2. Natural polysaccharides, properties, and their degradation mechanism.

Polysaccharide	Chain Characteristics	Properties	Crosslinker	Degradation Mechanism	Comments	Reference
Hyaluronic Acid	D-glucuronic acid and N-acetyl-D- glucosamine linked by β-(1-3) and β-(1-4) bond	-Anionic -Water-absorption -Water soluble	pH temperature ionic	pH-responsive and enzymatic	Degradation by hyaluronidases	[53]
Arabinoxylan	Xylose β -1-4 linkages and α -L-arabinose substitutions (α -1-3 and α -1-2)	-Highly ferulated -Water-soluble -Neutral	Laccase Peroxidase	Enzymatic	Degradation by bacteria (Bifi- dobacterium and Bacteroides)	[34,54]
(TDD) - diama tain shark - a kata						

Table 2. Cont.

(TPP) sodium tripolyphosphate.

3.1. Physicochemical Characteristics of Polysaccharide-Based Nanoparticles

3.1.1. Size and Shape

The development of polysaccharide-based nanoparticles has progressed significantly in biomedicine in applications of drug, protein, and gene control delivery [55] and tissue engineering [56]. These nanomaterials have contributed to therapies for the treatment of various diseases such as cancer [57], diabetes [58], wound treatment [59], and inflammation [60]. These nanoparticles present a rapid absorption and release behavior, which allows a high diffusion capacity and volume change. Furthermore, their size and shape can be easily adapted and controlled to reduce desirable side effects [61]. The characterization of nanoparticles (size, surface charge, shape, and agglomeration) is highly important during their fabrication, because they have to overcome barriers in the human body, such as cell internalization, tissue penetration, localization directed at tumors, and circulation through blood vessels (Figure 1). The properties of nanoparticles related to their size and shape are their colloidal stability, specific surface, optical properties, in vivo behavior, and cellular absorption [62,63]. Boverhof et al. [14] analyzed 14 different regulatory authorities for the use of nanomaterials with a particular emphasis on safety and impact on human health and the environment. These authorities agreed to define the word "nanomaterial", which is a representative characteristic of its size. As a definition, a nanomaterial has at least one external dimension or surface structure in the nanometric range (~1–100 nm). Furthermore, these regulatory authorities indicated that the upper field of the nanoscale can be more than 100 nm and must present physical or chemical properties, and their biological effects are attributable to its size (e.g., bioavailability, lower toxicity, and lower dose) [64-67].



Figure 1. Illustration of the main properties of nanoparticles. Adapted from [68]. Open access.

The nanoparticle's size/shape are important parameters because they contribute to the biological processes. Small-sized particles present a higher surface-to-volume ratio, leading to the faster release of the drug. This size allows the particles to cross cell membranes, organs, and tissues. These include transport through the vasculature, phagocytosis,

endocytosis, and intracellular transport. However, the size should not be too small because it has been established that the size threshold in renal excretion is usually 6 to 8 nm. In addition, sizes smaller than 6 nm can be coated with serum proteins circulating through the bloodstream, which favors an increase in the hydrodynamic diameter, blocking renal excretion [69]. Conversely, nanoparticles larger than 100 nm are more likely to accumulate at the injection site or be trapped by macrophages in the spleen, lung, and liver [70].

3.1.2. Superficial Charge and Aggregation

The surface charge measures the Z potential and represents a simple measurement of the particles electrical characteristics. This property determines the stability of colloidal dispersions and is affected by the ionic strength and pH of the dispersion. The Z potential indicates the degree of repulsion between adjacent particles, charged in a dispersion [19]. Aggregation occurs when several particles are joined together strongly, resulting in the sum of surface areas of the individual particles. The size and shape of the aggregation have a significant influence on the properties of the nanomaterial. These aggregates are indivisible structures and cannot be separated by external forces [14]. The surface charge determines the dispersion, aggregation, or flocculation of the particles. It can influence stability, and a neutral charge leads to instability of the nanoparticles, causing precipitation. A high Z potential confers stability to the particles, thus preventing their aggregation. When this value is low, the particles tend to attract each other and form flocs. Negatively charged nanoparticles have low phagocytic absorption, increasing their time in the bloodstream. Additionally, positively charged particles cause increased phagocytosis due to the interaction of the anionic cell membrane with the nanoparticles [71]. As the particle passes through the systemic circulation, a positive surface charge facilitates interaction with plasma proteins, resulting in aggregation. Therefore, the charge density in the nanoparticles when interacting with body fluids can be altered [18,69,72]. The use of nanotechnology in drug delivery enables better targeting of delivery to cells and tissues [55].

3.2. Cellular Uptake Mechanisms

The cellular uptake mechanisms to deliver biological substances and therapeutic agents require internalization to specific compartments or organelles to affect a cellular level. First, nutrient uptake, communication between cells, their microenvironment, and cell adhesion occur through the plasma membrane by different mechanisms. Due to of their concentration gradients, hydrophobic or nonpolar compounds can easily diffuse through the plasma membrane by active transport mechanisms, such as integral membrane protein pumps or ion channels. Finally, nanoscale hydrophilic biomacromolecules are conventionally transported into the cell by endocytosis [73].

Endocytosis is an active transport process in which a cell membrane wraps itself around an object and encloses it in vesicles or vacuoles pinched off from its cytoplasmic membrane. Endocytosis is divided into two main mechanisms: phagocytosis and pinocytosis (Figure 2). Phagocytosis is the primary uptake for larger particles by the cell membrane forming an internal phagosome, and occurs principally in certain types of phagocytes (macrophages and neutrophils). Phagocytes can take up relatively large particles (1 micron). This mechanism occurs when a particle, cells, and pathogens are recognized by phagocytes circulating through the blood and tissues. They attach to the phagocytes' surface receptors, allowing further internalization. Once recognized, the plasma membrane of phagocytes encloses the particles within a cup-shaped protrusion and fuses to produce a phagosome that is pinched off inward [74].

Pinocytosis highlights the uptake of smaller particles. Pinocytosis is subdivided into clathrin-dependent and clathrin-independent pathways, divided into caveolae-mediated endocytosis, micropinocytosis, and other endocytic mechanisms. The clathrin-mediated endocytosis mechanism is the main endocytosis pathway, through which cells absorb necessary materials. This is known as receptor-mediated uptake, which involves the capture of particular macromolecules (ligands) or ligand-coated particles after attaching to receptors on the cell membrane's surface. This binding (receptor-ligand) initiates a process in the plasma membrane. It results in the formation of "coated pits", where the receptor concentration is greater than that in the remaining area of the cell membrane. Non-specific adsorptive uptake is a receptor-independent mechanism, and the uptake occurs through hydrophobic or electrostatic interactions, eventually allowing the internalization process. The macropinocytosis mechanism is a process that allows the uptake of larger particles that cannot easily cross other endocytosis pathways. In this mechanism, the plasma membrane generates ruffles of diverse shapes that absorb extracellular fluids via closing the membrane ruffles and forming large vesicles (macropinosomes) [75].



Figure 2. Different mechanisms of endocytosis. Adapted from [76]. Open access.

3.3. Polysaccharide Based-Nanoparticles and their Impact on Cellular Uptake

The biological response of the nanoparticles is measured in animal cell culture studies before starting in vivo administration [77]. In a biological system, nanoparticles can easily pass through the membranes and most biological barriers, and easily penetrate the body, cells, organs, and tissues, presenting problems such as potential toxicity [8]. The physicochemical characteristics of the nanoparticles contribute to the cellular uptake mechanisms. The smaller sizes of the nanoparticles allow them to internalize capillaries and thus easily target the cell, providing greater therapeutic efficiency and less toxicity. Materials internalized by endocytosis are delivered to lysosomes and are subject to degradation after the cleavage of the budding vesicles. Delivery of nanoparticles into lysosomes via clathrin-mediated endocytosis has been a widely used route for drug delivery [78]. The intracellular disposition of chitosan nanoparticles in macrophages has been previously studied. Chitosan nanoparticles with a size of 250 nm were prepared and collocated in a murine macrophage cell line (RAW 264.7) as a model macrophage. The results showed that the nanoparticles follow two mechanisms of macrophage (clathrin-mediated endocytosis pathway and phagocytosis) internalization, indicating cellular digestion of the nanoparticles [79]. In another study, alginate-based nanoparticles were fabricated. The in vitro study was performed in Caco-2 cells. The results showed that nanoparticles with a size of 50/120 nm internalized via clathrin-mediated endocytosis, and the mechanisms for nanoparticles with sizes of 420 and 730 nm were caveolae-mediated endocytosis and macropinocytosis, respectively [80]. In addition, in a study using chitosan-based nanoparticles (CS-NPs), the CS-NPs showed an average diameter of 281 nm and a negative surface charge (-0.3 mV). The cell experiments indicated that CS-NPs presented higher cell internalization efficiency in Raw 264.7 macrophages. These results show excellent potential for the use of these CS-NPs as a macrophage-targeting drug delivery system in further studies [81].

Nanoparticle behavior and motion in blood flow, membrane adhesion strength, and cell uptake pathways and effectiveness are influenced by particle shape. The surface charge of nanoparticles can also influence the uptake mechanism. Positively charged nanoparticles can be successfully absorbed by cells through electrostatic interactions with the negatively charged cell plasma membrane (clathrin-mediated endocytosis). In contrast, negatively charged nanoparticles are more likely to use caveolae-mediated routes [82].

The cytotoxicity of chitosan nanoparticles in Caco-2 cells was evaluated previously [83]. The nanoparticles had sizes of 25 and 333 nm and a Z potential of 5.3 mV. The results showed that in vitro cytotoxicity was mainly influenced by the particle size and not by its surface charge. The Caco-2 cells were more sensitive to cytotoxicity with 25 nm nanoparticles because these were able to internalize the cells. Similarly, silica nanoparticles (SiNPs) coated with chitosan (CH) and alginate (A) for the encapsulation of insulin have been previously reported [13]. The SiNP-CHs were synthesized with sizes of ~600 nm and with a Z potential of -13 mV, and the SiNP-A presented sizes of ~390 nm and a Z potential of +19 mV. Cytotoxicity was evaluated by the viability of HepG2 and Caco-2 cells. At concentrations of 100 μ g/mL, nanoparticles coated with chitosan and alginate did not show toxicity. In the same manner, insulin-loaded chitosan-pectin nanoparticles were developed [84]. Particles with a size of 240–420 nm were obtained, and their cell viability was evaluated using Caco-2 cells. The particles loaded with insulin showed a moderate decrease in cell viability, which could be attributed to the degree of acetylation of chitosan and the presence of insulin; similarly, this did not represent a toxicity problem. In addition, chitosan-g-poly(methyl methacrylate) nanoparticles thiolate by conjugating N-acetyl cysteine were synthesized [85]. The authors obtained nanoparticle sizes ranging from 100 to 330 nm and a positive surface charge due to the presence of free amide groups protonated to chitosan. Cell viability results using Caco-2 and HT29-MTX cells showed that, after 4 h, nanoparticles had viability > 90%, making it acceptable for a non-cytotoxic material. The permeability studies showed that the chitosan-based nanoparticles penetrated the cell monolayer in which mucus is present.

4. Colon

The colon, or large intestine, extends from the ileum to the rectum, with a length of approximately 1.5 m and an average diameter of 6 cm. It is composed primarily of the cecum, ascending, transverse, descending, and sigmoid colons, the rectum, and anus. A complex layer of mucosa covers the colon. The mucus secreted by the large intestine mucosa is responsible for protecting the intestinal wall from excoriation and bacterial activity. It provides the medium for agglutination of feces [86,87]. The colon's main function is to absorb the remainder of the water and salts secreted from the small intestine. It is responsible for bacterial fermentation, synthesis of small amounts of vitamins, storage, and excretion of feces. The microbiota that resides in the colon ferments the dietary fiber from the carbohydrates in the diet. The products of its digestion are gases (e.g., hydrogen, carbon dioxide, nitrogen, and sometimes methane) and short-chain fatty acids (e.g., acetic, propionic, butyric, and lactic acid) [88,89]. The pH from the cecum to the rectum varies from 5.5 to 7; the change in pH is influenced by the diet and people's health status [22]. The colon is inhabited by abundant microbiota (400 different species) of aerobic and anaerobic microorganisms. The microbiota has a significant impact on the host physiology, energy homeostasis, immune system, digestion, and the synthesis of vitamins. The colonic microbiota is mostly anaerobic, and the main species belong to the genera Bacteroides, Bifidobacterium, Eubacterium, and Lactobacillus. The colon harbors a population of 1011 bacteria per gram of intestinal content. These microorganisms produce a large

panel of enzymes responsible for breaking dietary fibers that are not digested in the small intestine [88,90,91].

The colon does not have villi as observed in the small intestine; instead, it has a crescentic mucosal fold called the *plica semilunaris* which increases the internal surface area of the colon by about 1300 cm^2 . The colon is composed of four prominent layers: (1) Serous (squamous epithelium) is found in regions of the colon with a peritoneal surface. (2) Muscular regions are groups of autonomous muscles that control movement and mixing within the colon. (3) Submucosa comprises connective tissue containing blood and lymphatic vessels, fat cells, and nerve ganglia. The ganglia communicate with the innermost layer of the colon: the mucosa. (4) The mucosa is subdivided into three layers: the muscularis mucosa, lamina propria, and epithelium (Figure 3) [86]. The epithelium secretes the mucosa from the stomach to the colon. This secretion forms an adherent gel (mucus) on the surface of the mucous membranes. Its function is to act as a protective barrier between the underlying epithelium and the lumen due to harmful agents, destructive hydrolases, and microorganisms. It also acts as a lubricant to facilitate the passage of digestive material, protecting the underlying epithelium from mechanical stress and providing an essential environment for the enteric microbiota. The colon mucus is a very sticky translucent gel with a thickness of 830 μ m [92].



Figure 3. Histological layers of the colon. Adapted from [93]. Rights holder: © (2021) Terese Winslow LLC, U.S. Govt. has certain rights.

5. Colon-Targeted Drug Delivery

Colon-targeted drug delivery is widely studied for the local treatment of various diseases, such as ulcerative colitis, colonic pathologies [60] colon cancer [94], and in the systemic administration of drugs, including proteins and peptides [95]. The administration of drugs and peptides for colon-targeted delivery is a challenge because they have to pass through the upper GI tract and not be absorbed. Thus, it is imperative to protect the drug during its journey through the GI tract [96]. The colon has been studied as a suitable site for the absorption of drugs, peptides, and proteins because it has a neutral pH and less diversity of digestive enzymes. In addition, the mucosa of the colon facilitates the absorption of several drugs and the colon's proteolytic activity is less than that in the small intestine. The colon also has a long residence time of 5 days [90].

Polysaccharide-based nanoparticles protect drugs, peptides, and proteins from hydrolysis and enzymatic degradation in the duodenum and jejunum, thus enabling proper release of the drug in the ileum or colon, improving its systemic bioavailability [97]. The administration of a drug to the colon can be oral or rectal [98]. Oral administration is the most convenient and preferred route for people because it is painless, non-invasive, and easy to handle [94]. The best candidates for colon-targeted delivery are drugs that show poor stomach absorption to the intestine (peptides and proteins) [99]. The development of formulations and new technologies, such as nanoparticles, for the delivery of drugs, peptides, and proteins depends on the physicochemical nature of the drug, the biomaterial, and the therapy. Carriers such as biopolymer nanoparticles can be used as matrices, hydrogels, or coating agents. This type of carrier must protect the drug, peptides, and proteins from the challenging conditions of the stomach and intestine, increase intestinal absorption in the bloodstream, allow specific cells in the human body to be reached, and ensure controlled release [100,101].

6. Factors That Influence Colon-Targeted Drug Delivery

The colon as a drug delivery site offers excellent advantages due to its pH, residence time, and low enzyme activity (Figure 4). The administration of drugs to the colon depends on its physiological factors to ensure optimal efficiency after oral administration. The nanoparticles must be manufactured considering their residence time in the GI tract, the environment (pH, microorganism, food, etc.), and the intestinal fluid volume [102]. For oral administration of drugs, peptides, and proteins, it is necessary to cross multiple barriers present in the GI tract, which are designed to break down nutrients and prevent the entry of pathogens. The barriers that influence the arrival and absorption of drugs, peptides, and proteins in the colon are classified into chemical, enzymatic, and mucus barriers [15]. Each is described below, and nanoparticle studies are shown in Table 3.



Figure 4. Conditions in the GI tract. Reprinted by permission from John Wiley and Sons [103] Copyright (2021).

Polysaccharide	Drug	Barrier	Size (nm)	Mechanism	Study	Fabrication Technique	Reference
HA-GA/HA-His	Doxorubicin	Chemical	147.5-607.6	pH responsive	In vitro	Ultrasonic dispersion	[104]
Chitosan/Pectin	Insulin	Chemical	240-420	pH responsive	In vitro	Electrostatic self-assembly	[84]
Chitosan-Heparin	Oligonucleotides	Chemical	145	pH responsive	In vitro	Spontaneous polyelectrolyte complexation	[105]
Fucoidan/Chitosan	Quercetin	Chemical	300-400	pH responsive	In vitro	Polyelectrolyte self-assembly	[106]
Q-AmCs	Curcumin	Chemical	162	pH responsive	In vitro	Ionic gelation	[107]
Alginate-Chitosan	BSA	Enzymatic	320-340	Enzymatic	In vitro	Nano-emulsion	[108]
Chitosan	Resveratrol	Enzymatic	115	Enzymatic	In vitro	Synthesized block-copolymer	[109]
Modified Pectin		Enzymatic	64.11	Enzymatic	In vitro	Sonication	[110]
						Double emulsion solvent	
COS-PLGA	BSA	Mucoadhesive	170.7	Muchoadhesiveness	Mucoadhesive strength	evaporation by	[111]
						homogenization	
ACS		Mucoadhesive	356	Muchoadhesiveness	Mucoadhesive strength	Ionic gelation	[44]
Pectinate-Chitosan	Curcumin	Mucoadhesive	218.1	Muchoadhesiveness	Mucoadhesive strength	Ionic gelation	[112]
Thiolated Chitosan	Sitagliptin	Mucoadhesive	160.3	Muchoadhesiveness	Mucoadhesive strength	Ionic gelation	[113]
Chitosan	Insulin	Enzymatic	220	Enzymatic	In vivo	Ionic gelation	[114]
Chitosan	TDF	Enzymatic	156	Enzymatic	In vivo	Ionic gelation	[115]
Chitosan	Insulin	Chemical	45 and 115	pH-responsive	In vivo	Ionic gelation	[116]
Chitosan+Mucin	Insulin	Mucoadhesive	504.1	Muchoadhesiveness	In vivo	Self-gelation	[117]
Chitosan/Alginate	Naringenin	Chemical	150-300	pH responsive	In vivo	Ionic gelation	[118]
Chitosan	IFNα	Chemical	36	Epithelium	In vivo	Ionic gelation	[119]
PEGylated Chitosan	Rosuvastatin	Chemical	<200	Epithelium	In vivo	Mediated reaction	[120]
Chitosan	Vancomycin	Chemical	220	pH-responsive	In vivo	Ionic gelation	[121]
Succinyl Chitosan/Alginate	Quercetin	Chemical	90	pH-responsive	In vivo	Ionic crosslinking	[122]
Chitosan/Alginate	Lovastatin	Chemical	50-100	pH-responsive	In vivo	Ionic gelation	[123]
Chitosan-AntBiop		Enzymatic	162.7	Enzymatic	In vivo	Ionotropic gelation	[124]
Thiolated Hyaluronic Acid	Insulin	Mucoadhesive	75	Muchoadhesiveness	In vivo	Ring-opening reaction	[125]
Hyaluronic Acid	DOX	Chemical	238.1 to 156.7	pH-responsive	In vivo	Two-step reaction	[126]
Chitosan-modified	Curcumin	Chemical	281	Epithelium	In vivo	Ionic gelation	[81]

Table 3. Polysaccharide-based nanoparticle colon-specific drug delivery.

Q-AmCs: Quaternized aminated chitosan, HA–GA/HA–His: Hyaluronic acid-l-histidine conjugate, Chitosan-AntBiop-NPs: Chitosan-Antihypertensive Biopeptides Nanoparticles, BSA: bovine serum albumin, Cur/ALG-GANPs: Curcumin loaded gum Arabic-sodium alginate nanoparticles, COS-PLG ANPs: Chitosan oligosaccharide and poly (D, L-lactide-co-glycolide) (PLGA) NPs, ACS: Acrylated chitosan, TDF: Tenofovir Disoproxil Fumarate, IFNα: Interferon-alpha, PEG: polyethylene glycol.

6.1. Chemical Barrier

In colon-targeted oral drug delivery systems such as drugs, peptides, and proteins are exposed to aggressive environments when crossing through the stomach and small intestine until they reach the colon. The first barrier they must cross is chemical. The pH along the GI tract destroys the administered molecules, reducing their effectiveness. As previously mentioned, the colon has a neutral pH (5.5–7), unlike the stomach and small intestine. The GI tract is divided into the upper and lower tract. The upper tract includes the oral cavity, pharynx, esophagus, stomach (pH 1 and 3.5), and the small intestine's initial section (duodenum, pH 6). The lower GI tract includes the remainder of the small intestine (jejunum, pH 6 and ileum, pH 7), and the segments of the large intestine (cecum, colon, and rectum) [22,127]. Polysaccharide-based nanoparticles can be degraded in the stomach by electrostatic interactions, degrading the polymer network and destroying the nanoparticle. Therefore, for these conditions, the approach is to design pH-dependent nanoparticle systems using resistant matrices and enteric coatings to prevent drug release in the upper GI tract [49].

The fabrication of chitosan and guar gum-based nanoparticles to encapsulate curcumin was previously reported [128]. Highly charged, monodisperse, hydrophilic colloidal nanoparticles were formed with an average diameter range of 250–290 nm. Under gastric stimulation conditions using a simulated gastric fluid (SGF) (2 g/L sodium chloride; 3.2 g/L pepsin; pH 1.2), the nanoparticles released 16% of the load after 3 h and remained stable, preserving their structural integrity. The release of nanoparticles in this system was controlled by the diffusion and swelling of the particles, which was presented in an acidic environment due to the possible protonation of the chitosan amide group. In the simulated intestinal fluid (SIF) (6 g/L potassium dihydrogen phosphate; 3.2 g/L pancreatin; 5 g/L bile salts; pH 7.0), the release of curcumin of the nanoparticles was 43% after 2 h of incubation. This could be attributable to the decrease in the solubility of chitosan at an intestinal pH close to its pKa (6.5) and the swelling of guar gum at pH ~7, resulting in greater porosity in the nanoparticles which resulted in the release of curcumin. In another study, polylactic-co-glycolic acid (PLGA)-modified nanoparticles using chitosan were manufactured [129]. The size of the nanoparticles increased from 132.8 nm to 172.7 nm due to the use of chitosan. The increase in size helped to increase the encapsulation efficiency from 65.8 to 87.1%. In vitro studies showed that modified nanoparticles responded at pH 5.5 by accelerating drug release. In addition, the development of modified nanogels based on soy protein and dextran to encapsulate riboflavin has been studied [130]. Morphological characterization analyses showed spherical core-layer structures, with sizes within the range of 32 to 40 nm. The size allowed an encapsulation efficiency of 65.9%. In vitro studies simulating GI tract conditions showed that these nanogels are stable in SGF and presented greater release in SIF.

6.2. Enzymatic Barrier

Enzymatic degradation is a challenge for polysaccharide-based nanoparticles. Some polysaccharides are usually susceptible to various digestive enzymes found in the GI tract. The breakdown of lipids and proteins occurs in the stomach by gastric lipases and pepsins, and continuous degradation in the small intestine by pancreatic enzymes (lipase, trypsin, and elastase) [127]. The polysaccharides that are not degraded by gastric and intestinal enzymes are degraded and fermented in the colon by the colonic microbiota. This degradation reduces polysaccharides to oligosaccharides, and their fermentation results in the production of metabolites and short-chain fatty acids [131]. Polysaccharides are often used to deliver drugs by film coating and matrix formation. Matrices based on non-starch polysaccharides are not decomposed in the stomach because anaerobic bacteria ferment them in the colon [54]. These microbiota-activated systems have shown promise in colon-targeted drug delivery due to anaerobic microbiota and their specific enzymatic activity. Polysaccharides fermentable by these microorganisms and used to manufacture drug delivery systems are pectin, guar gum, inulin, chitosan, and arabinoxylan. These

polysaccharides can resist being passed through the upper GI tract and be metabolized in the colon to release the drug, peptides, or proteins [97].

In a previous study, citrus pectin-based nanoparticles were coated using Eudragit S100 [132]. These nanoparticles were designed to release 5-fluorouracil (5-FU) in the colon to attack colorectal cancer and presented a spherical shape, an average size of 218.12 nm, and a Z potential of 27.5 mV. In vitro matrix degradation studies showed that nanoparticles were degraded by pectinase, and the drug release started after 4 h in SIF (pH 6.8). In vivo studies to evaluate site specificity showed that a smaller quantity of the drug was lost in the upper GI tract. An increase in the concentration of 5-FU was observed in the colon due to degradation by its microbiota.

6.3. Mucus Barrier

The mucus present in the small and large intestine is a sticky, elastic, and viscous layer. This layer is produced by goblet cells and is responsible for protecting epithelial cells by capturing hydrophobic molecules; it also protects them from physical damage caused by eating food. Also, the mucus layer is water and mucins, which contains salts, bacteria, carbohydrates, enzymes, and immunoglobulins. These components, which are coated with proteoglycans, give the layer a negative charge [133]. Although the pore sizes of mucus gels are around 100 to 200 nm, it has been confirmed the mucus layer can entrap molecules with high molecular weight and low permeability (e.g., insulin) and positively charged nanoparticles with a hydrophobic nature. Hence, this is one of the obstacles that nanoparticles must cross when drugs are administered orally. Many nanomaterials tend to be immobilized by the mucus layer and cannot reach the intestinal epithelial cells of the colon [133,134]. Polysaccharides are mucoadhesive biomaterials due to their hydrophilic groups (hydroxyl, carboxyl, amide, and sulfate), which bind to mucus by hydrogen bonding and hydrophobic or electrostatic interactions [44,98]. The negatively charged mucosa attracts positively charged nanoparticles, such as cationic polymers (chitosan and derivatives). Nanoparticles coated with cationic groups or thiol groups have a longer residence time in the GI tract. The mucoadhesive and mucopenetration properties of nanoparticles are important for oral administration because mucoadherent particles are removed from mucosal tissue [15].

Mucoadhesion is defined as the adhesion between materials (mucosa-material). Formulations designed for mucoadhesive dosing allow prolonged retention at the site, maintaining the controlled release rate of the drug, therefore increasing its bioavailability. Mucoadhesion can be affected by various factors, such as hydrophobicity, molecular weight, cross-linking, swelling, spatial conformation, pH, and polymer concentration. The transit time of these systems depends on the physiological renewal time of the mucus layer. The intestinal mucin turnover time is between 50 and 270 min, so the particles adhere to the mucus for between 4 and 5 h [92]. In the mucus-penetrating mechanism, nanoparticles must avoid adhering to the mucosa. In the design of mucus-penetrating nanoparticles, the particle easily penetrates the luminal mucus layer to break through the underlying adherent mucus layer. Once the nanoparticles penetrate, they are closer to the cells, and a higher dose of the drug is exposed to later bind to the underlying epithelium, improving drug delivery [134].

The manufacture of a chitosan-based nanoparticle system (NPCS) to encapsulate insulin has been previously reported [135]. The nanoparticles had a size of 253.2 nm and a Z potential of 28.2 mV. In this study, the effect of nanoparticles to adhere to and infiltrate the mucus layer was evaluated. The results showed that the NPCS could penetrate through the epithelial mucus layer and approach the epithelial cells. In another study, insulin-loaded polyurethane-alginate (PU-AG)-based nanoparticles were evaluated as a model for oral administration [136]. The nanoparticles reached sizes of 73 nm and a Z potential of -27.3 mV. Ex vivo mucoadhesion studies report that these nanoparticles presented high mucoadhesiveness and were strongly attached to the intestinal lumen. This resulted from the mucoadhesive property of alginate due to its carboxyl groups and the negative charge of the nanoparticle.

6.4. Absorption Colon-Specific Drug Delivery

The transport of drugs across the colonic mucosa depends on different factors, including colonic residence time, susceptibility to destruction during uptake, diffusional characteristics, colonic motility, and integrity of the epithelium. Alterations on the mucus layer develop various pathological conditions affecting colonic delivery; thus, the colon requires an intact mucous layer (Figure 5) [102]. The absorption mechanism for oral drugs is complex because they must be soluble in gastric fluid and be absorbed by the stomach, small intestine, or colon [137]. There are two main routes for oral drug absorption in the colon: the transcellular and paracellular routes. The transcellular route is the method by which most drugs are successfully delivered in the colon. Small molecules readily diffuse across colonic epithelial cells by a series of partitions between membrane lipids and aqueous environments. The greatest passive absorption of the drug occurs in the proximal colon due to the lipid fluidity of the membrane in the proximal colonocytes. The active transcellular route initiates drug uptake at the luminal surface through carrier-mediated uptake, receptor-mediated endocytosis, or pinocytotic endocytosis. Paracellular transport is controlled by the sieving action of the intercellular occluding junctions found at the apical neck of cells in the GI tract. This barrier is cation selective, limits the transit of larger compounds, and provides a barrier for the uptake of pathogens found in the lumen of the GI tract. Transcellular ion gradients drive the movement of water and ions through the paracellular pathway. Some drugs have local effects on the GI tract, whereas most are delivered to the bloodstream via systemic circulation to act elsewhere in the human body [96,131].



Figure 5. Barriers to the colonic transport of drugs. Reprinted by permission from John Wiley and Sons [103]. Copyright (2021).

The size of the nanoparticles is an important factor because it affects or benefits their absorption. Studies to assess the specificity of the drug release site can be affected by factors such as the pH of biological fluids, the enzyme system, the interaction between the drug, and the biological environment. Previous studies that discuss this factor (size) include the manufacture of chitosan-based nanoparticles by nano-emulsion, modifying their surface with a penetrating peptide (CPP CS NP) to encapsulate and release insulin [138]. The formed monodisperse spherical particles had ultra-small sizes (12 nm). The nanopar-

ticles were able to penetrate Caco-2 cells in the invitro assay through paracellular and intracellular pathways.

Similarly, the manufacture of calcium phosphate nanoparticles coated with vitamin B12-chitosan and alginate (ViTB12-Chi-CPNPs) to encapsulate and release insulin was previously reported [139]. The nanoparticles were synthesized with particle sizes of <250 nm and Z potential of +32.56 mV. Studies in the monolayer of Caco-2 cells showed greater uptake of ViTB12-Chi-CPNPs. A reduction in transepithelial electrical resistance measurements (TEER) was observed, resulting in the paracellular transport of insulin due to the opening of tight epithelial junctions. In another study, the author's manufactured alginate/dextran-based nanoparticles were loaded with insulin and coated with chitosan and albumin to improve the permeability of insulin through the intestinal epithelium [140]. The nanoparticle sizes obtained were 300 nm. To study the permeability of the nanoparticles, they used Caco-2/HT29-MTX-/Raji B cells. This model mimics the intestinal cellular monolayer due to its content of erythrocytes, goblet cells, and M cells in the Peyer's patches. The results showed that these nanoparticles increased interactions with cell monolayers.

7. Conclusions and Perspectives

The use of polysaccharide-based nanoparticles for oral drug delivery to the colon has become an essential strategy for treating several diseases. The use of polysaccharides to fabricate these nanomaterials improves their biocompatibility and biodegradability properties. Targeted drug delivery offers attractive properties when materials are developed at a nanometric scale because their size is related to the particle's stability, surface area, toxicity, degradation, and absorption mechanisms. Similarly, its size depends mainly on the encapsulated drugs, the type of polysaccharide, and the manufacturing technique. In this review, various studies are discussed, focusing on the physicochemical properties of nanoparticles and their potential to cross the barriers present in the GI, especially in the colon region.

Polysaccharide-based nanoparticles have significant advantages over other nanoparticles (metallic, ceramic, synthetic polymers, etc.). Currently, polysaccharide-based nanoparticles are generally investigated in terms of their physicochemical properties, such as size, surface charge, and encapsulation efficiency, via in vitro and in vivo tests. However, once nanoparticles are administered as drug delivery systems or as part of medical interventions, they tend to present specific interactions with elements of the human body, such as organs, tissues, cells, or hormonal systems. Therefore, it is crucial to investigate the possible side effects of these materials to a greater extent; however, this may significantly impact their use, application, and safety. Also, due to the extensive use and manufacture of polysaccharide-based nanoparticles in the nanomedicine and pharmaceutical areas, more scrutiny is needed of regulatory safety as they continue developing this field. Currently, it is known that polysaccharide-based nanoparticles are considered to be non-toxic, biocompatible, and biodegradable; nonetheless, more in-depth research should be undertaken on the long-term toxicity risks they may pose. Nanoparticle safety is critical because this technology is increasingly playing an essential role in medicine and pharmaceuticals. Although regulatory organizations exist in each country, and each has established its criteria for using nanoparticles, it is necessary to develop international legislation for the applications, safety, and restrictions of nanoparticles. This regulation would open the door to more clinical trials and regulatory standards to set future precedents for the use and benefits of polysaccharide-based nanoparticles.

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