

Supplementary Materials:

Coordination Properties of the Fungal Metabolite Harzianic Acid toward Toxic Heavy Metals

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Figure S1. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of Cd(ClO₄)₂ in MeOH/H₂O 50:50 (*w/w*).

Figure S2. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of CoCl₂ in MeOH/H₂O 50:50 (*w/w*).

Figure S3. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of NiCl₂ in MeOH/H₂O 50:50 (*w/w*).

Figure S4. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of Pb(ClO₄)₂ in MeOH/H₂O 50:50 (*w/w*).

Figure S5. ¹H NMR spectra of harzianic acid (HA) (down), Cd(ClO₄)₂ : HA (middle) and Pb(ClO₄)₂ : HA (up) recorded in CD₃OD/D₂O (50:50 *w/w*) at 400 MHz. Solvent peaks are removed.

Figure S6. Solid-state FT-IR spectrum of harzianic acid.

Figure S7. Solid-state FT-IR spectrum of Cd (II)/harzianic acid complex.

Figure S8. Solid-state FT-IR spectrum of Co (II)/harzianic acid complex.

Figure S9. Solid-state FT-IR spectrum of Ni (II)/harzianic acid complex.

Figure S10. Solid-state FT-IR spectrum of Pb (II)/harzianic acid complex.

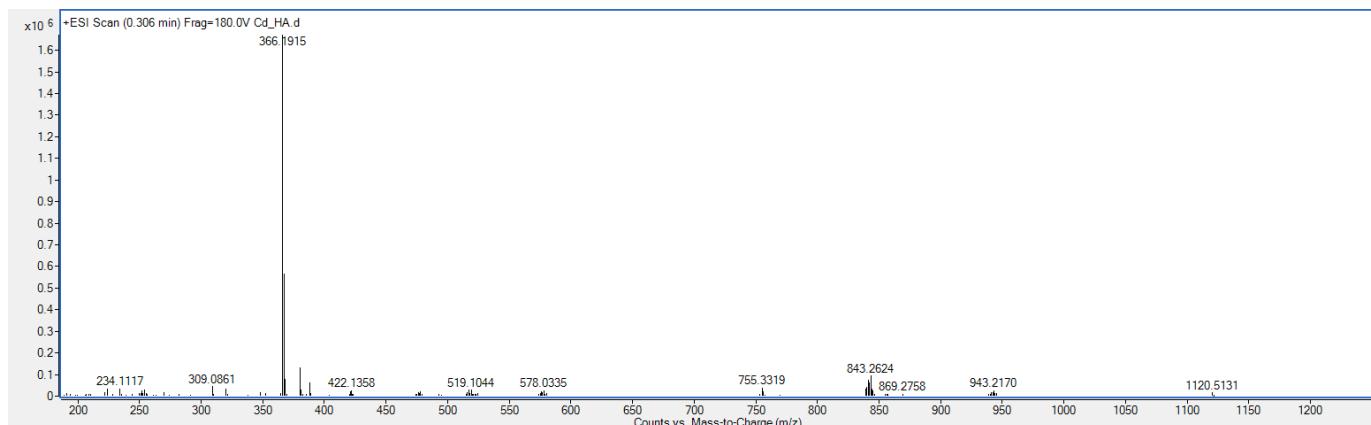


Figure S1. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of $\text{Cd}(\text{ClO}_4)_2$ in $\text{MeOH}/\text{H}_2\text{O}$ 50:50 (*w/w*).

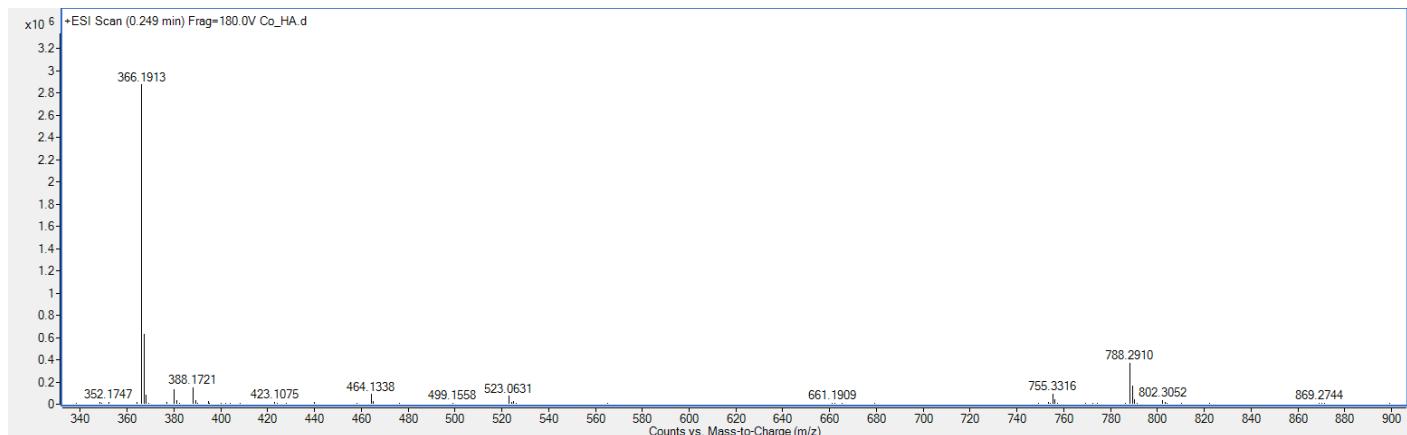


Figure S2. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of CoCl_2 in $\text{MeOH}/\text{H}_2\text{O}$ 50:50 (*w/w*).

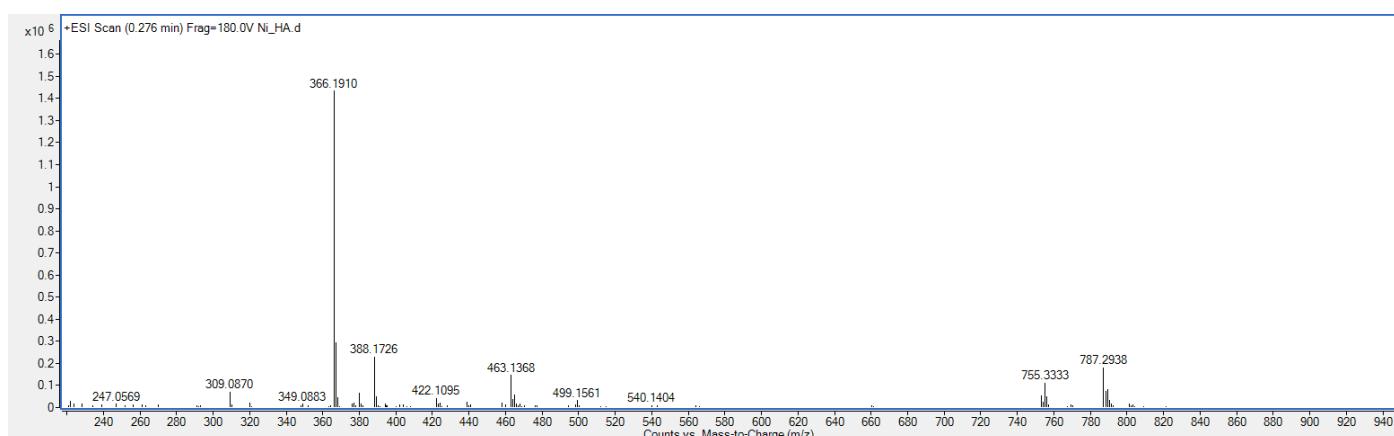


Figure S3. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of NiCl_2 in $\text{MeOH}/\text{H}_2\text{O}$ 50:50 (*w/w*).

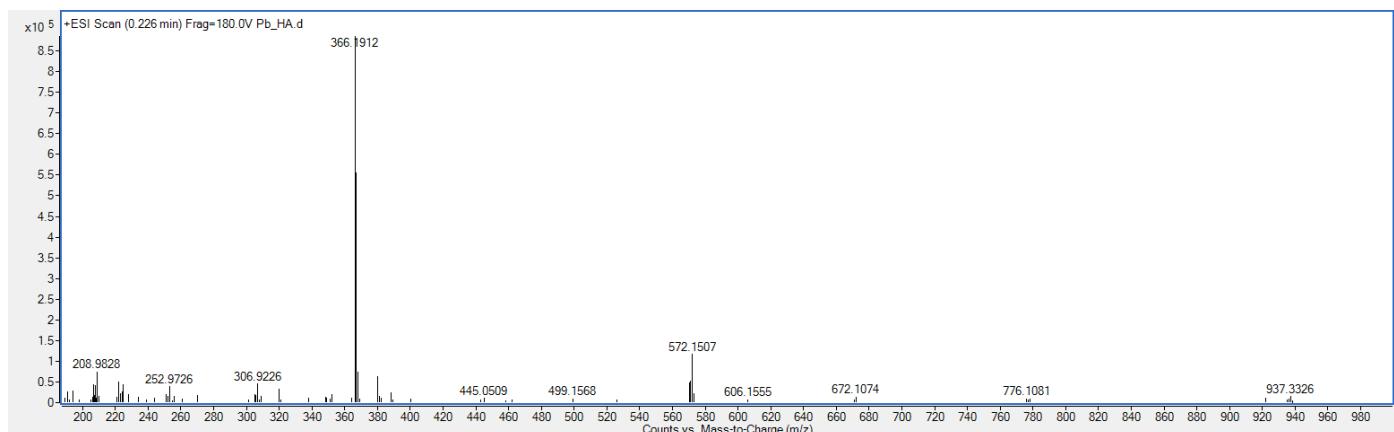


Figure S4. Mass spectrum ESIMS QTOF obtained by LC-MS analysis in ESI positive mode of a solution 1.0×10^{-3} M of harzianic acid and 1.0×10^{-3} M of $\text{Pb}(\text{ClO}_4)_2$ in $\text{MeOH}/\text{H}_2\text{O}$ 50:50 (*w/w*).

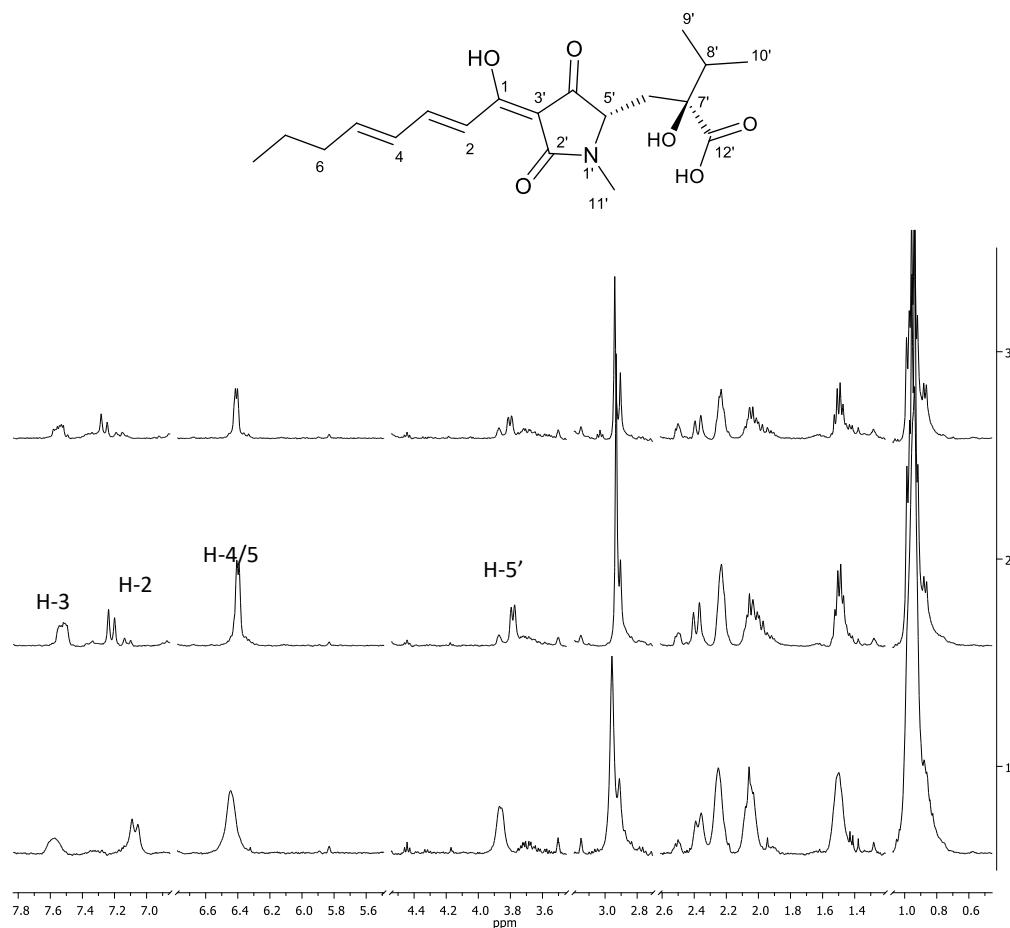


Figure S5. ^1H NMR spectra of harzianic acid (HA) (down), $\text{Cd}(\text{ClO}_4)_2 : \text{HA}$ (middle) and $\text{Pb}(\text{ClO}_4)_2 : \text{HA}$ (up) recorded in $\text{CD}_3\text{OD}/\text{D}_2\text{O}$ (50:50 *w/w*) at 400 MHz. Solvent peaks are removed.

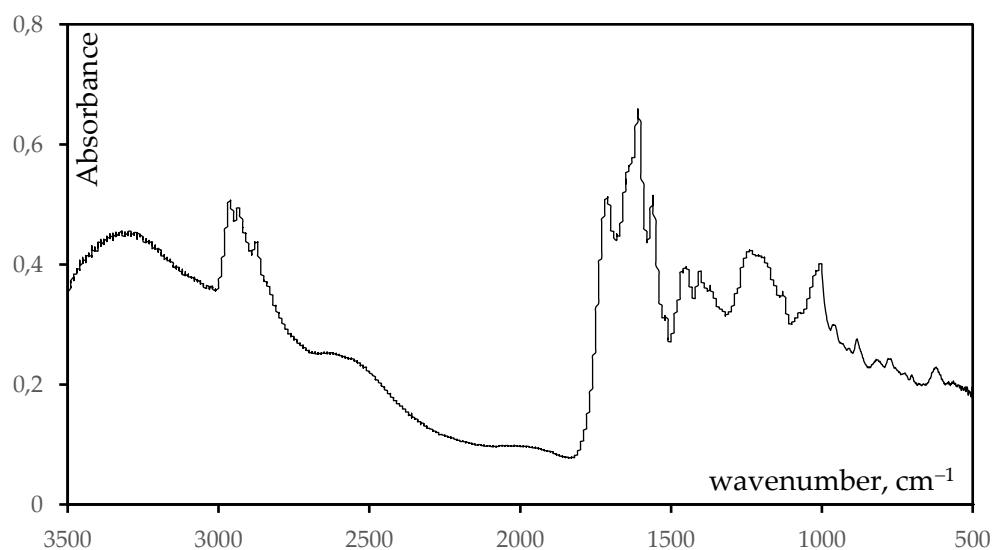


Figure S6. Solid-state FT-IR spectrum of harzianic acid.

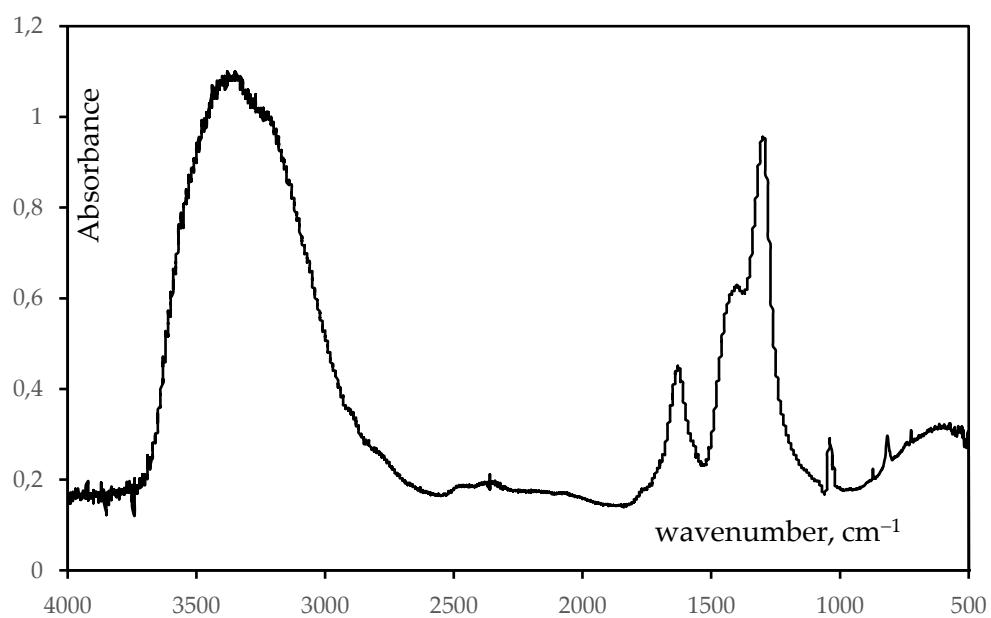


Figure S7. Solid-state FT-IR spectrum of Cd (II)/harzianic acid complex.

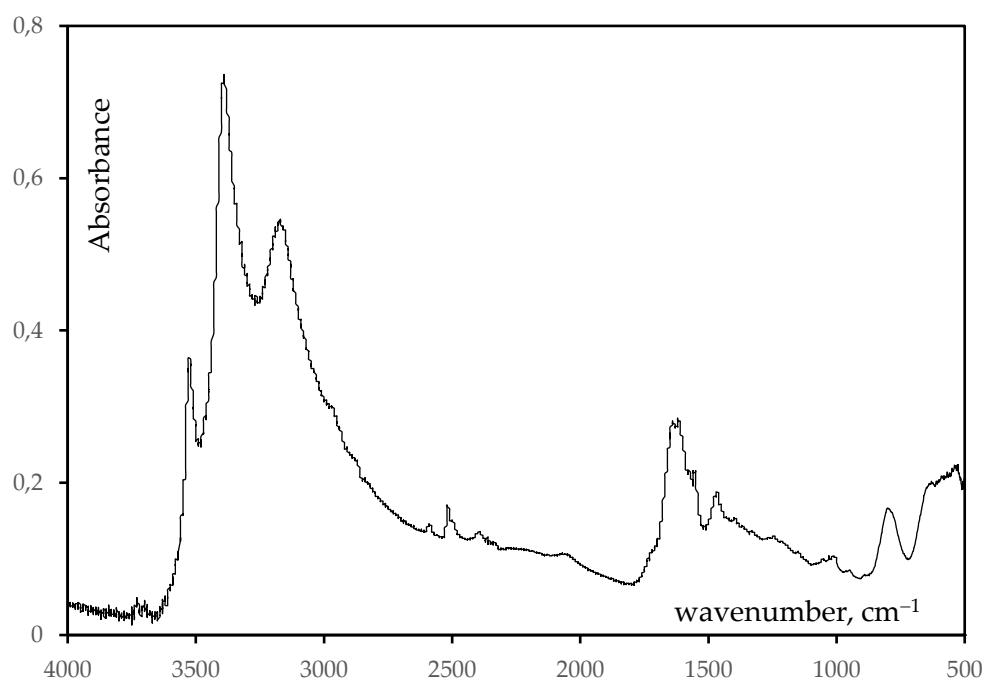


Figure S8. Solid-state FT-IR spectrum of Co (II)/harzianic acid complex.

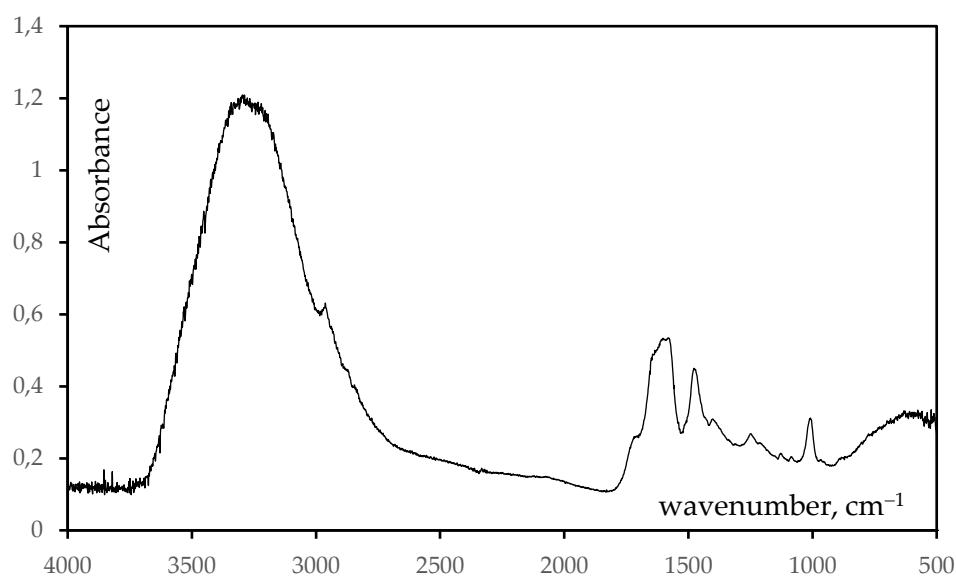


Figure S9. Solid-state FT-IR spectrum of Ni (II)/harzianic acid complex.

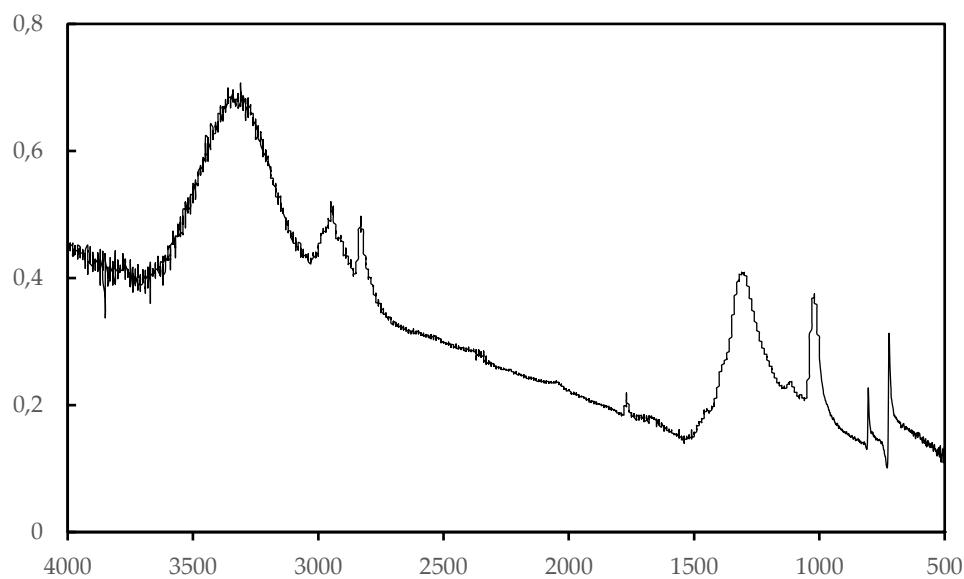


Figure S10. FT-IR spectrum of Pb (II)/harzianic acid complex