



Article Polarizabilities and Rydberg States in the Presence of a Debye Potential

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Abstract: Polarizabilities and hyperpolarizabilities, α_1 , β_1 , γ_1 , α_2 , β_2 , γ_2 , α_3 , β_3 , γ_3 , δ and ε of hydrogenic systems have been calculated in the presence of a Debye–Huckel potential, using pseudostates for the S, P, D and F states. All of these converge very quickly as the number of terms in the pseudostates is increased and are essentially independent of the nonlinear parameters. All the results are in good agreement with the results obtained for hydrogenic systems obtained by Drachman. The effective potential seen by the outer electron is $-\alpha_1/x^4 + (6\beta_1 - \alpha_2)/x^6$ + higher-order terms, where x is the distance from the outer electron to the nucleus. The exchange and electron-electron correlations are unimportant because the outer electron is far away from the nucleus. This implies that the conventional variational calculations are not necessary. The results agree well with the results of Drachman for the screening parameter equal to zero in the Debye-Huckel potential. We can calculate the energies of Rydberg states by using the polarizabilities and hyperpolarizabilities in the presence of Debye potential seen by the outer electron when the atoms are embedded in a plasma. Most calculations are carried out in the absence of the Debye-Huckel potential. However, it is not possible to carry out experiments when there is a complete absence of plasma at a particular electron temperature and density. The present calculations of polarizabilities and hyperpolarizabilities will provide accurate results for Rydberg states when the measurements for such states are carried out.

1. Introduction

Recently, Qi et al. [1] reported a calculation of the dipole polarizability α_1 of the hydrogenic systems in the screened Coulomb potential due to the systems being in hot dense plasmas. The screened potential in this case is the Debye–Huckel potential given by Equation (1).

$$V(r) = (-Ze^2/r) \exp(-\mu r),$$
 (1)

where $1/\mu = (k_B T_e/4\pi e^2 n_e)^{1/2}$ is the Debye screening length, k_B is the Boltzmann constant, T_e and n_e are the plasma electron temperature and density, respectively. There are other calculations of energy and polarizability by Zimmermann [2], Bahar et al. [3], Paul and Ho [4], Fowler [5] and Saha et al. [6]. The expression for polarizabilities is given by Equation (1).

$$S_{i,k} = \Sigma < 0 | v_i | N > | < N | v_i | 0 > / (E_N - E_0)^k,$$
(2)

where $v_i = r_1^i P_i(\cos \theta_i)$ and the various polarizabilities are given by Equation (2).

 $\alpha_i = S_{i,1}$, $\beta_i = S_{i,2}$ and $\gamma_i = S_{i,3}$. Ψ_N are intermediate states of the appropriate angular momentum and Σ indicates the sum of all such states.

2. Calculations and Results

Qi et al. [1] solved the Schrodinger equation in both the discrete and continuous spectrum of the potential given in Equation (1) by using the symplectic integration scheme. The contribution of the continuum states to the polarizability is particularly important for large μ when the electron binding energy is small and coupling to the continuum is



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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). strong. Saha et al. [6] solved the eigenvalue problem by the variational method using a large Slater-type orbitals basis for the discrete states. The contribution of the continuum states was not considered.

In the present calculation, we used pseudostates [7] for helium and the negative hydrogen ion to calculate polarizabilities and hyperpolarizabilities. In [7], the wave functions were of Hylleraas form because the systems consist of two electrons. Definitive results were obtained for the polarizabilities and hyperpolarizabilities. Now we are dealing with single-electron systems and therefore wave functions consist of one electron only and are very simple.

Pseudostates used in this calculation are given by

$$\Psi_{\rm S} = \Sigma_{i=0} C_i r^i \exp(-a_{\rm s} r) \, Y_{00}(\Omega), \tag{3}$$

$$\Psi_{\rm P} = \Sigma_{\rm i=1} D_{\rm i} r^{\rm i} \exp(-a_{\rm p} r) \, \Upsilon_{\rm 10}(\Omega), \tag{4}$$

$$\Psi_{\rm D} = \Sigma_{i=2} E_i r^i \exp(-a_d r) Y_{20}(\Omega), \tag{5}$$

$$\Psi_{\rm F} = \Sigma_{\rm i=3} F_{\rm i} r^{\rm i} \exp(-a_{\rm f} r) \, \Upsilon_{\rm 30}(\Omega), \tag{6}$$

where C_i , D_i , E_i , and F_i , are the eigenvectors. With a few terms and taking the nonlinear parameters equal to 1.0 for hydrogen atoms, we obtained results for α_i , β_i , and γ_i which agree very well for $\mu = 0$ with the known results. We varied all the nonlinear parameters in Equations (3)–(6). However, results were not sensitive to the variation of nonlinear parameters. Therefore, we kept it fixed at 1.0. It is certainly possible to use the hydrogenic functions. The use of pseudostates makes calculations very simple and straightforward. This reduces the efforts involved in the computation. We give in Table 1 the presently calculated results for α_i for various values of μ , using 20 terms in the expansion for wave functions (3) and (4). The results are very well converged for shorter expansions as well. We see that the agreement is very good. The present calculation is very easy to carry out compared to that of Qi et al. [1]. They have to calculate each bound state and the wave function corresponding to that state. They also included the continuum of the electronhydrogenic system. It should be noticed that only those values of μ can be used to calculate α_1 for which the ground state of the hydrogen atom remains bound.

Table 1. Comparison of the presently calculated results for α_i for hydrogen atom for various values of μ , using 20 terms in the wave functions.

М	Present	Qi et al. [1]
0.00	4.500	4.500
0.01	4.50220	Α
0.02	4.50868	4.50820
0.025	4.51346	4.51299
0.050	4.55220	4.55176
0.0625	4.58049	4.58003
0.10	4.69978	4.69933
0.20	5.27637	5.27661

A: No results are given for this value of μ by the authors.

In Table 2, we give β_1 , γ_1 , α_2 , β_2 , γ_2 , α_3 , β_3 , γ_3 , ε and δ for a few values of μ . The third-order polarizability δ has the following form [8]:

$$\delta = \Sigma_{N,M} < 0 | v_i | N > < N | v_j | M > < M | < M | v_k | 0 > /(E_N - E_0)(E_M - E_0),$$
(7)

	$\mu = 0.0$	0.02	0.025	0.05	0.0625	0.10	0.20
β_1	5.3750	5.3917	5.4009	5.4751	5.5305	5.7636	6.9384
γ_1	6.6458	6.6750	6.6911	6.8219	6.9182	7.3313	9.5128
α2	15.000	15.0540	15.0838	15.3246	15.5008	16.2498	20.0479
β ₂	13.3750	13.4453	13.4840	13.7977	14.0284	15.0195	20.3300
γ ₂	12.4948	12.5829	12.6315	13.0262	13.3179	14.5872	21.8128
α3	131.2500	131.989	132.396	135.6875	138.1049	148.490	204.341
β ₃	102.031	102.817	103.249	106.758	109.349	120.632	185.434
γ ₃	83.2044	84.0320	84.4974	88.1980	90.9547	103.142	178.390
E	59.2125	59.7110	59.9307	61.7172	63.0371	68.7756	101.226
Δ	106.500	107.036	107.331	109.724	111.484	119.044	159.391

Table 2. Polarizabilities and hyperpolarizabilities for hydrogen atoms for various values of µ.

The values that (ijk) can take are all the permutations of [1,2].

The fourth-order hyperpolarizability only involving dipole terms has the following form [8]:

 $\varepsilon = \Sigma_{N,M,P} < 0 | v_1 | N > < N | v_1 | M > < M | v_1 | P | 0 > /(E_N - E_0)(E_M - E_0)(E_P - E_0)$ (8)

In Equations (7) and (8), potentials v_1 , v_2 , and v_3 are given reference [8] and are not repeated here. The notation $|0\rangle$, $|N\rangle$, $|M\rangle$. and $|P\rangle$ represent wave functions for the angular momenta *S*, *P*, *D*, and *F* given in Equations (3)–(6) to be appropriately used to obtain nonzero matrix elements in Equations (7) and (8).

Similarly, by taking the nonlinear parameters equal to 2.0, we obtain results which agree with those obtained by Drachman [7] for $\mu = 0$, using the Dalgarno and Lewis method [9]. The present results are given in Table 3. It can be seen that the polarizabilities and hyperpolarizabilities increase as μ increases. Since the ground state of the He ion is very tightly bound compared to that of the hydrogen atom, it is possible to have much larger values of μ .

Table 3. Polarizabilities and hyperpolarizabilities for the helium ion for various values of μ .

	μ = 0.0	0.02	0.025	0.05	0.0625	0.10	0.20
α1	0.28125	0.28139	0.28146	0.28209	0.28255	0.28451	0.29374
β_1	0.08394	0.08406	0.08409	0.08439	0.08461	0.08556	0.09006
γ_1	0.02596	0.02599	0.02601	0.02614	0.02623	0.02665	0.02864
α2	0.23438	0.23459	0.02347	0.23568	0.23640	0.23945	0.25390
β2	0.05225	0.05232	0.05235	0.05267	0.05291	0.05389	0.58670
γ ₂	0.01220	0.01222	0.01224	0.01234	0.01241	0.01272	0.14245
α3	0.51269	0.51343	0.51384	0.51717	0.51962	0.53003	0.58004
β ₃	0.09964	0.09984	0.09994	0.10083	0.10148	0.10426	0.11780
γ ₃	0.02031	0.02037	0.02039	0.02063	0.02080	0.02153	0.02518
ε	0.05792	0.05802	0.05808	0.05853	0.05886	0.06027	0.06716
δ	0.41602	0.41655	0.41684	0.41926	0.42104	0.42861	0.46501

3. Rydberg States of He

The long-range potential, in terms of polarizabilities, seen by the outer electron is given by

$$U(x) = -\alpha_1/x^4 + (6\beta_1 - \alpha_2)/x^6 + (\delta + 16\gamma/5)/x^7 + (-\alpha_3 + 15\beta_2 - \epsilon - \alpha_1\beta_1 - 72\gamma_1 [1 + L(L+1)/10])/x^8,$$
(9)

where x is the distance of the outer electron from the nucleus. In Table 4, we give the expectation value of U(x) in MHz for the wave function of the outer electron in N = 10 and L = 7 and 8 states and compare the present results for μ = 0.0 with those obtained by Drachman [10] without the second-order corrections. We see that the agreement is very good even when very simple wave functions are used.

Table 4. Comparison of the presently calculated Rydberg states with the results of Drachman.

Ν	L	Present (MHz)	Drachman [6] (MHz)
10	7	-48.60605124	-48.60604738
10	8	-24.17853458	-24.17853458

The expectation values of $1/x^n$ for n = 4, 5, 6, 7, 8, 9 and 10 were calculated for $\mu = 0.0$ and therefore cannot be used when the helium atoms are embedded in the plasma.

4. Transition Rates

Transition rates are given by

$$A(np \to 1s) = 8.032 \times 10^9 \frac{(E_{np} - E_{1s})^3}{9} \int_0^\infty R_{np}(r) r^3 R_{1s}(r) dr$$
(10)

In the above equation, $R_{np}(r)$ and $R_{1s}(r)$ are the hydrogen functions for np and 1s states. In Table 5, we show how the transition rates from 4p, 3p and 2p states to the 1s state in hydrogen atoms change with the screening parameter. It can be seen that all the rates decrease with the increase in the screening parameter μ . Using the exact wave functions, we find the transition rates for 2p, 3p and 4p states to 1s state equal to 0.626, 0.167 and 0.0682 in units of 10^9 s^{-1} when there is no screening and exact hydrogenic functions are used. The last one is not in agreement with the one obtained from pseudostates while the first two are in good agreement with those obtained using pseudostates.

Table 5. Transition rates in sec $^{(-1)}$ 4p, 3p and 2p states of hydrogen atoms to the 1s state. The rates have been multiplied by $10^{(-9)}$.

μ	A (2p→1s)	A (3p $ ightarrow$ 1s)	A (4p→1s)
0.000	0.624293	0.166670	0.071760
0.020	0.618343	0.160013	0.066440
0.025	0.615137	0.156613	0.064150
0.050	0.590180	0.131877	0.053783
0.0625	0.572693	0.115647	0.051620
0.1000	0.503290	0.061607	0.058523
0.2000	0.215653	0.034923	0.070210

5. Conclusions

We showed that by using pseudostates, it is possible to obtain good results for polarizabilities and hyperpolarizabilities for hydrogenic systems, energies of Rydberg states in the helium atoms and transition rates in hydrogen atoms. The present results obtained using pseudostates for $\mu = 0$ agree with those obtained using elaborate wave functions. Since there is always the presence of a plasma at a particular electron temperature and density, the present results will be useful for comparison with observations. A detailed account of development of the field and possible applications is given in reference [10]. Using the perturbation theory, wave functions, polarizabilities and hyperpolarizabilities given here, the energies of various Rydberg states of any quantum numbers can easily be calculated in the Debye potential.

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