

Supplementary material

Facile Synthesizing Yolk–shelled Fe₃O₄@carbon Nanocavities with Balanced Physiochemical Synergism as Efficient Hosts for High-performance Lithium–sulfur Batteries



Figure S1. The optical photograph of Fe-BTC gel.

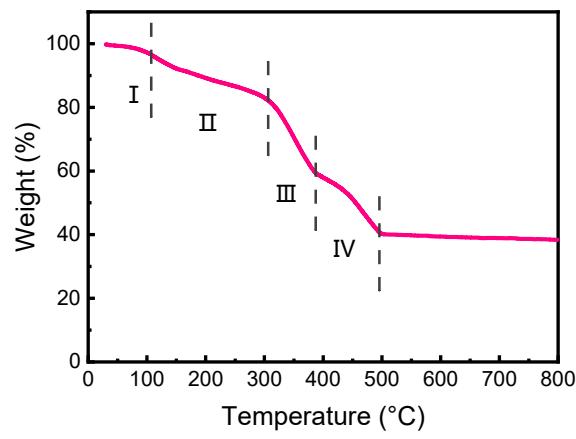


Figure S2. TGA curve of Fe-BTC aerogel acquired in a nitrogen atmosphere.

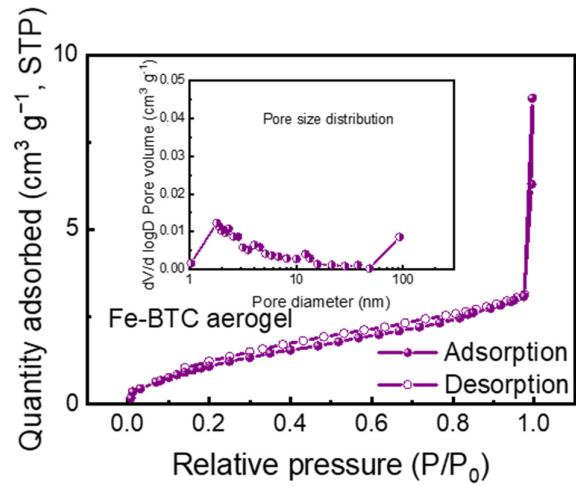


Figure S3. Nitrogen adsorption/desorption isotherm of Fe-BTC aerogel with an inset showing the corresponding size distribution of pores.

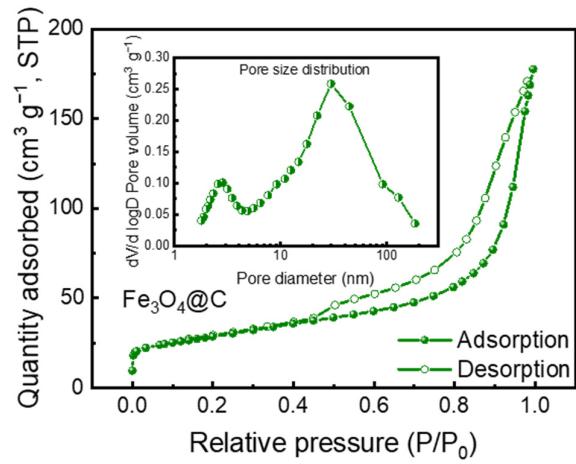


Figure S4. Nitrogen adsorption/desorption isotherm of $\text{Fe}_3\text{O}_4@\text{C}$ with an inset showing the corresponding size distribution of pores.

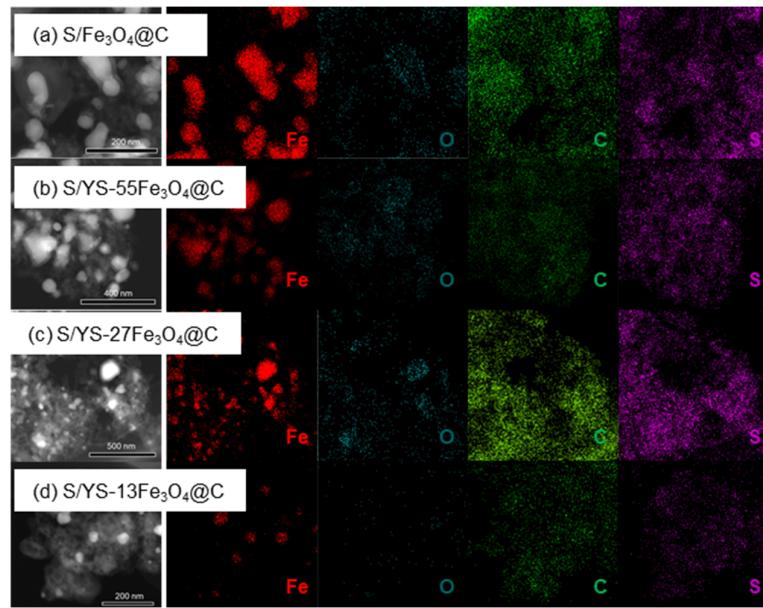


Figure S5. EDS elemental mapping images of (a) S/Fe₃O₄@C, (b) S/YS-55Fe₃O₄@C, (c) S/YS-27Fe₃O₄@C, and (d) S/YS-13Fe₃O₄@C.

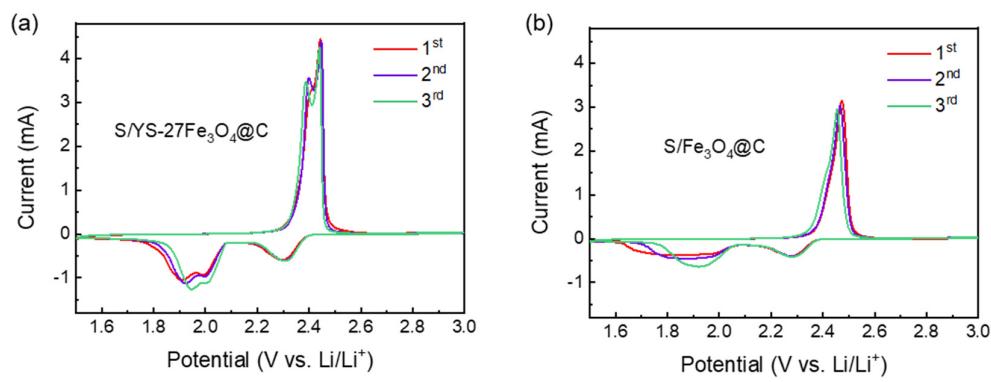


Figure S6. CV curves of (a) S/YS-27Fe₃O₄@C and (b) S/Fe₃O₄@C at 0.1 mV s⁻¹ for different cycles.

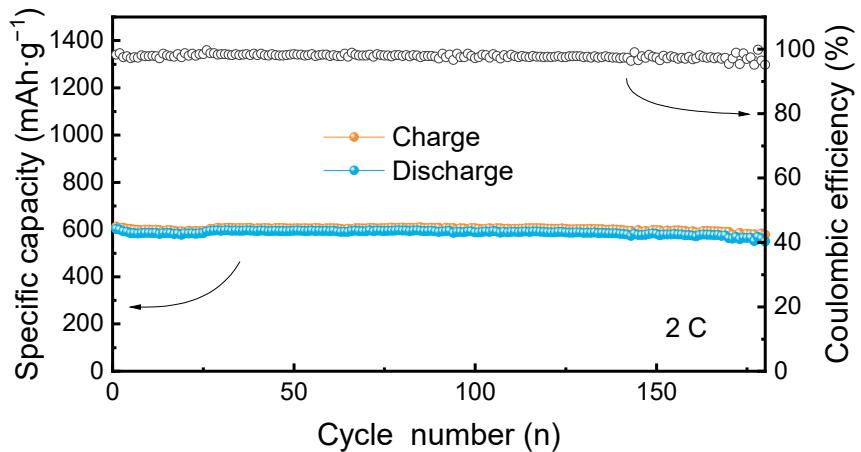


Figure S7. Long-term cycling stability of S/YS-27Fe₃O₄@C electrode at 2 C.

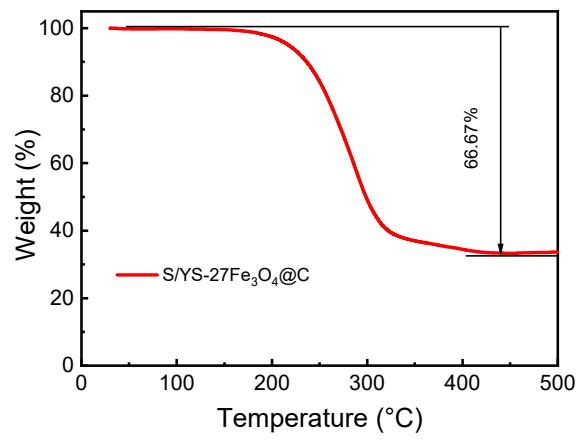


Figure S8. TGA curves of S/YS-27Fe₃O₄@C in N₂.

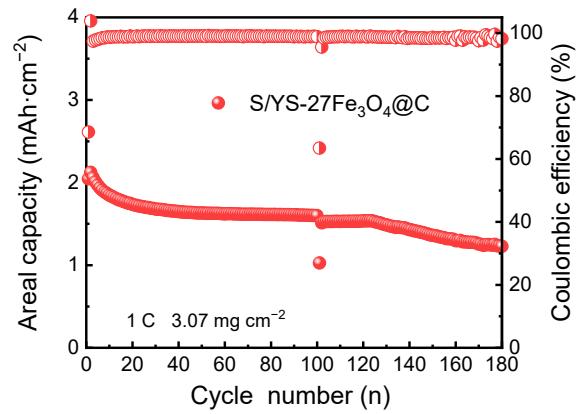


Figure S9. Cycle performance at 1 C of S/YS-27Fe₃O₄@C under high sulfur loadings.

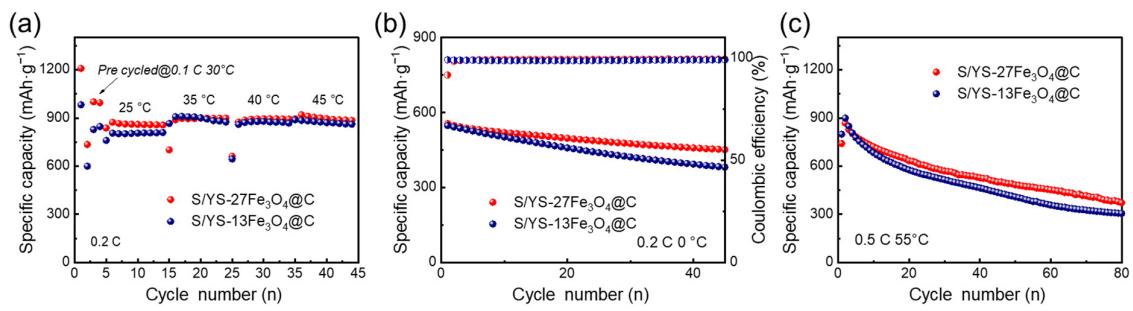


Figure S10. Cycle performance of S/YS-27Fe₃O₄@C and S/YS-13Fe₃O₄@C cathode with a wide temperature range: (a) 25 – 45 °C, (b) 0°C, (c) 55 °C.

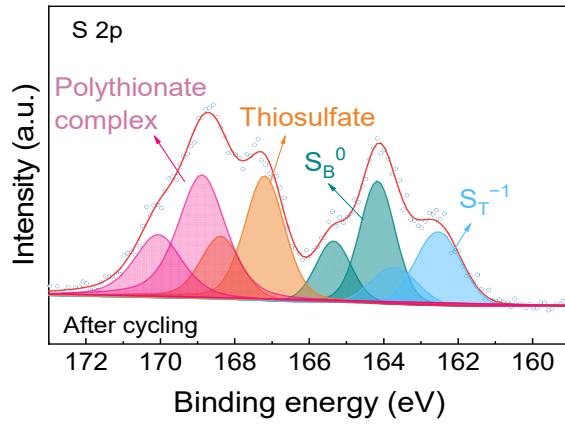


Figure S11. S 2p in S/YS-27Fe₃O₄@C after cycling.

Table S1. Pore structure parameters of Fe-BTC aerogel and Fe₃O₄@C.

	Specific surface area (m ² g ⁻¹)	Average pore diameter (nm)	Pore volume (cm ³ g ⁻¹)
Fe-BTC aerogel	36.1859	2.16362	0.019573
Fe ₃ O ₄ @C	99.3583	7.37165	0.183109