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Aging Investigation of Polyethylene-Coated Underground Steel Pipelines

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Abstract

The aging of three-layer polyethylene-coated buried steel pipelines for oil/gas and water transport poses significant challenges for public safety, environmental integrity, and economic sustainability. Over time, these pipelines become increasingly susceptible to corrosion and eventual failures, which can pose environmental hazards, safety risks, and costly repairs. Consequently, predicting the service life of polyethylene-coated steel pipelines is critical for mitigating corrosion risks, extending operational lifespan, and planning effective maintenance strategies. Current international standards lack clear methodologies and criteria for assessing the aging behavior of polyethylene-coated underground pipelines. Current studies have examined two techniques—Line Current Attenuation (LCA) and Drainage Test (DT)—to estimate aging rates in polyolefin-coated pipelines following soil exposure during service. The present study introduces an innovative approach for evaluating aging behavior. It includes a comprehensive analysis using an exponential aging model to estimate the coating's average specific electrical resistance at any service time, as well as quantitative criteria for the failure of oil/gas and water pipelines. Moreover, it is based on the modified LCA as the most suitable aging methodology with some limitations. Finally, the study concludes with a derived correlation between the coating's initial specific electrical resistance and its aging rates, and the prediction of the residual life of the polyethylene coating. This integrated framework provides a robust foundation for regulatory bodies, design engineers, maintenance planners, quality assurance/control teams, and researchers to ensure the long-term integrity and sustainability of underground polyethylene-coated steel pipelines.

Keywords: aging model; underground steel pipeline; coating's average specific electrical resistance; soil corrosion; three-layer high-density polyethylene coating (3LPE); line current attenuation test; drainage test (cathodic polarization)

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1. Introduction

The network of oil, gas, water, and hazardous liquid products pipelines plays a significant role in global infrastructure. One key challenge for underground pipelines is maintaining their integrity throughout their operational life while exposed to soil conditions.

Aging of polyethylene-coated underground steel pipelines poses significant risks and challenges to safety, environmental sustainability, and economic worth. As these pipelines age, they become increasingly susceptible to corrosion, leaks, and failures, leading to

environmental disasters and costly repairs. For instance, the USA has a well-developed network of natural gas pipelines totaling over 300,000 miles (483,000 km). Approximately 62% of these pipelines have been in operation for more than 50 years, nearing their maximum service life or having already surpassed their expected lifetime [1,2]. This situation heightens the likelihood of pipeline accidents and corrosion failures, as verified by statistical data [3,4]. Corrosion remains one of the most common causes of pipeline failures. Data from the Pipeline and Hazardous Materials Safety Administration (PHMSA) in the USA indicates that between 1998 and 2017, approximately 18% of incidents involving gas pipelines (gas transmission, gas gathering, and hazardous liquid pipelines) were caused by corrosion (internal and external) [2]. Another similar organization, the European Gas Pipeline Incident Data Group (EGIG), reported that over the last ten years, corrosion has been the predominant factor in reported pipeline incidents [3]. In water infrastructure, corrosive soil conditions are the most common cause of external corrosion, accounting for 59% of water transmission mains and 67% of distribution pipelines [4].

As a result, assessing and predicting the durability of polymer-coated steel pipelines, particularly in soil environments, is a key priority for minimizing corrosion risks and forecasting their periodic repairs.

Considerable resources are being devoted to extending the lifespan of pipelines made from various materials, such as steel or concrete (PCCP—Prestressed Concrete Cylinder Pipe) [5–7]. Significant achievements include the development of diagnostic methodologies, methods for assessing residual strength, repair methods, software, databases, and regulatory documents employed by operational organizations [8–12].

Progress in providing reliable protection for pipelines against soil corrosion has been limited due to several unresolved challenges, including predicting the condition and performance of the polymer coating over time [13,14].

The coating's performance is influenced by multiple factors, including service temperature, soil conditions and properties, cover depth, properties of local and backfilling materials, and other technical characteristics [15–18]. During long-term use of underground pipelines in soil conditions that are designed to last 50 years or more, the coating's dielectric properties tend to degrade over time, leading to a decrease in its protective properties and electrical resistance and, consequently, to increased electrical currents at the impressed cathodic protection stations and increased corrosion risks [19–23].

The efficiency of protective coatings for underground pipelines is evaluated based on their ability to sustain their properties throughout their service life. The predominant properties of the protective coatings include electrical resistance, dielectric continuity, specific defect ratio, and adhesion strength [24]. The evaluation results of the coating's effectiveness determine whether to proceed with its operation or undertake required repairs.

Forecasting the polymer coating's condition and performance over time should consider aging, periodic inspection findings, and technical characteristics along the surveyed pipeline [25]. This is essential for effective planning of rehabilitation, repair, or replacement [26,27]. Consequently, solving aging prediction tasks is unfeasible without modeling the aging processes, considering the aforementioned factors and characteristics.

Based on the review of technical literature, the following three potential approaches can be considered to achieve this objective:

- Excavation and assessment using laboratory testing methods of polymer coating properties, such as adhesion strength of the coating to the steel pipe, mechanical properties, specific electrical resistance, cathodic disbondment, etc. [28–31]. This approach has limited practical applicability because it neglects coating damage (discontinuities) and their impact on the coating's average specific electrical resistance. Consequently, the selected approach should be based on above-ground indirect inspection methods that

evaluate the degree of polymer coating damage, including the number of defects, their specific defect ratio, and their distribution along the pipeline. The galvanic relationship between the examined pipeline section and the overall network should not affect the selected inspection methods.

- Determination of coating aging by current demand based on coating breakdown factors, defined as the current density ratio required to polarize a coated steel surface compared to a bare steel surface, aimed to determine the pipeline's cathodic protection current consumption over various service times. Numerous studies and several leading international standards have been proposed, with defined threshold values [32–35]. Therefore, this direction is of less interest from an innovative research perspective.
- Determination of coating aging by coating average specific electrical resistance, which, according to two international standards [36,37], provides a methodology and criteria concerning the average specific electrical resistance (conductance) of newly buried polymer-coated steel pipelines and for predicting their durability over time. The above international standards specify threshold values (criteria) only for the specific electrical resistance (conductance) of newly polymer-coated pipelines [36,37], with no definite criteria or prediction methodology for assessing aging behavior over time. The latter standard [37] includes a single requirement: the insulation resistance for all types of coatings shall not decrease by more than three times after 10 years and by more than eight times after 20 years of operation.

Technical literature has scarcely explored the method of determining insulation aging using average electrical specific resistance over time. A prominent reference describes four long pipelines with HDPE coatings (factory-applied and field joint coating), where the resistance remained stable in three cases over 2, 7, and 12 years, with only a slight change observed in the fourth (in the range of electrical resistances of $1.3 \times 10^5 \div 3.0 \times 10^5\ \Omega{\cdot}m^2$, as shown in Figure 5.3 [38]).

Some references include prediction models based on insulation resistance of underground pipelines [39–41], which fundamentally contradict the conclusions of reference [38]. The present direction has been used as the objective of resolving the contradictions between various technical sources [38–41].

Three key standards define the technical specifications for factory-applied three-layer extruded HDPE (3LPE) coating for underground pipelines [42–44]. Among the standards, standard [43] is most suitable for underground water mains because it meets the technical requirements of the water industry. According to this standard, 3LPE coating consists of an epoxy primer (FBE) with a minimum thickness of 60 µm, a 140 µm copolymer adhesive layer, and a top layer made of high-density polyethylene (HDPE). The overall coating thickness depends on pipe diameter and type, ranging from 1.8 mm for diameters up to 100 mm to 3.7 mm for those exceeding 800 mm. Standard [45] specifies that pipes with diameters of 80″ (2032 mm) and 100″/108″ (2540/2743 mm) must have coatings no thinner than 4.2 mm and 5 mm, respectively.

The technical requirements for external protective field joint coatings (FJCs), such as cold-applied polymeric tape coatings (Type 12 [46]), two-layer heat-shrinkable sleeves (HSSs) (Type 14A [46]), or two-part liquid epoxy (Type 18A [46]), applied under field conditions at welds, T-joints, elbows, and other irregular shapes [46–48], differ from those specified for factory-applied coatings in the above-mentioned standards [42–45]. Consequently, they constitute pre-existing weak points where aging occurs faster [35,49].

Hence, this study aims to investigate and model the aging behavior of 3LPE-coated underground steel pipelines for oil/gas and water transmission. It focuses on indirect inspection methods for assessing polymer-coated steel pipes for long-term underground service and selecting preferred methods for evaluating coating degradation in field con-

ditions based on average specific electrical resistance. The aging prediction model for 3LPE-coated pipelines is based on field measurements from three to four consecutive years of underground exposure.

2. Degradation Mechanisms of Polyethylene

The polymer type, additive compositions and concentrations, structural morphology, and intended application strongly influence polymer properties [50]. Polyethylene (PE), a hydrocarbon polyolefin, is chemically composed of carbon and hydrogen atoms. Pure HDPE resin (not black compound) features a minimum density of 0.93 g/cm^3 (for class B according to standard [42]). HDPE compound comprises HDPE resin, carbon black, and additives such as stabilizers and long-term antioxidants.

Long-term exposure to detrimental factors, such as oxygen, heat, or UV radiation, can lead to irreversible changes in the polymer's structure and properties, affecting its original performance [51,52]. Under these degradative environments [53,54] polymers can lose their attributes, such as the coating's dielectric characteristics (specific electrical resistance and continuity), mechanical properties (elongation at break, yield and ultimate strengths), interfacial adhesion (peel strength), or other properties [16,21,22].

Polymer aging involves both physical and chemical changes [55].

During physical aging, the material may undergo stress relaxation and changes in crystallinity, while the polymer's chemical structure is preserved [56].

Chemical changes in polymers can involve chain scission, molecular crosslinking, or other reactions that may degrade their properties and ultimately cause material failure [57].

Various forms of degradation may affect HDPE concurrently during service due to thermal, mechanical, biological, environmental, and photodegradation, among others [58]. These processes can produce a synergistic action that accelerates material degradation [59]. Oxidative degradation is the most significant degradation mechanism for polyethylene compounds and may significantly impair their performance [59,60].

A free-radical chain mechanism is generally accepted as the key process responsible for the oxidation of HDPE and similar polymers [61–63]. Oxidation of polyethylene follows a chain reaction mechanism involving initiation, propagation, branching, and termination [64].

Radical initiation stage is as follows:

$$R - H \overset{UV}{\rightarrow} R\cdot + H\cdot \quad (1)$$

The initial step in polymer decomposition involves the creation of free alkyl radicals (R·), typically caused by energy, heat, or other environmental factors.

Oxidation chain propagation is as follows:

$$R\cdot + O_2 \rightarrow ROO\cdot \quad (2)$$

In the propagation phase, as illustrated in reaction (2), oxygen reacts with the newly generated alkyl radical (R·), forming a hydroperoxy radical (ROO·).

Hydroperoxy radicals are highly reactive and readily abstract labile hydrogen atoms from polymer chains, forming hydroperoxides. They may also interact with other polymer radicals, forming unstable intermediates that propagate the reaction.

$$ROO\cdot + RH \rightarrow ROOH + R\cdot \quad (3)$$

As oxidation progresses, the frequency of initiation increases because hydroperoxides (ROOHs) decompose, generating radicals via reactions (4), (5), and (6). As a result of propagation, polyethylene undergoes oxidative reactions that promote chain branching.

Chain branching is as follows:

$$ROOH \rightarrow RO\cdot + \cdot OH \quad (4)$$

$$2ROOH \rightarrow RO\cdot + ROO\cdot + H_2O \quad (5)$$

$$ROOH + RH \rightarrow RO\cdot + RO_2\cdot + H_2O \quad (6)$$

Termination is as follows:

$$2RO_2\cdot \rightarrow RO + ROH + O_2 \quad (7)$$

$$2R\cdot \rightarrow RR \quad (8)$$

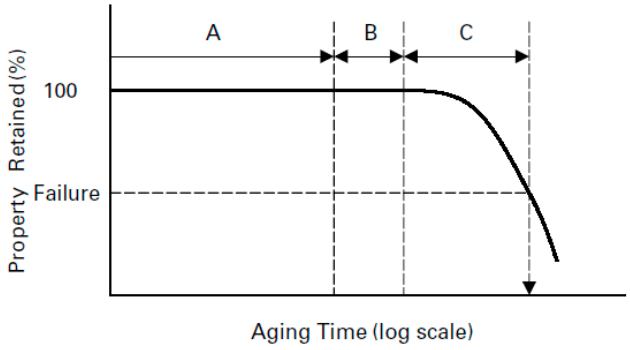
$$R\cdot + ROO\cdot \rightarrow ROOR \quad (9)$$

$$R\cdot + RO\cdot \rightarrow ROR \quad (10)$$

To minimize oxidation in polyolefins, they are commonly stabilized with antioxidants and stabilizers. Antioxidants are generally divided into the following two main categories: primary and secondary [56,61,65]. Primary antioxidants, like hindered phenols (Irganox 1010 or Irganox 1076 manufactured by BASF), act as free-radical scavengers by deactivating or trapping the free radicals. Some compounds deactivate free radicals—ROO·, RO·, and ·OH—by donating electrons, resulting in their conversion to ROOH, ROH, and water, respectively [66–68]. Secondary antioxidants, such as organic phosphites (e.g., Irgafos 168 produced by BASF), function by converting hydroperoxides (ROOH) into non-radical products, such as alcohols (ROH), thus preventing the formation of free radicals [61,69,70].

During the service life of stabilized polyethylene, antioxidants undergo chemical and physical depletion [55,56,71].

Oxidative degradation of polyethylene does not occur until all antioxidants within the coating are substantially depleted [72,73]. The oxidation degradation of polyolefin compounds is represented in three phases in Figure 1 [56,61]:



A—Depletion period of antioxidants;

B—Induction time to start polymer degradation;

C—Time to reach the failure level (50% or as defined) of deterioration of a specific property.

Figure 1. Three fundamental stages (A–C) describing the chemical degradation process of HDPE geomembranes [56,61].

Stage A represents the period during which antioxidants are consumed through oxygen reactions.

Stage B corresponds to the induction period required before the polymer begins to degrade through oxidation.

Stage C is characterized by the degradation of the polymer and a subsequent loss in its mechanical performance or other specific properties. Failure is typically defined as a 50% reduction in a key property (half-time degradation).

3. Experimental Procedure

3.1. General

This study is based on selected underground water and gas/oil steel pipelines. The selection was based on the following considerations:

1. Polyethylene-coated pipeline sections with high initial average specific electrical resistance exceeding 10^6 Ohm·m^2.
2. Pipeline segments preferably electrically separated from the pipeline network by insulating joints or pipeline ends with no continuation (or connection to other pipelines). In both cases, the electrical current is zero. Such a method makes it possible to isolate inspection zones from interference of cathodic protection currents present in the pipeline network, incorporating autonomous inspection capabilities, comparison of different inspection methods, and precise validation.
3. Selected pipeline sections with diverse technical characteristics (age, length, diameter, type and soil resistance, vicinity with high-voltage AC power lines (161/400 kV), etc.)
4. Selected pipeline sections monitored over three or more consecutive years (the research duration). This has allowed us to determine the aging rates of the polymer coating, compare the methods, and conclude on a suitable and reliable inspection method.
5. The oldest oil/gas and water pipelines with Drainage Test results of average specific electrical coating resistances that were tested in this study were 11 years old (from 2014).

3.2. Assessment Methods of Underground Polymer-Coated Steel Pipelines

An extensive literature review was undertaken to survey the assessment methods of aged underground polymer-coated steel pipelines [25,32–38,42,43,74–78]. The Drainage Test [38] and the Line Current Attenuation Test [36,79] were selected as the best methods for assessing the aging of buried polyethylene-coated steel pipes.

The Drainage Test is carried out by applying a current to the pipeline. The On- and Off-potentials and the electrical current are measured at consistent time intervals. The test is complete when the Off-potential is stable (no further negative change) or after a set duration (e.g., an hour). A Drainage Test can be conducted for different purposes, such as assessing the average electrical specific coating resistance or the consumed electrical current to protect an underground object from corrosion. Increased current consumption or decreased coating average specific electrical resistance implies more coating defects on the pipeline and/or larger defect areas, indicating lower coating quality. The results are validated if the Off-potential meets or exceeds the valid protective potential. This method is ineffective for identifying coating defects. To obtain the average specific electrical resistance of the entire surveyed polymer-coated pipeline section, the difference between the On- and Off-potentials is divided by the consumed electrical current to obtain the resistance of a pipeline section with an area of 1 m^2 and then multiplied by the entire surface area of the investigated coated pipeline section.

Figure 2 illustrates the DT's primary setup. The surveyed pipeline section must be electrically separated from other pipe sections using isolation joints [80] and connected to a temporary or permanent cathodic station. The current enters the pipeline at the coating defect sites through the soil (which is considered the electrolyte). For 3LPE coating, the

electrical current (I, [A]), On- (φ_{on} [mV]), and Off- (φ_{off} [mV]) potentials are recorded at regular time intervals, usually at 0, 3, 6, 9, 12, 15, 30, and 60 min.

For the LCA method, it is essential to utilize Line Current (LC) testing points, that are generally preinstalled and distributed along the length of gas/oil pipelines, as shown in Figure 3. They are usually installed before and after oil/gas stations, at isolation joints, at pipeline ends with no continuation, etc. In most water pipelines, the LC test points are not installed. Therefore, for this study, additional LC test points were designed and installed for water pipeline sections, following the same principles as for oil/gas pipelines.

In the LC test points, four copper wires are usually placed in the following configuration: the left pair of wires (LC test point A) is welded to the pipeline at a close distance from each other (typical distance is 15–30 cm). A trench is excavated 50–100 m away from the first pair of wires, and a second pair of copper wires is installed and connected to the pipeline, as shown in Figure 4. The outward pair is designed to measure electric current, and the inward pair is intended to measure electric voltage, including calibration of the pipeline's span against the defined length (50–100 m).

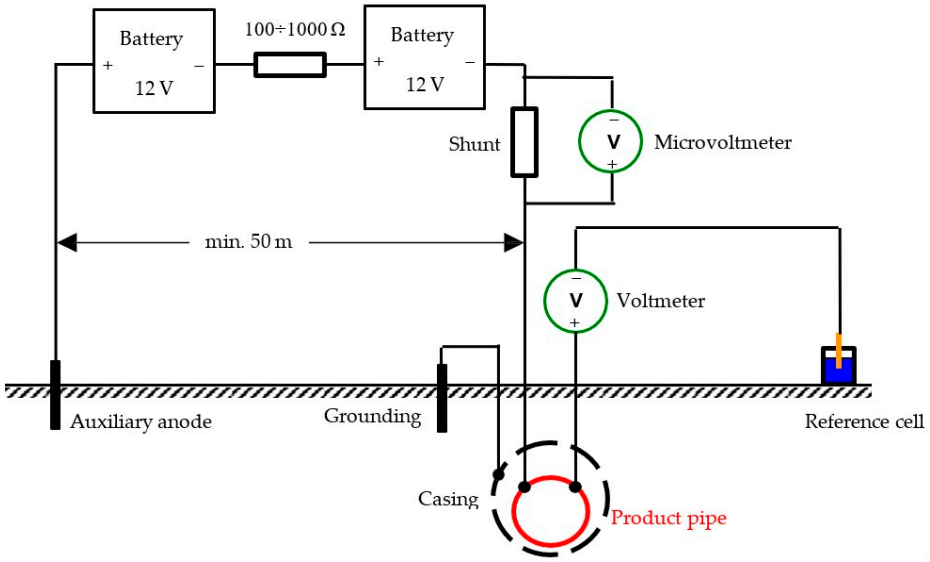
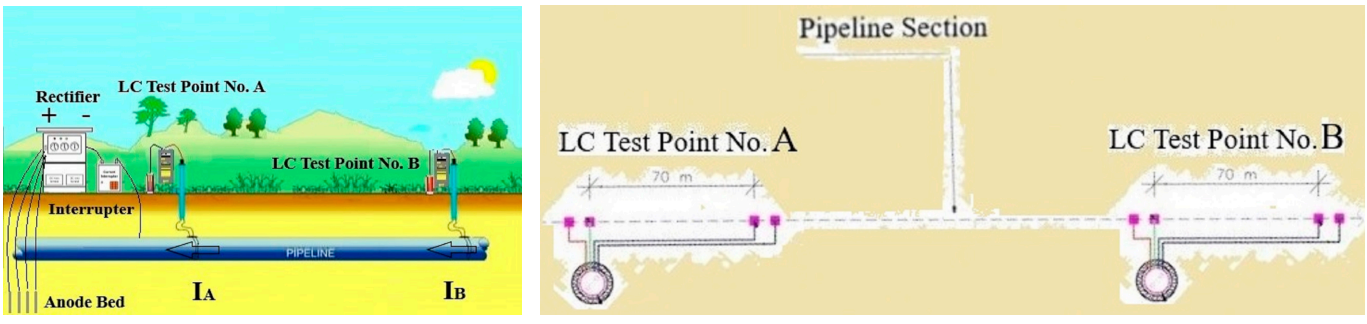


Figure 2. Schematics of the Drainage Test setup.



(**A**) (**B**)

Figure 3. Scheme of surveyed pipeline section by LCA method with two LC test points with rectifier and interrupter. (**A**) A general view of the LC test points. (**B**) The typical connection of the LC test points to the surveyed pipeline.

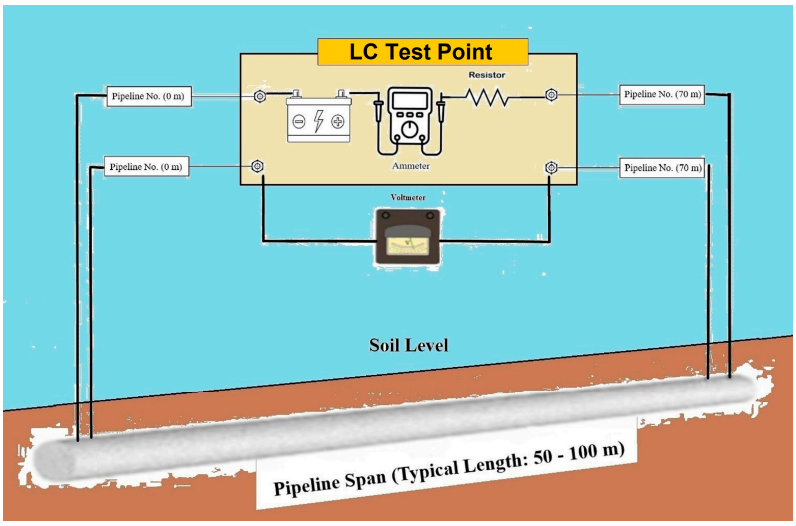


Figure 4. LC test point configuration for calibration of pipeline span (four wires).

The procedure for determining the coating electrical resistance from Line Current Attenuation measurements is detailed in the standard [36] and in the reference [78]. Following the review of the LCA procedure principles, which were derived from the two technical sources above, the modified procedure comprises a few stages:

a. First stage—Calibration of the line current (four-wire) test points for determining the electrical resistance of the tested pipeline sections with a defined LC length. The general arrangement for pipeline current measurement calibration is shown in Figure 4. An additional option for calculating the electrical resistance of the tested section is provided by the formula in Appendix B of the standard [36] (Standard Pipe Data Tables), but it is a less precise method compared with calibration. The main steps of the section's calibration are as follows:

   1. Measuring and recording the initial voltage (U_{0cal}, mV) between inward terminals, as shown in Figure 4, and noting the voltage polarity.
   2. Applying a test current I_{cal} (mA) between outward test leads.
   3. Measuring and recording the voltage (mV) change between inward terminals while interruption is applied with the chosen regime, like On/Off = 8:2 s, and noting the voltage polarity.

$$\Delta V_{CAL} = Vcal_{ON} - Vcal_{Off} \quad (11)$$

   4. Measuring and recording the difference in current (mA) between the outward terminals.

$$\Delta I_{CAL} = Ical_{ON} - Ical_{Off} \quad (12)$$

   Calculating resistance in span (μΩ) as follows:

$$R_{CAL} = \frac{\Delta V_{CAL}}{\Delta I_{CAL}} \quad (13)$$

   5. Steps 3, 4, and 5 are repeated with different electrical currents to verify results, obtain additional statistical data, and ensure repeatability.

b. Second stage—The surveyed pipe section is connected to either a temporary or permanent cathodic station with a connected current interrupter at a specific time regime,

like On/Off = 8:2 s. Measuring, recording, and calculating the potential change (ΔU) at each LC test location (μV or mV) between ON- and OFF- potentials are as follows:

$$\Delta U_{CP} = U_{CP_{ON}} - U_{CP_{Off}} \quad (14)$$

c. Third stage—Calculating the pipe current at each LC test location from the first and second stages.

$$I_{CP} = \frac{\Delta U_{CP}}{R_{CAL}} \quad (15)$$

d. Fourth stage—The surveyed pipe section is still connected to a temporary or permanent cathodic station with a current interrupter, like On/ Off = 8:2 s. The measurement of the "ON" (φon [mV]) and "Off" (φoff [mV]) structure-to-electrolyte potentials at each LC test location should be conducted, as in the standard [81]. Calculating the difference between ON- and OFF-potentials.

$$\Delta\varphi_{PIPE} = \varphi_{ON} - \varphi_{Off} \quad (16)$$

e. Fifth stage—Measuring the soil resistivity near each LC test location according to the Wenner four-pin or soil-box method [82]. Calculating the average soil resistivity of the surveyed pipeline section.

f. Sixth stage—Calculating the surface area (A) of the surveyed pipe section between LC test locations (m^2):

$$A = \pi \cdot D \cdot L \quad (17)$$

where D—pipe outside diameter and L—length of the pipe section.

g. Seventh stage—Calculate the average change in pipe-to-electrolyte potential (Δφ avg) for each pipeline section (between LC test points A and B).

$$\Delta\varphi_{Average} = \frac{\Delta\varphi_A + \Delta\varphi_B}{2} \quad (18)$$

where $\Delta\varphi_A = \Delta\varphi_{ON\text{-}A} - \Delta\varphi_{Off\text{-}A}$, the change in structure-to-electrolyte potential at point A, mV, and $\Delta\varphi_B = \Delta\varphi_{ON\text{-}B} - \Delta\varphi_{Off\text{-}B}$, the change in structure-to-electrolyte potential at point B, mV.

h. Eighth stage—Calculating the current pick-up (ΔI) for each pipeline section (between LC test points A and B):

$$\Delta I = \Delta I_A - \Delta I_B \quad (19)$$

i. Ninth stage — Finally, calculating the coating average specific electrical resistance (R_{coat}) for the pipeline section (between LC test points A and B) in $\Omega \cdot m^2$.

$$R_{coat} = \frac{\Delta\varphi_{Average}}{\Delta I} \cdot A \quad (20)$$

j. The obtained results have been normalized for a specific soil resistivity of 10 Ω·m, according to the requirements of the standard [36].

$$R_{Final} = \frac{10\Omega \cdot m}{\rho_{Average}} \cdot R_{coat} \quad (21)$$

To determine the aging rate of the polymer-coated underground steel pipelines, several sections of oil/gas (G1–G4) and water pipelines (N1–N4) were selected, as shown in Table 1.

Table 1. The selected gas/oil pipeline and water pipeline sections for 3LPE coating aging assessment by LCA and DT methods.

Pipeline Designation	Diameter (Inch)	Length (m)	Wall Thickness (mm)	Pipeline's Age (Year)	The Initial Average Specific Electrical Resistance of the Pipeline Section, ($\Omega \cdot m^2$) (*)
N1 (N1) South	16	4920	4.0	11	1.9×10^6
N2 (N2) South	20	4990	4.0	11	1.0×10^6
N3 (N3) South	24	4940	4.8	11	1.9×10^6
N4 (N11) Center	100	1200	15.9	11	4.3×10^6
G1 (G51) Center	18	9420	12.35	11	17×10^6
G2 (G52) Center	18	7760	12.35	11	11×10^6
G3 (G53) North	10	14,730	10.30	11	21×10^6
G4 (G54) North	18	12,540	12.35	11	19×10^6

(*)—The Drainage Test measured the initial average specific electrical resistance of the polymer-coated pipeline sections.

Additional remarks about oil/gas and water pipelines are as follows:

- The standard steel grade for pipes is X42, with a typical length of 12.2 m and varying wall thicknesses according to the standard [83].
- Standard factory-applied external coating is as follows: three-layer extruded HDPE with varying coating thicknesses according to the standard [43].
- Field joint coatings at the weld joints: for oil/gas pipelines—two-layer polymer tapes (applied cold polymeric tape field joint coatings—Class 12 according to the standard [46]); for water pipelines—two-layer heat-shrinkable sleeves (HSS—Class 14A according to the standard [46]).

The tests for oil/gas pipeline sections were conducted on the straight sections between oil/gas stations, which did not include all the components within the stations, such as epoxy-coated T-connections (T-joints) and elbows. This diminishes coating aging due to potential damage during field application. The absence of T-connections and elbows reduces the risk of premature aging. Pipeline sections with isolation joints at their edges were selected to enable electrical separation between the inspected pipeline sections and the entire pipeline network. This type of arrangement facilitated the execution of two concurrent inspections: Cathodic Polarization (Drainage Test) and upgraded Line Current tests.

The characteristics of water pipelines differ significantly from those of gas and oil pipelines. This includes numerous T-connections (T-joints) and elbows with diverse coating types applied under field conditions, primarily two-layer epoxy coatings from various manufacturers. Coating breakdown factors of two-part epoxy, particularly those applied in the field conditions on T-joints and elbows, are much higher than those of 3LPE factory-applied coating [33,35,49]. This directly contributes to the potential for faster degradation of the pipeline's overall coating. Locating pipeline sections with isolation joints in water pipelines is challenging, unlike in gas and oil pipelines. As a result, performing a Drainage Test (Cathodic Polarization) to determine the insulation's condition is not feasible, and only the modified LCA test might be appropriate for that aim.

The following main instruments and auxiliary accessories have been used for the measurements:

Universal measuring instrument with datalogger MiniLog2, Weilekes Elektronik GmbH, Gelsenkirchen, Germany.

Multimeters: Fluke 187/189/287, Fluke Europe B.V., Eindhoven, the Netherlands and Model LC-4.5 Voltmeter, M.C. Miller, Sebastian, FL, USA.

Reference electrode, type RE-5C, $Cu/CuSO_4$, M.C. Miller, Sebastian, FL, USA.

Synchronization MicroMax GPS360 Current Interrupter with GPS, American Innovations, Austin, TX, USA.

Metallic cables/leads from various manufacturers and with different section areas.

4. Aging Modeling

The new HDPE material is a hydrophobic and nonpolar polymer. During aging, oxidizing agents such as oxygen produce various hydrophilic chemical groups. These groups comprise alcohols (C-OH), aldehydes (R-CHO), ketones (C-O), ethers (R-O-R'), carboxyl (C-OOH), carbonyl (C=O), etc. [84,85]. The hydrophilic groups absorb water, reducing the electrical resistance of the nonpolar HDPE polymer matrix [79,86].

We assume that the generation of these functional groups follows first-order reaction kinetics. If the system's electrical resistance (or conductivity) is proportional to the concentration of the hydrophilic functional groups, it follows that first-order kinetic laws govern both conductivity and resistance.

For sufficiently small values of the product of the rate constant and time ($-\alpha \cdot t$), the exponential term can be approximated by its Maclaurin series (22), retaining only the first term.

$$e^x = 1 + \frac{x}{1!} + \frac{x^2}{2!} + \frac{x^3}{3!} + \cdots \tag{22}$$

Hence, the aging follows an exponential relationship (first-order kinetics), where the reaction rate diminishes at a linear rate proportional to the coating's average specific electrical resistance R over time t (23). The standard first-order differential equation describes this relationship.

$$Reaction\ Rate = -\frac{dR}{dt} = \alpha[\mathrm{R}] \tag{23}$$

For the first-order reactions, the units of α are $year^{-1}$.

From Equation (23)

$$\frac{dR}{R} = -\alpha \mathrm{dt} \tag{24}$$

Integration of Equation (24) leads to Equation (25):

From

$$\int \frac{1}{x} = \ln(x) \quad \Rightarrow \quad \ln(\mathrm{R}) - \ln(\mathrm{R_0}) = -\alpha \mathrm{t} \quad \Rightarrow \quad \ln(\mathrm{R}) = \ln(\mathrm{R_0}) - \alpha \mathrm{t} \tag{25}$$

Or

$$Rc(t) = Rc(0)e^{-\alpha t} \tag{26}$$

A plot of $\ln[\mathrm{R}/\mathrm{R_0}]$ vs. t will be linear with a slope of $-\alpha$.

An alternative method for determining the reaction order is to analyze changes in the half-life (or any other defined service time) as the reaction progresses. For the first-order reaction, the half-life service time is as follows,

$$\mathrm{t_{1/2}} = \ln 2/\alpha \tag{27}$$

Based on Equation (27), there is no dependence of the half-life service time on the electrical resistance, so $t_{1/2}$ is a constant parameter for a first-order reaction.

5. Results and Discussion

Polyethylene-coated steel pipelines age with time [87–90]. The polymer coating degrades due to numerous factors [16–18,28,91], such as soil composition, static and dynamic stresses, groundwater, microorganisms, and temperature [20–23]. Aging leads to the formation of spot defects and/or group defects. This may result in loss of the coating's protective properties, manifested in a decrease in the overall polymer coating resistance [92].

A decline in the coating's electrical resistance during the pipeline's operational period means that the current and number of cathodic stations must be increased, or the insulation in that section must be repaired [26].

Selected field indirect inspection methods are designed to evaluate the average specific electrical resistance of long polymer-coated buried pipelines, due to the presence of defects at the external 3LPE surface. As the defective exposed area increases with service time, the coating's electrical resistance decreases, resulting in diminished protective properties.

The minimum criterion for initial coating average specific electrical resistance was set at $3 \times 10^6\ \Omega{\cdot}m^2$, obtained and determined after analysis of DT results after long-term internal study.

5.1. Oil/Gas Pipelines

To assess the coating average specific electrical resistance of underground oil/gas pipelines over different service periods, two types of inspections have been performed: LCA and DT. Four oil/gas pipelines from our initial study were selected for this purpose (G1–G4). Each section pair, G1–G2 and G3–G4, was arranged sequentially and connected via an oil/gas station.

The oldest of these polymer-coated pipelines was 11 years old. To establish the aging model based on the average specific electrical resistance, evaluations were conducted during 3–4 annual measurements. LCA inspections were performed after 8–9, 10, and 11 years in the underground service. DT measurements were conducted only after 10 years. The results were compared with the initial average specific electrical resistance from long-term internal study, and the appropriate model was established based on the initial and aged specific electrical resistance results.

5.2. Water Pipelines

The LCA method was only used to determine the average specific electrical resistance of coatings on underground water pipelines, as no pipeline sections were electrically separated from the water pipeline network by isolation joints, and thus were not suitable for the execution of the DT method. For this purpose, four water pipelines with various technical characteristics were selected. Like the underground oil/gas pipelines, the maximum age of these pipelines was 11 years. The aging model was based on four time periods of inspections conducted—after construction, and after 9, 10, and 11 years of underground exposure.

Table 2 summarizes the average specific electrical resistance results over various service periods of 3LPE-coated oil/gas and water pipelines over time, conducted by the LCA and DT methods.

The analysis and comparison of the data from the LCA and DT methods for the G1–G2 and G3–G4 oil/gas pipeline sections show a significant difference between the results of the two techniques. This can be attributed to the DT method assessing the overall average specific electrical resistance of the coating, including two pipeline sections in sequence and the oil/gas station located between them. Many irregular shapes, such as T-joints, elbows, and similar components at oil/gas stations, are field-applied and protected with coatings, such as two-part liquid epoxies. The aging rates of field-applied epoxy coatings, based on

coating breakdown factors, are considerably higher than those of factory-applied 3LPE coating [35,49]. At the same time, the LCA method assesses the average specific electrical coating resistance of straight pipeline sections between defined LC test points, without considering oil/gas stations with irregularly shaped epoxy coatings. Thus, the coating aging rates of straight pipeline sections coated with factory-applied 3LPE and field-applied 2LPE coatings only are much lower than those of epoxy-coated sections. Accordingly, the LCA method's results are more appropriate to model the aging of polyethylene coating.

Table 2. Summary of coating average specific electrical resistance results of oil/gas and water pipelines using LCA and DT methods.

Pipeline Designation (*)	Initial Coating Electrical Resistance, $\Omega \cdot m^2$ (**)	Type of Test	Coating Electrical Resistance, $\Omega \cdot m^2$, vs. Service Time, Years			
			8 Years	9 Years	10 Years	11 Years
N1	2.0×10^6	LCA Test	-	1.2×10^6	0.9×10^6	0.7×10^6
N2	1.0×10^6	LCA Test	-	0.6×10^6	0.5×10^6	0.4×10^6
N3	19.2×10^6	LCA Test	-	8.7×10^6	8.4×10^6	8.7×10^6
N4	4.3×10^6	LCA Test	-	-	(***)	-
G1	16.9×10^6	LCA Test	10.0×10^6	-	9.6×10^6	9.4×10^6
	16.9×10^6	Drainage Test	-	-	1.4×10^6	-
G2	10.9×10^6	LCA Test	7.1×10^6	-	6.4×10^6	5.4×10^6
	10.9×10^6	Drainage Test	-	-	1.4×10^6	-
G3	21.0×10^6	LCA Test	-	11.8×10^6	-	9.1×10^6
	21.0×10^6	Drainage Test	-	-	0.7×10^6	-
G4	19.5×10^6	LCA Test	-	11.6×10^6	-	9.2×10^6
	19.5×10^6	Drainage Test	-	-	0.7×10^6	-

(*)— "N"—for water pipelines; "G"—for oil/gas pipelines. (**)—The initial average specific electrical resistance of the polymer-coated pipeline underground sections after installation and backfilling conducted by the Drainage Test. (***)—The N4 water pipeline has been crossed with the HVAC power lines (400 kV). The unreliable results obtained with the LCA method make it inappropriate for this technique.

Figure 5 and Table 3 provide analysis of the average specific electrical resistance of the 3LPE-coated oil/gas pipelines over time, based on the experimental results of the LCA method only.

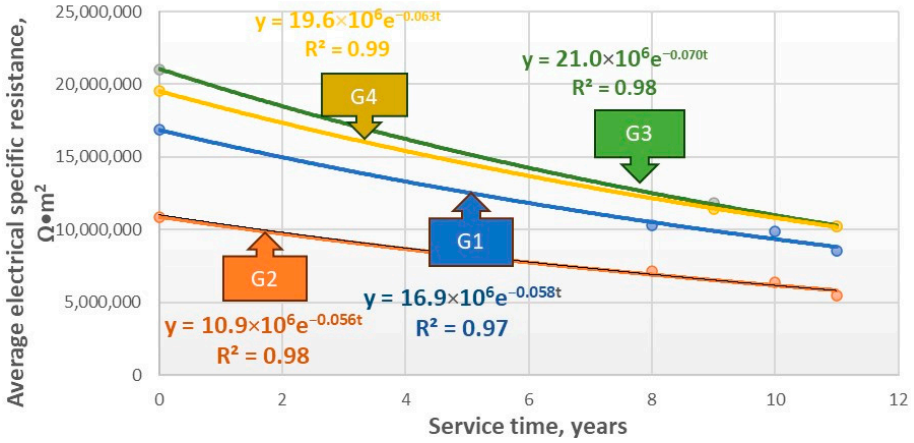


Figure 5. Aging exponential model based on average specific electrical coating resistance for G1, G2, G3, and G4 oil/gas pipeline sections based on the LCA method.

Table 3. Prediction model, coefficients of determinations, and calculated average aging coefficients of oil/gas pipelines based on the LCA method.

Pipeline Designation	Prediction Model	Calculated Average Aging Coefficient, α, 1/Year
G51	$R_C(t) = 16.9 \times 10^6 e^{-0.058t}$	0.058
G52	$R_C(t) = 10.9 \times 10^6 e^{-0.056t}$	0.056
G53	$R_C(t) = 21.0 \times 10^6 e^{-0.070t}$	0.070
G54	$R_C(t) = 19.6 \times 10^6 e^{-0.063t}$	0.063

From the data obtained using the investigated oil/gas pipeline sections, one can identify the following key patterns and dependencies:

a. The exponential model demonstrated a high determination fitting coefficient (R^2) for predicting the aging of 3LPE-coated steel pipelines.
b. Aging coefficients were determined and defined in the range from 0.05 to 0.07. Thus, the data suggests that 3LPE-coated pipelines exhibit minimal aging and are expected to have a long service life.
c. The initial average specific electrical resistance of the coating system is a key factor affecting the aging coefficient. The higher it is, the faster the degradation.

The evaluation of average specific electrical resistance over time for 3LPE-coated water pipelines is provided in Figure 6 and Table 4, using results from LCA experiments.

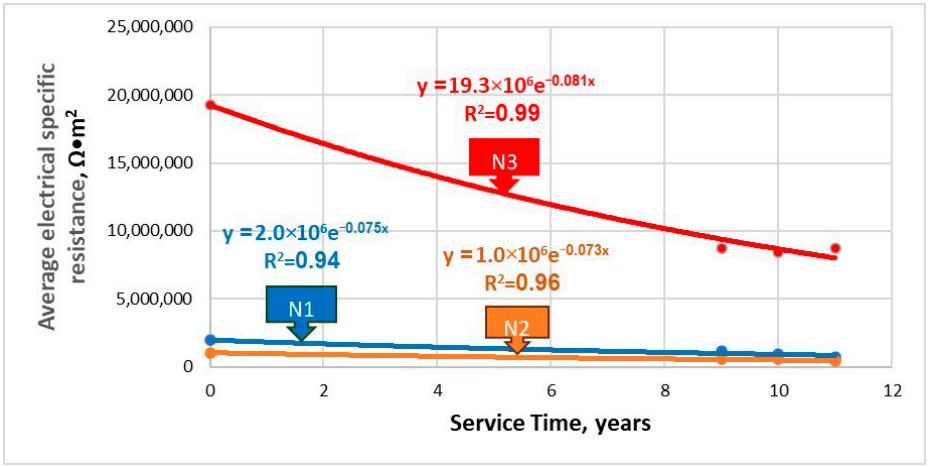


Figure 6. Aging exponential model based on average specific electrical coating resistance for N1, N2, and N3 water pipelines.

Table 4. Prediction model, coefficients of determination, and calculated average aging coefficients of water pipelines.

Pipeline Designation	Prediction Model	Calculated Average Aging Coefficient, α, 1/Year
N1	$R_C(t) = 2.0 \times 10^6 e^{-0.075t}$	0.075
N2	$R_C(t) = 1.0 \times 10^6 e^{-0.073t}$	0.073
N3	$R_C(t) = 19.3 \times 10^6 e^{-0.081t}$	0.081

Analysis of the data collected from the investigated water pipeline sections reveals the following conclusions:

d. The predictive model based on the exponential model was shown to estimate the aging of 3LPE-coated steel pipelines, with high determination fitting coefficients.
e. The aging coefficients fall within the range of 0.07 to 0.09, which exceeds the aging coefficient determined in oil/gas pipelines, suggesting a higher aging rate. However, the aging coefficient range is also low, proposing a relatively long service life.
f. Coating systems with high initial electrical resistance tend to exhibit higher aging rates.
g. An LCA-based inspection was conducted on one of the water pipelines intersecting a 400 kV AC high-voltage power line (HVAC). Unreliable results were obtained, making it inapt for such method. This conclusion is also relevant to oil/gas pipelines.

From the above-derived exponential aging phenomena, a general exponential correlation can be proposed for predicting the aging behavior of polyethylene-coated pipelines. The aging model is based on the exponential decay of coating average specific electrical resistance over time (26):

$$R_c(t) = R_c(0)e^{-\alpha t}$$

where:

$R_c(t)$—the average coating electrical resistance after service time t in underground exposure [$\Omega \cdot m^2$];
$R_c(0)$—the initial average coating electrical resistance after installation and backfilling (t = 0) [$\Omega \cdot m^2$];
α—the aging rate coefficient [1/year];
t—service time [years].

Analysis of the aging curves plotted in first-order coordinates reveals a strong correlation between the concentration of the various hydrophilic functional chemical groups and the electrical resistance, as indicated by a high correlation coefficient (R^2).

Under the assumptions noted in paragraph 4, an exponential correlation between electrical resistance and time was obtained, consistent with experimental findings.

The aging coefficients, the calculated range, average, minimum, and maximum criteria for oil/gas and water pipelines, are summarized in Tables 5 and 6 and Figure 7.

After establishing the exponential aging prediction model (26), Figure 7 shows the predictive average specific electrical resistance of the protective coating up to 50 years in underground service.

Figure 7 points out the following conclusions:

a. 3LPE-coated buried pipelines used for water and oil/gas exhibit low aging rates.
b. Coatings that initially have higher average specific electrical resistance are more prone to faster aging than those with lower initial electrical resistance.
c. For oil/gas pipelines, the aging coefficient α [1/year] changes in the range of 0.05–0.07; for water pipelines, in the range of 0.07–0.09. This indicates that 3LPE external coatings in oil/gas pipelines age more slowly than in water pipelines. The higher aging rates of polymer-coated water pipelines are primarily due to differences in coating technical characteristics compared to oil/gas pipelines, which contain numerous irregular geometrical connections (T-joints, elbows, consumer connections, air and drainage points). Most of the connections are coated with epoxy coatings applied in field conditions, which have significantly higher aging rates (coating breakdown factors) than the factory-applied 3LPE coating [35,49]. For oil/gas pipelines, the polymer-coated pipeline sections are usually constructed without epoxy-coated irregular geometrical shapes and adhere to strict quality control methods and pipeline installation procedures.

Table 5. Summary of aging coefficients for oil/gas and water pipelines.

The Aging Coefficient, 1/Year	Oil/Gas Pipelines	Water Pipelines
Calculated Range	0.06 ± 0.01	0.08 ± 0.01
Average	0.06	0.08
Minimum	0.05	0.07
Maximum	0.07	0.09
The General Prediction Model	$Rc(t) = Rc(0)e^{-0.05 \div 0.07t}$	$Rc(t) = Rc(0)e^{-0.07 \div 0.09t}$

Table 6. Summarized calculated prediction results of coating average specific electrical resistances ($\Omega \cdot m^2$) over time for oil/gas and water pipelines with different aging coefficients (*).

Service Time, Years	Oil/Gas Pipelines			Water Pipelines		
	$\alpha = 0.05$	$\alpha = 0.06$	$\alpha = 0.07$	$\alpha = 0.07$	$\alpha = 0.08$	$\alpha = 0.09$
0	1.0×10^7	1.0×10^7	1.0×10^7	3.0×10^6	3.0×10^6	3.0×10^6
10	6.1×10^6	5.5×10^6	5.0×10^6	1.5×10^6	1.3×10^6	1.2×10^6
20	3.7×10^6	3.0×10^6	2.5×10^6	0.7×10^6	0.6×10^6	0.5×10^6
30	2.2×10^6	1.7×10^6	1.2×10^6	0.4×10^6	0.3×10^6	0.2×10^6
40	1.4×10^6	0.9×10^6	0.6×10^6	0.2×10^6	0.1×10^6	0.8×10^5
50	0.8×10^6	0.5×10^6	0.3×10^6	0.9×10^5	0.5×10^5	0.3×10^5

(*)—The following assumptions have been made: the prediction exponential aging model: $Rc(t) = Rc(0) \cdot \exp(-\alpha t)$ [$\Omega \cdot m^2$]. For oil/gas pipelines, the initial coating average specific electrical resistance is -1.0×10^7 $\Omega \cdot m^2$. The initial coating average specific electrical resistance for water pipelines is 3.0×10^6 $\Omega \cdot m^2$.

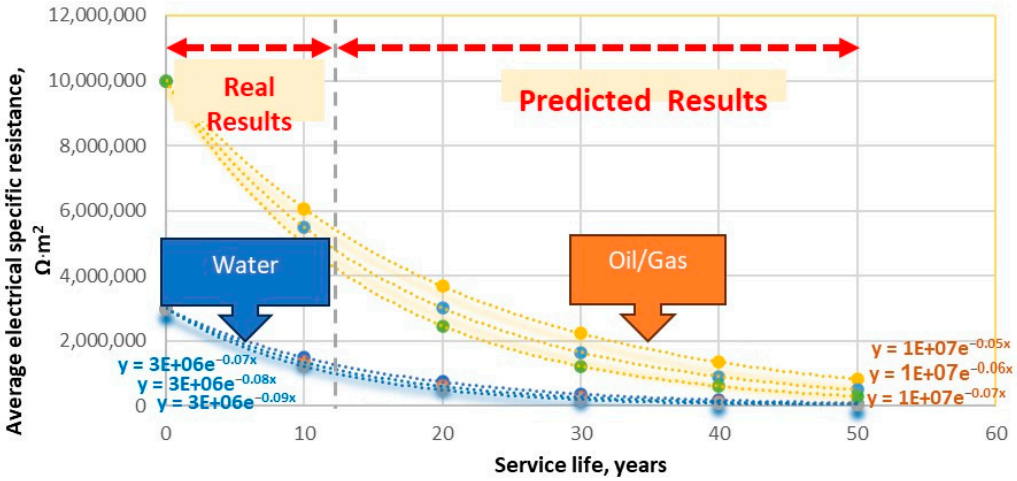


Figure 7. Aging prediction model of coating average specific electrical resistances ($\Omega \cdot m^2$) for oil/gas and water pipelines with different aging coefficients.

It should be emphasized that some of the results are inaccurate due to the influence of AC-induced voltage; therefore, underground pipelines coated with high-dielectric characteristics and located under parallel or across high-voltage power lines (HVAC: 161 or 400 kV) typically yield unreliable results by LCA testing methods. Hence, the attempt to test the 100-inch diameter water pipeline (N4) resulted in a wide range of results, which were used to establish, albeit ineffectively, the aging coefficients for the prediction model.

Another source of inaccuracies has been traced to instrument limitations due to voltage drops in the pipeline, as the measurements are around 1 μV. Hence, sensitive amplifier

voltmeters paired with data loggers are necessary. Furthermore, performing several LCA tests is recommended to ensure consistent results. The input impedance of the voltmeters should be at least 10 MΩ.

The exponential prediction aging model (26) of the 3LPE-coated buried pipelines contradicts the conclusions of the reference [38] and supports the aging model proposed in references [39–41], with several significant differences:

a. The aging coefficient spans across a broader range (0.05 to 0.07 year^{-1}) for oil and gas pipelines, indicating a potentially higher aging rate than the above-cited sources.
b. Water pipelines exhibit a higher aging rate coefficient range (0.07–0.09 year^{-1}) than oil/gas pipelines. This is primarily attributed to the frequent presence of field irregular geometry joints, such as T-connections and elbows, where two-part epoxy coatings are often applied in the field.
c. Epoxy coatings have significantly higher aging rates, based on coating breakdown factors, than the factory-applied 3LPE coating [33].

Consequently, it is possible to calculate the residual lifetime for each coated pipeline section at any given operational time. In Appendix A, the examples for calculations of water and gas/oil pipelines are given:

- For the water pipeline, section N1 (L = 4920 m; Ø = 16″, initial average specific electrical resistance is 2.0 × 10^6 Ω·m^2, minimum threshold of average electrical resistance for repair or replacement is 3 × 10^4 Ω·m^2), the selected calculated aging coefficient α is 0.08 year $^{-1}$. Since the operational time of this pipeline section is 10 years, the residual lifetime is 42.3 years.
- For the oil/gas pipeline, section G2 (L= 7757 m; Ø = 18′, the initial average specific electrical resistance is 10.9 × 10^6 Ω·m^2, minimum threshold of average specific electrical resistance for repair or replacement is 3 × 10^4 Ω·m^2), the selected aging coefficient α = 0.06 year $^{-1}$. Since the operational time of the pipeline section is 10 years, the residual lifetime is 88.2 years.

It should be noted that the aging rate of protective coatings is usually higher than that of structural materials (steel). Therefore, insulation degradation does not necessarily indicate corrosion or deterioration of the pipe itself.

6. Conclusions

The new HDPE material, initially hydrophobic and nonpolar, undergoes oxidation during aging, forming hydrophilic functional groups (e.g., alcohols, aldehydes, ketones, ethers, carboxyl, carbonyl). These groups absorb water, reducing the coating's electrical resistance. Their formation is assumed to follow first-order kinetics, implying that the electrical resistance correlates with their concentration. Aging exhibits an exponential trend governed by a standard first-order differential equation, where the reaction rate decreases proportionally to the coating's average specific electrical resistance over time.

The aging model for polyethylene-coated underground pipelines was proposed and validated. The model was based on the modified Line Current Attenuation test results used for water and oil/gas pipelines following various service periods and technical parameters. The oldest underground pipeline investigated in this study was 11 years old.

The following main insights have been derived from this study:

The Line Current Attenuation (LCA) method has proven to be an accurate technique for estimating the specific electrical resistances of 3LPE-coated steel pipelines over their service life (besides the cases of AC-induced voltage)

A first-order exponential aging model was developed based on the coating's average specific electrical resistance for water and oil/gas pipelines.

Increased initial electrical resistances of polyethylene coatings are directly associated with higher aging coefficients and aging rates.

The study demonstrated that the coating aging rates of 3LPE-coated water pipelines and oil/gas pipelines are comparatively low. Thus, it can be concluded that degradation in the polyethylene coating exceeding the allowable resistance criterion is caused by localized defects rather than overall coating aging [32,33]. As a result, detecting and repairing local sections with low coating electrical resistance is vital. This can be achieved by utilizing External Corrosion Direct Assessment (ECDA), indirect inspection methods such as Alternating Current Attenuation Survey (Electromagnetic Method) and/or by Alternating Current Voltage Gradient/ Direct Current Voltage Gradient (ACVG/DCVG), as well as by direct examinations, during which the insulation is inspected in prioritized test pits according to the standards [25,76]. Post-assessment stage should be combined as the final step of the coating's and steel's condition assessment [25].

Precision equipment capable of recording and analyzing results is essential, especially in oil/gas pipelines where insulation exhibits high-dielectric characteristics.

To establish a reliable method for determining the average specific electrical resistance of 3LPE-coated buried pipelines under the influence of high-voltage AC power lines (HVAC), further research is required.

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Appendix A. Examples of the Residual Isolation Lifetime Calculations for Oil/Gas and Water Pipelines

- The calculation for water pipeline section N1.

The general formula for predicting the aging behavior of coating average specific electrical resistance over time (for N1 water pipeline section) is as follows:

$$R_C(t) = 2.0 \times 10^6 e^{-0.080t}$$

After 10 years of service, the average specific electrical resistance of coating system, $R_c(10)$ will be calculated as follows:

$$R_C(t) = 2.0 \times 10^6 e^{-0.080 \times 10} = 8.8 \times 10^5\ \Omega \cdot m^2$$

The calibrated line current passed at the starting point is 922.25 mA, and at the end LC point, it is 917.77 mA. The difference between them is $\Delta I = 4.48$ mA.

The formula for calculating the pipeline section area is $A = \pi \cdot D \cdot L \Rightarrow$, $A = 6273\ m^2$.

The current density calculation is as follows (after 10 years): $j = \Delta I/A = 4.48/6273 = 0.71\ \mu A/m^2$. The current density measured on coated pipelines after a decade is better than

the standard's recommended range of 1 to 20 μA/m^2 for optimized designs, as indicated in Table 3 [36].

The initial current density (0 years) is as follows: 0.58 μA/m^2 (from Drainage Test results).

Specific ratio of coating defects is as follows: For an average specific electrical resistance of 8.8×10^5 Ω·m^2, the coating condition is good, with a range of specific area ratio of coating defects $-0.01 \div 1.50$ mm^2/m^2 and a defect size classification of several very small (IR < 1), minor ($1 \leq$ %IR < 3), or single moderate defects ($3 \leq$ IR < 15).

The expected whole lifetime of the protective coating until overall repair is as follows:

$$R_c(t) = R_c(0)e^{-\alpha t} \Rightarrow \frac{R_c(t)}{R_c(0)} = e^{-\alpha t} \Rightarrow ln\,\frac{R_c(t)}{R_c(0)} = -\alpha t \Rightarrow ln\,\frac{30{,}000}{2.0 \times 10^6} = -0.08t$$

$$t = 52.5 \text{ years}$$

The expected remaining duration of the protective coating until overall repair is as follows:

$$52.5 - 10 = 42.5 \text{ years.}$$

- The calculation for oil/gas pipeline section G2.

The general formula for predicting the aging behavior of coating average specific electrical resistance over time (for G2 oil/gas pipeline section) is as follows:

$$R_C(t) = 10.9 \times 10^6 e^{-0.060t}$$

After 10 years of service, the average specific electrical resistance $R_c(10)$ will be calculated as follows:

$$R_C(t) = 10.9 \times 10^6 e^{-0.060 \times 10} = 6.0 \cdot 10^6 \text{ Ω·m}^2$$

The calibrated line current passed at the starting point is 3.45 mA; at the end, the LC point is zero mA (isolation joint). The difference between them is ΔI = 3.45 mA.

The formula for calculating the pipeline section area is A = π·D·L ⇛ A = 11.136 m^2.

The current density calculation (after 10 years) is j = is ΔI/A = 3.45/11.136 = 0.31 μA/m^2. The measured current density for coated pipelines after 10 years is much better than the standard's recommended range of 1 to 20 μA/m^2 for optimized designs, as shown in Table [36].

The initial current density (0 years) is 0.07 μA/m^2 (from Drainage Test results).

Specific ratio of coating defects: For a coating with an average specific electrical resistance of $6.0 \cdot 10^6$ Ω·m^2, the coating condition is excellent, with the specific area ratio of coating defects below 0.01 mm^2/m^2, with defect size classified as minimal single defects (IR < 1) in internal technical documents of the water company.

The expected whole lifetime of the protective coating until repair is calculated as follows:

$$R_c(t) = R_c(0)e^{-\alpha t} \Rightarrow \frac{R_c(t)}{R_c(0)} = e^{-\alpha t} \Rightarrow ln\,\frac{R_c(t)}{R_c(0)} = -\alpha t \Rightarrow ln\,\frac{30{,}000}{6.0 \times 10^6} = -0.06t$$

where t = 98.2 years, and the residual lifetime is 88.2 years

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