

Article



Development of Heavy Metal-Free Photocatalytic RhB Decomposition System Using a Biodegradable Plastic Substrate

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Substrate+photocatalyst	Environmental friendliness	Ref.
Stainless steel +TiO ₂	×	[18]
Pebbles + TiO ₂	Δ	[16]
Galass+TiO ₂	Δ	[14]
Plastic+Ag	×	[19]
PLA+C ₃ N ₄	0	This work

Table S1. Comparison of photocatalyst and substrate environmental friendliness.

Table S2. Experimental conditions for degradation RhB by batch method under visible light irradiation.

Sample	Rhodamine B (5 ppm, 42 ml)		
Temperature	Room temperature (25 $^{\circ}$ C)		
Photocatalyst	M-500 or NT-500 (30 mg)		
Light source	Xenon lamp with cut filter		
	(6000-7000 μ W/cm ² , $\lambda \ge$ 420 nm)		
Irradiation time	0-60min		
Analysis	UV-visible spectrometer (554 nm)		

Table S3. BET areas, total pore volume and average pore diameter of M-500 and NT-500.

Photocatalyst	Sbet (cm²/g)	$V_{p}(P/P_{0}=0.99, cm^{3}/g)$	d _p (nm)
M-500	14.1	0.11	30.9
NT-500	52.4	0.20	15.6

Photocatalyst	Irradiation light source	Target material	Flow rate (mL/min)	Degradation re- sult (%)	Ref.
BiOI/porous	Xenon lamp	Methylene blue	1	28.1	[46]
g-C3N4/graphene	$\lambda \ge 420 \text{ nm}$	$1 \times 10^{-5} \text{ mol/L}$	1		
TiO ₂ /ACF	UV-C lamp	Methylene blue	10	75	[47]
	λ = 254 nm	10 mg/L			
WO3/graphene -	high-pressure mer-	RhB	5	66.7	[48]
oxide/TiO ₂	cury lamp	$1.2 \times 10^{-4} \text{ mol/L}$			
TiO2/ Glass plate	UV-A lamps	Carbamazepine	2.7	87	[49]
	$\lambda \leq 365 \text{ nm}$	50 µg/L			
C ₃ N ₄ /PLAring	Xenon lamp	RhB	0.35	50	This
	$\lambda \ge 420 \text{ nm}$	5 mg/L			work

Table S4. Comparison of photocatalytic performance under continuous flow conditions.



Figure S1. Photoluminescence spectra of M-500 and NT-500.



Figure S2. SEM images of (a) M-500 and (b) NT-500.



Figure S3. Photocatalytic RhB degradation of the C₃N₄ (M-500 and NT-500).