

Article

Novel 4,6-Disubstituted *s*-Triazin-2-yl Amino Acid Derivatives as Promising Antifungal Agents

Rakia Abd Alhameed ¹, Zainab Almarhoon ¹, Essam N. Sholkamy ², Salman Ali Khan ³, Zaheer Ul-Haq ³, Anamika Sharma ^{4,5}, Beatriz G. de la Torre ^{4,5}, Fernando Albericio ^{1,5,6,7,*} and Ayman El-Faham ^{1,8,*}

- ¹ Department of Chemistry, College of Science, King Saud University, P.O. Box 2455, Riyadh 11451, Saudi Arabia; Roki.ahmed@yahoo.com (R.A.A.); zalmarhoon@ksu.edu.sa (Z.A.)
- ² Department of Botany and Microbiology, College of Science, King Saud University, P.O. Box 2455, Riyadh 11451, Saudi Arabia; essam_92003@yahoo.com
- ³ Dr. Panjwani Center for Molecular medicine and Drug Research, International Center for Chemical and Biological Sciences, University of Karachi, Karachi 75270, Pakistan; salmanali9329176@gmail.com (S.A.K.); zaheer_qasmi@hotmail.com (Z.U.-H.)
- ⁴ KwaZulu-Natal Research Innovation and Sequencing Platform (KRISP), School of Laboratory Medicine and Medical Sciences, College of Health Sciences, University of KwaZulu-Natal, Durban 4041, South Africa; anamika.aug14@gmail.com (A.S.); garciadelatorreb@ukzn.ac.za (B.G.d.l.T.)
- ⁵ Peptide Science Laboratory, School of Chemistry and Physics, University of KwaZulu-Natal, Durban 4001, South Africa
- ⁶ CIBER-BBN (Networking Centre on Bioengineering, Biomaterials and Nanomedicine) and Department of Organic Chemistry, University of Barcelona, 08028 Barcelona, Spain
- ⁷ Institute for Advanced Chemistry of Catalonia (IQAC-CSIC), 08034 Barcelona, Spain
- ⁸ Chemistry Department, Faculty of Science, Alexandria University, P.O. Box 426, Ibrahimia 12321, Alexandria, Egypt
- * Correspondence: albericio@ub.edu or Albericio@ukzn.ac.za (F.A.); aelfaham@ksu.edu.sa or aymanel_faham@hotmail.com (A.E.-F.); Tel.: +966-114673195 (A.E.-F.)

Received: 9 September 2020; Accepted: 15 October 2020; Published: 21 October 2020



MDP

Abstract: A novel series of 4,6-disubstituted *s*-triazin-2-yl amino acid derivatives was prepared and characterized. Most of them showed antifungal activity against *Candida albicans* compared to clotrimazole (standard drug). Compounds bearing aniline derivatives, piperidine and glycine on the triazine core showed the highest inhibition zones at concentrations of 50, 100, 200, and 300 µg per disc. In addition, docking studies revealed that all the compounds accommodated well in the active site residues of *N*-myristoltransferase (NMT) and exhibited complementarity, which explains the observed antifungal activity. Interestingly, none of these compounds showed antibacterial activity.

Keywords: 2,4,6-trichlorotriazine; TCT; triazine amino acid; antifungal; *Candida albicans*; *N*-myristoltransferase

1. Introduction

An important component of a drug design program's quest for new leads is the synthesis of molecules that are novel but mimic recognized biologically active molecules by virtue of the existence of certain crucial structural characteristics, especially in the development of several specific synthetic protocols for the preparation of a range of substituted *s*-triazine derivatives owing to their involvement in several applications in the medicinal field and the synthesis of numerous pharmaceutical agents [1,2]. The reported literature reveals that substituted *s*-triazine derivatives are consistent with a number of

marked antimicrobial activities against *Staphylococcus aureus*, *Bacillus cereus*, *Bacillus subtilis*, *Salmonella typhimurium*, *Escherichia coli Klebsiella aerogenes*, and *Candida albicans* as human microbial pathogens [3].

C. albicans is a constituent of the normal part of the human commensal flora that lives in the human mouth and gastrointestinal tract. However, it is also is a diploid fungus that produces as both yeast and filamentous cells [4]. Systemic fungal infections (fungemias) including those by *C. albicans* have emerged as important causes of morbidity and mortality in immunocompromised patients (such as those with AIDS or under antitumor chemotherapy) [5,6]. In addition, hospital-acquired infections by *C. albicans* have become a cause of major health concerns [7–10].

In medicinal chemistry, the *s*-triazine ring has proved to be a privileged structure, and, therefore, its derivatives have been extensively studied against a broad number of biological targets. In this respect, *s*-triazine derivatives have shown antiprotozoal [11], anti-HIV [12], anticancer [13,14], antimalarial [15,16], antibacterial [17–19], antifungal [20–22], and antileishmanial activity [23], carbonic anhydrase inhibitors [24,25], and human monoamine oxidase (MAO) inhibitors [26].

Cyanuric chloride (TCT) is the starting reagent for the synthesis of *s*-triazine derivatives. TCT is a symmetrical tridentate compound (it has three reactive Cl) with the particularity from a synthetic point of view that once the first Cl has reacted, the two remaining Cl show different reactivity. This allows TCT to undergo sequential nucleophilic substitution by different nucleophiles (S, O, N) under a controlled temperature to furnish a full array of *s*-triazine derivatives [1,27,28].

Based on the previously reported data, here, we focused on the preparation of a novel series of disubstituted *s*-triazin-2-yl amino acid derivatives (**5a**–**q**) containing cyclic amines (morpholine, piperidine, or pyrrolidine), and 4-substituted aniline derivatives which are biologically interesting moieties [29]. To this end, we used different techniques, including conventional heating, microwave irradiation and ultrasound. The resulted disubstituted *s*-triazin-2-yl amino acid derivatives were screened against bacteria and fungi as a preliminary biological screening. In addition, molecular docking studies with *N*-Myristoltransferase (NMT) as an anti-fungal target will be discussed.

2. Materials and Methods

All reagents, chemicals and solvents were purchased from commercial suppliers. Reactions were monitored by using thin-layer chromatography (TLC, silica gel 60-F254 protected aluminum sheets). Melting points were conducted in open capillary tubes using a Gallenkamp melting point apparatus (Sigma-Aldrich Chemie GmbH, Taufkirchen, Germany) and were uncorrected. Fourier transform infrared spectroscopy (FTIR) spectra were recorded on a Nicolet 6700 spectrometer (Thermo Electron Scientific Instruments Corporation, Madison, WI, USA) using KBr discs. ¹H- and ¹³C-NMR spectra were recorded on a Varian-Agilent-NMR 600 MHz spectrometer (Varian, Inc., Palo Altro, CA, USA). Elemental analyses were recorded on a Perkin-Elmer 2400 elemental analyzer (Perkin-Elmer Inc. Waltham, MA, USA). The ultrasonic bath was purchased from Selecta (Barcelona, Spain). Microwave experiments were performed in a multimode reactor (Monowave 300, Aton Paar GmbH, 1400 W maximum magnetron, Germany). Heating time (5 min at 70 °C and 350 watt) and reaction time were held for 12–15 min at identical watt and temperature with stirring speed (800 rpm); then, cooling was performed until 25 °C.

2.1. General Procedure for the Synthesis of Disubstituted Amino Acids-Triazine Derivatives

2.1.1. One-Pot Synthesis

A mixture of amino acid (10 mmol) and NaHCO₃ (22 mmol) in 50 mL distilled water was added dropwise to a solution of cyanuric chloride (TCT; 1, 10 mmol) in 50 mL acetone at 0 °C. The reaction mixture was kept under stirring at 0 °C for 2 h, and then a solution of the amine (10 mmol) in 10 mL acetone was added dropwise (5 min) at 0 °C, followed by addition of a solution of NaHCO₃ (11 mmol) in 20 mL distilled water. After complete addition, the reaction mixture was stirred overnight at room temperature. The solvent was concentrated under vacuum and then extracted with dichloromethane

(10 mL). The aqueous layer was collected and acidified with 1N HCl. The white precipitate of the products (**4a**–**f**) was collected by filtration.

2.1.2. Stepwise Synthesis

The method was performed in the following two steps: Step (i) synthesis of monosubstituted *s*-triazine **3**: a solution of the amine (1 equiv.) in 10 mL acetone was added dropwise at 0 °C in 5 min to a mixture of cyanuric chloride (1 equiv.) in 50 mL acetone, followed by the addition of a solution of NaHCO₃ (1.5 Equation.) in 20 mL distilled water. The reaction mixture was kept under stirring at 0 °C for 2 h to afford mono substituted *s*-triazine derivatives. Step (ii) synthesis of disubstituted amino acid s-triazine: a mixture of amino acid (1 equiv.) and NaHCO₃ (2.5 eq.) in 50 mL distilled water was added dropwise (10 min) to a solution of the previously prepared monosubstituted *s*-triazine **3** (1 equiv.) in 50 mL acetone; the reaction mixture was stirred at room temperature overnight and then acidified with 1N HCl to furnish the target compounds **4a–f**.

2.2. General Procedures for the Synthesis of 4,6,-Disubstituted s-Triazin-2-yl Amino Acid Derivatives 5a-q

Method A: Conventional heating

Disubstituted *s*-triazine (1 mmol) was reacted with different amines (morpholine, piperidine and pyrrolidine; 1.1 mmol) in tetrahydrofuran (THF, 20 mL) and in the presence of diisopropylethylamine (DIEA; 2.2 mmol) under reflux for 24 h. The solvent was evaporated under vacuum to produce a white precipitate, which was neutralized by 1N HCl to afford trisubstituted *s*-triazine derivatives in 89–93% yield.

Method B: Microwave technique

Disubstituted *s*-triazine (1 mmol) was dissolved in 5 mL THF, and then 1.1 mmol of amine was added, followed by DIEA (2.2 mmol). This mixture was then subjected to microwave irradiation using a multimode reactor (Monowave 300, Aton Paar GmbH), where heating time (5 min, at 70 °C, 350 watt) and the reaction were held for 12–15 min at the same wattage and temperature with a stirring speed of 800 rpm. After cooling, the solvent was evaporated, and the resulting residue was subjected to neutralization with 1N HCl to obtain trisubstituted *s*-triazine derivatives in excellent yields and purities.

Method C: Ultrasonication method

An amine (1.1 mmol) was added to a solution of disubstituted-triazine (1 mmol) in 20 mL THF in the presence of DIEA (2.2 mmol), and the reaction mixture was subjected to ultrasound irradiation (1.5 h) at room temperature. TLC was used to monitor the completion of reaction, followed by workup similar to conventional methodology to obtain the product with highly yield (90–93%) and purity.

All compounds obtained from the three methods were purified via washing with a hot mixture of THF and a few drops of acetone.

2.2.1. (4-Morpholino-6-(Phenylamino)-1,3,5-Triazin-2-yl) Glycine (5a)

White solid in 89% (Method A), 91% (Method B) 92% (Method C) yield; mp 245–247 °C. IR (KBr, cm⁻¹): 3341 (NH); 1669 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 3.60 (4H, s, CH₂-O-CH₂), 3.66 (4H, s, CH₂-N-CH₂), 3.87 (2H, dd, *J* = 5.4 Hz, CH₂COOH), 6.90 (1H, s.br, H_{4'}), 7.19–7.23 (3H, m, H_{3'}, H_{5'} & NH), 7.68 (2H, dd, *J* = 7.2 Hz, H_{2'} & H_{6'}), 9.08 (1H, s, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 42.5 (CH₂COOH); 43.4 (CH₂-N-CH₂), 66.2 (CH₂-O-CH₂), 119.7 (C_{2'} & C_{6'}); 121.6 (C_{4'}); 128.4 (C_{3'} & C_{5'}); 140.4 (C_{1'}); 164.2, 164.8 & 165.9 (triazine), 172.3 (CO). Anal. Calcd for C₁₅H₁₈N₆O₃ (330.35): C, 54.54; H, 5.49; N, 25.44. Found: C, 54.22; H, 5.56; N, 25.63; HRMS: *m/z*: calcd. 331.35[M+H]⁺; Found: 331.08.

2.2.2. (4-((4-Chlorophenyl) Amino)-6-Morpholino-1,3,5-Triazin-2-yl) Glycine (5b)

White powder in 87% (Method A), 90% (Method B), 92% (Method C) yield; mp 246–248 °C. IR (KBr, cm⁻¹): 3350 & 3412 (NH); 1678 (CO). ¹H-NMR (DMSO-d₆; ppm): δ 3.60 (4H, s, CH₂-O-CH₂), 3.65 (4H, s, CH₂-N-CH₂), 3.87 (2H, dd, *J* = 4.8, 5.4 Hz, CH₂COOH), 7.23–7.27 (3H, m, H₃', H₅' & NH), 7.71

(2H, dd, J = 7.2, 8.4 Hz, $H_{2'} \& H_{6'}$), 9.25 (1H, s, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 42.7 (CH₂COOH); 43.7 (CH₂-N-CH₂), 66.4 (CH₂-O-CH₂), 121.5 (C_{2'} & C_{6'}); 125.4 (C_{4'}); 128.5 (C_{3'} & C_{5'}); 139.7 (C_{1'}); 164.2, 164.9 & 165.9 (triazine), 172.5 (CO). Anal. Calcd for C₁₅H₁₇ClN₆O₃ (364.79): C, 49.39; H, 4.70; N, 23.04. Found: C, 49.54; H, 4.81; N, 23.19; HRMS: *m*/*z*: calcd. 365.80[M+H]⁺, Found: 365.64.

2.2.3. (4-((4-Methoxyphenyl) Amino)-6-Morpholino-1,3,5-Triazin-2-yl) Glycine (5c)

Off-white powder in 86% (Method A), 90% (Method B), 90% (Method C) yield; mp 246–248 °C. IR (KBr, cm⁻¹): 3349 (NH); 1674 (CO);¹H-NMR (DMSO-d₆; ppm): δ 3.59 (4H, s, CH₂-O-CH₂), 3.64 (4H, s, CH₂-N-CH₂), 3.69 (3H, s, OCH₃), 3.83 (2H, d, *J* = 5.4 Hz, CH₂COOH), 6.80 (2H, t, *J* = 7.8, 10.8 Hz, H₃·& H₅·), 7.54 (2H, d, *J* = 9 Hz, H₂· & H₆·), 7.57 (1H, s, NH), 8.91 (1H, s, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 42.7 (CH₂COOH); 43.6 (CH₂-N-CH₂), 55.4 (OCH₃), 66.3 (CH₂-O-CH₂), 113.8 (C₃· & C₅·); 121.6 (C₂·& C₆·); 133.7 (C₁·); 154.5 (C₄·); 164.1, 165.0 & 166.0 (triazine), 172.6 (CO). Anal. Calcd for C₁₆H₂₀N₆O₄ (360.37): C, 53.33; H, 5.59; N, 23.32. Found: C, 53.54; H, 5.71; N, 23.17; HRMS: *m*/*z*: calcd. 361.38 [M+H]⁺; Found: 361.13.

2.2.4. (4-(Phenylamino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Glycine (5d)

White powder in 89% (Method A), 93% (Method B), 92% (Method C) yield; mp 268–270 °C. IR (KBr, cm⁻¹): 3339 (NH); 1674 (CO);¹H-NMR (DMSO-d₆; ppm): δ 1.46 (4H, s, H_{3'} & H_{5'}), 1.59 (2H, s, H_{4'}), 3.67 (4H, s, CH₂-N-CH₂), 3.86 (2H, dd, *J* = 5.4, 6 Hz, NHCH₂COOH), 6.88 (1H, t, *J* = 7.2, 7.8 Hz, H_{4"}), 7.08 (1H, s, NH), 7.20 (2H, q, *J* = 7.8, 8.4 Hz, H_{3"} & H_{5"}), 7.69 (2H, dd, *J* = 7.2, 8.4 Hz, H_{2"} & H_{6"}), 8.96 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 24.5 (C_{4'}), 25.6(C_{3'}& C_{5'}), 42.5 (HOOC-CH₂NH), 43.7 (C_{2'}& C_{6'}); 119.6 (C_{2"} & C_{6"}); 121.4 (C_{4"}); 128.4 (C_{3"} & C_{5"}); 140.6 (C_{1"}); 164.1, 165. 8, 166.2 (triazine), 172.4 (CO). Anal. Calcd for C₁₆H₂₀N₆O₂ (328.38): C, 58.52; H, 6.14; N, 25.59. Found: C, 58.41; H, 6.10; N, 25.71. HRMS: *m/z*: calcd. 329.38 [M+H]⁺, Found: 329.13.

2.2.5. (4-((4-Chlorophenyl) Amino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Glycine (5e)

White powder in 88% (Method A), 91% (Method B), 92% (Method C) yield; mp 260–262 °C. IR (KBr, cm⁻¹): 3396 (NH); 1654 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 1.46 (4H, s, H₃'& H₅'), 1.59 (2H, s, H₄'), 3.67 (4H, s, CH₂-N-CH₂), 3.85 (2H, dd, *J* = 4.8 Hz, CH₂COOH), 7.15 (1H, s, NH), 7.24 (2H, t, *J* = 9,12.6 Hz, H_{3"} & _{H5"}), 7.73 (2H, dd, *J* = 7.2, 7.8Hz, H_{2"} & H_{6"}), 9.13 (1H, s, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 24.8 (C_{4'}), 25.8 (C_{3'}& C_{5'}), 42.8 (HOOC-CH₂NH), 44.0 (C_{2'}& C_{6'}); 121.3 (C_{2"} & C_{6"}); 125.1 (C_{4"}); 128.5 (C_{3"}& C_{5"}); 139.9 (C_{1"}); 164. 3, 166.1, 166.5 (triazine), 172.6 (CO).Anal. Calcd for C₁₆H₁₉ClN₆O₂ (362.82): C, 52.97; H, 5.28; N, 23.16. Found: C, 53.09; H, 5.32; N, 23.40. HRMS: *m/z*: calcd. 363.83 [M+H]⁺; Found: 363.80.

2.2.6. (4-((4-Methoxyphenyl) Amino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Glycine (5f)

Off-white powder in 87% (Method A), 90% (Method B), 90% (Method C) yield; mp 261–263 °C. IR (KBr, cm⁻¹): 3260 (NH); 1667 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 1.45 (4H, s, H₃' & H₅'), 1.59 (2H, s, H₄'), 3. 66 (4H, s, CH₂-N-CH₂), 3.68 (3H, s, OCH₃), 3.84 (2H, dd, CH₂COOH), 6.80 (2H, dd, *J* = 9, 10.8 Hz, H₃" & H₅"), 7.01 (1H, s, NH), 7.57 (2H, dd, *J* = 6, 8.4Hz, H₂" & H₆"), 8.81 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 24.8 (C₄'), 25.8 (C₃' & C₅'), 42.8 (CH₂COOH), 44.0 (C₂' & C₆'); 55.6 (OCH₃), 113.9 (C₃" & C₅"), 121.39 (C₂" & C₆"); 134.0 (C₁"); 154.4 (C₄"), 164. 3, 164.5, 166.4 (triazine), 172.7 (CO). Anal. Calcd for C₁₇H₂₂N₆O₃ (358.40): C, 56.97; H, 6.19; N, 23.45. Found C, 56.72; H, 6.30; N, 23.67. HRMS: *m/z*: calcd. 359.41 [M+H]⁺; Found: 359.10.

2.2.7. (4-((4-Chlorophenyl) Amino)-6-(Pyrrolidin-1-yl)-1,3,5-Triazin-2-yl) Glycine (5g)

White powder in 86% (Method A), 92% (Method B), 91% (Method C) yield; mp 262–264 °C. IR (KBr, cm⁻¹): 3371, 3246 (NH); 1661 (CO). ¹H-NMR (DMSO-d₆; ppm): δ 1.86 (4H, s, H_{3"} & H_{4"}), 3.39–3.46 (4H, m, H_{2"} & H_{5"}), 3.86 (2H, d, *J* = 5.4 Hz, NH CH₂COOH), 7.10 (1H, d, NH). 7.23 (2H, t, *J* = 9, 9.6 Hz,

 $H_{3'}$ & $H_{5'}$), 7.80 (2H, dd, *J* = 7.2, 8.4 Hz, $H_{2'}$ & $H_{6'}$), 9.13 (1H, d, NH);¹³C-NMR (DMSO-d₆; ppm): δ 25.3 (C_{3"} & C_{4"}), 42.7 (NHCH₂COOH), 46.5 (C_{2"} & C_{5"}), 121.6 (C_{2'} & C_{6'}), 125.8 (C_{4'}), 128.6 (C_{3'} & C_{5'}), 139.8 (C_{1'}); 161.3, 163.1, 164.7 (triazine), 172.5 (CO).Anal. Calcd for C₁₅H₁₇ClN₆O₂ (348.79): C, 51.65; H, 4.91; N, 24.10. Found C, 51.81; H, 5.02; N, 24.01. HRMS: *m*/*z*: calcd. 349.80 [M+H]⁺; Found: 349.63.

2.2.8. (4-((4-Methoxyphenyl) Amino)-6-(Pyrrolidin-1-yl)-1,3,5-Triazin-2-yl) Glycine (5h)

White powder in 86% (Method A), 90% (Method B), 93% (Method C) yield; mp 270–272 °C. IR (KBr, cm⁻¹): 3262 (NH); 1668 (CO);¹H-NMR (DMSO-d₆; ppm): δ 1.85 (4H, s, H_{3"} & H_{4"}), 3.44 (4H, s, H_{2"} & H_{5"}), 3.68 (3H, s, OMe), 3.86 (2H, s, NH-CH₂COOH), 6.79 (2H, d, *J* = 7.8 Hz, H_{3′} & H_{5′}), 6.90 (1H, s, NH), 7.66 (2H, d, *J* = 7.8 Hz, H_{2′} & H_{6′}), 8.78 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 25.3 (C_{3"} & C_{4"}), 42.7 (CH₂COOH), 46.1 (C_{2"} & C_{5"}), 55.6 (OMe), 113.9 (C_{3′} & C_{5′}), 121.2 (C_{2′} & C_{6′}), 134.4 (C_{1′}); 154.3 (C_{4′}), 163.5, 165.8, 166.1 (triazine), 172.7 (CO). Anal. Calcd for C₁₆H₂₀N₆O₃ (344.38): C, 55.80; H, 5.85; N, 24.40. Found: C, 55.99; H, 5.97; N, 24.64. HRMS: *m/z*: calcd. 345.38 [M+H]⁺; Found: 345.56.

2.2.9. 3-((4-Morpholino-6-(Phenylamino)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5i)

White powder in 84% (Method A), 91% (Method B), 93% (Method C) yield; mp 267–269 °C. IR (KBr, cm⁻¹): 3285 (NH); 1666 (CO). ¹H-NMR (DMSO-d₆; ppm): δ 2.50 (2H, t, *J* = 7.2 Hz, CH₂COOH), 3.45 (2H, q, *J* = 10.8 HZ, NH-CH₂CH₂), 3.60 (4H, dd, *J* = 7.2 Hz, CH₂-O-CH₂), 3.66 (4H, dd, CH₂-N-CH₂), 6.89 (1H, s, H_{4'}), 6.93 (1H, s, NH). 7.21 (2H, d, *J* = 7.8 Hz, H_{3'} & H_{5'}), 7.71(2H, d, *J* = 7.8 Hz, H_{2'} & H_{6'}), 9.00 (1H, d, NH), 12.19 (1H, s, COOH);¹³C-NMR (DMSO-d₆; ppm): δ 34.5 (CH₂CH₂COOH); 36.8 (NH CH₂CH₂COOH), 43.7 (CH₂-N-CH₂), 66.5 (CH₂-O-CH₂), 119.9 (C_{2'} & C_{6'}); 121.7 (C_{4'}); 128.7 (C_{3'} & C_{5'}); 140.9 (C_{1'}); 164.4, 165.1 & 165.9 (triazine), 173.6 (CO). Anal. Calcd for C₁₆H₂₀N₆O₃ (344.38): C, 55.80; H, 5.85; N, 24.40. Found: C, 55.80; H, 5.85; N, 24.40. HRMS: *m/z*: calcd. 345.38 [M+H]⁺; Found: 345.44.

2.2.10. 3-((4-((4-Chlorophenyl) Amino)-6-Morpholino-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5j)

White powder in 86% (Method A), 93% (Method B), 90% (Method C) yield; mp 285–287 °C. IR (KBr, cm⁻¹): 3277 (NH); 1678 (CO). ¹H-NMR (DMSO-d₆; ppm): δ 2.48 (2H, s, CH₂COOH), 3.44 (2H, q, *J* = 7.2 Hz, NH-CH₂CH₂), 3.60 (4H, d, *J* = 7.2 Hz, CH₂-O-CH₂), 3.64 (4H, dd, CH₂-N-CH₂), 6.99 (1H, s, NH), 7.24 (2H, d, *J* = 7.8 Hz, H₃, & H₅), 7.75 (2H, d, *J* = 7.2 HZ, H₂, & H₆), 9.17 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 34.4 (CH₂CH₂COOH), 36.8 (NHCH₂CH₂); 43.7 (CH₂-N-CH₂), 66.4 (CH₂-O-CH₂), 121.3 (C₂, C₆); 125.3 (C₄); 128.6 (C₃, C₅); 140.0 (C₁); 164.3, 165.0 & 165.8 (triazine), 173.5 (CO). Anal. Calcd for C₁₆H₁₉ClN₆O₃ (378.82): C, 50.73; H, 5.06; N, 22.19. Found: C, 50.95; H, 5.23; N, 22.45. HRMS: *m/z*: calcd. 379.82 [M+H]⁺; Found: 379.76.

2.2.11. 3-((4-((4-Methoxyphenyl) Amino)-6-Morpholino-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5k)

Off-white powder in 88% (Method A), 90% (Method B), 92% (Method C) yield; mp 269–271 °C. IR (KBr, cm⁻¹): 3273 (NH); 1667 (CO).¹H-NMR (DMSO-d₆; ppm): δ 2.49 (2H, s, CH₂COOH), 3.43 (2H, d, *J* = 4.2 Hz, NHCH₂CH₂), 3.59–3.66 (8H, m, CH₂-O-CH₂ & CH₂-N-CH₂), 3.69 (3H, s, OCH₃), 6.79 (1H, s, NH), 6.80 (2H, d, *J* = 8.4 Hz, H₃' & H₅'), 7.57 (2H, s, H₂' & H₆'), 8.83 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 34.7 (CH₂COOH); 37.0 (NHCH₂CH₂COOH), 44.0 (CH₂-N-CH₂), 55.7 (OCH₃), 66.7 (CH₂-O-CH₂), 114.1 (C₃' & C₅'); 121.9 (C₂' & C₆'); 134.1 (C₁'); 154. 8 (C₄'); 164.5, 165.3 & 165.9 (triazine), 173.89 (CO). Anal. Calcd for C₁₇H₂₂N₆O₄ (374.40): C, 54.54; H, 5.92; N, 22.45. Found: C, 54.75; H, 6.08; N, 22.69. HRMS: *m/z*: calcd. 375.40 [M+H]⁺; Found: 375.61.

2.2.12. 3-((4-(Phenylamino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (51)

White powder in 89% (Method A); 92% (Method B), 94% (Method C) yield; mp 276–278 °C. IR (KBr, cm⁻¹): 3275 (NH); 1663 (CO).¹H-NMR (DMSO-d₆; ppm): δ 1.46 (4H, s, H_{3'} & H_{5'}), 1.59 (2H,

s, H4[']), 2.49 2H, s, CH₂ COOH), 3.45 (2H, t, *J* = 6, 10.8 Hz, CH₂NH), 3.66 (4H, d, CH₂-N-CH₂), 6.83 (1H, s, NH), 6.87 (1H, s, H_{4"}), 7.20 (2H, t, *J* = 7.2,7.8 Hz, H_{3"} & H_{5"}), 7.72(2H, d, *J* = 7.8 Hz, H_{2"} & H_{6"}), 8.92 (1H, d, NH);¹³C-NMR (DMSO-d₆; ppm): δ 24.6 (C_{4'}), 25.3 (C_{3'}& C_{5'}), 34.3 (CH₂-COOH), 36.5 (NHCH₂CH₂COOH), 43.7 (C_{2'}& C_{6'}); 119.5 (C_{2"} & C_{6"}); 121.3 (C_{4"}); 128.4 (C_{3"}& C_{5"}); 140.8 (C_{1"}); 164. 4, 165.7, 165.9 (triazine), 173.4 (CO). Anal. Calcd for C₁₇H₂₂N₆O₂ (342.40): C, 59.63; H, 6.48; N, 24.54. Found: C, 59.89; H, 6.66; N, 24.87. HRMS: *m/z*: calcd. 343.40 [M+H]⁺; Found: 343.61.

2.2.13. 3-((4-((4-Chlorophenyl) Amino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5m)

White powder in 86% (Method A) 92% (Method B) 92% (Method C) yield; mp 266–268 °C. IR (KBr, cm⁻¹): 3262 (NH); 1665 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 1.46 (4H, s, H₃' & H₅'), 1.59 (2H, s, H₄'), 2.50 (2H, s, CH₂COOH), 3.43 (2H, q, *J* = 6.6 Hz, CH₂CH₂NH), 3.67 (4H, d, CH₂-N-CH₂), 6.89 (1H, d, NH), 7.24 (2H, d, *J* = 7.8 Hz, H₃" & H₅"), 7.76 (2H, d, *J* = 8.4 Hz, H₂" & H₆"), 9.09 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 24.8 (C₄'), 25.8 (C₃' & C₅'), 34.4 (CH₂CH₂COOH), 36.8 (CH₂CH₂NH), 44.0 (C₂' & C₆'); 121.2 (C₂" & C₆"); 125.0 (C₄"); 128.5 (C₃" & C₅"); 140.1 (C₁"); 164. 5, 165.8, 166.1 (triazine), 173.5 (CO). Anal. Calcd for C₁₇H₂₁ClN₆O₂ (376.85): C, 54.18; H, 5.62; N, 22.30. Found: C, 54.33; H, 5.76; N, 22.54. HRMS: *m/z*: calcd. 377.85 [M+H]⁺; Found: 377.63.

2.2.14. 3-((4-((4-Methoxyphenyl) Amino)-6-(Piperidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5n)

Off-white powder in 89% (Method A) 94% (Method B) 93% (Method C) yield; mp 275–277 °C. IR (KBr, cm⁻¹): 3259 (NH); 1667 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 1.45 (4H, s, H₃, & H₅), 1.58 (2H, s, H₄), 2.50 (2H, s, CH₂COOH), 3. 43 (2H, s, HOOC-CH₂CH₂NH), 3.65 (4H, s, CH₂-N-CH₂), 3.68 (3H, s, OCH₃), 6.79 (2H, d, *J* = 9 Hz, H₃, & H₅), 7.59 (2H, s, H₂, & H₆), 8.84 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 24.9 (C₄), 25.8 (C₃, C₅), 34.6 (CH₂COOH), 36.8 (CH₂CH₂NH) 43.9 (C₂, C₆); 55.5 (OCH₃), 113.9 (C₃, C₅), 121.2 (C₂, & C₆); 134.2 (C₁); 154.4 (C₄), 164. 4, 164.7, 166.0 (triazine), 173.6 (CO). Anal. Calcd for C₁₈H₂₄N₆O₃ (372.43): C, 58.05; H, 6.50; N, 22.57. Found: C, 58.23; H, 6.62; N, 22.81. HRMS: *m/z*: calcd. 373.43 [M+H]⁺; Found: 373.55.

2.2.15. 3-((4-(Phenylamino)-6-(Pyrrolidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (50)

White powder in 87% (Method A) 91% (Method B) 92% (Method C) yield, mp 291–293°C. IR (KBr, cm⁻¹): 3268 (NH); 1670 (CO);¹H-NMR (DMSO-d₆): δ 1.86 (4H, s, H_{3"} & H_{4"}), 2.51 (2H, s, CH₂COOH), 3.46 (6H, s, H_{2"}, H_{5"} & NHCH₂CH₂COOH), 6.87 (1H, d, *J* = 6 Hz, H_{4'}), 7.19 (2H, d, *J* = 7.8 Hz, H_{3'} & H_{5'}), 7.79 (2H, s, H_{2'} & H_{6'}), 8.91 (1H, s, NH), 12.18 (1H, s, COOH); ¹³C-NMR (DMSO-d₆; ppm): δ 25.3 (C_{3"}& C_{4"}), 34.5 (CH₂COOH), 36.8 (NHCH₂CH₂COOH), 46.1 (C_{2"} & C_{5"}), 119.6 (C_{2'} & C_{6'}), 121.4 (C_{4'}), 128.7 (C_{3'}& C_{5'}), 141.3 (C_{1'}), 163.7, 164. 2 (triazine), 173.6 (CO). Anal. Calcd for C₁₆H₂₀N₆O₂ (328.38): C, 58.52; H, 6.14; N, 25.59. Found: C, 58.52; H, 6.14; N, 25.59. HRMS: *m/z*: calcd. 329.38 [M+H]⁺; Found: 329.35.

2.2.16. 3-((4-((4-Chlorophenyl) Amino)-6-(Pyrrolidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5p)

White powder in 89% (Method A), 92% (Method B) 90% (Method C) yield; mp 289–291 °C. IR (KBr, cm⁻¹): 3270 (NH); 1676 (CO);¹H-NMR (DMSO-d₆; ppm): δ 1.85 (4H, s, H_{3"} & H_{4"}), 2.48 (2H, s, CH₂COOH), 3.38 (6H, s, CH₂CH₂NH, H_{2"} & H_{5"}), 3.45 (4H, d, *J* = 5.4 Hz, H_{2"} & H_{5"}), 6.85 (1H, d, NH), 7.23 (2H, d, *J* = 7.2 Hz, H₃' & H₅'), 7.82 (2H, d, *J* = 8.4 Hz, H₂' & H₆'), 9.09 (1H, d, NH); ¹³C-NMR (DMSO-d₆; ppm): δ 25.3 (C_{3"} & C_{4"}),34.5 (CH₂COOH), 36.8 (CH₂CH₂NH), 46.1 (C_{2"} & C_{5"}), 121.1 (C_{2'} & C_{6'}), 124.9 (C_{4'}), 128.5 (C_{3'} & C_{5'}), 140.3 (C₁'); 163.6, 164.1, 165.8 (triazine), 173.5 (CO). Anal. Calcd for C₁₆H₁₉ClN₆O₂ (362.82): C, 52.97; H, 5.28; N, 23.16. C, 52.76; H, 5.34; N, 23.43. HRMS: *m/z*: calcd. 363.82 [M+H]⁺; Found: 363.71.

2.2.17. 3-((4-((4-Methoxyphenyl) Amino)-6-(Pyrrolidin-1-yl)-1,3,5-Triazin-2-yl) Amino) Propanoic Acid (5q)

White powder in 86% (Method A), 90% (Method B), 92% (Method C) yield; mp 273–275 °C. IR (KBr, cm⁻¹): 3261 (NH); 1660 (CO); ¹H-NMR (DMSO-d₆; ppm): δ 1.84 (4H, s, H_{3"} & H_{4"}), 2.49 (CH₂COOH), 3.39–344 (6H, m, H_{2"}, H_{5"}& NHCH₂CH₂COOH), 3.68 (3H, s, OMe), 6.79 (2H, d, *J* = 9 Hz, H_{3'} & H_{5'}), 7.66 (2H, d, *J* = 9.6 Hz, H_{2'} & H_{6'}), 8.73 (1H, d, NH); ¹³C-NMR (DMSO-d₆): δ 25.2 (C_{3"}& C_{4"}), 34.6 (CH₂COOH), 36.8 (NHCH₂CH₂COOH), 46.1 (C_{2"} & C_{5"}), 55.5 (OMe), 113.9 (C_{3'} & C_{5'}), 121.1 (C_{2'} & C_{6'}), 134.4 (C_{1'}), 154.3 (C_{4'}); 163.7, 164. 0, 165.7 (triazine), 173.6 (CO).Anal. Calcd for C₁₇H₂₂N₆O₃ (358.40): C, 56.97; H, 6.19; N, 23.45. Found: C, 57.12; H, 6.31; N, 23.29. HRMS: *m/z*: calcd. 359.40 [M+H]⁺; Found: 359.53.

2.3. Antimicrobial Activity

2.3.1. Preparation of Microbial Inoculums

The antibacterial activity of **5a–q** was evaluated against Gram-positive bacteria "*S. aureus* ATCC 29213 and *S. epidermidis* ATCC 12228", Gram-negative bacteria "*E. coli* ATCC 25922 and *S. typhimurium* ATCC 14028" and *C. albicans* ATCC 60193.

2.3.2. Antimicrobial Assay Using the Disc Diffusion Method

The microbial inoculums were prepared as described in our previous work [30]. The antibacterial activity of **5a–q** was studied by employing a micro-dilution method, using Mueller–Hinton broth and following the guidelines of the Clinical and Laboratory Standards Institute (CLSI) (M07, 11th Edition, 2018) [31], and serial dilution was performed using dimethyl sulfoxide (DMSO) to prepare 200 μ g/mL of each compound. Paper discs (6 mm diameter) previously dipped into each compound were placed gently onto the microorganism-seeded plates. The antibacterial assay plates were then incubated at 24 h at 37 ± 1 °C. The diameters of the inhibition zones were expressed and measured in mm. Each assay was repeated three times in this experiment, and the results were expressed as a mean value ± standard deviation.

2.3.3. Minimum Inhibitory (MIC) and Minimum Fungicidal Concentration (MFC)

The MICs and MFCs were carried out as described by Rasadah and Muharnad [32]. In this regard, the nutrient agar and Sabouraud agar medium were inoculated with freshly prepared cells of *C. albicans*. The discs were dipped into the DMSO containing each compound and placed on the plates at concentration ranging between 50 and 300 μ g per disc. Next, the discs were placed gently onto the surface of the Petri dishes after the bacterial growth and evaporation of DMSO (where it was used as a negative control). After the incubation of the Petri dishes at 37 °C for 24 h, antifungal activity was measured. Each assay was repeated three times in this experiment, and the results were expressed as a mean value.

2.3.4. Methodology of Docking Studies

In the present study, the crystal structure of *N*-myristoltransferase (NMT) was retrieved under the accession code 1IYL [33]. The protein structure was prepared using the protein preparation wizard in Molecular Operating Environment (MOE) Version 2018.0101 [34]. During preparation, the structures were subjected to addition of missing atoms and residues, bond order, formal charge correction, and tautomer adjustment. Hydrogens atoms were added using the protonate 3D algorithm. For protonation, GB/VI was used as the electrostatics function, with a dielectric value of 80 (for solvent). The van der Waals and electrostatics cutoffs were set at 15 and 10 A, respectively. Partial charges were applied using the AMBER10: EHT force field. The water molecules involved in the bridging ligand and protein were retained and kept rigid during energy minimization. The ligands were charged and minimized using the MMFF94x force field in MOE. The graphics were obtained using the MOE software suite.

3. Results and Discussion

3.1. Chemistry

Disubstituted *s*-triazinyl amino acid derivatives (**5a**–**q**) were prepared as demonstrated in Scheme 1 via either a one-pot synthesis (Route A) or stepwise synthetic pathway (Route B). In the one-pot synthesis, the amino acid was added to cyanuric chloride at 0 °C using aq. NaHCO₃ as base. After 2 h, substituted aniline derivatives were added in the presence of aq. NaHCO₃ as base, and the reaction was stirred overnight at rt to afford **4a**–**f**. In contrast, the stepwise strategy (Route B, Scheme 1) involved the addition of substituted aniline derivative to cyanuric chloride at 0 °C for 2 h using NaHCO₃ as base to form **3a–c**. After isolation of **3a–c**, underwent a nucleophilic substitution reaction with different amino acids to afford the products **4a–f**.



Scheme 1. Synthesis of di- and tri-substituted *s*-triazine amino acid derivatives.

No marked difference in the yield of **4a–f** between the two routes (A and B) was observed. However, the stepwise reaction decreased the probability of side product formation from the reagents remaining in the one-pot reaction. It is important to highlight that sufficient amount of NaHCO₃ was used during the synthesis to overcome/neutralize traces of cyanuric acid, which might be present in the crude cyanuric chloride.

The synthesis of **5a–q** required elevated temperatures, which were achieved in three ways, namely by conventional heating, microwave irradiation or ultrasonication. Disubstituted *s*-triazine derivatives **4a–f** (1 equiv.) and secondary amines (1.1 equiv.) were reacted in the presence of diisopropylethylamine (DIEA) as a base and THF as solvent in the three conditions mentioned above. Using conventional heating (reflux for 20–24h), **5a–q** were obtained in good yields and purities. In contrast, the use of

microwave irradiation afforded these compounds in better yields in only 12–15 min of reaction time, while the application of ultrasonic irradiation allowed their synthesis at room temperature in 1.5h with a greater yield and purity than conventional heating (Table 1). Of note, the prolonged reaction under conventional heating can be markedly reduced by means of microwave or ultrasound irradiation. The structures of all compounds **5a–q** were confirmed by NMR spectra (¹H and ¹³C) as shown in Supporting information (Figures S1–S17).

Compound	Conventional (20–24 h)	Microwave (12–15 min)	Ultrasound (1.5 h)	
5a	89	91	92	
5b	87	90	92	
5c	86	90	90	
5d	89	93	92	
5e	88	91	92	
5f	87	90	90	
5g	86	92	91	
5h	86	90	93	
5i	84	91	93	
5j	86	93	90	
5k	88	90	92	
51	89	92	94	
5m	86	92	92	
5n	89	94	93	
50	87	91	92	
5p	89	92	90	
5q	86	90	92	

Table 1. Yield (%) of the synthesis of **5a–q** derivatives via conventional heating, microwave irradiation and ultrasonic methods.

3.2. In Vitro Antimicrobial Assays

The antibacterial and antifungal activity of **5a–q** was tested. None of the tested compounds showed antibacterial activity against the two Gram-positive bacteria (*S. aureus* and *S. epidermidis*) or two Gram-negative bacteria (*E. coli* and *S. typhimurium*). However, they showed promising antifungal activity against the human pathogenic *C. albicans* compared with the reference standard drug clotrimazole (Table 2). The effective bioactivity of 4,6-disubstituted s-triazin-2-yl amino acid derivatives with fungal strain, and no activity on bacterial strains may be dependent on several features of both species like cell wall, cell membrane or metabolic activity and the different amino acid moiety used for the preparation of these derivatives [35]. In addition, the proteinaceous transfer system for these bioactive compounds, primarily in the fungal strain, is responsible for the process [36].

The growth-inhibiting effects of 5a-q against *C. albicans* were quantitatively determined by means of the disc diffusion method at concentrations of 50, 100, 200, and 300 µg per disc. The results are shown in Table 2 and Figure 1. Compounds 5d, 5e, and 5f showed the best inhibitory capacity at 50 µg per disc of 15, 13, and 14 mm, respectively. These results revealed that the presence of the piperidine is crucial for the activity. Thus, the combination of piperidine with aniline derivatives on the triazine core bearing the glycine amino acid 5d gave the best results. In addition, among the aniline derivatives, chloro appeared to be more active than methoxy. These results are consistent with the notion that compounds bearing electron-withdrawing groups, such as chlorine atoms, have a potent inhibitory effect on the examined microbe [19,37]. This observation was also noticed in all tested compounds at different concentrations, as shown in Figure 1.

Compound	C. albicans (mm)			MIC	MFC	
	50 µg	100 µg	200 µg	300 µg	(μM)	(μM)
5a	7 ± 0.1	9 ± 0.4	9 ± 0.2	10 ± 0.1	37.76 ± 0.7	75.52 ± 0.7
5b	7 ± 0.3	8 ± 0.2	10 ± 0.1	12 ± 0.3	34.14 ± 0.1	68.28 ± 0.1
5c	7 ± 0.3	8 ± 0.3	9 ± 0.3	10 ± 0.1	34.58 ± 0.8	69.16 ± 0.7
5d	15 ± 0.2	18 ± 0.4	21 ± 0.3	23 ± 0.4	37.95 ± 0.2	75.90 ± 0.9
5e	13 ± 0.1	16 ± 0.1	18 ± 0.3	20 ± 0.5	34.36 ± 0.6	68.72 ± 0.8
5f	14 ± 0.4	16 ± 0.7	17 ± 0.4	19 ± 0.1	34.74 ± 0.3	69.48 ± 0.3
5g	7 ± 0.1	8 ± 0.3	10 ± 0.3	11 ± 0.2	35.73 ± 0.7	71.46 ± 0.5
5h	7 ± 0.2	9 ± 0.1	9 ± 0.3	10 ± 0.1	36.19 ± 0.2	72.38 ± 1.0
5i	8 ± 0.3	10 ± 0.2	12 ± 0.4	12 ± 0.1	36.19 ± 0.4	72.38 ± 0.9
5j	10 ± 0.4	13 ± 0.5	14 ± 0.3	15 ± 0.6	33.29 ± 0.3	66.58 ± 0.6
5k	7 ± 0.2	8 ± 0.1	9 ± 0.3	10 ± 0.1	36.19 ± 0.2	72.38 ± 1.2
51	8 ± 0.2	10 ± 0.2	11 ± 0.3	12 ± 0.2	36.40 ± 0.3	72.8 ± 0.4
5m	7 ± 0.3	14 ± 0.3	15 ± 0.2	15 ± 0.3	33.09 ± 0.1	66.18 ± 0.9
5n	7 ± 0.2	9 ± 0.2	9 ± 0.1	10 ± 0.1	33.47 ± 0.5	66.94 ± 1.2
50	7 ± 0.1	8 ± 0.3	10 ± 0.2	10 ± 0.3	37.95 ± 0.7	75.90 ± 0.9
5p	7 ± 0.1	8 ± 0.1	11 ± 0.3	11 ± 0.1	34.35 ± 0.4	68.70 ± 0.6
5q	7 ± 0.2	8 ± 0.1	9 ± 0.4	10 ± 0.1	34.78 ± 0.6	69.56 ± 0.4
Clotrimazole **	-	-	-	23 ± 0.2	-	-

Table 2. Zone of inhibition (mm) *, Minimum Inhibitory Concentration (MIC) and Minimum Fungicidal

 Concentration (MFC) of the target compounds against *Candida albicans* growth.

* Inhibition zone (mm) at 300 μ g per disc; ** 100 μ g per disc, ± means Standard deviation (STDV) of triplicate experiments.



Figure 1. Inhibition zones of **5a–q** at concentrations of 50,100, 200 and 300 μ g per disc against *C. albicans*. Error bars mean the STDV of the average values of the triplicate experiments.

The individual minimum inhibitory concentration (MIC, μ M) and minimum fungicidal concentration (MFC, μ M) values of **5a–q** against *C. albicans* are listed in Table 2. In general, all the compounds showed good antifungal activities, with a MIC value falling between 33.47 and 37.95 μ M and MFC value between 66.58 and 75.90 μ M. Again, this assay supports the previous findings illustrated inTable 2.

3.3. Docking Studies

The observed antifungal activity of the compounds can be attributed to the inhibition of *N*-myristoltransferase (NMT) enzyme, which has been validated as an anti-fungal target [32].

NMT activity is essential for the vegetative growth of *C. albicans*. NMT is a promising target enzyme for the development of novel fungicidal drugs with a broad anti-fungal spectrum. To date, various types of NMT inhibitors have been reported, including peptidomimetics, benzothiazoles, tetrahydro- carbazoles, sulphonamides and benzofurans. Recently, we have carried out an extensive virtual screening to identify NMT inhibitors effective against kinetoplastids. Therefore, we examined the protein–ligand interaction profile of **5b**, **5d**, **5j**, **5n** and **5o** with NMT. The docking profile of compound **5d** suggests that the compound is well accommodated in the binding site of *C. albicans* NMT. The *s*-triazine moiety mediates hydrogen bonds with the side chain hydroxyl of Y225. The Y225 also mediates another hydrogen bond with ligand **5o** (Figure 2). The benzene ring of Y225 exhibits pi stacking interaction with the benzene ring of the ligands. Furthermore, the hydrogen bonding between the ligand and backbone amide of L355 and L394 provides anchorage. Moreover, the ligand also demonstrates hydrophobic contact with surrounding residues, which explains the inhibitory potential of **5o** observed.



Figure 2. The protein–ligand interaction profile of the ligands (5b, 5d, 5j, 5n, and 5o).

A slightly different binding mode was observed in the triazine moiety of **5d** mediate hydrogen bond interaction with L451 (Figure 2). Y225 exhibits pi stacking interaction with the benzene ring. The compound also demonstrates polar contact with the side chain of T211. An extensive network of hydrogen and hydrophobic contacts provides further anchorage.

The protein–ligand interaction profile of **5***j*, **5***b* and **5***n* demonstrated that the triazine and benzene moieties of the ligands are stacked against Y359 (Figure 2). The methoxy substitution shows polar contact with L394. However, the oxygen atom of the oxathiane ring mediates the hydrogen bond with N392.

Our results demonstrate that all the compounds are well accommodated and exhibit complementarity with the active site residues of *C. albicans* NMT, which explains the observed antifungal activity.

Furthermore, to explore the antifungal activity against other pathogenic fungi, the active site of *Cryptococcus neoformans, Aspergillus fumigatus* and *Histoplasma capsulatum* NMTs were aligned to *C. albicans* NMT, as known crystal structures of NMTs from different fungi indicate that all NMTs have similar substrate binding sites despite their low sequence identity. The sequence similarity is obtained from blastp between *C. albicans* NMT with *C. neoformans* NMt (46%), *A. fumigatus* NMT (49%) and *H. capsulatum* NMT (46%), and, interestingly, they share a similar substrate binding site. As depicted in Figure 3, the alignment of the active sites of NMTs from different fungi reveals that all the active site residues in *C. albicans* were well aligned with *C. neoformans*, *A. fumigatus* and *H. capsulatum*. However,

Tyr256 in *C. albicans* replaced Asn249 and Asn331 in *A. fumigatus* and *H. capsulatum*, respectively, while Phe339 in *C. albicans* replaced Ser378, Ser351 and Ser415 in *A. fumigatus*, *C. neoformans* and *H. capsulatum*, respectively. This high degree of similarity between binding sites may be attributed to the broad spectrum antifungal activities, and these compounds could be active against other pathogenic fungi species.



Figure 3. Alignment of active site of *C. albicans N*-myristoltransferase (NMT) (firebrick color) with *C. neoformans* NMT (grey), *Aspergillus fumigatus* (spring green color) NMT and *Histoplasma capsulatum* NMT (salmon color).

4. Conclusions

Novel 4,6-disubstituted s-triazin-2-yl amino acid derivatives were prepared via conventional heating, microwave irradiation and ultrasonication. The two latter techniques gave **5a–q** in good purity and better yield within shorter time intervals compared to the conventional heating method. The synthesized derivatives were not found to be active against two Gram-positive bacteria (*S. aureus ATCC* 29213, *S. epidermidis ATCC* 12228) and two Gram-negative bacteria (*E. coli ATCC* 25922 and *S. typhimurium ATCC* 14028). However, they showed promising antifungal activities against the human pathogen *C. albicans* (*ATCC* 60193).

The key to the antifungal activity is the presence of the piperidine ring. Thus, the compounds containing piperidine were the most active, which agreed with the previously reported data [29]. Interestingly, morpholine (six-membered ring containing one O instead of C as in the case of piperidine) and pyrrolidine (five-membered ring) showed less activity. This supports the need for a degree of hydrophobicity in this position.

On the other hand, aniline without any substituent proved to be more promising than chloro (electron-withdrawing) and methoxy (electron-donating) analogs. In addition, the synthesized compounds were found to be well accommodated in the active site residues of *C. albicans* NMT, which explains the observed antifungal activities.

The known crystal structures of NMTs from different pathogenic fungi, namely, *C. neoformans*, *A.* fumigatus and *H. capsulatum*, indicate that all NMTs share similar substrate binding sites. This high degree of similarity between binding sites may be attributed to the broad-spectrum antifungal activities, and these compounds could be active against other pathogenic fungi species.

Finally, we can summarize the structure–activity relationship of our target compounds as indicated in Figure 4.



Figure 4. The structure–activity relationship.

To gain a complete picture of the effect of structural modification or functionalization on antifungal activities, further development for a new series based on *s*-triazine amino acid derivatives with different substituents will be considered in our laboratories.

Supplementary Materials: The following are available online at http://www.mdpi.com/2309-608X/6/4/237/s1, Figures S1–S17: ¹H-NMR and ¹³C-NMR for compound **5a–q**.

Author Contributions: The work was designed and supervised by A.E.-F., Z.A., B.G.d.I.T., and F.A.; the synthesis and characterization of the reported compounds were carried out by R.A.A. and Z.A.; E.N.S. and A.S. performed the biological activity assay; S.A.K. and Z.U.-H. carried out the docking study; R.A.A., Z.A., E.N.S., and S.A.K. prepared the first drafts of the manuscript. The final version includes contributions from all authors. All authors have read and agreed to the published version of the manuscript.

Funding: Deanship of Scientific Research at King Saud University, Saudi Arabia, research group no. (RGP-1441-234). National Research Foundation (NRF) (# 105892 and Blue Sky's Research Programme # 120386).

Acknowledgments: The authors extend their thanks to the Deanship of Scientific Research at King Saud University for funding this work through research group no. (RGP -1441-234, Saudi Arabia). The authors express their deep gratitude to one of the referees for his/her more-than-valuable comments.

Conflicts of Interest: The authors declare no conflict of interest.

References

- 1. Blotny, G. Recent applications of 2,4,6-trichloro-1,3,5-triazine and its derivatives in organic synthesis. *Tetrahedron* **2006**, *62*, 9507–9522. [CrossRef]
- 2. Desai, N.C.; Makwana, A.H.; Rajpara, K.M. Synthesis and study of 1,3,5-triazine based thiazole derivatives as antimicrobial agents. *J. Saudi Chem. Soc.* **2016**, *20*, S334–S341. [CrossRef]
- 3. Srinivas, K.; Srinivas, U.; Bhanuprakash, K.; Harakishore, K.; Murthy, U.S.; Rao, V.J. Synthesis and antibacterial activity of various substituted s-triazines. *Eur. J. Med. Chem.* **2006**, *41*, 1240–1246. [CrossRef] [PubMed]
- 4. Ryan, K.J.; Ray, C.G. *Sherris Medical Microbiology*; McGraw Hill: New York, NY, USA, 2004; Volume 4, ISBN 0-8385-8529-9.
- 5. Kullberg, B.J.; Oude Lashof, A.M. Epidemiology of opportunistic invasive mycoses. *Eur. J. Med. Res.* **2002**, *7*, 183–191. [PubMed]
- Weig, M.; Groß, U.; Mühlschlegel, F. Clinical aspects and pathogenesis of Candida infection. *Trends Microbiol.* 1998, 6, 468–470. [CrossRef]
- Zadik, Y.; Burnstein, S.; Derazne, E.; Sandler, V.; Ianculovici, C.; Halperin, T. Colonization of Candida: Prevalence among tongue-pierced and non-pierced immunocompetent adults. *Oral Dis.* 2010, *16*, 172–175. [CrossRef] [PubMed]
- Achkar, J.M.; Fries, B.C. Candida infections of the genitourinary tract. *Clin. Microbiol. Rev.* 2010, 23, 253–273. [CrossRef]

- 9. Kelly, M.T.; MacCallum, D.M.; Clancy, S.D.; Odds, F.C.; Brown, A.J.P.; Butler, G. The Candida albicans CaACE2 gene affects morphogenesis, adherence and virulence. *Mol. Microbiol.* **2004**, *53*, 969–983. [CrossRef]
- Kumamoto, C.A. Inflammation and gastrointestinal Candida colonization. *Curr. Opin. Microbiol.* 2011, 14, 386–391.
 [CrossRef]
- 11. Melato, S.; Prosperi, D.; Coghi, P.; Basilico, N.; Monti, D. A Combinatorial Approach to 2,4,6-trisubstituted triazines with potent antimalarial activity: Combining conventional synthesis and microwave-assistance. *Chem. Med. Chem.* **2008**, *3*, 873–876. [CrossRef]
- 12. Xiong, Y.-Z.; Chen, F.-E.; Balzarini, J.; De Clercq, E.; Pannecouque, C. Non-nucleoside HIV-1 reverse transcriptase inhibitors. Part 11: Structural modulations of diaryltriazines with potent anti-HIV activity. *Eur. J. Med. Chem.* **2008**, *43*, 1230–1236. [CrossRef] [PubMed]
- El-Faham, A.; Farooq, M.; Almarhoon, Z.; Alhameed, R.A.; Wadaan, M.A.M.; de la Torre, B.G.; Albericio, F. Di- and tri-substituted s-triazine derivatives: Synthesis, characterization, anticancer activity in human breast-cancer cell lines, and developmental toxicity in zebrafish embryos. *Bioorg. Chem.* 2020, 94, 103397. [CrossRef] [PubMed]
- Al-Rasheed, H.H.A.; Malebari, A.M.M.; Dahlous, K.A.A.; El-Faham, A. Synthesis and Characterization of New Series of 1,3-5-Triazine Hydrazone Derivatives with Promising Antiproliferative Activity. *Molecules* 2020, 25, 2708. [CrossRef]
- 15. Kumar, A.; Srivastava, K.; Kumar, S.R.; Puri, S.K.; Chauhan, P.M.S. Synthesis of 9-anilinoacridine triazines as new class of hybrid antimalarial agents. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 6996–6999. [CrossRef] [PubMed]
- Bhat, H.R.; Singh, U.P.; Gahtori, P.; Ghosh, S.K.; Gogoi, K.; Prakash, A.; Singh, R.K. 4-Aminoquinoline -1,3,5-triazine: Design, synthesis, in vitro antimalarial activity and docking studies. *New J. Chem.* 2013, 37, 2654–2662. [CrossRef]
- 17. Pathak, P.; Thakur, A.; Bhat, H.R.; Singh, U.P. Hybrid 4-Aminoquinoline-1,3,5-triazine Derivatives: Design, Synthesis, Characterization, and Antibacterial Evaluation. *J. Heterocycl. Chem.* **2015**, *52*, 1108–1113. [CrossRef]
- Haibaa, N.S.; Khalil, H.H.; Abdel Moniem, M.; El-Wakil, M.H.; Bekhit, A.A.; Khattab, S.N. Design, synthesis and molecular modeling studies of new series of *s*-triazine derivatives as antimicrobial agents against multi-drug resistant clinicalisolates. *Bioorg. Chem.* 2019, *89*, 103013. [CrossRef]
- Ramadan, D.R.; Elbardan, A.A.; Bekhit, A.A.; El-Faham, A.; Khattab, S.N. Synthesis and characterization of novel dimeric s-triazine derivatives as potential anti-bacterial agents against MDR clinical isolates. *New J. Chem.* 2018, 42, 10676–10688. [CrossRef]
- 20. Sarmah, K.; Sarmah, N.; Kurmi, K.; Patel, T. Synthesis and studies of antifungal activity of 2,4,6-trisubstituted 1,3,5-triazines. *Adv. Appl. Sci. Res.* **2012**, *3*, 1459–1462.
- 21. Saeed, S.; Rashid, N.; Jones, P.G.; Yunas, U. 2-substituted 4H-[1,3]thiazolo[3,2-a][1,3,5]triazine-4-thiones: Synthesis, crystal structure, and antifungal activity. *J. Heterocycl. Chem.* **2010**, *47*, 908–912. [CrossRef]
- 22. Romani, L. Immunity to fungal infections. Nat. Rev. Immunol. 2011, 11, 275-288. [CrossRef] [PubMed]
- 23. Khattab, S.N.; Khalil, H.H.; Bekhit, A.A.; Abd El-Rahman, M.M.; de la Torre, B.G.; El-Faham, A.; Albericio, F. 1,3,5-triazino peptide derivatives: Synthesis, characterization, and preliminary antileishmanial activity. *ChemMedChem* **2018**, *13*, 725–735. [CrossRef]
- 24. Mikulová, M.B.; Kružlicová, D.; Pecher, D.; Supuran, C.T.; Mikuš, P. Synthetic Strategies and Computational Inhibition Activity Study for Triazinyl-Substituted Benzenesulfonamide Conjugates with Polar and Hydrophobic Amino Acids as Inhibitors of Carbonic Anhydrases. *Int. J. Mol. Sci.* **2020**, *21*, 3661. [CrossRef] [PubMed]
- Mikuš, P.; Krajčiová, D.; Mikulová, M.; Horváth, B.; Pecher, D.; Garaj, V.; Bua, S.; Angeli, A.; Supuran, C.T. Novel sulfonamides incorporating 1,3,5-triazine and amino acid structural motifs as inhibitors of the physiological carbonic anhydrase isozymes I, II and IV and tumor-associated isozyme IX. *Bioorg. Chem.* 2018, *81*, 241–252. [CrossRef] [PubMed]
- 26. Khattab, S.N.; Khalil, H.H.; Bekhit, A.A.; El-Rahman, M.M.A.; El-Faham, A.; Albericio, F. Synthesis and preliminary biological evaluation of 1,3,5-triazine amino acid derivatives to study their MAO inhibitors. *Molecules* **2015**, *20*, 15976–15988. [CrossRef] [PubMed]
- 27. Sharma, A.; El-Faham, A.; de la Torre, B.G.; Albericio, F. Exploring the Orthogonal Chemoselectivity of 2,4,6-Trichloro-1,3,5-Triazine (TCT) as a Trifunctional Linker With Different Nucleophiles: Rules of the Game. *Front. Chem.* **2018**, *6*, 516. [CrossRef]

- 28. Calvete, M.J.F.; Pinto, S.M.A.; Burrows, H.D.; Castro, M.M.C.A.; Geraldes, C.F.G.C.; Pereira, M.M. Multifunctionalization of cyanuric chloride for the stepwise synthesis of potential multimodal imaging chemical entities. *Arab. J. Chem.* **2020**, *13*, 2517–2525. [CrossRef]
- 29. Desai, N.C.; Makwana, A.H.; Senta, R.D. Synthesis, characterization and antimicrobial activity of some novel 4-(4-(arylamino)-6-(piperidin-1-yl)-1,3,5-triazine-2-ylamino)-*N*-(pyrimidin-2-yl)benzenesulfon amides. *J. Saudi Chem. Soc.* **2016**, *20*, 686–694. [CrossRef]
- Al-Marhoon, Z.; Abdel-Megeed, A.; Sholkamy, E.N.; Siddiqui, M.R.H.; El-Faham, A. Synthesis of Phenylcarbamic Acid and 2-[2-Oxo-3-(4-substituted phenylimino)-indolin-1-yl] acetohydrazide Derivatives as Promising Antifungal Agents. *Asian J. Chem.* 2014, *26*, 7665–7672. [CrossRef]
- 31. Wayne, P.A. *Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria That Grow Aerobically*, 11th ed.; Clinical and Laboratory Standards Institute: Wayne, PA, USA, 2018.
- 32. Rasadah, M.A.; Muharnad, Z. Prosid. In *PerubatanTraditional Malaysia Ke-5*; Universiti Malaya: Kuala Lumpur, Malaysia, 1988; p. 173.
- Toraskar, M.; Prasad, K.; Kadam, V. N-myristoyltransferase: A novel target. *Mini. Rev. Med. Chem.* 2008, *8*, 142–149.
 [CrossRef]
- 34. *Molecular Operating Environment (MOE), Version 2018.0101;* Chemical Computing Group Inc.: Montreal, QC, Canada, 2016.
- 35. Scheffers, D.-J.; Pinho, M.G. Bacterial Cell Wall Synthesis: New Insights from Localization Studies. *Microbiol. Mol. Biol. Rev.* 2005, *69*, 585–607. [CrossRef] [PubMed]
- 36. Dare, K.; Ibba, M. Roles of tRNA in cell wall biosynthesis. *Wiley Interdiscip. Rev. RNA* 2012, *3*, 247–264. [CrossRef] [PubMed]
- 37. Mewada, N.S.; Shah, D.R.; Lakum, H.P.; Chikhalia, K.H. Synthesis and biological evaluation of novel s-triazine based aryl/heteroaryl entities: Design, rationale and comparative study. *J. Assoc. Arab Univ. Basic Appl. Sci.* **2016**, *20*, 8–18. [CrossRef]

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



© 2020 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).