

## **Slow Magnetic Relaxation in a One-Dimensional Coordination Polymer Constructed from Hepta-coordinate Cobalt(II) Nodes**

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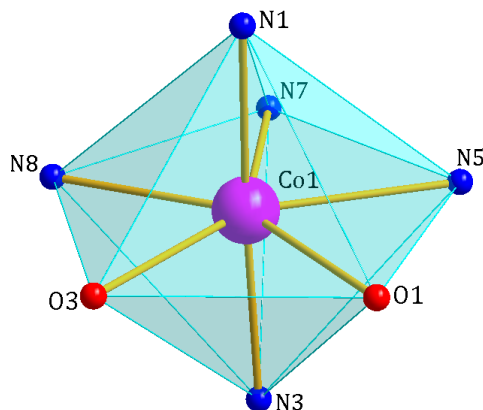
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**Table S1.** X-ray Crystallographic Data and Refinement Parameters for complex **1**.

<b>1</b>	
Formula	C <sub>29</sub> H <sub>31</sub> CoN <sub>9</sub> O <sub>6</sub>
M <sub>w</sub> (g mol <sup>-1</sup> )	660.56
Crystal size (mm)	0.45×0.18×0.16
Crystal system	Orthorhombic
Space group	<i>Pbca</i>
T (K)	293(2)
a (Å)	19.6105(6)
b (Å)	15.8275(5)
c (Å)	20.3903(6)
α (°)	90
β (°)	90
γ (°)	90
V (Å <sup>3</sup> )	6328.9(3)
Z	8
ρ <sub>calcd</sub> (g cm <sup>-3</sup> )	1.387
μ(MoKα) (mm <sup>-1</sup> )	0.598
<i>F</i> (000)	2744.0
T <sub>max</sub> , T <sub>min</sub>	0.919, 0.869
h, k, l range	-26 ≤ <i>h</i> ≤ 26, -21 ≤ <i>k</i> ≤ 21, -27 ≤ <i>l</i> ≤ 27
Collected reflections	7676
Independent reflections	5767
Goodness-of-fit (GOF) on F <sup>2</sup>	1.095
R1, wR2 ( <i>I</i> > 2σ <i>I</i> )	0.0607, 0.1813

R1, wR2 (all data)	0.0812, 0.1970
CCDC Number	1487267
$R1 = \Sigma  Fo  -  Fc  /\Sigma Fo $ and $wR2 =  \Sigma w( Fo ^2 -  Fc ^2) /\Sigma w(Fo)^2 ^{1/2}$	



**Figure S1.** Distorted pentagonal bipyramid coordination geometry around the Co<sup>II</sup> center in **1**.

**Table S2.** Summary of SHAPE analysis for complex **1**.

HP-7	1	D <sub>7h</sub>	Heptagon
HPY-7	2	C <sub>6v</sub>	Hexagonal pyramid
PBPY-7	3	D <sub>5h</sub>	Pentagonal bipyramid
COC-7	4	C <sub>3v</sub>	Capped octahedron
CTPR-7	5	C <sub>2v</sub>	Capped trigonal prism
JPBPY-7	6	D <sub>5h</sub>	Johnson pentagonal bipyramid J13
JETPY-7	7	C <sub>3v</sub>	Johnson elongated triangular pyramid J7

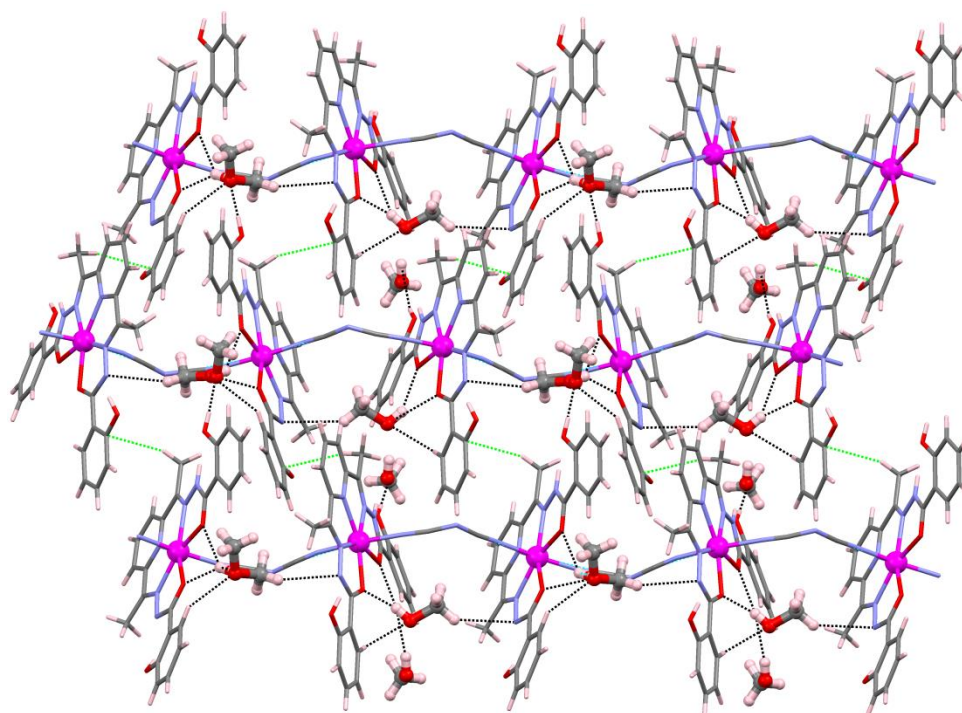
Structure [ML <sub>7</sub> ]	HP-7	HPY-7	PBPY-7	COC-7	CTPR-7	JPBPY-7	JETPY-7
Complex <b>1</b>	33.909	23.175	<b>0.244</b>	7.271	5.752	2.952	23.537

**Table S3.** Bond distances (Å) and bond angles (°) around Co<sup>II</sup> center found in complex **1**.

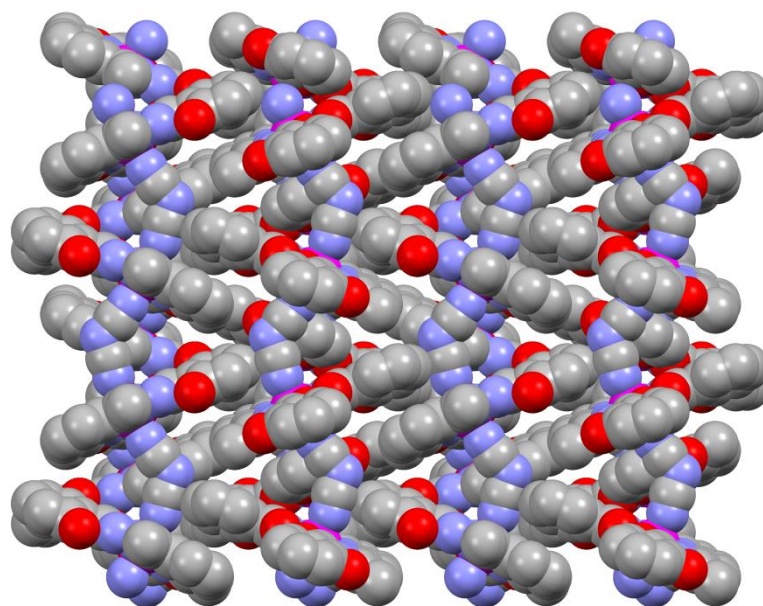
Bond distance (Å)			Bond angle (°)		
Complex <b>1</b>			Complex <b>1</b>		
	Co1—O1	2.2315(1)		O1—Co1—O3	76.82(5)
	Co1—O3	2.1897(1)		O1—Co1—N1	87.80(5)
	Co1—N1	2.0829(1)		O1—Co1—N5	70.62(5)
	Co1—N5	2.2149(1)		O1—Co1—N7	139.44(5)
	Co1—N7	2.2035(1)		O1—Co1—N8	149.21(5)
	Co1—N8	2.1408(1)		O1—Co1—N3	85.51(5)
	Co1—N3	2.0813(1)		O3—Co1—N1	89.61(5)
				O3—Co1—N5	147.38(5)
				O3—Co1—N7	143.73(6)
				O3—Co1—N8	72.40(6)
				O3—Co1—N3	87.57(6)
				N1—Co1—N5	91.11(6)
				N1—Co1—N7	92.04(6)
				N1—Co1—N8	92.80(6)



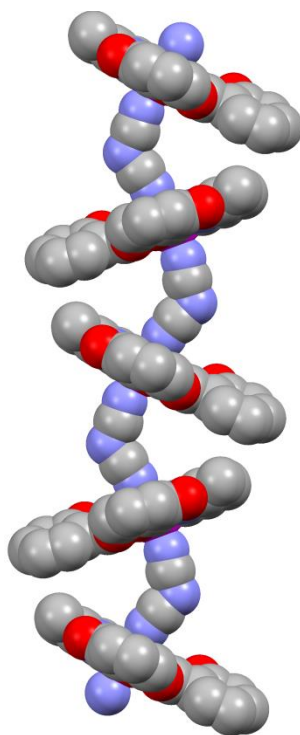
N1—Co1—N3	173.18(6)
N5—Co1—N7	68.84(6)
N5—Co1—N8	140.08(6)
N5—Co1—N3	87.93(6)
N7—Co1—N8	71.33(5)
N7—Co1—N3	93.91(5)
N8—Co1—N3	92.27(5)



**Figure S2.** A view of supramolecular 2D arrangement of complex **1** through intermolecular H-bonding and CH $\cdots$  $\pi$  interactions.



**Figure S3.** A view of de-solvated framework of **1** emphasizing the supramolecular interactions.

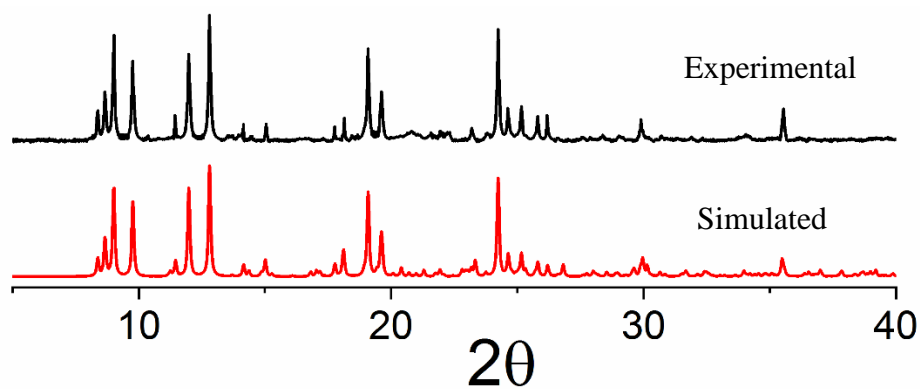


**Figure S4.** Helical 1D arrangement of complex **1** along the *c* axis.

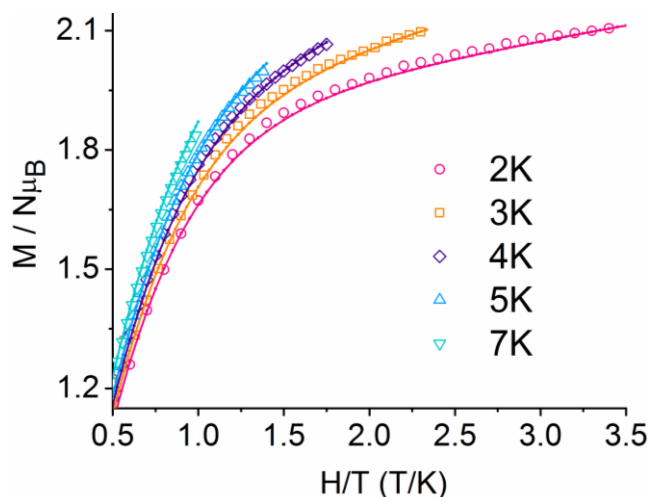
**Table S4.** H-bond parameters found in complex **1**.

D—H $\cdots$ A	D—H(Å)	H $\cdots$ A(Å)	D $\cdots$ A (Å)	<D—H—A(°)	Symmetry <sup>#</sup>
O5—H5 $\cdots$ O1	0.82	2.39	3.070(4)	141	0
O5—H5 $\cdots$ O3	0.82	2.14	2.834(4)	143	0
C54—H54A $\cdots$ O3	0.96	2.51	3.395(6)	154	0
O2—H2 $\cdots$ O32A	0.82	1.76	2.550(4)	162	1
C2—H2A $\cdots$ N4	0.93	2.48	3.397(7)	168	2
O32A—H32A $\cdots$ O5	0.82	1.87	2.668(4)	165	3

<sup>#</sup> (0) *x*,*y*,*z*; (1) 1/2+*x*,*y*,1/2-*z*; (2) 1/2-*x*,1-*y*,-1/2+*z*; (3) -1/2+*x*,1/2-*y*,1-*z*.



**Figure S5.** PXRD for complex **1**.



**Figure S6.**  $M/N\mu_B$  vs.  $H/T$  plots at the indicated temperatures for complex **1**. The solid lines are the best fit.

**Table S5.** Magnetic anisotropy ( $D$  parameters) and SIM parameters for previously reported seven-coordinated  $\text{Co}^{\text{II}}$  SIMs in the literature.

Complex	$D$ ( $\text{cm}^{-1}$ )	$\tau_0$ (s)	$U_{\text{eff}}$ (K)	Ref.
$[\text{Co}(\text{dapbhH}_2)(\text{H}_2\text{O})(\text{NO}_3)](\text{NO}_3)$	32.4	$6.0 \times 10^{-10}$	81.2	1, 2
$[\text{CoL}_{\text{N5}}(\text{H}_2\text{O})_2]\text{Cl}_2 \cdot 4\text{H}_2\text{O}$	24.6	$1.2 \times 10^{-6}$	29.8	2
$[\text{Co}(\text{dapbhH}_2)(\text{im})_2] \cdot \text{H}_2\text{O}$	24.8	$8.7 \times 10^{-11}$	89.6	2
$[\text{Co}(\text{dapbhH}_2)\text{I}(\text{NO}_3)]\text{I}$	30	nr	nr	3
$[\text{Co}(\text{dapbhH}_2)\text{Br}(\text{NO}_3)]\text{Br}$	30	nr	nr	3
$[\text{Co}(\text{L}_1)\text{Cl}_2]$	40	nr	nr	4
$[\text{Co}(\text{L}_2)](\text{ClO}_4)_2$	26	nr	nr	5
$[\text{Co}(\text{L}_2)](\text{NO}_3)_2$	25	nr	nr	5
$[\text{Co}(\text{dapbhH}_2)(\text{SCN})_2]$	15.9	nr	nr	6
$[\text{Co}(\text{dapbh})(\text{H}_2\text{O})_2]$	13.1	nr	nr	6
$[\text{Co}(\text{L})\text{Cl}_2] \cdot 2\text{CH}_3\text{OH}$	38	$1.1 \times 10^{-5}$	7.9	7
$[\text{Co}(\text{L})\text{Br}_2]$	41	$1.1 \times 10^{-5}$	6.1	7
$[\text{Co}(\text{L})\text{I}_2]$	35	$1.1 \times 10^{-5}$	6.5	7
$[\text{Co}(\text{tdmmb})(\text{H}_2\text{O})_2][\text{BF}_4]_2$	25.6	$1.1 \times 10^{-6}$	42.2	8
$[\text{Co}(\text{tdmmb})(\text{CN})_2] \cdot 2\text{H}_2\text{O}$	17.4	$3.2 \times 10^{-7}$	48.9	8
$[\text{Co}(\text{tdmmb})(\text{NCS})_2]$	26.3	$1.0 \times 10^{-6}$	49.2	8
$[\text{Co}(\text{tdmmb})(\text{SPh})_2]$	34.5	$2.1 \times 10^{-7}$	54.7	8
$[\text{Co}(\text{4-tert-butylpyridine})_3(\text{NO}_3)_2]$	35.8	$3.7 \times 10^{-9}$	52.6	9
$[\text{Co}(\text{isoquinoline})_3(\text{NO}_3)_2]$	35.7	$9.1 \times 10^{-8}$	28.8	9
$[(\text{OTfpy})\text{Co}(\kappa_2\text{-O}_2\text{NO})_2]$	-41.4	nr	nr	10
$[\text{Co}(\text{H}_2\text{daps})(\text{MeOH})_2]$	43.1	$7.4 \times 10^{-6}$	33.5	11
$[\text{Co}(\text{H}_4\text{daps})(\text{NCS})(\text{MeOH})] \cdot (\text{ClO}_4) \cdot (\text{MeOH})$	41.5	$5.6 \times 10^{-6}$	28.4	11
$[\text{Co}(\text{H}_4\text{daps})(\text{NCS})_2] \cdot (\text{MeOH})_2$	38.8	$4.8 \times 10^{-6}$	23.6	11
$[\text{Co}(\text{-DAPBH})(\text{NO}_3)(\text{H}_2\text{O})](\text{NO}_3)$	30	nr	50	12
$[\text{Co}(\text{H}_4\text{L})(\text{DMF})(\text{H}_2\text{O})](\text{NO}_3)_2 \cdot (\text{DMF})$	35.9	$6.5 \times 10^{-6}$	25	13
$[\text{Co}(\text{H}_4\text{L})(\text{MeOH})(\text{H}_2\text{O})](\text{NO}_3)_2 \cdot (\text{MeOH})$	37.2	$3.5 \times 10^{-6}$	15	13
$[\text{Co}(\text{H}_4\text{L})(\text{DEF})(\text{H}_2\text{O})](\text{NO}_3)_2$	43.7	$1.9 \times 10^{-6}$	4	13
$[\text{Co}(\text{H}_3\text{daps})(\text{dca})] \cdot (\text{MeOH})_2 \cdot (\text{MeCN})$	41.3	$5.3 \times 10^{-6}$	9.9	This paper

nr = not reported.

**Table S6.** Ab initio calculated energies (cm<sup>-1</sup>) of the lowest states ( $S = 3/2$ ) of complex **1**.

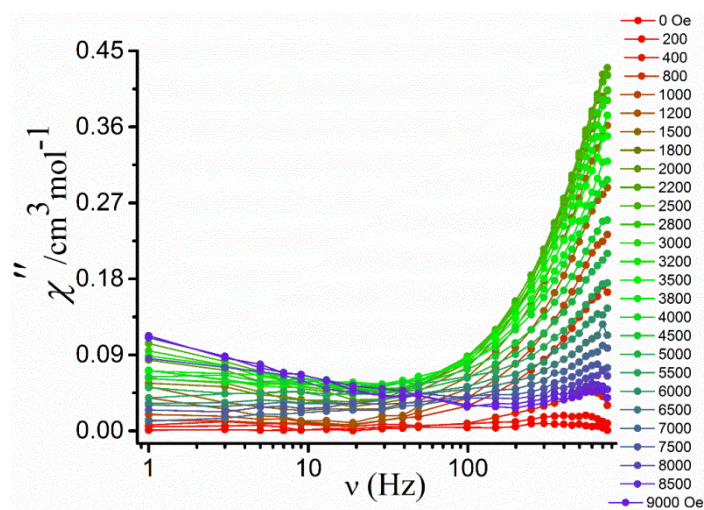
spin-free states	Spin-orbit states
0.000	0.000
2667.229	0.000
2917.222	88.169
4420.566	88.169
5045.421	2189.501
11036.401	2189.501
12350.207	2662.497
16705.986	2662.497
17772.836	3175.830
19500.981	3175.830
	3529.968
	3529.968

**Table S7.** Ab initio calculated magnetic anisotropy in the ground state and first excited state (w.r.t.  $\hat{s} = 1/2$ ) for complex **1**.

Ground state	First excited state
$g_x = 5.21$	$g_x = 0.37$
$g_y = 4.37$	$g_y = 0.37$
$g_z = 1.97$	$g_z = 5.96$

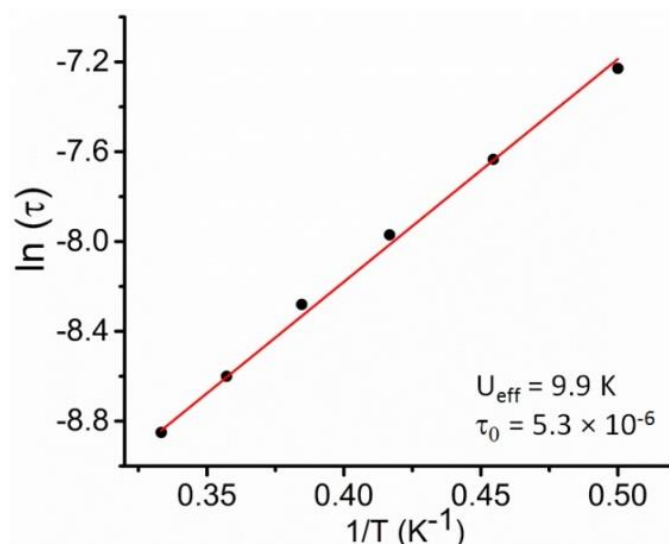
**Table S8.** Energy of the first four excited states (cm<sup>-1</sup>) and their contribution to the  $D$  and  $E$  values in cm<sup>-1</sup> at CAS(7,5) NEVPT2 level by ORCA.

State	Mult	Complex <b>1</b>		
		Energy	$D$	$E$
1st	4	3663.5	2.955	-0.454
2nd	4	4049.8	0.868	0.385
3rd	4	5839.4	13.713	11.159
4th	4	6652.6	11.105	-9.458



**Figure S7.** Frequency dependency of out-of-phase susceptibility at different external magnetic field (0-0.9 T) and 2 K temperature.





**Figure S8.**  $\ln(\tau)$  vs  $1/T$  plot for complex **1**.

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