

## Article

# Extraction, Separation and Purification of Bioactive Anticancer Components from *Peganum harmala* against Six Cancer Cell Lines Using Spectroscopic Techniques

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**Abstract:** Conventional cancer treatments normally involve chemotherapy or a combination of radio- and chemotherapy. However, the adverse effects of synthetic medicines encouraged the exploration of novel therapeutic medications of a bio-friendly nature. In an effort to explore anticancer compounds from natural resources, crude extract of *Peganum harmala* (seeds) was fractionated on the basis of polarity, and the fractions were further tested for anticancer activity. Brine shrimp lethality assays and potato disc antitumor assays were used to test each fraction for cytotoxic and antitumor potential. The ethyl acetate fraction was found to be most potent, with LC<sub>50</sub> and IC<sub>50</sub> values of 34.25 µg/mL and 38.58 µg/mL, respectively. Further activity-guided fractionation led to the isolation of the bioactive compound PH-HM-10 which was identified and characterized by Mass Spectroscopy (MS), Infrared Spectroscopy (IR), Proton Nuclear Magnetic Resonance Spectroscopy (<sup>1</sup>HNMR), Carbon Nuclear Magnetic Resonance Spectroscopy (<sup>13</sup>CNMR) and Heteronuclear Single Quantum Correlation (HSQC). Anticancer aspects in the isolated compound were determined against six human cancer cell lines with a maximum anticancer effect (IC<sub>50</sub> = 36.99 µg/mL) against the tested human myeloid leukemia (HL-60) cell line, followed by the human lung adenocarcinoma epithelial cell line (A549) and the breast cancer cell line (MCF-7) with an IC<sub>50</sub> of 63.5 µg/mL and 85.9 µg/mL, respectively). The findings of the current study suggest that the isolated compound (Pegaharmine E) is significantly active against the tested cancer cell lines and can be further investigated to develop future novel anticancer chemotherapeutic agents.

**Keywords:** *Peganum harmala*; anticancer; cell lines; compound isolation; NMR

## 1. Introduction

Cancer is a primary public health crisis for mankind. Cancer patients frequently experience various unpleasant side effects from chemotherapy and radiotherapy [1]. Although plants and plant-based medication were used for centuries, the toxicity issues linked with synthetic chemotherapeutic agents further increased the interest of the scientific community in this field. Another reason for using plant derivatives in therapeutic applications is their availability, potentiality and low cost in comparison with modern therapeutic medicines [2]. Almost 20% of plants located in different regions worldwide were tested biologically or pharmacologically, with considerable proportions being introduced into the market as new medicines [3]. Currently, about 50% of therapies in use are directly derived from plants

and 25% of the prescribed drugs have their source in tropical plants. These noteworthy attributes improved their importance as precursor substrates for the development of other drugs [4].

After a twenty-year hiatus, natural product research is now assuming fresh prominence. Natural products are a considerable source of useful active compounds. Modern analytical and genetic techniques such as metabolomics, molecular docking, and molecular networking, etc., make it significantly easier to find compounds for various uses—for example, antibiotics and pesticides. Equipped with a collection of sensitive and quick bioassays and analytical procedures, the natural product researcher is now more able than ever to explore nature's huge frontier of bioactive chemical wonders [5]. The bioassay-guided isolation approach to obtain bioactive compounds is time consuming and involves exhaustive efforts compared with non bioassay-guided isolation; this is because it requires testing activity for each individual fraction to trace the most active one. Additionally, active fractions may present in very minute quantities—too small for spectroscopic analysis and bioassays. This approach is still a basic but economical practice for characterizing natural products with distinct biological potential. Bioassay-guided isolation, however, can trace the most potent compounds responsible for the bioactivity of an extract. This approach also led to the development of methods for the isolation of active compounds. This approach provided a new recipe and established a protocol for other researchers [6].

Modern anticancer therapeutics are also vital for minimizing the numerous complications faced by cancer patients, but the prevalence of drug resistance resulted in the development of a growing interest in natural products [7,8]. *Peganum harmala* is an herbaceous, perennial plant with many reported pharmacological activities: it is carminative, diuretic, antithrombotic and analgesic. It also demonstrated numerous medicinal effects and shows antidiabetic, cardiovascular, neurologic, antimicrobial, gastrointestinal, insecticidal, antineoplastic and antiproliferative effects [9]. The present study assessed the cytotoxic and antitumor activity of different fractions of *P. harmala*, followed by isolation and characterization of the bioactive compounds from the most bioactive fraction showing potential anticancer activity.

## 2. Material and Methodology

### 2.1. Sample Preparation

Seeds of *P. harmala* were washed, rinsed and air dried at a temperature of  $25 \pm 2$  °C under shade by spreading in thin layers. Dried plant material was then homogenized to fine powder and stored for further utilization.

### 2.2. Fractionation

Fractionation was performed through the suspension of extracts in 250 mL water, followed by separation using organic solvents, including hexane, chloroform, ethyl acetate and methanol in a separating funnel, and by changing the polarity. Methanol was separated from the aqueous fraction by simple distillation [10]. Each of the fractions was dried through rotary evaporation of the solvent and stored at 4 °C for future analysis [11].

### 2.3. Brine Shrimp Lethality Test (BSLT)

The *Artemia salina* lethality bioassays were performed in accordance with the methodology proposed by Meyer et al., [12] to discover the toxicity of the plant extracts. The assays were performed in 0.45 µm multiwell plates. Seawater (5 mL) was poured into individual wells as the saltwater solution and evaporated. These bioassays were conducted in a temperature-controlled room at 28 °C under a continuous light regime. Various concentrations (1000–15.625 µg/mL) of fractions were tested with vincristine, potassium dichromate and etoposide as the positive controls. Hatched nauplii (10 per vial, with 12 h age) were exposed to *P. harmala* extract for 12 and 24 h. The mortality (100%) of the nauplii was calculated for those receiving treatment and the controls, through the given formula:

$$\text{Mortality (\%)} = \text{Survival in control (\%)} - \text{Survival in treatment (\%)}$$

Logarithmic regression analysis was performed to determine the  $LC_{50}$  values for various fractions and the positive controls used.

#### 2.4. Antitumor Assays

The antitumor assays were executed using the standard procedure of Yildirim et al. [13]. *Agrobacterium tumefaciens* was cultured on Yeast Extract Media (YEM) and a 48 h old culture was used to test the samples. Different concentrations 15.625–1000  $\mu\text{g}/\text{mL}$  of all fractions of *P. harmala* were used to determine antitumor activity, while vincristine and etoposide were used as positive controls. Briefly, red-skinned potatoes were surface-sterilized with 0.1%  $\text{HgCl}_2$  solution and disks were prepared. A total of 10 discs were placed in each Petri plate and 50  $\mu\text{L}$  inoculum was poured on each disc and incubated for 21 days in the dark at 28 °C. After the incubation period (21 days), the potato discs were stained with Lugol's solution (10% KI, 5%  $\text{I}_2$ ). Tumor inhibition was calculated using the following formula:

$$\text{Tumor inhibition (\%)} = (1 - \text{Number of tumors in the sample} / \text{Number of tumors in control}) \times 100$$

#### 2.5. MTT Cytotoxic Assays

The MTT assays were performed to determine the cytotoxic effects of different concentrations of the isolated compound against 6 human cancer cell lines HL-60, PC-3, SGC-7901, MCF-7, HCT116 and Lung A549 [14]. After exposure to different concentrations of bioactive compounds, the metabolically active cells were determined by the intensity of purple color (formazan product), and quantitative assessment was conducted using a spectrophotometer at the 590 nm wavelength [15]. Cell viability (%) and inhibition (%) were calculated using the following formula:

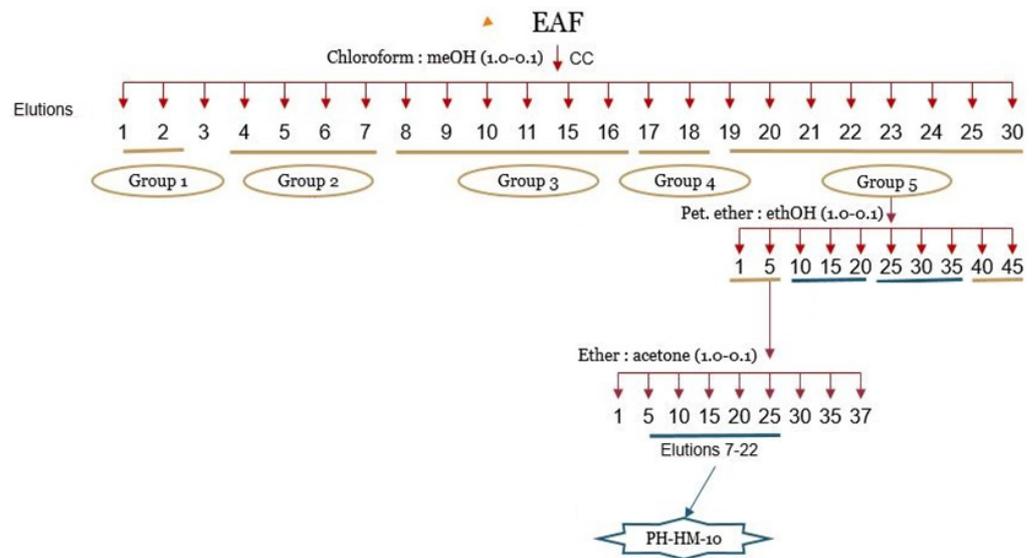
$$\text{Cell viability (\%)} = (\text{Absorbance of treated cells} / \text{Absorbance of cells with vehicle solvent}) \times 100$$

$$\text{Percentage inhibition} = 100 - \% \text{ cell viability}$$

#### 2.6. Isolation and Characterization of Compound (PH-HM-10)

Column chromatography was used for the purification of the bioactive components [16]. The most bioactive fraction, the ethyl acetate fraction, was further purified by a silica gel column sequentially eluted with a stepwise gradient of increasing solvent. Different components eluted by the column were analyzed using TLC. Isolated fractions were examined on a TLC plate using the solvent-vapor-saturated TLC chamber, and air dried in a fume hood to visualize spots within 1–2 min. TLC plates were visualized at UV-254 nm. Elutions showing the same patterns were pooled together and used for future analysis [17]. Being the most active, Group No. 5 was processed for HPLC analysis with a photodiode array detector and a column of 150  $\times$  4.6 mm. Petroleum ether as mobile phase A and ethanol as mobile phase B were used with a 0.5 mL/min flow rate. Analysis was performed as gradient with mobile phase A (90%), decreasing to 10% for a 5 min period; the column was equilibrated to initial conditions after the elution was completed. Elute with a concentration of 22.75  $\mu\text{g}/\text{mL}$  ( $IC_{50}$  value) was subjected to HPLC analysis. A total of 45 eluents were collected in vials, rotary evaporated and then subjected to thin-layer chromatography. Finally, the eluents 1–6 were also subjected to chromatography with ether and the ethyl acetate solvent mixture; 37 eluents were collected in a vial and elutions 7–22 yielded a white amorphous powdered form of the purified compound PH-HM-10. A summary of the isolation of compound PH-HM-10 is given in Figure 1.

The isolated components were characterized by spectroscopic techniques, specifically, Mass spectra (MS), Infrared (IR) Spectroscopy and Nuclear Magnetic Resonance Spectroscopy (NMR) [18].



**Figure 1.** Scheme used for the isolation of PH-HM-10. EAF = Ethyl acetate fraction; meOH = Methanol; ethOH = Ethanol.

### 3. Results

#### 3.1. Brine Shrimp Cytotoxicity

The cytotoxicity of all the tested fractions (n-hexane, chloroform, ethyl acetate, methanol and aqueous) indicated a dose-dependent effect after 12 h and 24 h of exposure (Figure 2). The n-hexane and ethyl acetate fractions showed 100% mortality in the brine shrimps after 12 h and 24 h of exposure at concentrations of 1000 µg/mL and 500 µg/mL. At a concentration of 250 µg/mL, 91 and 92% mortality were observed after 12 h and 24 h of treatment for the hexane extract, and 75 and 81% mortality were recorded for the ethyl acetate fraction. Similarly, the chloroform and methanol fractions of *P. harmala* seeds also exhibited 100% mortality in brine shrimp nauplii. At a concentration of 500 µg/mL of chloroform fraction, 90 and 98% mortality in brine shrimp larvae were observed after 12 and 24 h of exposure, while 90 and 92% mortality were observed for the methanol extract of *P. harmala* (s) at a concentration of 500 µg/mL. The aqueous fraction of the plant was found to be the least toxic with a maximum of 90 and 92% cytotoxic effects at a concentration of 1000 µg/mL after 12 and 24 h of treatment. However, the cytotoxicity decreases significantly at concentrations of 500 µg/mL to 31.25 µg/mL, indicating 67 to 24% and 67 to 30% after exposure to 12 and 24 h of treatment.

Of the five fractions of *P. harmala* seed extract, the ethyl acetate fraction was found to be the most active with an LC<sub>50</sub> value of 34.25 µg/mL, followed by methanol and hexane fractions with IC<sub>50</sub> values of 38.14 and 40.66 µg/mL, respectively (Table 1). The results of the study correspond to the findings of Khan et al., [19] who reported the cytotoxicity of the n-hexane extract of *P. harmala* seed extract. The chloroform and aqueous fractions showed LC<sub>50</sub> values of 42.63 and 74.29 µg/mL, respectively, in comparison with the positive control vincristine sulphate (LC<sub>50</sub> = 2.28 µg/mL), etoposide (LC<sub>50</sub> = 3.49 µg/mL) and potassium dichromate (LC<sub>50</sub> = 16.55 µg/mL). The findings of this study accord with the reported literature that potassium dichromate, etoposide and vincristine sulphate are extremely toxic as per Clarkson’s lethality criterion [20]. The LC<sub>50</sub> values for different concentrations of *P. harmala* seed extract fall within the category of bioactive metabolites, confirming their use as pharmacological agents [21].

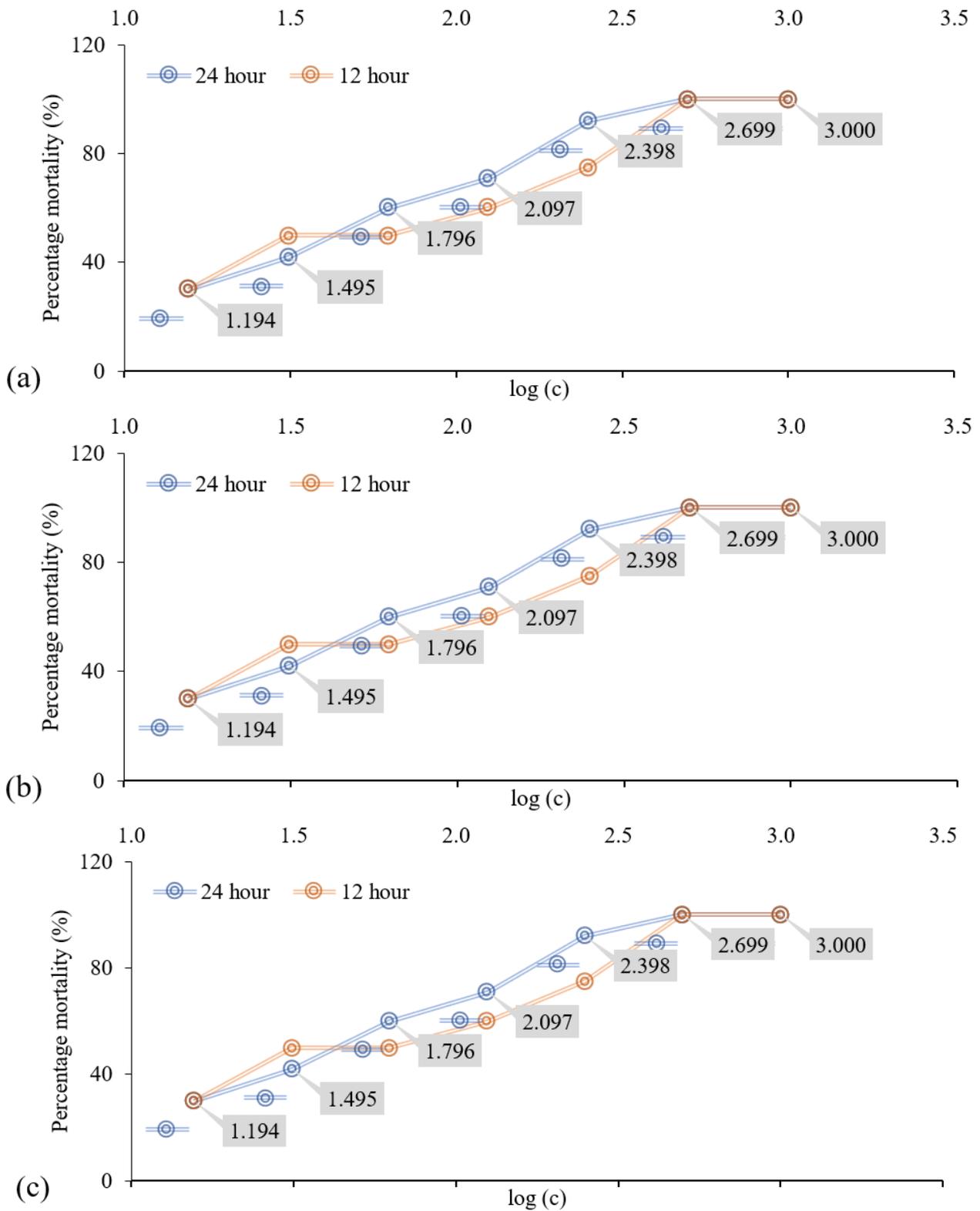


Figure 2. Cont.

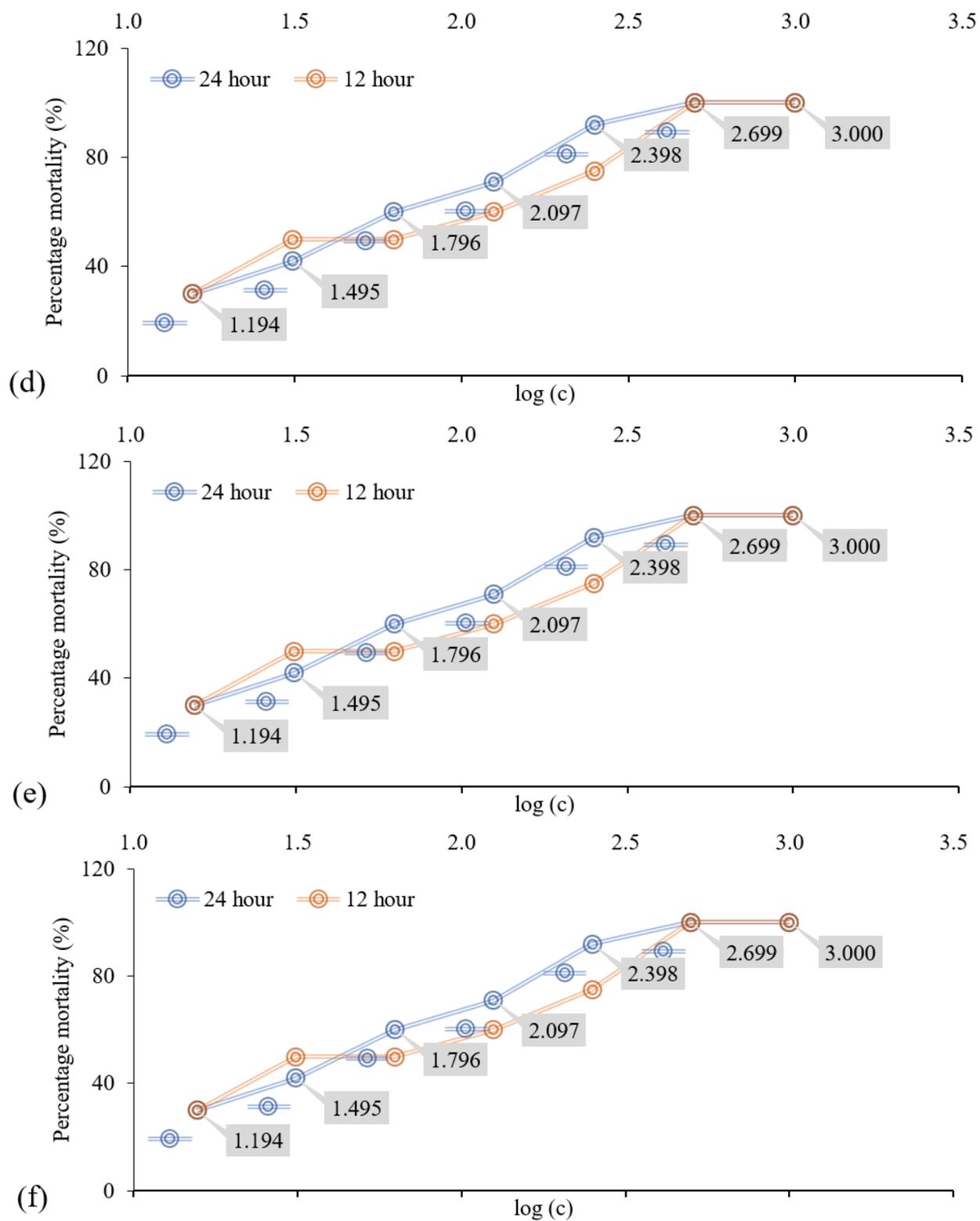
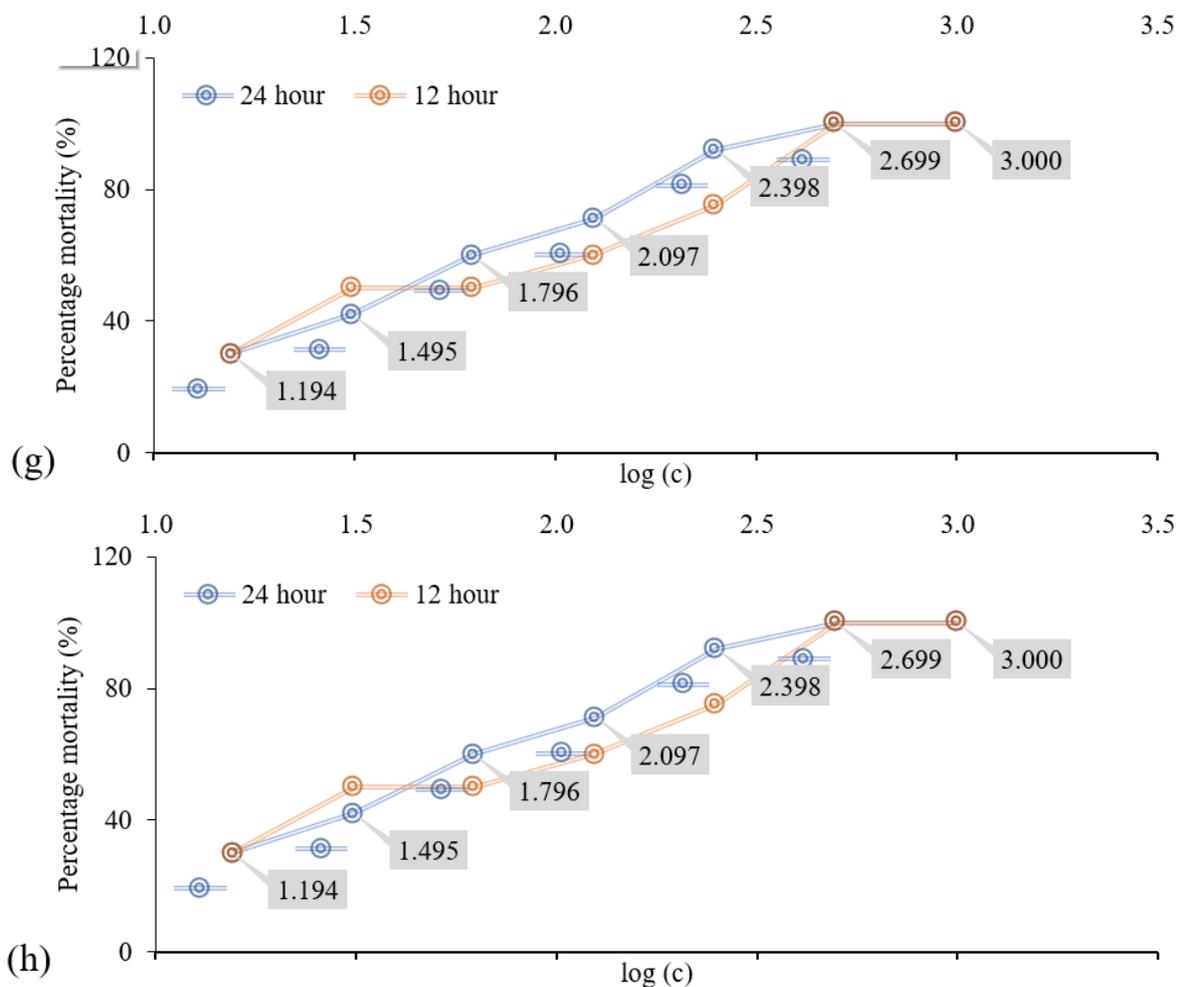


Figure 2. Cont.



**Figure 2.** Cytotoxicity of (a) n-hexane, (b) ethyl acetate, (c) chloroform, (d) methanol, (e) aqueous, (f) vincristine sulphate, (g) etoposide and (h) potassium dichromate against logarithmic concentration after 12 and 24 h of exposure.

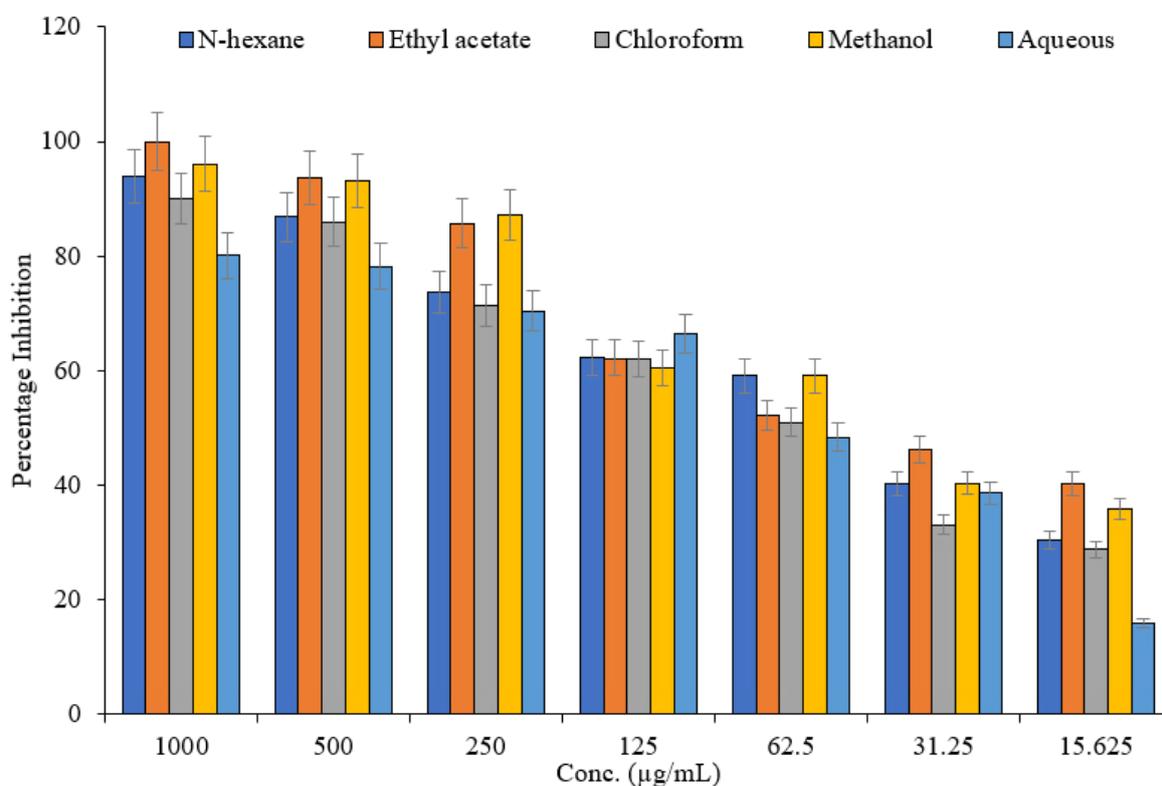
**Table 1.** Cytotoxicity of different fractions of *P. harmala* (S) extract; LC<sub>50</sub> and R<sup>2</sup> values determined through logarithmic regression analysis.

Sample	LC <sub>50</sub> µg/mL	Regression Equation	R <sup>2</sup>
n-hexane	40.66	$y = 18.446\ln(x) - 18.348$	0.961
Ethyl acetate	34.25	$y = 15.56\ln(x) - 4.9881$	0.956
Chloroform	42.63	$y = 17.931\ln(x) - 17.289$	0.947
Methanol	38.14	$y = 15.767\ln(x) - 7.4118$	0.977
Aqueous	74.29	$y = 17.57\ln(x) - 25.69$	0.962
Vincristine	2.28	$y = 9.1714\ln(x) + 42.432$	0.925
Etoposide	3.49	$y = 9.5836\ln(x) + 38.013$	0.951
K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	16.55	$y = 12.366\ln(x) + 15.293$	0.950

### 3.2. Antitumor Activity of *P. harmala* Fractions

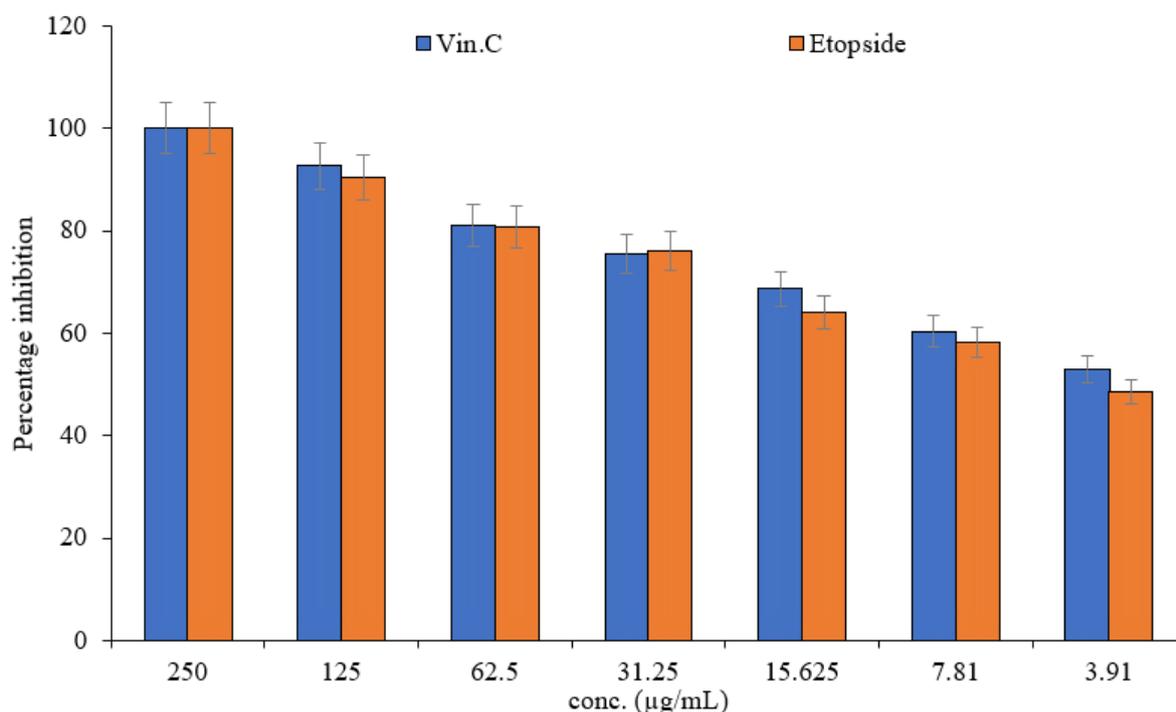
The results for the antitumor activity of *P. harmala* extract indicate dose-dependent antitumor activity for all fractions. The ethyl acetate fraction exhibits the highest percentage of tumor inhibition, i.e., 100% at a concentration of 1000 µg/mL, followed by 93.74% tumor growth inhibition at a concentration of 500 µg/mL. At lower concentrations, i.e., 250,

125, 62.5, 31.25 and 15.625  $\mu\text{g}/\text{mL}$ , 85.69, 62.19, 52.14, 46.33 and 40.28%, inhibition of tumor galls was observed. The n-hexane fraction exhibits 93.85 and 86.83% inhibition and 10 induced crown gall tumors at dosage concentrations of 1000 and 500  $\mu\text{g}/\text{mL}$ , respectively. The inhibition potential fell from 73.66 to 30.37% with a decrease in concentration from 250 to 15.625  $\mu\text{g}/\text{mL}$ . This indicates that the inhibition of tumor development is strongly correlated with the concentration of the extract (Figure 3). The same trend was observed throughout the antitumor assay among all the tested fractions. Similarly, the chloroform fraction also indicates strong potential to control tumor growth in a dose-dependent manner, with its best activity being 90.06% inhibition at 1000  $\mu\text{g}/\text{mL}$  of extract concentration. At the lowest concentration of chloroform extract, i.e., 15.625  $\mu\text{g}/\text{mL}$ , 28.84% inhibition of tumor development was observed. The methanolic fraction also exhibits 96.10% control of tumor development at 1000  $\mu\text{g}/\text{mL}$ , indicating assimilation of bioactive principles with antitumor activity in this solvent. The aqueous fraction of *P. harmala* seed extract exhibits the least activity among all fractions, with 80.18% inhibition of tumor growth at 1000  $\mu\text{g}/\text{mL}$  concentration, indicating the nature and solubility of bioactive components.



**Figure 3.** Percentage inhibition of tumors by different fractions of *P. harmala* (S) extract at different concentrations.

Positive controls were significantly able to inhibit crown gall tumors completely (100% mortality) at a range of concentrations between 1000 and 250  $\mu\text{g}/\text{mL}$  (Figure 4). The current bioassays show no substantial difference was detected in the bioactivity of vincristine and etoposide, although minor variations in the percentage inhibition were noticed along different concentrations. The 7.81  $\mu\text{g}/\text{mL}$  concentrations of both positive controls were capable of more than 50% inhibition of crown gall tumors. A minimum tumor percentage inhibition of 52.84% was observed against vincristine sulphate at a concentration of 3.91  $\mu\text{g}/\text{mL}$ , while the same concentration of etoposide exhibited 48.55% inhibition of crown gall tumors.



**Figure 4.** Percentage Inhibition of tumors by positive controls at different concentrations.

IC<sub>50</sub> values were calculated through logarithmic regression analysis for all the five fractions of *P. harmala* (S) extracts to check their crown gall tumor inhibition potential. The difference in the IC<sub>50</sub> values of the five different fractions of the *P. harmala* (S) extracts indicate a clear difference in their potential to inhibit crown gall tumors. The minimum IC<sub>50</sub> value (38.58 µg/mL) was shown by the ethyl acetate fraction, while the maximum IC<sub>50</sub> (81.36 µg/mL) was noted for aqueous fractions, indicating the lowest effectiveness. The chloroform fraction indicated an IC<sub>50</sub> value of 65.3 µg/mL, while the methanol fraction was able to inhibit 50 % tumor growth at a concentration of 42.23 µg/mL. The positive controls, vincristine sulphate and etoposide, exhibited 50% inhibition of tumor formation at 3.14 µg/mL and 4.31 µg/mL, respectively (Table 2).

**Table 2.** Antitumor activity IC<sub>50</sub> and R<sup>2</sup> values determined through logarithmic regression analysis.

Samples	Regression Equation	R <sup>2</sup> Value	IC 50
N-hexane	$y = 15.36\ln(x) + 10.917$	0.9837	50.94
Ethyl acetate	$y = 15.847\ln(x) + 14.078$	0.9596	38.58
Chloroform	$y = 15.966\ln(x) + 5.408$	0.9830	65.30
Methanol	$y = 16.206\ln(x) + 11.725$	0.9483	42.43
Aqueous	$y = 15.163\ln(x) + 4.6878$	0.9188	81.36
Vincristine	$y = -4.042\ln(x) + 22.582$	0.9945	3.14
Etoposide	$y = -4.353\ln(x) + 24.316$	0.9945	4.31

### 3.3. Isolation and Characterization of Compounds

#### 3.3.1. Isolation of PH-HM-10

A silica gel column (Sephadex, 75 × 3) packed in chloroform was used for the chromatography of the ethyl acetate fraction of *Peganum harmala*. A total of thirty elutions were collected in a conical flask with chloroform and methanol solvents as the mobile phase, with a gradual change in the polarity of the mobile phase (100:0, 90:10, 80:20, 70:30, 60:40, 50:50, 40:60, 30:70, 40:60, 30:70, 20:80, 10:90, 100:0). All the eluents were vaporized to

dryness using a rotary evaporator at 40 °C under reduced pressure and then subjected to thin layer chromatography using chloroform: methanol as the solvent system. Based on TLC profiling, the elutions were combined into Groups 1 to 5. These five groups were tested for their bioactivity; Group 5 was the most cytotoxic, with an LC<sub>50</sub> value of 17.31 µg/mL and demonstrating significant effectiveness in inhibiting crown gall tumorigenesis. Having an IC<sub>50</sub> value of 22.75 µg/mL, it was subjected to HPLC analysis that led to the isolation of compound PH-HM-10.

### 3.3.2. Characterization of Isolated Compound PH-HM-10

PH-HM-10 was a white amorphous powder with an observed rosy odor. The PH-HM-10 compound was soluble in methanol, water and DMSO. The mass spectra of PH-HM-10 were recorded and the molecular formula was assigned as C<sub>15</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub> in agreement with the [M + Na]<sup>+</sup> ion peak at *m/z* 305.61340 by HRESIMS (Figure 5). The IR spectrophotometer indicated a broad band at 3048 cm<sup>-1</sup>, suggesting the presence of aromatic hydrogen and revealing the presence of an amide (3502 cm<sup>-1</sup>) in the structure (Figure S1). The analysis of the <sup>1</sup>H NMR spectrum of PH-HM-10 measured in DMSO-d<sub>6</sub> indicated low-field signals at δ<sub>H</sub> 6.74 (H-8, dd), δ<sub>H</sub> 6.87 (H-10, d), δ<sub>H</sub> 7.24 (H-2, t) and δ<sub>H</sub> 7.56 (H-7, d), which are peculiar to aromatic protons in the molecule (Figure S2). One exchangeable proton signal appeared at δ<sub>H</sub> 11.30 (s) and was due to the NH-12 group present in the compound. Furthermore, a proton NMR X signal at δ<sub>H</sub> 2.52 (s) was assigned to the CH<sub>3</sub>-15; one methoxy group showed a singlet at δ<sub>H</sub> 3.51 (OCH<sub>3</sub>-1); another methoxy group δ<sub>H</sub> 3.78 (s, OCH<sub>3</sub>-9) was observed in the spectrum. <sup>13</sup>C NMR spectroscopic examination (DMSO-d<sub>6</sub>) of PH-HM-10 showed signals down field in the spectrum: C-8 (δ<sub>C</sub> 111.5), C-5 (δ<sub>C</sub> 120.3), C-7 (δ<sub>C</sub> 122.3), C-6 (δ<sub>C</sub> 121.7), C-13 (δ<sub>C</sub> 131.4), C-11 (δ<sub>C</sub> 137.4), C-1 (δ<sub>C</sub> 156.8) and C-9 (δ<sub>C</sub> 158.7); these were attributed to the aromatic carbons in the compound (Figure S3). HSQC analysis indicated a heteronuclear single quantum correlation or heteronuclear single quantum coherence. The experiment revealed the number of particular protons in the compound attached to specific carbon atoms. The HSQC of the isolated compound PH-HM-10 is provided in Figure S4. The HSQC of the PH-HM-10 revealed that C-5 with resonance at δ<sub>C</sub> 120.3, C-6 (δ<sub>C</sub> 121.7), C-9 (δ<sub>C</sub> 158.7), and C-14 with a chemical shift at δ<sub>C</sub> 189.7 were not attached to any of the protons [22]. Moreover, C-7 (δ<sub>C</sub> 122.3) was linked with two protons, showing signals at δ<sub>H</sub> 7.56 (H-7, d). C-8 (δ<sub>C</sub> 111.5) was attached to a proton with a signal at δ<sub>H</sub> 6.78 (H-8, dd) and C-10 showed resonance at δ<sub>C</sub> 93.7. In addition, CH correlates with a proton signal at δ<sub>H</sub> 6.87 (H-10, d), showing resonance in the downfield region in the isolated molecule of PH-HM-10 in the HSQC correlation (Table 3). The structure of the isolated compound was established (Figure 6) with the help of vin characterization techniques and identified as Pegaharmine E [23].

### 3.4. Anticancer Potential of PH-HM-10

The cytotoxicity of the isolated compound PH-HM-10 was examined at five different concentrations ranging from 31.25 µg/mL to 500 µg/mL (Figure 7). Results reveal that compound PH-HM-10 was the most effective against human myeloid leukemia (HL-60), with an IC<sub>50</sub> value of 36.99 µg/mL; it was also effective against the human lung adenocarcinoma epithelial cell line (A549), with 50% inhibition at 63.5 µg/mL. PH-HM-10 was least active against human gastric cancer (SGC-7901), with a maximum IC<sub>50</sub> value of 123.44 µg/mL. The isolated compound PH-HM-10 showed moderate activity on the human colorectal tumor cell line (HCT-116). A percentage inhibition of the human colorectal tumor cell line was observed from 30.44 to 76%, exhibiting moderate effectiveness in the compound. PH-HM-10 also showed moderate activity against the breast cancer cell line (MCF7), with a minimum absorbance value of 0.218 nm at the 500 µg/mL concentration of PH-HM-10 and with 26.65% cell viability of the breast cancer cell line. The maximum percentage inhibition against the MCF-7 cell line, observed at this concentration, was 73.35%. The IC<sub>50</sub> value against the breast cancer cell line (MCF7) was observed to be 85.90 µg/mL (Table 4). This result implies that the PH-HM-10 compound has considerable anticancer activity against

multiple cancer cell lines, indicating that PH-HM-10 is a prominent candidate for future drug development.

### Mass Spectrum

Analysis name	E:\Data\Huma\HM10	Source Type	ESI
Scan Begin	250 m/z	Focus	Not Active
Scan End	350 m/z	Ion Polarity	Positive
Ion	[M+H] <sup>+</sup>	Nebulizer	0.3 Bar
Measured m/z	305.61340	Dry Heater	180 °C
M	C <sub>15</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub> Na	Dry Gas	4.0/min

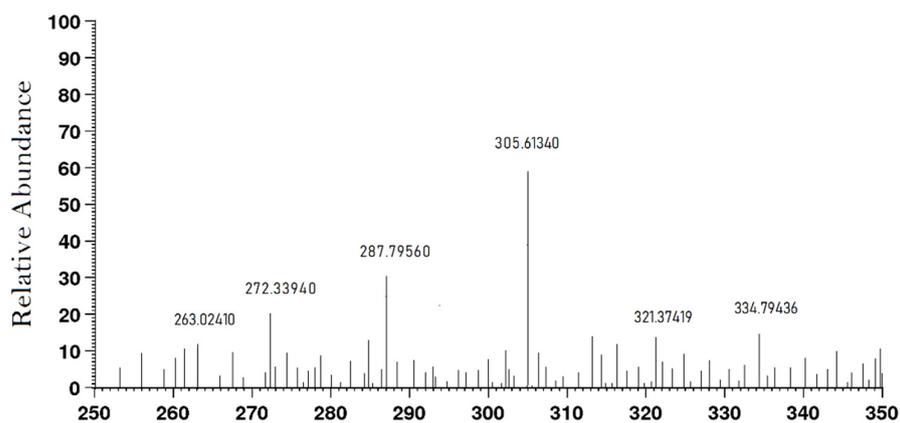


Figure 5. Mass spectrum of PH-HM-10.

Table 3. NMR (<sup>1</sup>H and <sup>13</sup>C) chemical shifts value of PH-HM-10.

Position	<sup>13</sup> C δC (ppm)	<sup>1</sup> H δH (ppm)	HSQC
1	156.822	-	C
2	-	7.244 (t)	NH
3	41.403	3.165 (m)	CH <sub>2</sub>
4	25.527	3.155 (m)	CH <sub>2</sub>
5	120.319	-	C
6	121.713	-	C
7	122.263	7.562 (d)	CH
8	111.468	6.738 (dd)	CH
9	158.681	-	C
10	93.702	6.872 (d)	CH
11	137.409	-	C
NH-12	-	11.302 (s)	
13	131.351	-	C
14	189.729	-	C
15	28.173	2.523 (s)	CH <sub>3</sub>
OCH3-9	55.568	3.784 (s)	CH <sub>3</sub>
OCH3-1	51.209	3.512 (s)	CH <sub>3</sub>

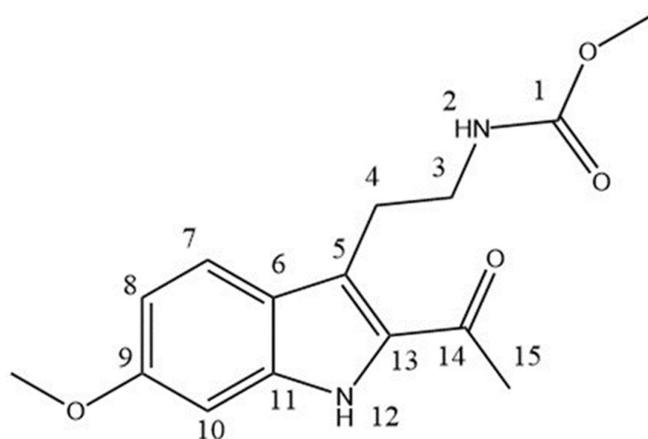


Figure 6. Structure of PH-HM-10 (Pegaharmine E).

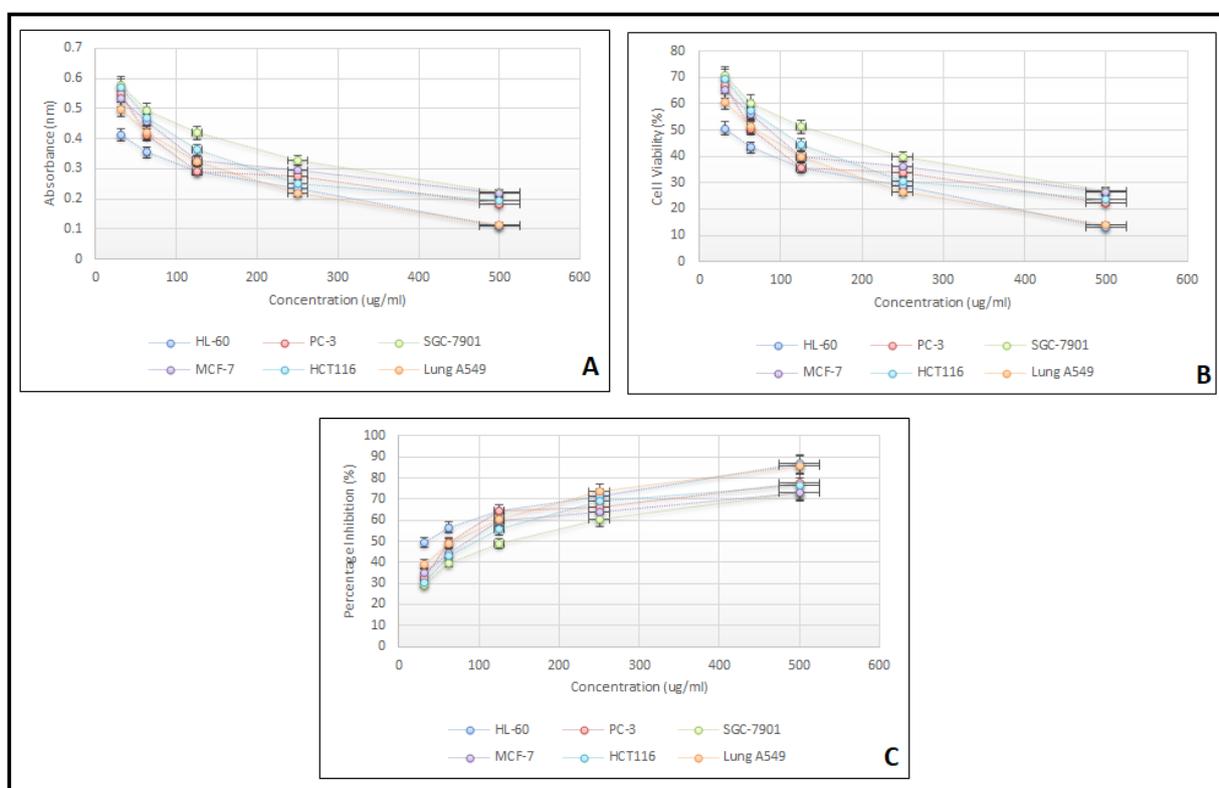


Figure 7. Absorbance (A), Percentage cell viability (B), and Percentage inhibition (C) against cancer cell lines at five different concentrations of PH-HM-10.

Table 4. Regression analysis and IC<sub>50</sub> values of anticancer activity.

Cell Lines	LC <sub>50</sub>	Regression Equation	R <sup>2</sup>
HL-60	36.99	$y = 12.893\ln(x) + 3.4476$	0.9684
PC-3	73.61	$y = 15.326\ln(x) - 15.883$	0.9437
SGC-7901	123.44	$y = 15.538\ln(x) - 24.827$	0.9959
MCF-7	85.9	$y = 13.951\ln(x) - 12.126$	0.9701
HCT116	93.84	$y = 17.055\ln(x) - 27.456$	0.9907
Lung A549	63.5	$y = 17.037\ln(x) - 20.721$	0.9965

#### 4. Discussion

Plant-derived bioactive components proved to be effective medication for treatment of various ailments. Based on ethno-medicinal and several earlier scientific reports on *P. harmala*, seeds were subjected to fractionation. All five fractions obtained after solvent-solvent fractionation were tested for cytotoxicity against brine shrimp nauplii with significant results, suggesting the role of seeds in the toxic properties of *P. harmala*. This further strengthened the speculation about the capability of the plant to produce anticancer agents [24]. Earlier findings confirmed plants with cytotoxic potential as a chief source of bioactive principles [25]. Chloroform extract significantly increases nauplii mortality so current findings assist in the prediction of bioactive compounds and anticancer potential [26]. The results of the cytotoxicity assays were in accordance with Khan et al. [27], indicating a dose-dependent response. Potassium dichromate, vincristine sulphate and etoposide were used as positive controls for the BSLT and were categorized as extremely cytotoxic in these assays, as per Clarkson's lethality criterion [20]. Brine shrimp lethality is helpful for analyzing different plant extracts for confirmation of their cytotoxic potential. The pharmacological perspective suggests a strong relationship between the brine shrimp lethality test and the discovery of bioactive principles [28]. The ethyl acetate fraction is known to be significantly cytotoxic, which helps predict the presence of the bioactive principles responsible for anticancer activity [26]. A strong relationship exists between the extract samples and the capability of an extract to control the development of crown gall tumors on potato discs [27]. The most potent ethyl acetate fraction led to the isolation of the bioactive compound PH-HM-10, following which, characterization was performed. ESI-MS spectral analysis of PH-HM-10 revealed that the dynamic energy of MS2 from protonated PH-HM-10  $[M + Na]^+$  at 305.61340  $m/z$  afforded the fragment ion at  $m/z$  287.79560 by the loss of one water molecule (18Da). C-O stretching mostly appears in the range of 1500–1800  $cm^{-1}$ , which is the characteristic band range in organic compounds [29]; thus, the characteristic signal at 1780  $cm^{-1}$  might signify the presence of saturated carbonyl conformation in PH-HM-10; stretching olefinic groups (1643  $cm^{-1}$ ) were identified from the IR spectrum. The C-N group in the structure of the isolated bioactive compound was confirmed through Fourier Transform infrared spectroscopy by the characteristic absorbance peak at 1081  $cm^{-1}$  in the spectrum [30]. Methyl, methylene and aliphatic sharp asymmetric and symmetric stretching were mostly observed in the range of 2900–2800  $cm^{-1}$ , which validates current output [31]. The values for the chemical shifts were shifted in the downfield regions  $\delta H$  6.00 ppm to  $\delta H$  9.00 ppm for the aromatic hydrogen in the H-NMR spectrum, confirming the current results [32]. A diverse range of isolated bioactive compounds is clear evidence of the singlet signal existing at the chemical shift  $\delta H$  10–11 ppm range and which corresponds to the N-H proton of the indole nucleus, reinforcing the analysis of the current spectrum [33]. The signal that appeared in the upfield region of the  $^{13}C$  NMR spectra was assignable to one methyl carbon resonance at the chemical shift value of  $\delta C$  28.173 ppm, appearing in the aliphatic region at the (CH3-15) position. The peaks observed in the upfield of the  $^{13}C$  NMR spectra in the aliphatic region between  $\delta C$  10 and 25 ppm can be fairly assigned to carbon atoms in the compound with a methyl group, so this authenticates the presence of methyl in the structure [34].  $^{13}C$  NMR spectra also revealed the presence of two carbons (OCH<sub>3</sub>-9 and OCH<sub>3</sub>1) in the structure, bearing one methoxy group at  $\delta C$  55.568 ppm and another at  $\delta C$  51.209 ppm, which endorsed the C-O bond in the isolated compound. On the basis of current output and literature values, the resonance signal in the  $^{13}C$  NMR spectrum in the range of  $\delta C$  50–58 ppm could be due to the carbon bearing one methoxy group present in the structure of the compound, so this coincides with the current spectrum [35]. PH-HM-10 compound (Pegaharmine E) was analyzed for its anticancer activity against six selected human cancer cell lines using 3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) assays [36]. The results of the study suggest the already reported trend of a strong correlation between dosage and the inhibition of cancer cell growth [37]. The absorbance level was affected by the concentration of the compound, hence it validated the relationship between concentration and absorbance [38]. The current

results coincide with former research outcomes related to anticancer potentiality tested against breast cancer cell lines [39]. The anticancer activity of isolated compounds may be due to the alteration in the redox balance that is essential for the survival of cancer cells, or it might be due to the induction of the ROS level or inhibiting the ROS level in selected cancer cells [40]. The results of the study indicate a difference in the cytotoxic effects of PH-HM-10 against different cell lines, with the best effects being observed against human myeloid leukemia (HL-60); this may be due to the differential sensitivity of cancer cells that results in different responses. Many cancers carry individual markers; therefore, the relatively higher sensitivity of some cells, such as HL-60, to this extract of some selected cells, is a reflection of their unique genetic nature [41]. The results of the present study provide dependable evidence that the *P. harmala* extract carries promising anticancer compounds, worthy of further investigation for the development of anticancer drugs.

## 5. Conclusions

The findings of the present study reveal that the isolated compound of *P. harmala* is Pegaharmine E, and bioassays proved its significant anticancer properties against multiple human cancer cell lines. Extensive in vivo and mechanistic studies are suggested in future to validate and empower its use towards anticancer drug discovery.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/separations9110355/s1>, Figure S1: Infrared spectrum of PH-HM-10; Figure S2: H NMR of PH-HM-10; Figure S3: C NMR of PH-HM-10; Figure S4: HSQC spectrum of PH-HM-10.

**Author Contributions:** Y.B. conceived the idea. H.M.S. conducted the experiment. S.A.A., N.S., D.H., K.M. and S.N. conducted the literature review. A.Q. provided technical expertise to strengthen the basic idea. W.A., A.S. and S.K. helped in statistical analysis. Y.B. proofread and provided intellectual guidance. All authors read the first draft, helped in revision, and approved the article. All authors have read and agreed to the published version of the manuscript.

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