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Studies on the Reactivity of (9-Methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-*d*]pyrimidin-11-yl)hydrazine Towards Some Reagents for Biological Evaluation

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Received:

doi:10.3797/scipharm.0910-11

October 20th 2009

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Sci Pharm. 2010; 78: 1–12

Published: December 11th 2009 Accepted: November 27th 2009

This article is available from: http://dx.doi.org/10.3797/scipharm.0910-11

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Abstract

(9-Methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-*d*]pyrimidin-11-yl)hydrazine (**1**) was used as a precursor for preparation of some novel 1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-*d*]pyrimidin-11-yl)-1*H*-pyrazoles **2–7**, -1*H*-isoindole-1,3(2*H*)-dione **8**, and -pyridazin-3(2*H*)-one **9**. Moreover, the acyclic *C*-nucleosides **10** and **11** were prepared by treating compound **1** with D-glucose. The *in vitro* antimicrobial activity of the tested compounds was evaluated by measuring the zone diameters and some of the prepared products showed potent antimicrobial activity in compared with those of well known drugs (standard). In general, the non-acetylated sugar hydrazone derivative **10** showed the highest antibacterial and antifungal potency among the tested compounds and standard with IZ = 22, 21 and 22 mm and MIC = 62.5 and 31.25 µg/ml, respectively.

Keywords

Thieno[2,3-*d*]pyrimidine • Pyrazole • Pyridazine • C-Nucleosides • Antimicrobial activity

Introduction

Pyrimidine and fused heterocyclic pyrimidine derivatives have attracted a great deal of interest in particular 4-hydrazinopyrimidine derivatives, which were tested for their medicinal, bactericidal and fungicidal activity [1–4]. Pyrimidine and heterocyclic annulated pyrimidine derivatives attracted great interest due to the wide variety of interesting biological activities observed for these compounds, such as anticancer [4], antiviral [5], Anti-HIV-1 Activity [6], anti-inflammatory [7] and antimicrobial activities [8]. In addition, several substituted dihydronaphthothienopyrimidine derivatives showed antimicrobial activities against *Bacillus subtilis, Escherichia coli, Aspergillus niger* and *Candida albicans* [9], and their ester-containing derivatives demonstrated more antimicrobial activities than the corresponding cyano-containing analogs. In view of the above and in continuation of our research program concerned with structural modification of certain biologically active heterocyclic nuclei with the purpose of enhancing their biological activity [10–16], we aimed to incorporate a fused pyrimidine moiety with other heterocyclic ring system to obtain new functions in an attempt to improve the antimicrobial profile of compounds containing the dihydronaphthothienopyrimidine ring system.

Results and Discussion

From the view of biological activity, polycondensed fused heteroaromatic systems are often of much greater interest than the constituent monocyclic compounds. The appearance of gualitatively new properties of an annelated molecule, enlargement of the possibility of varying pharmacophore groups in different positions of the molecule and the ability of the latter to interact with various receptors adopting several conformations are apparently of crucial importance [17]. In this investigation, compound 1 [18] was dissolved in ethanol and refluxed with (ethoxymethylene)malononitrile, tetracyanoethylene, [bis(methylthio)methylene]malononitrile, ethyl (ethoxymethylene)cyanoacetate or methyl [bis(methylthio)methylene]cyanoacetate to afford the corresponding substituted pyrazole derivatives 2-6, respectively (Sch. 1). The structures of the latter compounds were confirmed on the basis of their elemental analysis and spectral data (cf. Exp.). The IR spectra of compounds 2-4 showed absorption bands characteristic for NH₂ and C≡N groups, while those of compounds 5 and 6 revealed absorption bands characteristic for NH₂ and C=O. Also, the ¹H NMR spectra showed signals at δ = 7.10, 6.55, 6.60, 6.81 and 7.00 ppm due to NH₂ (exchangeable with D_2O) for compounds **2–6**, respectively. The MS gave the molecular ion peaks at m/z (%) = 358 (100), 383 (89), 404 (100), 405 (100), and 437 (100) for compounds 2-6, respectively. Similarly, when compound 1 was refluxed with acetyl acetone the pyrazole derivative 7 was obtained (Sch. 1). The ¹H NMR and ¹³C NMR spectra of the latter compound showed signals at δ = 2.05, 2.19 ppm and 11.97, 13.29 ppm for (2CH₃), while its IR spectrum revealed the absence of NH, NH₂ groups. The MS, gave the molecular ion peak at m/z (%) = 346 (78).

On the other hand, compound **1** was refluxed with tetrachlorophthalic anhydride in glacial acetic acid to afford *N*-imido derivative **8** (Sch. 2). The presence of NH and two C=O groups in the IR spectra and the NH-proton (D₂O exchangeable) in the ¹H-NMR spectrum confirmed its structure. Also, the MS of compound **8** gave fragments showing the isotopic pattern due to the presence of chlorine atoms (cf. Exp.).



Sch. 1.



Sch. 2.

2-(9-Methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-6-phenyl-

4,5-dihydropyridazin-3(2*H*)-one (**9**) was prepared by heating the hydrazine derivative **1** with β -benzoylpropionic acid. Inspection of the IR spectrum of product **9** showed the presence of the C=O group and its ¹H NMR spectrum revealed signals at δ 2.43, 3.61 ppm for 2CH₂ of pyridazine ring. In addition, the ¹³C NMR spectrum showed signal at δ 173.80 accountable for the C=O group (cf. Exp.).

The hydrazone derivative **10** was prepared by reacting compound **1** with D-glucose in the presence of a catalytic amount of glacial acetic acid. The product **10** revealed absorption bands for OH and NH groups in IR spectra and its ¹H NMR spectrum showed the presence of the sugar protons, NH, and azo-methine (C<u>H</u>=N) (cf. Exp.).

Acetylation of the hydrazone derivative **10** with acetic anhydride/pyridine at room temperature gave unexpected O-acetylated cyclic C-nucleosides **11** (Sch. 2). The absence of NH as well as the azo-methine (C<u>H</u>=N) in ¹H NMR spectra confirmed its structure (cf. Exp.). Also, the ¹³C NMR spectrum showed signals accountable for the acetylated sugar residue (cf. Exp.).

The formation of compound **11** might have taken place *via* oxidative cyclization of the *O*-acetylated sugar. In general, the Dimroth type rearrangement of *S*-triazolopyrimidines was intensively discussed and verified with X-ray diffraction by Rashad et al [16]. So, the triazolo[1,5-*c*]pyrimidine derivative **11** was obtained directly via Dimroth type rearrangement of its triazolo[4,3-*c*]pyrimidine derivative.

Antimicrobial activity

Compound No.	Escherichia coli	Bacillus subtilis	Candida albicans
2	16	16	11
3	16.5	16	12
4	16	18	16
5	12	11	11
6	11	10	10
7	11	12	10
8	12	12	10
9	20	20	15
10	22	21	22
11	12	11	22
Chloramphenicol	24	24	_
fluconazole	_	_	26

Tab. 1.The inhibition zones diameter (IZ) in mm.

Highly active (inhibition zone > 20 mm); Moderately active (inhibition zone16–19 mm); Slightly active (inhibition zone 11–15 mm); (–) no inhibition zone The *in vitro* antimicrobial activity of the tested compounds was evaluated by measuring the zone diameters and the results were compared with those of well known drugs (standard). Among the tested compounds, for gram-positive and gram-negative bacteria, it was noticed that ß-enaminonitriles of pyrazole ring system **2–4** (IZ = 16–18 mm and MIC = 65 μ g/ml) demonstrated inhibitory activities more than β -enaminoesters **5** and **6** (IZ = 10–12 mm and MIC = 125 μ g/ml). However, the pyridazine derivative **9** and non-acetylated sugar **10** revealed the most significant antibacterial activities (IZ = 20-22 mm and MIC = 62.5 μ g/ml). On the other hand, the non-acetylated sugar **10** and the acetylated cyclic *C*-nucleoside **11** revealed more effective antifungal activity than the other tested compounds showing IZ = 22 mm and MIC = 31.25 μ g/ml. However, replacement of the hydroxyl moiety in **10** by acetyl group in **11** led to decrease the antibacterial potency of **10** showing IZ = 11, 12 mm, MIC = 62.5 μ g/ml and the antifungal activity was not affected. In general, the non-acetylated sugar hydrazone derivative **10** showed the highest antibacterial and antifungal potency among the tested compounds and standard with IZ = 22, 21 and 22 mm and MIC = 62.5 μ g/ml, respectively.

Compound No.	Escherichia coli	Candida albicans	Bacillus subtilis
4	65	31.25	65
5	125	31.25	125
9	62.5	15	62.5
10	62.5	31.25	62.5
11	12	31.25	11
Chloramphenicol	10	_	10
Fluconazole	_	0.5	_

Tab. 2. MIC in μ g/ml of the most active compounds.

Experimental

All melting points were uncorrected and measured using an Electro-thermal IA 9100 apparatus (Shimadzu, Japan). Microanalytical data were performed by Vario El-Mentar apparatus (Shimadzu, Japan), National Research Centre, Cairo, Egypt. The IR spectra (KBr) were recorded on a Perkin-Elmer 1650 spectrophotometer, National Research Centre, Cairo, Egypt. ¹H NMR and ¹³C NMR spectra were determined on a Varian Mercury (300 MHz) spectrometer (Varian, UK) and the chemical shifts were expressed in ppm relative to TMS as internal reference, Faculty of science, Cairo University, Egypt. Mass spectra were recorded on 70 eV EI Ms-QP 1000 EX (Shimadzu, Japan), National Research Centre, Cairo, Egypt.

Compound **1** was prepared according to a reported method [18].

Preparation of compounds 2–6.

General procedure

To a solution of compound **1** (2.82g, 1mmol) in (20 ml) dry ethanol, (ethoxymethylene)malononitrile, tetracyanoethylene, [bis(methylthio)methylene]malononitrile, ethyl (ethoxymethylene)cyanoacetate or methyl [bis(methylthio)methylene]cyanoacetate (1 mmol) was added and the reaction mixtures were refluxed for 2, 5, 4, 3 and 4 h, respectively. The products, which separated on cooling, were collected by filtration and recrystallized from ethanol to give compounds 2-6.

5-Amino-1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-1Hpyrazole-4-carbonitrile (**2**)

Yield (95%), M.p. 260–262°C; IR (KBr) v = 3407, 3200 (NH₂), 2209 (CN) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 2.77$ (s, 3H, CH₃), 2.95–3.00 (m, 4H, 2CH₂), 6.34 (d, J = 8 Hz, 1H, Ar-H), 6.91–7.31 (m, 5H, 3Ar-H and NH₂, D₂O exchangeable), 7.55 (s, 1H, C₃-H); ¹³C NMR (DMSO-d₆) $\delta = 25.32$ (CH₃), 25.67 (C-5'), 29.62 (C-6'), 115.29 (C-4), 118.72 (CN), 125.21, 126.83, 127.29, 128.12, 128.47 (Ar-C), 131.89 (C-3), 135.48 (C-6a'), 141.92 (C-11c'), 143.35 (C-11b'), 150.45 (C-7a'), 153.64 (C-5), 162.16 (C-9'), 172.10 (C-11'); MS, *m/z* (%): 358 (M⁺, 100). Anal. calcd for C₁₉H₁₄N₆S: C, 63.67; H, 3.94; N, 23.45; S, 8.95. Found: C, 63.62; H, 4.00; N, 23.39; S, 9.01.

5-Amino-1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-1Hpyrazole-3,4-dicarbonitrile (**3**)

Yield (65%), M.p. 166–168 °C; IR (KBr) v = 3407, 3200 (NH₂), 2219 (CN) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 2.80$ (s, 3H, CH₃), 3.00-3.15 (m, 4H, 2CH₂), 6.25 (d, J = 8 Hz, 1H, Ar-H), 6.55 (s, 2H, NH₂, D₂O exchangeable), 7.00-7.26 (m, 3H, Ar-H); ¹³C NMR (DMSO-d₆) $\delta = 25.14$ (CH₃), 25.23 (C-5'), 29.18 (C-6'), 110.68 (CN), 116.69 (CN), 124.34, 125.90, 127.25, 128.05 (Ar-C), 131.05 (C-3), 134.59 (C-6a'), 142.71 (C-11c'), 148.35 (C-11b'), 150.40 (C-7a'), 152.69 (C-5), 162.16 (C-9'), 173.10 (C-11'); MS, *m*/*z* (%): 383 (M⁺, 89). Anal. calcd for C₂₀H₁₃N₇S: C, 62.65; H, 3.42; N, 25.57; S, 8.36. Found: C, 62.60; H, 3.48; N, 25.54; S, 8.40.

5-Amino-1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-3-(methylsulfanyl)-1H-pyrazole-4-carbonitrile (**4**)

Yield (90%), M.p. 124–126 °C; IR (KBr) v = 3460, 3346 (NH₂), 2215 (CN) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 1.58$ (s, 3H, SCH₃), 2.80 (s, 3H, CH₃), 2.90-3.05 (m, 4H, 2CH₂), 6.50 (d, J = 8 Hz, 1H, Ar-H), 6.60 (s, 2H, NH₂, D₂O exchangeable), 7.00-7.25 (m, 3H, Ar-H); ¹³C NMR (DMSO-d₆) $\delta = 11.64$ (CH₃), 25.15 (CH₃), 25.34 (C-5'), 29.32 (C-6'), 113.04 (C-4), 115.59 (CN), 126.03, 126.16, 126.47, 127.13, 128.51 (Ar-C), 132.50 (C-3), 133.82 (C-6a'), 140.67 (C-11c'), 150.75 (C-11b'), 151.25 (C-7a'), 153.83 (C-5), 160.17 (C-9'), 172.31 (C-11'); MS, *m*/*z* (%): 404 (M⁺, 100). Anal. calcd for C₂₀H₁₆N₆S₂: C, 59.38; H, 3.99; N, 20.78; S, 15.85. Found: C, 59.29; H, 4.05; N, 20.71; S, 15.92.

Methyl 5-amino-1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-1H-pyrazole-4-carboxylate (**5**)

Yield (90%), M.p. 139–141 °C; IR (KBr) v = 3464, 3354 (NH₂), 1685 (C=O) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 1.34$ (t, J = 6.9 Hz, 3H, CH₃), 2.81 (s, 3H, CH₃), 2.96–3.07 (m, 4H, 2CH₂), 4.28 (q, J = 7.5 Hz, 2H, CH₂), 6.43 (d, J = 8 Hz, 1H, Ar-H), 6.81 (s, 2H, NH₂, D₂O exchangeable), 6.90-7.35 (m, 4H, 3Ar-H and C₃-H); ¹³C NMR (DMSO-d₆) δ 14.45 (CH₃), 25.17 (CH₃), 25.29 (C-5'), 29.32 (C-6'), 59.82 (OCH₂), 96.49 (C-4), 125.65, 126.03, 126.25, 127.33, 128.30 (Ar-C), 131.85 (C-3), 134.14 (C-6a'), 140.77 (C-11c'), 141.41 (C-11b'), 151.35 (C-7a'), 151.83 (C-5), 160.39 (C-9'), 164.31 (C-11'), 172.29 (C=O); MS, *m/z*

(%): 405 (M^{+} , 100). Anal. calcd for $C_{21}H_{19}N_5O_2S$: C, 62.20; H, 4.72; N, 17.27; S, 7.91. Found: C, 62.30; H, 4.65; N, 17.33; S, 7.81.

Ethyl 5-amino-1-(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-3- (*methylsulfanyl)-1*H-pyrazole-4-carboxylate (**6**)

Yield (90%), M.p. 250–252 °C; IR (KBr) v = 3460, 3346 (NH₂), 1685 (C=O) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 1.45$ (s, 3H, SCH₃), 2.77 (s, 3H, CH₃), 2.97–3.10 (m, 4H, 2CH₂), 3.83 (s, 3H, OCH₃), 6.51 (d, J = 8 Hz, 1H, Ar-H), 6.97–7.23 (m, 5H, 3Ar-H and NH₂, D₂O exchangeable); ¹³C NMR (DMSO-d₆) $\delta = 11.16$ (SCH₃), 25.12 (CH₃), 25.34 (C-5'), 29.37 (C-6'), 51.03 (OCH₃), 93.95 (C-4), 125.92, 126.27, 126.53, 126.85, 128.78 (Ar-C), 132.76 (C-3), 133.59 (C-6a'), 139.92 (C-11c'), 150.86 (C-11b'), 151.46 (C-7a'), 153.56 (C-5), 160.06 (C-9'), 164.52 (C-11'), 172.04 (C=O); MS, *m*/*z* (%): 437 (M⁺, 100). Anal. calcd for C₂₁H₁₉N₅O₂S₂: C, 57.65; H, 4.38; N, 16.01; S, 14.66. Found: C, 57.71; H, 4.33; N, 16.09; S, 14.60.

11-(3,5-Dimethyl-1H-pyrazol-1-yl)-9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidine (7)

To a solution of compound **1** (2.82 g, 1 mmol) in ethanol (20 ml), acetyl acetone (1 mmol) was added and the reaction mixture was refluxed for 10 h. The solvent was then removed under reduced pressure and the residue was recrystallized from ethanol to give compound **7**. Yield (70%), M.p. 129–131 °C; ¹H NMR (DMSO-d₆) δ = 2.05 (s, 3H, CH₃), 2.19 (s, 3H, CH₃), 2.86 (s, 3H, CH₃), 2.95–3.03 (m, 4H, 2CH₂), 5.91 (s, 1H, C₄-H), 6.10 (d, *J* = 8 Hz, 1H, Ar-H), 6.79-7.19 (m, 3H, Ar-H); ¹³C NMR (DMSO-d₆) δ = 11.97 (CH₃), 13.29 (CH₃), 25.04 (CH₃), 25.36 (C-5), 29.41 (C-6), 108.56 (C-4'), 119.21, 124.89, 126.19, 126.33, 127.15, 127.82 (Ar-C), 131.06 (C-3'), 134.49 (C-6a), 140.49 (C-11c), 141.58 (C-11b), 151.03 (C-7a), 161.85 (C-9), 172.04 (C-11); MS, *m/z* (%): 346 (M⁺, 78). Anal. calcd for C₂₀H₁₈N₄S: C, 69.34; H, 5.24; N, 16.17; S, 9.26. Found: C, 69.30; H, 5.29; N, 16.09; S, 9.33.

4,5,6,7-Tetrachloro-2-[(9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)amino]-1H-isoindole-1,3(2H)-dione (8)

A mixture of compound **1** (2.82 g, 1 mmol) and tetrachlorophthalic anhydride (2.85 g, 1 mmol) in glacial acetic acid (50 ml) was refluxed for 2 h. The formed precipitate was filtered on hot, dried and recrystallized from dioxane to give compound **8**. Yield (73%), M.p. 285–287 °C; IR (KBr) v = 3326 (NH) and 1791, 1724 (anhydride CO) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 2.35$ (s, 3H, CH₃), 2.83-3.28 (m, 4H, 2CH₂), 7.08-7.35 (m, 2H, Ar-H), 7.85 (d, J = 7.2 Hz, 1H, Ar-H), 8.37 (d, J = 7.6 Hz, 1H, Ar-H), 9.60 (s, 1H, NH, D₂O exchangeable); ¹³C NMR (DMSO-d₆) $\delta = 25.10$ (CH₃), 26.02 (C-5'), 30.13 (C-6'), 98.95 (C-2a), 125.90, 126.27, 126.71, 126.85, 128.78 (Ar-C), 132.35 (C-3), 134.82 (C-6a'), 139.92 (C-11c'), 150.68 (C-11b'), 154.46 (C-7a'), 155.50 (C-6a), 160.06 (C-9'), 162.52 (C-11'), 164.10 (C=O), 168.50 (C=O); MS, m/z (%): 552 (M⁺, Cl³⁷, 11.32), 550 (M⁺, Cl³⁵, 4.72). Anal. calcd for C₂₃H₁₂Cl₄N₄O₂S: C, 50.20; H, 2.20; Cl, 25.77; N, 5.82; S, 5.83. Found: C, 50.31; H, 2.12; Cl, 25.81; N, 5.78; S, 5.90.

2-(9-Methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)-6-phenyl-4,5-dihydropyridazin-3(2H)-one (9)

A mixture of compound **1** (2.82 g, 1 mmol) and β -benzoylpropionic acid (1.78 g, 1 mmol) in

ethanol (50 ml) was refluxed for 4 h. The formed precipitate was filtered off hot, dried and recrystallized from dioxane to give compound **9**. Yield (97%), M.p. 154–156 °C; IR (KBr) v = 1691 (C=O) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 2.43$ (m, 5H, CH₂ and CH₃), 2.81–2.95 (m, 4H, 2CH₂), 3.61 (t, J = 7.5 Hz, 2H, CH₂), 7.10–7.45 (m, 9H, Ar-H); ¹³C NMR (DMSO-d₆) $\delta = 17.74$ (CH₃), 23.87 (C-5'), 29.35 (C-6'), 39.92 (C-5), 56.64 (C-4), 125.30, 125.85, 126.53, 127.62, 128.45 (Ar-C), 131.24 (C-3), 134.66 (C-6a'), 139.58 (C-11c'), 150.21 (C-11b'), 154.40 (C-7a'), 160.25 (C-9'), 162.51 (C-11'), 173.80 (C=O); MS, *m/z* (%): 424 (M⁺, 100). Anal. calcd for C₂₅H₂₀N₄OS: C, 70.73; H, 4.75; N, 13.20; S, 7.55. Found: C, 70.66; H, 4.81; N, 13.12; S, 7.62.

D-Glucose (9-methyl-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-d]pyrimidin-11-yl)hydrazone (10)

A mixture of compound **1** (2.82 g, 1 mmol), D-glucose (1.80 g, 1 mmol), ethanol (30 ml), and a catalytic amount of glacial acetic acid (3 drops) was heated at 80 °C for 2 h. The formed precipitate was filtered off, dried and recrystallized from ethanol to give compound **10**. Yield (60%), M.p. 142–144 °C; IR (KBr) v = 3353-3220 (broad, OH+NH) cm⁻¹; ¹H NMR (DMSO-d₆) $\delta = 2.40$ (m, 3H, CH₃), 2.80–3.02 (m, 4H, 2CH₂), 3.20–3.60 (protons of the alditol congregated with the solvent absorption) [14], 3.70–3.80 (m, 2H, *CH*₂OH), 4.40-5.11 (m, 5H, 5OH, D₂O exchangeable), 7.05–7.42 (m, 5H, Ar-H and NH, D₂O exchangeable), 8.30 (s, 1H, N=CH); ¹³C-NMR (DMSO-d₆) $\delta = 18.82$ (CH₃), 25.60 (C-5), 29.63 (C-6), 61.26, 70.71, 72.78, 77.12, 92.65 (C-alditol), 125.91, 126.20, 126.75, 126.80, 128.74 (Ar-C), 134.62 (C-6a'), 139.91 (C-11c'), 150.52 (C-11b'), 153.42 (C-7a'), 160.06 (C-9'), 162.51 (C-11'), 162.90 (N=CH). Anal. calcd for C₂₁H₂₄N₄O₅S: C, 56.74; H, 5.44; N, 12.60; S, 7.21. Found: C, 56.84; H, 5.39; N, 12.69; S, 7.11.

(1S)-1,2,3,4,5-Penta-O-acetyl-1-C-(5-methyl-8,9-dihydronaphtho[1',2':4,5]thieno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)-D-arabinitol (11)

A solution of compounds **10** (1 mmol) in a mixture of acetic anhydride (10 mL) and anhydrous pyridine (10 ml) was stirred at room temperature for 24 h. The reaction mixture was poured into ice-water with stirring and the solids that precipitated were collected by filtration, washed with water, dried and recrystallized from ethanol to give compounds **11**. Yield (70%), M.p. 109–111 °C; IR (KBr) v = 1748 (OAc), 1602 (C=N); ¹H NMR (DMSO-d₆) $\delta = 1.80-2.10$ (m, 15H, 5OAc), 2.40 (m, 3H, CH₃), 2.80–3.10 (m, 6H, 3CH₂), 4.20–5.50 (m, 4H, 4CHOAc), 7.10–7.40 (m, 4H, Ar-H); ¹³C-NMR (DMSO-d₆) $\delta = 20.50$, 20.62, 20.66, 20.74, 20.88 (5CH₃), 22.70 (C-5'), 29.86 (C-6'), 61.41, 65.90, 67.08, 67.52, 70.52, 73.99 (C-alditol), 109.91 (C-4), 126.81, 127.24, 127.38, 127.53 (Ar-C), 135.13 (C-3), 136.62 (C-6a'), 142.12 (C-11c'), 149.52 (C-11b'), 153.40 (C-7a'), 160.10 (C-9'), 162.51 (C-11'), 169.56, 170.18, 170.59, 170.62, 170.83 (5CO). Anal. calcd for C₃₂H₃₄N₄O₁₀S: C, 57.65; H, 5.14; N, 8.40; S, 4.81. Found: C, 57.55; H, 5.20; N, 8.45; S, 4.80.

Antimicrobial activity

In vitro antimicrobial screening

The newly synthesized compounds were screened for their antibacterial activity against one gram positive bacteria, *Bacillus subtilis* NRRL B-543 and one gram negative bacteria, *Escherichia coli* NRRL B-210 and yeast *Candida albicans* NRRL Y-477. These microorganisms were obtained from Northern Utilisation Research and Development Division, U.S. Departement of Agricultural Peoria, Illinois, USA. Chloramphenicol and Fluconazole were purchased (pure form) from Egyptian market and used in a concentration of 2 mg/ml as references for antibacterial and antifungal activities. These compounds were assayed by the agar diffusion method [19]. The assay medium flasks containing 50 ml of nutrient agar medium for bacteria and Czapek's-Dox agar media for yeast. The holes each of 9 mm diameter were made by scooping out medium with a sterilized cork borer in a petri dish which was seeded with the organisms. The solutions of each test compound (0.10 ml) were added separately in the holes and Petri dishes were subsequently incubated. The incubation was carried out at 30°C for 24h. Simultaneously, controls were maintained by employing 0.10 ml of dimethylsulfoxide (DMSO) which did not reveal any inhibition and zones of inhibition produced by each compound was measured in mm. The results of antimicrobial studies are given in Table 1.

Minimal inhibitory concentration (MIC) measurement

The bacteriostatic activity of the active compounds (having inhibition zones (IZ) \ge 16 mm) was then evaluated using the two fold serial dilution technique [20]. Two fold serial dilutions of the test compounds and reference drugs solutions were prepared using the proper nutrient broth. The final concentration of the solutions varied between 500 and 7.81 µg/ml with the concentration of DMF not exceeding 2.5%. Each 0.10 ml from the tested compounds in DMF was mixed with 1 ml, 2 ml, and 3 ml sterilized distilled water and 0.10 ml from each diluted samples was added to test tubes. The tubes were then inoculated with the test organisms, grown in their suitable broth at 37 °C for 24 hours for bacteria and 48 hours for fungi (about 1×10^6 cells/ml), each 5 ml received 0.10 ml of the above inoculum and were incubated at 37 °C for 48 hours. The lowest concentration showing no growth was taken as the minimum inhibitory concentration (MIC) (Table 2).

Conclusions

The overall results indicated that, the tested compounds showed promising antimicrobial activity against bacteria and Fungi. Among the tested compounds, for gram-positive and gram-negative bacteria, it was noticed that ß-enaminonitriles of pyrazole ring system 2, 3 and 4 demonstrated inhibitory activities more than ß-enaminoesters 5 and 6. However, the pyridazine derivative 9 and non-acetylated sugar 10 revealed the most significant antibacterial activities. On the other hand, the non-acetylated sugar 9 and the acetylated cyclic *C*-nucleoside 10 revealed more effective activity against yeast than the other tested compounds.

Acknowledgement

We are grateful to Dr. Klaus Banert, Chemistry Department, Technical University Chemnitz, Germany for facilities and support.

Supporting Information

The scanned ¹H NMR spectra of compounds **2** and **4**, and the scanned ¹³C NMR spectra of compounds **2**, **4** and **7** are available in the online version (Format: PDF, Seize: ca. 0.3 MB): http://dx.doi.org/10.3797/scipharm.0910-11.

10

Authors' Statement

Competing Interests

The authors declare no conflict of interest.

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