

Supplementary Materials

Novel Enrofloxacin Schiff base metal complexes: Synthesis, spectroscopic characterization, computational simulation and antimicrobial investigation against some food and phyto-pathogens

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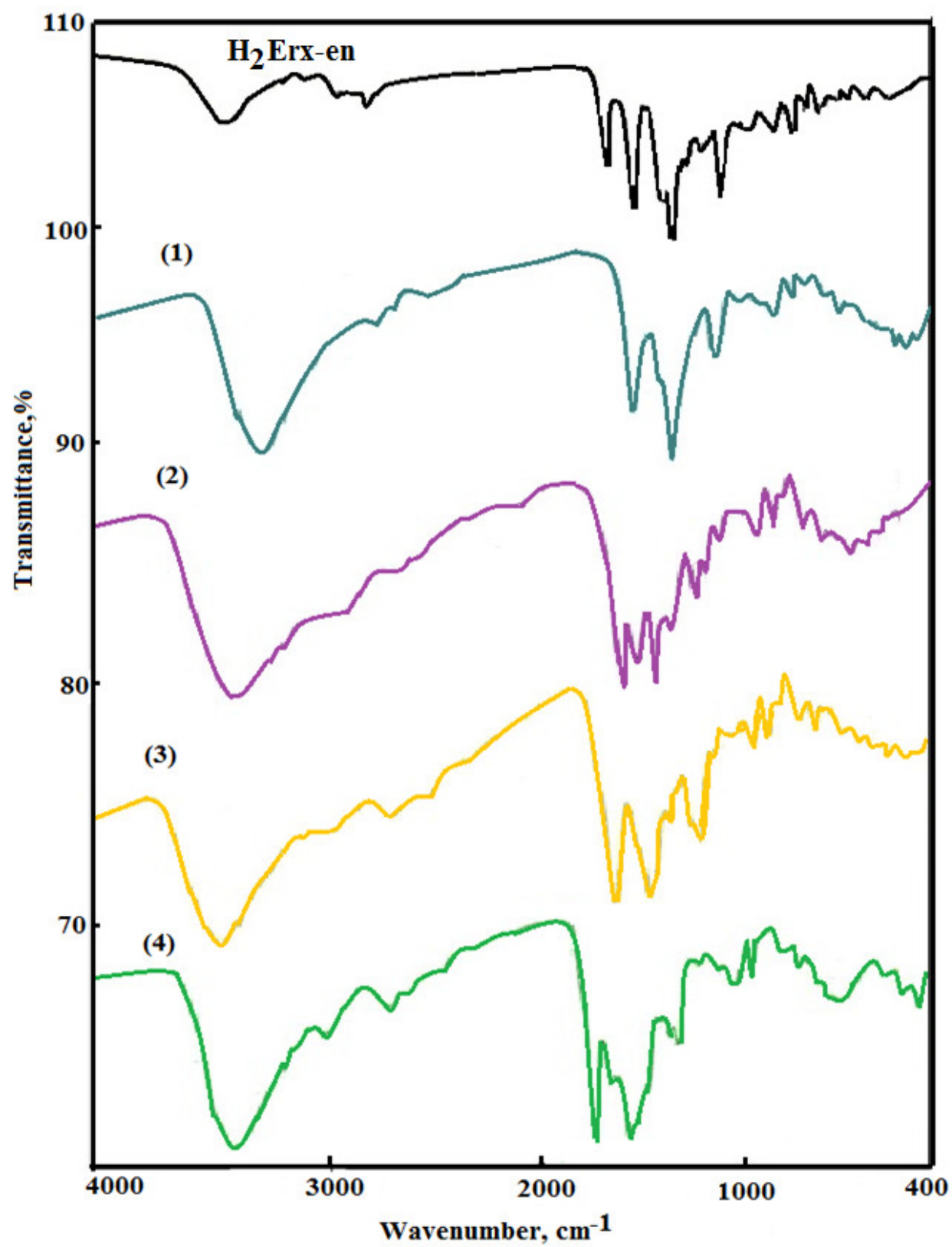


Figure S1. Infrared spectra for **H₂Er_x-en** and its metal complexes

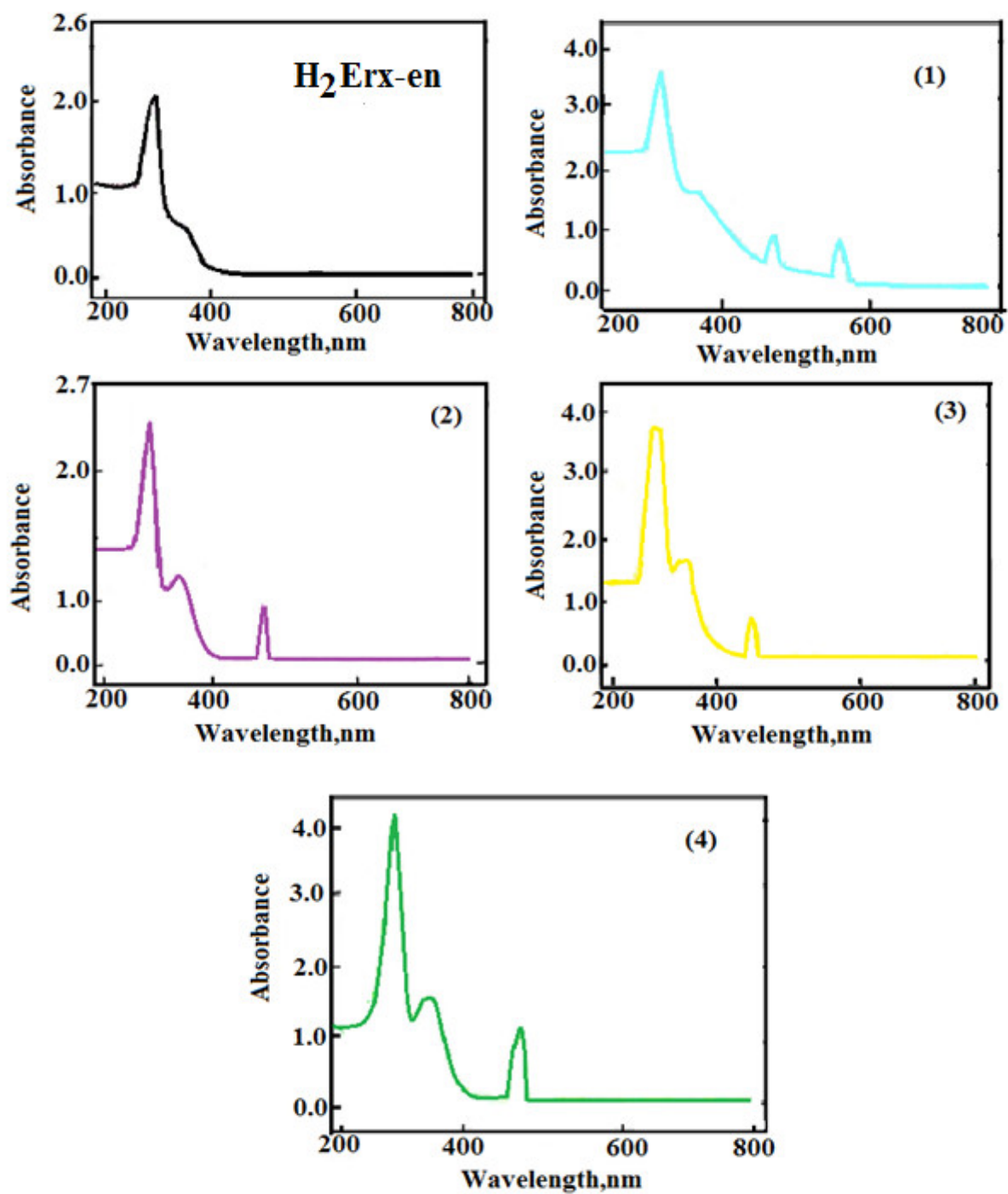


Figure S2. Electronic absorption spectra for $H_2Erx-en$ and its metal complexes

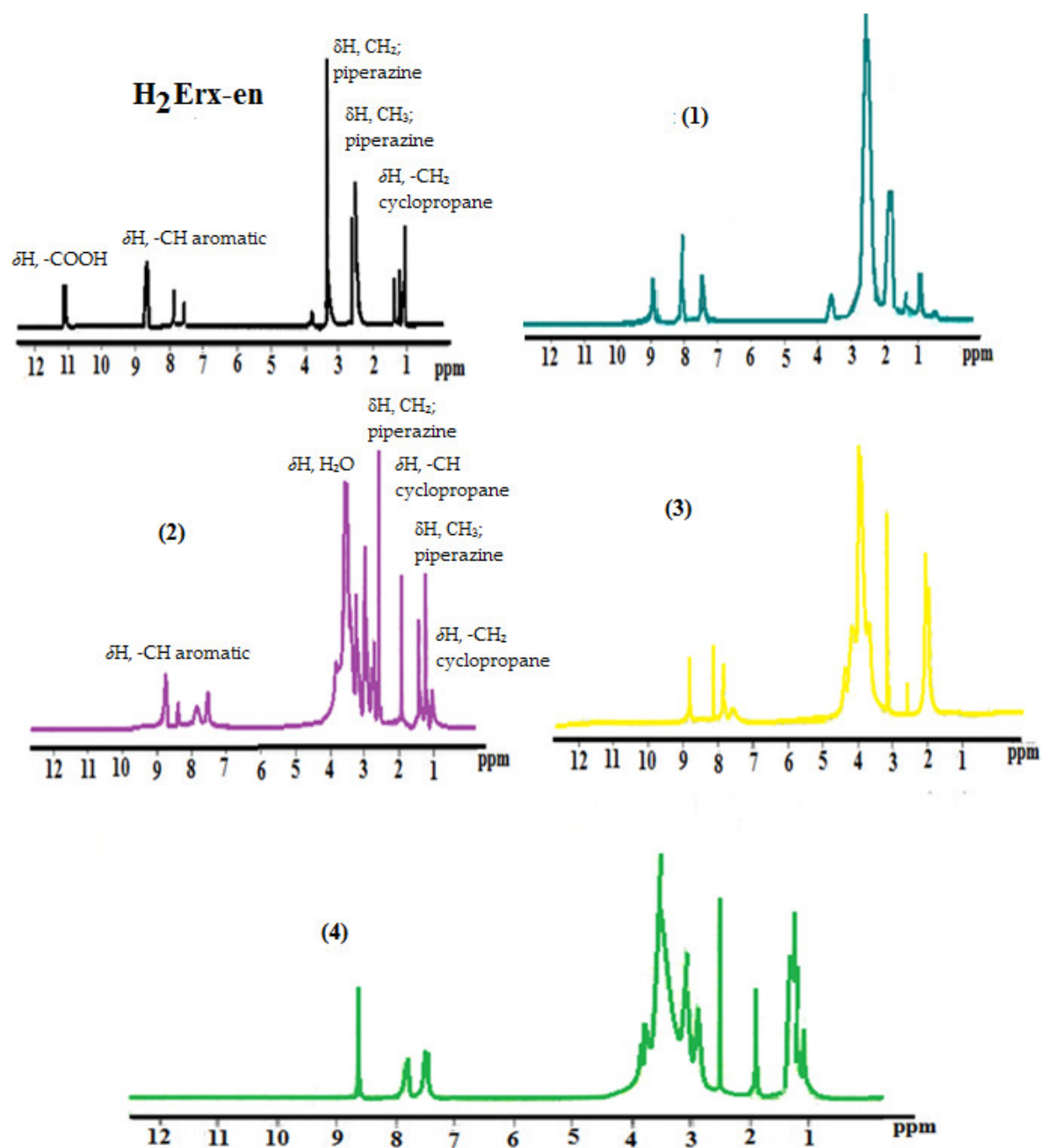


Figure S3. 1H NMR spectra for **H₂Erx-en** and its metal complexes

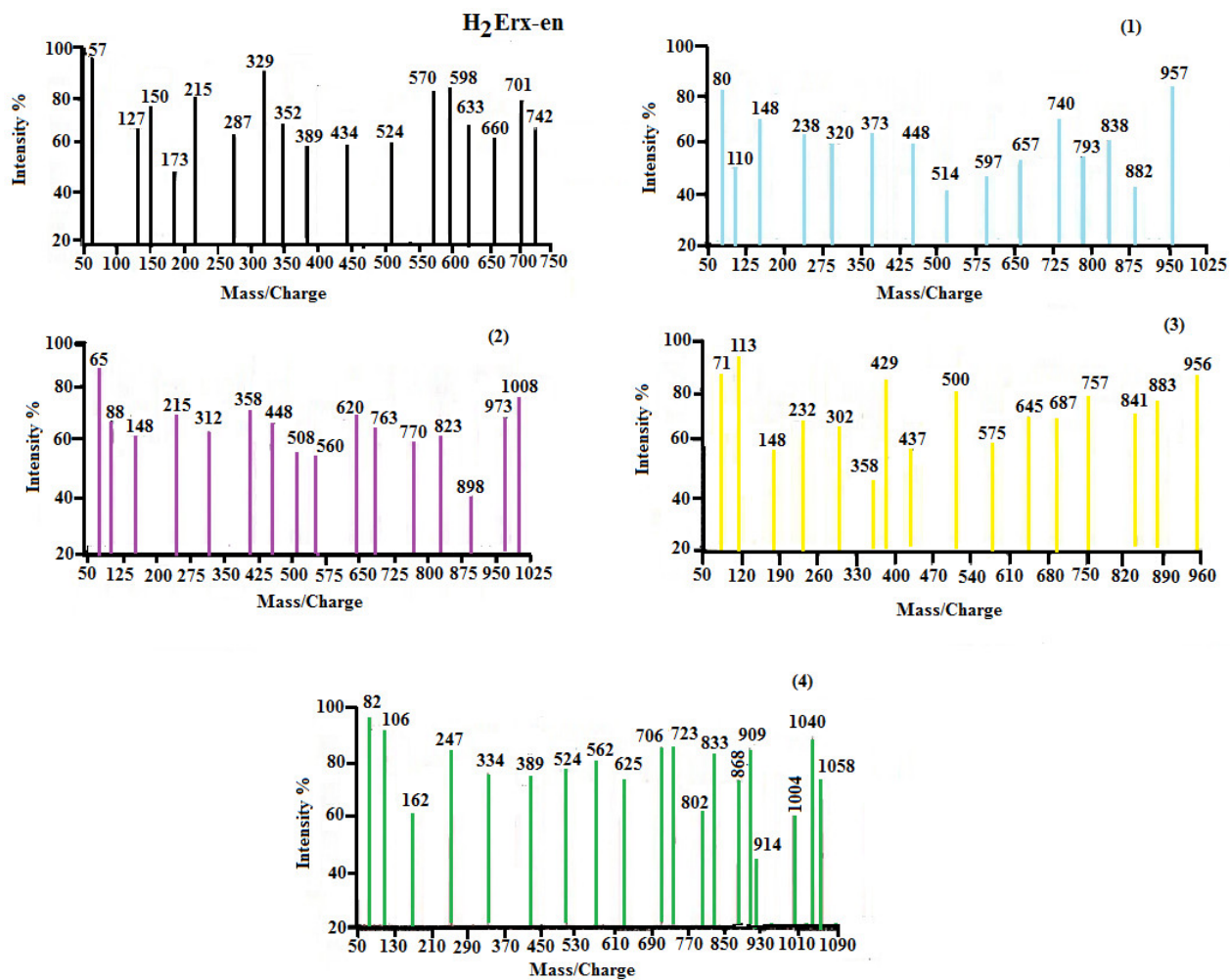


Figure S4. Mass spectra diagrams for **H₂Erx-en** and its metal complexes

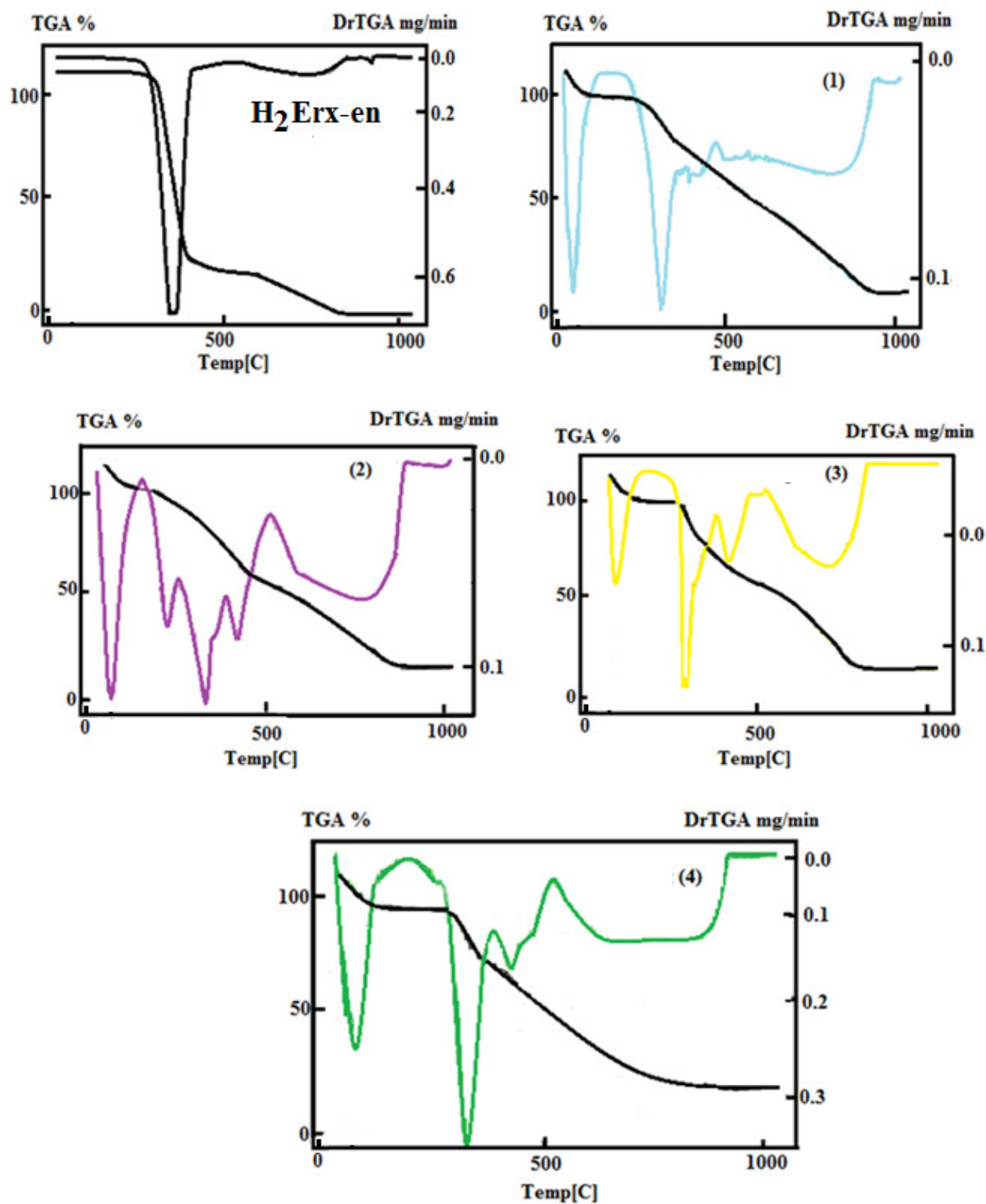
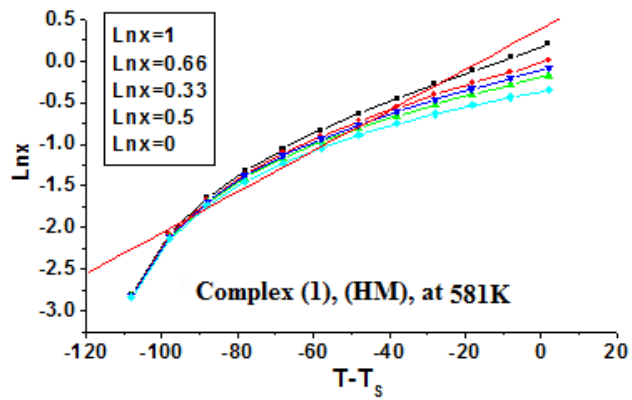
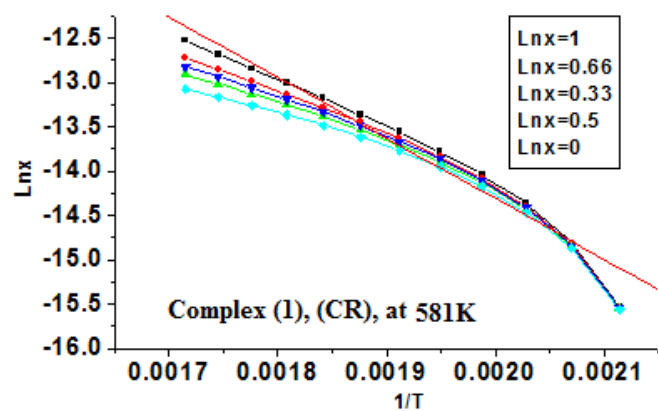
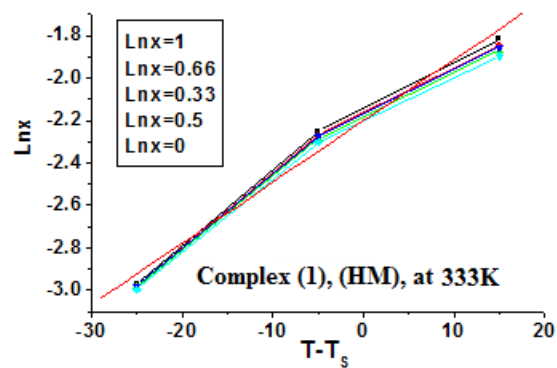
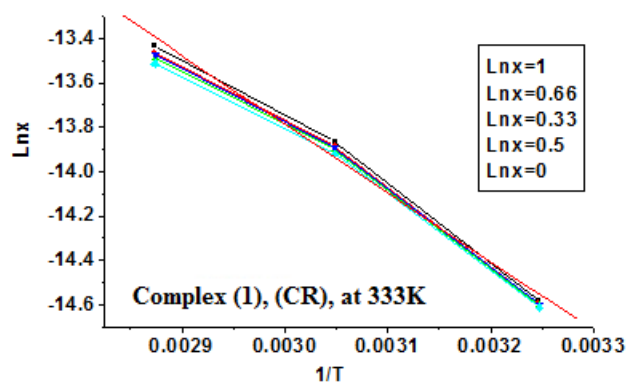
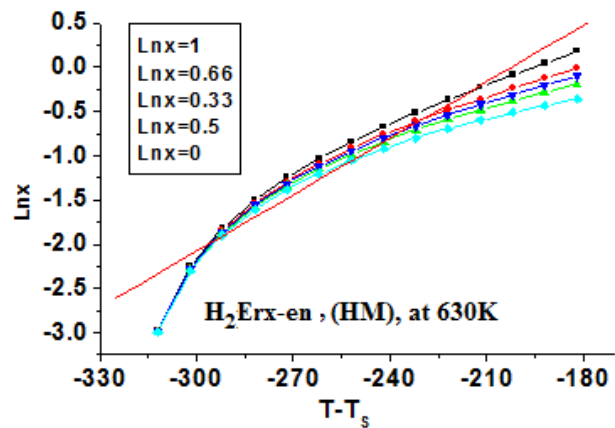
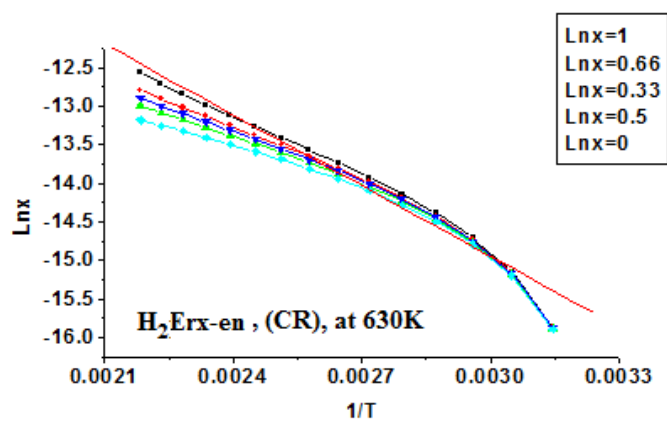
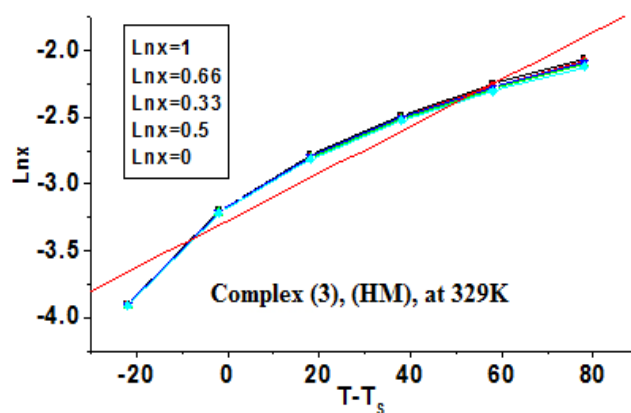
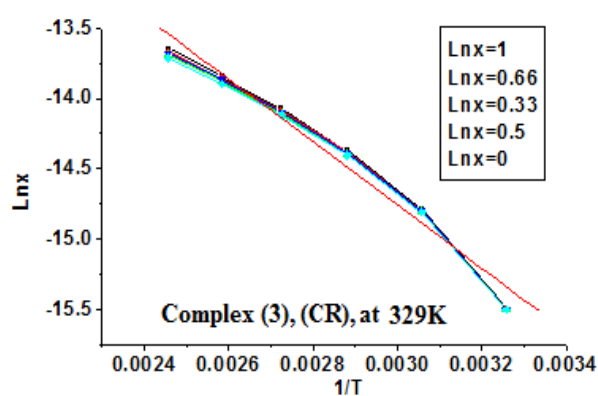
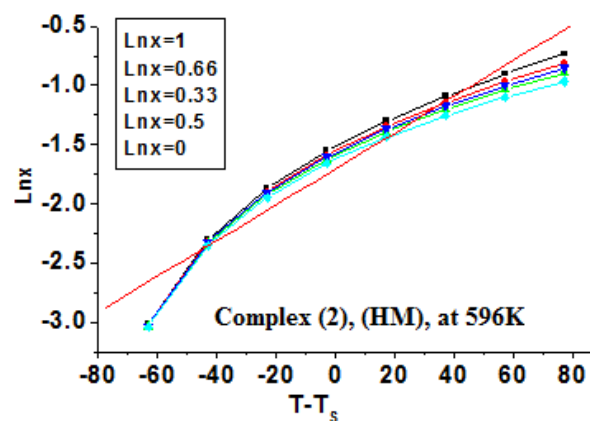
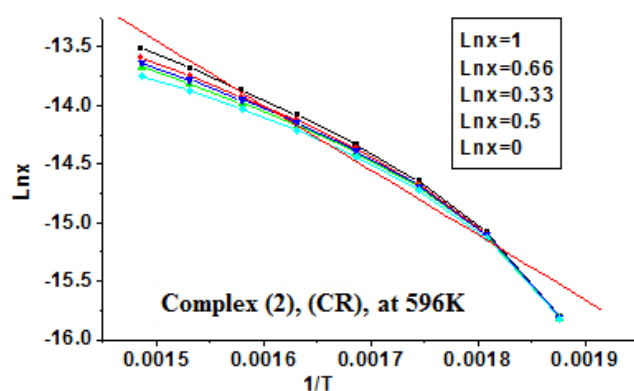
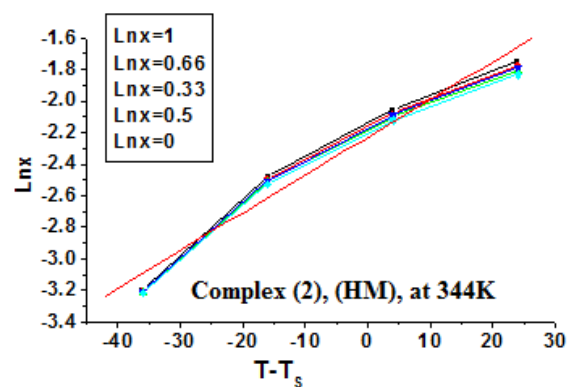
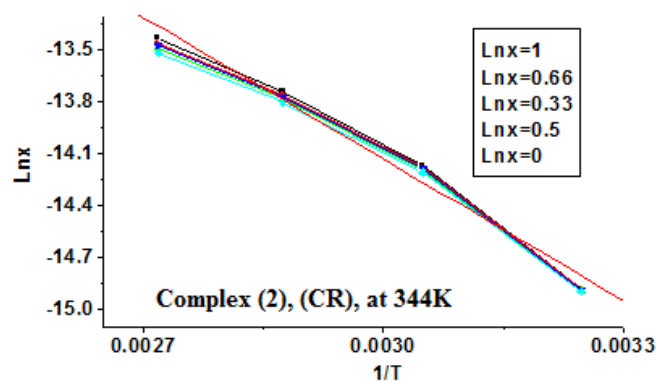


Figure S5. TG and DTG diagrams for $H_2Erx-en$ and its metal complexes.





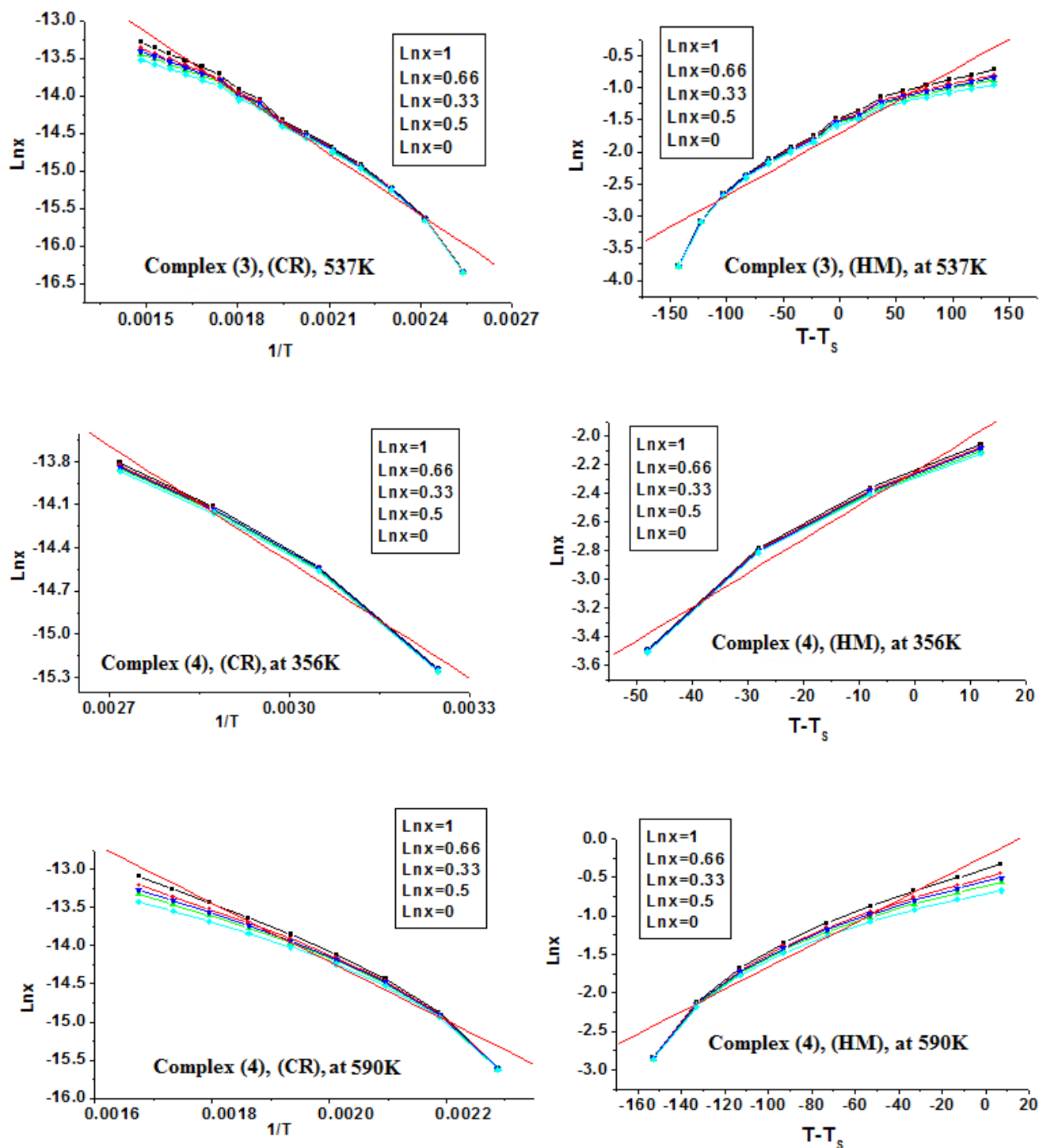


Figure S6. The diagrams of kinetic parameters of $\text{H}_2\text{Er}_x\text{-en}$ and its metal complexes using Coats-Redfern(CR) and Horowitz-Metzger (HM) equations

Table S1. The inhibition diameter zone (mm) values for **H₂Erx-en** and its metal complexes

Tested compounds	Microbial species													
	Bacteria											Fungi		
	<i>E. coli</i>	MIC	AI	<i>Salm. typhi</i>	MIC	AI	<i>S. aureus</i>	MIC	AI	<i>B. cereus</i>	MIC	AI	<i>C. albicans</i>	<i>P. vulpinum</i>
H₂Erx-en	0	0	0	30 ⁺³ ±1	0.75 ±0.01	184.0	28.6 ⁺³ ±0.8	1.00 ±0.03	210.2	0	0	0	0	0
(1)	0	0	0	0	0		31.6 ⁺³ ±2	0.75 ±0.03	232.3	3.3 ⁺² ±0.1	0.50 ±0.007	20.2	0	0
(2)	0	0	0	0	0	0	33.6 ⁺³ ±0.8	0.75 ±0.006	247.0	0	0	0	0	0
(3)	0	0	0	0	0	0	27.6 ⁺³ ±0.8	0.75 ±0.01	202.9	0	0	0	0	0
(4)	0	0	0	27.6 ⁺³ ±1	0.50 ±0.005	169.3	0	0	0	2.3 ⁺³ ±0.1	0.75 ±0.03	0	0	0
Ciprofloxacin	20 ⁺³ ±1	-		0	0		13.6 ⁺³ ±0.8						0	0
Amikacin	14 ⁺³ ±1	-		16.3 ⁺³ ±0.3	-		8.6 ⁺³ ±0.8	-		5.3 ⁺³ ±0.3			0	0
Cefuroxime	0	-		14 ⁺³ ±1	-		18.3 ⁺³ ±0.8	-		0			0	0
Cefpodexime	0	-		7 ⁺³ ±0.5	-		19 ⁺³ ±1	-		0			0	0
Ceftazidime	0	-		15.3 ⁺³ ±2	-		3 ⁺³ ±0.5	-		0			0	0
Ethylene diamine	0	-		15 ⁺³ ±0.5	-		0	-	0	0			0	0
DMF	0	-		0	-		0	-	0	0	-		0	0

Statistical significance PNS P not significant, P <0.05; P+1 P significant, P>0.05; P+2 P highly significant, P>0.01; P+3 P very highly significant, P>0.001; student's t-test (Paired).

Table S2. Equilibrium geometric parameters bond lengths (Å), bond angles (°),dihedral angles (°), total energy (k cal/mol) and dipole moment of **H₂Erx-en** using DFT calculations.

Bond length (Å)							
C2-C3	1.343	N4-C5	1.345	C7-O8	1.209	C13-C14	1.346
N1-C2	1.351	C5-C6	1.346	C7-O9	1.337	C11-C13	1.364
C3-N4	1.345	C6-C7	1.364	N1-C14	1.348	C11-O10	1.335
						C11-O12	1.210
Bond angle (°)							
N1C2C3	128.19	C6C7O8	123.05	C14C13C11		122.17	
C2C3N4	120.99	C6C7O9	121.75	C13C11O10		123.89	
C3N4C5	121.83	O8C7O9	115.19	C13C11O12		120.32	
N4C5C6	121.26	C2N1C14	116.47	O12C11O10		115.69	
C5C6C7	123.58	N1C14C13	120.39				
Dihedral angles (°) of (H ₂ Erx-en)							
C5C6C7O9	-179.29	O8C7C6C5		1.29			
C7C6C5N4	2.44	C6C5N4C3		-87.03			
C5N4C3C2	-177.92	N4C3C2N1		-0.344			
C3C2N1C14	3.87	C2N1C14C13		-94.46			
N1C14C13C11	-4.51	C14C13C11O10		0.458			
C14C13C11O12	-178.89						
Dihedral angles (°) of enrofloxacin molecule (Enro)							
C8C12C16O17	-4.02	C6N9C13C14		126.91			
C21N20C1C4	-81.72	C6N9C13C15		61.16			
C4C1N20C25	90.62	C10C12C16O18		0.25			
C1N20C25C24	136.88	C10C12C16O17		-179.55			
N20C25C24N23	49.68	O11C8C12C16		7.46			
C25C24N23C26	87.48	C27C26N23C24		-81.48			
Total energy/ k cal/mol		-311211.5866330					
Total dipole moment/D		10.56					

Table S3. Equilibrium geometric parameters bond lengths (Å), bond angles (°),dihedral angles (°), Total energy (k cal/mol) and Dipole moment of the studied complexes using DFT calculations.

Bond lengths/ Å	(1)	(2)	(3)	(4)
M-N1	1.982	2.595	2.315	2.751
M-O10	1.930	2.393	2.193	2.439
M-N4	1.973	2.607	2.318	2.753
M-O9	1.935	2.406	2.207	2.499
M-O15	1.999	2.357	2.139	2.448
M-O16	2.111	2.361	2.097	2.455
N1-C2	1.265	1.269	1.271	1.269
C2-C3	1.346	1.343	1.344	1.345
C3-N4	1.269	1.268	1.270	1.269
N1-C14	1.276	1.280	1.279	1.281
C13-C14	1.357	1.358	1.356	1.359
C11-C13	1.362	1.363	1.361	1.364
C11-O10	1.346	1.342	1.343	1.345
C11-O12	1.209	1.211	1.210	1.212
N4-C5	1.275	1.281	1.279	1.280
C5-C6	1.355	1.357	1.356	1.359
C6-O7	1.363	1.361	1.362	1.360
C7-O9	1.342	1.341	1.343	1.342
C7-O8	1.211	1.210	1.212	1.211
N1-M-N4	79.71	70.37	74.87	69.54
N1-M-O9	166.07	129.83	130.34	132.79
N1-M-O10	85.54	75.96	81.83	73.03
N1-M-O15	100.54	150.98	83.21	130.21
N1-M-O16	105.69	78.94	145.39	92.21
N4-M-O9	96.34	75.37	79.53	74.76
N4-M-O10	113.26	130.03	139.28	127.97
N4-M-O15	157.17	134.35	89.78	151.68
N4-M-O16	81.26	119.24	133.31	82.59
O9-M-O10	83.72	101.02	91.32	108.73
O9-M-O15	88.32	77.57	138.75	95.18
O9-M-O16	86.79	151.06	80.73	112.81
O10-M-O15	89.43	90.77	120.35	80.23
O10-M-O16	163.41	88.09	82.79	133.92
O15-M-O16	76.69	74.88	78.20	77.09
Total energy, k cal/mol	-347128.827	-333986.825	-333922.301	-335273.485
Dipole moment, D	12.852	14.597	22.480	13.286

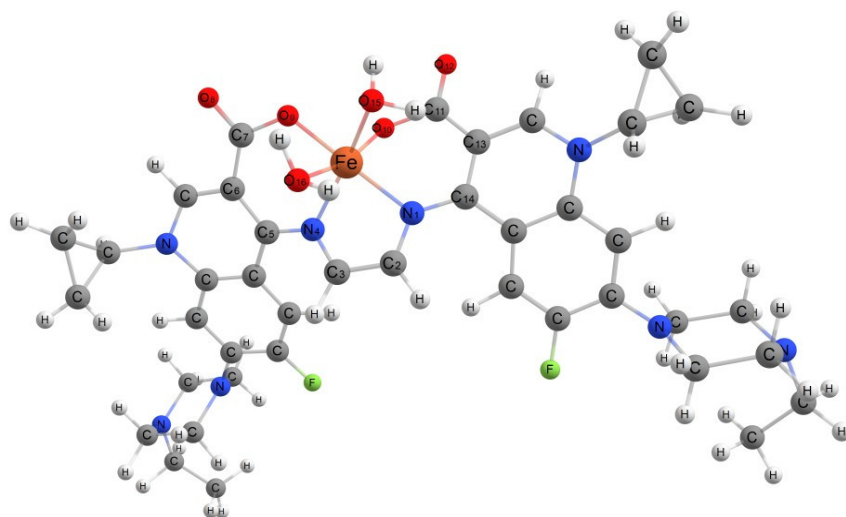
Table S4. Calculated charges on donating sites and energy values (HOMO, LUMO, Energy gap ΔE /eV, hardness (η), global softness (S), electro negativity (χ), absolute softness (σ), chemical potential (Pi), global electrophilicity (ω) and additional electronic charge (ΔN_{\max}) of H₂Er_x-en and its metal complexes using DFT calculation

Parameters	H ₂ Er _x -en	(1)	(2)	(3)	(4)
M	-	0.089	0.448	0.307	0.406
N1	-0.285	-0.094	-0.172	-0.143	-0.173
N4	-0.087	-0.116	-0.175	-0.154	-0.187
O9	-0.362	-0.450	-0.433	-0.373	-0.422
O10	-0.391	-0.445	-0.417	-0.379	-0.426
O15	-	-0.326	-0.299	-0.531	-0.335
O16	-	-0.332	-0.324	-0.305	-0.313
C2	0.165	0.086	0.017	0.018	0.007
C3	-0.043	0.066	0.021	0.037	0.039
C5	-0.120	0.228	0.255	0.248	0.267
C6	-0.057	-0.153	-0.186	-0.172	-0.183
C7	0.482	0.472	0.489	0.493	0.492
O8	-0.454	-0.450	-0.464	-0.461	-0.443
C14	0.302	0.249	0.253	0.239	0.250
C13	-0.146	-0.136	-0.183	-0.167	-0.195
C11	0.517	0.483	0.491	0.498	0.481
O12	-0.429	-0.451	-0.456	-0.456	-0.446
HOMO,H	-0.308	-0.309	-0.310	-0.311	-0.315
LUMO,L	-0.245	-0.297	-0.289	-0.302	-0.310
I = -H	0.308	0.309	0.310	0.311	0.315
A = -L	0.245	0.297	0.289	0.302	0.310
$\Delta E = L-H$	0.063	0.012	0.021	0.009	0.005
$\eta = (I-A)/2$	0.0315	0.006	0.0105	0.0045	0.0025
$\chi = -(H+L)/2$	0.2765	0.303	0.2995	0.3065	0.3125
$\sigma = 1/\eta$	31.476	166.667	95.238	222.22	400.00
$S = 1/2 \eta$	15.873	83.333	47.619	111.11	200.00
Pi = - χ	-0.2765	-0.303	-0.2995	-0.3065	-0.3125
$\omega = (Pi)^2/2 \eta$	1.2135	7.651	4.271	10.438	19.531
$\Delta N_{\max} = \chi/\eta$	8.778	50.5	28.524	68.111	125.00

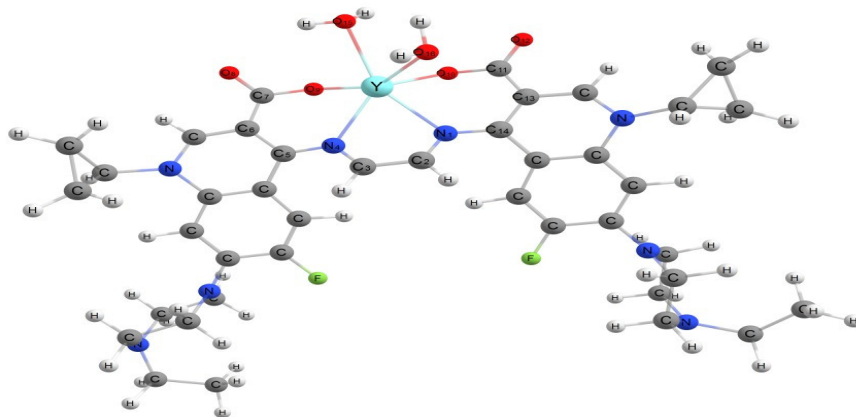
(I) is ionization energy, (A) is an electron affinity

Table S5. Composition of the frontier molecular orbital for the obtained stable complexes using DFT/B3LYP/Cep-31G.

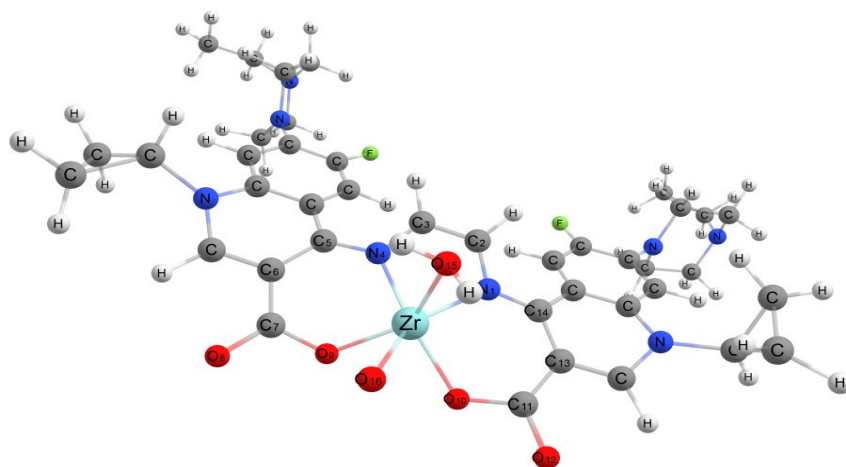
	(1)			(2)	
	H	L		H	L
Fe(III)	0.3%	0.0%	Y(III)	0.0%	0.0%
Enerofloxacin	99.7%	100%	Enerofloxacin	100%	72.4%
ethylenediamine	0.0%	0.0%	ethylenediamine	0.0%	27.6%
O15	0.0%	0.0%	O15	0.0%	0.0%
O16	0.0%	0.0%	O16	0.0%	0.0%
	(3)			(4)	
	H	L		H	L
Zr(IV)	0.0%	0.0%	La(III)	0.0%	66.8%
Enerofloxacin	100%	100%	Enerofloxacin	100%	10.4%
ethylenediamine	0.0%	0.0%	ethylenediamine	0.0%	22.8%
O15	0.0%	0.0%	O15	0.0%	0.0%
O16	0.0%	0.0%	O16	0.0%	0.0%



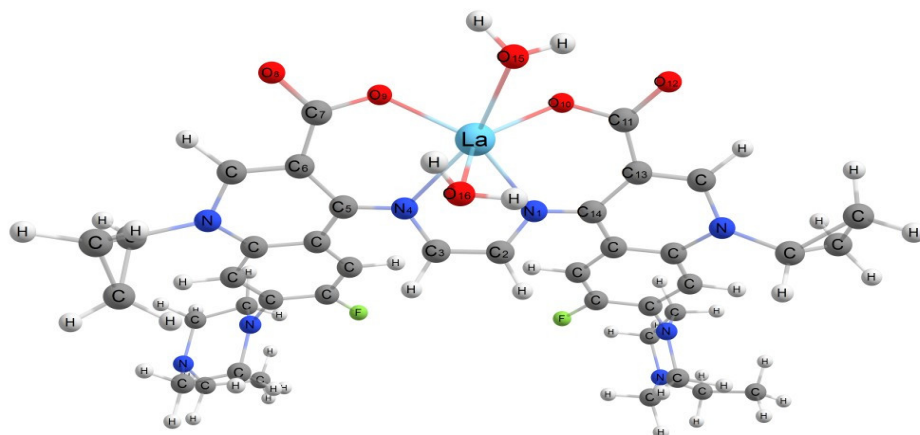
Scheme S1. Optimized geometrical structure of complex **(1)** using DFT calculations.



Scheme S2. Optimized geometrical structure of complex **(2)** using DFT calculations.



Scheme S3. Optimized geometrical structure of complex **(3)** using DFT calculations.



Scheme S4. Optimized geometrical structure of complex **(4)** using DFT calculations.