

## Supplementary

**Table 1S.** List of Chemical Species Appearing in Methane Thermochlorination Kinetic Scheme from ref. [14,15].

### Unimolecular Decomposition Reactions

No.	Reaction	An	En
		(1/s)	(cal/mol)
1	$\text{Cl}_2 \rightleftharpoons 2\text{Cl}\cdot$	0.115E+12	0.580E+5
2	$\text{CH}_3\text{Cl} \rightleftharpoons \text{CH}_3\cdot + \text{Cl}\cdot$	0.1E+17	0.844E+5
3	$\text{CH}_2\text{Cl}_2 \rightleftharpoons \text{CH}_2\text{Cl}\cdot + \text{Cl}\cdot$	0.1E+17	0.811E+5
4	$\text{CHCl}_2-\text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2\cdot + \text{CH}_2\text{Cl}\cdot$	0.1E+18	0.884E+5
5	$\text{CH}_2\text{Cl}-\text{CH}_2\text{Cl} \rightleftharpoons 2\text{CH}_2\text{Cl}\cdot$	0.1E+18	0.901E+5
6	$\text{CHCl}_3 \rightleftharpoons \text{CHCl}_2\cdot + \text{Cl}\cdot$	0.1E+17	0.776E+5
7	$\text{CHCl}_2-\text{CHCl}_2 \rightleftharpoons 2\text{CHCl}_2\cdot$	0.1E+18	0.824E+5
8	$\text{CCl}_3-\text{CCl}_3 \rightleftharpoons 2\text{CCl}_3\cdot$	0.1E+18	0.714E+5
9	$\text{CCl}_3-\text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_2\text{Cl}\cdot$	0.1E+18	0.827E+5
10	$\text{CCl}_3-\text{CHCl}_2 \rightleftharpoons \text{CCl}_3\cdot + \text{CHCl}_2\cdot$	0.1E+18	0.759E+5
11	$\text{CCl}_4 \rightleftharpoons \text{CCl}_3\cdot + \text{Cl}\cdot$	0.1E+17	0.709E+5
12	$\text{CH}_2\text{Cl}-\text{CH}_2\cdot \rightleftharpoons \text{CH}_2\text{CH}_2 + \text{Cl}\cdot$	0.1E+15	0.203E+5
13	$\text{CH}_2\text{Cl}-\text{CCl}_2\cdot \rightleftharpoons \text{CCl}_2\text{CH}_2 + \text{Cl}\cdot$	0.1E+15	0.259E+5
14	$\text{CHCl}_2-\text{CHCl}\cdot \rightleftharpoons t\text{-CHClCHCl} + \text{Cl}\cdot$	0.1E+15	0.216E+5
15	$\text{CHCl}_2-\text{CHCl}\cdot \rightleftharpoons c\text{-CHClCHCl} + \text{Cl}\cdot$	0.1E+15	0.211E+5
16	$\text{CH}_2\text{Cl}-\text{CHCl}\cdot \rightleftharpoons \text{CHClCH}_2 + \text{Cl}\cdot$	0.1E+15	0.229E+5
17	$\text{CHCl}_2-\text{CH}_2\cdot \rightleftharpoons \text{CHClCH}_2 + \text{Cl}\cdot$	0.1E+15	0.192E+5
18	$\text{CHCl}_2-\text{CCl}_2\cdot \rightleftharpoons \text{CCl}_2\text{CHCl} + \text{Cl}\cdot$	0.1E+15	0.212E+5
19	$\text{CCl}_3-\text{CCl}_2\cdot \rightleftharpoons \text{CCl}_2\text{CCl}_2 + \text{Cl}\cdot$	0.1E+15	0.194E+5
20	$\text{CCl}_3-\text{CH}_2\cdot \rightleftharpoons \text{CCl}_2\text{CH}_2 + \text{Cl}\cdot$	0.1E+15	0.175E+5
21	$\text{CCl}_3-\text{CHCl}\cdot \rightleftharpoons \text{CCl}_2\text{CHCl} + \text{Cl}\cdot$	0.1E+15	0.185E+5

### Metathesis Reactions

No.	Reaction	An	En
1	$c\text{-CHClCHCl} + \text{Cl}\cdot \rightleftharpoons c\text{-CHClCCl}\cdot + \text{HCl}$	0.1E+14	0.196E+4
2	$t\text{-CHClCHCl} + \text{Cl}\cdot \rightleftharpoons t\text{-CHClCCl}\cdot + \text{HCl}$	0.1E+14	0.151E+4
3	$\text{CH}_3\text{Cl} + \text{Cl}\cdot \rightleftharpoons \text{CH}_2\text{Cl}\cdot + \text{HCl}$	0.1E+14	0.460E+3
4	$\text{CH}_3\text{Cl} + \text{Cl}\cdot \rightleftharpoons \text{CH}_3\cdot + \text{Cl}_2$	0.1E+15	0.256E+5

5	$\text{CH}_4 + \text{Cl}\cdot \rightleftharpoons \text{CH}_3\cdot + \text{HCl}$	0.1E+14	0.211E+4
6	$\text{CH}_4 + \text{CH}_2\text{Cl}\cdot \rightleftharpoons \text{CH}_3\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.158E+5
7	$\text{CH}_2\text{Cl}_2 + \text{Cl}\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{HCl}$	0.1E+14	0.460E+3
8	$\text{CH}_2\text{Cl}_2 + \text{CH}_3\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{CH}_4$	0.3E+12	0.110E+5
9	$\text{CH}_2\text{Cl}_2 + \text{CH}_2\text{Cl}\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.110E+5
10	$\text{CH}_2\text{Cl}_2 + \text{Cl}\cdot \rightleftharpoons \text{CH}_2\text{Cl}\cdot + \text{Cl}_2$	0.1E+15	0.224E+5
11	$\text{CH}_2\text{Cl}_2 + \text{CH}_3\cdot \rightleftharpoons \text{CH}_2\text{Cl}\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.110E+5
12	$\text{CHCl}_2-\text{CH}_2\text{Cl} + \text{Cl}\cdot \rightleftharpoons \text{CHCl}_2\cdot-\text{CHCl}\cdot + \text{HCl}$	0.1E+14	0.460E+3
13	$\text{CHCl}_2-\text{CH}_2\text{Cl} + \text{Cl}\cdot \rightleftharpoons \text{CH}_2\text{Cl}-\text{CCl}_2\cdot + \text{HCl}$	0.1E+14	0.460E+3
14	$\text{CH}_2\text{Cl}-\text{CH}_2\text{Cl} + \text{Cl}\cdot \rightleftharpoons \text{CH}_2\text{Cl}-\text{CHCl}\cdot + \text{HCl}$	0.1E+14	0.460E+3
15	$\text{CHCl}_3 + \text{Cl}\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{HCl}$	0.1E+14	0.460E+3
16	$\text{CHCl}_3 + \text{CH}_3\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_4$	0.3E+12	0.110E+5
17	$\text{CHCl}_3 + \text{CH}_2\text{Cl}\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.110E+5
18	$\text{CHCl}_3 + \text{CHCl}_2\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_2\text{Cl}_2$	0.3E+12	0.110E+5
19	$\text{CHCl}_3 + \text{Cl}\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{Cl}_2$	0.1E+15	0.189E+5
20	$\text{CHCl}_3 + \text{CH}_3\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.110E+5
21	$\text{CHCl}_3 + \text{CH}_2\text{Cl}\cdot \rightleftharpoons \text{CHCl}_2\cdot + \text{CH}_2\text{Cl}_2$	0.3E+12	0.110E+5
22	$\text{CHCl}_2-\text{CHCl}_2 + \text{Cl}\cdot \rightleftharpoons \text{CHCl}_2-\text{CCl}_2\cdot + \text{HCl}$	0.1E+14	0.460E+3
23	$\text{CCl}_4 + \text{Cl}\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{Cl}_2$	0.1E+15	0.122E+5
24	$\text{CCl}_4 + \text{CH}_3\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_3\text{Cl}$	0.3E+12	0.110E+5
25	$\text{CCl}_4 + \text{CH}_2\text{Cl}\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CH}_2\text{Cl}_2$	0.3E+12	0.110E+5
26	$\text{CCl}_4 + \text{CHCl}_2\cdot \rightleftharpoons \text{CCl}_3\cdot + \text{CHCl}_3$	0.3E+12	0.110E+5

### Chaperon Reactions

No.	Reaction	An	En
1	$\text{CCl}_2-\text{CCl}_2 + \text{Cl}_2 \rightleftharpoons \text{CCl}_3-\text{CCl}_2\cdot + \text{Cl}\cdot$	0.1E+14	0.399E+5
2	$\text{CCl}_2-\text{CHCl} + \text{Cl}_2 \rightleftharpoons \text{CHCl}_2-\text{CCl}_2\cdot + \text{Cl}\cdot$	0.1E+14	0.380E+5
3	$\text{CCl}_2-\text{CHCl} + \text{Cl}_2 \rightleftharpoons \text{CCl}_3-\text{CHCl}\cdot + \text{Cl}\cdot$	0.1E+14	0.408E+5
4	$\text{c-CHClCHCl} + \text{Cl}_2 \rightleftharpoons \text{CHCl}_2-\text{CHCl}\cdot + \text{Cl}\cdot$	0.1E+14	0.375E+5
5	$\text{t-CHClCHCl} + \text{Cl}_2 \rightleftharpoons \text{CHCl}_2-\text{CHCl}\cdot + \text{Cl}\cdot$	0.1E+14	0.374E+5
6	$\text{CCl}_2\text{CH}_2 + \text{Cl}_2 \rightleftharpoons \text{CH}_2\text{Cl}-\text{CCl}_2\cdot + \text{Cl}\cdot$	0.1E+14	0.346E+5
7	$\text{CCl}_2\text{CH}_2 + \text{Cl}_2 \rightleftharpoons \text{CCl}_3-\text{CH}_2\cdot + \text{Cl}\cdot$	0.1E+14	0.429E+5
8	$\text{CHClCH}_2 + \text{Cl}_2 \rightleftharpoons \text{CH}_2\text{Cl}-\text{CHCl}\cdot + \text{Cl}\cdot$	0.1E+14	0.373E+5
9	$\text{CHClCH}_2 + \text{Cl}_2 \rightleftharpoons \text{CHCl}_2-\text{CH}_2\cdot + \text{Cl}\cdot$	0.1E+14	0.410E+5





76	$\text{CCl}_2\text{CHCl} + \text{CHCl}_2 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CH}_2\text{Cl} - \text{CHCl}\cdot$	0.3E+15	0.608E+5
77	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_3 \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CH}_3 - \text{CCl}_2\cdot$	0.3E+15	0.564E+5
78	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CCl}_3 - \text{CH}_2\cdot$	0.3E+15	0.612E+5
79	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CH}_2\text{Cl} - \text{CCl}_2\cdot$	0.3E+15	0.528E+5
80	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CHCl}_2 \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CCl}_3 - \text{CHCl}\cdot$	0.3E+15	0.551E+5
81	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CCl}_3 \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CCl}_3 - \text{CCl}_2\cdot$	0.3E+15	0.539E+5
82	$\text{CCl}_2\text{CHCl} + \text{CCl}_4 \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CCl}_3\cdot$	0.3E+15	0.525E+5
83	$\text{CCl}_2\text{CHCl} + \text{CHCl}_2 - \text{CHCl}_2 \rightleftharpoons \text{CCl}_3 - \text{CHCl}\cdot + \text{CHCl}_2 - \text{CHCl}\cdot$	0.3E+15	0.566E+5
84	$\text{CCl}_2\text{CHCl} + \text{CH}_2\text{Cl} - \text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CH}_2\cdot$	0.3E+15	0.634E+5
85	$\text{CCl}_2\text{CHCl} + \text{CHCl}_2 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CH}_2\cdot$	0.3E+15	0.617E+5
86	$\text{CCl}_2\text{CHCl} + \text{CHCl}_2 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CHCl}\cdot$	0.3E+15	0.580E+5
87	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_3 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CH}_3 - \text{CCl}_2\cdot$	0.3E+15	0.536E+5
88	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CH}_2\cdot$	0.3E+15	0.584E+5
89	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CCl}_2\cdot$	0.3E+15	0.500E+5
90	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CHCl}_2 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CHCl}\cdot$	0.3E+15	0.523E+5
91	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CHCl}_2 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CCl}_2\cdot$	0.3E+15	0.495E+5
92	$\text{CCl}_2\text{CHCl} + \text{CCl}_3 - \text{CCl}_3 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CCl}_2\cdot$	0.3E+15	0.512E+5
93	$\text{CCl}_2\text{CHCl} + \text{CCl}_4 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CCl}_3\cdot$	0.3E+15	0.497E+5
94	$\text{CCl}_2\text{CHCl} + \text{CHCl}_2 - \text{CHCl}_2 \rightleftharpoons \text{CHCl}_2 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CHCl}\cdot$	0.3E+15	0.538E+5
95	$\text{CCl}_2\text{CCl}_2 + \text{CH}_2\text{Cl} - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CH}_2\cdot$	0.3E+15	0.652E+5
96	$\text{CCl}_2\text{CCl}_2 + \text{CHCl}_2 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CH}_2\cdot$	0.3E+15	0.636E+5
97	$\text{CCl}_2\text{CCl}_2 + \text{CHCl}_2 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CHCl}\cdot$	0.3E+15	0.599E+5
98	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CH}_3 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CH}_3 - \text{CCl}_2\cdot$	0.3E+15	0.555E+5
99	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CH}_2\cdot$	0.3E+15	0.603E+5
100	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CH}_2\text{Cl} \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CH}_2\text{Cl} - \text{CCl}_2\cdot$	0.3E+15	0.519E+5
101	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CHCl}_2 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CHCl}\cdot$	0.3E+15	0.542E+5
102	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CHCl}_2 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CCl}_2\cdot$	0.3E+15	0.514E+5
103	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_3 - \text{CCl}_3 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CCl}_3 - \text{CCl}_2\cdot$	0.3E+15	0.530E+5
104	$\text{CCl}_2\text{CCl}_2 + \text{CCl}_4 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CCl}_3\cdot$	0.3E+15	0.516E+5
105	$\text{CCl}_2\text{CCl}_2 + \text{CHCl}_2 - \text{CHCl}_2 \rightleftharpoons \text{CCl}_3 - \text{CCl}_2\cdot + \text{CHCl}_2 - \text{CHCl}\cdot$	0.3E+15	0.557E+5

**Table 2S.** Microkinetic model of methane halogenation using iodine and bromine in the gas phase from ref [17].

No.	Reaction	An ( $10^{13}$ cm $^3$ /mole s)	Ea (kJ/mole)
1	$\text{Br}_2 \rightarrow \text{Br} + \text{Br}$	422	185
2	$\text{Br} + \text{H}_2 \rightarrow \text{HBr} + \text{H}$	4.8	74.6
3	$\text{H} + \text{Br}_2 \rightarrow \text{HBr} + \text{Br}$	6.8	1.83
4	$\text{H} + \text{HBr} \rightarrow \text{H}_2 + \text{Br}$	1.32	4.6
5	$2\text{Br} \rightarrow \text{Br}_2$	14.8	-7.12
6	$\text{Br} + \text{HBr} \rightarrow \text{Br}_2 + \text{H}$	27.2	185.0
7	$\text{Br} + \text{CH}_4 \rightarrow \text{HBr} + \text{CH}_3$	9.81	77.7
8	$\text{Br}_2 + \text{CH}_3 \rightarrow \text{CH}_3\text{Br} + \text{Br}$	1.16	0.3
9	$\text{CH}_3\text{Br} + \text{Br} \rightarrow \text{CH}_3 + \text{Br}_2$	*	*
10	$\text{CH}_3 + \text{HBr} \rightarrow \text{Br} + \text{CH}_4$	0.537	6.7
11	$\text{I}_2 \rightarrow \text{I} + \text{I}$	8.43	126.0
12	$\text{I} + \text{H}_2 \rightarrow \text{HI} + \text{H}$	16.9	141.0
13	$\text{H} + \text{I}_2 \rightarrow \text{HI} + \text{I}$	43.1	1.8
14	$\text{H} + \text{HI} \rightarrow \text{H}_2 + \text{I}$	4.74	2.7
15	$\text{I} + \text{I} \rightarrow \text{I}_2$	20.0	-4.8
16	$\text{I} + \text{HI} \rightarrow \text{I}_2 + \text{H}$	80.1	155.0
17	$\text{I} + \text{CH}_4 \rightarrow \text{HI} + \text{CH}_3$	89.1	146.0
18	$\text{I}_2 + \text{CH}_3 \rightarrow \text{CH}_3\text{I} + \text{I}$	0.999	6.3
19	$\text{CH}_3\text{I} + \text{I} \rightarrow \text{CH}_3 + \text{I}_2$	20.0	82.8
20	$\text{CH}_3 + \text{HI} \rightarrow \text{I} + \text{CH}_4$	0.415	5.4
21	$\text{CH}_3\text{I} \rightarrow \text{CH}_3 + \text{I}$	1.00	226
22	$\text{IBr} \rightarrow \text{I} + \text{Br}$	111 [a]	157.55 [a]
23	$\text{H} + \text{IBr} \rightarrow \text{HI} + \text{Br}$	21.6 [b]	1.8 [c]
24	$\text{H} + \text{IBr} \rightarrow \text{HBr} + \text{I}$	3.87 [d]	5.0 [e]
25	$\text{I} + \text{Br} \rightarrow \text{IBr}$	*	*
26	$\text{HBr} + \text{I} \rightarrow \text{H} + \text{IBr}$	*	*
27	$\text{HI} + \text{Br} \rightarrow \text{HBr} + \text{I}$	0.746	0.27
28	$\text{I} + \text{HBr} \rightarrow \text{HI} + \text{Br}$	*	*
29	$\text{CH}_3 + \text{IBr} \rightarrow \text{CH}_3\text{Br} + \text{I}$	1.16 [f]	0.3 [g]
30	$\text{CH}_3 + \text{IBr} \rightarrow \text{CH}_3\text{I} + \text{Br}$	[h]	[h]

31	$\text{CH}_3\text{Br} + \text{I} \rightarrow \text{CH}_3 + \text{IBr}$	*	*
32	$\text{CH}_3\text{I} + \text{Br} \rightarrow \text{CH}_3 + \text{IBr}$	*	*
33	$\text{IBr} + \text{Br} \rightarrow \text{Br}_2 + \text{I}$	2.11	0
34	$\text{IBr} + \text{I} \rightarrow \text{I}_2 + \text{Br}$	*	*
35	$\text{Br}_2 + \text{I} \rightarrow \text{IBr} + \text{Br}$	*	*
36	$\text{I}_2 + \text{Br} \rightarrow \text{IBr} + \text{I}$	7.22	0.0
37	$\text{CH}_3\text{Br} + \text{Br} \rightarrow \text{CH}_2\text{Br} + \text{HBr}$	9.99	68.26
38	$\text{CH}_2\text{Br} + \text{HBr} \rightarrow \text{CH}_3\text{Br} + \text{Br}$	0.045	2.53
39	$\text{CH}_2\text{Br} + \text{HI} \rightarrow \text{CH}_3\text{Br} + \text{I}$	0.090	-0.8
40	$\text{CH}_2\text{Br} + \text{Br}_2 \rightarrow \text{CH}_2\text{Br}_2 + \text{Br}$	0.331	-3.01
41	$\text{CH}_2\text{Br}_2 + \text{Br} \rightarrow \text{CH}_2\text{Br} + \text{Br}_2$	*	*

\*the rate constants were calculated using the reverse reaction rate constant and the equilibrium constant according to  $k = k_{\text{reverse}} \times \exp[-\Delta G/RT]$ .

[a] average of reactions #1 and #11; [b]  $A_{23} = \frac{1}{2}A_{13}$ ; [c]  $Ea_{23} = Ea_{13}$ ; [d]  $A_{24} = \frac{1}{2}A_3$ ; [e]  $Ea_{24} = Ea_3$ ; [f]  $A_{29} = \frac{1}{2}A_8$ ; [g]  $Ea_{29} = Ea_8$ ; [h]  $k_{30} = 10 \cdot k_{29}$ .