

Supplementary Information

Defects and Calcium Diffusion in Wollastonite

Sumudu Nimasha ¹, Sashikesh Ganeshalingam ¹, Navaratnarajah Kuganathan^{2,3*} and Alexander Chroneos^{2,3}

¹Department of Chemistry, University of Jaffna, Sir. Pon Ramanathan Road, Thirunelvely, Jaffna, 40000, Srilanka

²Department of Materials, Imperial College London, London, SW7 2AZ, United Kingdom

³Faculty of Engineering, Environment and Computing, Coventry University, Priory Street, Coventry CV1 5FB, United Kingdom

*Correspondence: n.kuganathan@imperial.ac.uk; ad0636@coventry.ac.uk (N.K)

Table S1. Interatomic potential parameters used in the atomistic simulations of CaSiO₃

Two-body [$\Phi_{ij}(r_{ij}) = A_{ij} \exp(-r_{ij}/\rho_{ij}) - C_{ij}/r_{ij}^6$]					
Interaction	A/eV	$\rho/\text{\AA}$	C/eV·Å ⁶	Y/e	K/eV·Å ⁻²
Co ²⁺ – O ²⁻	696.3	0.3362	0.00	2.00	10.74
Mn ²⁺ – O ²⁻	715.80	0.3464	0.00	3.42	81.20
Ni ²⁺ – O ²⁻	683.5	0.3332	0.00	2.00	8.77
Mg ²⁺ – O ²⁻	946.627	0.31813	0.00	2.00	99999
Zn ²⁺ – O ²⁻	499.6	0.3595	0.00	2.05	10.28
Sr ²⁺ – O ²⁻	776.84	0.35867	0.00	1.526	11.406
Ba ²⁺ – O ²⁻	931.79	0.3949	0.00	1.46	14.78
Al ³⁺ – O ²⁻	1114.9	0.3118	0.00	3.00	99999
Ga ³⁺ – O ²⁻	2901.12	0.2742	0.00	3.00	99999
Fe ³⁺ – O ²⁻	1156.36	0.3299	0.00	4.97	304.7
In ³⁺ – O ²⁻	1495.65	0.3327	4.33	3.00	99999
Sc ³⁺ – O ²⁻	1575.85	0.3211	0.000	3.00	99999
Y ³⁺ – O ²⁻	1345.10	0.3491	0.000	3.00	99999
Gd ³⁺ – O ²⁻	1885.75	0.3399	20.34	3.00	99999
La ³⁺ – O ²⁻	1545.21	0.3590	0.000	-0.25	145.0
Ge ⁴⁺ – O ²⁻	1497.3996	0.325646	16.00	4.00	99999
Ti ⁴⁺ – O ²⁻	877.200	0.380960	9.00	-35.863	65974.0
Sn ⁴⁺ – O ²⁻	1414.32	0.3479	13.66	4.00	99999
Zr ⁴⁺ – O ²⁻	1502.11	0.3477	5.10	1.65	189.70
Ce ⁴⁺ – O ²⁻	1986.83	0.3511	20.40	7.70	291.75