

## Article

# Enhanced Upconversion Photoluminescence of $\text{LiYF}_4$ : $\text{Yb}^{3+}/\text{Ho}^{3+}$ Crystals by Introducing $\text{Mg}^{2+}$ Ions for Anti-Counterfeiting Recognition

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**Abstract:** By doping appropriate lanthanide ions,  $\text{LiYF}_4$  as a host luminescent material can simultaneously exhibit bright visible-light emission. A series of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  microparticles with different  $\text{Mg}^{2+}$  doping concentrations were synthesized and investigated. The crystal structure of the synthesized microparticles was tested by X-ray diffraction (XRD). Notably, a significant increase in the upconversion photoluminescence intensity of upconversion microparticles (UCMPs) was obtained by introducing  $\text{Mg}^{2+}$  ions under 980 nm laser excitation, and achieved a maximum level when the concentration of  $\text{Mg}^{2+}$  ions was 8 mol%. Additionally, the practicality of the resultant UCMPs used as the raw material of anti-counterfeiting ink was systematically investigated. These results prove that the  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  are very promising as screen-printing materials for anti-counterfeiting recognition labels.

**Keywords:** hydrothermal synthesis;  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$ ; anti-counterfeiting identification; screen printing



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## 1. Introduction

Compared with other luminescent materials (e.g., organic dye, quantum dots, etc.), rare-earth-doped luminescent materials have many advantages, such as sharp emission and long lifetime, good photochemical and thermal stability, etc. [1]. The luminescent properties of lanthanide-doped fluoride materials are mainly subject to the host matrix, sensitizers, and activators [2–5]. Hence, optimal host–dopant combinations are necessary for an efficient UC luminescence process. It is universally known that the UC efficiency of rare-earth ions depends largely on the phonon energy of the matrix.  $\text{LiYF}_4$  is also considered to be an ideal alternative method, which is likely to inhibit non-radiative relaxation due to its low phonon energy [6]. In addition, the  $\text{LiYF}_4$  host can simultaneously exhibit bright visible-light emission by doping with appropriate lanthanides. Herein,  $\text{LiYF}_4$  was selected as the host material, and the combination  $\text{Yb}^{3+}/\text{Ho}^{3+}$  was used as a sensitizer/activator to increase the excitation of the photons.

The doping of non-rare-earth ions is regarded as a promising approach for the improvement of UC [6–10]. Zhou et al. reported the enhancement of green upconversion luminescence in tetrahedral  $\text{LiYF}_4:\text{Yb}/\text{Er}$  nanoparticles by doping with  $\text{Mn}^{2+}$  ions [6]. Li et al. synthesized  $\text{NaLnF}_4$  UCNPs via a simple  $\text{M}^{2+}$  ( $\text{Mg}^{2+}$ ,  $\text{Co}^{2+}$ ) doping method [7]. Yang et al. synthesized  $\text{Mg}^{2+}$ -doped hexagonal-phase  $\text{NaGdF}_4:\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$  nanocrystals; the results show that the intensity of UC emission increased significantly as a result of  $\text{Mg}^{2+}$  doping [8]. The results show that overall UCL is enormously enhanced after doping with  $\text{Mg}^{2+}$  ( $\approx 27.5$ -fold). Ding et al. synthesized  $\text{Li}^+$ -doped  $\beta\text{-NaGdF}_4:\text{Ln}^{3+}$  nanocrystals, and a significant enhancement of UC emission in the  $\beta\text{-NaGdF}_4:\text{Ln}^{3+}$  nanocrystals was observed [9]. Ding et al. synthesized a  $\text{Mn}^{2+}$ -doped core–shell–shell nanoarchitecture, and

the NIR-to-red upconversion emission was efficiently promoted [10]. Dopant-enhanced up-conversion nanoparticles (UCNPs) have attracted great interest for anti-counterfeiting for their specific UC luminescence property of emitting higher-efficiency shorter wavelengths via near-infrared (NIR) excitation [11–15].

In this work, a series of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different  $\text{Mg}^{2+}$  doping concentrations were successfully prepared via hydrothermal synthesis methods. In  $\text{LiYF}_4$  crystals, because the radius of  $\text{Mg}^{2+}$  (0.72 Å) ions is less than that of  $\text{Y}^{3+}$  (0.90 Å), the incorporation of  $\text{Mg}^{2+}$  ions replaces some of the  $\text{Y}^{3+}$  ions or fills the crystal gap, resulting in the shrinkage and expansion of the main lattice and the asymmetry of local crystal field. The symmetry of the  $\text{LiYF}_4$  local crystal field is changed by the doping with  $\text{Mg}^{2+}$  ions, so as to affect the UC luminescence efficiency of the product [16,17]. The crystal structure and photoluminescence intensity of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs were characterized. Finally, the fluorescence imaging experiment was carried out. The imaging results prove that the prepared UCMPs can be applied to screen printing.

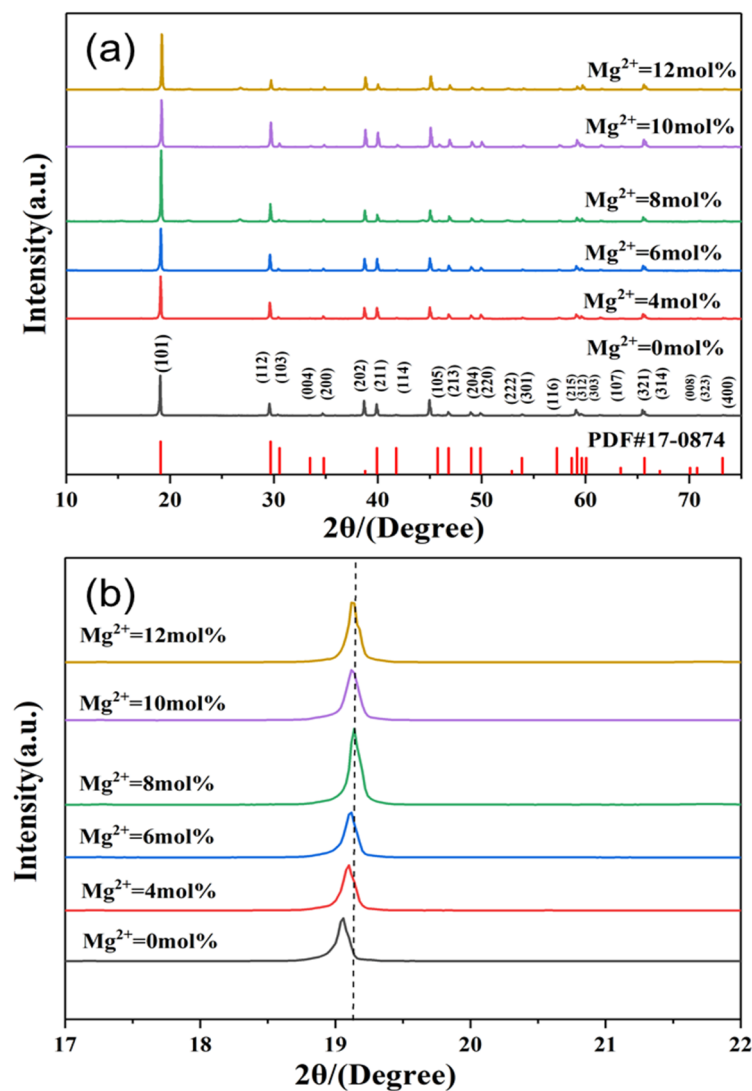
## 2. Experimental Section

$\text{Mg}^{2+}$ -ion-doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  micron crystals were prepared via hydrothermal synthesis. The specific preparation process was as follows: First, 1 mmol ethylenediaminetetraacetic acid (EDTA) powder was dissolved in 20 mL of deionized water in a beaker and stirred continuously under the auxiliary condition of constant-temperature heating at 60 °C to form solution A. Secondly, 1 mmol rare-earth nitrate was dissolved in 10 mL of deionized water and subjected to ultrasonic treatment for 10 min to form a mixed solution B. The rare-earth nitrate included  $\text{Y}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ,  $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{Yb}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ , and  $\text{Ho}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ , with the proportions of 0.79- $x$ ,  $x$ , 0.2, and 0.01, respectively.  $x$  is the doping amount of  $\text{Mg}^{2+}$ . Solution A and solution B were mixed and stirred well. A mixed solution of 20 mmol LiF and 5 mmol  $\text{NH}_4\text{F}$  was dissolved in water to form solution C (5 mL). Finally, solution C was dropped slowly into the mixed solution and continuously stirred for about 30 min to form a milky white liquid. Finally, the mixture was put into the autoclave, and maintained for 48 h at 230 °C. The products were washed with a mixture of deionized water and ethanol (volume, 1:1) and centrifuged several times. Then, they were dried in a vacuum-drying oven at 60 °C for 12 h and, finally, the samples were ground and collected.

The X-ray diffraction (XRD) patterns of the products demonstrated their crystal structure and phase identification, and were recorded using a BRUKER D8 ADVANCE type X-ray diffractometer with Cu K $\alpha$  radiation ( $\lambda$  is 0.1546 nm). The elemental content of the sample was measured using an inductively coupled plasma optical emission spectrometer (ICP-OES, Agilent 5110), see Table S1. A power-adjustable 980 nm laser diode (0 to 2W) was selected as a pump light source. The photoluminescence spectra were obtained using a Hitachi F-4500 fluorescence spectrophotometer (1.0 nm for spectral resolution (FWHM) of the spectrophotometer and 400 V for PMT voltage) at room temperature. The camera employed for taking photographs in the imaging experiments was a high-performance CCD camera produced by Andorra, UK.

## 3. Results and Discussion

The XRD results of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different  $\text{Mg}^{2+}$  doping concentrations are shown in Figure 1. The concentrations of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with  $\text{Mg}^{2+}$  ion doping were 0 mol%, 4 mol%, 6 mol%, 8 mol%, 10 mol%, and 12 mol%. The main diffraction peaks were 100, 110, 101, 200, 111, 201, 210, 002, 300, 211, 102, 112, 220, 311, 312, and 302. The diffraction peaks of the  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs generally coincided with the PDF standard card (PDF: 17-0874). There were no other miscellaneous peaks, even with high-concentration  $\text{Mg}^{2+}$  ion doping, indicating that pure UCMPs were prepared.



**Figure 1.** (a) XRD diffraction pattern of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs; (b) diffraction peak of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs around  $19^\circ$ .

As a result, Figure 1a shows the XRD pattern of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  around  $19^\circ$ . As seen from the local magnification of the XRD images, the diffraction peak shifts to the right with the increase in  $\text{Mg}^{2+}$  concentration. The angular movement of the main diffraction peak reaches the limit when the  $\text{Mg}^{2+}$  doping concentration is 8 mol%. When the concentration of  $\text{Mg}^{2+}$  ions exceeds 8 mol%, the diffraction peak of the nanocrystal starts to move to the left again.

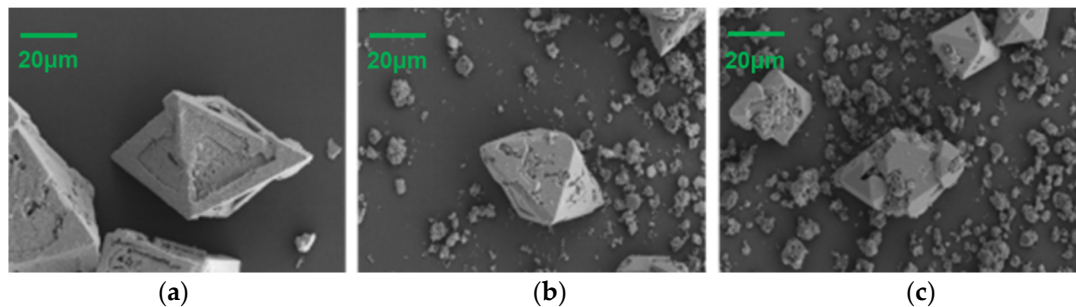
In  $\text{LiYF}_4$  crystal, because the radius of  $\text{Mg}^{2+}$  ( $0.72 \text{ \AA}$ ) ions is less than that of  $\text{Y}^{3+}$  ( $0.90 \text{ \AA}$ ), the incorporation of  $\text{Mg}^{2+}$  ions replaces some of the  $\text{Y}^{3+}$  ions or fills the crystal gap, resulting in the shrinkage and expansion of the main lattice and the asymmetry of the local crystal field. When the doping concentration of  $\text{Mg}^{2+}$  ions is 0–8 mol%, the crystal shrinks, and the main diffraction peak of XRD moves to the right. When the doping concentration of  $\text{Mg}^{2+}$  ions is greater than 8 mol%, the lattice structure expands, and the main diffraction peak begins to move to the left. The symmetry of the  $\text{LiYF}_4$  local crystal field is changed by the doping of  $\text{Mg}^{2+}$  ions, so as to affect the UC luminescence efficiency of the product [16,17].

$$2d\sin\theta = n\lambda \quad (1)$$

In the Bragg's law formula,  $\lambda$ ,  $d$ , and  $\theta$  are the diffraction wavelength, the distance between planes, and the diffraction angle, respectively. According to Formula (1), when

a certain amount of  $\text{Mg}^{2+}$  ions are added, some  $\text{Mg}^{2+}$  ions will enter the lattice gap of  $\text{LiYF}_4$ , increasing the distance  $d$  between planes and reducing the diffraction angle  $\theta$ . As the concentration of  $\text{Mg}^{2+}$  ions decreases, the  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  lattice expands, so the XRD moves to a low angle.

As shown in Figure 2, the  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different  $\text{Mg}^{2+}$  impurity concentrations were tested by SEM. In Figure 2, the samples have good crystallinity, and the morphology of the samples is similar to a biconical octahedron structure. The morphology is regular and the dispersion is mainly uniform. The average size of the microcrystals is about 55  $\mu\text{m}$  in length and 30  $\mu\text{m}$  in width under different  $\text{Mg}^{2+}$  ion doping concentrations.

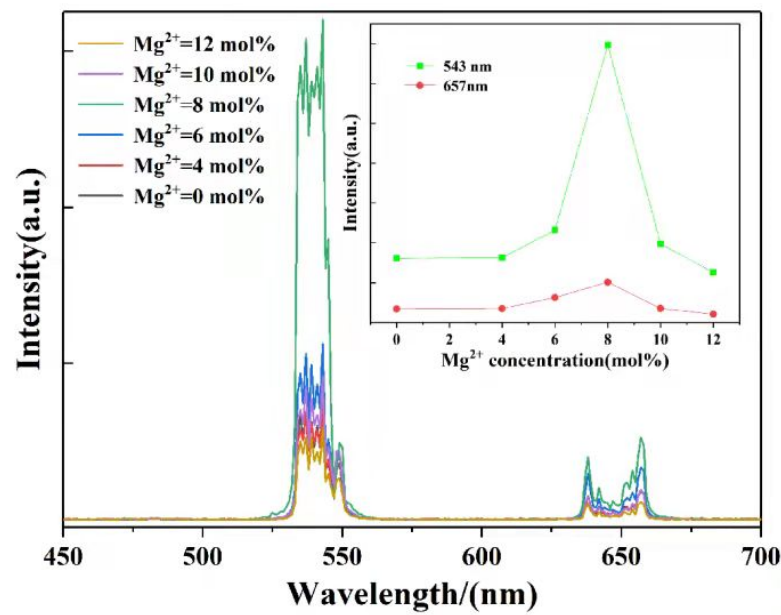


**Figure 2.** SEM of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs. (a)  $\text{Mg}^{2+}=0$  mol%, (b)  $\text{Mg}^{2+}=8$  mol%, (c)  $\text{Mg}^{2+}=12$  mol%.

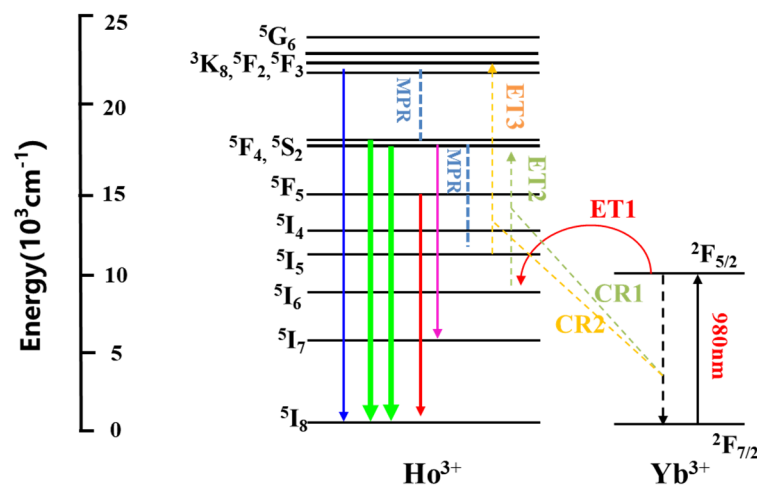
The photoluminescence was investigated as a function of the change in  $\text{Mg}^{2+}$  doping concentration. The  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with  $\text{Mg}^{2+}$  doping concentrations of 0%, 4%, 6%, 8%, 10%, and 12% were selected for fluorescence spectroscopy tests.

Figure 3 shows the photoluminescence spectra of different  $\text{Mg}^{2+}$  doping concentrations of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs under 980 nm laser excitation. As shown in Figure 3, the photoluminescence intensity increased continuously as the  $\text{Mg}^{2+}$  doping concentration changed from 0 mol% to 8 mol%. The photoluminescence intensity reached the maximum as the  $\text{Mg}^{2+}$  doping concentration reached 8 mol%. However, when the  $\text{Mg}^{2+}$  doping concentration was greater than 8 mol%, the photoluminescence intensity gradually weakened with the increase in  $\text{Mg}^{2+}$  doping concentration. Compared with the samples without  $\text{Mg}^{2+}$  doping, the UC luminescence intensity of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with a doping concentration of 8 mol% was increased by about fourfold. It was thus proven that the adding  $\text{Mg}^{2+}$  ions into  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  micron crystals can enhance the upconversion luminescence intensity effectively. Because the incorporation of  $\text{Mg}^{2+}$  ions changes the symmetry of the crystal field, the crystal field distortion also affects the UC luminescence efficiency of  $\text{Ho}^{3+}$  ions, improving the UC luminescence performance of the  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  crystals [7,8].

The energy level transition diagram of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  under the irradiation of 980 nm lasers is shown in Figure 4. After the  $\text{Yb}^{3+}$  ions absorb laser energy, they first transition from the  $^2\text{F}_{7/2}$  energy level to the  $^2\text{F}_{5/2}$  energy level, followed by cross-relaxation and energy transfer with  $\text{Ho}^{3+}$  ions. Then, a series of radiative transitions of  $\text{Ho}^{3+}$  ions are formed through the energy transfer ( $\text{ET}_1$ ) process, and through energy transfer ( $\text{ET}_2$ ,  $\text{ET}_3$ ) caused by cross-relaxation processes ( $\text{CR}_1$  and  $\text{CR}_2$ ) between two close  $\text{Yb}^{3+}$  and  $\text{Ho}^{3+}$  ions [18]. This results in the green UC emission of  $\text{Ho}^{3+}$  ions at 535 nm ( $^5\text{F}_4 \rightarrow ^5\text{I}_8$ ) and 543 nm ( $^5\text{S}_2 \rightarrow ^5\text{I}_8$ ), the red UC emission of  $\text{Ho}^{3+}$  at 638 nm ( $^3\text{K}_8$ ,  $^5\text{F}_2$ ,  $^5\text{F}_3 \rightarrow ^5\text{I}_7$ ) and 657 nm ( $^5\text{F}_5 \rightarrow ^5\text{I}_8$ ), and near-infrared UC emission of  $\text{Ho}^{3+}$  ions at 750 nm ( $^5\text{F}_4$ ,  $^5\text{S}_2 \rightarrow ^5\text{I}_7$ ).

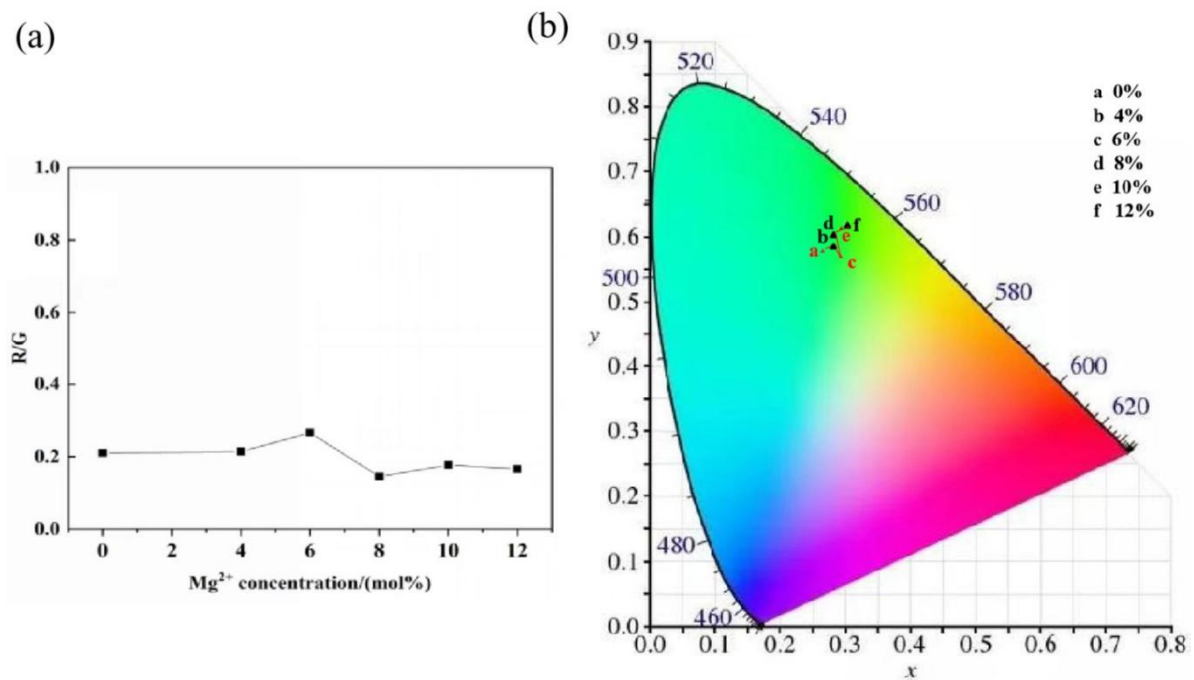


**Figure 3.** Photoluminescence spectra of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different  $\text{Mg}^{2+}$  doping concentrations.



**Figure 4.** Schematic diagram of energy level transition of  $\text{Mg}^{2+}$ -doped  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs.

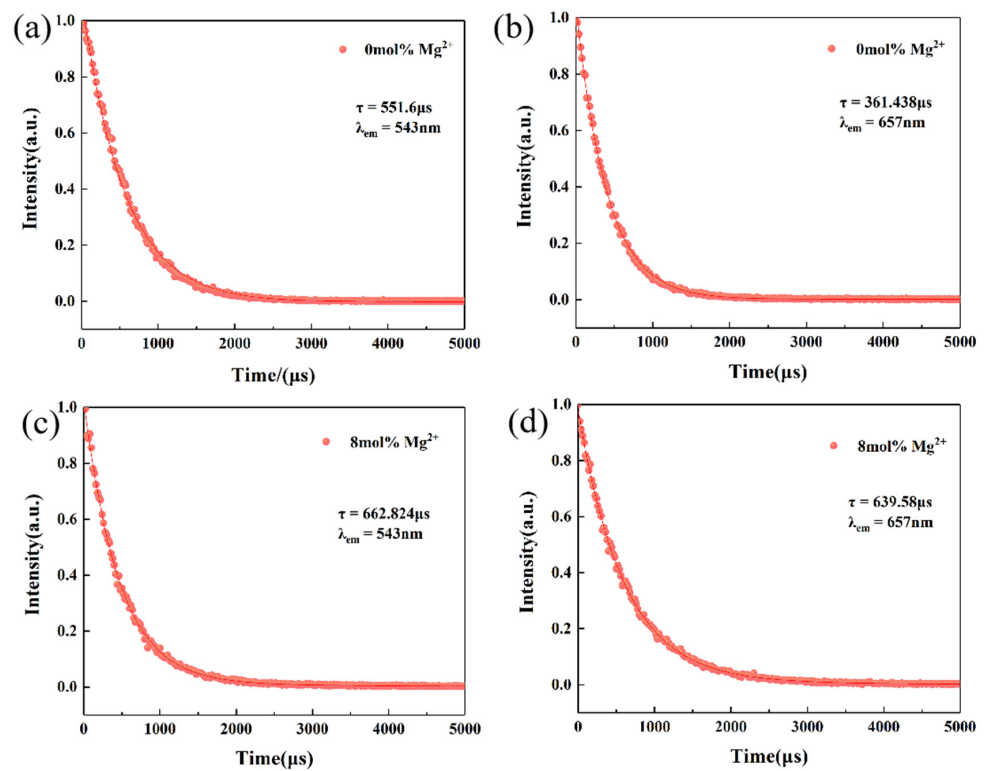
The red–green light intensity ratio is an essential standard to evaluate the luminescence performance of luminescent materials. Figure 5a shows the changes in the upconversion red–green light intensity ratio ( $R/G = I_{657\text{nm}}/I_{543\text{nm}}$ ) of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs doped with different  $\text{Mg}^{2+}$  ion concentrations. With the change in  $\text{Mg}^{2+}$  ion concentration, the conversion of the ratio of red light to green light of UC luminescence is only about 10%, indicating that the UC luminescence of the sample has good monochromaticity, and is affected only slightly by the concentration of  $\text{Mg}^{2+}$  ion doping. In Figure 5b, the visible light colors of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  with different concentrations of  $\text{Mg}^{2+}$  ion doping are exhibited in the CIE color coordinates (without interference by UV/IR light). The coordinates of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different concentrations  $\text{Mg}^{2+}$  ion doping were (0.2653, 0.5731), (0.2809, 0.5792), (0.2977, 0.5563), (0.2844, 0.6024), (0.2990, 0.6127), and (0.3151, 0.6204), respectively.



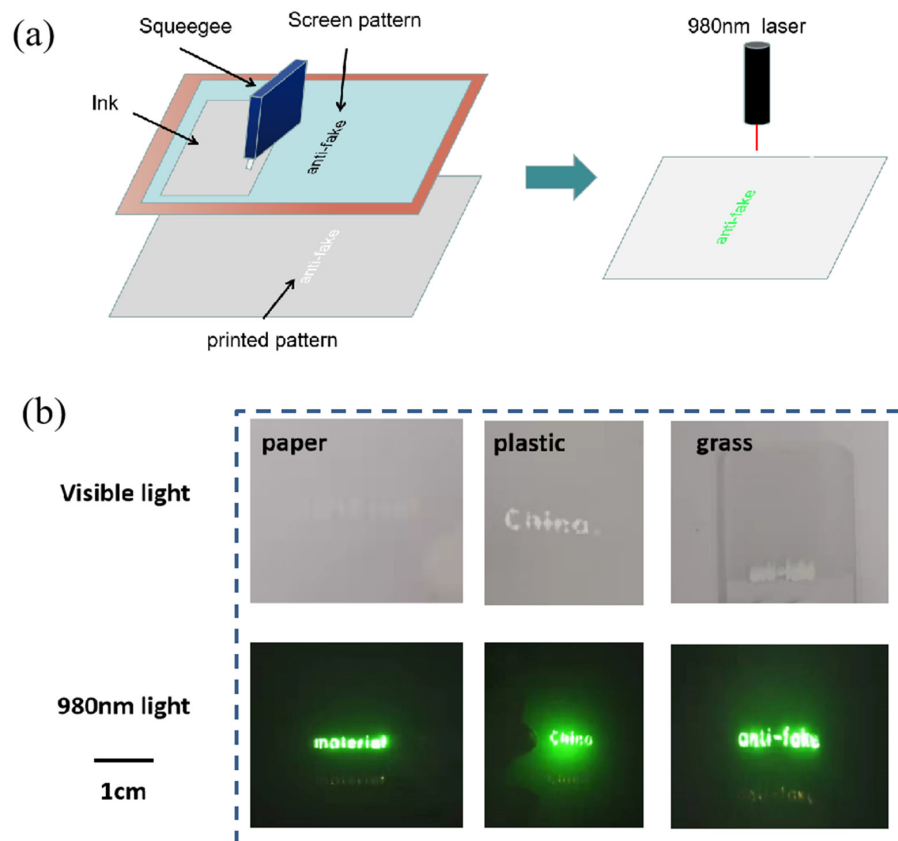
**Figure 5.** (a) Upconversion red–green light intensity ratio of Mg<sup>2+</sup>-doped LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs; (b) CIE of Mg<sup>2+</sup>-doped LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs.

In order to further study the UC luminescence enhancement effect of Mg<sup>2+</sup> ions on LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs, the UC luminescence lifetimes at 543 nm and 657 nm of LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs doped with and without Mg<sup>2+</sup> ions were measured under 980 nm laser excitation. Then, the corresponding fluorescence lifetimes were calculated by fitting the test results. As shown in Figure 6a,b, the UC fluorescence lifetime of LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs without Mg<sup>2+</sup> ion doping at 543 nm was 551.6  $\mu$ s, and the UC fluorescence lifetime at 657 nm was 361.438  $\mu$ s. As shown in Figure 6c,d, the UC fluorescence lifetime of 8 mol% Mg<sup>2+</sup>-ion-doped LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs at 543 nm was 662.824  $\mu$ s, and the UC fluorescence lifetime at 657 nm was 639.58  $\mu$ s. This shows that the UC fluorescence lifetime at 543 nm and 657 nm is increased by about 20% and 70%, respectively, due to doping with Mg<sup>2+</sup> ions. These results demonstrate that doping with Mg<sup>2+</sup> ions can significantly enhance the fluorescence lifetime of the samples.

Screen printings were made on various materials, including paper, plastic sheet, and glass sheet. The photographs of the imaging experiments are shown in Figure 7b. All characters' patterns on different materials were almost invisible under daylight conditions. However, the as-printed characters on various materials became green and clearly visible under the excitation of the 980 nm laser. This shows that the LiYF<sub>4</sub>:Yb<sup>3+</sup>/Ho<sup>3+</sup> UCMPs with different Mg<sup>2+</sup> ion doping have good application potential in anti-counterfeiting recognition.



**Figure 6.** The upconversion fluorescence lifetime decay curves of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs doped with (a) 0 mol% and (c) 8 mol%  $\text{Mg}^{2+}$  at 543 nm. The upconversion fluorescence lifetime decay curves of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs doped with (b) 0 mol% and (d) 8 mol%  $\text{Mg}^{2+}$  at 657 nm.



**Figure 7.** (a) Demonstration of the screen-printing process and the anti-counterfeiting application

of 980 nm excitation. (b) Photographs of different kinds of anti-counterfeiting characters that were printed onto various materials with the composites as inks under the excitation of a 980 nm laser.

Figure 7a shows the screen-printing process of making anti-counterfeiting characters. Firstly,  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  micron crystal particles doped with a  $\text{Mg}^{2+}$  ion concentration of 8 mol% were selected as the raw material for anti-counterfeiting ink. The ground  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}/\text{Mg}^{2+}$  micron crystal powder was evenly mixed with mesh metal ink to make anti-counterfeiting ink, which was prepared at a ratio of 5:1 (samples:ink). Finally, the prepared anti-counterfeiting ink was poured onto the screen-printing template, and then brushed with a scraper. The specific anti-counterfeiting characters were successfully printed on the base material below the screen-printing template, and the upconversion luminous anti-counterfeiting characters were obtained.

#### 4. Conclusions

In summary,  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}$  UCMPs with different  $\text{Mg}^{2+}$  ion doping concentrations were synthesized and investigated. The emission peaks of  $\text{Ho}^{3+}$  ions were 480 nm, 535 nm, 543 nm, 638 nm, 657 nm, and 750 nm under the excitation of a 980 nm laser when different concentrations of  $\text{Mg}^{2+}$  ions were incorporated. The photoluminescence intensity reached its maximum when the  $\text{Mg}^{2+}$  ion doping concentration was 8 mol%. The prepared UCMPs were incorporated into a screen-printing agent. Characters were printed on different materials using the engraved screen-printing template. The printed characters were clearly visible under the irradiation of a 980 nm laser. The synthesized  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}/\text{Mg}^{2+}$  UCMPs have potential application prospects in anti-counterfeiting recognition.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/photonics9090614/s1>, Table S1: ICP-OES results of  $\text{LiYF}_4:\text{Yb}^{3+}/\text{Ho}^{3+}/\text{Mg}^{2+}$ .

**Author Contributions:** Experimental investigation and analysis, D.L. and J.W.; writing—original draft preparation, D.L.; validation, D.L. and J.W.; writing—review and editing, J.S. and D.L.; supervision, J.S. and C.W. All authors have read and agreed to the published version of the manuscript.

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