

Supplementary Materials

Evaluation of Antimicrobial Activity and Cytotoxicity Effects of Extracts of *Piper nigrum* L. and Piperine

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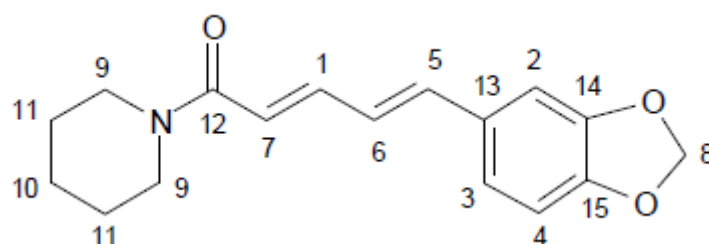


Figure S1. Molecular structure of Piperine.

The purity of the piperine compound (Figure S1) was confirmed by the ¹H and ¹³C NMR data. In the spectra of the piperine component, the following signals were observed: δ 24.69 (C-10), δ 25.53 (C11), δ 26.68 (C-11), δ 43.17 (C-9), δ 46.83 (C-9), δ 101.21 (C-8), δ 105.60 (C-2), δ 108.41 (C-4), δ 120.01 (C-7), δ 125.30 (C-3), δ 138.13 (C-5), δ 130.95 (C-13), δ 122.42 (C-6), δ 142.39 (C-1), δ 148.05 (C-14), δ 148.13 (C-15), δ 165.35 (C-12). The solvent signal was observed at around 76.5 ppm. All ¹³C-NMR values correlated accurately with the data reported in the literature (Figure S2).

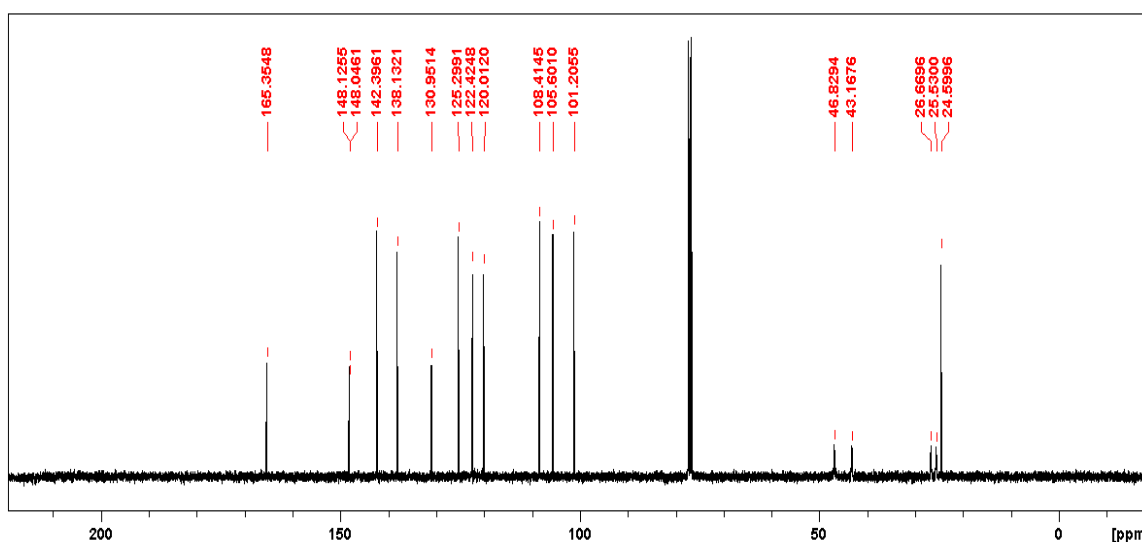


Figure S2. ¹³C NMR spectrum obtained from analysis of piperine crystals.

The ¹H-NMR indicated 19 protons within the piperine compound, 12 are in the aliphatic region. In the ¹H-NMR spectra of the piperine component, the following signals were observed: δ 1.58–1.65 (4 H, m, 11-H), δ 1.56–1.59 (2 H, m, 10-H), δ 3.52–3.63 (4 H, m, 9-H), δ 5.95 (2 H, s, 8-H), δ 6.42 (1 H, d, 7-H), δ 6.73 (1 H, m, 6-H), δ 6.74 (1 H, m, 5-H), δ

6.76 (1 H, d, 4-H), δ 6.87 (1 H, dd, 3-H), δ 6.96 (1 H, m, 2-H), δ 7.36–7.40 (1 H, ddd, 1-H). Also, the solvent signal was observed at around 7.26 ppm. All ^1H -NMR values corresponded to the data reported in the literature (Muthurajan et al., 2014) (Figura S3).

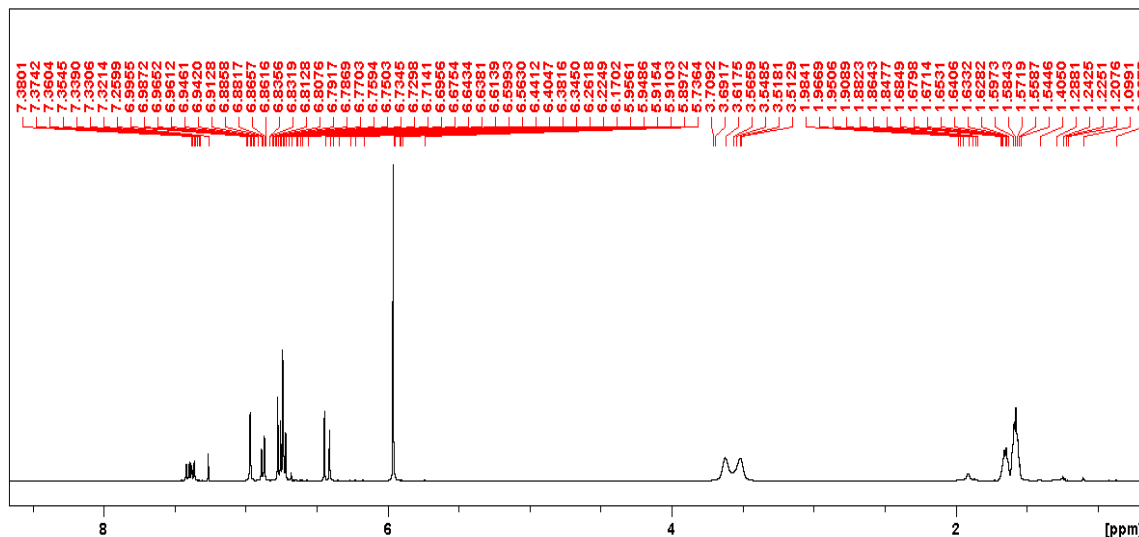


Figure S3. ^1H NMR spectrum obtained from analysis of piperine crystals.

The absorptions of the carbon atoms were verified in the bands of 24–165 ppm (Table S1). According to Thenmozhi and coworkers (2014) (Muthurajan, Mohanraj, Aravindan, Jayaprakash, & Thenmozhi, 2014) these bands suggest the presence of only aromatic and unsaturated carbon atoms in the molecule. Peaks in the 0–50 ppm range correspond to carbon–carbon single bonds. The peak in the range of 101 ppm refers to the carbon–oxygen atom in the molecule. The bands of 100–150 ppm are due to the carbon–carbon double bonds and the absorption peak in the 165.35 ppm range is the carbon present in the carbon–oxygen double bond (Figure S1).

Table S1. NMR data (δ_{C} and δ_{H}) experimental (Exp.) in ppm for piperine.

Position	^{13}C	^1H
	Exp.	Exp.
1	142,4	7,4
2	105,6	7,0
3	125,3	6,9
4	108,4	6,8
5	138,1	6,7
6	122,4	6,7
7	120,0	6,4
8	101,2	6,0
9	43,17	3,6
9	46,83	-
10	24,69	1,6
11	25,53	1,6
11	26,68	-
12	165,4	-
13	131,0	-
14	148,1	-
15	148,1	-

(–) Data not available.