



Communication Relative Humidity Optical Sensor Based on Self-Assembled Gold Nanoparticles Covered with Nafion

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Abstract: The detection of humidity plays a vital role in healthcare, industrial, and scientific areas, and the development of an ideal sensor is in continuous progress. In this work, a relative humidity (RH) optical sensor based on localized surface plasmon resonance of self-assembled gold nanoparticles formed by thermal dewetting and coated with Nafion fluoropolymer is under study. Sensor performance has been found to substantially depend on Nafion layer thickness. The best sensing element—an array of gold nanoparticles covered with a 300 nm-thick Nafion—has been shown to possess a linear response in a wide dynamic range of 0–85% RH with a limit of detection down to 0.12%. Thus, a simple and low-cost method for high-accuracy RH detection has been demonstrated.

Keywords: gold nanoparticles; humidity sensor; Nafion; optical sensor

1. Introduction

The detection of relative humidity (RH) is a critical aspect of many industrial, commercial, and scientific applications, including environment monitoring, healthcare, agriculture, transportation, chemical and food processing, etc. Humidity may dramatically affect the performance of gas sensors due to cross-sensitivity [1,2]. This is why, for improving the accuracy of gas sensors, RH detection is very important, particularly for detection in the ppm or ppb range. There are several types of humidity sensors, including quartz crystal microbalance [3,4], capacitive [5,6], resistive [7–11], and optical ones, depending on the measured property. Moreover, self-powered RH sensors have been intensively studied in recent years [12–14]. Optical humidity sensors are typically more sensitive and immune to electromagnetic interference compared to their electronic counterparts. To date, a broad variety of optical humidity sensors based on photoluminescence [15–17], Brillouin scattering [18,19], Bragg grating [20–22], Fabry-Perot interferometer [23,24], photonic crystal [25,26], and plasmon resonance have been proposed. Plasmonic humidity sensors are widely studied because of the possibility of using self-assembled [27] and patterned [28] nanoparticle arrays possessing localized and lattice plasmon resonances, respectively. Moreover, surface plasmon-polariton waves in thin metal films [29] and hybrid plasmonic-photonic approaches [30,31] are suitable for optical RH detection.

Optical humidity sensors, as well as other types, require a functional material that efficiently adsorbs/desorbs water vapor from the atmosphere and changes in some way the properties of the material. The latter results in sensor responses depending on adsorbed water content. There are many functional materials used for optical humidity sensors, both inorganic (for example, ZnO [32], CsPbBr₃ [33], and carbon nanomaterials [34,35]) and organic materials (including polyvinyl alcohol (PVA) [36], chitosan [37], polyethylene glycol (PEG) [38], Nafion [39–42], etc.) Nafion is the trade name for sulfonated tetrafluoroethylene copolymer. It has a hydrophobic main chain and sidechains terminating in hydrophilic sulfonic ($-SO_3H$) groups that give the material its ion-transport properties. Being a fluoropolymer, it has excellent chemical, thermal, and mechanical stability. Nafion



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). membranes are highly conductive for cations, which are transported along the hydrophilic and negatively charged channels made of sulfonic groups. These channels, 3–6 nm wide, penetrate throughout the polymer, interconnect, and compose a significant volume fraction [43]. With these channels, Nafion readily adsorbs water molecules, which leads to an increase in volume (swelling) and conductivity as well as a change in refractive index. According to [24], Nafion's refractive index is 1.38 in the dry state and lowers as a result of water adsorption. Other reports on Nafion refractive index give 1.36 in the dry state [44] or 1.37–1.39 for sufficiently thick films [45]. The author of [46] observed that the refractive index of a 70-nanometer-thick Nafion film on SiO₂ substrate decreases from 1.338 in the dry state to 1.327 in the wet state; the latter is even lower than the water refractive index. Such discrepancies between different studies can be attributed to the method of Nafion film deposition, substrate interactions, and film thickness. Films with thicknesses in the tens of nanometer range differ by their structure from films thicker than 50 nm [43,45,47] and change water uptake amounts.

Electric sensors (resistive and capacitive) dominate the RH sensor market due to their low cost (starting at \$1), short response/recovery times (tens of seconds and below), moderate accuracy (typically in the range ± 1 –4%), and long-term stability. The disadvantages of such sensors are their susceptibility to strong electromagnetic interference, hysteresis (1–3%), and flammability. Requirements for humidity sensors strongly depend on potential applications; however, an ideal sensor should possess fast response/recovery times, a wide dynamic range, a low limit of detection, a linear response without hysteresis caused by different behavior during adsorption/desorption processes, and long-term stability. On the other hand, in real-world applications, there is typically a tradeoff between these parameters. The aim of this work was to develop a simple, low-cost, and robust optical humidity sensor for a wide range of potential applications. For this purpose, self-assembled gold nanoparticle arrays obtained via thermal dewetting of e-beam evaporated gold thin films were used as substrates due to the low-cost and versatile fabrication method and the high sensitivity of plasmonic structures to changes in the refractive index of an environmental medium. Nafion was chosen as a humidity-sensitive cover for the plasmonic gold nanoparticle array because of the large change in its refractive index in dry and humid air, as mentioned above, and its long-term stability.

2. Materials and Methods

Fused silica substrates ($12 \times 12 \times 1$ mm) from Siegert Wafer were cleaned in Piranha solution, washed with deionized water, and dried with a nitrogen gun. The substrates were then cleaned by corona discharge inside an e-beam evaporator (VATT-700, Ferri Vatt Ltd., Kazan, Russia), and a 6 nm gold thin film was deposited at 5×10^{-3} Pa vacuum. The obtained films were then heat-treated in a tube furnace at 400–500 °C to promote the formation of self-assembled gold nanoparticles and were characterized with a double-beam spectrophotometer (UV-3600, Shimadzu, Kyoto, Japan) in the spectral range 350–850 nm. A 5 wt.% Nafion 1100 W solution was purchased from Sigma Aldrich and employed as received. Nafion solution was spin-coated at speeds of 1200–10,000 rpm to obtain 130–300 nm thin films over gold nanoparticle arrays. The thickness of the deposited films was measured with a stylus profilometer (P-7, KLA Tencor, Milpitas, CA, USA). The topography of the samples after gold nanoparticle formation and Nafion coating deposition was measured by an atomic force microscope (NT-MDT).

The transmission spectra of the sample were measured at normal incidence using the setup shown in Figure 1. A collimated beam from a white light source (XWS-65, ISTEQ, Eindhoven, The Netherlands) passed through an aperture and a sample placed in a gas cell with quartz windows and was directed to a spectrometer with a TE-cooled CCD (iDus 401, Andor, Belfast, UK). Transmitted spectra were integrated for 10 s. A constant 400 sccm nitrogen gas flow was humidified with a controlled evaporation and mixing system (W-101A, Bronkhorst, Ruurlo, The Netherlands), injecting a controlled mass of water microdroplets and evaporating them in the nitrogen flow. The mass of injected water

and gas flow were controlled with liquid flow and mass flow controllers (Bronkhorst), respectively. The relative humidity of the output nitrogen flow was measured with an IPVT-08-D1 (Eksis, Moscow, Russia) commercial temperature and relative humidity sensor [48].

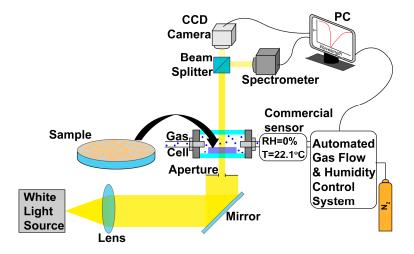


Figure 1. Scheme of the measurement setup. Transmission spectra of the samples were recorded in nitrogen gas flow with controlled RH.

3. Results and Discussion

3.1. Optimization of Heat Treatment Parameters

To promote the formation of nanoparticles on quartz substrates, thermal annealing of e-beam-evaporated thin gold films was used. Morphology and optical properties of such dewetted films determined by localized surface plasmon resonance (LSPR) of gold nanoparticles are known to strongly depend on substrate, thickness of the gold film, and thermal treatment parameters [49–51]. For optical sensor applications, the quality factor of a sensitive element is a key characteristic. That was why heat treatment parameters were varied to optimize the quality factor of the annealed gold films. The quality factor was calculated as a ratio of the full width at half maximum of the LSPR-related absorption band to its center wavelength. Based on literature data and preliminary experiments, the temperature range of 400–500 °C was chosen for further experiments. Heat treatment at 400 °C for 15 min was found to induce the formation of an LSPR band centered at 558 nm (Figure 2a). Prolonged annealing resulted in the narrowing and blue shift of the LSPR band as well as a decrease in the band amplitude. The same trends were observed for samples annealed at 450 °C and 500 °C (Figure 2b,c).

The full width at half maximum of the LSPR band of dewetted thin gold films substantially depends on particle size distribution, aspect ratio of nanoparticles, and their shape [51–54]. In our case, the same behavior was observed for all studied temperatures. The aspect ratio of nanoparticles should lower with the rising annealing temperature, resulting in a narrowing of the LSPR band due to a more homogeneous particle size distribution. Growth of average nanoparticle size with continuation of annealing, typical for dewetted gold films [55], led to an increment in the distance between neighboring nanoparticles and a decrease in the area covered with gold nanoparticles [51]. The latter two are reasons for the blue shift of the LSPR band and some increase in transmission, respectively.

Figure 2d shows the dependence of the quality factor on annealing time at different temperatures. The quality factor of thin films annealed at 400 °C and 450 °C did not exceed 6, while it reached 7 after annealing at 500 °C for 90 min, with a slight decrease with further treatment. Due to high optical losses in noble metals, typical quality factor magnitudes of LSPR in such nanostructures are on the order of ten [56–58]. Based on the obtained results, gold nanoparticle arrays annealed at 500 °C for 90 min were chosen for further deposition of Nafion films.

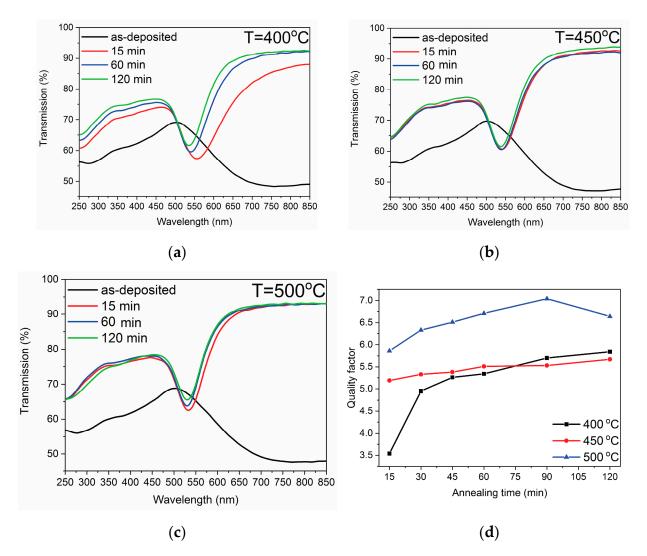


Figure 2. Transmission spectra of gold thin films heat-treated at (a) 400 °C, (b) 450 °C, and (c) 500 °C for different times. (d) Dependence of the LSPR quality factor on heat treatment time at different temperatures.

3.2. Nation Deposition and Characterization

The thickness of humidity-sensitive material has a significant effect on the performance of optical humidity sensors [59,60]. To study the effect of Nafion thickness, three samples annealed under the chosen optimal conditions were prepared for Nafion deposition. The thicknesses of the Nafion films were found to be 130 nm (Sample A), 190 nm (Sample B), and 300 nm (Sample C). Transmission spectra before and after Nafion deposition are shown in Figure 3. Deposition of Nafion led to a long-wavelength shift of the LSPR band due to an increase in the refractive index of the environment. Moreover, enhancement of LSPR band amplitude was also detected for samples B and C covered with 190 and 300 nm Nafion films, respectively. For the thinnest Nafion sample deposited at 10,000 rpm, weakening of the LSPR amplitude was observed. The latter was most likely due to removing some gold nanoparticles during Nafion deposition at such a high rate due to poor adhesion of gold to the quartz substrate.

To study the topology of the samples surfaces, they were characterized with AFM. Figure 4a displays a typical surface profile observed after dewetting of the gold film at 500 °C for 90 min. The surface contains closely packed, self-assembled gold nanoparticles with a height below 25 nm. Very smooth surfaces with RMS roughness below 1 nm were observed after Nafion deposition (Figure 4b–d).

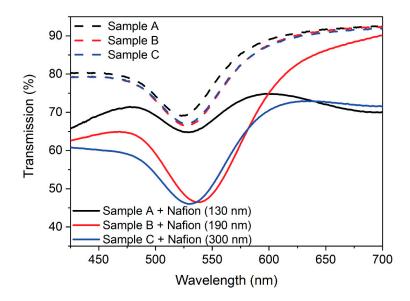


Figure 3. Transmission spectra of gold thin films annealed at 500 °C for 90 min (dash lines) and covered with Nafion (solid lines).

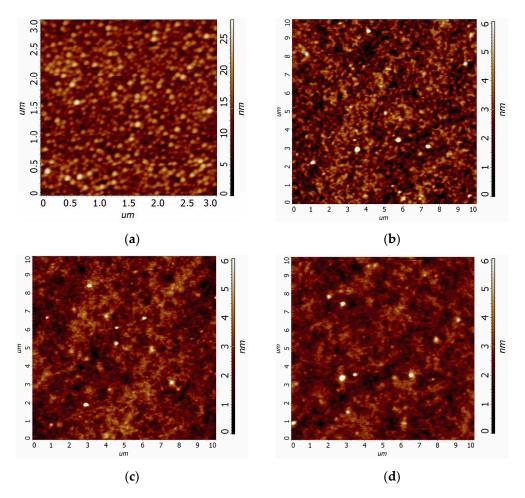


Figure 4. (a) Typical AFM image of gold self-assembled nanoparticles after heat treatment at 500 $^{\circ}$ C for 90 min; AFM images of the samples A (b), B (c), and C (d).

3.3. Relative Humidity Detection

Transmission spectra of the fabricated samples covered with Nafion films of different thicknesses were recorded during a controlled change of humidity in nitrogen gas flow.

Selected spectra obtained at varied RH in the range of 0-85% are shown in Figure 5a-c. Sample A possessed a very weak dependence of the LSPR band location on humidity, while the most prominent changes in spectra were observed in the range of 600-750 nm due to a shift of interference fringes resulting from swelling Nafion film with humidity increase. Sample B showed no interference fringes, and only a weak long-wavelength shift of the LSPR band was observed. The LSPR band location in Sample C was also red-shifted with rising humidity levels. A conventional characteristic allowing direct comparison of various plasmonic RH sensors is a shift of the SPR band normalized to RH change. The dependence of the LSPR spectral shift on humidity is shown in Figure 5d, and it reveals a non-linear character for all samples. It should be noted that Sample A showed a blue shift, while the other samples possessed a red shift of the LSPR band with an increase in RH. The absolute average LSPR shifts over the 0–85% RH range were 0.020 nm/% RH, 0.024 nm/% RH, and 0.097 nm/% RH for samples A, B, and C, respectively. The obtained values are comparable with some publications (see [61-64] and tables therein), but far below the top results exceeding 2 nm/% RH for fiber optic humidity sensors [39,65,66] and reaching 5 nm/% RH for a side-polished SPR-based optical fiber sensor [29]. Most SPR-based sensors require a high-resolution spectrometer for the detection of SPR band shifts with changes in RH. This leads to an increase in fabrication costs and limits the miniaturization of the device.

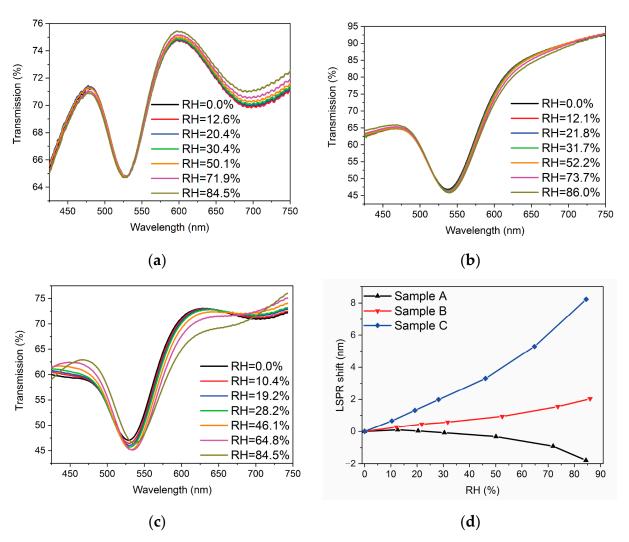


Figure 5. Transmission spectra of the samples with Nafion thicknesses of 130 nm (**a**), 190 nm (**b**), and 300 nm (**c**) measured at different RHs. (**d**) Dependence of the LSPR shift on RH for samples covered with different Nafion thicknesses.

Another typical method to detect humidity levels is the measurement of transmitted power in some spectral range [67,68]. To identify the spectral range with the most significant changes, differential spectra obtained from transmission spectra at variable RH by subtracting the spectrum of the corresponding sample in dry nitrogen were calculated (Figure 6a–c). In the case of Sample A, characterized by the lowest shift of the LSPR band with humidity, the largest response was observed in the red part of the visible spectrum, reaching an increment of transmission around 1.4% with a growth RH from 0% up to 84.5%. However, the transmission changes were mostly caused by the shift of interference fringes, with additional contributions from the LSPR band in the range between 525 nm and 625 nm. Samples B and C possessed maximal changes in transmission, around 3.5% and 7%, respectively, in the spectral range 550–600 nm associated with the long-wavelength shift of the LSPR band.

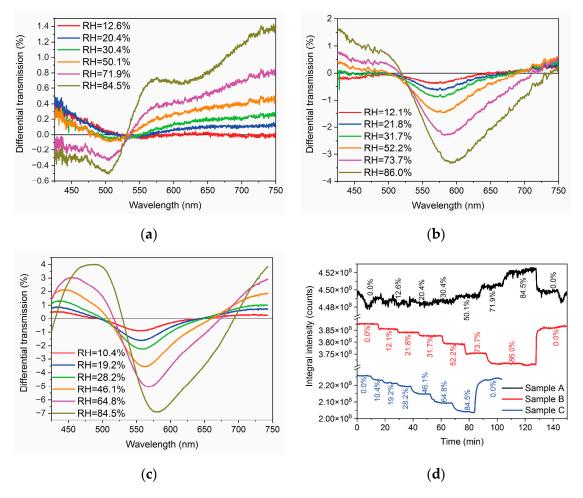


Figure 6. Differential transmission spectra of samples A (**a**), B (**b**), and C (**c**) measured at different RHs. (**d**) Temporal evolution of integral intensity in the spectral range 550–600 nm for the samples.

Figure 6d shows the temporal dependence of the change in integral intensity registered in the spectral range 550–600 nm (in the case of a sensor prototype, this value would be proportional to the optical power registered with a conventional photodetector). Due to minor changes in the transmission of Sample A, the response was undetectable in the range of RH below 30%. Moreover, the sample did not show a reversible response since the signal at dry nitrogen after being subjected to a wet atmosphere significantly differed from the initial one. On the other hand, the samples covered with thicker Nafion films possessed a linear response in all the studied RH ranges from 0% to 85%, with linearities of 98.3% and 99.7% (Figure 7). The limit of detection for sensors is typically calculated as $3\sigma/S$, where σ is the standard deviation calculated in our case at dry nitrogen over 5 min, and S stands for sensitivity, defined as the slope of the response curve. The limits of detection of the samples were found to be 1.8% and 1.1% for samples B and C, respectively. The amplitude of the response for a change in RH from 0% to 85% was about 20% greater for Sample C compared to Sample B.

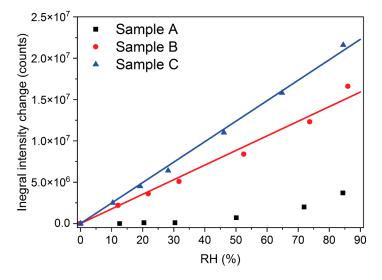


Figure 7. Dependence of absolute changes in integral intensity of transmitted light in the spectral range 550–600 nm on RH. Lines represent a linear fitting of the experimental data for samples B and C.

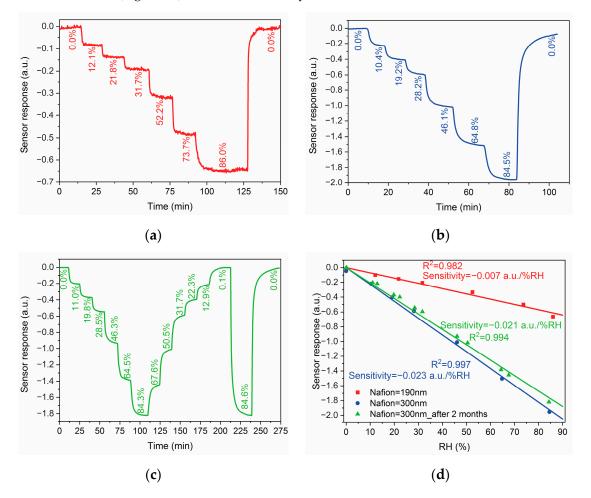
The limit of detection of the proposed sensitive elements can be enhanced by increasing the signal-to-noise ratio. The response curves shown in Figure 6d contained noise from the light source, detector, and other factors like mechanical vibration, for example. The noise of the detector was significantly reduced by a relatively long integration time of 10 s. To decrease noise from the light source related to temporal instability of emitted power, the remaining part of light (out of the 550–600 nm range related to the SPR band) transmitted through the sample can be detected in the second optical channel. Thus, the sensor response can be defined through the measured integral intensity (power in the case of a device) of transmitted light detected in two channels with the following expression:

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SR = ([I_{550-600} / (I_{425-550} + I_{600-750})]_{RH=Z\%} - [I_{550-600} / (I_{425-550} + I_{600-750})]_{RH=0\%}) \times 100
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where SR is the sensor response, I_{X-Y} is integral intensity of transmitted light in the wavelength range from X to Y nm, Z is the measured RH value.

Temporal dependencies of the sensors response for Samples B and C are shown in Figure 8a,b. Examples of the measured spectra of transmitted light at variable RHs are presented in Figure S1 in the Supplementary Materials. Dependence of the sensors response on humidity level also revealed linear character with sensitivity equaled to -0.007 a.u./% RH and -0.023 a.u./% RH (Figure 8d). Due to increase in the signal to noise ratio, the limit of detection of samples B and C was improved to 1.2% and 0.12% RH respectively.

As mentioned above, the long-term stability and hysteresis-free operation of a humidity sensor are important parameters for real applications. To analyze these characteristics, response of Sample C was measured after being stored for 2 months at ambient conditions. The sensors showed excellent stability as follows from the results presented in Figure 8c,d. Sensitivity of the stored sensor was found to be -0.021 a.u./% RH with limit of detection 0.13% RH that are in agreement with characteristics of as-fabricated sensor. Weak decrease in response amplitude of the Sample C after 2 months may be related to some inhomogeneity of the sample. The results of the experiment also showed almost hysteresis-free



work with maximum difference at high humidity around 0.5% between linear fit curves (Figure S2), that below accuracy of the reference sensor.

Figure 8. Sensor response vs. time with change of RH for samples B (**a**), C (**b**), and C after 2 months (**c**). (**d**) Dependence of the response amplitude on RH.

Response (t_{90}) and recovery (t_{10}) times were estimated with a change in RH from 0% to about 85% and vice versa. Recovery times were calculated to be 150 s and 400 s for as-fabricated samples B and C, respectively. Sample C after 2 months showed a response time of 320 s and a recovery time of 400 s.

The possible implementation of the proposed sensor is presented in Figure S3. The sensor includes a white light source that can be a conventional, low-cost white LED, a gas cell with the sensitive element based on a self-assembled gold nanoparticle array covered with Nafion, a Bragg mirror reflecting light in the 550–650 nm wavelength range, two Si photodetectors, and electronics to divide output signals from the photodetectors to each other. Moreover, optical fiber implementation of the proposed sensor is also possible with the deposition of gold on bare optical fibers with subsequent annealing and deposition of Nafion, similar to other SPR-based optical fiber humidity sensors.

A comparison of the performance characteristics of the developed sensor with those of other optical humidity sensors is presented in Table 1. As mentioned in the introduction, the pursuit of significant improvement in one parameter often results in the worsening of the others. That is why sensors with fast response and a low limit of detection are usually characterized by a narrow dynamic range, and vice versa. Our results show that a very simple and low-cost optical humidity sensor with a wide dynamic range, a linear response, and a low limit of detection can be successfully realized with a low cost of response/recovery time. The proposed sensor may find applications where high accuracy of RH detection is required but there is no need for a fast response; for example, calibration or correction of other gas sensor readouts can be performed. However, it is worth noting that further experiments on the sensor's deterioration during cycles of humidifying/drying in various atmospheres, studies on its temperature sensitivity, etc. must be performed.

Functional Dynamic Limit of **Response/Recovery** Reference Material Range Detection Time (RH Range) 115 ms (50→90%) PDDA/PSS 60-80% 0.007% [69] 200 ms (90→50%) Nafion 45-80% 1% 200 ms (50→67%) [42] Agarose 40 - 80%0.7% $1 \, s/4 \, s$ [70] 140 s (0→1%) SU-8 0-50% 0.03% [71] 40 s (1→0%) Polyimide 20-95% 5% 180 s (55→60%) [72] 7.4 s (35→85%) 35-85% [73] Agarose 2.6% 8.0 s (85→35%) 370 ms (50→90%) SnO₂ 20-90% 0.04% [74]380 ms (90→50%) Tungsten 4.5 s (20→80%) 2-98% 5% [75] filament 5.0 s (80→20%) Nafion This work 0-85% 1.2% 150 s (85→0%) 320 s (0→85%) Nafion 0-85% 0.12% This work 400 s (85→0%)

Table 1. Comparison of the obtained results with the selected published works on optical humidity sensors.

4. Conclusions

Optical humidity sensors based on LSPR of self-assembled gold nanoparticles obtained by thermal annealing of gold thin films and covered with Nafion thin films were fabricated and studied. With simple measurement of transmitted light integral intensity in two optical channels, the proposed sensor was shown to possess a reversible linear response in the wide dynamic range from 0% to 85% RH and an excellent limit of detection down to 0.12%, even with the relatively low absolute spectral shift of LSPR 0.097 nm/% RH. Further improvement of the sensor characteristics is possible with an increase in the quality factor of the gold nanoparticle array by repeated cycles of thin gold deposition and annealing [55] or periodic nanostructures [69,70]. We believe that the results of our study will not only provide a new convenient pathway for the design of gas sensors but also demonstrate a new concept for the development of LSPR-based humidity detection.

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/photonics10090975/s1. Figure S1: Transmitted light spectra measured at different relative humidity levels in a nitrogen atmosphere for Nafion film thicknesses of 130 nm (a), 190 nm (b), and 300 nm (c). The spectra were used to calculate the sensor response. Figure S2: Response of Sample C for 2 months stored in ambient conditions to RH from dry to humidified nitrogen and in the reverse direction. Figure S3: Schema of the proposed optical humidity sensor based on LSPR of self-assembled gold nanoparticles covered with Nafion thin film.

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Conflicts of Interest: The authors declare no conflict of interest.

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