



Article Mathematical Modeling and Stability Analysis of a Two-Phase Biosystem

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Abstract: We propose a new mathematical model describing a biotechnological process of simultaneous production of hydrogen and methane by anaerobic digestion. The process is carried out in two connected continuously stirred bioreactors. The proposed model is developed by adapting and reducing the well known Anaerobic Digester Model No 1 (ADM1). Mathematical analysis of the model is carried out, involving existence and uniqueness of positive and uniformly bounded solutions, computation of equilibrium points, investigation of their local stability with respect to practically important input parameters. Existence of maxima of the input–output static characteristics with respect to hydrogen and methane is established. Numerical simulations using a specially elaborated web-based software environment are presented to demonstrate the dynamic behavior of the model solutions.

Keywords: anaerobic digestion; two-phases process; hydrogen; methane; mathematical model; equilibrium points; stability analysis; numerical simulation

1. Introduction

Anaerobic digestion (AD) is a multi-step biotechnological process widely applied for the treatment of wastes and wastewater, where usually hydrogen (H_2) is not accumulated as an intermediary product [1–3]. Recently, the interest in H_2 production by modifying operating conditions of the AD has increased. This process has been denominated as dark fermentative H_2 production in contrast with light derived H_2 producing processes [4–8]. In traditional AD, H_2 is not detected as it is consumed immediately e.g., by hydrogenotrophic methanogens to produce methane CH_4 and carbon dioxide CO_2 [9]. Conversely, H_2 can be produced as main component in biogas by engineering the process conditions in a way that favors Clostridium and Enterobacter metabolism against the predominance of archaeal microflora [10,11]. However, the main limitation of dark fermentative H_2 production is the rather low energy recovery, cf. [7,8,12]. In order to completely utilize the organic acids produced during dark fermentation and improve the overall energy conversion efficiency, a two-stage AD (TSAD) concept consisting of a hydrogenic process followed by a methanogenic process has been suggested (see e.g., [12–14]).

A lot of models describing separately the fermentative hydrogen production (see, e.g., [15,16]) and the AD for methane production (cf. e.g., [17–20]) are known. However, only a few models of TSAD processes are available; see, for example, [21–24] and the references therein. In [25], both bio-hydrogen and bio-methane productions are optimized in two bioreactors regarding two operating parameters, organic loading rate, and dilution rate; however, the first bioreactor is running in semi-continuous and the second—in batch operation mode because the working volumes of both bioreactors are

equal. In [26], steady-state analysis for the TSAD process is presented using a very simple model without hydrolysis and without taking into account the most important fact for this scheme—the energy recovery.

In this paper, we propose a mathematical model, describing the process of simultaneous H_2 and CH_4 production by AD of organic wastes in a cascade of two connected continuously stirred bioreactors. The biochemical processes in the first bioreactor include disintegration of organic wastes, hydrolysis, and acidogenesis with hydrogen production. The methane production from acetate (methanogenesis) is separated in the second bioreactor. The proposed model is developed by reducing the well known Anaerobic Digester Model No 1 (ADM1) basic structure elaborated by the International Water Association (IWA) [17]. Modeling of TSAD using ADM1 is also presented in [27], where, however, methane is obtained from both bioreactors.

Our mathematical model is presented for the first time in the conference paper [21]. The present article represents an extension of the latter, providing (i) more detailed assumptions used to construct the model, which are based on practical experiments (Section 3); (ii) more precise and deeper theoretical investigations of the model dynamics such as existence and uniqueness of positive and uniformly bounded solutions, more precise computation of practically important equilibrium points of the model and study of their local asymptotic stability (Section 4); (iii) more detailed investigation concerning the optimization of hydrogen and methane production on steady state operation (Section 5), and showing the existence of maxima, which is important for the applications; (iv) numerical simulations in Section 6, based on a newly developed, so called one-page application "(H2,CH4) Bioreactors ver. 1.2" in the SmoWeb platform.

The main goal of this article is to present rigorous mathematical analysis of the proposed model dynamics, to detect and provide some model-based predictions, which could be useful in designing and engineering real laboratory experiments. The theoretical studies are supported and complimented by a variety of computer simulations.

2. Process Description

In the TSAD system, relatively fast growing acidogens and H_2 producing microorganisms are developed in the first-stage hydrogenic bioreactor (BR_1 with working volume V_1 , see Figure 1) and are involved in the production of volatile fatty acids (VFA–mainly acetate and butyrate, [8]) and H_2 . Some recent studies on the microbial community in the hydrogen producing BR_1 show that dominant are microorganisms of the genus Clostridium [28]. In BR_1 , pH is kept in the range 5.0–5.5, and acetate degrading methanogens are eliminated. On the other hand, the slow growing acetogens and methanogens are developed in the second-stage methanogenic bioreactor (BR_2 with working volume V_2 , and pH in the range 6.5–8.5), in which the produced VFA are further converted to CH_4 and CO_2 by methane-forming bacteria from the genus Methanobacterium, Methanococcusect, see, for details, [9] and the references therein.

It is known that in the two-stage $H_2 + CH_4$ system the energy yields are up to 43% more in comparison to the traditional one-stage CH_4 production process (cf. [8,25]).

Assume that the volumes V_1 and V_2 of the bioreactors are constant. Let F_1 and F_2 be the inflows in the first and second bioreactor, respectively, and let $F_1 = F_2 = F$ be valid. It is well known that the dilution rates D_1 and D_2 are defined as

$$D_1 = \frac{F}{V_1}, \quad D_2 = \frac{F}{V_2}.$$

Then,

$$D_2 = \frac{V_1}{V_2} D_1 := \gamma D_1 \text{ within } \gamma := \frac{V_1}{V_2}.$$
(1)

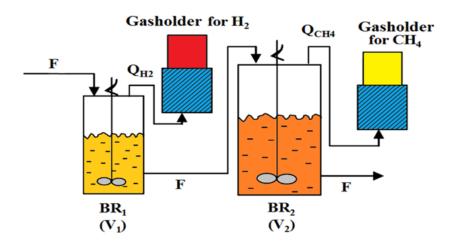


Figure 1. Two-phases process of AD with production of hydrogen (H_2) and methane (CH_4) .

Later on in the paper (see Section 5), by analyzing points of maximum biogas production in both bioreactors, biohydrogen in BR_1 and biomethane in BR_2 , we find theoretical values for the hydraulic retention time (HRT) $1/D_{1,\text{max}}$ and $1/D_{2,\text{max}}$, which show that the volume V_2 of the second bioreactor for methane production should be larger than the volume V_1 of the first bioreactor. Therefore, $\gamma < 1$ should be valid. We determine the constant γ using the proposed model equations.

3. Model Description

In this section, we present the mathematical model describing the process of simultaneous H_2 and CH_4 production by AD of organic wastes in a cascade of two continuously stirred bioreactors. The model is derived on the basis of the ADM1 basic structure as well as on our experience with the two-phase AD process with hydrogen and methane production, see [29]. The ADM1 is a complex model, describing all known (till its creation) microbiological and biochemical reactions occurring in AD of organic wastes. ADM1 is suitable mainly for process knowledge and simulation, but not appropriate for theoretical studies and model-based control design due to the plenty of input parameters which are difficult to obtain. The reduction of ADM1 is done here in order to simplify the latter and to provide an opportunity for studying the asymptotic stability of the dynamics with respect to two essential indicators: (i) the separation of the model equations into two groups, corresponding to the reactions taking place in each one of the two bioreactors, and (ii) the type of the organic wastes used.

For simplification of the model, the following assumptions have been accepted:

- In *BR*₁ only the main VFA (acetate) is formed, whereas propionate and butyrate are omitted. In this way, the slow acetogenic phase has been omitted and consequently in *BR*₂ only methanogenesis occurs.
- In both bioreactors, the existence of inhibition phenomena is not taken into account.
- Balance equations of the hydrogen and methane in the liquid phases have been neglected because they are practically not dissolved in liquids.
- Hydrogenotrophic bacteria, producing methane from hydrogen, are eliminated previously and do not exist in this process.
- Equations describing balances of inorganic components and some biochemical equations have been neglected in view of simplifying the model.
- pH is considered in the range between 5.0 to 5.5 in the first bioreactor BR_1 , and in the interval 6.5–8.5 in the second bioreactor BR_2 . Within these ranges, it is assumed that the negative influence of pH could be disregarded.

These ranges of pH were obtained during our own laboratory experiments with TSAD of waste from maize treatment (not published yet).

The proposed models of the two bioreactors are of balancing type for continuously stirred tank reactors (CSTR) like the original ADM1 model. For each bioreactor, the ADM1 equations are adapted to describe the dynamics of the most important variables for the AD of organic waste.

Following the above assumptions, we present now the mathematical model for hydrogen and methane production in the two bioreactors. The definitions of the phase variables and parameters in the model equations are given in Tables 1 and 2 below.

The dynamics in the first bioreactor are described by the following set of 10 nonlinear ordinary differential equations:

$$\frac{d}{dt}S_{su}(t) = D_1(S_{su}^{in} - S_{su}) + k_{hyd,ch}X_{ch} + f_{su,li}k_{hyd,li}X_{li} - \mu_{suaa,su}X_{suaa}$$
(2)

$$\frac{d}{dt}S_{aa}(t) = D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}X_{pr} - \mu_{suaa,aa}X_{suaa}$$
(3)

$$\frac{d}{dt}S_{fa}(t) = D_1(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa}$$
(4)

$$\frac{a}{dt}S_{ac}(t) = -D_1S_{ac} + (1 - Y_{suaa}) [f_{ac,su}\mu_{suaa,su} + f_{ac,aa}\mu_{suaa,aa}] X_{suaa} + 0.7 (1 - Y_{fa})\mu_{fa}X_{fa}$$
(5)

$$\frac{d}{dt}X_c(t) = D_1(X_c^{in} - X_c) - k_{dis}X_c$$
(6)

$$\frac{d}{dt}X_{ch}(t) = -(D_1 + k_{hyd,ch})X_{ch} + f_{ch,xc}k_{dis}X_c$$
(7)

$$\frac{d}{dt}X_{pr}(t) = -(D_1 + k_{hyd,pr})X_{pr} + f_{pr,xc}k_{dis}X_c$$
(8)

$$\frac{d}{dt}X_{li}(t) = -(D_1 + k_{hyd,li})X_{li} + f_{li,xc}k_{dis}X_c$$
(9)

$$\frac{d}{dt}X_{suaa}(t) = \left[-D_1 + Y_{suaa}(\mu_{suaa,su} + \mu_{suaa,aa})\right]X_{suaa}$$
(10)

$$\frac{d}{dt}X_{fa}(t) = -(D_1 - Y_{fa}\mu_{fa})X_{fa},$$
(11)

with gaseous output

$$Q_{h2} = (Y_{h2,su} \mu_{suaa,su} + Y_{h2,aa} \mu_{suaa,aa}) X_{suaa} + Y_{h2,fa} \mu_{fa} X_{fa}$$

where Q_{h2} is the hydrogen flow rate [L/h]. In (2)–(11), the biomass specific growth rates $\mu_{suaa,su}$, $\mu_{suaa,aa}$ and μ_{fa} are of the form

$$\begin{split} \mu_{suaa,su} &= \mu_{suaa,su}(S_{su},S_{aa}) = k_{m,suaa} \frac{S_{su}}{k_{s,suaa} + S_{su}} \cdot \frac{S_{su}}{S_{su} + S_{aa}}, \\ \mu_{suaa,aa} &= \mu_{suaa,aa}(S_{su},S_{aa}) = k_{m,suaa} \frac{S_{aa}}{k_{s,suaa} + S_{aa}} \cdot \frac{S_{aa}}{S_{su} + S_{aa}}, \\ \mu_{fa} &= \mu_{fa}(S_{fa}) = k_{m,fa} \frac{S_{fa}}{k_{s,fa} + S_{fa}}. \end{split}$$

	Definition of the Model Variables		
S _{su}	concentration of monosacharides $[gCOD/L]$		
S _{aa}	concentration of amino acids (AA) $[gCOD/L]$		
S _{fa}	concentration of fatty acids (LCFA) $[gCOD/L]$		
S _{ac}	concentration of total acetate in $BR_1 [gCOD/L]$		
$S_{ac,ch4}$	concentration of total acetate in $BR_2 \left[gCOD/L\right]$		
X _c	concentration of composites $[gCOD/L]$		
X_{ch}	concentration of carbohydrates $[gCOD/L]$		
X_{pr}	concentration of proteins $[gCOD/L]$		
X_{li}	concentration of lipids $[gCOD/L]$		
X _{suaa}	concentration of sugar and AA degraders $[g/L]$		
X _{fa}	concentration of LCFA degraders $[g/L]$		
X _{ac}	concentration of acetate degraders $[g/L]$		
D_1	dilution rate in the first bioreactor $[h^{-1}]$		
D_2	dilution rate in the second bioreactor $[h^{-1}]$		

Table 1. Model Variables

	Definition of the Model Parameters	Values
S_{su}^{in}	input concentration of $S_{su} [gCOD/L]$	0.01
S ⁱⁿ _{aa}	input concentration of S_{aa} [gCOD/L]	0.001
S_{fa}^{in}	input concentration of $S_{fa} [gCOD/L]$	0.001
S_{su}^{in} S_{aa}^{in} S_{fa}^{in} X_{c}^{in}	input concentration of $X_c [gCOD/L]$	50
f _{ch,xc}	stoichiometric parameter [–]	0.2
f _{pr,xc}	stoichiometric parameter [-]	0.2
f _{li,xc}	stoichiometric parameter [–]	0.3
f _{su,li}	stoichiometric parameter [-]	0.05
f _{fa,li}	stoichiometric parameter [–]	0.95
fac,su	stoichiometric parameter [-]	0.41
fac,aa	stoichiometric parameter [-]	0.4
Y _{suaa}	stoichiometric parameter [–]	0.1
Y_{ac}	stoichiometric parameter [–]	27.3
Y _{fa}	stoichiometric parameter [–]	0.06
$Y_{h2,su}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{h2,aa}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{h2,fa}$	physicochemical parameter $[L^2/g]$	0.7
$Y_{ch4.ac}$	physicochemical parameter $[L^2/g]$	75
k _{dis}	biochemical parameter $[h^{-1}]$	0.0208
$k_{hyd,ch}$	biochemical parameter $[h^{-1}]$	0.417
k _{hyd,pr}	biochemical parameter $[h^{-1}]$	0.417
$k_{hyd,li}$	biochemical parameter $[h^{-1}]$	0.417
k _{m,suaa}	biochemical parameter $[h^{-1}]$	1.25
k _{s,suaa}	biochemical parameter $\left[g/L\right]$	0.5
$k_{m,fa}$	biochemical parameter $[h^{-1}]$	0.15
$k_{m,ac}$	biochemical parameter $[h^{-1}]$	0.0167
$k_{s,fa}$	biochemical parameter $\left[g/L\right]$	0.67
$k_{s,ac}$	biochemical parameter $\left[g/L\right]$	0.4

Table 2. Model Parameters

We extend the model (2)–(11) by the following two equations describing the process in the second bioreactor (cf. [30]):

$$\frac{d}{dt}S_{ac,ch4}(t) = D_2(S_{ac} - S_{ac,ch4}) - Y_{ac} \mu_{ac,ch4} X_{ac}$$
(12)

$$\frac{d}{dt}X_{ac}(t) = (-D_2 + \mu_{ac,ch4})X_{ac}$$
(13)

with gaseous output

$$Q_{ch4} = Y_{ch4,ac} \mu_{ac,ch4} X_{ac}$$

$$\mu_{ac,ch4} = \mu_{ac,ch4}(S_{ac,ch4}) = k_{m,ac} \frac{S_{ac,ch4}}{k_{s,ac} + S_{ac,ch4}}.$$

All model coefficients in the Equations (2)–(13) are assumed to be positive. The dilution rates D_1 and D_2 are the decision or control variables. The numerical values in the rightmost column of Table 2 are taken from [31]. In the model variables and parameters, the subscripts h2 and ch4 indicate hydrogen (H_2) and methane (CH_4) , respectively.

4. Investigation of the Model Solutions

This section presents a mathematical analysis of the dynamics (2)-(13). First, we establish existence, positivity, uniform boundedness, and uniqueness of the model solutions. The results are presented in Theorem 1. The interested reader can find a detailed proof of Theorem 1 in Appendix A.1 of the Appendix A.

Theorem 1. If $0 < Y_{suaa} < 1$ and $0 < Y_{fa} < 1$, then all solutions of the dynamics (2)–(13) are positive and uniformly bounded, and thus exist for all time $t \in [0, +\infty)$.

Note that the assumptions $0 < Y_{suaa} < 1$ and $0 < Y_{fa} < 1$ in Theorem 1 are not restrictive. Since Y_{suaa} and Y_{fa} are yield coefficients, their values are always less than 1 in practice; see last column in Table 2.

In the next two subsections, we find the simplest solutions of the model, namely the equilibrium points, and investigate their local asymptotic stability.

4.1. Equilibrium Points of the Model

The equilibrium points of the model are solutions of the algebraic equations obtained from (2)–(13)by setting the right-hand sides equal to zero:

$$D_1(S_{su}^{in} - S_{su}) + k_{hyd,ch}X_{ch} + f_{su,li}k_{hyd,li}X_{li} - \mu_{suaa,su}X_{suaa} = 0$$
(14)

$$D_1(S_{aa}^{ia} - S_{aa}) + k_{hyd,pr}X_{pr} - \mu_{suaa,aa}X_{suaa} = 0$$

$$(15)$$

$$D_1(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa} = 0$$

$$D_1(S_{fa}^{in} - S_{fa}) + f_{fa,li}k_{hyd,li}X_{li} - \mu_{fa}X_{fa} = 0$$

$$(16)$$

$$D_1 S_{ac} - (1 - Y_{suaa}) \left[f_{ac,su} \mu_{suaa,su} + f_{ac,aa} \mu_{suaa,aa} \right] X_{suaa} - 0.7(1 - Y_{fa}) \mu_{fa} X_{fa} = 0$$
(17)

$$D_1(X_c^{in} - X_c) - k_{dis}X_c = 0 (18)$$

$$(D_1 + k_{hyd,ch})X_{ch} - f_{ch,xc}k_{dis}X_c = 0$$
⁽¹⁹⁾

$$(D_1 + k_{hyd,pr})X_{pr} - f_{pr,xc}k_{dis}X_c = 0$$
(20)

$$(D_1 + k_{hyd,li})X_{li} - f_{li,xc}k_{dis}X_c = 0$$
⁽²¹⁾

$$\left[D_1 - Y_{suaa}(\mu_{suaa,su} + \mu_{suaa,aa})\right] X_{suaa} = 0$$
⁽²²⁾

$$(D_1 - Y_{fa}\mu_{fa})X_{fa} = 0. (23)$$

$$(D_1 - Y_{fa}\mu_{fa})X_{fa} = 0.$$

$$D_2(S_{ac} - S_{ac,ch4}) - Y_{ac} \mu_{ac,ch4}X_{ac} = 0$$
(23)
(24)

$$(D_2 - \mu_{ac,ch4})X_{ac} = 0. (25)$$

We are looking for nonnegative solutions of (14)-(25) since only these equilibrium points correspond to the physical and biological meaning of the model variables. The equilibrium points are computed as functions of the control variables, the dilution rates D_1 and D_2 .

Using Equation (18), we find the equilibrium component \bar{X}_c with respect to the phase variable X_c , which is obviously positive:

$$\bar{X}_c = \frac{D_1 X_c^{in}}{D_1 + k_{dis}}.$$

Furthermore, Equations (19)–(21) imply the equilibrium components \bar{X}_{ch} , \bar{X}_{pr} and \bar{X}_{li} respectively:

$$\bar{X}_{ch} = \frac{f_{ch,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hyd,ch}}, \quad \bar{X}_{pr} = \frac{f_{pr,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hyd,pr}}, \quad \bar{X}_{li} = \frac{f_{li,xc} \, k_{dis} \, \bar{X}_c}{D_1 + k_{hyd,li}}.$$
(26)

Equations (16) and (23) are used to determine the steady state components \bar{S}_{fa} and \bar{X}_{fa} with respect to the phase variables S_{fa} and X_{fa} . Obviously, Equation (23) suggests that the trivial solution $X_{fa} \equiv 0$ always exists, but is not of practical interest. Assuming that $X_{fa} \neq 0$ and using the expression of μ_{fa} , we obtain from (23) the following steady state component \bar{S}_{fa}

$$\bar{S}_{fa} = \frac{D_1 k_{s,fa}}{Y_{fa} k_{m,fa} - D_1}.$$
(27)

It is clear that \bar{S}_{fa} exists and $\bar{S}_{fa} > 0$ if and only if

$$D_1 < Y_{fa} k_{m,fa}. \tag{28}$$

Equation (16) then delivers the equilibrium component for X_{fa} ,

$$\bar{X}_{fa} = \frac{D_1(S_{fa}^{in} - \bar{S}_{fa}) + f_{fa,li}k_{hyd,li}\bar{X}_{li}}{\mu_{fa}(\bar{S}_{fa})}.$$
(29)

Obviously, \bar{X}_{fa} from (29) satisfies $\bar{X}_{fa} > 0$ if and only if

$$\bar{S}_{fa} < S_{fa}^{in} + \frac{1}{D_1} f_{fa,li} k_{hyd,li} \bar{X}_{li}.$$
 (30)

Remark 1. The quantity $S_{fa}^{in} + \frac{1}{D_1} f_{fa,li} k_{hyd,li} \bar{X}_{li}$ on the right-hand side of (30) can be considered as a worst-case upper bound of the total concentration of fatty acids (LCFA) S_{fa} , which accumulation in BR₁ might occur as a result of some imbalances in the degradation phase, leading to wash-out of the LCFA degraders X_{fa} [32].

The inequality (30) is equivalent with

$$D_{1}^{3}\left(k_{s,fa} + S_{fa}^{in}\right) + D_{1}^{2}\left[\left(k_{s,fa} + S_{fa}^{in}\right)\left(k_{hyd,li} + k_{dis}\right) - Y_{fa}k_{m,fa}S_{fa}^{in}\right] \\ + D_{1}\left[f_{fa,li}k_{hyd,li}f_{li,xc}k_{dis}X_{c}^{in} - Y_{fa}k_{m,fa}S_{fa}^{in}\left(k_{hyd,li} + k_{dis}\right) + k_{s,fa} + S_{fa}^{in}\right] \\ - Y_{fa}k_{m,fa}k_{hyd,li}k_{dis}\left(S_{fa}^{in} + f_{fa,li}f_{li,xc}X_{c}^{in}\right) < 0.$$
(31)

The left hand-side cubic polynomial in (31) always possesses one positive real root $D_1^{(1)}$, so the inequality (30) is satisfied if $D_1 \in (0, D_1^{(1)})$. Therefore, the equilibrium components \bar{S}_{fa} and \bar{X}_{fa} exist and satisfy $\bar{S}_{fa} > 0$ and $\bar{X}_{fa} > 0$ if $D_1 \in (0, \min\{Y_{fa}k_{m,fa}, D_1^{(1)}\})$ holds true.

If $D_1 = D_1^{(1)}$, then \bar{X}_{fa} becomes equal to zero; this means that $D_1 = D_1^{(1)}$ is a (transcritical) bifurcation value of the parameter D_1 , leading to wash-out of the biomass X_{fa} for $D_1 > D_1^{(1)}$. We shall

pay special attention to the case $\bar{X}_{fa} = 0$ motivating our considerations later on in the paper (see Section 5). If $\bar{X}_{fa} = 0$, then we obtain the following steady state component for S_{fa}

$$\bar{S}_{fa}^{0} = S_{fa}^{in} + \frac{1}{D_1} f_{fa,li} k_{hyd,li} \bar{X}_{li},$$
(32)

which always exists and is positive if $D_1 > D_1^{(1)}$.

Here, and in what follows, the superscript 0 indicates $\bar{X}_{fa} = 0$ in the steady state components.

Furthermore, Equations (14), (15) and (22) are used to compute the steady state components \bar{S}_{su} , \bar{S}_{aa} , and \bar{X}_{suaa} in case positive solutions do exist. It is clear from Equation (22) that the trivial solution $X_{suaa} \equiv 0$ always exists, but we exclude this case for biological evidence. Let $X_{suaa} \neq 0$. Then, Equation (22) reduces to

$$D_1 - Y_{suaa} \left[\mu_{suaa,su}(S_{su}, S_{aa}) + \mu_{suaa,aa}(S_{su}, S_{aa}) \right] = 0.$$
(33)

Using Equation (15), we find

$$X_{suaa} = \frac{D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}\bar{X}_{pr}}{\mu_{suaa,aa}(S_{su}, S_{aa})}.$$
(34)

The equilibrium component with respect to X_{suaa} should be positive and this will be fulfilled if

$$S_{aa} < S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}.$$
 (35)

A similar remark to Remark 1 can be made to the above inequality (35) concerning the concentrations of the amino acids S_{aa} and of the degraders X_{suaa} .

Substituting X_{suaa} from (34) into Equation (14) and using Equation (33), we obtain the following nonlinear algebraic system with respect to S_{aa} and S_{su} :

$$\begin{bmatrix} D_1(S_{su}^{in} - S_{su}) + k_{hyd,ch}\bar{X}_{ch} + f_{su,li}k_{hyd,li}\bar{X}_{li} \end{bmatrix} \mu_{suaa,aa}(S_{su}, S_{aa}) \\ - \begin{bmatrix} D_1(S_{aa}^{in} - S_{aa}) + k_{hyd,pr}\bar{X}_{pr} \end{bmatrix} \mu_{suaa,su}(S_{su}, S_{aa}) = 0 \tag{36}$$
$$D_1 - Y_{suaa} [\mu_{suaa,su}(S_{su}, S_{aa}) + \mu_{suaa,aa}(S_{su}, S_{aa})] = 0.$$

It is straightforward to see that system (36) is equivalent to the following one:

$$A_{1}S_{aa}(S_{aa} + S_{su}) - A_{2}S_{su}^{2}(1 + S_{aa}) - A_{3}S_{aa}S_{su}(S_{aa} + S_{su}) = 0$$

$$B_{1}(S_{aa}^{2} + S_{su}^{2}) + B_{2}S_{aa}S_{su}(S_{aa} + S_{su}) - B_{3}S_{aa}S_{su} - B_{4}(S_{aa} + S_{su}) = 0,$$
(37)

where

$$\begin{aligned} A_1 &= D_1 S_{su}^{in} + \frac{D_1 k_{dis} X_c^{in}}{D_1 + k_{dis}} \left[\frac{k_{hyd,ch} f_{ch,xc}}{D_1 + k_{hyd,ch}} + \frac{f_{su,li} k_{hyd,li} f_{li,xc}}{D_1 + k_{hyd,li}} \right], \\ A_2 &= D_1 S_{aa}^{in} + \frac{D_1 k_{hyd,pr} f_{pr,xc} k_{dis} X_c^{in}}{(D_1 + k_{dis})(D_1 + k_{hyd,pr})}, \quad A_3 = D_1 k_{s,suaa}; \\ B_1 &= k_{s,suaa} (Y_{suaa} k_{m,suaa} - D_1), \quad B_2 = Y_{suaa} k_{m,suaa} - D_1, \\ B_3 &= 2D_1 k_{s,suaa}, \quad B_4 = D_1 k_{s,suaa}^2. \end{aligned}$$

All coefficients A_i , i = 1, 2, 3, and B_j , j = 1, 2, 3, 4, depend on D_1 (the dilution rate in the first bioreactor BR_1). Assume that there exists a value $D_1^{(2)} > 0$ of D_1 such that system (37) possesses

positive solutions \bar{S}_{aa} and \bar{S}_{su} for all $D_1 \in (0, D_1^{(2)})$. Denote by $D_1^{(3)}$ a third possible upper bound for D_1 , defined by

$$D_1^{(3)} = \min\{D_1 > 0: \ \bar{S}_{aa} < S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}\}.$$
(38)

Then, the equilibrium component \bar{X}_{suaa} will exist and will be positive for $D_1 \in (0, \min\{D_1^{(2)}, D_1^{(3)}\})$:

$$\bar{X}_{suaa} = \frac{D_1(S_{aa}^{in} - \bar{S}_{aa}) + k_{hyd,pr}\bar{X}_{pr}}{\mu_{suaa,aa}(\bar{S}_{su}, \bar{S}_{aa})}.$$
(39)

Note that an equality $\bar{S}_{aa} = S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}$ for some value of D_1 will lead to $\bar{X}_{suaa} = 0$ and may cause wash-out of the biomass X_{suaa} , which is practically undesirable, and we shall exclude this case.

Equation (17) is used to find the steady state component

$$\bar{S}_{ac} = \frac{1}{D_1} \left[(1 - Y_{suaa}) \left(f_{ac,su} \mu_{suaa,su} (\bar{S}_{su}, \bar{S}_{aa}) + f_{ac,aa} \mu_{suaa,aa} (\bar{S}_{su}, \bar{S}_{aa}) \right) \bar{X}_{suaa} + 0.7 (1 - Y_{fa}) \mu_{fa} (\bar{S}_{fa}) \bar{X}_{fa} \right].$$
(40)

The latter is positive if $Y_{suaa} < 1$ and $Y_{fa} < 1$ are fulfilled.

If $\bar{X}_{fa} = 0$, then the equilibrium component with respect to S_{ac} becomes

$$\bar{S}_{ac}^{0} = \frac{1}{D_{1}} (1 - Y_{suaa}) \left(f_{ac,su} \mu_{suaa,su}(\bar{S}_{su}, \bar{S}_{aa}) + f_{ac,aa} \mu_{suaa,aa}(\bar{S}_{su}, \bar{S}_{aa}) \right) \bar{X}_{suaa};$$
(41)

 \bar{S}_{ac}^0 exists and is positive if $Y_{suaa} < 1$ and $D_1 \in \left(D_1^{(1)}, \min\{D_1^{(2)}, D_1^{(3)}\}\right)$ hold true.

The above calculations deliver the following equilibrium points of the model (2)–(11):

$$\begin{split} E_1(D_1) &= (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}, \bar{S}_{ac}, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa}) \\ &\quad \text{if } 0 < D_1 < \min\{Y_{fa}k_{m,fa}, D_1^{(1)}, D_1^{(2)}, D_1^{(3)}\}; \\ E_1^0(D_1) &= (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}^0, \bar{S}_{ac}^0, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa} = 0) \\ &\quad \text{if } D_1^{(1)} < D_1 < \min\{D_1^{(2)}, D_1^{(3)}\}. \end{split}$$

Consider now Equations (12) and (13) to compute the equilibrium components with respect to X_{ac} and $S_{ac,ch4}$. Thereby, we assume that \bar{S}_{ac} and \bar{S}_{ac}^0 are known; see (40) and (41), respectively.

Equation (25) has the obvious solution $X_{ac} \equiv 0$, which is not of practical interest. Assuming that $X_{ac} \neq 0$, the equality $\mu_{ac,ch4}(S_{ac,ch4}) = D_2$ delivers the following steady state component for $S_{ac,ch4}$

$$\bar{S}_{ac,ch4} = \frac{k_{s,ac} D_2}{k_{m,ac} - D_2}.$$
(42)

Obviously, there exists $\bar{S}_{ac,ch4} > 0$ if $D_2 < k_{m,ac}$ holds true. Then, the corresponding equilibrium component \bar{X}_{ac} is

$$\bar{X}_{ac} = \frac{1}{Y_{ac}}(\bar{S}_{ac} - \bar{S}_{ac,ch4}).$$
 (43)

The latter is positive if

$$\bar{S}_{ac} > \bar{S}_{ac,ch4}.\tag{44}$$

If we consider \bar{S}_{ac}^0 instead of \bar{S}_{ac} we obtain

$$\bar{X}_{ac}^{0} = \frac{1}{\gamma_{ac}} (\bar{S}_{ac}^{0} - \bar{S}_{ac,ch4}), \tag{45}$$

which is positive if

$$\bar{S}_{ac}^0 > \bar{S}_{ac,ch4}.\tag{46}$$

The inequalities (44) and (46) give relationships between D_1 and D_2 because \bar{S}_{ac} and \bar{S}_{ac}^0 depend also on D_1 ; see (40) and (41), respectively.

Therefore, the extended model (2)–(13) can possess the following two equilibrium points, parameterized on D_1 and D_2 :

$$E(D_1, D_2) = E_1(D_1) \cup (S_{ac,ch4}, \bar{X}_{ac}),$$

$$E^0(D_1, D_2) = E_1^0(D_1) \cup (\bar{S}_{ac,ch4}, \bar{X}_{ac}^0).$$

We summarize the above calculations in the following theorem.

Theorem 2. Let $0 < Y_{suaa} < 1$ and $0 < Y_{fa} < 1$ be fulfilled in the model Equations (2)–(13). Assume that there exists a value $D_1^{(2)} > 0$ such that positive solutions \bar{S}_{aa} and \bar{S}_{su} of system (37) do exist for $D_1 \in (0, D_1^{(2)})$. Then,

- (i) the equilibrium point $E(D_1, D_2)$ exists if $0 < D_1 < \min\{Y_{fa}k_{m,fa}, D_1^{(1)}, D_1^{(2)}, D_1^{(3)}\}$ and $0 < D_2 < \min\{k_{m,ac}, \min\{D_2 > 0 : \bar{S}_{ac} > \bar{S}_{ac,ch4}\}\}$;
- (ii) the equilibrium point $E^0(D_1, D_2)$ exists if $D_1^{(1)} < D_1 < \min\{D_1^{(2)}, D_1^{(3)}\}$ and $0 < D_2 < \min\{k_{m,ac}, \min\{D_2 > 0 : \bar{S}_{ac}^0 > \bar{S}_{ac,ch4}\}\}$.

Theorem 2 can be specified using the coefficient values in Table 2. Numerical calculations produce the following values:

$$Y_m k_{m,fa} = 0.009, \ D_1^{(1)} \approx 0.00843, \ D_1^{(2)} \approx 0.0942.$$
 (47)

The left plot in Figure 2 visualizes the equilibrium component \bar{S}_{aa} , which exists for $D_1 \in (0, D_1^{(2)})$, as well as the upper bound $S_{aa}^{in} + \frac{1}{D_1} k_{hyd,pr} \bar{X}_{pr}$ of \bar{S}_{aa} according to (35). Obviously, (35) is satisfied for $D_1 \in (0, D_1^{(2)})$ so that $D_1^{(2)} \leq D_1^{(3)}$. The right plot in Figure 2 shows the equilibrium component $\bar{X}_{suaa}(D_1)$, which vanishes at $D_1 = D_1^{(2)}$.

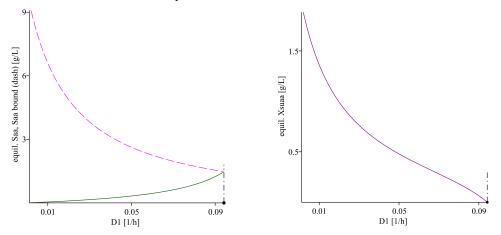


Figure 2. Left plot: the equilibrium point \bar{S}_{aa} (solid line) and the upper bound of \bar{S}_{aa} (dash line) according to (35) as functions of D_1 . Right plot: the graph of the equilibrium point \bar{X}_{suaa} . The vertical dash-dot lines pass through $D_1^{(2)}$.

Furthermore, we have $k_{m,ac} = 0.0167$ and

$$\bar{S}_{ac} > \bar{S}_{ac,ch4} \iff D_2 \in (0, 0.0156) =: (0, D_2^{(2)}).$$
 (48)

This relation is demonstrated in the left plot of Figure 3. The steady state component $\bar{S}_{ac} = \bar{S}_{ac}(D_1)$ (solid line) exists for $D_1 \in (0, D_1^{(1)})$. The dash line represents the equilibrium component $\bar{S}_{ac,ch4} = \bar{S}_{ac,ch4}(D_2)$. The horizontal dot line passes through the points $(D_1^{(1)}, \bar{S}_{ac}(D_1^{(1)}))$ and $(D_2^{(2)}, \bar{S}_{ac,ch4}(D_2^{(2)})$. The vertical dash-dot lines pass through $D_1^{(1)}$ and $D_2^{(2)}$. This implies the following relation:

 $E(D_1, D_2)$ exists $\iff D_1 \in (0, D_1^{(1)}) = (0, 0.00843), \quad D_2 \in (0, D_2^{(2)}) = (0, 0.0156).$

Similarly, the right plot in Figure 3 visualizes the relation

$$\bar{S}_{ac}^0 > \bar{S}_{ac,ch4} \iff D_2 \in (0, 0.0152) =: (0, D_2^{(1)})$$

because the horizontal dot line passes through the points $(D_1^{(1)}, \bar{S}_{ac}^0(D_1^{(1)}))$ and $(D_2^{(1)}, \bar{S}_{ac,ch4}(D_2^{(1)}))$. This leads to the following relation:

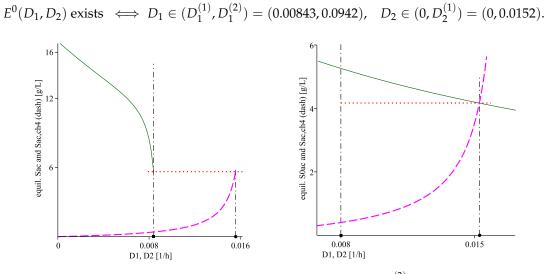


Figure 3. Existence of the equilibrium component $\bar{S}_{ac,ch4}$ for $D_2 \in (0, D_2^{(2)})$ (left plot) and for $D_2 \in (0, D_2^{(1)})$ (right plot).

The equilibrium $E(D_1, D_2)$ corresponds to coexistence of all microorganisms in both bioreactors BR_1 and BR_2 , whereas $E^0(D_1, D_2)$ is related to wash-out of the LCFA degrader X_{fa} in BR_1 for hydrogen production.

4.2. Local Asymptotic Stability of the Equilibrium Points

We shall first investigate the local asymptotic stability of the coexistence equilibrium $E_1(D_1)$ for $D_1 \in (0, D_1^{(1)})$ and of the wash-out steady state (within $\bar{X}_{fa} = 0$) $E_1^0(D_1)$ for $D_1 \in (D_1^{(1)}, D_1^{(2)})$, i.e., of the equilibrium points corresponding to the model (2)–(11) of the first bioreactor BR_1 for hydrogen production.

It is well known that an equilibrium point is locally asymptotically stable, if all eigenvalues of the Jacobi matrix evaluated at this equilibrium have negative real parts, cf. e.g., [33]. The eigenvalues of this Jacobi matrix coincide with the roots of the corresponding characteristic polynomial.

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Detailed calculations of the characteristic polynomials $P(E_1; \lambda)$ and $P(E_1^0; \lambda)$ with respect to the equilibrium points $E_1 = E_1(D_1)$ and $E_1^0 = E_1^0(D_1)$ are given in Appendix A.2 of the Appendix A. We present them here for reader's convenience:

$$\begin{split} P(E_{1};\lambda) &= (-D_{1}-\lambda)(-D_{1}-k_{dis}-\lambda)(-D_{1}-k_{hyd,ch}-\lambda) \\ &\times (-D_{1}-k_{hyd,pr}-\lambda)(-D_{1}-k_{hyd,li}-\lambda) \\ &\times \left[\lambda^{2} + \left(D_{1} + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_{1}\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right] \\ &\times R(E_{1};\lambda), \quad D_{1} \in (0,D_{1}^{(1)}); \\ P(E_{1}^{0};\lambda) &= (-D_{1}-\lambda)^{2}(-D_{1}-k_{dis}-\lambda)(-D_{1}-k_{hyd,ch}-\lambda) \\ &\times (-D_{1}-k_{hyd,pr}-\lambda)(-D_{1}-k_{hyd,li}-\lambda)(-D_{1}+Y_{fa}\mu_{fa}(\bar{S}_{fa}^{0})-\lambda) \\ &\times R(E_{1}^{0};\lambda), \quad D_{1} \in (D_{1}^{(1)},D_{1}^{(2)}), \end{split}$$

where $R(E_1; \lambda)$ is the characteristic polynomial of the sub-matrix $\Delta^{(3)}$ evaluated at the steady state components $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (0, D_1^{(1)})$, and $R(E_1^0; \lambda)$ is the characteristic polynomial of the sub-matrix $\Delta^{(3)}$ evaluated at the steady state components $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (D_1^{(1)}, D_1^{(2)})$, see Appendix A.2 in the Appendix A.

The linear quantities in $P(E_1; \lambda) = 0$ and $P(E_1^0; \lambda) = 0$ produce negative real eigenvalues of the Jacobi matrices. In the first polynomial $P(E_1; \lambda)$, the quadratic equation

$$\lambda^2 + \left(D_1 + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_1\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa} = 0$$

possesses two real negative roots. The multiplier $-D_1 + Y_{fa}\mu_{fa}(S_{fa}^0) - \lambda$ in the second polynomial $P(E_1^0;\lambda)$ delivers a negative root $\lambda = -D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}^0)$ because, in this case, $D_1 > Y_{fa}\mu_{fa}(\bar{S}_{fa}^0)$ holds true.

It remains to show that $R(E_1; \lambda) = 0$ and $R(E_1^0; \lambda) = 0$ have roots with negative real parts. These two problems are solved numerically using the coefficient values in Table 2. The real parts of the eigenvalues are presented graphically in Figures 4 and 5, and are obviously negative. Therefore, the steady states $E_1 = E_1(D_1)$, $D_1 \in (0, D_1^{(1)})$ and $E_1^0 = E_1^0(D_1)$, $D_1 \in (D_1^{(1)}, D_1^{(2)})$, are locally asymptotically stable.

Consider again the polynomial $P(E_1; \lambda)$. We know that at $D_1 = D_1^{(1)}$ the equilibrium component \bar{X}_{fa} becomes equal to zero, and thus $P(E_1; \lambda) = 0$ has a root $\lambda = 0$. This means that a transcritical bifurcation at $D_1 = D_1^{(1)}$ occurs, leading to wash-out of the LCFA degraders X_{fa} . As a result, the steady state $E_1(D_1)$ no longer exists for $D_1 > D_1^{(1)}$, and $E_1^0(D_1)$ remains the locally asymptotically stable equilibrium for $D_1 \in (D_1^{(1)}, D_1^{(2)})$.

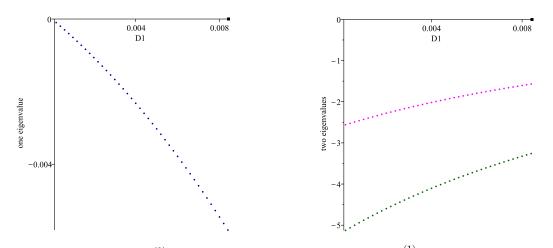


Figure 4. Eigenvalues of $\Delta^{(3)}$ evaluated at $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (0, D_1^{(1)})$. The solid box on the horizontal axis denotes $D_1^{(1)}$.

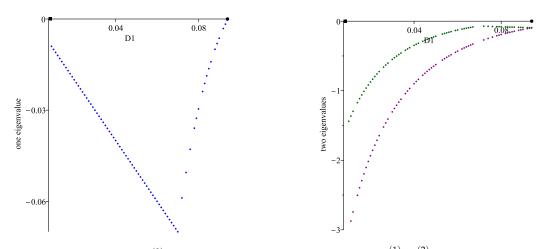


Figure 5. Eigenvalues of $\Delta^{(3)}$ evaluated at $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (D_1^{(1)}, D_1^{(2)})$. The solid box and solid circle on the horizontal axis denote $D_1^{(1)}$ and $D_1^{(2)}$, respectively.

The equilibrium points of the model (12)–(13) of the second bioreactor BR_2 for methane production are known to be locally asymptotically stable for any admissible value of D_2 (see, e.g., [20]). Indeed, the characteristic polynomials of the Jacobi matrices of (12)–(13) evaluated at the equilibrium points $(\bar{S}_{ac,ch4}, \bar{X}_{ac})$ and $(\bar{S}_{ac,ch4}, \bar{X}_{ac}^0)$, have the form

$$\lambda^2 + A \lambda + B = 0$$
 and $\lambda^2 + A_0 \lambda + B_0 = 0$,

where

$$A = D_{2} + Y_{ac} \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac} > 0, \quad B = Y_{ac} \mu_{ac,ch4} (\bar{S}_{ac,ch4}) \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac} > 0,$$

$$A_{0} = D_{2} + Y_{ac} \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac}^{0} > 0, \quad B_{0} = Y_{ac} \mu_{ac,ch4} (\bar{S}_{ac,ch4}) \frac{d\mu_{ac,ch4}}{dS_{ac,ch4}} (\bar{S}_{ac,ch4}) \bar{X}_{ac}^{0} > 0$$

The positive signs of *A*, *B* and *A*₀, *B*₀ imply that the above quadratics possess two roots with negative real parts. This proves the local asymptotic stability of the equilibrium points $E(D_1, D_2)$ for $D_1 \in (0, D_1^{(1)}), D_2 \in (0, D_2^{(2)})$ and $E^0(D_1, D_2)$ for $D_1 \in (D_1^{(1)}, D_1^{(2)}), D_2 \in (0, D_2^{(1)})$.

5. Optimization of Hydrogen and Methane Flow Rates

In this section, we shall find values of the dilution rates D_1 and D_2 , so that the hydrogen and methane flow rates Q_{h2} and Q_{ch4} evaluated at the model equilibrium points achieve maximal values.

We compute Q_{h2} on the two sets of steady states $E_1(D_1)$, $D_1 \in (0, D_1^{(1)})$ and $E_1^0(D_1)$, $D_1 \in (D_1^{(1)}, D_1^{(2)})$:

$$Q_{h2} = Q_{h2}(D_1) = Y_{h2,su} \,\mu_{suaa,su}(\bar{S}_{su}, \bar{S}_{aa})\bar{X}_{suaa} + Y_{h2,aa}\mu_{suaa,aa}(\bar{S}_{su}, \bar{S}_{aa})\bar{X}_{suaa} + Y_{h2,fa} \,\mu_{fa}(\bar{S}_{fa})\bar{X}_{fa}, \quad D_1 \in \left(0, D_1^{(1)}\right) Q_{h2}^0 = Q_{h2}^0(D_1) = Y_{h2,su} \,\mu_{suaa,su}(\bar{S}_{su}, \bar{S}_{aa})\bar{X}_{suaa}$$

$$+ Y_{h2,aa} \mu_{suaa,aa}(\bar{S}_{su}, \bar{S}_{aa}) \bar{X}_{suaa}, \quad D_1 \in \left(D_1^{(1)}, D_1^{(2)}\right)$$

Recall that the superscript 0 in the equilibrium $E_1^0(D_1)$ indicates $\bar{X}_{fa} = 0$.

The functions $Q_{h2}(D_1)$ and $Q_{h2}^0(D_1)$ depend on the decision (manipulated) input D_1 and are called input–output static characteristics with respect to the hydrogen production.

The important (from practical point of view) question is whether the functions $Q_{h2}(D_1)$ and $Q_{h2}^0(D_1)$ are unimodal with respect to D_1 in the admissible intervals for D_1 . For example, the function $Q_{h2}(D_1)$ is called unimodal if there exists a unique (admissible) point $D_{1,\text{max}}$, such that $Q_{h2}(D_1)$ possesses maximum $Q_{h2,\text{max}} = Q_{h2}(D_{1,\text{max}})$, $Q_{h2}(D_1)$ is strongly increasing if $D_1 < D_{1,\text{max}}$ and $Q_{h2}(D_1)$ is strongly decreasing if $D_1 > D_{1,\text{max}}$.

Figure 6 (left plot) presents the graphs of the input-output static characteristics $Q_{h2}(D_1)$ for $D_1 \in (0, D_1^{(1)})$ (solid line) and $Q_{h2}^0(D_1)$ for $D_1 \in (D_1^{(1)}, D_1^{(2)})$ (dash line). The function $Q_{h2}(D_1)$ achieves its maximum at $D_{1,\max} \approx 0.0072 \in (0, D_1^{(1)})$ and $Q_{h2,\max} \approx 0.115 [L/h]$. The maximum of $Q_{h2}^0(D_1)$ is achieved at $D_{1,\max}^0 \approx 0.0439 \in (D_1^{(1)}, D_1^{(2)})$ and $Q_{h2,\max}^0 \approx 0.169 [L/h]$. Therefore, the absolute maximum of $Q_{h2}(D_1) \cup Q_{h2}^0(D_1)$, $D_1 \in (0, D_1^{(2)})$ coincides with the maximum of $Q_{h2}^0(D_1)$. From a practical point of view, this means that the maximal flow rate of H_2 could be achieved at a operation mode, related to wash-out of the LCFA degraders X_{fa} in BR_1 .

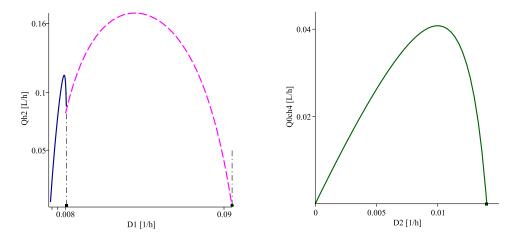


Figure 6. The input–output static characteristics $Q_{h2}(D_1) \cup Q_{h2}^0(D_1)$ for $D_1 \in (0, D_1^{(2)})$ (left) and $Q_{ch4}^0(D_{1,max}^0, D_2)$ for $D_2 \in (0, D_2^{(1)})$ (right). The vertical dash-dot lines in the left plot pass through $D_1^{(1)}$ (solid box) and $D_1^{(2)}$ (solid circle). In the right plot, $D_2^{(1)}$ is marked by a sold box.

A similar question of maximizing the biogas production at steady states has been discussed in [34] for the well known four-dimensional AM2 model. There, two maxima of the input–output static characteristics with respect to methane flow rate have also been determined, related to coexistence or wash-out of the acidogenic bacteria in the bioreactor.

Using the methane flow rate Q_{ch4} , we compute further the input–output static characteristics Q_{ch4} and Q_{ch4}^0 on the set of the steady states $E(D_1, D_2)$ and $E^0(D_1, D_2)$ respectively, namely:

$$Q_{ch4} = Q_{ch4}(D_1, D_2) = Y_{ch4,ac} \,\mu_{ac,ch4}(\bar{S}_{ac,ch4}) \,\bar{X}_{ac},$$
$$Q_{ch4}^0 = Q_{ch4}(D_1, D_2) = Y_{ch4,ac} \,\mu_{ac,ch4}(\bar{S}_{ac,ch4}) \,\bar{X}_{ac}^0.$$

The latter are parameterized on the two inputs D_1 and D_2 , since \bar{X}_{ac} and \bar{X}_{ac}^0 depend on both D_1 and D_2 ; see (43) and (45), respectively.

With $D_1 = D_{1,\max}^0 \approx 0.0439$, we compute the equilibrium components $\bar{S}_{ac,ch4}$ and \bar{X}_{ac}^0 according to (42) and (45), as well as Q_{ch4}^0 , all of them as functions of D_2 .

Figure 6 (right plot) visualizes the graph of $Q_{ch4}^0(D_{1,\max}^0, D_2)$ for $D_2 \in (0, D_2^{(1)})$; the latter is a unimodal function, taking its maximum at $D_{2,\max}^0 \approx 0.00991$ [1/*h*] with $Q_{ch4,\max}^0 = Q_{ch4}(D_{1,\max}^0, D_{2,\max}^0) \approx 0.0388$ [*L*/*h*].

The latter results are theoretical and predicted by the model studies. Experimental data that confirm these results by considering H_2 and CH_4 volumetric production under similar conditions for which our model was developed (i.e., two bioreactors in a cascade with a similar ratio of working volumes and operating in continuous mode) can be found in [35].

Using the presentation (1), we define and compute the constant γ and in this way the relationship between the volumes V_1 and V_2 of the two bioreactors:

$$\gamma = \frac{D_{2,\max}^0}{D_{1,\max}^0} \approx 0.226 \quad \Longrightarrow \quad V_2 \approx 4.42 \, V_1.$$

The last relation implies that the volume of the second bioreactor BR_2 for methane production should be at least 4.4 times greater than the volume of the first bioreactor BR_1 for hydrogen production.

6. Dynamic Behavior of the Model Solutions

The dynamic model has been implemented as one-page application "(H2,CH4) Bioreactors ver. 1.2" in a web-based simulation software SmoWeb, see http://platform.sysmoltd.com/. SmoWeb is an open computational platform in *Python* and a library of applications is built on top of this platform. It does not require installation of any software by the user, since all computations are performed in a web cloud. *Python* is used as the implementation language because it is a powerful modern general purpose object-oriented language. Nowadays, *Python* has become an inseparable part of the scientific computing due to the vast abundance of open source libraries and interfaces to major software tools. A number of one-page applications have already been and are continuously being developed on top of the functionality provided by the platform, including bioprocess Modeling as well.

All figures below are developed using the one-page application "(H2,CH4) Bioreactors ver. 1.2". The plots illustrate the dynamic behavior of the model solutions when D_1 , D_2 and the input concentration of the composites X_c^{in} take different values from Tables 3 and 4, respectively.

Time [h]	0–5000	5000-10,000	10,000–15,000	15,000-20,000
$D_1 [1/h]$	0.005	0.01	0.015	0.02
$D_2 \ [1/h]$	0.00113	0.00226	0.00339	0.00452

Figure 7 presents the time evolution of the flow rates $Q_{h2}(t)$ and $Q_{ch4}(t)$ when D_1 and D_2 take values from Table 3. Figures 8 and 9 display the model solutions by variable D_1 and D_2 given in Table 3. It is seen that, if $D_1 \in (0, D_1^{(1)}) = (0, 0.00843)$ and $D_2 \in (0, D_2^{(2)}) = (0, 0.0156)$, then the model solutions tend to the corresponding components of the locally stable coexistence equilibrium

 $E(D_1, D_2)$. When $D_1 \in (D_1^{(1)}, D_1^{(2)}) = (0.00843, 0.0942)$ and $D_2 \in (0, D_2^{(1)}) = (0, 0.0152)$, then the solutions approach the locally stable wash-out equilibrium (with $\bar{X}_{fa} = 0$) after sufficiently large time t [h]; in particular, one can see in the left plot of Figure 9 that $X_{fa}(t)$ approaches 0 for sufficiently large time t [h].

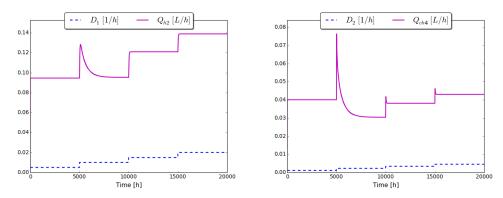


Figure 7. Time evolution of Q_{h2} and Q_{ch4} with variable D_1 and D_2 from Table 3.

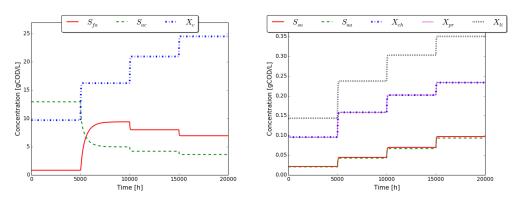


Figure 8. Dynamic behavior of the model solutions with variable D_1 from Table 3.

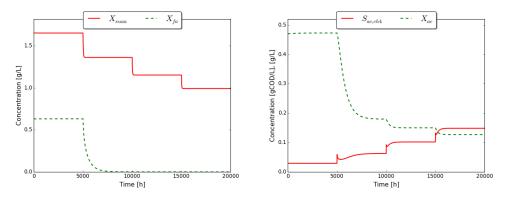


Figure 9. Dynamic behavior of the model solutions with variable D_1 from Table 3.

Table 4. Values of X_c^{in} .

Time [h]	0–5000	5000-10,000	10,000–15,000	15,000-20,000
$X_c^{in} \ [gCOD/L]$	40	50	75	60

Figures 10–12 demonstrate the dynamics of $Q_{h2}(t)$, $Q_{ch4}(t)$ as well as of the model solutions, when D_1 and D_2 are fixed, $D_1 = 0.005 [1/h]$, $D_2 = 0.00113 [1/h]$, but the input concentration of composites X_c^{in} varies taking values from Table 4. The model solutions tend to the corresponding

components of the coexistence equilibrium $E(D_1, D_2)$, which is the locally stable steady state. These numerical simulations demonstrate the fact that uncertainties in the input X_c^{in} do not affect the stability of the model dynamics.

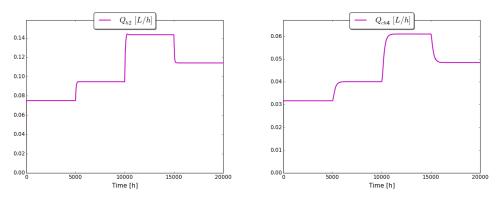


Figure 10. Time evolution of Q_{h2} and Q_{ch4} with variable X_c^{in} from Table 4.

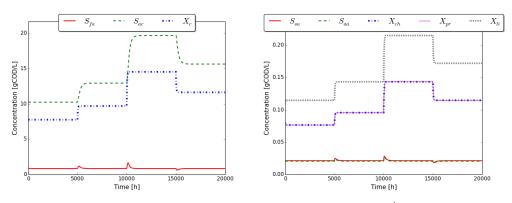


Figure 11. Dynamic behavior of the model solutions with variable X_c^{in} from Table 4.

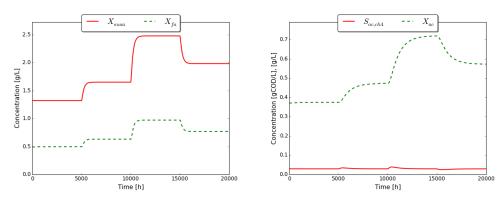


Figure 12. Dynamic behavior of the model solutions with variable X_c^{in} from Table 4.

7. Discussion

In this paper, we propose a mathematical model describing the process of simultaneous production of hydrogen and methane by anaerobic digestion of organic wastes in a cascade of two connected continuously stirred bioreactors with different volumes. The proposed model is developed by adapting and reducing the universal Anaerobic Digester Model No 1 (ADM1). The equations in our model are separated in two groups, corresponding to the processes in the two bioreactors, for hydrogen and for methane production, respectively. The simplifications made in the two models naturally lead to some limitations compared to the original ADM1. However, as already shown, the latter also does not cover all possible reactions, e.g., possible inclusion in the model of the activity of the so called syntrophicacetate oxidizers [36]. Some new, more complex models have also been developed [37], which however have the same main disadvantage as ADM1—great complexity and thus impossibility to carry out some mathematical analytical studies.

The investigation of the hydrogen and methane flow rates, Q_{h2} and Q_{ch4} respectively, on steady state operation, shows existence of maxima for some values of the dilution rates D_1 and D_2 in the two bioreactors BR_1 and BR_2 . The local maximum for D_1 in BR_1 reflects the maximal yield of hydrogen from lipids at very long hydraulic retention time HRT = $1/D_{1,max} \approx 1/0.0072 \approx 139$ [hours] ≈ 5.8 [days]. This maximum is much less than the maximum yield of hydrogen derived from monosaccharides and aminoacids (proteins) for the accepted combination of coefficients and initial conditions reflecting the biodegradation of organic wastes. In [35] under similar conditions, results close to our theoretical model-based predictions are obtained.

The model also allows us to find the optimal ratio between the volumes V_1 and V_2 of the two bioreactors subject to the same optimization goal. Many studies in the literature report on TSAD processes; however, most of them are operated manually and the ratio of working volumes of CSTRs is not discussed. In some references, a ratio of 10 is accepted without comments. In [38], a TSAD system consisting of two CSTRs operating at mesophilic conditions are used to investigate the effect of HRT on hydrogen and methane production; the ratio of working volumes of the bioreactors is equal to 8 (without comments). In [39], optimization of TSAD of separately collected municipality bio-waste is carried out in pilot scale CSTRs at thermophilic temperature, using recirculation of the digestate in the second bioreactor to maintain pH in the first bioreactor at optimal value. There, the ratio of working volumes of the bioreactors is equal to 3.8 (without comments). In [40], optimal loading rate is obtained providing maximum H_2 and CH_4 productions in TSAD of cassava wastewaters using specific thermophilic bioreactors and constant recycling ratio of 1:1 with automatic control of pH in the hydrogenic bioreactor. The ratio of working volumes of both bioreactors is equal to 6 (without explanations). The review article [41] is the first one that combines optimization approaches for three possible products from AD-methane, hydrogen and volatile fatty acids (VFA), taking into account different process parameters and types of bioreactors, including acidogenesis and methanogenesis separation in two different bioreactors. However, the ratio of the bioreactors working volumes is not given and discussed.

Using the adapted values of the coefficients in our model, we obtain an optimal ratio of the working volumes of both bioreactors equal to 4.42 using a criterion for maximizing the bioenergy production. This value falls into the range of the above reported values; however, it is theoretically derived.

8. Conclusions

In this paper, we consider a mathematical model describing a biotechnological process of hydrogen and methane production by anaerobic digestion. The process is carried out in two connected continuously stirred bioreactors, thus the model equations are separated in two groups, describing the processes in each bioreactor. We establish existence and uniqueness of positive and uniformly bounded solutions of the model dynamics. Two equilibrium points of the model are computed and their local asymptotic stability is studied with respect to the practically important input parameters, the dilution rates D_1 and D_2 in the two bioreactors. The first equilibrium point corresponds to coexistence of substrates and biomass, and the second one is related to wash-out of the LCFA degraders in the first bioreactor. We show that there exists a value of the dilution rate D_1 , at which the coexistence equilibrium disappears as a result of bifurcation of the steady state and the wash-out equilibrium remains the only locally asymptotically stable equilibrium. The investigation of the hydrogen and methane flow rates, Q_{h2} and Q_{ch4} , on steady state operation, shows existence of maxima at some values of the dilution rates D_1 and D_2 . It is worth noting that the maximal hydrogen flow rate is achieved at the steady state where the LCFA degrader X_{fa} is washed out. This is established here by taking numerical values for the model parameters from the literature [31]. Nevertheless, this fact could

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help in the design of an AD process to achieve maximal production of either hydrogen and methane. The future work envisages extending the model of the second bioreactor by equations describing an acetogenesis phase, where other VFA (propionate, butyrate and valeriate) will be transformed into acetate, as well as including inhibitory kinetics of the microorganisms growth. The elaborated web-based software will allow us to check different model-based optimization and control strategies that will contribute to designing and engineering a real process.

Author Contributions: M.B.: software development, simulation and visualization, writing and editing; N.D.: theoretical investigation of the model—existence and uniqueness of solutions, stability analysis of equilibrium points, writing and editing; I.S.: model development and validation, discussion and evaluation of the results, editing. All authors have read and agreed to the published version of the manuscript.

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Conflicts of Interest: The authors declare no conflict of interest.

Appendix A

Appendix A.1. Proof of Theorem 1

Equation (6) is linear with respect to X_c and can be solved explicitly:

$$X_{c}(t) = X_{c}(0)e^{-(D_{1}+k_{dis})t} + \frac{D_{1}X_{c}^{in}}{D_{1}+k_{dis}}.$$

The solution $X_c(t)$ obviously exists for all $t \ge 0$, is positive, and is uniformly bounded. In practice, it is reasonable to consider only positive initial conditions $X_c(0) > 0$.

Assume that $X_{ch}(0) = 0$. Then, Equation (7) implies $\dot{X}_{ch}(0) = f_{ch,xc}k_{dis}X_c(0) > 0$ and so there exists $\varepsilon > 0$ such that $X_{ch}(t) > 0$ for each $t \in (0, \varepsilon)$. Let $X_{ch}(0) \ge 0$ and suppose that there exists $\bar{t}_{ch} > 0$ such that $X_{ch}(t) > 0$ for each $t \in (0, \bar{t}_{ch})$ and $X_{ch}(\bar{t}_{ch}) = 0$. Then, $\dot{X}_{ch}(\bar{t}_{ch}) \le 0$, but it follows from (7) that $\dot{X}_{ch}(\bar{t}_{ch}) > 0$, a contradiction. Thus, $X_{ch}(t) > 0$ for all $t \ge 0$.

Similar arguments as above can be applied to X_{pr} and X_{li} from Equations (8) and (9), respectively, so that $X_{pr}(t) > 0$ and $X_{li}(t) > 0$ for all $t \ge 0$.

Equation (10) implies that, if $X_{suaa}(0) = 0$, then $X_{suaa}(t) = 0$ for all $t \in [0, \infty)$ due to the uniqueness of solutions of Cauchy problems. Assuming that $X_{suaa}(0) > 0$, we obtain

$$X_{suaa}(t) = X_{suaa}(0) e^{\int_0^t [-D_1 + Ysuaa(\mu_{suaa,su}(S_{su}(\xi), S_{aa}(\xi)) + \mu_{suaa,aa}(S_{su}(\xi), S_{aa}(\xi)))]d\xi} > 0 \text{ for all } t \ge 0.$$

Similarly, if $X_{fa}(0) = 0$, then it follows from Equation (11) that $X_{fa}(t) = 0$ for all $t \in [0, \infty)$. Thus, it is reasonable to consider only positive initial conditions $X_{fa}(0) > 0$. Then, we have

$$X_{fa}(t) = X_{fa}(0) e^{-\int_0^t \left[D_1 - Y_{fa} \mu_{fa}(S_{fa}(\xi)) \right] d\xi} > 0 \text{ for all } t \ge 0.$$

Assume now that $X_{fa}(0) > 0$, $X_{suaa}(0) > 0$ and $X_j(0) \ge 0$ for $j \in \{c, ch, pr, li\}$. If $S_{su}(0) = 0$, then it follows from Equation (2) that

$$\dot{S}_{su}(0) = D_1 S_{su}^{in} + k_{hyd,ch} X_{ch}(0) + f_{su,li} k_{hyd,li} X_{li}(0) > 0,$$

thus there exists $\varepsilon > 0$ with $S_{su}(t) > 0$ for $t \in (0, \varepsilon)$. Let $S_{su}(0) \ge 0$. Assume that there exists $\bar{t}_{su} > 0$ such that $S_{su}(t) > 0$ for $t \in (0, \bar{t}_{su})$ and $S_{su}(\bar{t}_{su}) = 0$. Then, $\dot{S}_{su}(\bar{t}_{su}) \le 0$, but, from (2), it follows that

 $\dot{S}_{su}(\bar{t}_{su}) > 0$, a contradiction. Therefore, $S_{su}(t) > 0$ for all t > 0. Using Equations (3) and (4), one can show in a similar way that $S_{aa}(t) > 0$ and $S_{fa}(t) > 0$ for all $t \ge 0$.

Denote

$$\Sigma_1 = X_{ch} + f_{ch,xc} X_c, \ \Sigma_1^{in} = f_{ch,xc} X_c^{in}$$

Since $X_{ch}(t) > 0$ for all $t \ge 0$, it follows from Equations (6) and (7) that

$$\dot{\Sigma}_{1} = \dot{X}_{ch} + f_{ch,xc}\dot{X}_{c} = D_{1}\left(f_{ch,xc}X_{c}^{in} - (X_{ch} + f_{ch,xc}X_{c})\right) - k_{hyd,ch}X_{ch} \le D_{1}\left(\Sigma_{1}^{in} - \Sigma_{1}\right).$$

Multiplying both sides of the inequality by $e^{D_1 t} > 0$, we obtain $e^{D_1 t} \dot{\Sigma}_1 + D_1 e^{D_1 t} \Sigma_1 \leq e^{D_1 t} D_1 \Sigma_1^{in}$, which is equivalent to $\frac{d}{dt} (e^{D_1 t} \Sigma_1) \leq e^{D_1 t} D_1 \Sigma_1^{in}$, and further

$$\int_0^t \frac{d}{d\xi} e^{D_1\xi} \Sigma_1(\xi) d\xi \le D_1 \Sigma_1^{in} \int_0^t e^{D_1\xi} d\xi,$$

thus finally

$$\Sigma_1 \le e^{-D_1 t} \Sigma_1(0) + \Sigma_1^{in} \left(1 - e^{-D_1 t} \right).$$

The latter inequality means that $\limsup_{t\to\infty} \Sigma_1(t) \leq \Sigma_1^{in}$. Since $X_c(t)$ is positive and bounded, and $X_{ch}(t)$ is positive, this implies that $X_{ch}(t)$ is uniformly bounded and exists for all $t \in [0, \infty)$.

Denote $\Sigma_2 = X_{pr} + f_{pr,xc}X_c$, $\Sigma_2^{in} = f_{pr,xc}X_c^{in}$. Then, using the fact that $X_{ch}(t) \ge 0$ for all $t \ge 0$, we obtain from Equations (6) and (9)

$$\dot{\Sigma}_2 = D_1 \left(f_{pr,xc} X_c^{in} - X_{pr} - f_{pr,xc} X_c \right) - k_{hyd,pr} X_{pr} \le D_1 \left(\Sigma_2^{in} - \Sigma_2 \right),$$

which means that $\limsup_{t\to\infty} \Sigma_2(t) \le \Sigma_2^{in}$. Since $X_c(t)$ is positive and bounded, the latter inequality implies that $X_{pr}(t)$ is also uniformly bounded, and exists for all $t \in [0, \infty)$.

Define $\Sigma_3 = X_{li} + f_{li,xc}X_c$ and $\Sigma_3^{in} = f_{li,xc}X_c^{in}$. Then, using Equations (6) and (9) and the fact that $X_{li}(t) \ge 0$ for all $t \ge 0$, we obtain

$$\dot{\Sigma}_{3} = D_{1} \left(f_{li,xc} X_{c}^{in} - X_{li} - f_{li,xc} X_{c} \right) - k_{hyd,li} X_{li} \leq D_{1} \left(\Sigma_{3}^{in} - \Sigma_{3} \right),$$

which implies $\limsup_{t\to\infty} \Sigma_3(t) \leq \Sigma_3^{in}$. Since $X_c(t)$ is positive and bounded, the latter inequality means that $X_{li}(t)$ is also uniformly bounded, and exists for all $t \in [0, \infty)$.

Consider now

$$\Sigma_4 = S_{fa} + \frac{1}{Y_{fa}} X_{fa} + f_{fa,li} X_{li} + f_{fa,li} f_{li,xc} X_c, \quad \Sigma_4^{in} = S_{fa}^{in} + f_{fa,li} f_{li,xc} X_c^{in}.$$

Then, we obtain from Equations (6), (9), and (11)

$$\dot{\Sigma}_4 = D_1 \left(\Sigma_4^{in} - \Sigma_4 \right)$$
, thus $\Sigma_4(t) = \Sigma_4(0) e^{-D_1 t} + \Sigma_4^{in} \left(1 - e^{-D_1 t} \right) > 0$.

Since $X_c(t)$ and $X_{li}(t)$ are positive and bounded, $S_{fa}(t)$ and $X_{fa}(t)$ are positive, the latter inequality implies that $S_{fa}(t)$ and $X_{fa}(t)$ are uniformly bounded and exists for all $t \in [0, \infty)$.

Further consider

$$\Sigma_5 = S_{su} + S_{aa} + \frac{1}{Y_{suaa}} X_{suaa} + X_{ch} + X_{pr} + f_{su,li} X_{li} + (f_{ch,xc}k_{dis} + f_{li,xc}f_{su,li} + f_{pr,xc}) X_c,$$

$$\Sigma_5^{in} = S_{su}^{in} + S_{aa,in} + (f_{ch,xc}k_{dis} + f_{li,xc}f_{su,li} + f_{pr,xc}) X_c^{in}.$$

Then, we obtain

$$\dot{\Sigma}_5 = D_1 \left(\Sigma_5^{in} - \Sigma_5 \right) \ \Rightarrow \ \Sigma_5(t) = \Sigma_5(0) e^{-D_1 t} + \Sigma_5^{in} \left(1 - e^{-D_1 t} \right).$$

Since $X_c(t)$, $X_{ch}(t)$, $X_{li}(t)$ and $X_{pr}(t)$ are positive and bounded, the above presentation implies that $S_{su}(t)$, $S_{aa}(t)$ and $X_{suaa}(t)$ are also uniformly bounded, and exist for all $t \in [0, \infty)$.

Consider now Equation (5). Since $S_{ac}(t)$ serves as input into the second bioreactor, see Equation (12), it is reasonable to consider only positive initial values $S_{ac}(0) > 0$. The positivity of the phase variables included on the right-hand side of Equation (5) imply that $\dot{S}_{ac}(t) \ge -D_1 S_{ac}(t)$ and thus $S_{ac}(t) \ge S_{ac}(0)e^{-D_1t} > 0$ for all $t \in [0, \infty)$.

Furthermore, the boundedness of the specific growth rate functions imply that there exists a constant $\Gamma_1 > 0$ such that $\dot{S}_{ac}(t) \leq -D_1 S_{ac}(t) + \Gamma_1$ for each $t \geq 0$. Then, we have consecutively

$$\begin{aligned} \frac{d}{dt} \left(e^{D_1 t} S_{ac}(t) \right) &\leq \Gamma_1 e^{D_1 t} \\ e^{D_1 t} S_{ac}(t) &\leq S_{ac}(0) + \Gamma_1 \int_0^t e^{D_1 \xi} d\xi \\ S_{ac}(t) &\leq e^{-D_1 t} S_{ac}(0) + \frac{\Gamma_1}{D_1} \left(1 - e^{-D_1 t} \right) \end{aligned}$$

which means that $\limsup_{t\to\infty} S_{ac}(t) \leq \frac{\Gamma_1}{D_1}$.

Therefore, all solutions of the model (2)–(11) are positive, uniformly bounded, and thus exist for all $t \ge 0$.

Consider now the models (12) and (13). Obviously, if $X_{ac}(0) = 0$, then $X_{ac}(t) = 0$ for all $t \ge 0$. Therefore, it is reasonable to consider only positive initial conditions $X_{ac}(0) > 0$. Then,

$$X_{ac}(t) = X_{ac}(0)e^{\int_0^t \left[-D_2 + \mu_{ac,ch4}(S_{ac,ch4}(\xi))\right]d\xi} > 0 \text{ for all } t \ge 0.$$

If $S_{ac,ch4}(0) = 0$, then $\dot{S}_{ac,ch4}(0) = D_2 S_{ac}(0) > 0$. Assume that there exists $\hat{t} > 0$ such that $S_{ac,ch4}(t) \ge 0$ for $t \in (0, \hat{t})$ but $S_{ac,ch4}(\hat{t}) = 0$. Then, there exists $\varepsilon > 0$ such that $\dot{S}_{ac,ch4}(t) > 0$ for each $t \in (\hat{t} - \varepsilon, \hat{t})$, and so

$$S_{ac,ch4}(\hat{t}) = S_{ac,ch4}(t) + \int_t^t \dot{S}_{ac,ch4}(\xi) d\xi > 0,$$

which is a contradiction. Thus, $S_{ac,ch4}(t) > 0$ for all $t \ge 0$.

Finally, denote $\Sigma_6 = S_{ac,ch4} + Y_{ac}X_{ac}$. Since $S_{ac}(t)$ is positive and uniformly upper bounded, there exists a constant $\Gamma_2 > 0$ such that $S_{ac}(t) \le \Gamma_2$ for all $t \ge 0$. Then, it is straightforward to see that

$$\dot{\Sigma}_6 = D_2(S_{ac} - \Sigma_6) \le D_2(\Gamma_2 - \Sigma_6),$$

which means that $\Sigma_6(t) \leq \Sigma_6(0)e^{-D_2t} + \Gamma_2(1 - e^{-D_2t})$. Thus, $\lim \sup_{t\to\infty} \Sigma_6(t) \leq \Gamma_2$. Since $S_{ac,ch4}$ and X_{ac} are positive, it follows that they are bounded as well. This proves Theorem 1.

We introduce the following notations for simplicity and better readability:

The Jacobi matrix *J* related to g_j , j = 1, 2, ..., 10, with respect to $(S_{su}, S_{aa}, S_{fa}, S_{ac}, X_c, X_{ch}, X_{pr}, X_{li}, X_{suaa}, X_{fa})$ has the form

$$J = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & \frac{\partial g_1}{\partial S_{fa}} & \frac{\partial g_1}{\partial S_{fa}} & \frac{\partial g_1}{\partial X_{cc}} & \frac{\partial g_1}{\partial X_{ch}} & \frac{\partial g_1}{\partial X_{pr}} & \frac{\partial g_1}{\partial X_{li}} & \frac{\partial g_1}{\partial X_{suaa}} & \frac{\partial g_1}{\partial X_{fa}} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & \frac{\partial g_2}{\partial S_{fa}} & \frac{\partial g_2}{\partial S_{ac}} & \frac{\partial g_2}{\partial X_{c}} & \frac{\partial g_2}{\partial X_{ch}} & \frac{\partial g_2}{\partial X_{pr}} & \frac{\partial g_2}{\partial X_{li}} & \frac{\partial g_2}{\partial X_{suaa}} & \frac{\partial g_2}{\partial X_{fa}} \\ \vdots \\ \frac{\partial g_{10}}{\partial S_{su}} & \frac{\partial g_{10}}{\partial S_{aa}} & \frac{\partial g_{10}}{\partial S_{fa}} & \frac{\partial g_{10}}{\partial S_{ac}} & \frac{\partial g_{10}}{\partial X_{ch}} & \frac{\partial g_{10}}{\partial X_{ch}} & \frac{\partial g_{10}}{\partial X_{pr}} & \frac{\partial g_{10}}{\partial X_{li}} & \frac{\partial g_{10}}{\partial X_{suaa}} & \frac{\partial g_{10}}{\partial X_{fa}} \end{pmatrix}$$

Using the explicit dependance of g_j , j = 1, 2, ..., 10, on the above variables, we obtain

The characteristic polynomial corresponding to the Jacobi matrix *J* is defined by $det(J - \lambda I_{10})$, where λ is any complex number and I_{10} is the (10 × 10)–identity matrix.

We note that the Jacobian $J = (J_{ij})$, i, j = 1, 2, ..., 10, is a sparse matrix, whose fourth column and fifth row contain only one nonzero element, i.e., $J_{44} = \frac{\partial g_4}{S_{ac}} = -D_1$ and $J_{55} = \frac{\partial g_5}{X_c} = -D_1 - k_{dis}$, respectively. Then, we obtain

 $det(J - \lambda I_{10}) = (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda) \times$

$$= (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda)(-D_1 - k_{hyd,pr} - \lambda)(-D_1 - k_{hyd,li} - \lambda)$$

$$\times \begin{vmatrix} \frac{\partial g_1}{\partial S_{su}} - \lambda & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} & 0 \\ \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} & 0 \\ \\ \times & 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} - \lambda & 0 & -\mu_{fa} \\ \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} - \lambda & 0 \\ \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 & \frac{\partial g_{10}}{\partial X_{fa}} - \lambda \end{vmatrix}$$

$$=(-D_1-\lambda)(-D_1-k_{\textit{dis}}-\lambda)(-D_1-k_{\textit{hyd},\textit{ch}}-\lambda)(-D_1-k_{\textit{hyd},\textit{pr}}-\lambda)(-D_1-k_{\textit{hyd},\textit{li}}-\lambda)$$

$$\times det(\Delta^{(5)} - \lambda I_5),$$

where I_5 is the (5×5) -identity matrix and $\Delta^{(5)}$ denotes the matrix

$$\Delta^{(5)} = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} & 0 \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & 0 & -\mu_{suaa,aa} & 0 \\ 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} & 0 & -\mu_{fa} \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} & 0 \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 & \frac{\partial g_{10}}{\partial X_{fa}} \end{pmatrix}$$

Furthermore, we obtain

$$det(\Delta^{(5)} - \lambda I_5) = -\mu_{fa} \begin{vmatrix} \frac{\partial g_1}{\partial S_{su}} - \lambda & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} - \lambda & 0 \\ 0 & 0 & \frac{\partial g_{10}}{\partial S_{fa}} & 0 \end{vmatrix}$$

$$+ \left(\frac{\partial g_{10}}{\partial X_{fa}} - \lambda\right) \begin{vmatrix} \frac{\partial g_1}{\partial S_{su}} - \lambda & \frac{\partial g_1}{\partial S_{aa}} & 0 & -\mu_{suaa,su} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} - \lambda & 0 & -\mu_{suaa,aa} \\ 0 & 0 & \frac{\partial g_3}{\partial S_{fa}} - \lambda & 0 \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & 0 & \frac{\partial g_9}{\partial X_{suaa}} - \lambda \end{vmatrix}$$
$$= \left[\mu_{fa} \frac{\partial g_{10}}{\partial S_{fa}} + \left(\frac{\partial g_{10}}{\partial X_{fa}} - \lambda\right) \left(\frac{\partial g_3}{\partial S_{fa}} - \lambda\right) \right] \times det(\Delta^{(3)} - \lambda I_3),$$

where I_3 is the (3×3) -identity matrix and $\Delta^{(3)}$ denotes the matrix

$$\Delta^{(3)} = \begin{pmatrix} \frac{\partial g_1}{\partial S_{su}} & \frac{\partial g_1}{\partial S_{aa}} & -\mu_{suaa,su} \\ \frac{\partial g_2}{\partial S_{su}} & \frac{\partial g_2}{\partial S_{aa}} & -\mu_{suaa,aa} \\ \frac{\partial g_9}{\partial S_{su}} & \frac{\partial g_9}{\partial S_{aa}} & \frac{\partial g_9}{\partial X_{suaa}} \end{pmatrix}.$$

Replacing the derivatives $\frac{\partial g_{10}}{\partial S_{fa}}$, $\frac{\partial g_{10}}{\partial X_{fa}}$ and $\frac{\partial g_3}{\partial S_{fa}}$ by their corresponding explicit expressions, the characteristic polynomial $det(J - \lambda I_{10})$ is presented by

$$det(J - \lambda I_{10}) = (-D_1 - \lambda)(-D_1 - k_{dis} - \lambda)(-D_1 - k_{hyd,ch} - \lambda)(-D_1 - k_{hyd,pr} - \lambda)$$

$$\times (-D_1 - k_{hyd,li} - \lambda) \left[\mu_{fa} Y_{fa} X_{fa} \frac{d\mu_{fa}}{dS_{fa}} + (-D_1 + Y_{fa} \mu_{fa} - \lambda) \left(-D_1 - X_{fa} \frac{d\mu_{fa}}{dS_{fa}} - \lambda \right) \right]$$

$$\times det(\Delta^{(3)} - \lambda I_3).$$

The corresponding partial derivatives in $\Delta^{(3)}$ are as follows:

(i). The characteristic polynomial of the coexistence equilibrium

$$E_1 = E_1(D_1) = (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}, \bar{S}_{ac}, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa}) > 0 \text{ for } D_1 \in \left(0, D_1^{(1)}\right).$$

Denote by $P(E_1; \lambda)$ the characteristic polynomial $det(J - \lambda I_{10})$ evaluated at the equilibrium E_1 , and by $R(E_1; \lambda)$ the characteristic polynomial $det(\Delta^{(3)} - \lambda I_3)$ evaluated at the equilibrium components $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (0, D_1^{(1)})$. Taking into account that $-D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}) = 0$, we obtain for $P(E_1; \lambda)$ the presentation

$$P(E_{1};\lambda) = (-D_{1} - \lambda)(-D_{1} - k_{dis} - \lambda)(-D_{1} - k_{hyd,ch} - \lambda)$$
$$\times (-D_{1} - k_{hyd,pr} - \lambda)(-D_{1} - k_{hyd,li} - \lambda)$$
$$\times \left[\lambda^{2} + \left(D_{1} + \frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right)\lambda + D_{1}\frac{d\mu_{fa}}{dS_{fa}}(\bar{S}_{fa})\bar{X}_{fa}\right] \times R(E_{1};\lambda).$$

(ii). The characteristic polynomial of the equilibrium

$$E_1^0 = E_1^0(D_1) = (\bar{S}_{su}, \bar{S}_{aa}, \bar{S}_{fa}^0, \bar{S}_{ac}^0, \bar{X}_c, \bar{X}_{ch}, \bar{X}_{pr}, \bar{X}_{li}, \bar{X}_{suaa}, \bar{X}_{fa} = 0) \text{ for } D_1 \in \left(D_1^{(1)}, D_1^{(2)}\right).$$

Denote by $P(E_1^0; \lambda)$ the characteristic polynomial $det(J - \lambda I_{10})$ evaluated at the equilibrium E_1^0 , and by $R(E_1^0; \lambda)$ the characteristic polynomial $det(\Delta^{(3)} - \lambda I_3)$ evaluated at the equilibrium components $(\bar{S}_{su}, \bar{S}_{aa}, \bar{X}_{suaa})$ for $D_1 \in (D_1^{(1)}, D_1^{(2)})$. Taking into account that $\bar{X}_{fa} = 0$ and that $-D_1 + Y_{fa}\mu_{fa}(\bar{S}_{fa}^0) \neq 0$ in this case, we obtain for $P(E_1^0; \lambda)$ the presentation

$$P(E_{1}^{0};\lambda) = (-D_{1} - \lambda)^{2}(-D_{1} - k_{dis} - \lambda)(-D_{1} - k_{hyd,li} - \lambda)(-D_{1} - k_{hyd,pr} - \lambda)$$
$$\times (-D_{1} - k_{hyd,li} - \lambda)(-D_{1} + Y_{fa}\mu_{fa}(\bar{S}_{fa}^{0}) - \lambda) \times R(E_{1}^{0};\lambda).$$

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