



Article

The Effect of Cu and Ga Doped ZnIn₂S₄ under Visible Light on the High Generation of H₂ Production

Ikki Tateishi ^{1,*}, Mai Furukawa ², Hideyuki Katsumata ² and Satoshi Kaneko ^{1,2}¹ Global Environment Center for Education & Research, Mie University, Mie 514-8507, Japan² Department of Chemistry for Materials, Graduate School of Engineering, Mie University, Mie 514-8507, Japan; maif@chem.mie-u.ac.jp (M.F.); hidek@chem.mie-u.ac.jp (H.K.); kaneco@chem.mie-u.ac.jp (S.K.)

* Correspondence: tateishi@gecer.mie-u.ac.jp; Tel.: +81-59-231-9647

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Table S1. Starting material for the preparation of photocatalyst.

Catalyst	Starting Material (mmol)				
	ZnSO ₄ ·7H ₂ O	GaCl ₃	CuCl	InCl ₃ 4H ₂ O	TAA
ZnIn ₂ S ₄	1	0	0	2	8
Zn _{0.87} Cu _{0.13} In ₂ S _{3.935}	0.87	0	0.13	2	8
Zn _{0.87} Ga _{0.13} In ₂ S _{4.065}	0.87	0.13	0	2	8
Zn _{0.86} (CuGa) _{0.07} In ₂ S ₄	0.86	0.07	0.07	2	8
Zn _{0.80} (CuGa) _{0.10} In ₂ S ₄	0.80	0.1	0.1	2	8
Zn _{0.74} (CuGa) _{0.13} In ₂ S ₄	0.74	0.13	0.13	2	8
Zn _{0.68} (CuGa) _{0.16} In ₂ S ₄	0.68	0.16	0.16	2	8
Zn _{0.62} (CuGa) _{0.19} In ₂ S ₄	0.62	0.19	0.19	2	8

Table S2. Expected photocatalyst components (molar ratio).

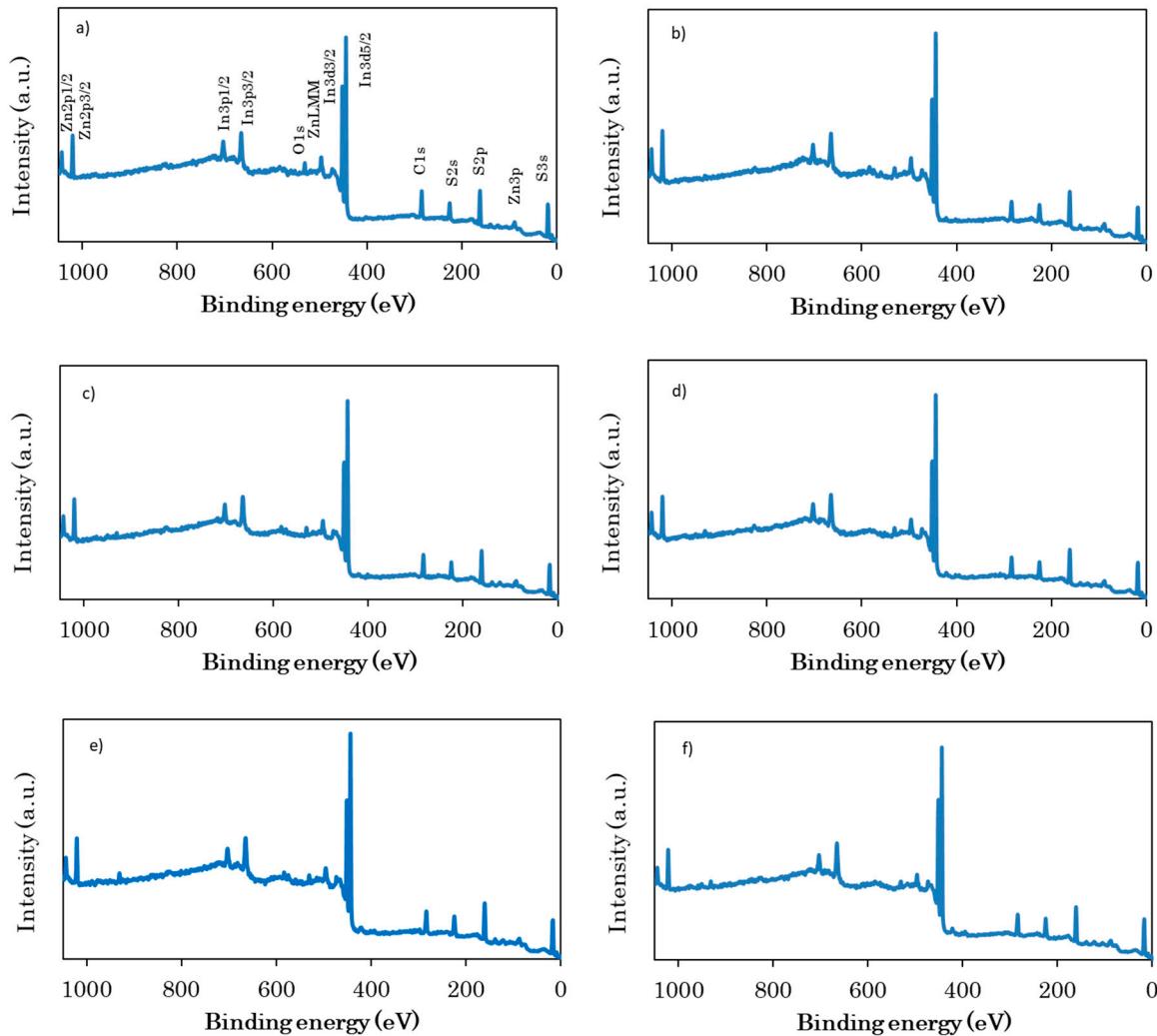
	ZnIn ₂ S ₄	In ₂ S ₃	CuInS ₂	Ga ₂ S ₃
ZnIn ₂ S ₄	1	0	0	0
Zn _{0.87} Cu _{0.13} In ₂ S _{3.935}	0.87	0.065	0.130	0
Zn _{0.87} Ga _{0.13} In ₂ S _{4.065}	0.87	0.130	0	0.065
Zn _{0.86} (CuGa) _{0.07} In ₂ S ₄	0.86	0.105	0.070	0.035
Zn _{0.80} (CuGa) _{0.10} In ₂ S ₄	0.80	0.150	0.100	0.050
Zn _{0.74} (CuGa) _{0.13} In ₂ S ₄	0.74	0.195	0.130	0.065
Zn _{0.68} (CuGa) _{0.16} In ₂ S ₄	0.68	0.240	0.160	0.080
Zn _{0.62} (CuGa) _{0.19} In ₂ S ₄	0.62	0.285	0.190	0.095

Table S3. Elemental ratios of Zn_(1-2x)(CuGa)_xIn₂S₄ from XPS result.

Catalyst	Elemental Component (Mole Ratio)				
	Zn	Ga	Cu	In	S
ZnIn ₂ S ₄	1	-	-	2.25	3.99
Zn _{0.86} (CuGa) _{0.07} In ₂ S ₄	0.86	0.06	0.06	2.09	4.11
Zn _{0.80} (CuGa) _{0.10} In ₂ S ₄	0.74	0.09	0.09	1.91	3.77
Zn _{0.74} (CuGa) _{0.13} In ₂ S ₄	0.74	0.11	0.12	2.31	3.88
Zn _{0.68} (CuGa) _{0.16} In ₂ S ₄	0.68	0.15	0.14	2.09	3.61
Zn _{0.62} (CuGa) _{0.19} In ₂ S ₄	0.62	0.19	0.19	1.83	3.70

Table S4. Specific surface area of photocatalysts.

Photocatalyst	Specific Surface Area [$\text{m}^2 \cdot \text{g}^{-1}$]
ZnIn_2S_4	5.5
$\text{Zn}_{0.74}(\text{CuGa})_{0.13}\text{In}_2\text{S}_4$	32.7
$\text{Zn}_{0.62}(\text{CuGa})_{0.19}\text{In}_2\text{S}_4$	18.6

**Figure S1.** XPS survey spectra of $\text{Zn}_{(1-2x)}(\text{CuGa})_x\text{In}_2\text{S}_4$. (a) $x = 0$, (b) $x = 0.07$, (c) $x = 0.10$, (d) $x = 0.13$, (e) $x = 0.16$ and (f) $x = 0.19$.

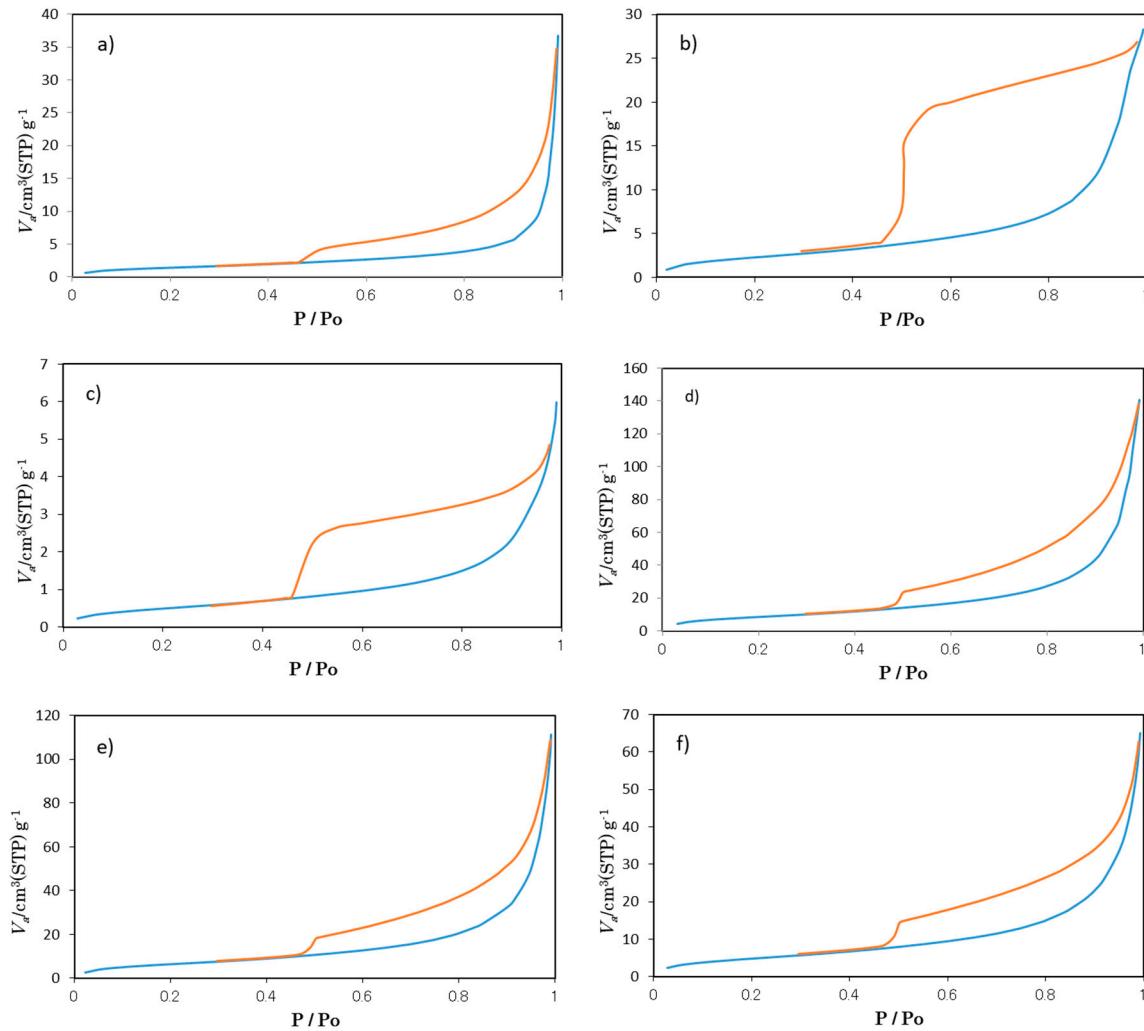


Figure S2. Adsorption/desorption isotherm of $\text{Zn}_{(1-2x)}(\text{CuGa})_x\text{In}_2\text{S}_4$. **(a)** $x = 0$, **(b)** $x = 0.07$, **(c)** $x = 0.10$, **(d)** $x = 0.13$, **(e)** $x = 0.16$ and **(f)** $x = 0.19$. Blue line: adsorption, orange line: desorption.

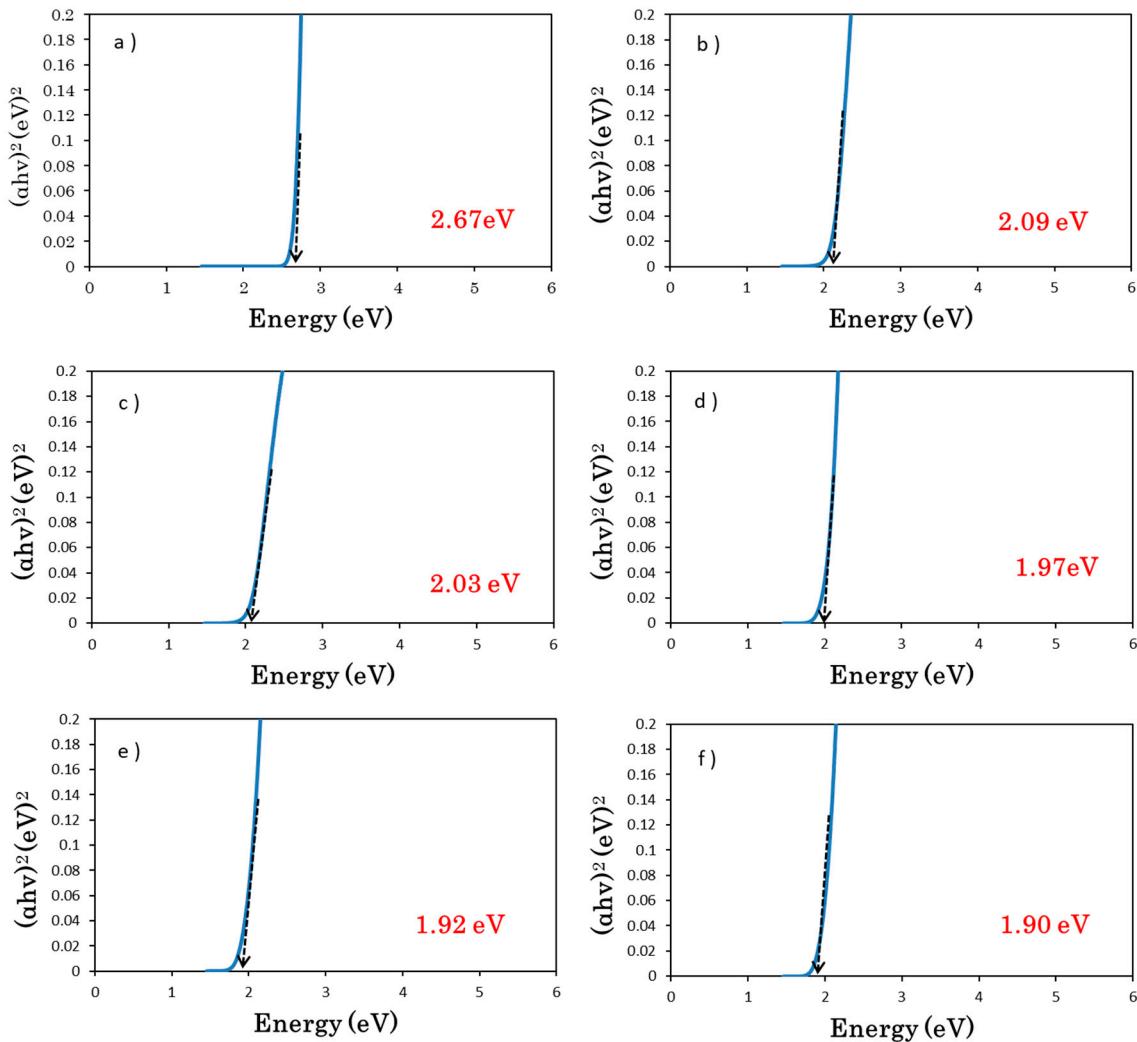


Figure S3. Tauc plots of $\text{Zn}_{(1-2x)}(\text{CuGa})_x\text{In}_2\text{S}_4$. (a) $x = 0$, (b) $x = 0.07$, (c) $x = 0.10$, (d) $x = 0.13$, (e) $x = 0.16$ and (f) $x = 0.19$.

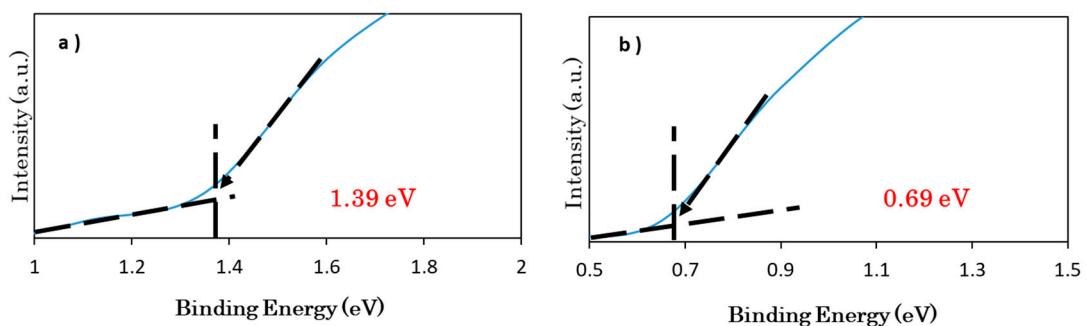


Figure S4. Valence-band edge XPS spectra of (a) ZnIn_2S_4 and (b) $\text{Zn}_{0.74}(\text{CuGa})_{0.13}\text{In}_2\text{S}_4$.

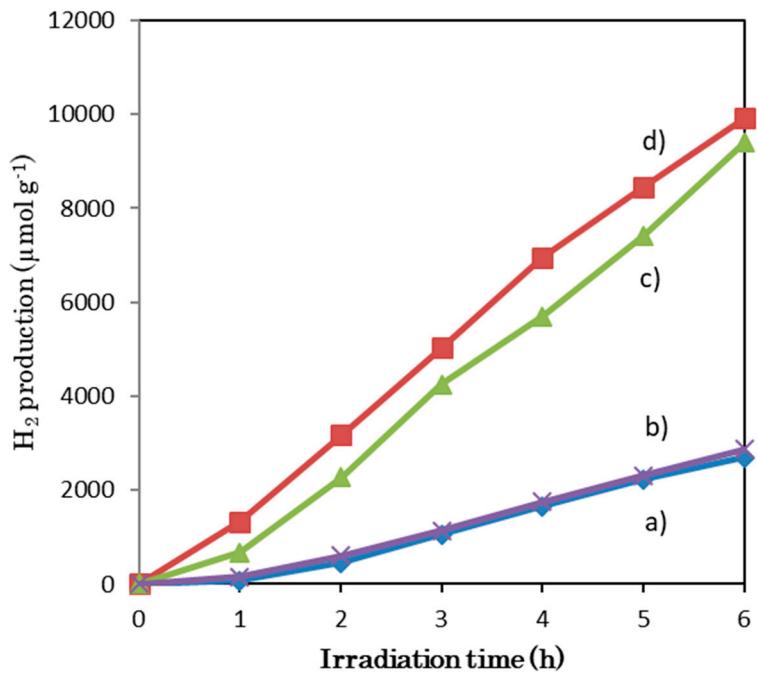


Figure S5. Photocatalytic hydrogen production with (a) ZnIn₂S₄ (rhombus), (b) Zn_{0.87}Ga_{0.13}In₂S_{4.065} (cross), (c) Zn_{0.87}Cu_{0.13}In₂S_{3.935} (triangle) and (d) Zn_{0.74}(CuGa)_{0.13}In₂S₄ (square). Pt doping: 1.0 wt%.

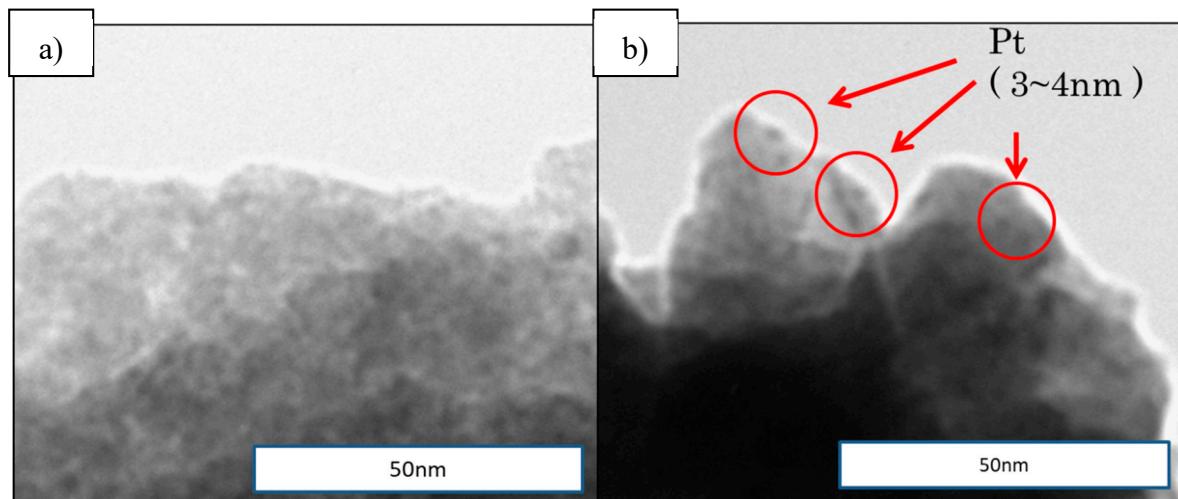


Figure S6. TEM images of Zn_{0.74}(CuGa)_{0.13}In₂S₄. (a) before and (b) after irradiation.