

Supporting Material

Manipulating the conformation of 3,2':6',3"-terpyridine in $[\text{Cu}_2(\mu\text{-OAc})_4(3,2':6',3''\text{-tpy})]_n$ 1D-polymers

Dalila Rocco¹, Samantha Novak, Alessandro Prescimone¹, Edwin C. Constable¹ and Catherine E. Housecroft^{1*}

Department of Chemistry, University of Basel, BPR 1096, Mattenstrasse 24a, CH-4058 Basel, Switzerland

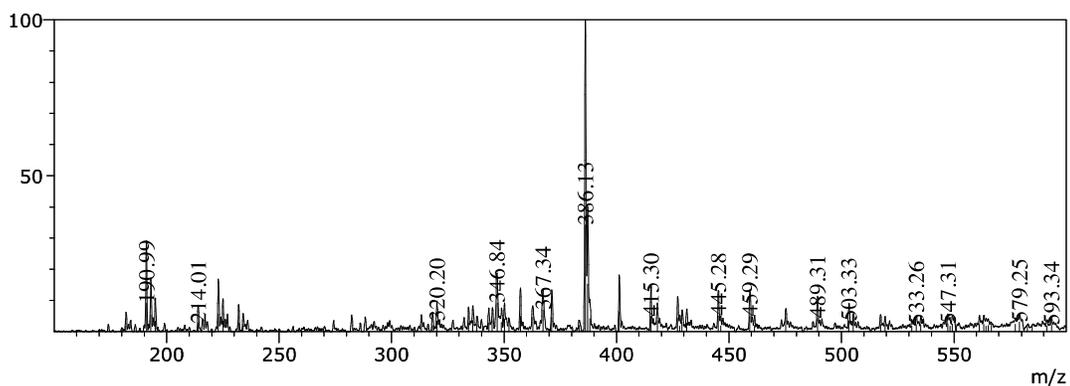


Figure S1. Electrospray mass spectrum of **1**.

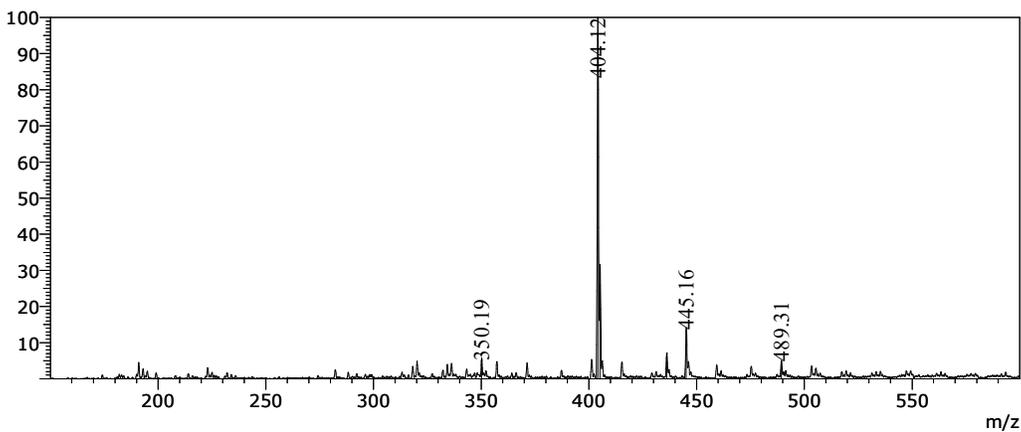
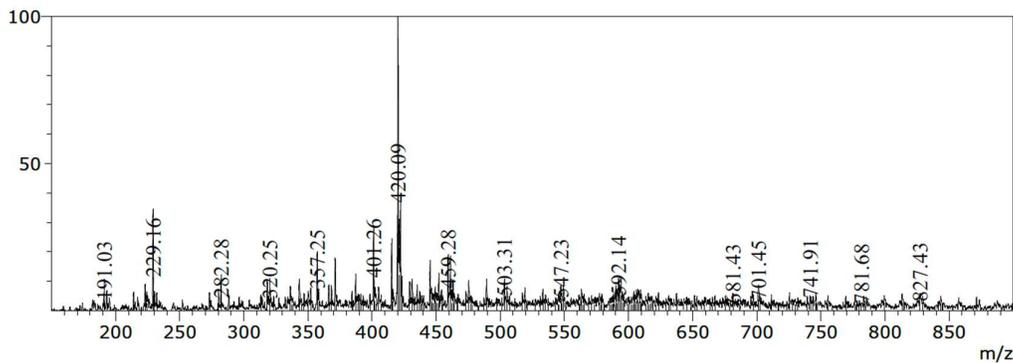


Figure S2. Electrospray mass spectrum of **2**.

MS Spectrum Positive Full Scan
Line#:1 R.Time:----(Scan#:----)
MassPeaks:176
Spectrum Mode:Averaged 0.000-0.133(0-17) Base Peak:420.09(586508)
BG Mode:Averaged 0.567-2.950(69-355) Segment 1 - Event 1



MS Spectrum
Line#:1 R.Time:----(Scan#:----)
MassPeaks:176
Spectrum Mode:Averaged 0.000-0.133(0-17) Base Peak:420(586508)
BG Mode:Averaged 0.567-2.950(69-355) Segment 1 - Event 1

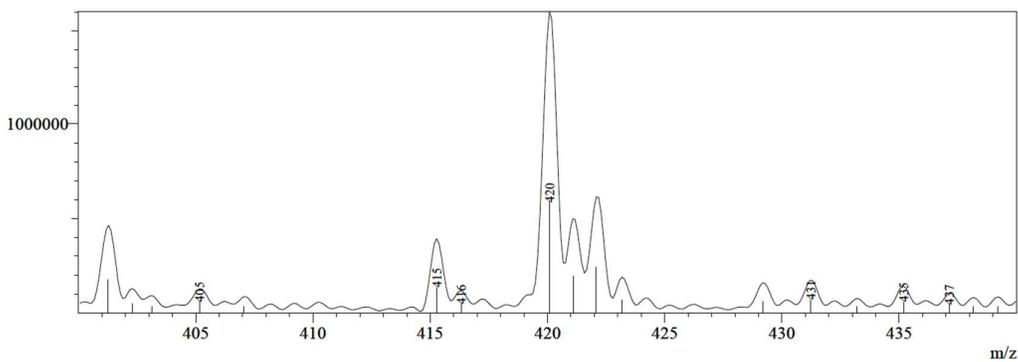


Figure S3. Electrospray mass spectrum of **3**.

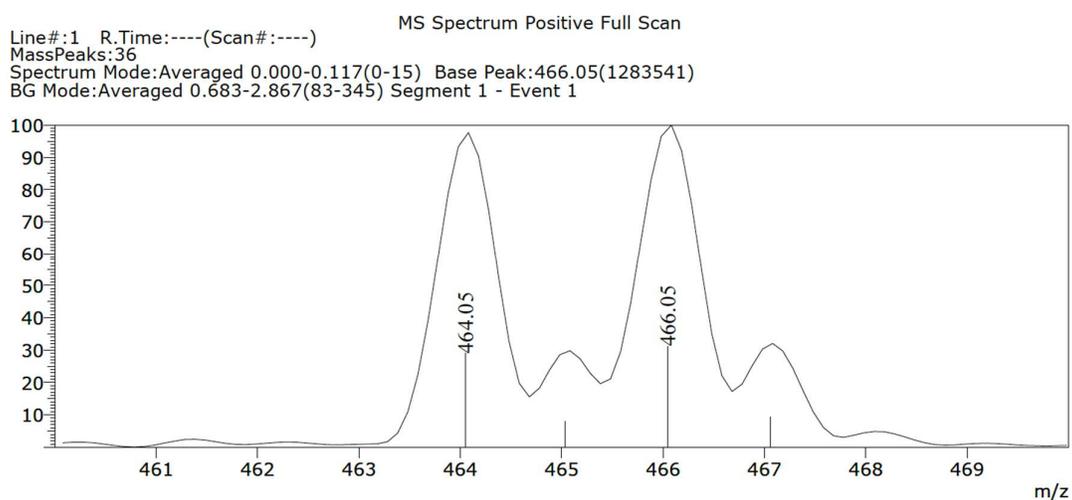
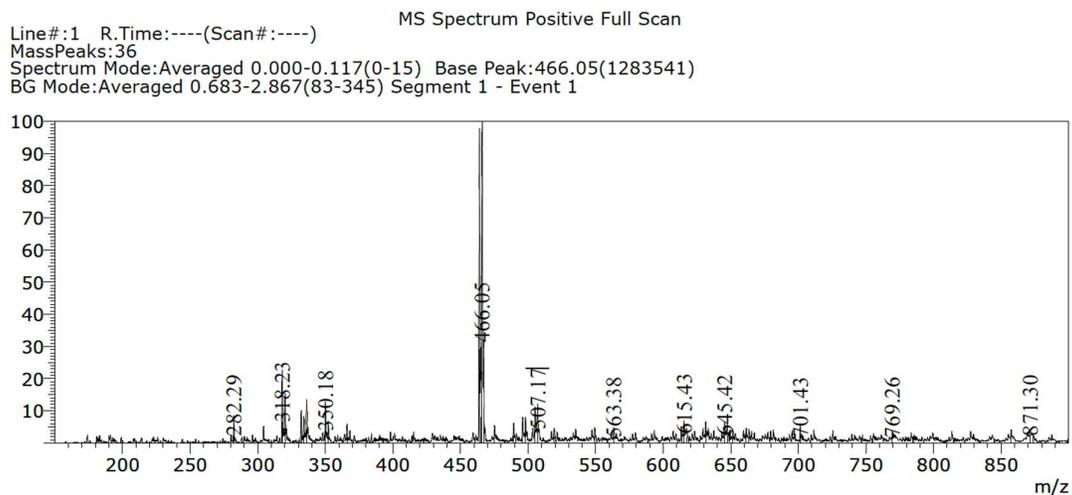


Figure S4. Electrospray mass spectrum of 4.

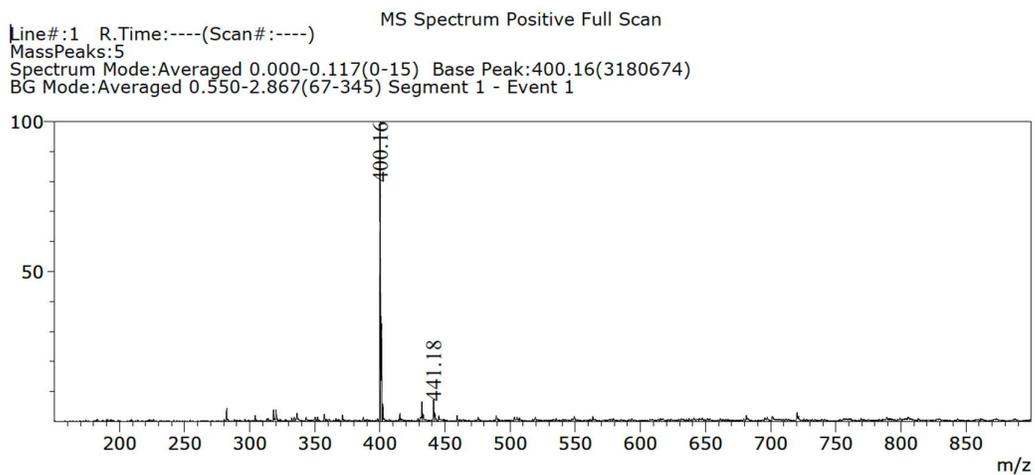


Figure S5. Electrospray mass spectrum of 5.

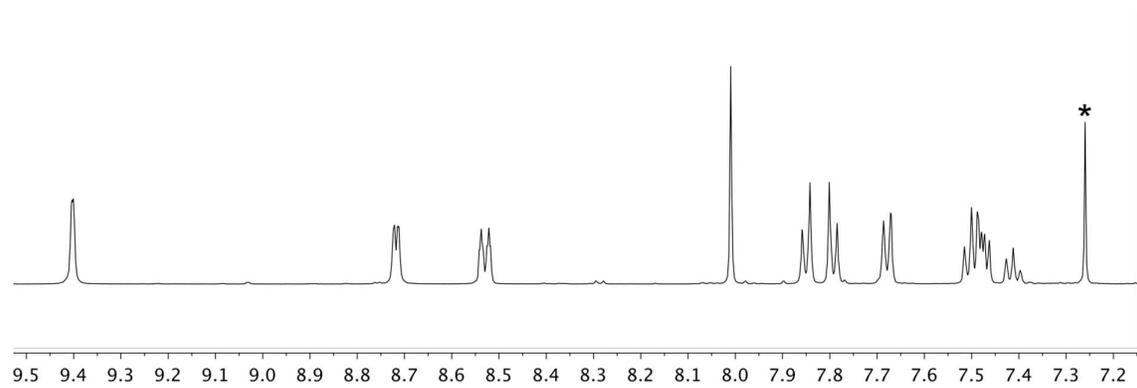


Figure S6. ^1H NMR spectrum of **1** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

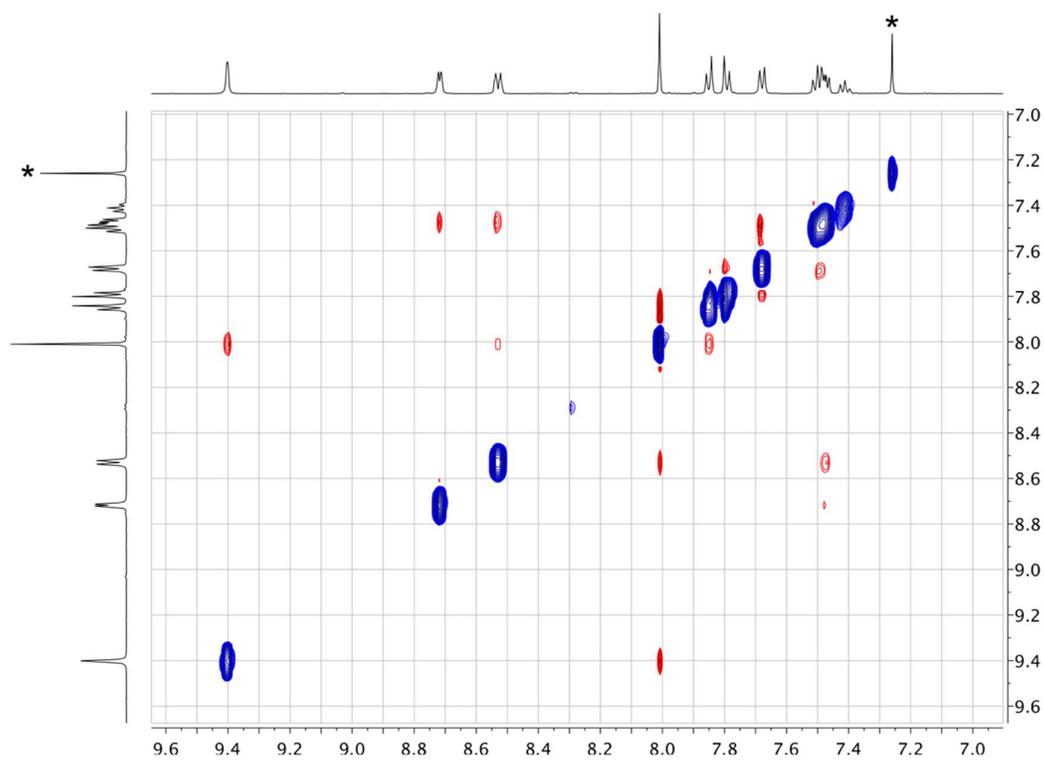


Figure S7. NOESY spectrum of **1** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

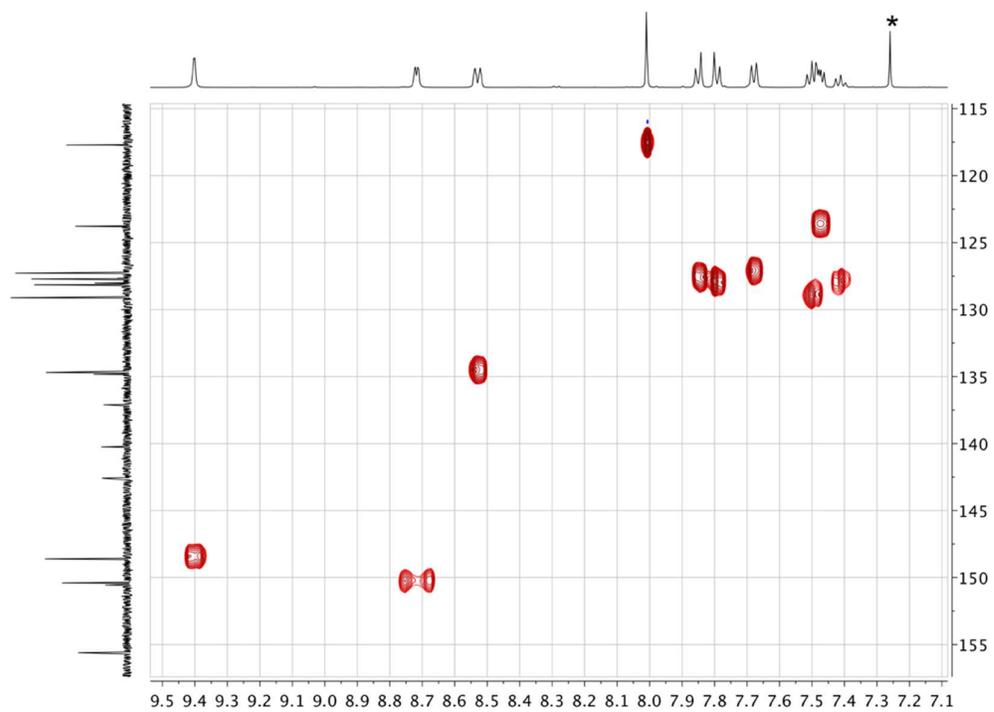


Figure S8. HMQC spectrum of **1** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

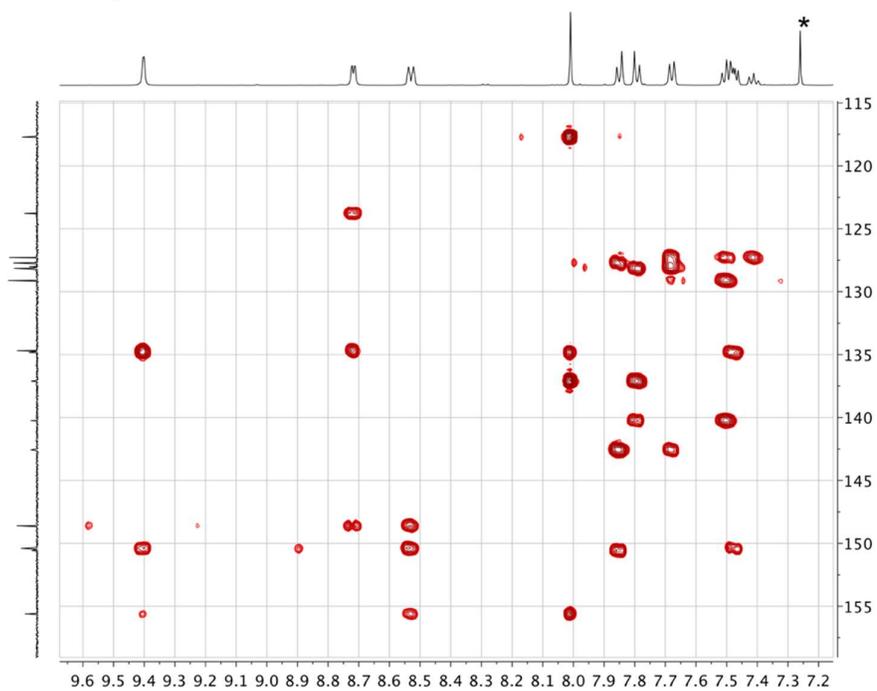


Figure S9. HMBC spectrum of **1** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

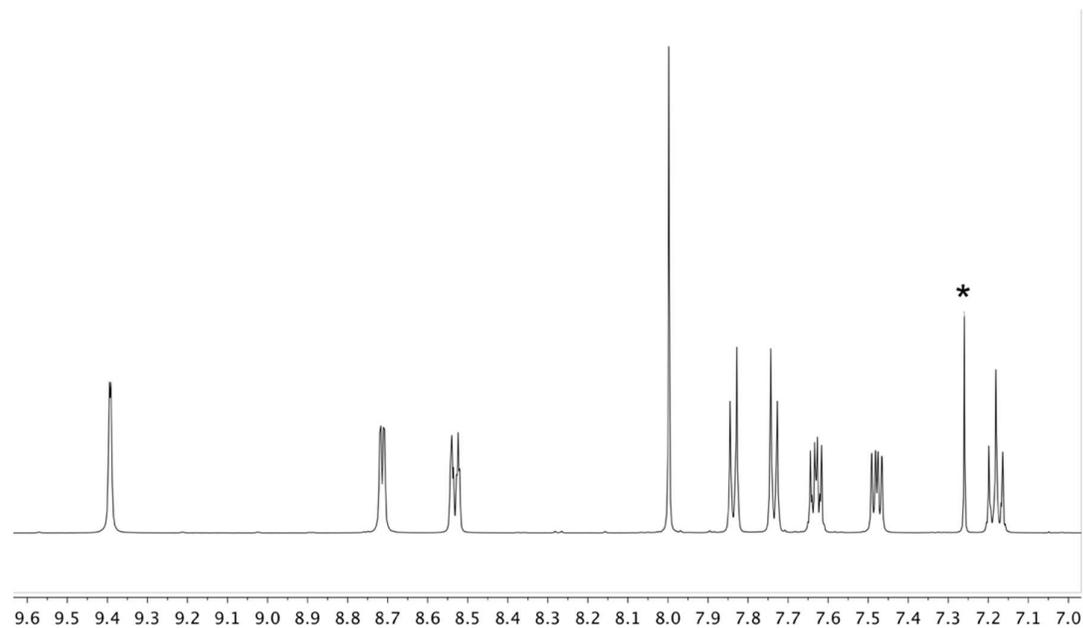


Figure S10. ^1H NMR spectrum of **2** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

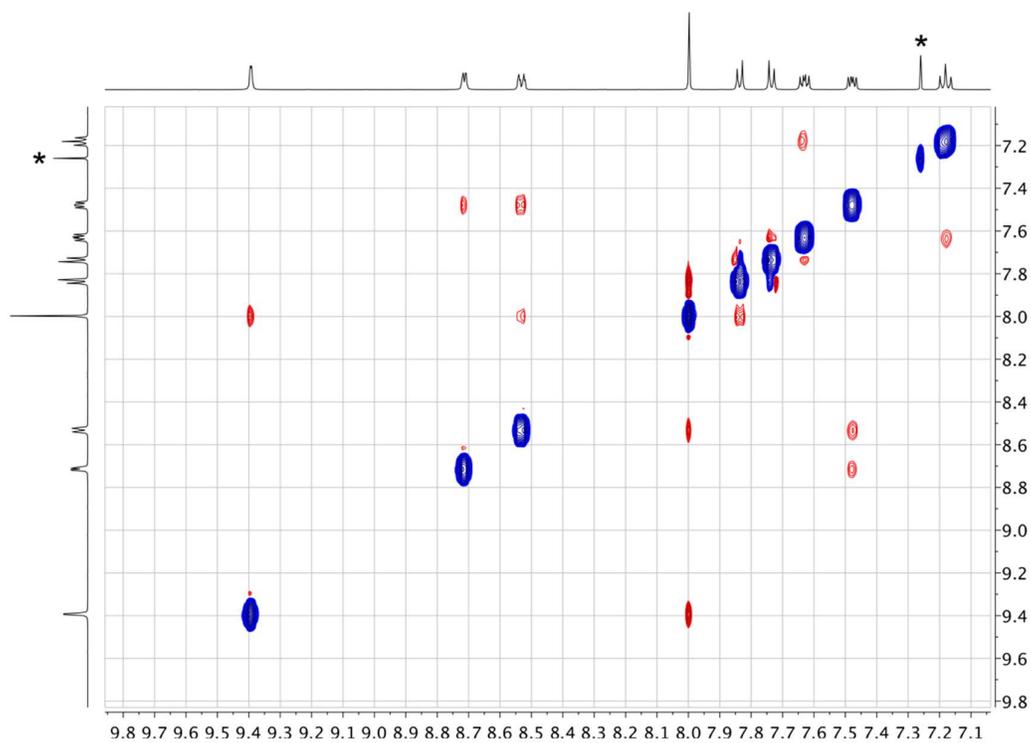
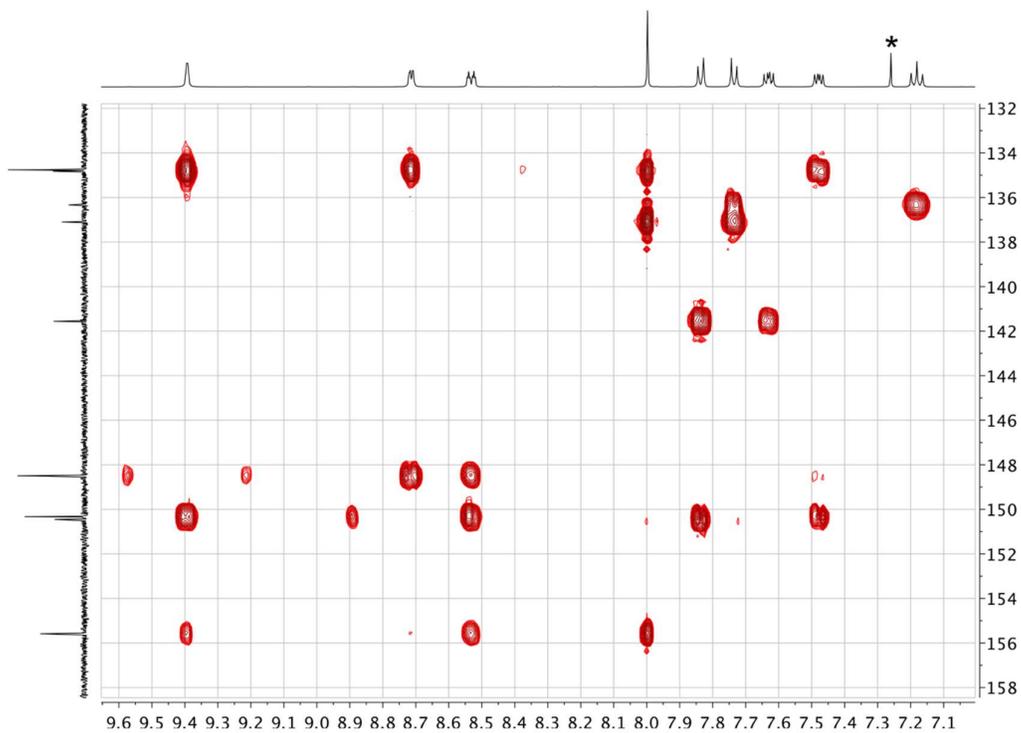
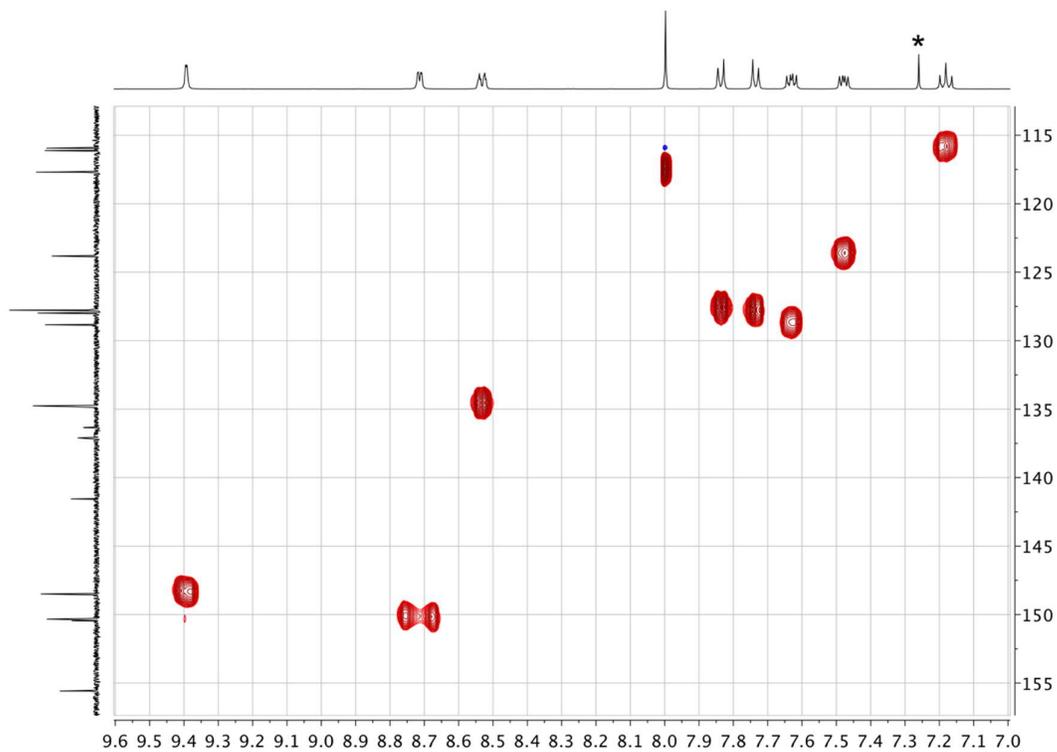


Figure S11. NOESY spectrum of **2** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .



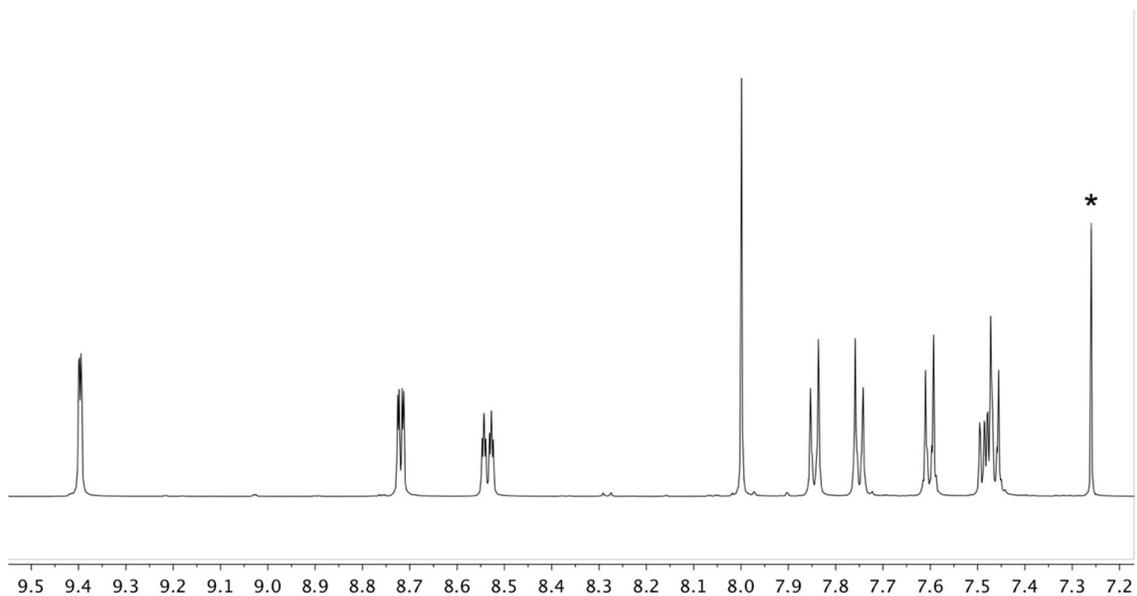


Figure S14. ^1H NMR spectrum of **3** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

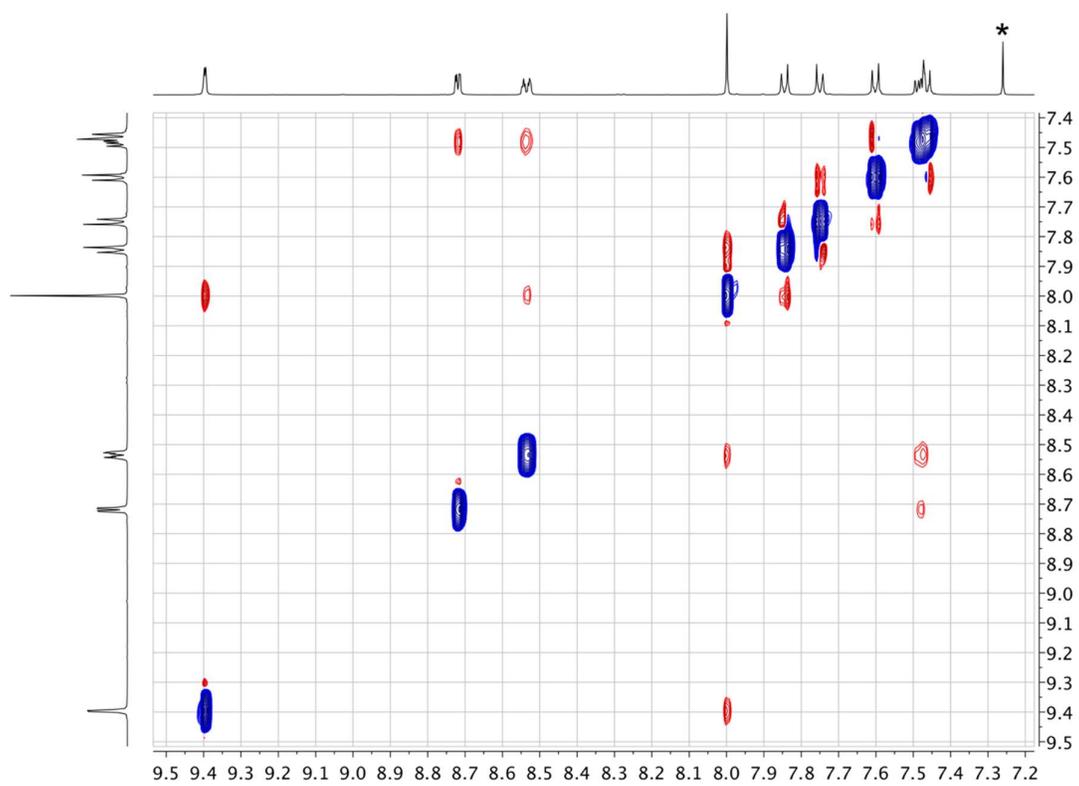


Figure S15. NOESY spectrum of **3** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

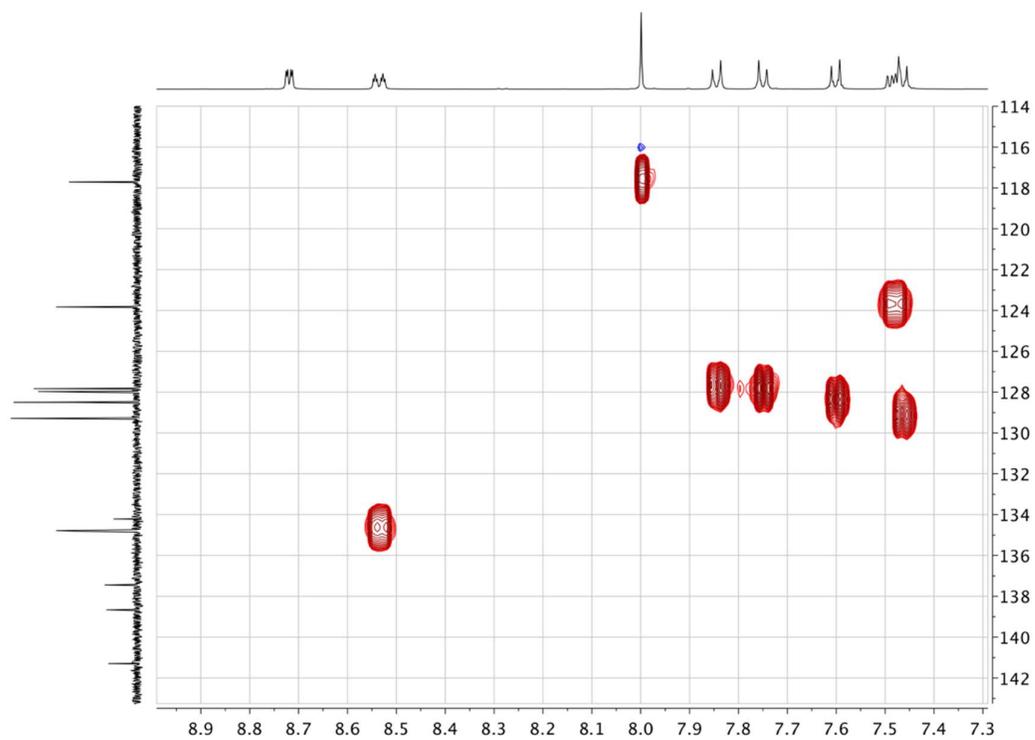


Figure S16. HMBC spectrum of **3** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

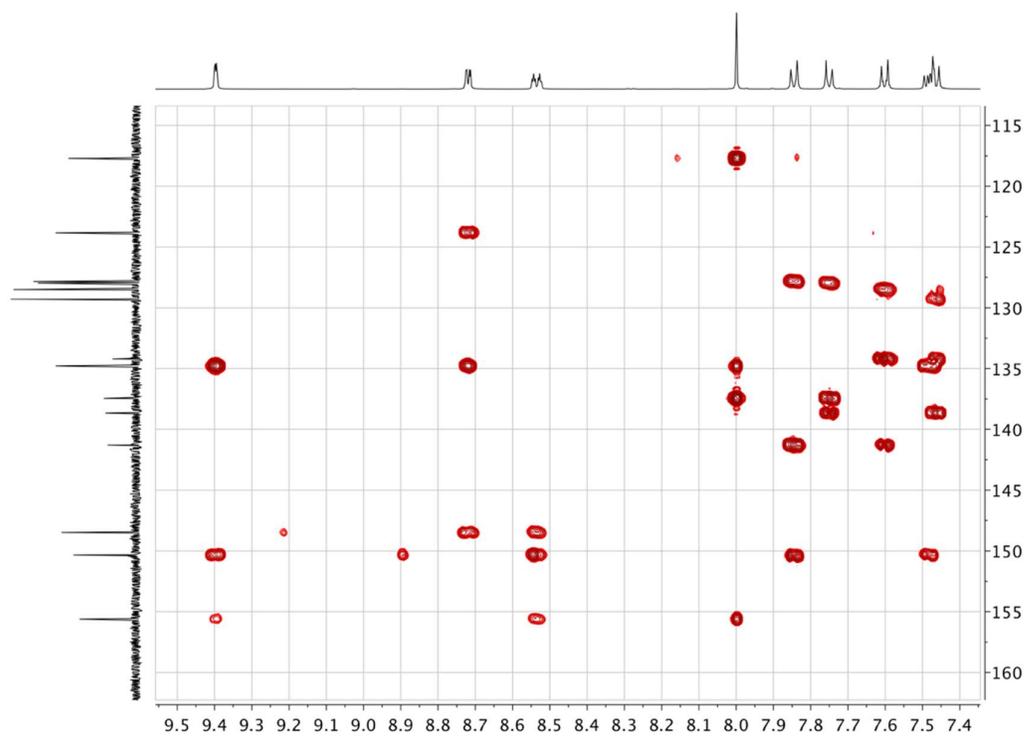


Figure S17. HMBC spectrum of **3** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

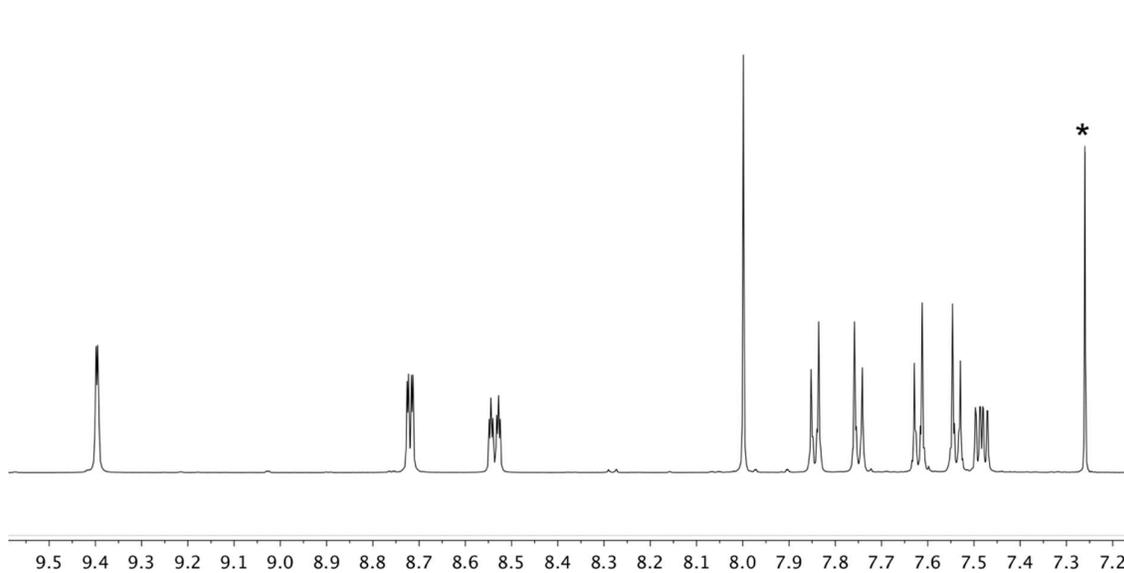


Figure S18. ^1H NMR spectrum of **4** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

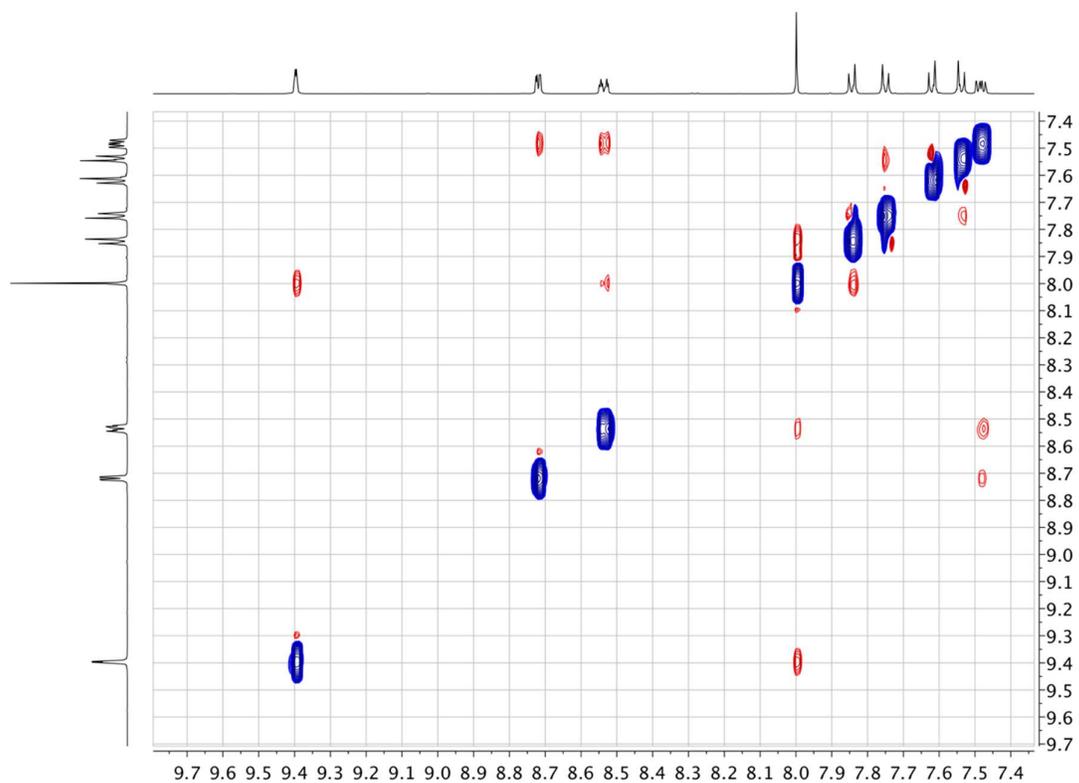


Figure S19. NOESY spectrum of **4** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

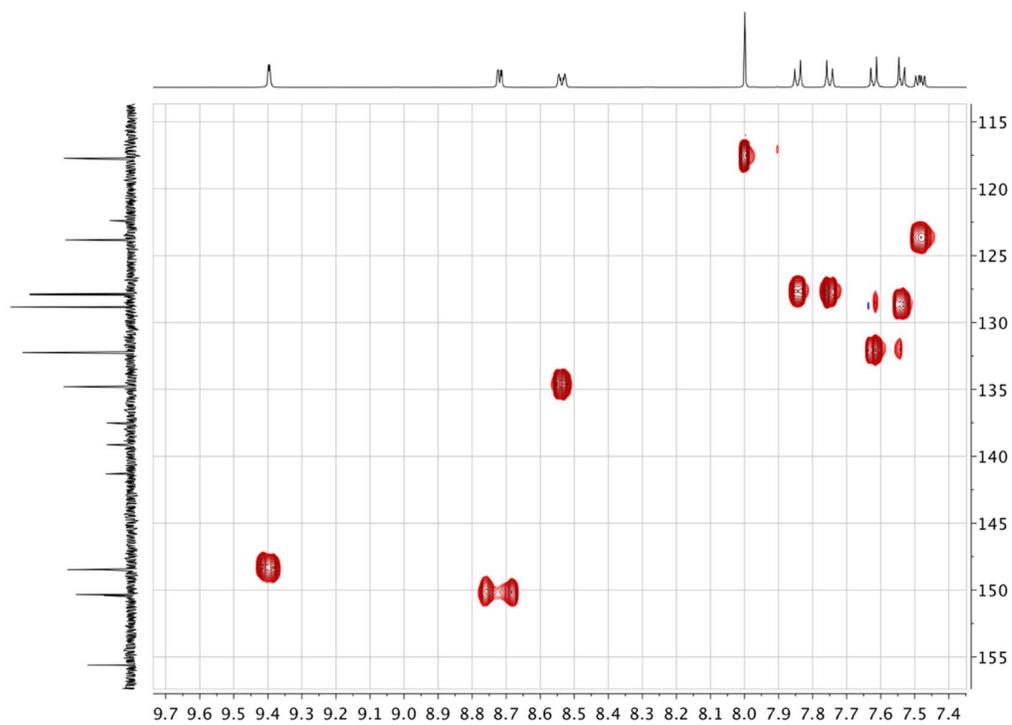


Figure S20. HMQC spectrum of **4** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

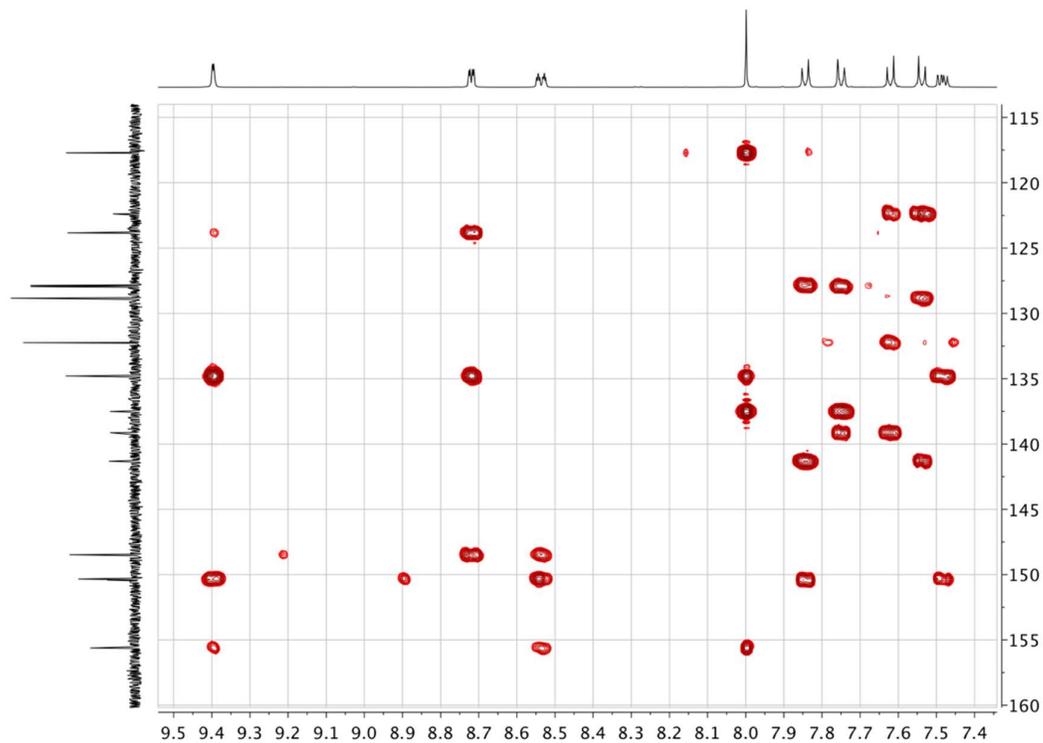


Figure S21. HMBC spectrum of **4** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

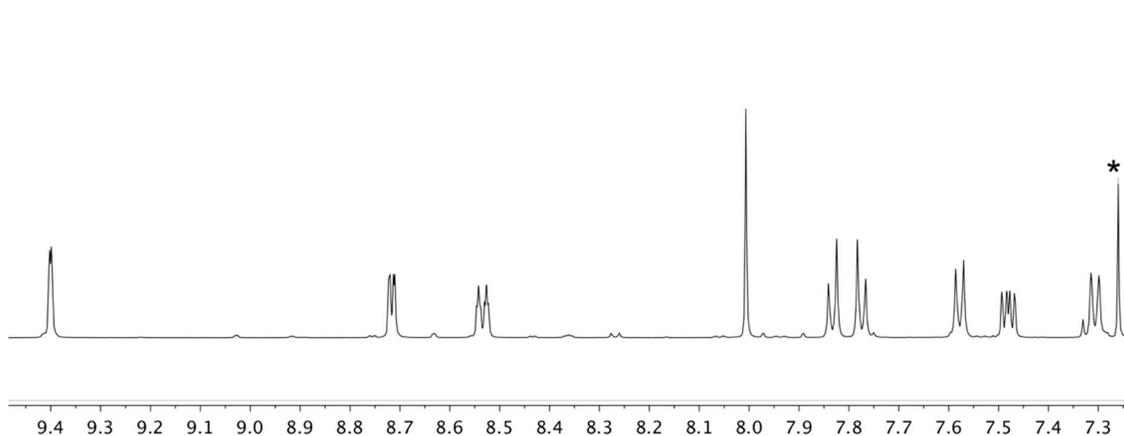


Figure S22. ^1H NMR spectrum (aromatic region) of **5** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

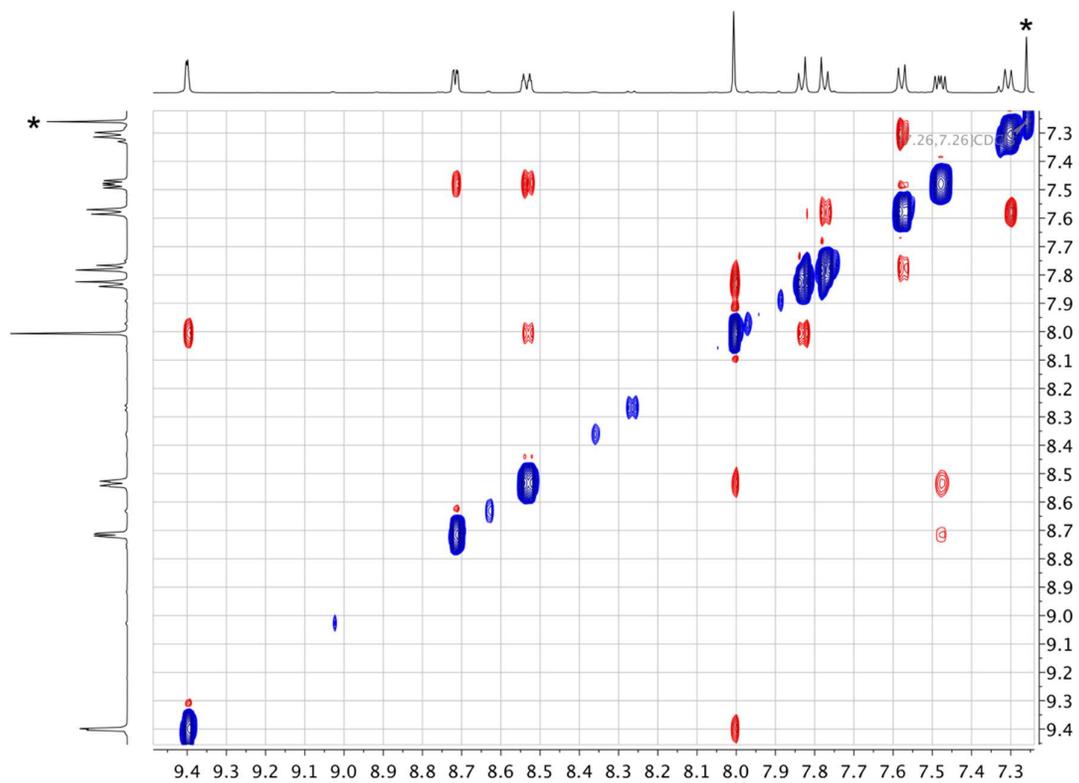


Figure S23. NOESY spectrum (aromatic region) of **5** (500 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

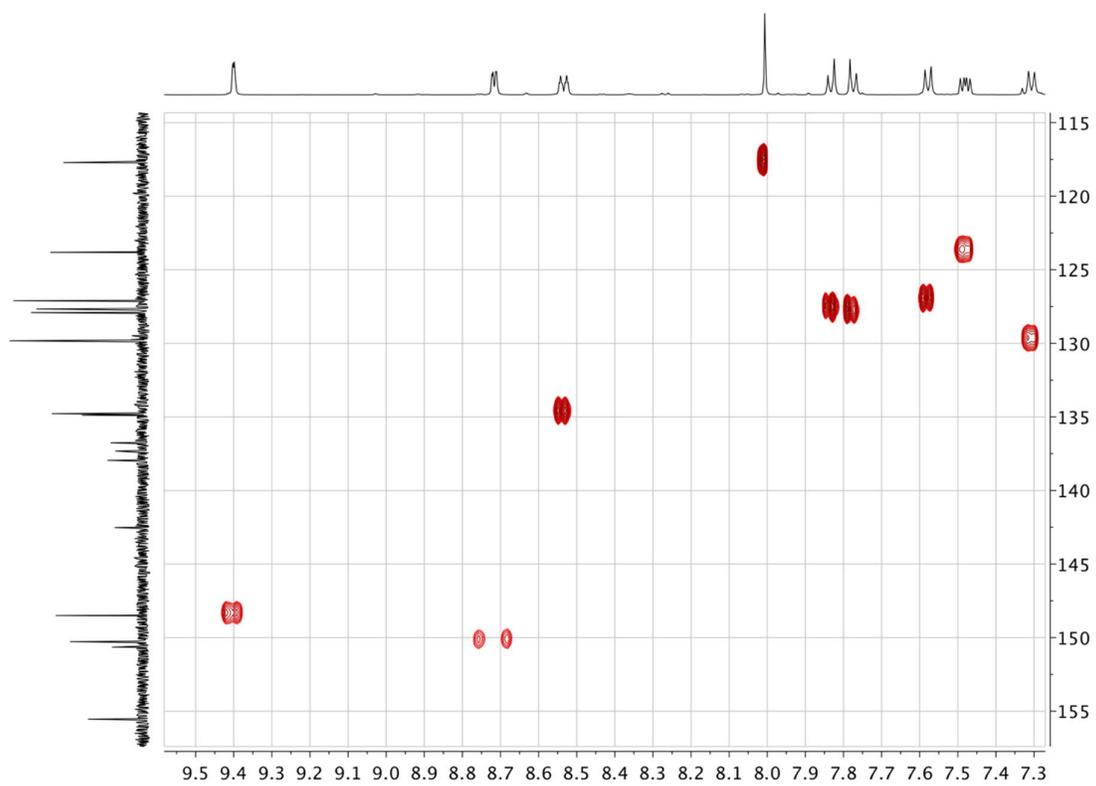


Figure S24. HMQC spectrum (aromatic region) of **5** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

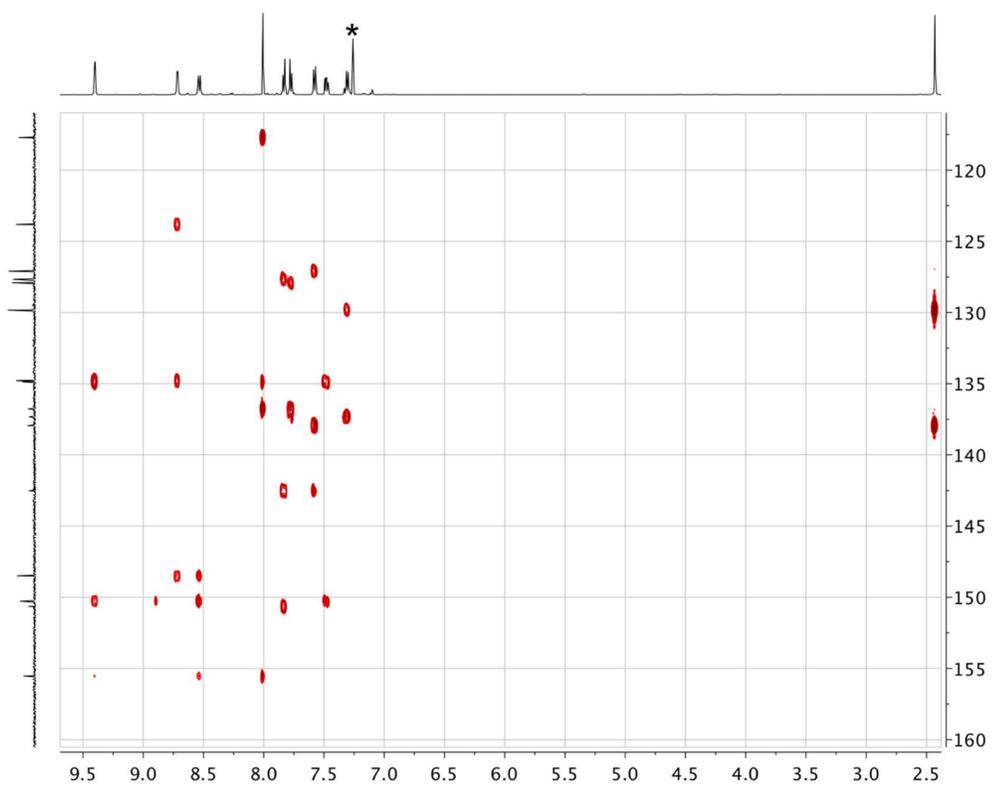


Figure S25. HMBC spectrum of **5** (^1H 500 MHz, $^{13}\text{C}\{^1\text{H}\}$ 126 MHz, CDCl_3 , 298 K). * = residual CHCl_3 .

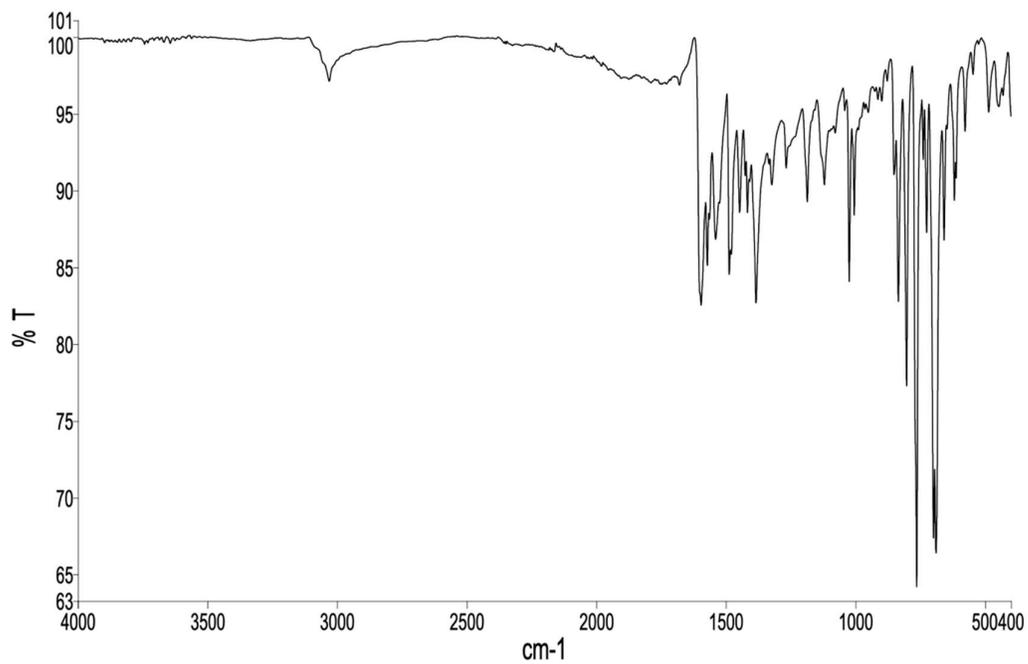


Figure S26. FT-IR spectrum of solid 1.

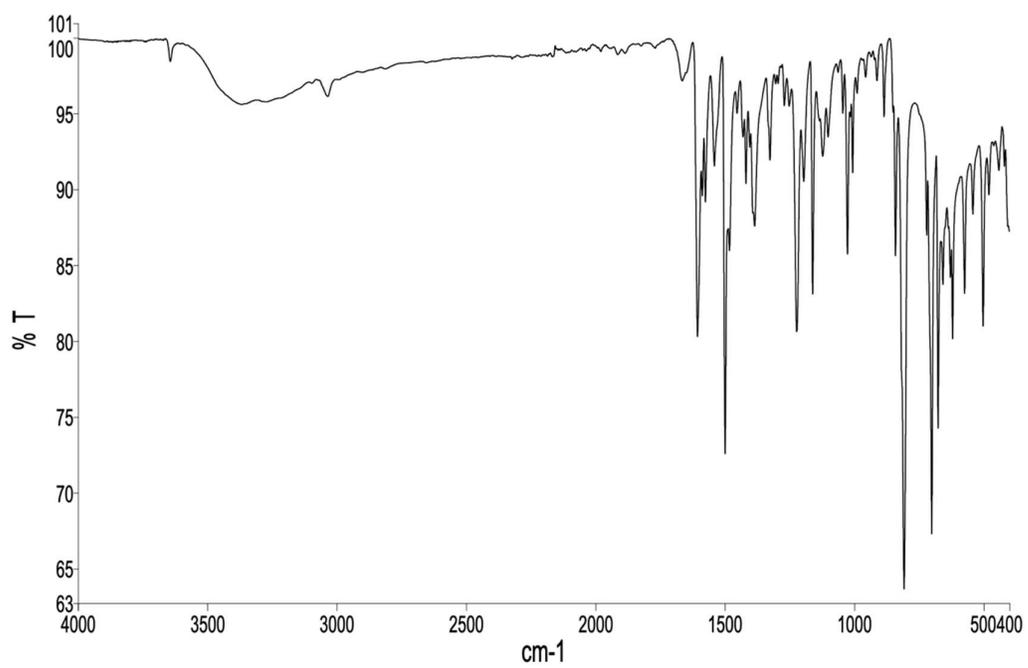


Figure S27. FT-IR spectrum of solid 2.

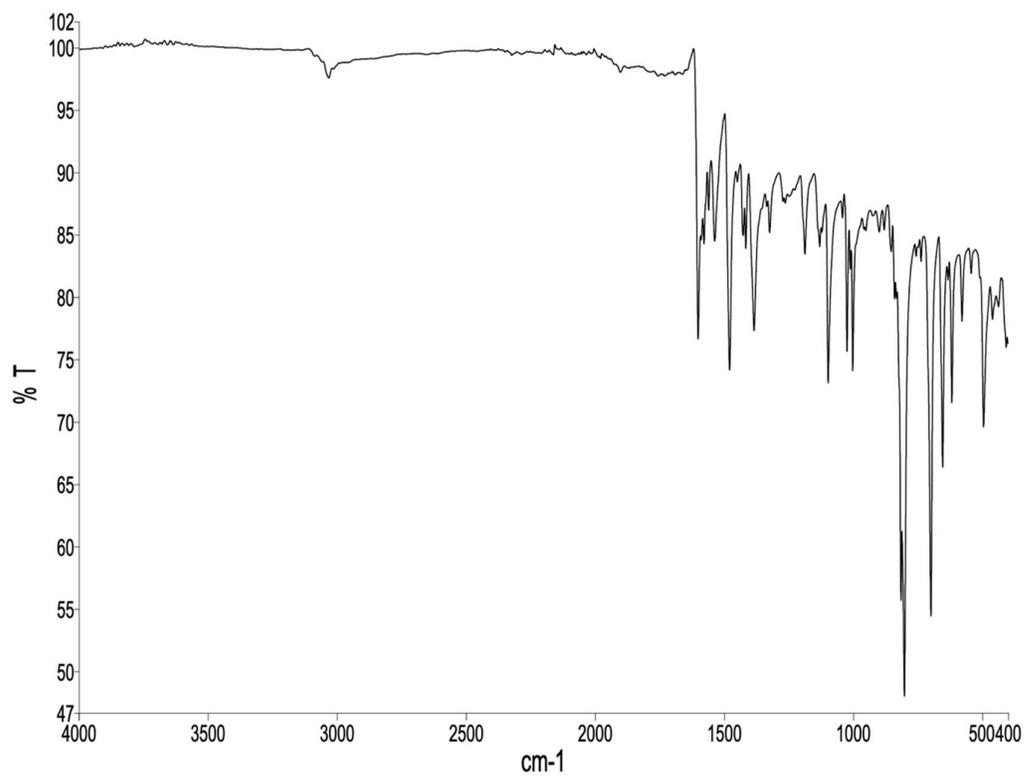


Figure S28. FT-IR spectrum of solid 3.

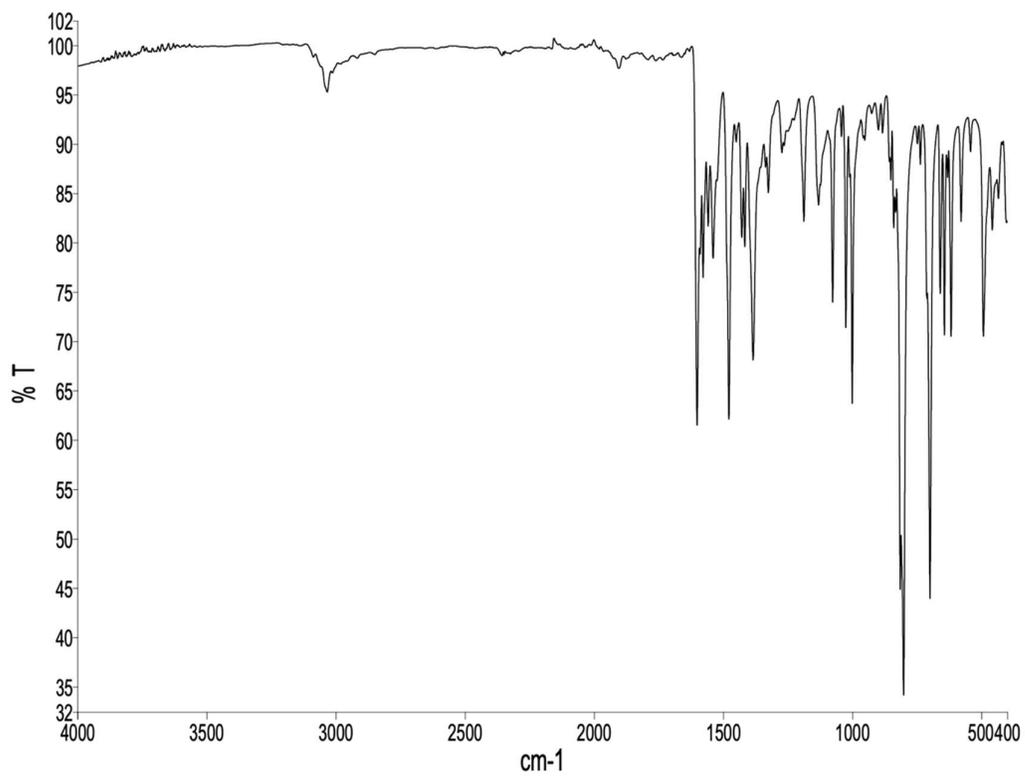


Figure S29. FT-IR spectrum of solid 4.

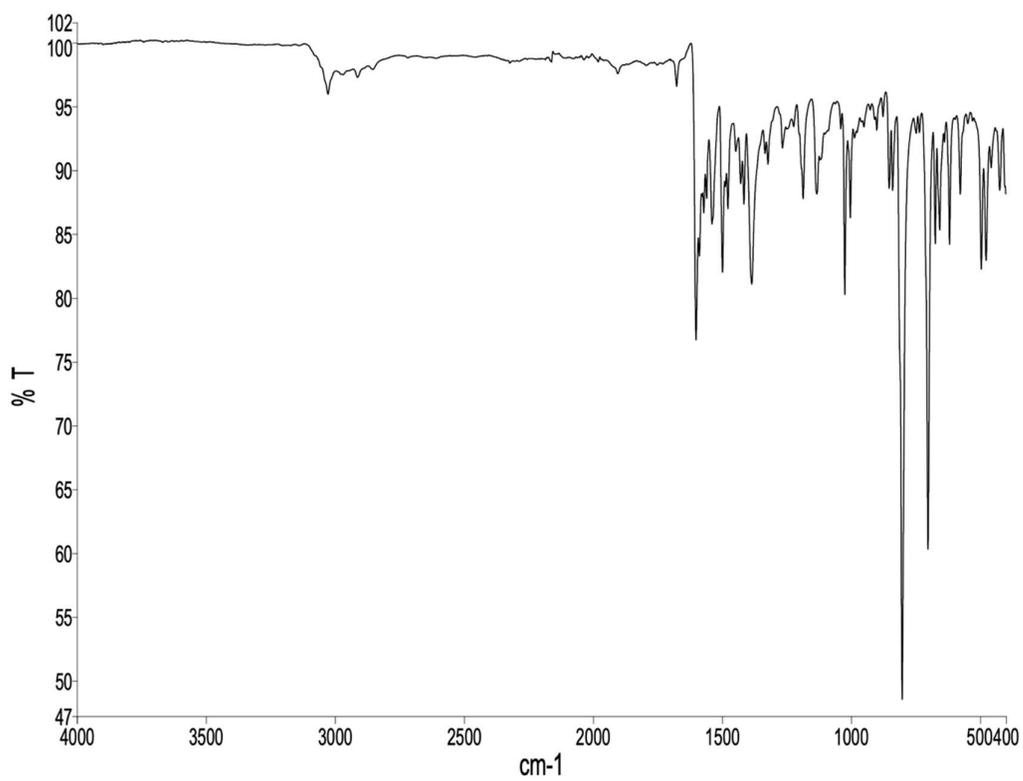


Figure S30. FT-IR spectrum of solid **5**.

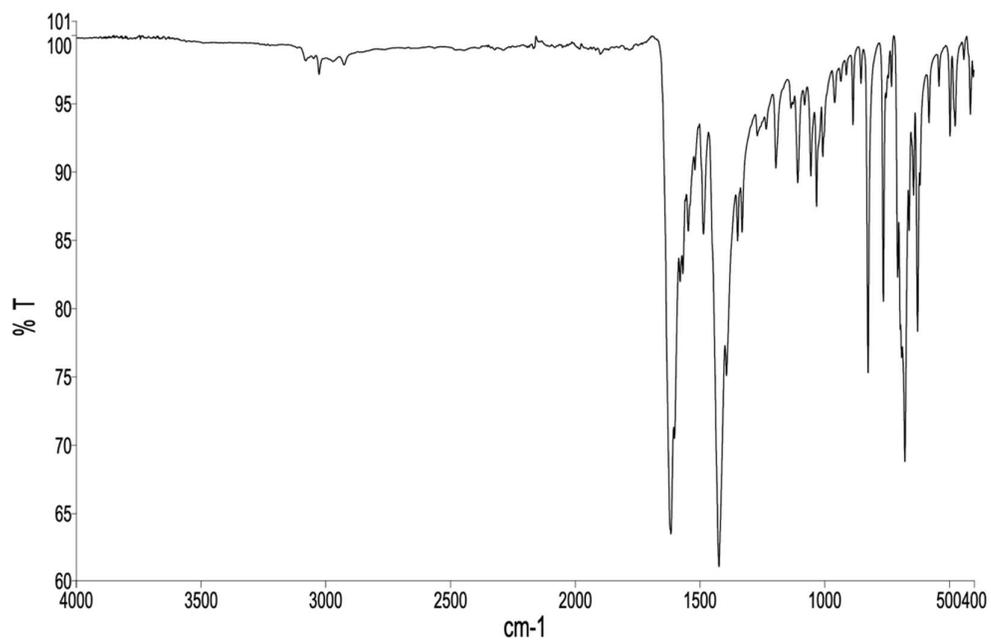


Figure S31. FT-IR spectrum of solid [Cu₂(μ-OAc)₄(**1**)_n].

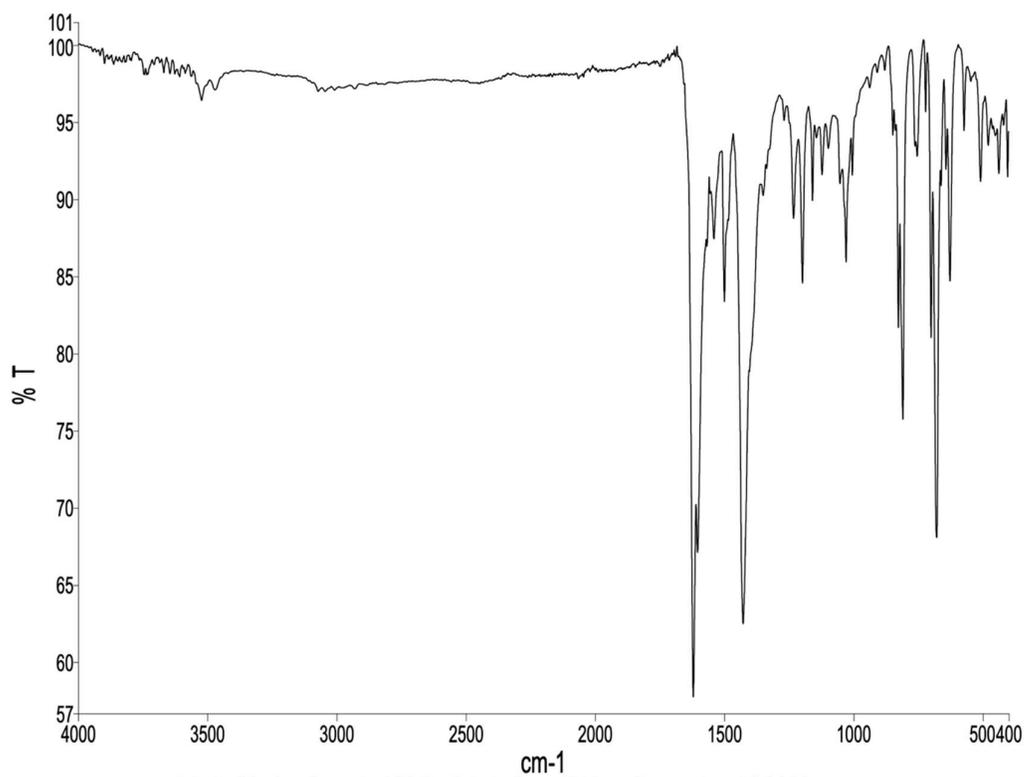


Figure S32. FT-IR spectrum of solid $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{2})]_n$.

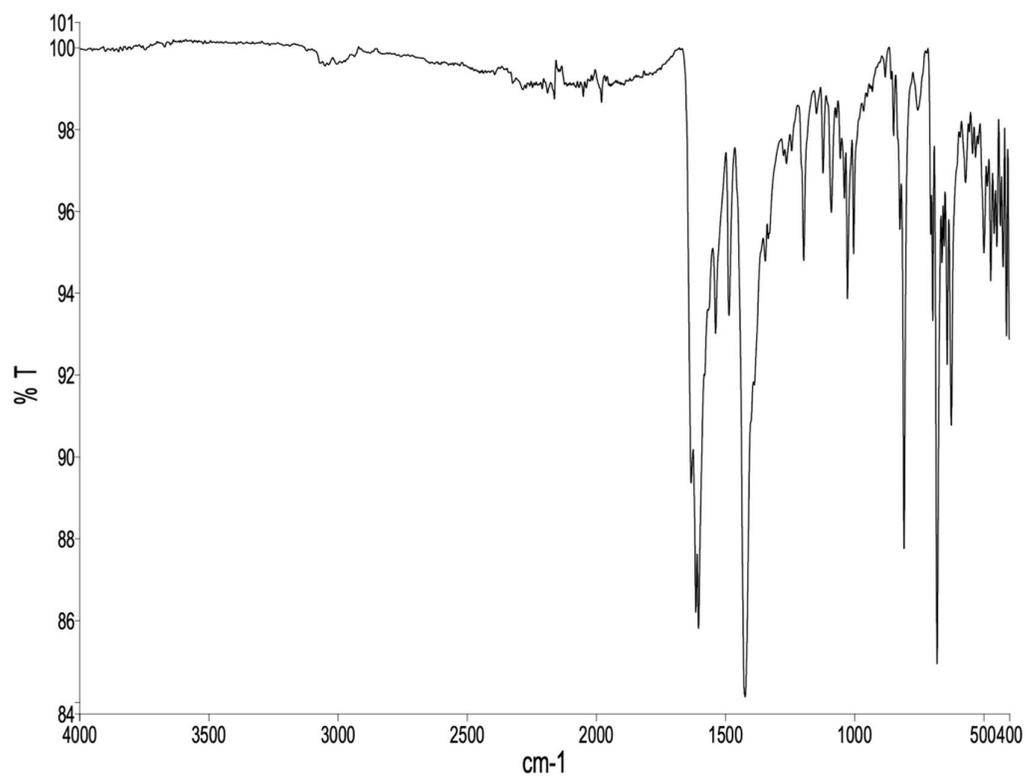


Figure S33. FT-IR spectrum of solid $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{3})]_n$.

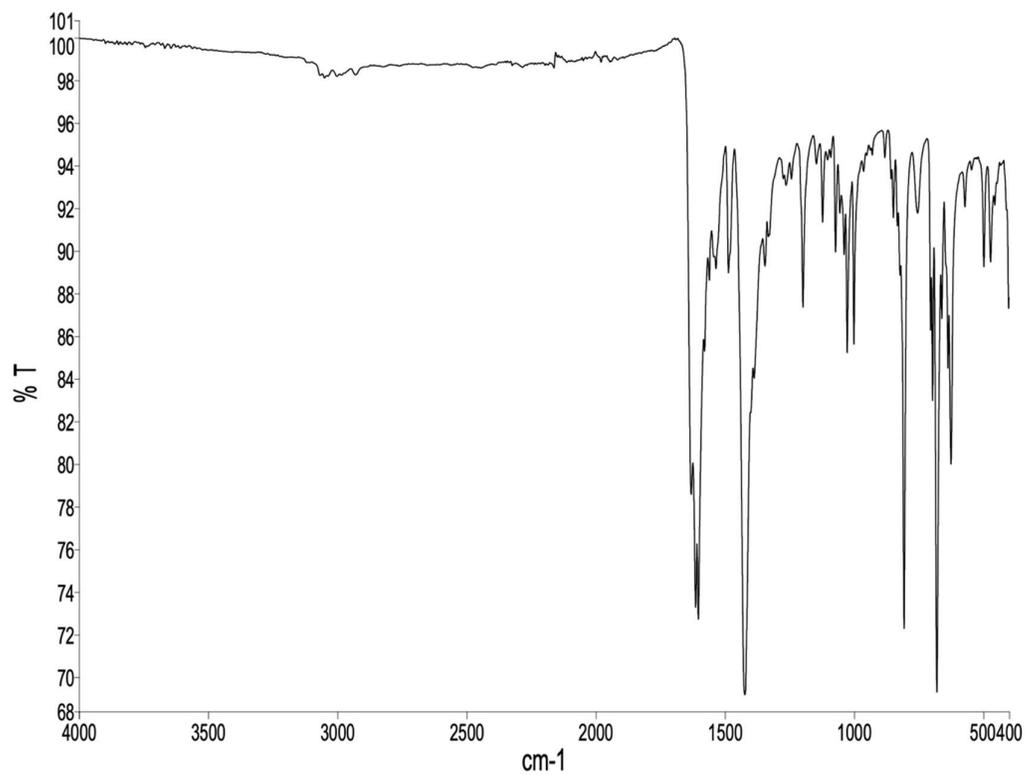


Figure S34. FT-IR spectrum of solid $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{4})]_n$.

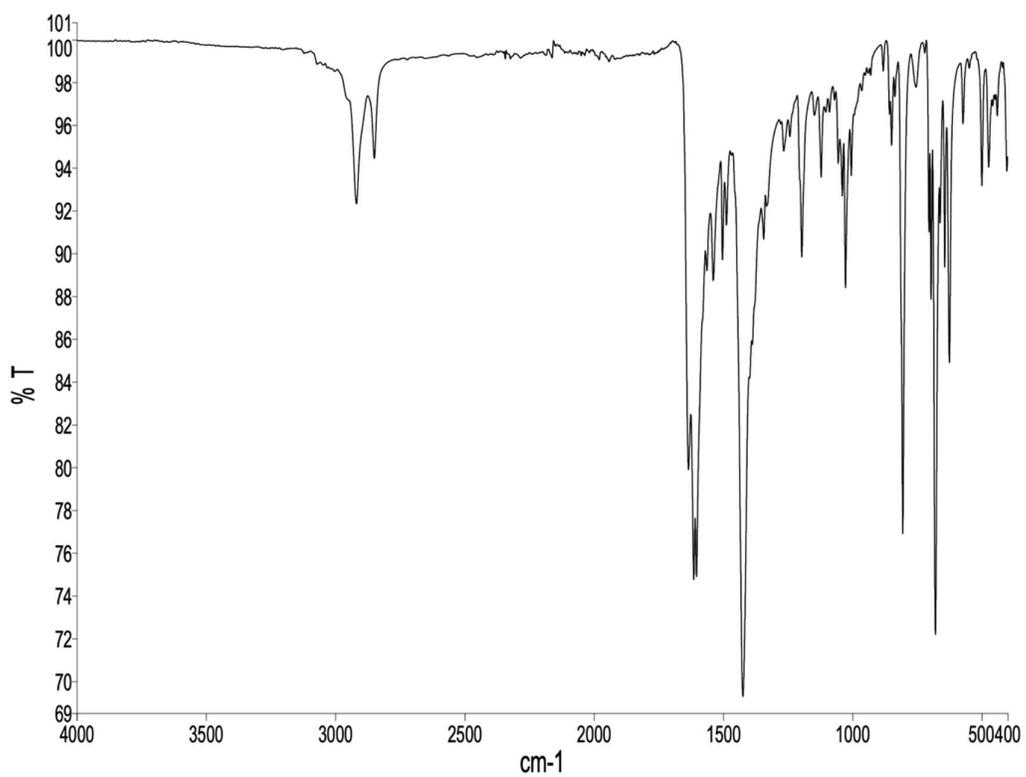


Figure S35. FT-IR spectrum of solid $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{5})]_n \cdot n\text{MeOH}$.

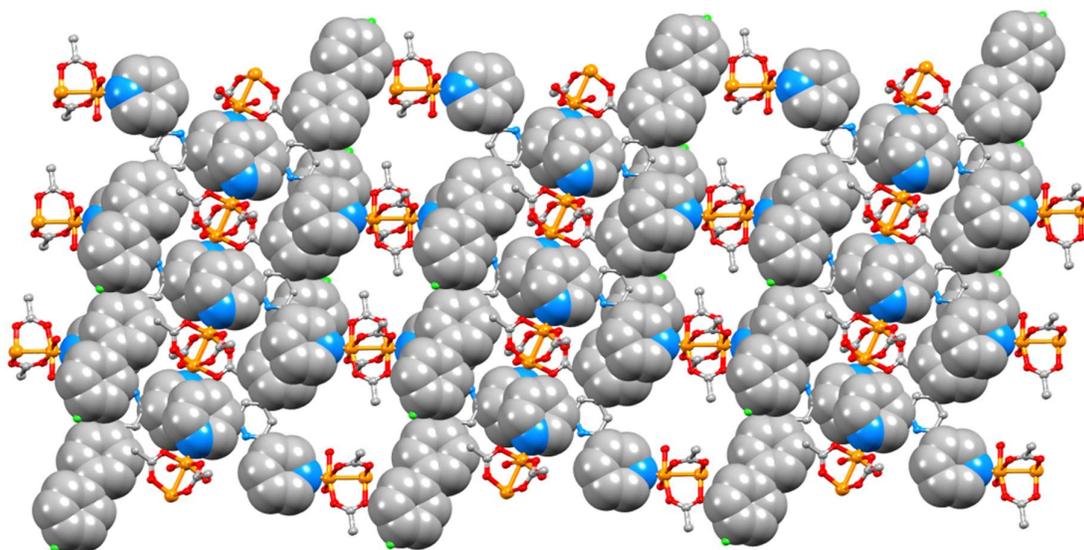


Figure S36. Packing of 1D-coordination polymers in $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{3})]_n$.

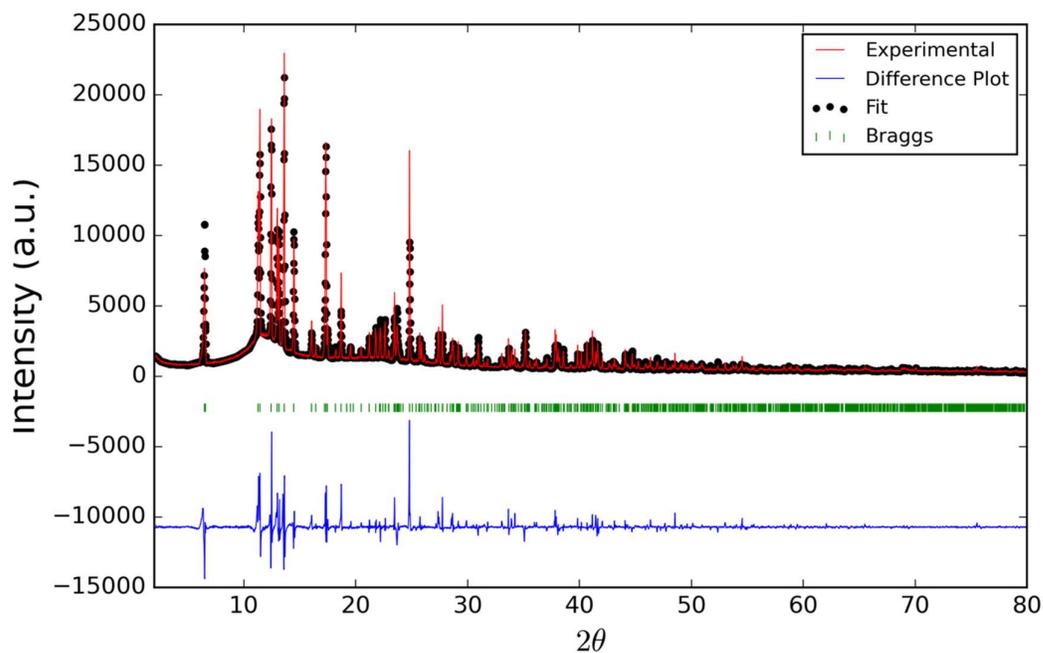


Figure S37. X-Ray diffraction ($\text{CuK}\alpha 1$ radiation) pattern (red lines) of the bulk crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{1})]_n$ with fitting to the predicted pattern from the single crystal determination. The black dots correspond to the best fits from the Rietveld refinements. Green vertical marks show the Bragg peak positions. The blue line in each plot shows the difference between experimental and calculated points.

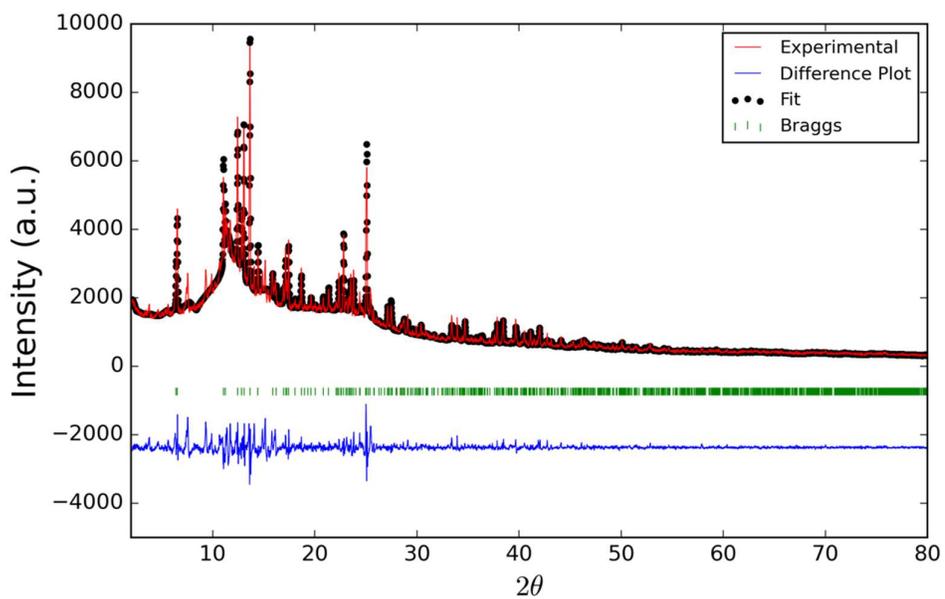


Figure S38. X-Ray diffraction (CuK α 1 radiation) pattern (red lines) of the bulk crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{2})]_n$ with fitting to the predicted pattern from the single crystal determination. The black dots correspond to the best fits from the Rietveld refinements. Green vertical marks show the Bragg peak positions. The blue line in each plot shows the difference between experimental and calculated points.

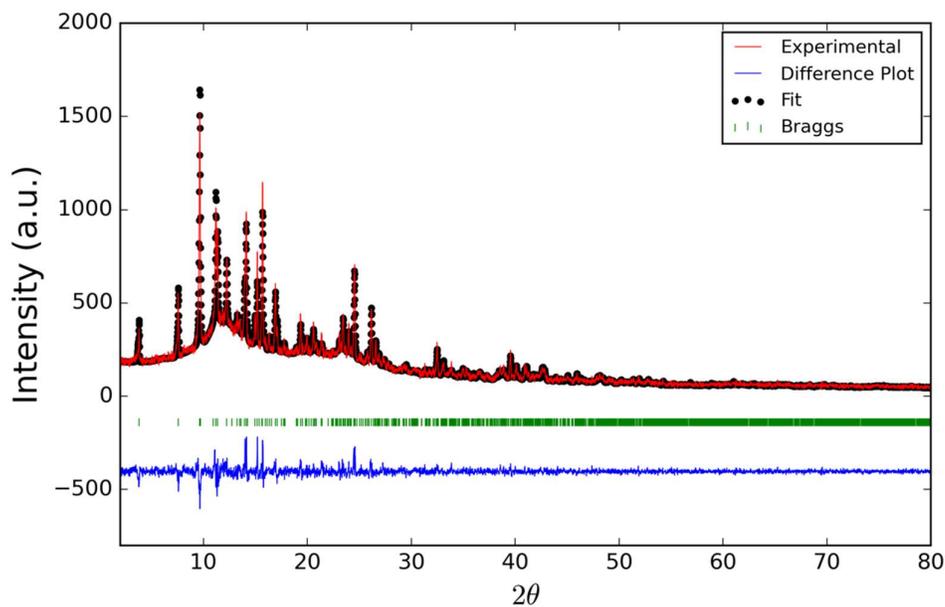


Figure S39. X-Ray diffraction (CuK α 1 radiation) pattern (red lines) of the bulk crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{3})]_n$ with fitting to the predicted pattern from the single crystal determination. The black dots correspond to the best fits from the Rietveld refinements. Green vertical marks show the Bragg peak positions. The blue line in each plot shows the difference between experimental and calculated points.

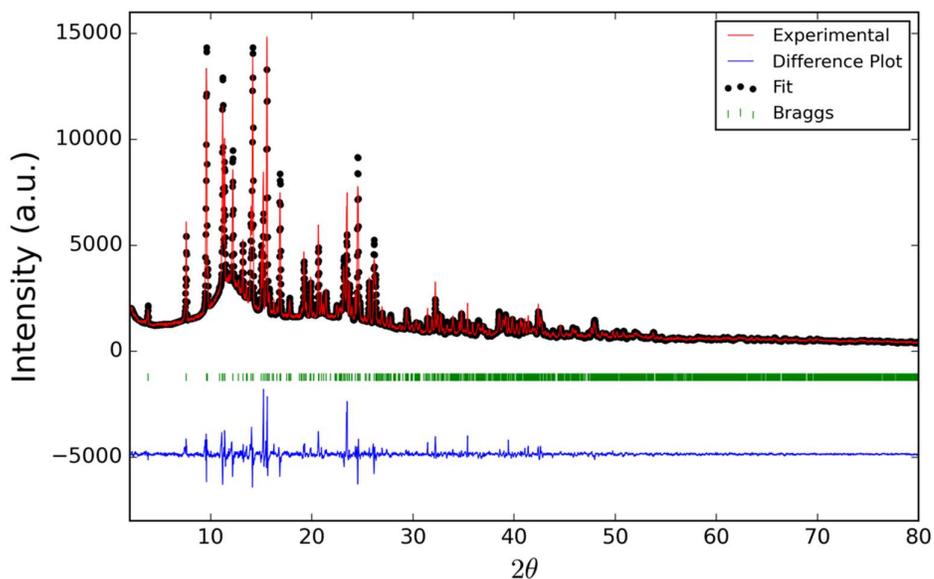


Figure S40. X-Ray diffraction (CuK α 1 radiation) pattern (red lines) of the bulk crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{4})]_n$ with fitting to the predicted pattern from the single crystal determination. The black dots correspond to the best fits from the Rietveld refinements. Green vertical marks show the Bragg peak positions. The blue line in each plot shows the difference between experimental and calculated points.

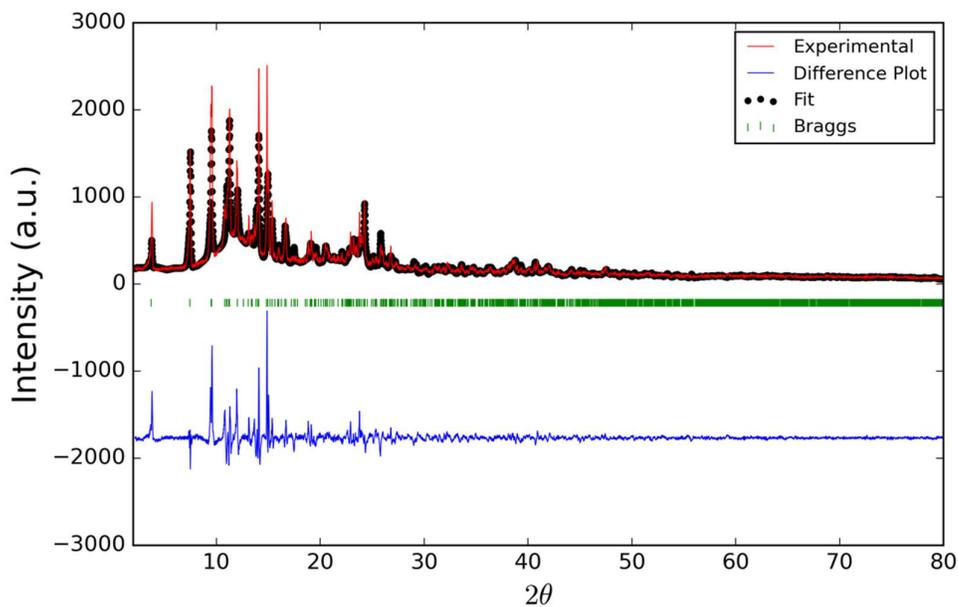


Figure S41. X-Ray diffraction (CuK α 1 radiation) pattern (red lines) of the bulk crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{5})]_n \cdot n\text{MeOH}$ with fitting to the predicted pattern from the single crystal determination. The black dots correspond to the best fits from the Rietveld refinements. Green vertical marks show the Bragg peak positions. The blue line in each plot shows the difference between experimental and calculated points.

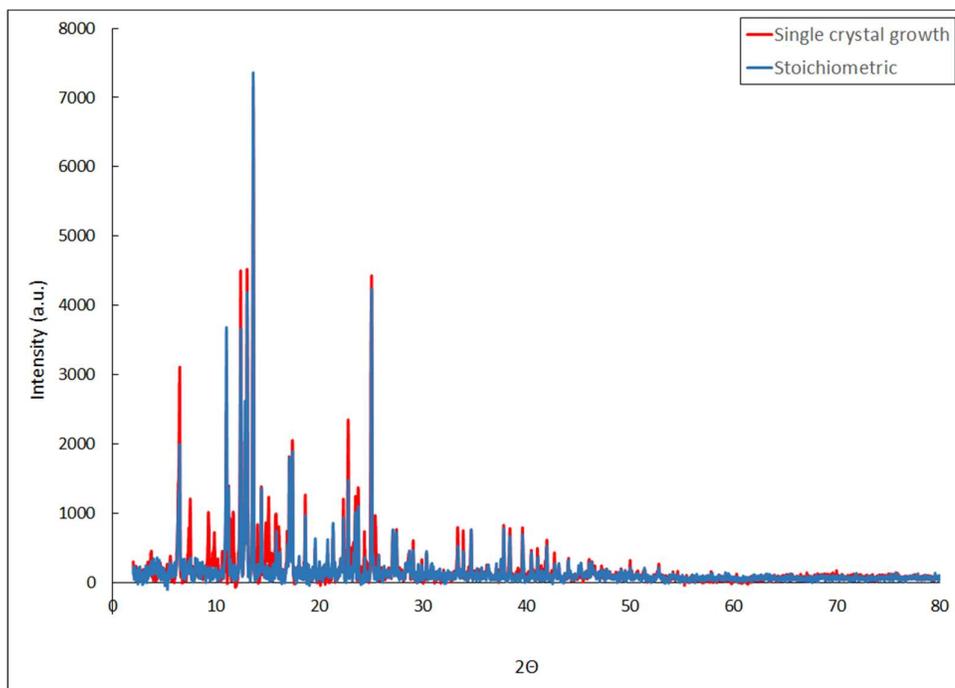


Figure S42. Superimposition of the PXR D patterns of the bulk single-crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{2})_n]$ (red line) and of the material from the preparative scale reaction (blue line). The red lines with no match in the blue pattern arise from an impurity in the bulk material from single-crystal growth (see text).

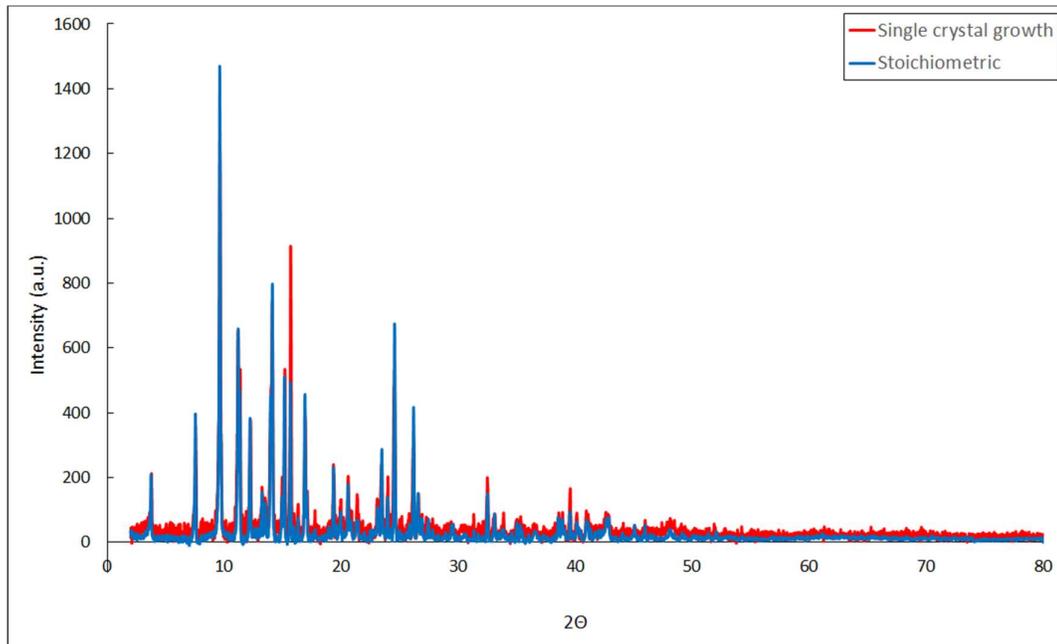


Figure S43. Superimposition of the PXR D patterns of the bulk single-crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(\mathbf{3})_n]$ (red line) and of the material from the preparative scale reaction (blue line).

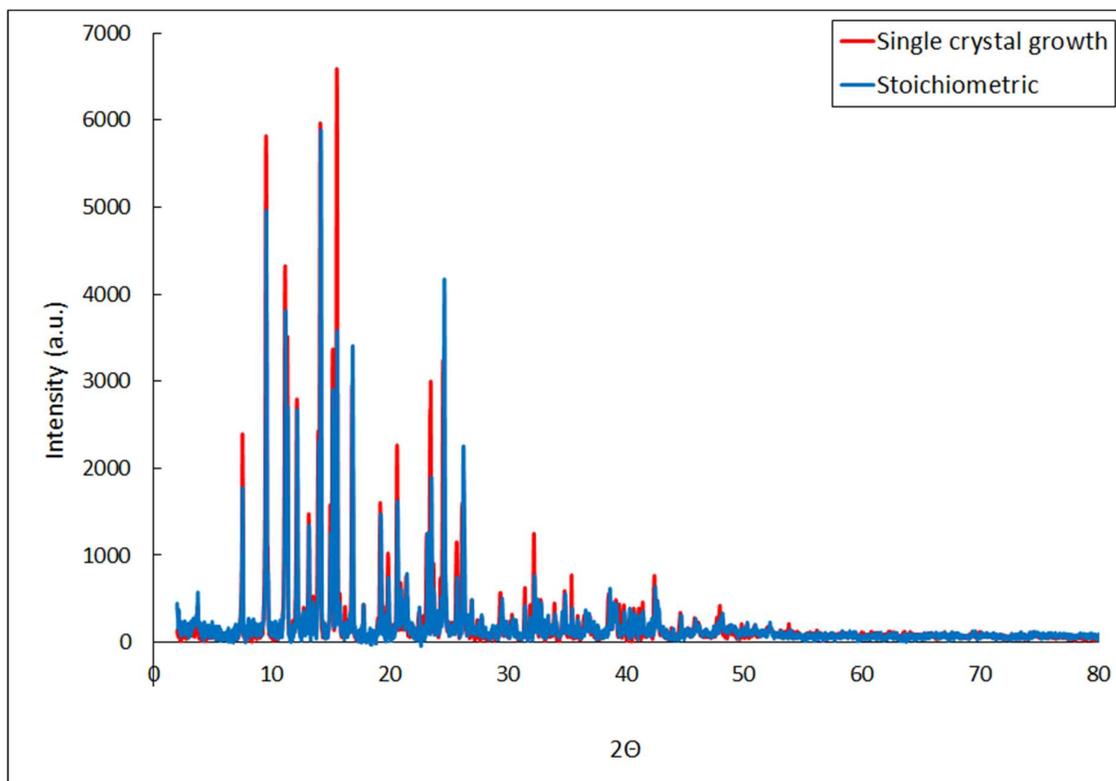


Figure S44. Superimposition of the PXR D patterns of the bulk single-crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(4)]_n$ (red line) and of the material from the preparative scale reaction (blue line).

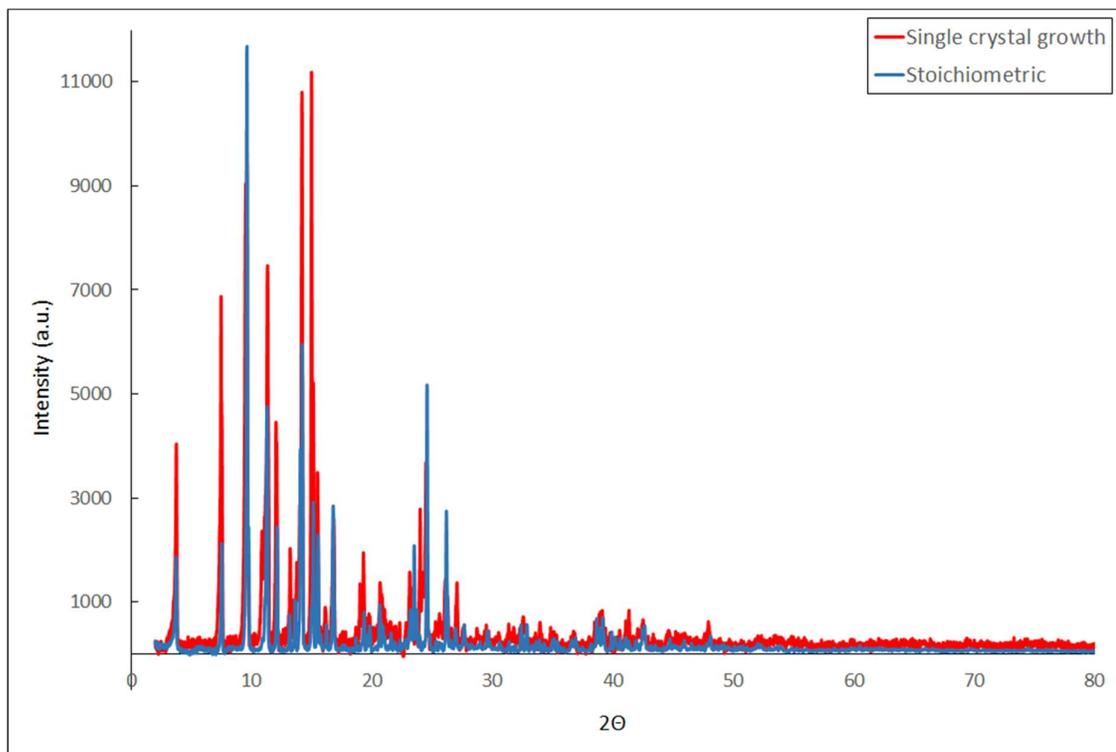


Figure S45. Superimposition of the PXR D patterns of the bulk single-crystalline material of $[\text{Cu}_2(\mu\text{-OAc})_4(5)]_n$ (red line) and of the material from the preparative scale reaction (blue line).