

## Supporting information

# Solid state phosphorescence enhancement of Pt<sup>II</sup>-based emitters via combination of $\pi$ -Hole(Isocyano Group)··· d<sub>z</sub><sup>2</sup>[Pt<sup>II</sup>] and I···Cl Halogen-Bonding Interactions

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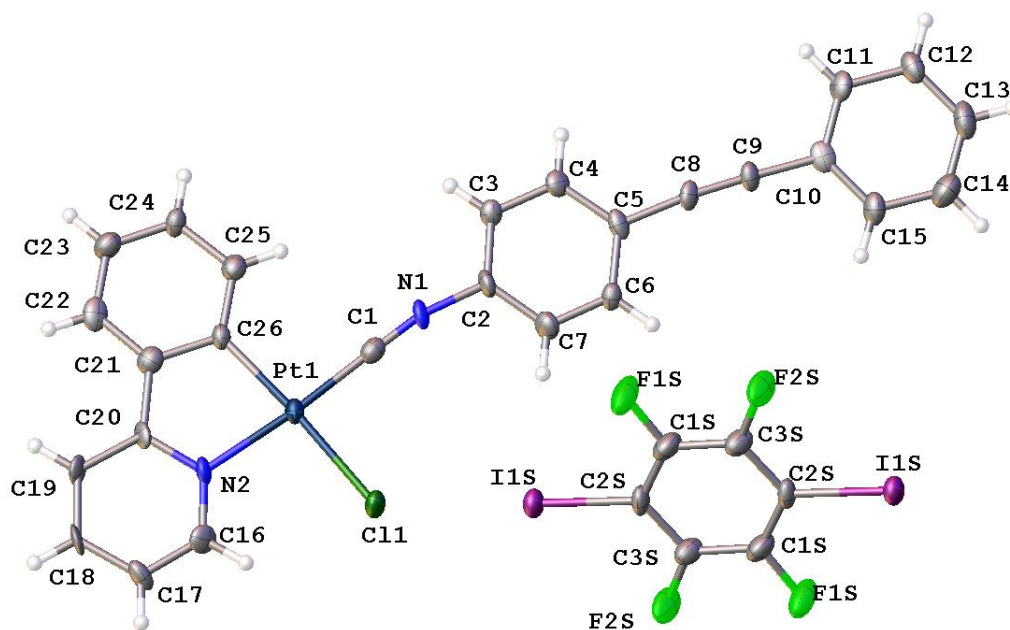
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## S1. X-ray diffraction studies

**Table S1.1** Crystal data and structure refinement for 1·½(1,4-DITFB) adduct.

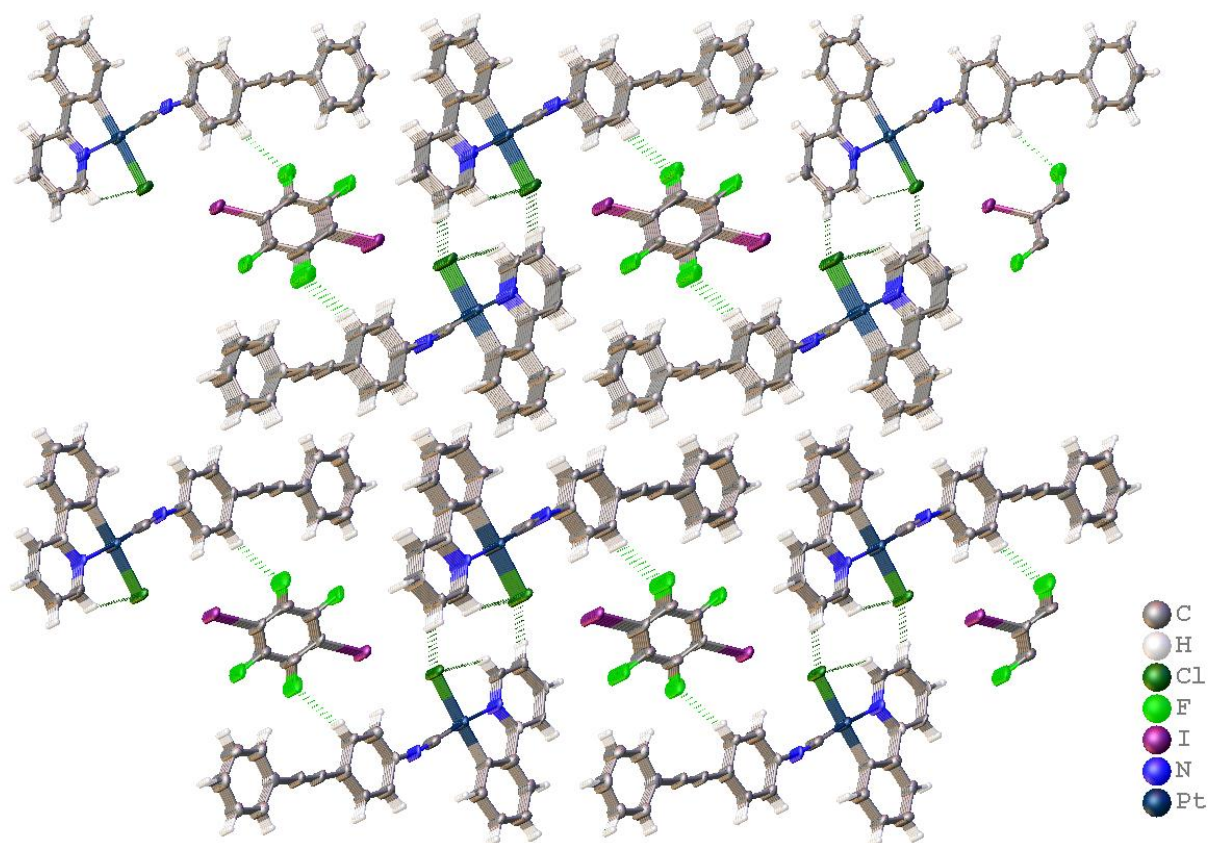
Identification code	1·½(1,4-DITFB)
CCDC code	2290328
Empirical formula	C <sub>29</sub> H <sub>17</sub> ClF <sub>2</sub> IN <sub>2</sub> Pt
Formula weight	788.89
Temperature/K	100.02(10)
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c
a/Å	15.9655(3)
b/Å	4.59870(10)
c/Å	34.1243(7)
α/°	90
β/°	91.549(2)
γ/°	90
Volume/Å <sup>3</sup>	2504.51(9)
Z	4
ρ <sub>calc</sub> /cm <sup>3</sup>	2.092
μ/mm <sup>-1</sup>	21.449
F(000)	1484.0
Crystal size/mm <sup>3</sup>	0.1 × 0.1 × 0.1
Radiation	CuKα (λ = 1.54184)
2Θ range for data collection/°	5.182 to 130
Index ranges	-17 ≤ h ≤ 18, -5 ≤ k ≤ 5, -37 ≤ l ≤ 40
Reflections collected	14691
Independent reflections	4281 [R <sub>int</sub> = 0.0516, R <sub>sigma</sub> = 0.0464]
Data/restraints/parameters	4281/0/319
Goodness-of-fit on F <sup>2</sup>	1.221
Final R indexes [I>=2σ (I)]	R <sub>1</sub> = 0.0597, wR <sub>2</sub> = 0.1445
Final R indexes [all data]	R <sub>1</sub> = 0.0640, wR <sub>2</sub> = 0.1464
Largest diff. peak/hole / e Å <sup>-3</sup>	2.31/-1.46



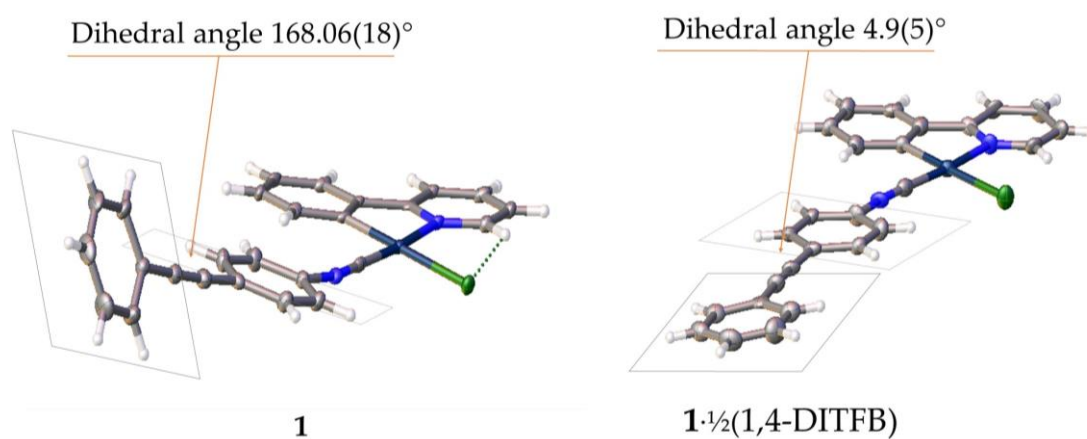
**Figure S1.1.** Molecular view of  $1 \cdot \frac{1}{2}(1,4\text{-DITFB})$ . Thermal ellipsoids for are drawn at the 50% probability level.

**Table S1.2.** Selected bond lengths (Å) and angles (°) for  $1 \cdot \frac{1}{2}(1,4\text{-DITFB})$ .

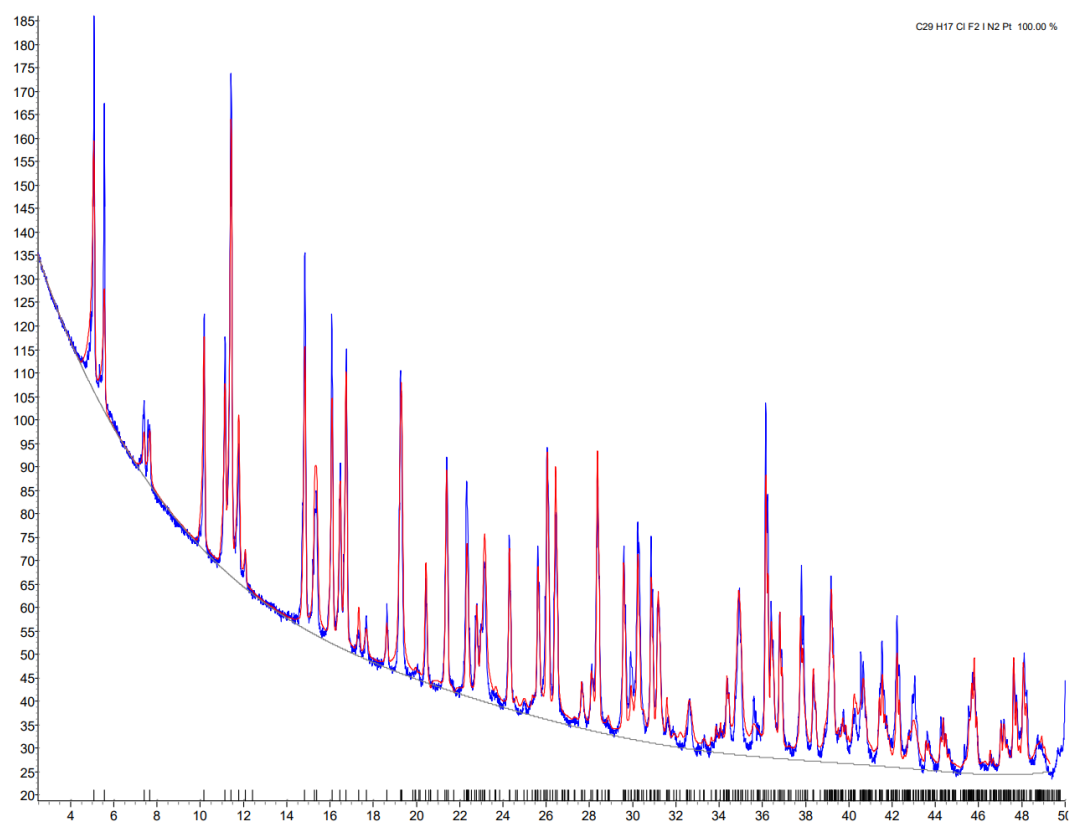
	Length, Å		Angle, °
Pt1–N2	2.077(3)	$\angle(\text{C26–Pt1–N2})$	81.2(5)
Pt1–C26	2.010(12)	$\angle(\text{N2–Pt1–Cl1})$	94.9(3)
Pt1–Cl1	2.396(3)	$\angle(\text{C1–Pt1–C26})$	94.2(5)
Pt1–C1	1.908(14)	$\angle(\text{N1–C1–Pt1})$	178.4(12)
N1–C1	1.147(17)	$\angle(\text{C1–N1–C2})$	168.4(13)
C8–C9	1.211(19)	$\angle(\text{C5–C8–C9})$	174.6(14)



**Figure S1.2.** A fragment of the crystal packing of  $1 \cdot \frac{1}{2}(1,4\text{-DITFB})$ .

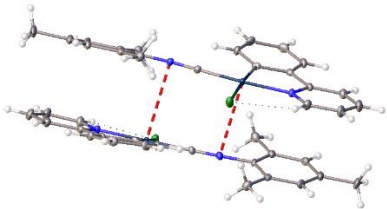
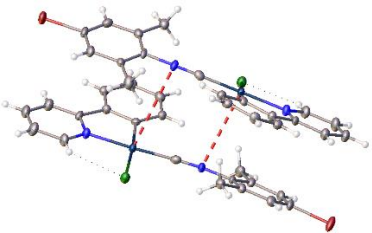
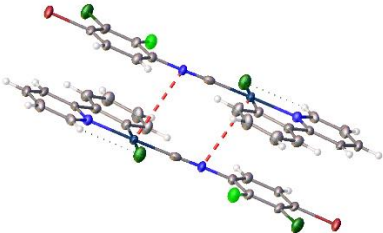
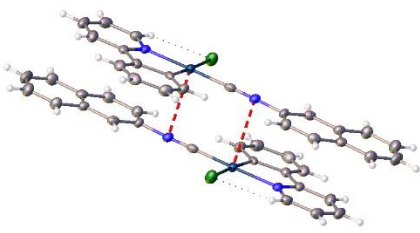
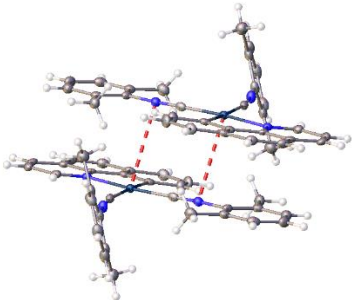
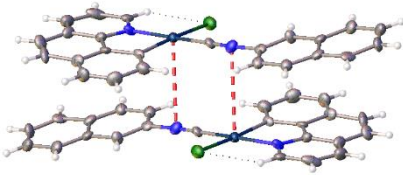


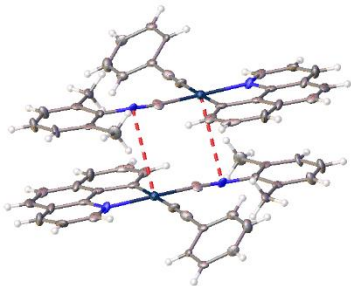
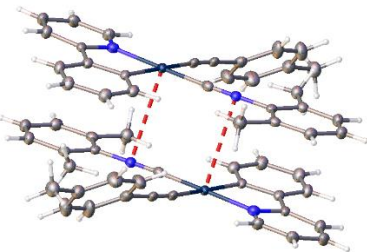
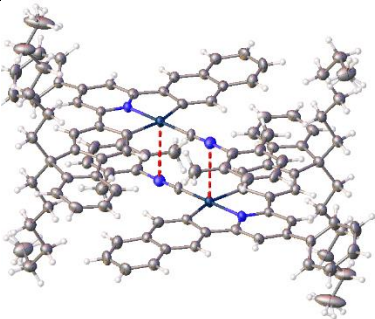
**Figure S1.3.** Molecule structures of complex **1** in the crystals of pure **1** [58] and adduct  $1 \cdot \frac{1}{2}(1,4\text{-DITFB})$ .



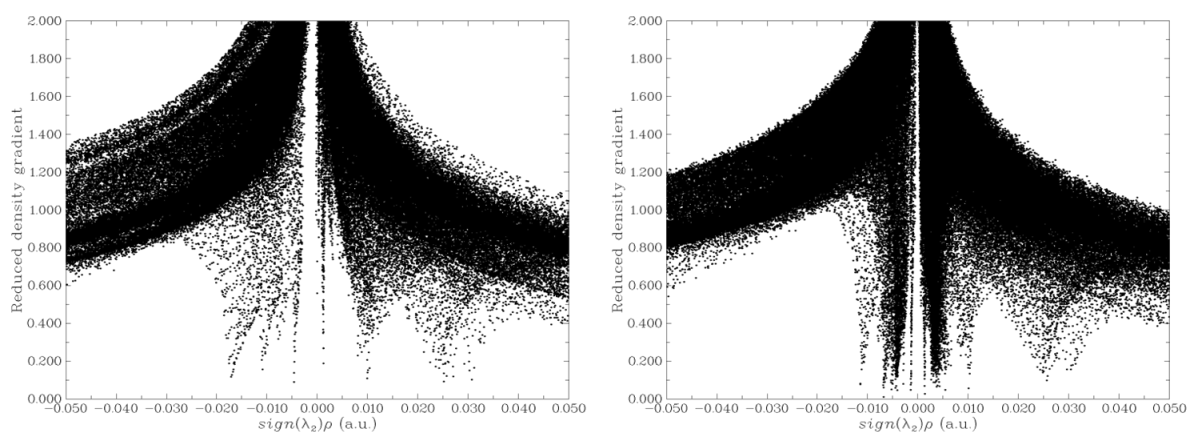
**Figure S1.4.** The measured powder X-ray diffraction patterns of crystalline  $1\cdot\frac{1}{2}(1,4\text{-DITFB})$ .

**Table S1.3** CSD data analysis.

CCDC code	Structure	Pt···N <sub>iso</sub> , Å	∠(C≡N...Pt)	Ref.
2044042		3.636(4), 3.832(4)	98.1(3), 93.4(3)	[1]
2075446		3.624(5), 3.938(6)	106.4(4), 95.1(5)	
2078554		3.475(4)	103.2(4)	
2042219		3.501(3)	86.1(3)	
732182		3.62536(5)	94.4034(14)	[2]
792590		3.654(12)	104.5(10)	[3]

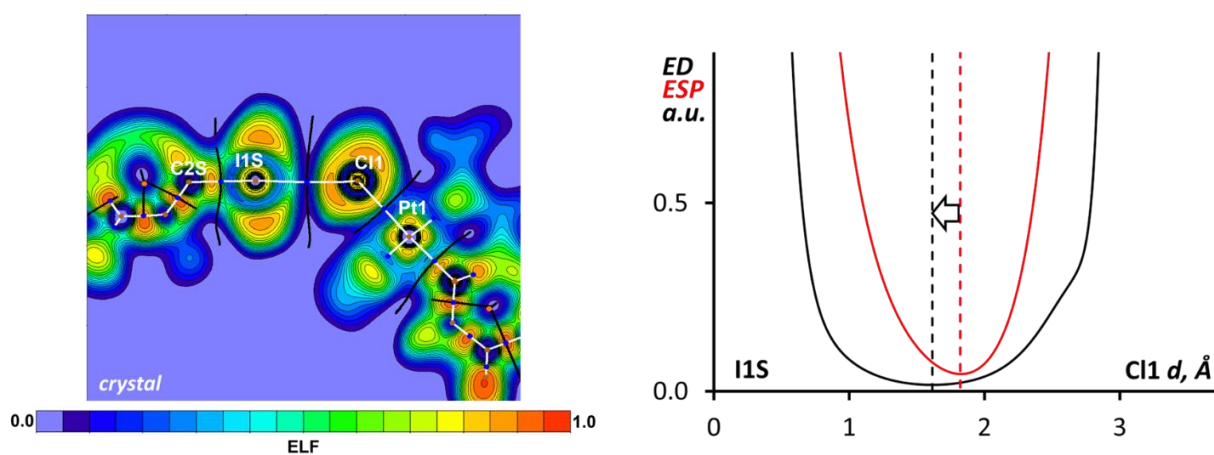
893736		3.696(11)	86.0(9)	[4]
1987102		3.523(4)	86.3(3)	[5]
820799		3.62064(7)	82.2962(19)	[6]
Comparison, Å				
Bondi VdW radii		3.30		
Batsanov VdW radii		3.65		
Alvarez VdW radii		3.95		

## S2. Theoretical calculations

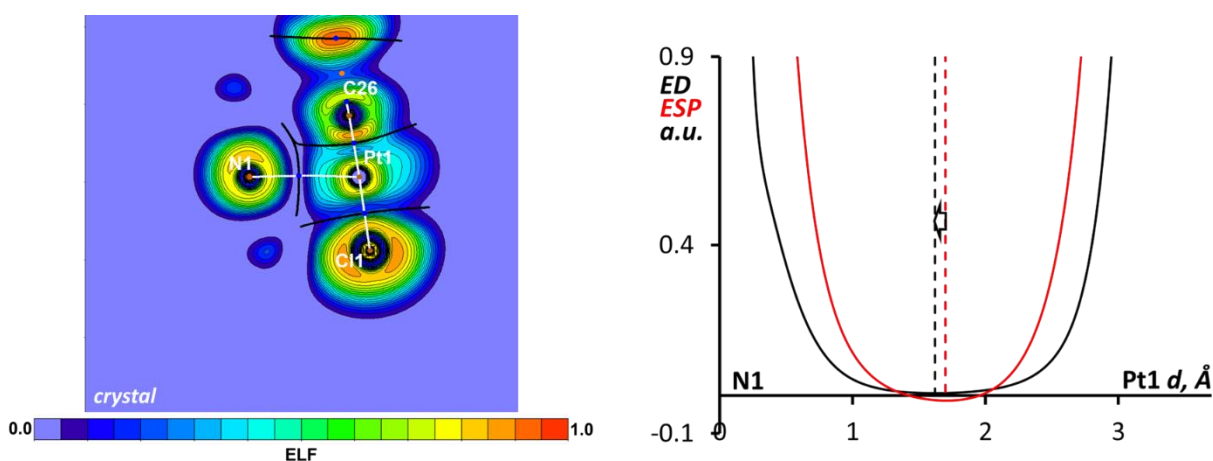


**Figure 2.1.** RDG( $\mathbf{r}$ ) –  $\text{sign}[\lambda_2(\mathbf{r})]\rho(\mathbf{r})$  plot for the I...Cl HaB in (left) and  $\text{N}_{\text{CN}}\cdots\text{Pt}$  contact (right) in the crystal structure of  $1\cdot\frac{1}{2}(1,4\text{-DITFB})$ .

### ELF and ED/ESP Minima Analysis



**Figure 2.2.** Left: ELF projections (contour lines with 0.05 step), bond paths (white lines), zero-flux surface projections (black lines), bond CPs (blue dots), nuclear CPs (brown dots), and ring CPs (orange dots). Right: The ED (black) vs. ESP (red) minima along the bond paths for the I...Cl HaB in the crystal structure of  $1\cdot\frac{1}{2}(1,4\text{-DITFB})$ .



**Figure 2.3.** Left: ELF projections (contour lines with 0.05 step), bond paths (white lines), zero-flux surface projections (black lines), bond CPs (blue dots), nuclear CPs (brown dots), and ring CPs (orange dots). Right: The ED (black) vs. ESP (red) minima along the bond paths for the  $\text{N}_{\text{CN}}\cdots\text{Pt}$  contact in the crystal structure of  $1\cdot\frac{1}{2}(1,4\text{-DITFB})$ .

### Literature

1. Katkova, S.A., et al., *Intermolecular (Isocyano group)···PtII interactions involving coordinated isocyanides in cyclometalated PtII complexes*. J. Mol. Struct., 2022. **1253**: p. 132230.
2. Fornies, J., et al., *Luminescent benzoquinolate-isocyanide platinum(II) complexes: effect of PtPt and pipi interactions on their photophysical properties*. Chem Asian J, 2012. **7**(12): p. 2813-23.
3. Díez, A., et al., *Structural and luminescence studies on pi...pi and Pt...Pt interactions in mixed chloro-isocyanide cyclometalated platinum(II) complexes*. Inorg Chem, 2010. **49**(7): p. 3239-51.
4. Díez, Á., et al., *Synthesis and Luminescence of Cyclometalated Compounds with Nitrile and Isocyanide Ligands*. Organometallics, 2009. **28**(6): p. 1705-1718.
5. Martínez-Junquera, M., et al., *Isomerism, aggregation-induced emission and mechanochromism of isocyanide cycloplatinated(ii) complexes*. Journal of Materials Chemistry C, 2020. **8**(21): p. 7221-7233.
6. Kui, S.C., et al., *Luminescent organoplatinum(II) complexes with functionalized cyclometalated C<sup>N</sup>C ligands: structures, photophysical properties, and material applications*. Chemistry, 2012. **18**(1): p. 96-109.