

# Supplementary Materials

## Rational fabrication of benzene-linked porous polymers for selective CO<sub>2</sub> capture

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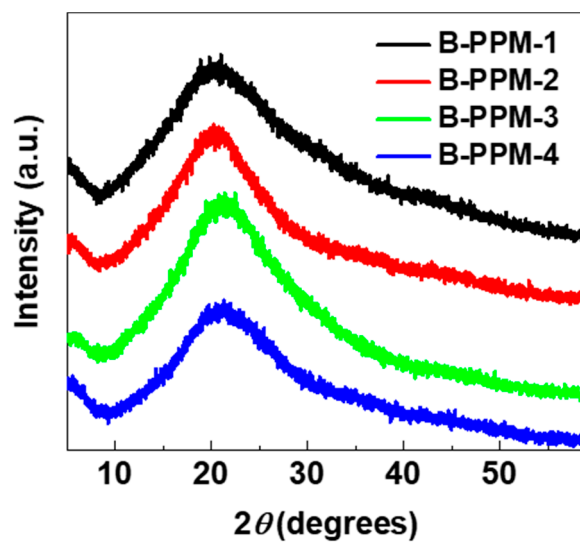
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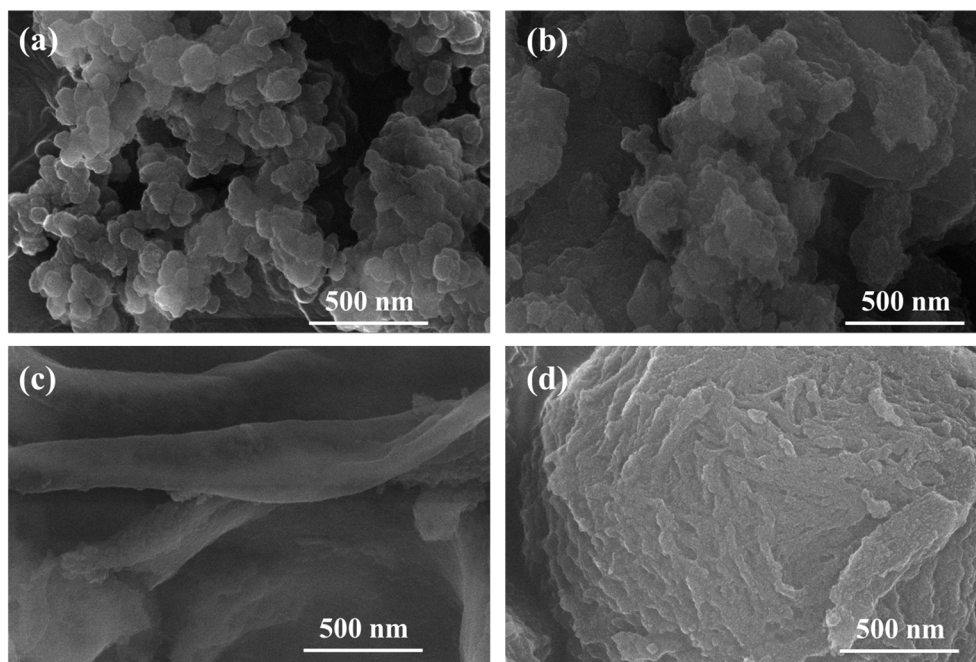
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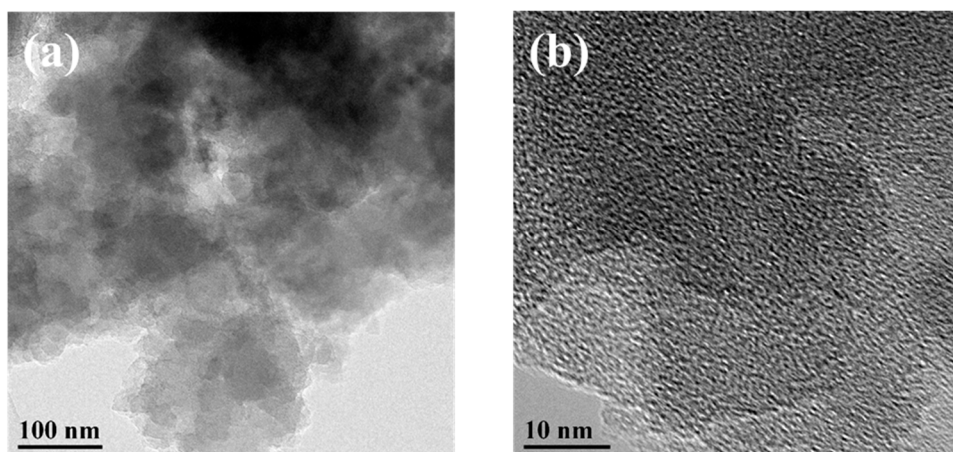
<sup>†</sup> These authors contributed equally to this work.



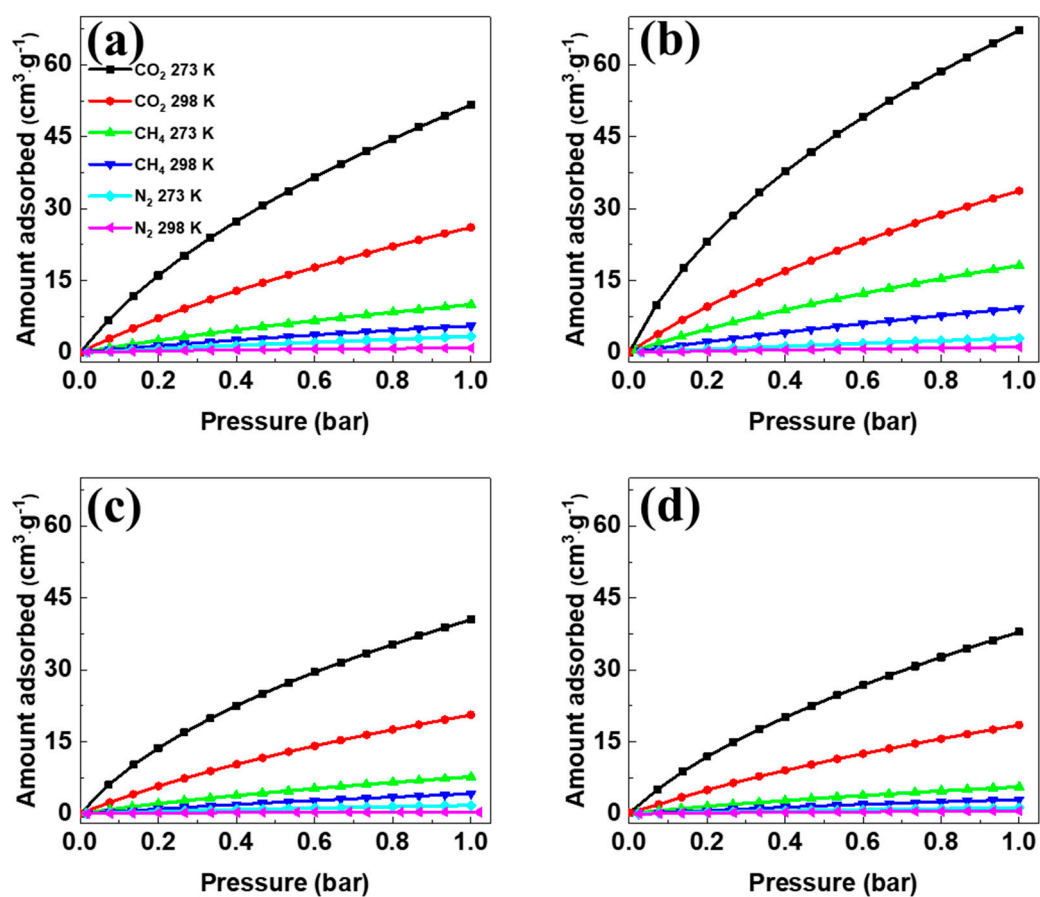
**Figurer S1.** XRD pattern of the B-PPMs.



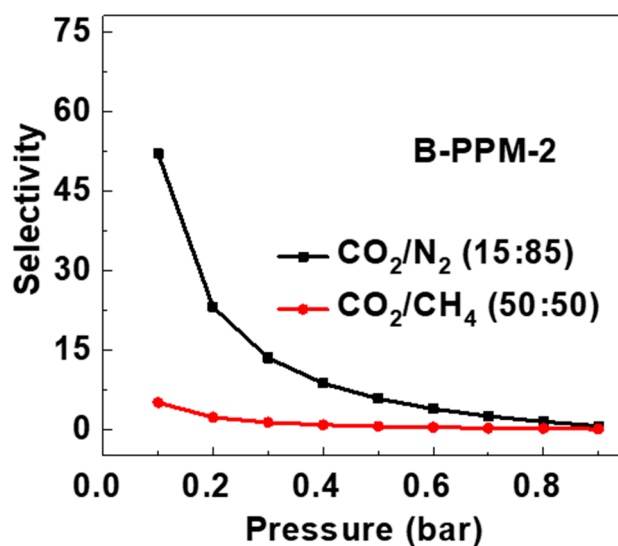
**Figure S2.** SEM images of the (a) B-PPM-1, (b) B-PPM-2, (c) B-PPM-3 and (d) B-PPM-4.



**Figure S3.** TEM images (a) 100 nm and (b) 10 nm of the B-PPM-2.

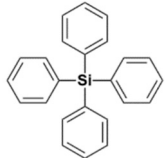
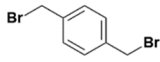
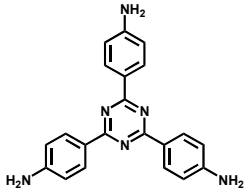
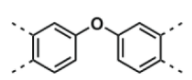
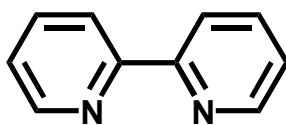
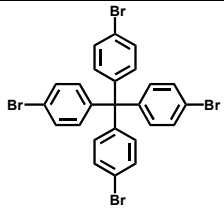


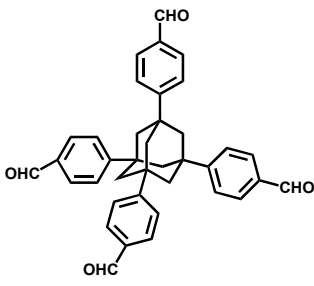
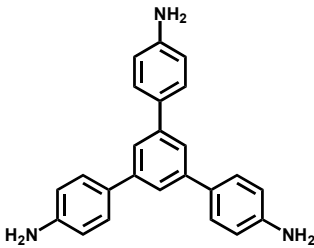
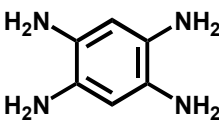
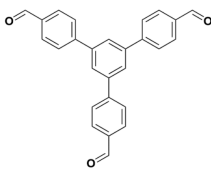
**Figure S4.** CO<sub>2</sub>, N<sub>2</sub> and CH<sub>4</sub> adsorption isotherms of the sample (a) B-PPM-1, (b) B-PPM-2, (c) B-PPM-3, and (d) B-PPM-4 at 273 K and 298 K and 1 bar.



**Figure S5.** IAST selectivity of CO<sub>2</sub>/N<sub>2</sub> (15:85) and CO<sub>2</sub>/CH<sub>4</sub> (50:50) on the B-PPM-2 at 298 K.

**Table S1.** Comparison of the cost of adsorbents based on the price of monomers and catalysts

Sample	Monomer 1 (Price)	Monomer 2 (Price)	Catalyst (Price)	Price /100g	Ref.
B-PPM-2	 tetraphenylsilane (\$743/100mL)	 $\alpha, \alpha'$ -Dibromo-p-xylene (\$82/100g)	—	\$792	This work
TPI-5	 4,4',4'-(1,3,5-Triazine-2,4,6-Triyl) Trianiline (\$1006/100g)	 4,4'-Oxybis(1,2-dimethylb enzene) <i>synthesized from</i> <i>precursor</i> (\$832/100g)	Pd(dba) <sub>2</sub> (\$230/2g)	\$964	[1]
PAF-1	 2,2'-Bipyridine (\$464/100g)	 Tetrakis(4-bromophenyl) methane (\$3528/100g)	[Ni(cod) <sub>2</sub> ] (\$159/2g)	\$4136	[2]

PSN-TAPB	 <p>1,3,5,7-Tetrakis(4-formylphenyl)adamantane (\$1094 /1g)</p>	 <p>1,3,5-Tris(4-aminophenyl)benzene (\$2093/100g)</p>	—	~\$2800/ 1g	[3]
CPCs	 <p>1,2,4,5-benzenetetramine tetrahydrochloride (\$210/1g)</p>	 <p>1,3,5-(4-formylphenyl)-benzene (\$1714/100g)</p>	—	\$2259	[4]

The price of chemicals is obtained from Sigma-Aldrich. The price is mostly based on 100 g (or 100 mL) of chemicals, and other units will be used if chemicals with 100 g (or 100 mL) is not available.

1. Liebl, M.R.; Senker, J. Microporous functionalized triazine-based polyimides with high CO<sub>2</sub> capture capacity. *Chem. Mater.* **2013**, *25*, 970-980, doi:10.1021/cm4000894.
2. Ben, T.; Ren, H.; Ma, S.; Cao, D.; Lan, J.; Jing, X.; Wang, W.; Xu, J.; Deng, F.; Simmons, J.M.; et al. Targeted synthesis of a porous aromatic framework with high stability and exceptionally high surface area. *Angew. Chem. Int. Ed.* **2009**, *48*, 9457-9460, doi:10.1002/anie.200904637.
3. Li, G.; Zhang, B.; Yan, J.; Wang, Z. Micro- and mesoporous poly(Schiff-base)s constructed from different building blocks and their adsorption behaviors towards organic vapors and CO<sub>2</sub> gas. *J. Mater. Chem. A* **2014**, *2*, 18881-18888, doi:10.1039/c4ta04429k.
4. Ashourirad, B.; Sekizkardes, A.K.; Altarawneh, S.; El-Kaderi, H.M. Exceptional gas adsorption properties by nitrogen-doped porous carbons derived from benzimidazole-linked polymers. *Chem. Mater.* **2015**, *27*, 1349-1358, doi:10.1021/cm504435m.