

Article

Hydrogen Peroxide and Superoxide Anion Radical Photoproduction in PSII Preparations at Various Modifications of the Water-Oxidizing Complex

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Received: 30 July 2019; Accepted: 29 August 2019; Published: 5 September 2019



Abstract: The photoproduction of superoxide anion radical ($O_2^{\cdot-}$) and hydrogen peroxide (H_2O_2) in photosystem II (PSII) preparations depending on the damage to the water-oxidizing complex (WOC) was investigated. The light-induced formation of $O_2^{\cdot-}$ and H_2O_2 in the PSII preparations rose with the increased destruction of the WOC. The photoproduction of superoxide both in the PSII preparations holding intact WOC and the samples with damage to the WOC was approximately two times higher than H_2O_2 . The rise of $O_2^{\cdot-}$ and H_2O_2 photoproduction in the PSII preparations in the course of the disassembly of the WOC correlated with the increase in the fraction of the low-potential (LP) Cyt b_{559} . The restoration of electron flow in the Mn-depleted PSII preparations by exogenous electron donors (diphenylcarbazide, Mn^{2+}) suppressed the light-induced formation of $O_2^{\cdot-}$ and H_2O_2 . The decrease of $O_2^{\cdot-}$ and H_2O_2 photoproduction upon the restoration of electron transport in the Mn-depleted PSII preparations could be due to the re-conversion of the LP Cyt b_{559} into higher potential forms. It is supposed that the conversion of the high potential Cyt b_{559} into its LP form upon damage to the WOC leads to the increase of photoproduction of $O_2^{\cdot-}$ and H_2O_2 in PSII.

Keywords: photosystem II; water-oxidizing complex; cytochrome b_{559} superoxide anion radical; hydrogen peroxide

1. Introduction

Photosystem II (PSII) is a pigment–protein complex built into the thylakoid membrane. The main function of PSII is the light-induced oxidation of water to molecular oxygen with a transfer of electrons to the pool of plastoquinones. Recent crystallographic investigations of cyanobacterial PSII showed that a minimal structure capable of photosynthetic water oxidation and oxygen evolution (the so-called core complex of PSII) contains at least 20 protein subunits, 35 chlorophyll (Chl) molecules, 12 molecules of carotenoids, and at least 14–20 integral lipid molecules per monomer [1–3]. The light-induced charge separation with the formation of an oxidized primary electron donor, $P_{680}^{+\bullet}$ (the strongest biological oxidant, with a redox potential of 1.1–1.27 V [4,5]), occurs in the photochemical reaction centre (RC) consisting of main proteins, D1 (PsbA) and D2 (PsbD), and cytochrome b_{559} (Cyt b_{559}). $P_{680}^{+\bullet}$ oxidizes TyrZ (tyrosine residue of D1 protein) with the formation of $TyrZ^{\bullet}$, which in turn takes an electron from the Mn_4CaO_5 cluster, the inorganic core of the water-oxidizing complex (WOC). The sequential absorption of photons and charge separation in the RC result in the formation of intermediate states (S_0 – S_4) of the WOC, and the transition from S_4 to S_0 is accompanied by the oxygen release.

An integral part of the reaction centre is Cyt b_{559} , which participates in redox reactions and, in comparison with other redox components of the RC, is not located inside the D1/D2 heterodimer. Cyt b_{559} can be found in at least four different redox forms: the Cyt b_{559} high-potential (HP) form ($E =$ from + 350 mV to + 450 mV), in intermediate-potential (IP) form ($E =$ from + 125 to +240 mV),

a low-potential (LP) form ($E =$ from -40 to $+80$ mV) (see [6]), and in a so-called very low-potential (VLP) form with $E_m =$ from -150 to -200 mV [7,8]. The ratio of the redox forms of Cyt b_{559} PSII preparations depends on the structural integrity and composition of PSII. It was shown that the perturbation of the WOC led to the decrease of HP Cyt b_{559} and the increase of IP and LP Cyt b_{559} [9,10]. It was also shown that the conversion of HP Cyt b_{559} to the LP Cyt b_{559} could be induced by acidification of the medium [11]. Cyt b_{559} is assumed to participate in cyclic electron transfer, which is considered to be a protective mechanism against the photoinhibition of PSII, but this photoprotective role of Cyt b_{559} is debated [12–14]. It has been shown that Cyt b_{559} shows the following enzymatic properties: oxygen reductase, superoxide reductase, superoxide oxidase, and plastoquinol oxidase (see review in [15]).

When electrons from water pass into the electron transport chain of PSII, compounds with low redox potential are formed. They are considered to be the essential sources for the production of superoxide anion radicals ($O_2^{\cdot-}$), which are subsequently converted to H_2O_2 and O_2 via spontaneous or enzyme-catalyzed dismutation. Using a luminol–peroxidase method for the detection of H_2O_2 , it was shown that the light-induced yield of the H_2O_2 in isolated oxygen-evolving PSII membrane fragments was slight (about 0.01 H_2O_2 molecules per RC and saturating flash) [16,17]. Possible donors of electrons to O_2 can be the reduced forms of the primary electron acceptor pheophytin ($Pheo^-$) [18], the primary (Q_A^-) and secondary (Q_B^-) quinone electron acceptors [19], plastoquinone ($PQH^{\cdot}$) (where $O_2^{\cdot-}$ is produced via the proportion between plastoquinone (PQ) and plastoquinol (PQH_2)) [20,21], and LP cytochrome Cyt b_{559} [22,23]. For a detailed description of $O_2^{\cdot-}$ and H_2O_2 photoproduction in PSII, see also [24,25].

It was shown that the treatments leading to the perturbation of the PSII donor side increased H_2O_2 photoproduction [16,17,26,27]. It was assumed that the increase of H_2O_2 photoproduction in the PSII after a partial injury of the WOC could be associated with the replacement of the four-electron (with the release of O_2) by the two-electron (with the production of H_2O_2) oxidation of water [16,27]. However, using isotope-labelled water in combination with a detection system for H_2O_2 showed that the oxygen in H_2O_2 formed during the illumination of NaCl-wash PSII membranes did not originate from water [26]. Thus, H_2O_2 photoproduction in PSII can occur both via the disproportionation of $O_2^{\cdot-}$ formed as a result of the one-electron reduction of O_2 on the PSII acceptor side and the incomplete photooxidation of water appearing after disturbance of the WOC.

In the present work, the effect of the step-by-step disassembly of the WOC on H_2O_2 and $O_2^{\cdot-}$ photoproduction in PSII membrane fragments and core complexes was investigated. The light-induced formation of $O_2^{\cdot-}$ and H_2O_2 in PSII was raised with the increasing destruction of the WOC. The comparison between H_2O_2 and $O_2^{\cdot-}$ photoproduction in PSII preparations showed that $O_2^{\cdot-}$ yield in all samples was approximately two times higher than H_2O_2 . It is suggested that the stimulation of H_2O_2 photoproduction caused by the destruction of the WOC is mainly due to the acceptor side of PSII rather than the donor side via the enhancement of the $O_2^{\cdot-}$ production, and Cyt b_{559} can play a crucial role in this.

2. Results

2.1. Functional Activity in PSII Preparations at Various Modifications of the WOC

The investigation of H_2O_2 and $O_2^{\cdot-}$ photoproduction in PSII was carried out on the PSII membranes and the PSII core complexes with different degrees of damage to the WOC: untreated, and NaCl-, $CaCl_2$ -, and NH_2OH -treated PSII. The step-by-step disassembly of the WOC led to the suppression of PSII activity (oxygen-evolving activity and photoinduced ΔF). The yield of photoinduced ΔF was decreased by 20% and 30% after NaCl and $CaCl_2$ treatments of PSII membranes, respectively (Figure 1(I)A, curves 2 and 3). The complete removal of Mn ions from the WOC by NH_2OH treatment led to a 5-fold decrease in the ΔF (Figure 1(I)A, curve 4) due to the loss of electron donation from the Mn-containing WOC to the PS II reaction centre (RC), which is in accordance with previous publications [28]. The photosynthetic oxygen evolution was more sensitive to the treatments in comparison with the ΔF (Figure 1(II) A).

The rate of photosynthetic oxygen evolution in the untreated PSII membranes was about $600 \mu\text{mol O}_2 (\text{mg Chl h})^{-1}$. The treatment of PSII membranes with NaCl and CaCl_2 resulted in a decrease in the rate of photosynthetic O_2 evolution by 30% and 90%, respectively. The Mn removal from the WOC completely inhibited the oxygen-evolving activity of PSII and resulted in O_2 photoconsumption which, as was shown earlier, was associated with both the photoformation of organic hydroperoxides on the donor side via a radical chain mechanism and with the photoproduction of H_2O_2 on the acceptor side of PSII [29–31].

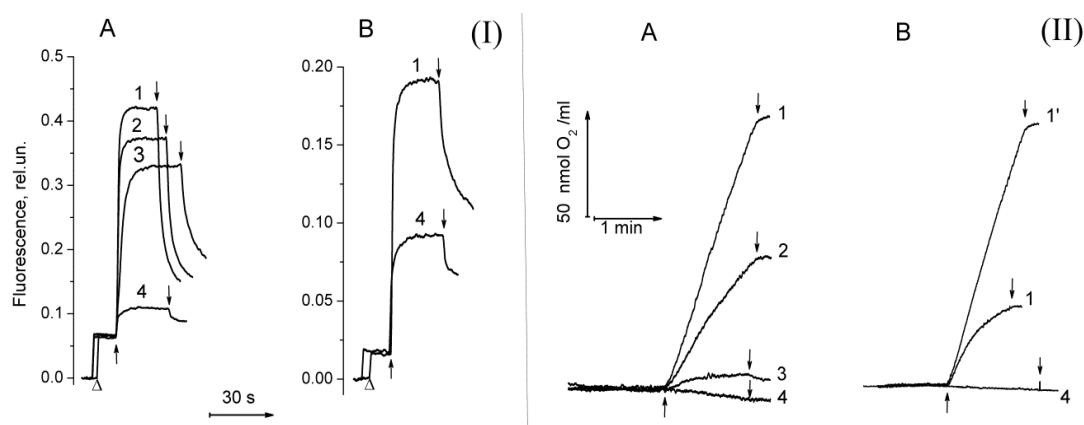


Figure 1. (I) Kinetics of photoinduced changes of chlorophyll fluorescence yield (ΔF) related to the photoreduction of the primary electron acceptor, Q_A , in photosystem II (PSII) membranes fragments (A) and PSII core complexes (B) before (1) and after modification of the water-oxidizing complex caused by treatments with NaCl (2), CaCl_2 (3), and NH_2OH (4). The measurements of ΔF were done in a medium containing 50 mM MES–NaOH (pH 6.5), 35 mM NaCl and 0.4 M sucrose at a Chl concentration of $10 \mu\text{g/mL}$. Δ , switching of the measuring light; $\uparrow$ and $\downarrow$, actinic light on and off, respectively. (II) Kinetics of oxygen evolution in PSII membranes (A) and PSII core complexes (B) before (1) and after modification of the water-oxidizing complex caused by treatments with NaCl (2), CaCl_2 (3), and NH_2OH (4). The measurements were made in the medium containing 50 mM MES–NaOH (pH 6.5), 35 mM NaCl, 0.4 M sucrose at a Chl concentration of $10 \mu\text{g/mL}$ for the PSII membranes and at $5 \mu\text{g/mL}$ for the PSII core complexes in the presence of 1 mM $\text{K}_3[\text{Fe}(\text{CN})_6]$ and 100 μM DCBQ. (1')—oxygen evolution in the PSII core complexes was done in the presence of 5 mM CaCl_2 . $\uparrow$ and $\downarrow$ —light ($\lambda > 650 \text{ nm}$, $1500 \mu\text{mol photon s}^{-1} \text{ m}^{-2}$) on and off, respectively.

The PSII core complexes showed maximal oxygen-evolving activity (about $1300 \mu\text{mol O}_2 (\text{mg Chl h})^{-1}$) only in the presence of exogenous Ca^{2+} (Figure 1(II)B, curves 1 and 1'). The CaCl_2 dependence of the oxygen-evolving activity in the core complexes can be associated with the partial removal of PsbP and PsbQ proteins during ion exchange chromatography, since the concentration of MgSO_4 used to elute the PSII cores was about 100 mM. It was shown that the release of PsbP and PsbQ proteins from the WOC suppressed PSII oxygen-evolving activity and the addition of CaCl_2 reconstituted high rates of oxygen evolution in the PS II preparations deprived of these proteins [32]. Due to this reason, only NH_2OH treatment was performed to modify the WOC in the PSII core complexes. In comparison with PSII membranes (where the release of Mn from the WOC resulted in a drastic decrease in the ΔF), the yield of ΔF in the Mn-depleted PSII core complexes was about two times less than in the untreated ones (Figure 1(I)B). A similar yield of ΔF was also observed in Mn-depleted PSII core complexes which were obtained by isolation from Mn-depleted PSII membranes. This may be due to the removal of a quinone from the Q_B site, since the Q_B quinone can release from its binding site during the isolation of PSII core complexes [33]. Even though the yield of ΔF in the Mn-depleted PSII core complexes was sufficiently high, the ability of the samples to perform photosynthetic oxygen evolution was completely lost (Figure 1(II)B, curve 4).

2.2. The Ratio in Redox Forms of Cyt b_{559} in PSII Preparations at Various Modifications of the WOC

In addition to the suppression of the PSII functional activity, the destruction of the WOC changed the ratio in redox forms of Cyt b_{559} in the PSII membranes (Table 1). The contents of HP, IP, and LP Cyt b_{559} in the untreated PSII membranes were 57%, 9%, and 34%, respectively. The treatment of PSII membranes with 1 M NaCl caused a slight decrease in the content of HP Cyt b_{559} and an increase of its IP form without changing the content of LP Cyt b_{559} . A much stronger disturbance of the WOC induced by the treatment of PSII membranes with 1 M CaCl_2 was accompanied by a significant decrease in the proportion of HP Cyt b_{559} and increase of IP and LP Cyt b_{559} ; thus, the ratio of the redox form of Cyt b_{559} in the samples was about 20% of the HP form, 35% of the IP form, and 45% of the LP form. In the Mn-depleted PSII membranes, most of Cyt b_{559} was in the LP (52%) and the IP (31%) forms, and only 17% was in the HP form. The similar interrelationship between the state of the WOC and the ratio in the redox forms of Cyt b_{559} in PSII preparations was shown previously [9,10]. In contrast to PSII membranes, untreated PSII core complexes contained about 12 % of HP Cyt b_{559} , and this percentage did not change after the removal of Mn from the WOC. However, the untreated and Mn-depleted PSII core complexes considerably differed in the content of IP and LP Cyt b_{559} : For the untreated samples, the contents of the IP and LP forms were 45% and 43%, respectively, while Mn-depleted samples contained 21% of the IP form and 67 % of the LP form (Table 1).

Table 1. The ratio of Cyt b_{559} redox forms in PSII preparations after various modifications of the water-oxidizing complex.

Redox Forms of Cyt b_{559}	PSII Membranes				PSII Core Complexes	
	Untreated	NaCl-Treated	CaCl_2 -Treated	Mn-Depleted	Untreated	Mn-Depleted
HP	56.6	43.7	19.8	17.4	12	11.8
IP	8.6	21.7	34.7	30.6	45.2	21
LP	34.8	34.6	45.5	52	42.8	67.2

2.3. Photoproduction of H_2O_2 in PSII Preparations at Various Modifications of the WOC

Figure 2A illustrates the dependence of H_2O_2 photoproduction in the PSII membranes, varying in the degree of damage to the WOC, on the duration of illumination. The photoproduction of H_2O_2 by PSII membranes increased with the increasing destruction of the WOC. If, before treatments, the PSII membranes produced about $0.014 \mu\text{mol H}_2\text{O}_2$ per mg Chl for 30 s of illumination ($\lambda > 600 \text{ nm}$, $1500 \mu\text{mol photon s}^{-1} \text{ m}^{-2}$), then after NaCl, CaCl_2 , and NH_2OH treatments, the yield of H_2O_2 was 0.014, 0.018, and $0.045 \mu\text{mol H}_2\text{O}_2$ per mg Chl, respectively. It appears from this that the Mn-depleted PSII membranes, in which the electron supply from water to the reaction centre was inhibited, produced three times more H_2O_2 than other samples. However, the capability of Mn-depleted PSII membranes to the light-induced production of H_2O_2 decreased during illumination. As a consequence, the amount of H_2O_2 produced by the Mn-depleted PSII membranes with 3 min of lighting was close to that generated by untreated samples. Ono and Inoue [34] showed that a gradual release of Mn from the WOC in the CaCl_2 -washed PSII membranes took place, and the Mn abundance in the samples decreased to about one half of the initial level after incubation in CaCl_2 -free medium at 0°C under darkness for 7 h. In our case, the incubation time of the CaCl_2 -treated PSII membranes at 0°C did not exceed 30 minutes, since a small aliquot of the samples was thawed for each series of measurements. In this regard, the number of reaction centres containing two manganese ions should be small based on the total number of reaction centres. Nevertheless, the CaCl_2 -treated samples containing about two Mn ions per RC were specially prepared. The rates of H_2O_2 and $\text{O}_2^{\bullet-}$ photoproduction in these samples were two times higher than those of the CaCl_2 -treated PSII membranes containing four Mn ions per RC (data not presented).

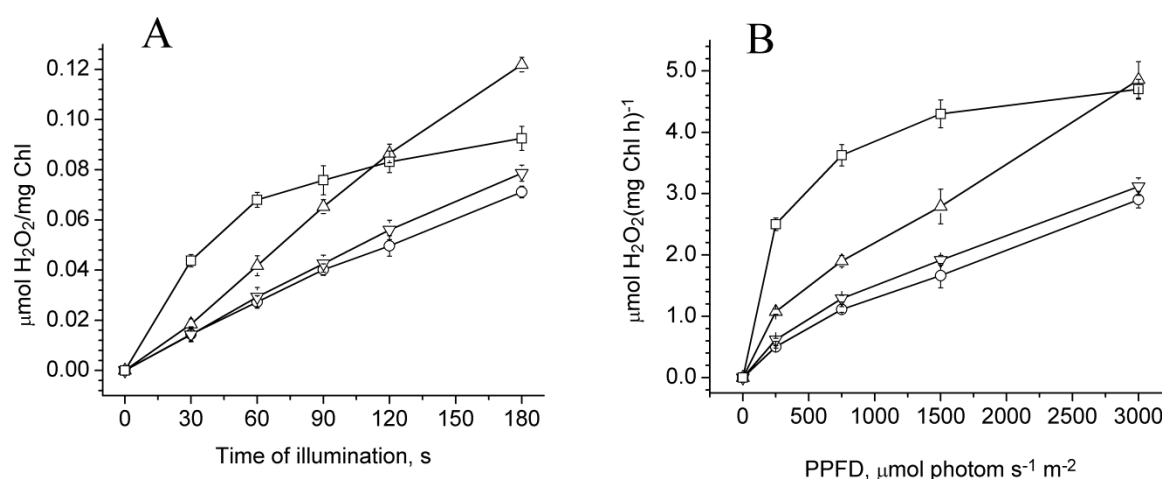


Figure 2. H₂O₂ photoproduction in PSII membranes before (○) and after NaCl (▽), CaCl₂ (Δ), and NH₂OH (□) treatments. **(A)** Dependence of H₂O₂ photoproduction in the PSII membranes on the duration of illumination ($\lambda > 600$ nm, 1500 $\mu\text{mol photon s}^{-1} \text{m}^{-2}$). **(B)** Dependence of H₂O₂ photoproduction in the PSII membranes on light intensity (the samples were illuminated at various light intensities for 1 min). The illumination of the samples was done in a medium containing 20 mM MES–NaOH (pH 6.5), 35 mM NaCl, and 0.4 M sucrose at 25 °C. The concentration of chlorophyll during illumination was 50 $\mu\text{g/mL}$.

Figure 2B shows the dependence of the rate of H₂O₂ photoproduction by the PSII membranes on light intensity. The rate of H₂O₂ production was calculated by monitoring the concentration of H₂O₂ formed upon 1 min illumination of the samples. The rate of H₂O₂ photoproduction in untreated PSII membranes at 250 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ was equal to 0.5 $\mu\text{mol H}_2\text{O}_2 (\text{mg Chl h})^{-1}$, and it increased two times after CaCl₂ treatment of the PSII membranes and five times after Mn removal. The difference in the rate of H₂O₂ photoproduction between the untreated and Mn-depleted PSII membranes gradually decreased with increasing light intensity, to the extent that at the photosynthetic photon flux density (PPFD) of 3000 $\mu\text{mol of photons m}^{-2} \text{s}^{-1}$, the rate of H₂O₂ photoproduction by Mn-depleted PSII membranes was only two times higher than in untreated ones (4.7 and 2.8 $\mu\text{mol H}_2\text{O}_2 (\text{mg Chl h})^{-1}$, respectively). At the same time, the difference in the rates of H₂O₂ production between untreated and NaCl- and CaCl₂-treated PSII membranes upon the increase of PPFD was practically unchanged. Similar to PSII membranes, the removal of Mn clusters from the PSII core complexes stimulated the photoproduction of H₂O₂ (Figure 3). The rate of H₂O₂ photoproduction in Mn-depleted PSII core complexes calculated for 30 s after the start of continuous illumination ($\lambda > 600$ nm, 1500 $\mu\text{mol photon s}^{-1} \text{m}^{-2}$) was four times higher than the untreated samples (16 μmol and 4 $\mu\text{mol H}_2\text{O}_2$ per mg Chl h, respectively). However, the suppression of H₂O₂ production in Mn-depleted PSII core complexes during illumination or at increasing light intensity occurred slower than in the Mn-depleted PSII membranes.

In addition to H₂O₂, other species of peroxides (such as organic hydroperoxides) also can be formed upon the illumination of PSII preparations, which is especially applicable to the Mn-depleted samples [29–31]. To obtain insight into the specificity of homovanilic acid (HVA) for other peroxide species, the reaction of the probe with two peroxides—m-chloroperbenzoic acid (MCPBA) as a model of a lipophilic hydroperoxide and tert-butyl hydroperoxide (TBHP) as a hydrophilic hydroperoxide—was examined. The addition of MCPBA or TBHP at a concentration even ten times higher than H₂O₂ resulted in only a slight increase in the fluorescence intensity of HVA, indicating that the contribution of hydroperoxides (which could be formed on the donor side of PSII) was negligible (Supplementary Materials Figure S1).

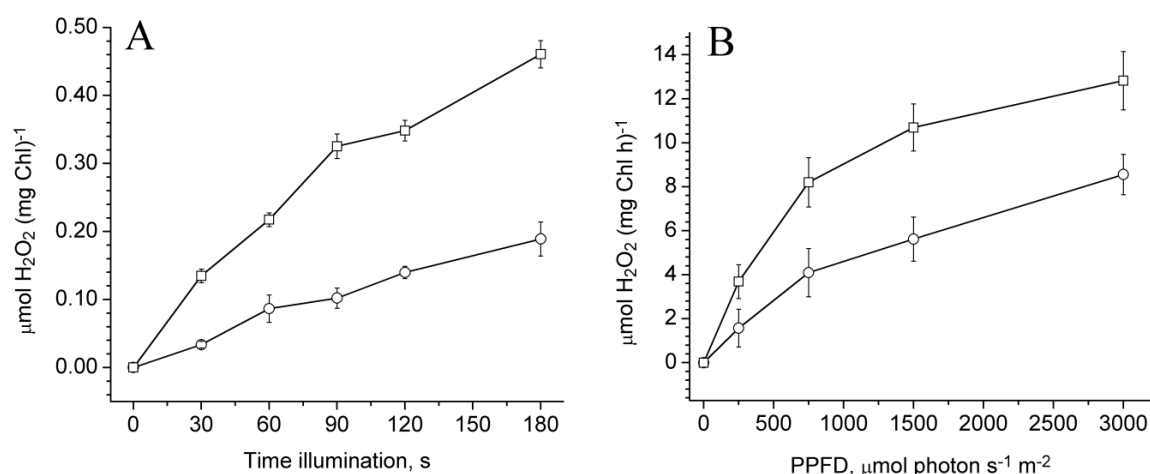


Figure 3. H₂O₂ photoproduction in untreated (○) and Mn-depleted PSII core complexes (□). (A) Dependence of H₂O₂ photoproduction in the PSII core complexes on the duration of illumination ($\lambda > 600$ nm, 1500 $\mu\text{mol photon s}^{-1} \text{m}^{-2}$). (B) Dependence of H₂O₂ photoproduction in the PSII core complexes on light intensity (the samples were illuminated at various light intensities for 1 min). The illumination of the samples was done in a medium containing 20 mM MES–NaOH (pH 6.5), 35 mM NaCl and 0.4 M sucrose at 25 °C. The concentration of chlorophyll during illumination was 50 $\mu\text{g/mL}$.

2.4. Photoproduction of O₂^{•−} in PSII Preparations at Various Modification of the WOC

The main path of H₂O₂ production in PSII is the disproportion of superoxide anion radicals, which are from the one-electron reduction of O₂ on the acceptor side of PSII. The photoproduction of O₂^{•−} in the PSII preparations was investigated using Cyt *c*. To distinguish the photoreduction of Cyt *c* related to O₂^{•−} from its reduction by reduced electron carriers on the acceptor side of PSII [35], the measurements were performed both in the absence and in the presence of superoxide dismutase (SOD). The photoreduction of Cyt *c* in untreated PSII membranes as well as in NaCl- and CaCl₂-treated PSII membranes in the absence of SOD occurred with equal rates (Figure 4A–C, curve 1). The rate of Cyt *c* photoreduction in the Mn-depleted PSII membranes was much higher in comparison with other samples (especially during the first 10 seconds of illumination (Figure 4D, curve 1). The SOD added to the PSII membranes suppressed the Cyt *c* photoreduction and degree of the suppression depending on the destruction of the WOC (Figure 4A–D, curve 2). The inhibition of the Cyt *c* photoreduction with SOD was equal to 50%, 60%, and 76% in the untreated PSII, NaCl-, and CaCl₂-treated PSII membranes, respectively. The addition of SOD completely suppressed Cyt *c* photoreduction by the Mn-depleted PSII membranes, and negative ΔA_{550} was observed (Figure 4D, curve 2) which, as was shown recently [35], is associated with photooxidation of reduced Cyt *c* on the donor side of PSII. Figure 4E shows the kinetics of the Cyt *c* photoreduction after the subtraction of the kinetics measured in the presence of SOD, which demonstrates O₂^{•−}-dependent Cyt *c* reduction. These data indicate that the increase in the damage to the WOC stimulates O₂^{•−} photoproduction by PSII membranes. The removal of Mn clusters from the PSII core complexes also led to a significant increase in the rate of Cyt *c* photoreduction. However, in contrast to PSII membranes, the addition of SOD completely suppressed the Cyt *c* photoreduction both in untreated and Mn-depleted PSII core complexes (Figure 5A, curves 3 and 4), indicating that the samples were not capable of reducing Cyt *c* by electron carriers. Figure 5B shows the Cyt *c* reduction associated with the light-induced O₂^{•−} formation in PSII core complexes. As can be seen from the figure, Mn removal from the WOC led to a significant (more than five times) stimulation of O₂^{•−} photoproduction in PSII core complexes.

The addition of 20 μM diuron led to the almost complete suppression of H₂O₂ and O₂^{•−} photoproduction in all the samples. This demonstrates that H₂O₂ and O₂^{•−} photoproduction is linked to electron transport in PSII.

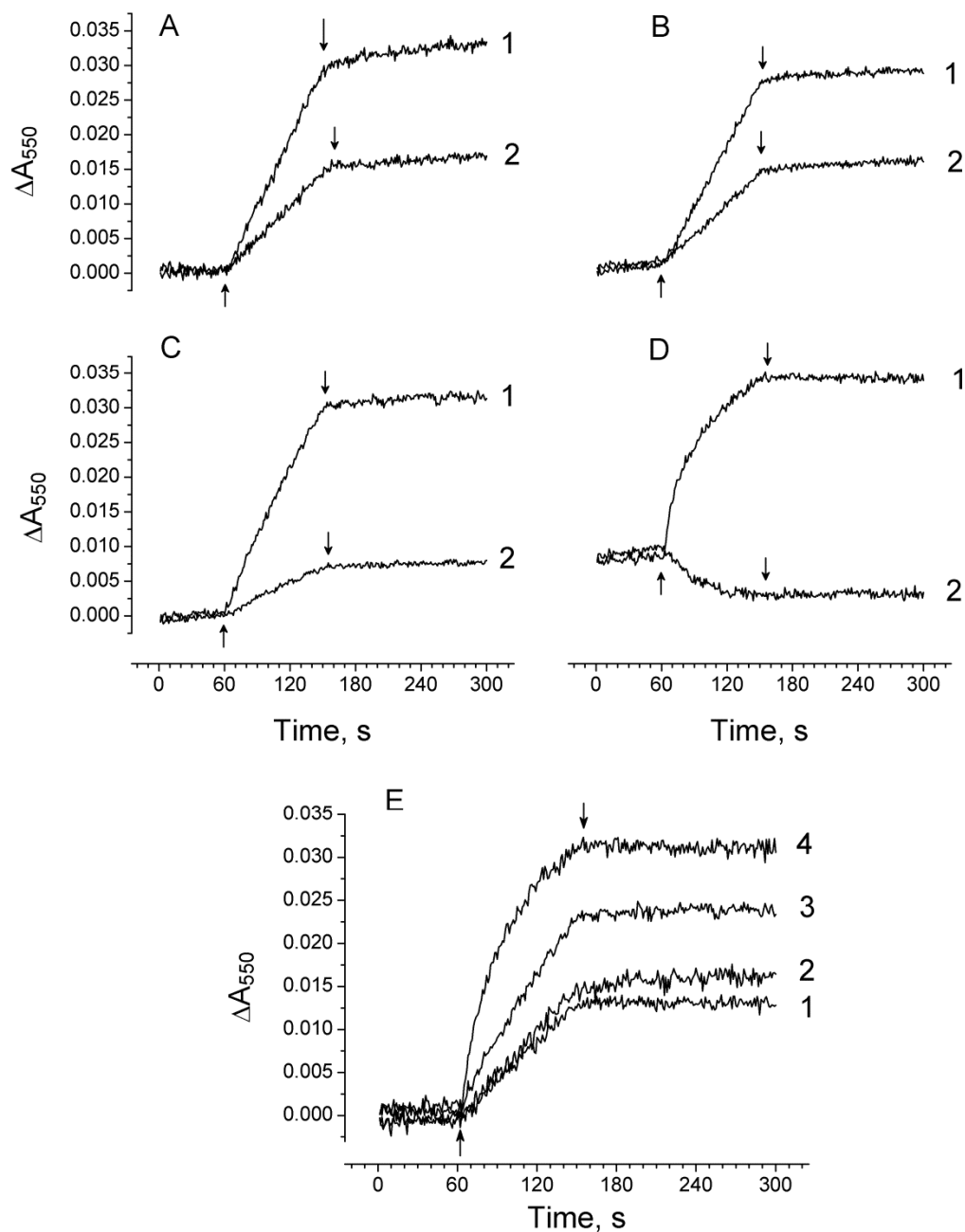


Figure 4. Kinetics of Cyt *c* photoreduction by PSII membranes before (A) and after modification of the water-oxidizing complex caused by treatments of NaCl (B), CaCl₂ (C), and NH₂OH (D). The measurements were done in the absence of additions (1) and after the addition of 50 U/mL SOD (2). Reaction medium contained 50 mM MES–NaOH (pH 6.5), 35 mM NaCl, 0.4 M sucrose, and 10 μ M Cyt *c*. The PSII membranes were illuminated ($\lambda > 600$ nm, 1500 μ mol photon s⁻¹ m⁻²) at chlorophyll concentration of 10 μ g/mL. Up and down arrows indicate light on and off, respectively. (E) Kinetics of Cyt *c* reduction associated with the light-induced O₂^{-•} formation in the PSII membranes before (1) and after modification of the water-oxidizing complex caused by treatments of NaCl (2), CaCl₂ (3) and NH₂OH (4). The kinetics was obtained by the subtraction of the kinetics of Cyt *c* photoreduction measured in the presence of SOD from that measured in the absence of SOD.

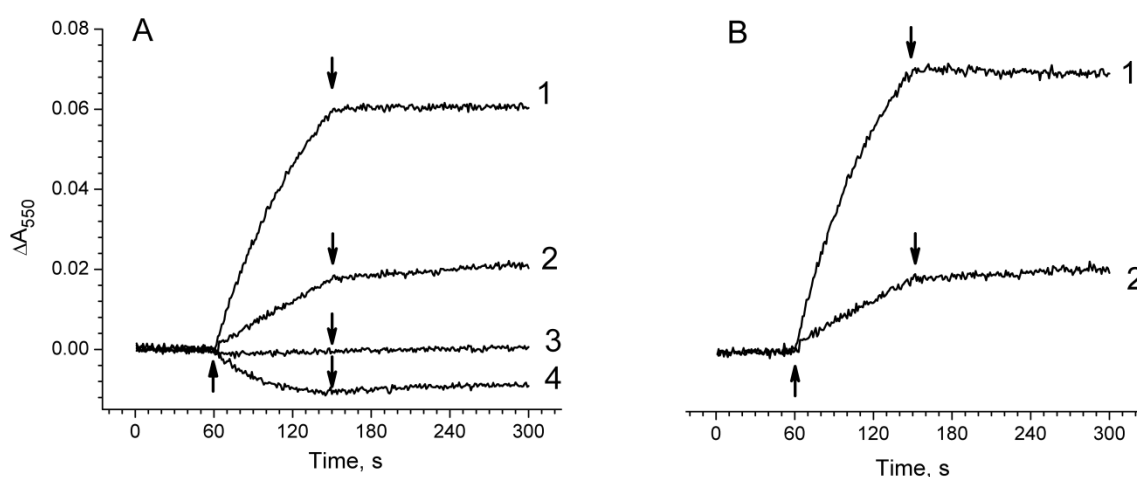


Figure 5. (A) Kinetics of Cyt *c* photoreduction by PSII core complexes before (2, 3) and after NH_2OH treatment (1, 4). The measurements were done in the absence of additions (1, 2) and after the addition of 50 μM SOD (3, 4). (B) Kinetics of Cyt *c* reduction associated with the light-induced $\text{O}_2^{\cdot-}$ formation in the PSII core complexes before (2) and after Mn removal (1). The kinetics was obtained by the subtraction of kinetics of Cyt *c* photoreduction measured in the presence of superoxide dismutase (SOD) from that measured in the absence of SOD. Reaction medium contained 50 mM MES–NaOH (pH 6.5), 35 mM NaCl, 0.4 M sucrose and 10 μM Cyt *c*. The samples were illuminated ($\lambda > 600$ nm, $1500 \mu\text{mol photon s}^{-1} \text{ m}^{-2}$) at chlorophyll concentration of 10 $\mu\text{g/mL}$. Up and down arrows indicate light on and off, respectively.

Table 2 shows the comparison in the rates of H_2O_2 and $\text{O}_2^{\cdot-}$ photoproduction in PSII preparations at various modifications of the WOC. As illustrated above, the ability of the Mn-depleted PSII preparations to produce H_2O_2 and $\text{O}_2^{\cdot-}$ was significantly decreased during illumination as a consequence of their sensitivity to light. Therefore, the rates were calculated for 30 s after the start of illumination ($\lambda > 600$ nm, $1500 \mu\text{mol photon m}^{-2} \text{ s}^{-1}$) of the PSII preparations. The rate of light-induced formation of $\text{O}_2^{\cdot-}$ and H_2O_2 in the PSII preparations rose with the increasing destruction of the WOC, and the photoproduction of $\text{O}_2^{\cdot-}$ in all samples was almost two times higher than H_2O_2 . The data suggest that all or most of the H_2O_2 comes from $\text{O}_2^{\cdot-}$ dismutation, where two molecules of $\text{O}_2^{\cdot-}$ form one peroxide molecule.

Table 2. The rate of H_2O_2 and $\text{O}_2^{\cdot-}$ photoproduction in PSII preparations after various modifications of the water-oxidizing complex. The rates were calculated for 30 s after the start of illumination ($\lambda > 600$ nm, $1500 \mu\text{mol photon m}^{-2} \text{ s}^{-1}$) of the samples.

The Rate of Photoproduction, $\mu\text{mol (mg Chl h)}^{-1}$	PSII Membranes				PSII Core Complexes	
	Untreated	NaCl-Treated	CaCl_2 -Treated	Mn-Depleted	Untreated	Mn-Depleted
H_2O_2	1.7 ± 0.36	1.7 ± 0.3	2.2 ± 0.25	5.25 ± 0.3	4.1 ± 0.6	16.2 ± 1.2
$\text{O}_2^{\cdot-}$	2.7 ± 0.1	2.6 ± 0.2	5.3 ± 0.15	11.3 ± 0.3	6.9 ± 0.25	37.8 ± 0.5

2.5. Effect of Exogenous Electron Donors on the Photoproduction of $\text{O}_2^{\cdot-}$ and H_2O_2 in Mn-Depleted PSII Preparations

Exogenous electron donors effectively restore photoinduced ΔF as a result of an increase in electron flow to the PSII reaction centre [28]. Figure 6I shows the Cyt *c* reduction associated with $\text{O}_2^{\cdot-}$ photoproduction in Mn-depleted PSII membranes (Figure 6IA) and in Mn-depleted PSII core complexes (Figure 6IB) upon the addition of 50 μM diphenylcarbazide (DPC). The restoration of electron flow in the Mn-depleted PSII preparations by DPC resulted in a three-fold suppression of $\text{O}_2^{\cdot-}$ photoproduction in PSII membranes, which was two-fold in PSII core complexes. The effect of the exogenous electron

donor, Mn^{2+} , on the photoproduction of H_2O_2 in the Mn-depleted PSII preparations was studied using an H_2O_2 -dependent couple reaction between 3-methyl-2-benzothiazolinone hydrazone (MBTH) and 3-(dimethylamino) benzoic acid (DMAB) catalyzed by peroxidase. The use of another system for the determination of H_2O_2 was due to the fact that the electron donors used for the restoration of electron flow in the Mn-depleted PSII preparations affected the reaction of H_2O_2 with HVA. In addition to this, Mn^{2+} was used instead of DPC since DPC also affected the determination of H_2O_2 by this measuring system. MnCl_2 (50 μM) added to the samples before illumination diminished the photoproduction of H_2O_2 in Mn-depleted PSII membranes and core complexes by 55% and 45%, respectively (Figure 6II, kinetics 1 and 2). Adding 50 μM MnCl_2 to the samples after illumination had practically no effect on the light-induced yield of H_2O_2 (Figure 6II, kinetics 1'), indicating that MnCl_2 did not affect the H_2O_2 -dependent couple reaction between MBTH and DMAB as well, not leading to H_2O_2 decomposition.

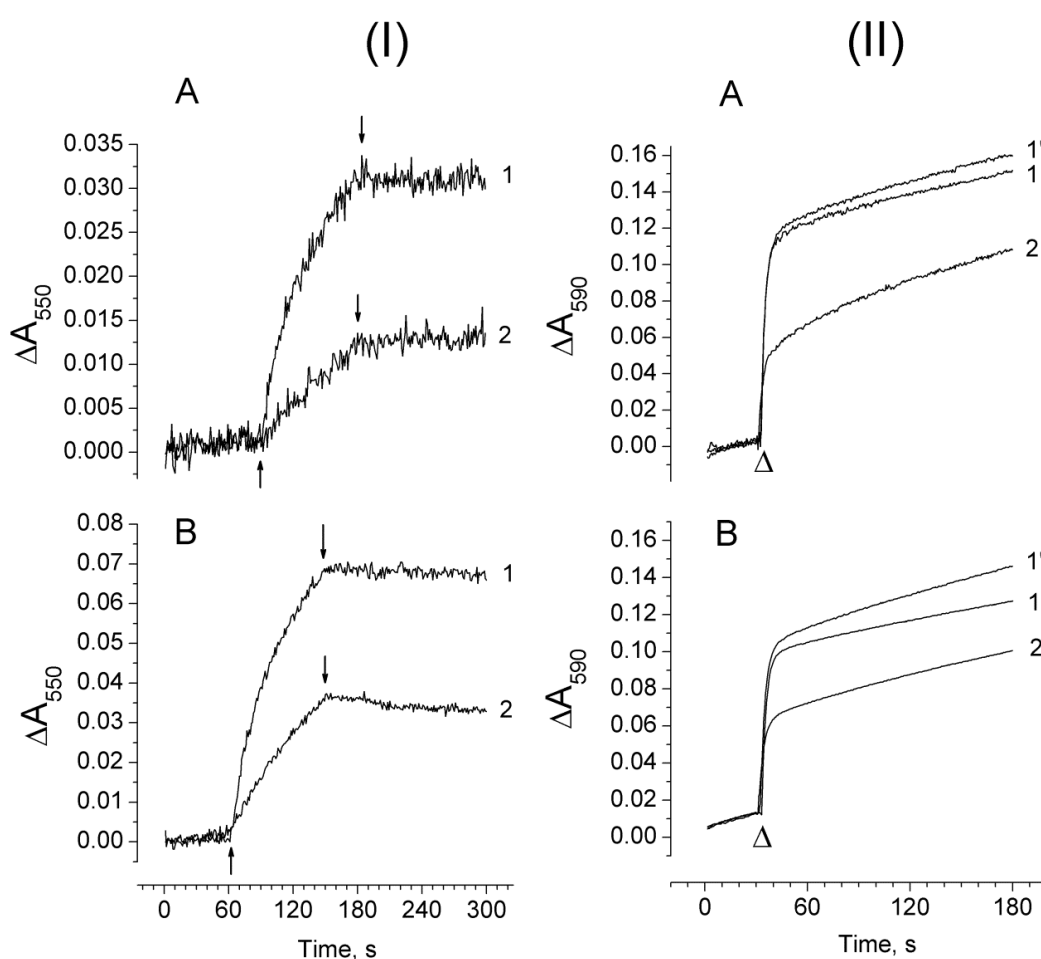


Figure 6. (I) Kinetics of Cyt *c* photoreduction related to the production of $\text{O}_2^{\cdot-}$ in Mn-depleted PSII membranes (A) and Mn-depleted PSII core complexes (B) in the absence (1) and the presence of 50 μM DPC (2). $\uparrow$ and $\downarrow$ light on and off, respectively. (II) Absorption changes at 590 nm measuring H_2O_2 production in the Mn-depleted PSII membranes (A) and the Mn-depleted PSII core complexes (B) after illumination for 90 s ($\lambda > 600$ nm, $1500 \mu\text{mol photon s}^{-1} \text{ m}^{-2}$) in the absence (1) and in the presence of 50 μM MnCl_2 (2). 1'—50 μM MnCl_2 was added after the illumination of the samples. Chlorophyll concentration of the PSII membranes and core complexes was 50 $\mu\text{g Chl/mL}$ and 20 $\mu\text{g Chl/mL}$, respectively. The light-induced yield of H_2O_2 in the samples was measured with 5 mM DMAB, 0.1 mM MBTH, and 3 unit/mL horseradish peroxidase (HRP) (see Material and Methods). Δ —injection of HRP into the suspension of the samples.

3. Discussion

The obtained results demonstrate that the step-by-step disassembly of the WOC leading to the suppression of electron transport from the WOC to RC stimulates H_2O_2 and $\text{O}_2^{\cdot-}$ photoproduction in PSII, and, among the samples, the Mn-depleted PSII preparations (which are not capable of water oxidation) show the highest rate of H_2O_2 and $\text{O}_2^{\cdot-}$ photoproduction. The photoproduction of H_2O_2 in PSII can be associated with both the univalent reduction of O_2 on the acceptor side to $\text{O}_2^{\cdot-}$ (see [24,25]) and H_2O_2 formed on the donor side when the WOC is perturbed without the release of manganese [16,27]. In our case, the stimulation of H_2O_2 photoproduction in the PSII preparations induced by the injury of the WOC was mainly due to the increase in the $\text{O}_2^{\cdot-}$ production on the acceptor side of PSII. This conclusion has been made based on the following observations: (1) The rate of $\text{O}_2^{\cdot-}$ photoproduction was approximately two times higher than H_2O_2 (Table 2), and in the reaction dismutation, two $\text{O}_2^{\cdot-}$ give the yield of one molecule of H_2O_2 (although the part of produced H_2O_2 can be oxidized by PSII during illumination, especially in the presence of exogenous Mn^{2+} [36,37]); (2) the PSII preparations deprived of Mn_4CaO_5 complex (when the water oxidation in PSII was lost entirely) showed maximal activity in $\text{O}_2^{\cdot-}$ and H_2O_2 photoproduction. However, in comparison with the samples holding the Mn cluster, the capability of Mn-depleted PSII membranes for H_2O_2 photoproduction drastically decreased during illumination or at high light intensity. This behavior of Mn-depleted PSII membranes can be attributed to the deficiency of the electron source and high sensitivity of Mn-depleted PSII preparations to photoinhibition. The sources of electrons for $\text{P}_{680}^{+\cdot}$ and $\text{TyrZ}^{\cdot}$ in the absence of an Mn cluster can be chlorophylls and carotenoids (their photooxidation has been shown in several works [38–42]), lipids in the lipid belt around D1 and D2 (their presence in the RC has been demonstrated [2,43]), the amino acid residues involved in coordination of the Mn_4CaO_5 cluster [3], and His located in the vicinity of TyrZ. Apparently, the changes of the acceptor side caused by the modification of the WOC facilitate the photoproduction of $\text{O}_2^{\cdot-}$. However, it cannot be excluded that the donor side of PSII also generates H_2O_2 , especially in the case of CaCl_2 -treated PSII membranes [27,44], but its contribution seems negligible. In order to accurately estimate the contribution of the donor side, it is necessary to separate the H_2O_2 formed on the acceptor side from the donor side.

Pool PQ, pheophytin, Q_A , and Cyt b_{559} are considered to be the primary sources involved in $\text{O}_2^{\cdot-}$ and H_2O_2 photoproduction on the acceptor side (see [24,25]). It is worthwhile to consider the role of these cofactors in the enhancement of $\text{O}_2^{\cdot-}$ and H_2O_2 photoproduction by PSII preparations after the destruction of the WOC.

The pool of PQ is shown to be involved in H_2O_2 formation within the thylakoid membrane [20,21]. The isolation of PSII preparations results in the deprivation of the PQ pool. It was shown that the PQ content was about 2.5 PQ/RC for PSII membranes [45], while the Q_B quinone could be released from its binding site during the isolation of PSII core complexes (these complexes did not emit the B-band arising from $\text{S}_2\text{Q}_\text{B}$ charge recombination, although the vacant Q_B pocket preserved a high affinity for 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU)) [33]. The analysis of PQ in the PSII core complexes isolated from cyanobacterium *Acaryochloris marina* MBIC 11017 showed that these complexes contained about 1.4 PQ per RC [46]. Since the increase of $\text{O}_2^{\cdot-}$ and H_2O_2 photoproduction after damage to the WOC took place in both PSII membranes and core complexes, the participation of the PQ pool in $\text{O}_2^{\cdot-}$ and H_2O_2 photoproduction seems to be vague, although we cannot exclude the possibility that some free PQ in PSII membranes could be involved $\text{O}_2^{\cdot-}$ photoproduction. It was suggested [47] that $\text{O}_2^{\cdot-}$ can be formed via the reduction of O_2 by plastoquinones formed through the one-electron reduction of plastoquinone at the Q_B site and one-electron oxidation of plastoquinol by Cyt b_{559} . Thus, it is possible that the involvement of PQ in $\text{O}_2^{\cdot-}$ photoproduction induced by damage to the WOC occurs via its interaction with Cyt b_{559} .

The redox potential of Pheo (its midpoint redox potential (E_m) of the redox couple $\text{Pheo}/\text{Pheo}^-$ at pH 7 is -610 mV [4,48]) favors the reduction of O_2 to $\text{O}_2^{\cdot-}$, since $E_\text{m}(\text{O}_2/\text{O}_2^{\cdot-})$ is about -160 mV. According to Allakhverdiev and co-workers [49], the $E_\text{m}(\text{Pheo}/\text{Pheo}^-)$ in PSII core complexes from

Synechocystis sp PCC 6803 was -525 mV for untreated and about -609 mV for Mn-depleted samples. Thus, the removal of manganese from the WOC shifts the E_m (Pheo/Pheo $^-$) towards negative values. It seems that this shift in the redox potential of Pheo would not lead to a significant increase of $O_2^- \bullet$ and H_2O_2 photoproduction when the electron transport from the WOC to the RC was inhibited. In addition, the rate of H_2O_2 photoproduction in the Mn-depleted PSII preparations at low light intensity was five times higher than that in the samples containing “native” WOC, i.e., when the accumulation of the long-lived state of Pheo $^-$ is less favorable. By contrast, the production of H_2O_2 in the Mn-depleted PSII preparations decreased with increasing light intensity or duration of illumination. It seems that the electron transfer directly from Pheo $^-$ to O_2 is not productive, although its reduction potential favors this reaction. Perhaps this is due to the recombination between P_{680}^+ and Pheo $^-$ (which is less 5 ns) proceeding much faster than the electron transfer from Pheo $^-$ to O_2 or the difficulty of the formation of $O_2^- \bullet$ within RC. If the enhancement of $O_2^- \bullet/H_2O_2$ photoproduction in Mn-depleted PSII preparations is mainly associated with Pheo, then the restoration of electron flow in the samples by exogenous electron donors (DPC and Mn^{2+}) would lead to the increase in production of $O_2^- \bullet$ and H_2O_2 . However, the restoration of electron flow in the samples diminishes the photoproduction of $O_2^- \bullet$ and H_2O_2 (Figure 6). Nevertheless, the stimulation of $O_2^- \bullet/H_2O_2$ photoproduction can be linked to the shift in the E_m of Pheo/Pheo $^-$ if it is assumed that the light-induced formation of $O_2^- \bullet$ occurs in an aprotic environment, where E_m for $O_2/O_2^- \bullet$ varies from -480 mV to -710 mV [50].

The E_m for Q_A/Q_A^- in intact PSII preparations isolated from spinach has values of -84 mV [51], about -162 mV [52] or -146 mV [49]. The discrepancy in the values is attributed to the removal of bicarbonate from PSII during the measurement procedure of E_m (Q_A/Q_A^-) [53]. Thus, the redox potential of Q_A is not sufficient for the effective reduction of O_2 to $O_2^- \bullet$ (although it is assumed that this reaction occurs because the ratio between O_2 and $O_2^- \bullet$ is strongly shifted towards O_2 [24]). In contrast to Pheo, the removal of the inorganic core of the WOC changes the E_m (Q_A/Q_A^-) by 150 mV to positive values [49,54,55]. Such a change of E_m (Q_A/Q_A^-) should lead to either to a slowdown or an interruption in the electronation of O_2 by Q_A^- . However, the opposite tendency is observed: The destruction of the WOC accompanied by the shift of E_m (Q_A/Q_A^-) to positive values stimulates the light-induced formation of $O_2^- \bullet$. This result may infer that the Q_A site is not directly responsible for the enhancement of $O_2^- \bullet$, and H_2O_2 photoproduction observed after the damage to the WOC. DCMU prevents electron transport between Q_A and Q_B by the competitive binding of the herbicide molecule in the Q_B site on the reaction centre. The blocking of the electron transport between Q_A and Q_B facilitates the light-induced accumulation of Q_A^- , as evidenced by the acceleration of Fv rise upon the addition of diuron [56]. Thus, the 90% suppression of the photoproduction of $O_2^- \bullet$ in the membranes and core complexes of PSII by diuron may also indicate that the main part of $O_2^- \bullet$ is not formed on the Q_A site. It is probably true that this effect of diuron can be associated with the effects on the another components of the PSII reaction centre. It was shown that DCMU influenced the functioning of the WOC, the light-induced accumulation of reduced pheophytin [57], and the redox potential of HP Cyt b_{559} [58].

The involvement of Cyt b_{559} in O_2 reduction is presented in several works (see [15]), and all of them confirm that only LP Cyt b_{559} can be involved in the reduction of O_2 to $O_2^- \bullet$. The redox potential of LP Cyt b_{559} varies from -40 mV to $+80$ mV (see [6]), which is not enough for the reduction of O_2 . However, it is assumed that E_m ($O_2/O_2^- \bullet$) can be close to 0 mV if the concentration of O_2 greatly exceeds the level of produced $O_2^- \bullet$ [24]. Considering this fact, the LP Cyt b_{559} is capable of reducing O_2 to $O_2^- \bullet$. In addition to this, Cyt b_{559} can also exist in the VLP form, having more redox power for the reduction of O_2 (the E_m of the VLP form is from -150 to -200 mV [7,8]). The results presented here show that the increase in the fraction of LP Cyt b_{559} (including its VLP form) induced by damage to the WOC correlates with the rise of $O_2^- \bullet$ and H_2O_2 photoproduction. It seems that the increase of the $O_2^- \bullet$ photoproduction in PSII after the destruction of the WOC occurs due to the increase in the fraction of LP Cyt b_{559} and/or its VLP form. The suppression of $O_2^- \bullet$ and H_2O_2 photoproduction in the Mn-depleted PSII preparations observed upon the addition of the exogenous electron donors (Figure 6)

can be ascribed to the conversion of LP Cyt b_{559} to higher-potential forms. The conversion of LP Cyt b_{559} to IP and HP forms during the illumination of Mn-depleted PSII preparations in the presence of exogenous electron donors to PS II was shown previously by Mizusawa and co-workers [9].

Thus, the destruction of the WOC leading to the suppression of electron transport within the reaction centre of PSII promotes $O_2^{\cdot-}$ and H_2O_2 photoproduction on the acceptor side of PSII through shifts in the redox potential of electron carriers of PSII. It seems that the conversion of HP and IP Cyt b_{559} to the LP form caused by the damage to the WOC makes a significant contribution to the enhancement of photoproduction of $O_2^{\cdot-}$ and H_2O_2 in PSII. However, it is not improbable that a shift in the E_m (Pheo/Pheo $^-$) towards negative values may play a facilitating role in $O_2^{\cdot-}$ photoproduction in terms of its formation in the aprotic environment. Perhaps the light-induced overproduction of $O_2^{\cdot-}/H_2O_2$ associated with damage to the WOC may be a signal for the activation of processes necessary for the repair of damaged PSII, since the photoformation of $O_2^{\cdot-}$ in native (undamaged) PSII is negligible.

4. Materials and Methods

4.1. Isolation of PS II Membranes and PSII Core Complexes

Oxygen-evolving PSII membrane preparations were isolated from spinach leaves according to the procedure in [59]. The samples were suspended in a medium containing 20 mM MES–NaOH (pH 6.5), 35 mM NaCl, 0.33 M sucrose, and 10% glycerol and stored at -76°C . The isolation of PSII core complexes was performed according to the method in [60] with some modification: Bis-Tris buffer was replaced by MES. The concentration of chlorophyll (Chl) was measured as described previously [61]. The manganese content in PSII preparations was determined with an atomic absorption spectrophotometer equipped with a Kvant2A flame atomizer (Cortec, Russia).

4.2. Preparation of PSII Membranes with a Different Degree of Disassembly of the WOC and Mn-Depleted PSII Core Complexes

To obtain PSII membrane preparations with different degrees of disassembly of the WOC, the samples were treated by 1 M NaCl [62], 1 M $CaCl_2$ [63], or 5 mM NH_2OH [64]. According to the literature, the first treatment results in the depletion of two extrinsic proteins (PsbP and PsbQ) of the WOC (NaCl-treated PSII), while the incubation of the PSII preparations in the presence of 1 M $CaCl_2$ releases all the external proteins (PsbP, PsbQ, and PsbO) from the WOC ($CaCl_2$ -treated PSII). Both these treatments do not extract manganese ions from the WOC, which suggests that the Mn cluster is relatively unaffected. The NH_2OH treatment removes PsbP, PsbQ, and PsbO proteins and Mn ions from the WOC, but some amount of PsbO protein remains (Mn-depleted PSII).

Mn-depleted PSII core complexes were obtained by two approaches: (1) PSII core complexes were incubated in the presence of 5 mM NH_2OH for 60 min, and then the samples were transferred to a Q-Sepharose column equilibrated with medium containing 20 mM MES–NaOH (pH 6.5), 35 mM NaCl, and 0.4 M sucrose with 0.03% (w/v) n-dodecyl- β -D-maltoside (medium A). After loading the samples, the column was washed with medium A with 1 mM ethylenediaminetetraacetic acid (EDTA) and then with medium A free from EDTA. The Mn-depleted PSII core complexes were eluted from the column by 100 mM $MgSO_4$ being added into medium A; (2) Mn-depleted PSII core complexes were obtained from Mn-depleted PSII membranes in accordance with the procedure of isolation of PSII core complexes [60].

Atomic absorption spectroscopy measurements of the manganese content in PSII membranes showed that untreated and NaCl-treated PSII preparations had 4.2 ± 0.2 atoms of manganese per PSII reaction centre, while its content was 3.8 ± 0.1 and less than 0.1 Mn per RC in the $CaCl_2$ -treated and the Mn-depleted PSII membranes, respectively. The content of Mn ions in PSII core complexes was 3.9 ± 0.2 for untreated and close to 0 for the NH_2OH -treated samples.

4.3. Measurements of Functional Activity of PSII Preparations

The functional activity of PSII preparations was estimated by photoinduced changes of chlorophyll fluorescence yield (ΔF) related to the photoreduction of the primary electron donor, Q_A , and oxygen evolution measurements. The kinetics of photoinduced ΔF were measured in a 10 mm cuvette at room temperature by using an XE-PAM fluorometer (Walz, Germany). The photosynthetic oxygen evolution was measured in a temperature-controlled chamber by a Clark-type oxygen electrode (Hansatech Instruments, UK) at continuous illumination ($\lambda > 600$ nm, $1500 \mu\text{mol photons s}^{-1} \text{ m}^{-2}$). The measurements were carried out at 25°C in the presence of artificial electron acceptors for PSII 0.1 mM 2,6-dichloro-p-benzoquinone (DCBQ) and 1 mM $\text{K}_3[\text{Fe}(\text{CN})_6]$.

4.4. Determination of H_2O_2 Photoproduction by PSII Preparations

The photoproduction of H_2O_2 in PSII membranes or core complexes was studied using the fluorescent probe homovanillic acid (HVA). The method is based on the H_2O_2 -dependent oxidation of HVA mediated by horseradish peroxidase (HRP) to a highly fluorescent dimer [65]. The PSII preparations, resuspended in medium containing 20 mM MES-NaOH (pH 6.5), 35 mM NaCl, and 0.4 M sucrose at $50 \mu\text{g}$ of Chl/mL, were illuminated or kept under darkness at 25°C . Then, an aliquot ($500 \mu\text{L}$) of the samples was added into the same volume of the reaction medium containing 100 mM Hepes (pH 7.6), $600 \mu\text{M}$ HVA, and 2 U/mL HRP. After 30 min incubation at 37°C , the PSII membranes were centrifuged at $12,000 \text{ g}$ for 2 min. The supernatant was collected, and its fluorescence spectrum ($350\text{--}500 \text{ nm}$, $\lambda_{\text{ex}} = 312 \text{ nm}$) was recorded with a Cary Eclipse fluorescence spectrophotometer (Agilent, USA). To remove the PSII core complexes from the solution, they were loaded on an Amicon Ultra centrifugal filter (Ultracel 30K, Merck Millipore, Germany) and centrifuged at 5000 g for 15 min. The fraction passing through the filter (free from PSII core complexes) was collected, and the fluorescence spectrum was recorded. The difference between the fluorescence spectra of illuminated and unilluminated samples, designated as the “light minus dark” fluorescence spectrum, represented the light-induced formation of H_2O_2 . The number of H_2O_2 formed under the illumination of the PSII preparations was calculated from the fluorescence intensity of HVA upon the addition of $5 \mu\text{M}$ H_2O_2 . The effect of the exogenous electron donor, Mn^{2+} , on the photoproduction of H_2O_2 in Mn-depleted PSII was examined by the method based on the oxidative coupling of 3-methyl-2-benzothiazolinone hydrazone (MBTH) and 3-(dimethylamino) benzoic acid (DMAB) in the presence of H_2O_2 peroxidase catalyzes, with the couple reaction between MBTH and DMAB with the formation of a deep purple compound having an absorption band between 575 and 600 nm with a peak at 590 nm [27,66]. The use of this method for detecting hydrogen peroxide was because Mn^{2+} did not interfere with the determination of H_2O_2 when using this system, while the presence of Mn^{2+} affected the detection of H_2O_2 by HVA. The measurements were performed as follows: 5 mM DMAB and 0.1 mM MBTH were added to the samples illuminated in the absence or the presence of $50 \mu\text{M}$ MnCl_2 , then the change at 590 nm was recorded before and after the injection of HRP (3 U/mL).

4.5. Determination of $\text{O}_2^{\bullet-}$ Photoproduction by PSII Preparations

The light-induced generation of $\text{O}_2^{\bullet-}$ in PSII was detected by cytochrome *c* (Cyt *c*) [67,68]. PSII membranes or core complexes were resuspended at $10 \mu\text{g}$ Chl/mL in a buffer solution containing 50 mM MES-NaOH (pH 6.5), 35 mM NaCl, 0.4 M sucrose, and $10 \mu\text{M}$ Cyt *c*. Kinetics of absorbance changes at 550 nm related to the reduction of Cyt *c* upon illumination of PSII preparations with red light ($\lambda > 600 \text{ nm}$, $1500 \mu\text{mol (photon) s}^{-1} \text{ m}^{-2}$) were measured in a 10 mm cuvette at room temperature using a spectrophotometer Agilent 8453 (USA). The rate of photoreduction of Cyt *c* was estimated by monitoring the concentration of reduced Cyt *c*. The amount of reduced Cyt *c* was calculated using the differential extinction coefficient between ferrocycytochrome *c* and ferricytochrome *c* at 550 nm (21.1 mM^{-1}).

4.6. Analysis of Redox Forms of Cyt b_{559} in PSII Preparations

Redox states of Cyt b_{559} in PSII preparations were determined by measuring the differential (reduced-minus-oxidized) absorption spectrum of Cyt b_{559} on a Shimadzu UV-1800 (Japan) spectrophotometer. To oxidize Cyt b_{559} , 50 μ M potassium ferricyanide was added. The reduction of the HP, IP, and LP (LP+VLP) forms of Cyt b_{559} was achieved by the stepwise addition of 5 mM hydroquinone, 5 mM sodium ascorbate, and sodium dithionite, respectively. After each addition of the redox agent, a differential absorption spectrum was recorded. The content of HP Cyt b_{559} was attributable to the spectra of Cyt b_{559} obtained upon the addition of hydroquinone to the samples with ferricyanide. The fraction of IP Cyt b_{559} was determined as the difference between the spectra of Cyt b_{559} reduced by ascorbate and the spectra of Cyt b_{559} reduced by hydroquinone, for the LP form of Cyt b_{559} , and from the spectra of dithionite-reduced Cyt b_{559} were subtracted the ascorbate-reduced spectra of Cyt b_{559} .

Supplementary Materials: The following are available online at <http://www.mdpi.com/2223-7747/8/9/329/s1>, Figure S1: Time course of homovanillic acid oxidation at 37 °C induced by 10 μ M H_2O_2 (○), 100 μ M tert-Butyl hydroperoxide, and 100 μ M m-Chloroperbenzoic acid (□).

Funding: The work was supported by the Russian Science Foundation (grant 14-14-00535); the results presented in Figures 3, 5 and 6 were obtained with support from the Ministry of Education and Science of the Russian Federation (theme AAAA-A17-117030110136-8).

Acknowledgments: I thank D. Yanykin for discussion of this work.

Conflicts of Interest: The authors declare no conflict of interest.

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