

# Synthetic Nitroimidazoles: Biological Activities and Mutagenicity Relationships

Alka MITAL

Department of Pharmaceutical Technology, National Institute of Pharmaceutical Education and Research, Sector 67, S. A. S. Nagar-160062, Punjab, India

E-mail: [alkamital@gmail.com](mailto:alkamital@gmail.com)

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## Abstract

Parasitic and bacterial infections affecting the gastrointestinal tract represent a significant cause of morbidity and mortality worldwide. Nitroheterocyclic drugs have been available since the early 1960s for the treatment of anaerobic protozoa. The application of these drugs has widened and they are presently used to treat anaerobic pathogenic bacteria and protozoa. 5-nitroimidazoles are a well-established group of antiprotozoal and antibacterial agents that inhibit the growth of both anaerobic bacteria and certain anaerobic protozoa, such as *Trichomonas vaginalis*, *Entamoeba histolytica* and *Giardia lamblia*. The important antibacterial and antiprotozoal activities of nitroimidazoles are associated with reductive metabolism that has led to considerable interest in nitroimidazole reduction chemistry and the synthesis of new, highly effective drugs. The present review provides a brief account of various biological activities exhibited by synthetic nitroimidazole derivatives as well as their structure–mutagenicity relationships.

## Keywords

Antibacterial • Antifungal • Antimycobacterial • Trypanocidal • Anti-HIV activity • Antileishmanial agents

## Introduction

Nitroheterocyclic compounds have a wide variety of applications, ranging from food preservatives to antibiotics. Nitroimidazoles have therapeutic uses as anaerobic antibacterials and antiprotozoal agents. 5-nitroimidazoles are a well-established group of antiprotozoal and antibacterial agents [1]. The antimicrobial activity of these chemotherapeutic agents inhibits the growth of both anaerobic bacteria and certain anaerobic protozoa such as *Trichomonas vaginalis*, *Entamoeba histolytica* and *Giardia lamblia* [2]. They have other interesting biological activities of therapeutic potential such as, radiosensitizers in treatment of cancer [3–5], control of fertility [6], and antitubercular therapy [7, 8]. 5-nitroimidazole derivatives have been tested in cell-based assays and in enzyme assays against HIV-1 recombinant reverse transcriptase [9, 10]. 2-Nitroimidazoles play a major role as bioreductive markers for tumour hypoxia, as radiosensitizers [11–13], and some also demonstrate antiprotozoan activity [14]. Some dinitro and mono nitroimidazole derivatives have been predicted as notable radiosensitizers, antiprotozoal and antibacterial or antiepileptic agents [15].

## Antibacterial agents

The introduction of nitroheterocyclic drugs in the late 1950s and the 1960s heralded a new era in the treatment of infections caused by gram-negative and -positive bacteria and a range of pathogenic protozoan parasites. The antibiotic azomycin, a 2-nitroimidazole, isolated from a streptomycete, was the first active nitroimidazole to be discovered [16], which acted as the main impetus for the systematic search for drugs with activity against anaerobic protozoa. This led to the synthesis of the 5-nitroimidazole, metronidazole (1-hydroxyethyl-2-methyl-5-nitroimidazole, MTZ) and the demonstration of its activity against *T. vaginalis* [17]. Subsequently, MTZ was shown to cure giardiasis [18], amoebiasis [19], and *Balantidium* infections [20]. MTZ is most widely used in the treatment of anaerobic protozoan parasitic infections caused by *T. vaginalis*, *G. duodenalis*, and *E. histolytica* [21]. Metronidazole, tinidazole and ornidazole are the main synthetic agents of therapy for invasive amoebiasis (Figure 1) [22, 23]. The highly selective effect of these drugs is due to reduction of these drugs by nitroreductase enzymes resulting in the formation of highly reactive free radical species [24, 25]. Although there is no carcinogenicity or mutagenicity of metranidazole in human beings [26], it has been shown to be mutagenic in bacteria and carcinogenic in rodents [27–29]. Depending on the nature of substituents and the position of the nitro group, the nitroimidazole derivatives can show various pharmacological activities [30]. The compounds with nitro group at position 4 are usually less active than the corresponding 5-nitro derivatives.

Tinidazole and ornidazole have different side chains at the 1 position, do not differ markedly in their antimicrobial activity. Modifications at the 2 position, however, are known to interfere with both the activity and the microbial spectrum [31]. Compound GO 10213 [1-methylsulfonyl-3-(1-methyl-5-nitro-2-imidazolyl)-2-imidazolidinone], which has an imidazolidinone ring structure at the 2 position [32], exerts stronger antitrichomonal activity than MTZ (Figure 1) [33, 34]. GO 10213 is highly active against anaerobic bacteria, even more than metronidazole but less than niridazole. Thus, the modification of the 5-nitroimidazole at the 2 position increases not only its antitrichomonal activity but also its antibacterial activity [33, 34]. Although GO 10213 may be more active than MTZ, the imidazolidinone ring in this compound has been shown to be more carcinogenic than the nitro-

imidazole ring, making it less attractive therapeutically [35]. Various 5-(1-methyl-5-nitro-2-imidazolyl)-4*H*-1,2,4-triazoles were synthesized and tested *in vitro* for their antibacterial and antifungal activities. Few compounds exhibited significant effects against *Bacillus subtilis* at MIC ranges of 0.5–1 µg/ml and moderate effects against *Staph. aureus* [36].

The nitroimidazole compound **1**, 1-methyl-2-methylsulfonyl-4-nitroimidazole, is an anti-protozoal and bactericidal compound with the unique and surprising property of being totally non-mutagenic and thus of a much higher degree of safety than is found with other nitroimidazoles (Figure 1) [32, 37]. The invention of Miwa *et al.*, relates to the non-mutagenic 1,2-disubstituted-4-nitroimidazole compounds with structure **2**, useful as antiprotozoal agents (Figure 1) [38]. The compounds are 1,4-dimethyl-5-nitro-1*H*-imidazol-2-yl methylcarbamate, 1,4-dimethyl-2-(2-hydroxyethyl)-5-nitroimidazole, 1,2-dimethyl-4-(2-hydroxyethyl)-5-nitroimidazole, and pivaloyl esters. These novel compounds are not mutagenic in Ames strain TA100, and are highly potent against protozoal diseases. They also demonstrated that 1,2,4-substituted 5-nitroimidazoles are potent antiprotozoal and/or antibacterial agents with little or no mutagenic and drug residue level problems.

There was an attempt to differentiate among the structural elements contributing to mutagenicity and antitrichomonal activities as a basis for the rational design of safer nitroimidazoles [39]. Walsh *et al.*, presented two approaches for developing non-mutagenic 5-nitroimidazoles based on knowledge of the likely mechanisms by which reactive intermediates are metabolically formed [40]. These approaches are the incorporation of a substituent at the C-4 position of the imidazole ring and the addition of readily oxidizable functionalities (gallate ester derivatives) into the molecule. These structural features revealed that the addition of a 4-substituent significantly reduced or eliminated the mutagenicity, and always reduced the *in vitro* antitrichomonal activity.

Differently substituted 2-hydroxyaryl-(1-methyl-5-nitro-1*H*-imidazol-2-yl)methanols, **3, 4** with substituted phenolic rings in the ortho and/or para positions and containing bulky and strong lipophilic groups such as a *t*-butyl group in the carbocyclic ring have been reported in the patents [41, 42] for their antimicrobial activity and lack of mutagenicity (Fig. 1) [43].

The antimicrobial profile of a new nitroimidazole derivative EU 11100 (5-nitro-1-methylimidazol-2-yl-(2-hydroxy-3-*tert*-butylphenyl)carbinol) has been studied by Dubini *et al.*, (Figure 1) [44]. The *in vitro* activity has been evaluated against both aerobic and anaerobic bacteria, *T. vaginalis*, and mycetes, under suitable experimental conditions. The compound was compared with ampicillin against aerobic bacteria; with MTZ against anaerobic bacteria, lactobacilli and *T. vaginalis*; with nistatin and econazole against candida and with econazole and bifonazole against filamentous fungi. This derivative has been shown to be moderately active against some anaerobic bacteria belonging to both the Gram-positive and Gram-negative groups. Its inhibitory activity against *T. vaginalis* was similar to that of metronidazole [44].

A new series of substituted (*Z*)-2-arylidene-3(2*H*)-benzofuranones bearing 1-methyl-5-nitroimidazole **5** or 4-nitroimidazole **6** were synthesized and assayed for their antibacterial activity against Gram-positive and Gram-negative bacteria (Fig. 1) [45]. Most of the 5-nitroimidazole analogues showed a remarkable inhibition of a wide spectrum of Gram-positive bacteria (*Streptococcus aureus*, *S. epidermidis*, MRSA, and *Bacillus subtilis*) and Gram-negative *Klebsiella pneumoniae*, whereas 4-nitroimidazole analogues

exhibited no effect against selected bacteria. The quantitative structure–activity relationship investigation was applied to find a correlation between different physico-chemical parameters of the compounds studied and their biological activity. Their antibacterial activity, suggests that possibly negative charge of  $-\text{NO}_2$  group on C-5 of imidazole and partial charge on carbonyl oxygen in benzofuran are necessary for action. Moreover, there is a limited space available for this moiety of benzofuran system, thus the bulkier substituted analogues **7** ( $R = 5\text{-Br}, 5\text{-CH}_3$ ) were inactive (Figure 1). Changing the position of the substituent on the benzofuran appears to have little influence on the antibacterial activity. Khabnadideh *et al.*, synthesized a series of alkylimidazole **8** and investigated the relationship between the length of the alkyl chain and its antibacterial activity (Fig. 1) [46]. The compound **8** were investigated for antibacterial activity against *Staphylococcus aureus*, *Pseudomonas aeruginosa* and *Escherichia coli*. Several compounds showed significant *in vitro* activity against *E. coli*, *S. aureus* and *P. aeruginosa*. Antibacterial activity of 1-alkylimidazoles increases as the number of carbons in the alkyl chain rises to nine and then decreases. So 1-nonylimidazole **9** is the most effective compound (Figure 1), as it has the lowest MIC (minimum inhibitory concentration) and MBC (minimum bactericidal concentration) (MIC *S. aureus* = 10  $\mu\text{g/ml}$ ; MBC *S. aureus* = 19  $\mu\text{g/ml}$ ; MIC *P. aeruginosa* = 39  $\mu\text{g/ml}$ ; MBC *P. aeruginosa* = 78  $\mu\text{g/ml}$ ; MIC *E. coli* = 19  $\mu\text{g/ml}$ ; MBC *E. coli* = 39  $\mu\text{g/ml}$ ).

The antiparasitic activity of nitroimidazoles **10** bearing an arylsulfonylmethyl group was assessed against *T. vaginalis*, and the *in vitro* cytotoxicity was evaluated on human monocytes and the mutagenicity was determined by the Salmonella mutagenicity assay (Figure 1) [47]. Molecules, bearing an additional methyl group on the 2-position, showed a lower mutagenicity than metronidazole. All the tested molecules were found to be mutagenic in the Salmonella mutagenicity assay using the most sensitive strain TA100 [48, 49]. Moreover, three derivatives were characterized by a low mutagenicity and an efficient antitrichomonal activity. The methyl group at the 2-position and the arylsulfonylmethyl group at the 4-position of imidazole ring modulated the nitro reduction at the 5-position as in **10**. They have shown that the replacement of the methyl group at the 2-position by a lactam group could double the mutagenicity of metronidazole [50]. In this study, the methyl group at the 2-position combined with the arylsulfonylmethyl group at the 4-position lowered the  $\text{MP}_{\text{TA100}}$  of dimetridazole, metronidazole and secnidazole series [47]. A good correlation was demonstrated between the antiprotozoal activity and the mutagenicity for these molecules which reflects their ability to damage DNA through covalent binding and induction of DNA breaks [51]. They were characterized by a low mutagenicity and high antiparasitic activity. Although the mechanism of action explaining their biological activity remains to be elucidated, these newly synthesized compounds allow the disconnection between mutagenicity and biological activity for the first time. This approach offers the possibility of synthesizing new and potentially safer 5-nitroimidazoles.

A series of *N*-[5-(5-nitro-2-heteroaryl)-1,3,4-thiadiazol-2-yl]thiomorpholine derivatives were synthesized and evaluated for *in vitro* anti-*Helicobacter pylori* activity (Figure 1) [52]. Nitrofurans analog **11** containing thiomorpholine S,S-dioxide moiety was the most potent compound displaying very strong activity at 8 mg/disc (inhibition zone diameter >40 mm) against both MTZ-sensitive and resistant strains. Among nitrothiophenes, compound **12** showed moderate activity and compound **13** showed strong activity against *H. pylori* at three different concentrations (Figure 1).

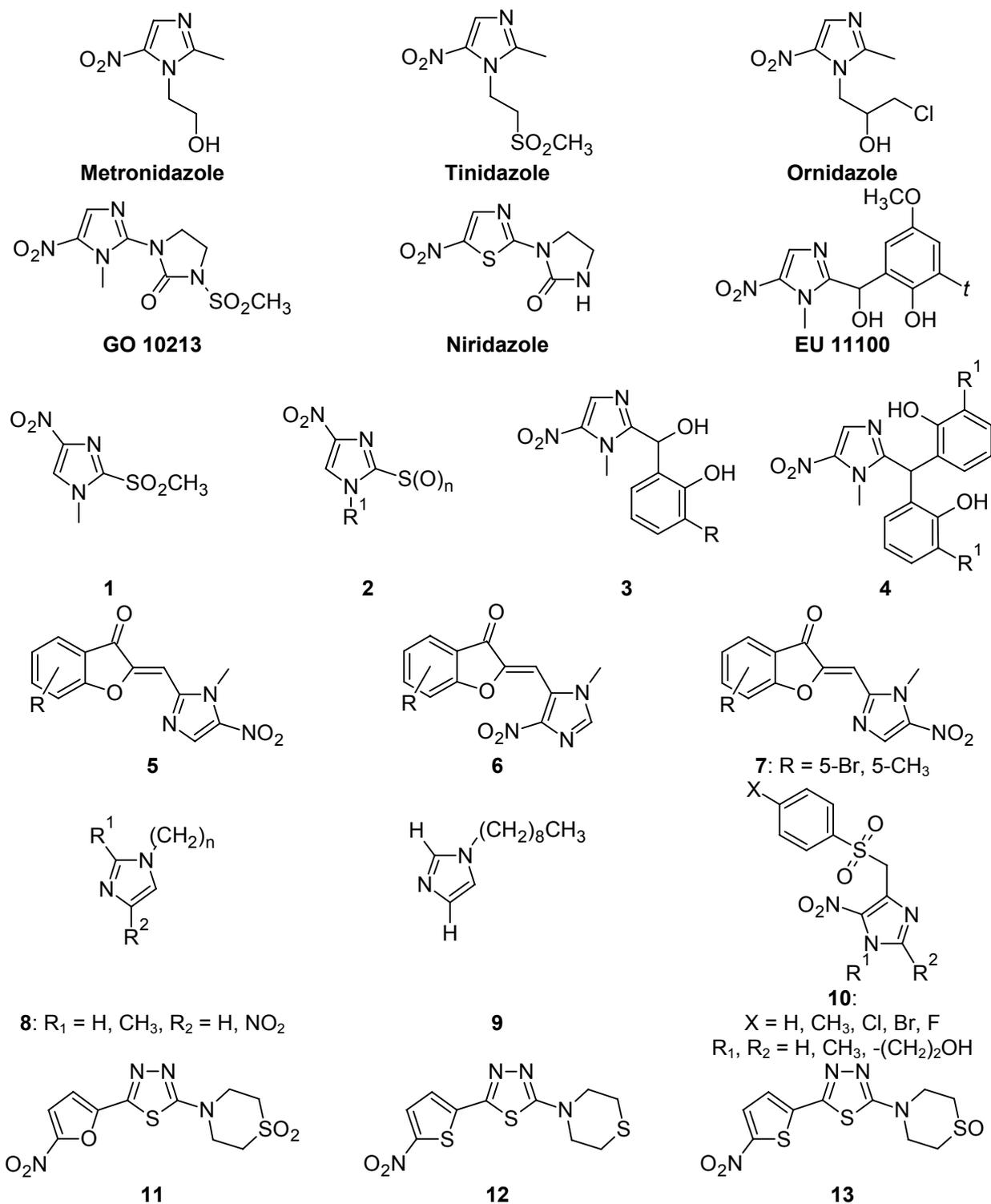


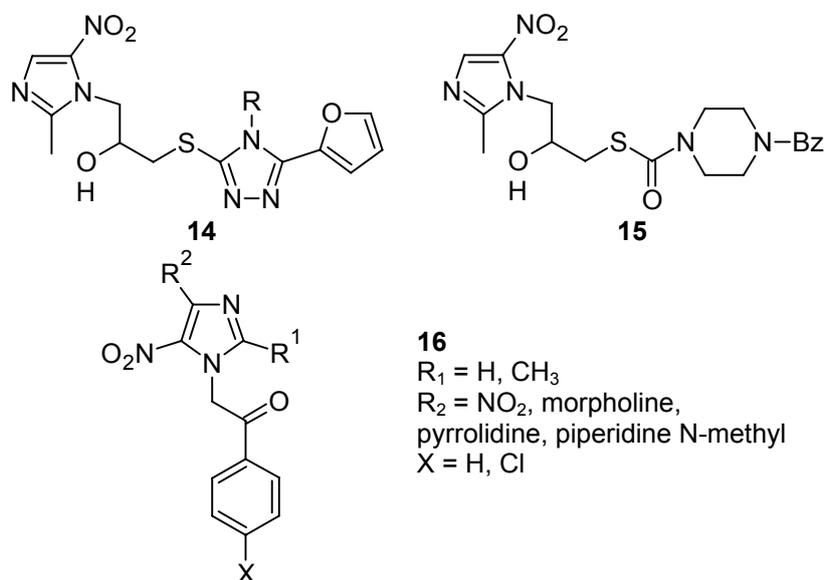
Fig. 1. Antibacterial agents

### Antifungal nitroimidazoles

The imidazoles are a class of antifungal azole derivatives having a broad spectrum of activities both *in vitro* and *in vivo*. Four 5-nitroimidazoles derivatives, satranidazole, S75 0400 A, flunidazole and ronidazole were tested and found to be more active than MTZ, the

drug commonly used to treat infections caused by *Blastocystis hominis* in humans [53]. Ketoconazole and iodoquinol have been reported to have therapeutic activity in infections caused by *B. hominis*, and were found to be less active than MTZ.

Some new nitroimidazole derivatives bearing a 1,2,4-triazolythioethyl, 1,2,4-triazolythiopropyl or *N,N*-disubstituted dithiocarbamoyl-propyl residue at N-1 were synthesized and evaluated for *in vitro* antibacterial and antifungal activity (Figure 2) [54]. Although not as active as the standard ampicillin, compound **14** were found to be active against *S. aureus* ATCC 6538 and/ or *S. epidermidis* ATCC 12228 where ornidazole was devoid of activity. The most active compound was **15** (antifungal activity against *Trichophyton tonsurans* NCPF 245; MIC = 3 µg/ml) exerting about one half the activity of ornidazole. A panel of antifungal and antihelminthic drugs was tested for activity against *Mycobacterium tuberculosis* (*M. tb*) *in vitro*. Antifungal drugs, miconazole, 2-nitroimidazole, clotrimazole, and the antihelminthic drug niclosamide were found to have significant antituberculosis activity, with MICs between 1 and 10 µg/ml [55]. Niclosamide and 2-nitroimidazole also had activity against stationary phase tubercle bacilli.



**Fig. 2.** Antifungal agents

Some derivatives have been synthesized by treating 4,5-dinitro- and 2-methyl-4,5-dinitroimidazoles with epoxypropane, epichlorohydrin or phenacyl bromide and tested for their antioxidant and antifungal properties against fungi species *Sclerophoma pityophila* [56]. The nitro group in *N*-substituted 4,5-dinitro- and 2-methyl-4,5-dinitroimidazoles has been replaced with primary and secondary amines to afford 4-amino-5-nitroimidazole derivatives. Nearly all of them have shown significant antioxidant activity in comparison with that of tocopherol, which is used as a reference substance. Amongst the most active derivatives **16**, bearing chlorine atom in 2-hydroxypropyl and 2-oxopropyl chains or phenacyl group at N-1 position of the imidazole ring (Figure 2), two compounds have very strong fungistatic activity against *S. pityophila*. High effectiveness was induced by the displacement of nitro group at 4-position on the imidazole ring to morpholine or piperidine and by the presence of a chlorine atom at 4-position on the phenyl ring.

## Antimycobacterial nitroimidazoles

One of the first reports to note the antimycobacterial activity of 2-nitroimidazoles compounds were a series of compounds synthesized with a variety of substituents at the 1- and 5-positions [57]. They were tested for activity against a panel of Gram-negative and Gram-positive organisms, as well as fungi. The majority of compounds with alkyl, amide, or alcohol substituents were found to be inactive against all organisms tested. There was improvement in activity, however, when the 5-substituent was replaced with a vinyl group. A four-fold increase in *M. tb* activity occurred when the N-1 substituent was changed from methyl to ethyl. Activity was the highest against *M. tb* when the N-1 substituent was ethyl and the 5-vinyl group was left unsubstituted. These vinyl substituted 2-nitroimidazoles were explored further in a subsequent report by the same group [58] and the best activity against this strain (2 µg/ml) was for 1-methyl-2-nitro-5-(2-nitrohex-1-enyl)-1*H*-imidazole, a compound with a vinyl group substituted with both an *n*-butyl group and a second nitro group. Several series of 2-nitro and 2-aminoimidazoles with hydrophilic substitutions at the 5-positions such as oximes and hydrazones were synthesized and tested for antibacterial activity [59–61]. Most of these compounds had moderate activity against *M. tb* (10–200 µg/ml), but they displayed enhanced activity against many other organisms. The most potent compound against *M. tb* (5 µg/ml) was bearing an *n*-decyl substituted oxime at the 5-position of the imidazole ring. The authors noted that lipophilic substituents at the 5-position increased activity against Gram positive organisms, including *M. tb*.

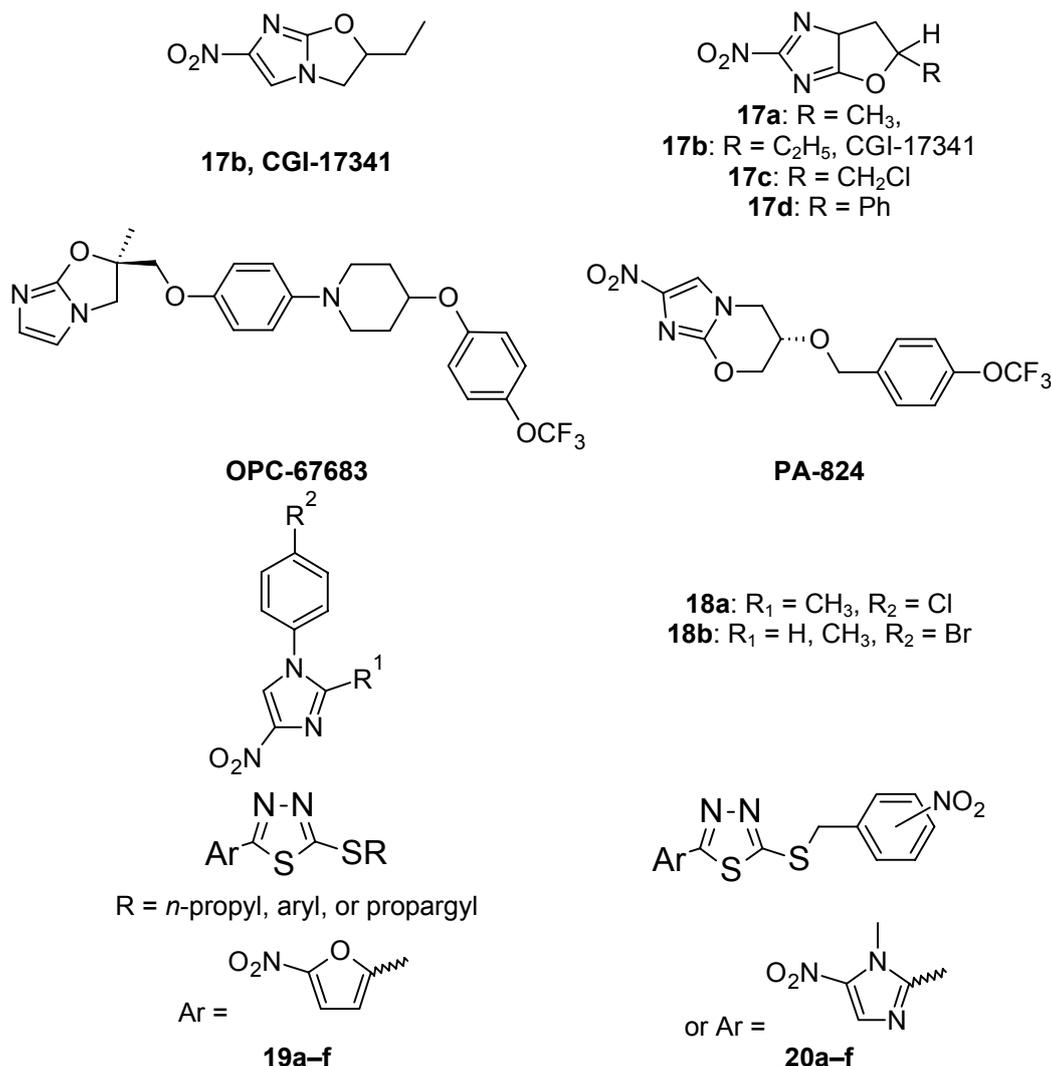
4- and 5-nitroimidazoles have been the subject of significant efforts to optimize anti-tubercular activity. One particular class of nitroimidazoles, containing an oxazole (or oxazine) ring structure fused to the imidazole, has shown particular promise. These compounds were originally discovered to create novel dinitroimidazoles that might have improved characteristics as radiosensitizers. Dinitroimidazoles, nitroimidazo[2,1-*b*]oxazoles were synthesized by reaction of the unsubstituted dinitro precursor with oxiranes, spontaneous cyclization, accompanied by loss of the 2-nitro group [62]. These compounds were later found to have potent antitubercular activity and analogued by chemists from Hindustan Ciba-Geigy, India for antitubercular optimization [7, 63]. Three discrete series of structural analogs were explored: 4-nitroimidazo [2,1-*b*]oxazoles, 5-nitroimidazo[2,1-*b*] oxazoles, and 4-nitroimidazo-1-ethanols. The uncyclized nitroimidazo-1-ethanols were not as active as the corresponding imidazo[2,1-*b*]oxazoles. The lead compound was **17a**, with a methyl group on the 2-position of the imidazo-oxazole ring, and a MIC of 1.95 µg/ml against *M. tb* H37Rv (Figure 3). Variations in the imidazo[2,1-*b*]oxazole series with a single substitution at the 2-position, increasing chain length slightly (to ethyl to give compound CGI 17341, **17b**) or adding a single halogen atom (**17c**) increased *in vitro* activity dramatically (30 and 16-fold, respectively). Replacement of the methyl group with a benzene ring (**17d**) increased activity only slightly (2-fold) while replacement with long, straight-chain alkyl groups decreased *in vitro* activity.

CGI 17341 (2-ethyl-5-nitro-2,3-dihydroimidazo[2,1-*b*][1,3]oxazole) is a novel lipophilic and orally active representative of the 5-nitroimidazole series of antimicrobial agents. This nitroimidazooxazole derivative is a promising novel antituberculosis compound with potent *in vitro* and *in vivo* activities [7, 63], but is not developed because of its mutagenic properties. CGI 17341 inhibited the drug-susceptible and multi-drug-resistant strains of *M. tb* and had no cross-resistance with isoniazid, rifampin, streptomycin, or ethambutol. While

its *in vitro* activity against *M. tb* was comparable to those of isoniazid and rifampin, it was superior to those of streptomycin, ciprofloxacin or norfloxacin, and oxazolidinone DuP 721. Bicyclic nitroimidazoles such as PA-824 and OPC-67683 are currently in clinical development as a promising new class of therapeutics for tuberculosis (Figure 3) [64]. CGI-17341, OPC-67683, both nitroimidazo-oxazoles, and PA-824, a nitroimidazo-oxazine, have activity against aerobic and anaerobic populations of *M. tb* [63, 65, 66]. PA-824 has many attractive characteristics as a TB therapy, most notably its novel mechanism of action, its activity *in vitro* against all tested drug-resistant clinical isolates, and its activity as both a potent bactericidal and a sterilizing agent in mice. PA-824 shows no evidence of mutagenicity in genotoxicity studies, no significant cytochrome P450 interactions, and is inactive against a broad range of Gram-positive and Gram-negative bacteria. OPC-67683 had both inhibitory activity on mycolic acid biosynthesis and potent *in vitro* activity against *M. tb*, as indicated by its low MIC range across many strains, including MDR-TB [65]. The IC<sub>50</sub> values of OPC-67683 for mycolic acid subclasses were lower than those of isoniazid (INH), and these results correlated well with the *in vitro* anti-tubercular activity of OPC-67683 and INH. The anti-tubercular activity of nitro-imidazooxazole derivatives correlated well with their inhibitory activity against mycolic acid biosynthesis [65]. Thus they concluded that the inhibitory activity of OPC-67683 against mycolic acid synthesis was a mechanism of action attributable to killing mycobacteria at least as potently as INH. PA-824 has remarkably reduced mutagenicity compared with its parent nitroimidazole compound, CGI-17341, and it has recently been reported that PA-824 is not mutagenic in the Ames test and does not seem to be metabolized by cytochrome P450 into potentially carcinogenic substances [67]. This suggested the possibility that its mutagenic activity will continue to be the major obstacle to its development as an anti-TB drug.

The anti-tuberculosis activity of various imidazole derivatives has been reported [68, 69]. Among 1-aryl-4-nitroimidazoles without nitro groups at 1-aryl substituent only 1-(4-chlorophenyl)-2-methyl-4-nitroimidazole **18a** demonstrated sufficient activity (99% of inhibition at 6.25 µg/ml), others derivatives were inactive though bromo derivatives **18b** were characterized by 76% and 85% inhibition, respectively (Figure 3) [70]. It was clear that electron-donating substituent on 1-nitrogen atom in C-nitroimidazoles diminishes inhibition activity of the compounds. Surprisingly, it also appears that 2-methyl-4-nitroimidazoles are more active than 4-nitroimidazoles. Tuberculosis inhibition activity and cytotoxicity results of these derivatives were not better or at least similar to the control drugs INH and rifampicin (RMP).

Two series of 2-(5-nitro-2-furyl)- and 2-(1-methyl-5-nitro-1*H*-imidazol-2-yl)-5-propyl, allyl and propargylthio-1,3,4-thiadiazoles (**19a-f**) and 2-(5-nitro-2-furyl)- and 2-(1-methyl-5-nitro-1*H*-imidazol-2-yl)-5-(nitrobenzyl)thio-1,3,4-thiadiazole derivatives (**20a-f**) were synthesized and evaluated against *M. tb* (Figure 3) [71]. Among the nitrobenzylthio derivatives (**20a-f**), all the ortho, meta and para nitrobenzyl isomers in the nitrofurans series exhibited good antituberculosis activity, while the corresponding nitroimidazole analogues were completely inactive (Inhibition=0%). The compounds that bear a primary alkylthio substitution, displayed good antituberculosis activities, in the following order: *n*-propyl > ethyl > methyl (MIC=1.56-6.25 µg/ml) [72]. Some nitroimidazole and nitrofurans derivatives have been also claimed to possess *in vitro* antibacterial, antifungal and antituberculosis activities [54].



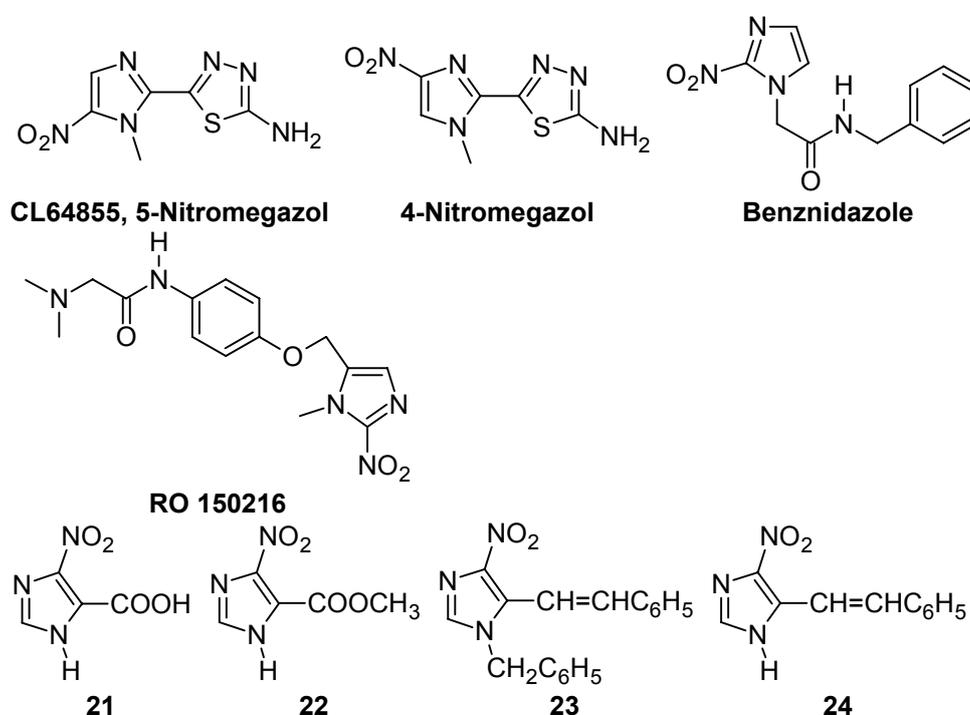
**Fig. 3.** Antimycobacterial agents

### Trypanocidal nitroimidazoles

The nitroimidazole-thiadiazole derivative CL64855 (2-amino-5-(1-methyl-5-nitro-2-imidazolyl)-1,3,4-thiadiazole, 5-nitromegazol) has a pronounced trypanocidal activity [73–75] which may be due to the triggering of radical production by the compound (Figure 4) [74, 76, 77]. Benznidazole was found mutagenic in *Salmonella* [78, 79]. The location of the nitro group in position 4 makes the nitro heterocyclic compounds (such as 4-nitromegazol) totally inactive against parasite (Figure 4) [77]. The treatment with 5-nitromegazol entails the production of reactive oxygen species in aerobic conditions [80], it can induce redox cycling which explains its toxic effects, through the production of superoxide radical anion and then hydroxyl radicals [81], that are highly damaging for cellular structures. Most nitroimidazoles are mutagenic in bacteria, and this mutagenicity has been attributed to nitroreductases present in these organisms. It has been argued that the lower capacity of mammalian cells to perform nitro reduction decreases the genotoxic risk of 5-nitromegazol. However, some studies [82, 83] have demonstrated its genotoxicity in mammalian cells. Because of its efficacy against several strains of *Trypanosoma cruzi* of diverse sensitivity,

it has a potential role in treatment of Chagas' disease. However, 5-nitromegazol has shown mutagenicity [84] according to the Ames assay. RO 150216, another nitroimidazole derivative, has been shown to be effective in inhibiting the culture growth of different strains of the African trypanosome [85], and *in vivo* using various animal models [80].

In an effort to synthesize imidazole analogs which retain biological but not toxic activities, a variety of new compounds with different substitutions on the bioisosteric imidazole and pyrazole structural frame were synthesized [48]. Structures bear substituents able to provide some variations in the lipophilic, steric and electronic properties of the molecules. In particular, substituents with increasing hydrophobic and steric properties, such as  $-\text{COOH}$ ,  $-\text{COOCH}_3$ ,  $-\text{CH}_3$ ,  $-\text{CH}_2\text{C}_6\text{H}_5$  and  $-\text{CH}=\text{CHC}_6\text{H}_5$  groups were considered. Compounds were tested on a set of tester strains of *Salmonella typhimurium*; in particular, classical nitroreductase- or *O*-acetyltransferase deficient derivatives have been incorporated into the assay in order to study the metabolism and mutagenicity of nitro compounds [86, 87].



**Fig. 4.** Trypanocidal agents

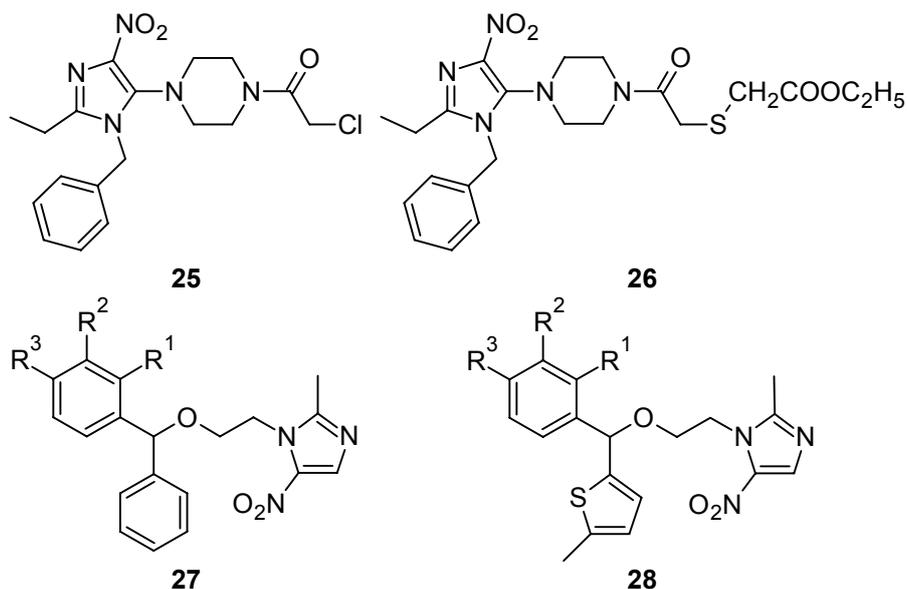
The structure–activity relationship study found that the following structural parameters are correlated with mutagenic potency: the presence of a methyl or a benzylic group on the imidazole ring, whereas the absence of a substituent in  $\text{N}_1$  or  $\text{N}_3$  leads to non-mutagenic compounds (derivative **21** and **22**) or to a marked decrease in mutagenicity (85% decrease with derivative **24** with respect to derivative **23**) (Figure 4). The presence of bulky substituents, such as benzylic or styryl groups, on the imidazole ring leads to increased mutagenic activity, which implies that such substituents do not sterically hinder  $\text{NO}_2$  metabolic activation, which would have caused inhibition of the nitroreduction or DNA binding steps critical to mutagenicity.

Some nitroimidazoles are reported as potent and selective histamine H-3 receptor agonists [88, 89], mitogen-activated protein (MAP) kinases inhibitors [90], nitric-oxide synthase inhibitors [91] and antibacterial agents [92]. Furthermore, 5-nitro-substituted haloimidazoles, showed important biological activity as potential radiosensitizers [93] and other imidazole derivatives having 5-alkylsulfanyl residues exhibited remarkable antitumor activity [94].

### Anti-HIV active nitroimidazoles

Non-nucleosides reverse transcriptase inhibitors (NNRTIs), a group of structurally diverse compounds, have been reported to directly inhibit the enzyme Reverse transcriptase (RT), which plays an essential and multifunctional role in the replication of HIV-1 and thus is considered to be an attractive target for inhibition of HIV-1 replication.

Al-Soud *et al.*, reported the synthesis of new 5-substituted piperazinyl-4-nitroimidazole derivatives and their evaluation for anti-HIV activity [95]. The target was the synthesis of new 4-nitroimidazoles, leading to inhibition of HIV by inhibition of RT and reduction of the drug-resistance strains [95]. Compounds **25** and **26** were found to be most active among the tested compounds inhibiting HIV replication in cell culture (Figure 5). The structure-activity relationships of 4-nitroimidazole derivatives have suggested the importance of a piperazine group on the imidazole ring substituted by aliphatic carbonyl groups such as  $-\text{COCH}_2\text{Cl}$ ,  $-\text{CO}(\text{CH}_2)_n\text{R}$  or sulphonamides for potent inhibitory activity against RT.



**Fig. 5.** Anti-HIV agents

1-[2-(Diarylmethoxy) ethyl]-2-methyl-5-nitroimidazoles (DAMNIs) is a novel family of HIV-1 non-nucleoside reverse transcriptase inhibitors (NNRTIs) active at submicromolar concentration [96, 97]. Replacement of one phenyl ring of 1-[2-(diphenylmethoxy) ethyl]-2-methyl-5-nitroimidazole **27** with heterocyclic rings, such as 2-thienyl or 3-pyridinyl, led to novel DAMNIs with increased activity (Figure 5). In HIV-1 WT cell-based assay the racemic 1-{2-[ $\alpha$ -(thiophen-2-yl)phenylmethoxy]ethyl}-2-methyl-5-nitroimidazole **28** ( $\text{EC}_{50}$  =

0.03  $\mu\text{M}$ ) proved 5 times more active than compound **27**. Docking experiments showed that the introduction of a chiral center would not affect the binding of both (*R*)-**28** and (*S*)-**28**. The internal scoring function of the Autodock program calculated the same inhibition constant ( $K_i = 7.9 \text{ nM}$ ) for the two enantiomers. Compounds **28** ( $\text{ID}_{50} = 8.25 \mu\text{M}$ ) were found more active than efavirenz ( $\text{ID}_{50} = 25 \mu\text{M}$ ) against the viral RT carrying the K103N mutation, suggesting for these compounds a potential use in efavirenz based anti-AIDS regimens.

### Antileishmanial agents

A series of 1-[5-(1-methyl-5-nitro-1*H*-imidazol-2-yl)-1,3,4-thiadiazol-2-yl]-4-arylpiperazines were synthesized and evaluated *in vitro* against *Leishmania major* [98]. The most active compound was 1-[(5-chloro-2-thienyl)...carbonyl]-4-[5-(1-methyl-5-nitro-1*H*-imidazol-2-yl)-1,3,4-thiadiazol-2-yl] piperazine **29** with an  $\text{IC}_{50}$  value of  $9.35 \pm 0.67 \text{ mM}$  against *L. major* promastigotes (Figure 6). In addition, this compound was effective against intracellular *L. major* and significantly decreased the infectivity index.

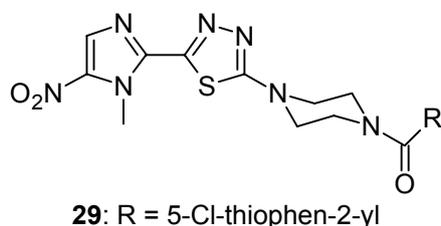


Fig. 6. Antileishmanial agent

### Nitroimidazoles as radiosensitizers

The 2-nitroimidazoles have been studied extensively for their use as radiosensitizers, hypoxic cytotoxins, and molecular markers of hypoxic regions in solid tumours [99-101]. Their selective hypoxic cytotoxicity and use as imaging agents for hypoxia are dependent on the bio-reduction of these compounds to reactive intermediates and the binding of these reductive species to intracellular macromolecules. Bio-reduction occurs only under extremely low oxygen tensions and, therefore, is selective for hypoxic regions [102]. In contrast, the use of 2-nitroimidazoles as radiosensitizers requires the intact compound to act as an oxygen mimic and potentiate the lethal effects of ionizing radiation in hypoxic but not aerobic cells [103].

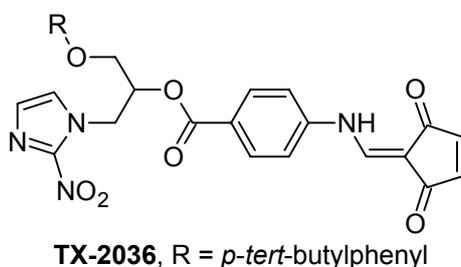


Fig. 7. Radiosensitizers

Chiral 2-nitroimidazole derivatives containing a 2-aminomethylene-4-cyclopentene-1,3-dione moiety were designed and synthesized as antiangiogenic hypoxic cell radiosensitizers [104]. All of these bifunctional derivatives proved to have activity as antiangiogenic hypoxic cell radiosensitizing agents and protein tyrosine kinase (PTK) inhibitory activities. TX-2036 was the most promising candidate for further development as an antiangiogenic hypoxic cell radiosensitizer (Figure 7).

2-Nitroimidazoles play a major role as bioreductive markers for tumour hypoxia and as radiosensitizers [11–13]. Several 2-nitroimidazoles with immunologically-identifiable side-chains have been described and conventional immuno-staining procedures can be used to locate their metabolites, bound to hypoxic cells in histological sections. Use of fluorescent immuno-reagents allows flow cytometric assessment of hypoxia and multiple colour fluorescent staining allows hypoxia to be correlated with other markers on a cell by cell basis. 2-nitroimidazole markers show considerable promise for clinical use in diagnosing hypoxia and allow rational application of hypoxia-related therapies [11].

### **Mechanism of action of antibacterial nitroimidazoles**

MTZ and other nitroimidazoles are only toxic once they become reduced, and bioreduction requires low redox potential electron-transfer systems. In general, the single-electron reduction potential of nitroimidazoles lies outside the normal range of mammalian redox systems. One-electron reduction of nitroimidazoles leads to an unstable nitro radical anion that can either decompose to a nitrite anion and an imidazole radical or be further reduced by accepting a second electron [105]. The constant reduction of the nitroimidazole creates a favourable concentration gradient resulting in continuous influx of the drug across the cell membrane. In the presence of oxygen, the electron from the nitro radical anion can be abstracted by oxygen thereby regenerating the parent compound and forming superoxide by a process known as futile cycling. Two-electron reduction of nitroimidazoles leads to somewhat more stable nitroso intermediates. These intermediates may be the biologically active form of nitroimidazoles.

Unfortunately, the nitroso compounds are also 1000-fold more toxic to mammalian cells and 1000-fold more mutagenic in bacterial systems than the parental nitro compounds, supporting the concept that mammalian cells are not susceptible to nitroimidazoles primarily because of the lack of ability to reduce such molecules [106]. The mechanism of the trichomonocidal activity of metronidazole and other 5-nitroimidazoles appears to depend on the ferredoxin-mediated reduction of their nitro group, with generation of a reactive metabolite or metabolites which interact with DNA leading to a subsequent inhibition of nucleic acid and protein synthesis. Redox cycling of these compounds under aerobic conditions appears to be a detoxification reaction by inhibiting net reduction of the drugs, thereby inhibiting their uptake. On the other hand, redox cycling of nitrofurans or other compounds with more positive reduction potential results in formation of high steady-state concentrations of oxygen-derived metabolites that might be of toxicological significance. It seems likely that reduced metabolites of nitroimidazoles (perhaps through covalent binding to tissue macromolecules and/or thiols depletion) are also involved in the nitroimidazoles' toxic effects to animal tissues and in their mutagenic and carcinogenic action.

## Toxicity, carcinogenicity and mutagenicity of nitroimidazoles

MTZ is relatively well tolerated in animals, with an LD<sub>50</sub> of 1 to 5g/kg in rats in acute studies and no signs of chronic toxicity problems with doses up to 150 mg/kg in rats for up to 80 weeks. Monkeys appear to be less sensitive and even at doses up to 225 mg/kg showed no adverse events [107, 108]. In humans, MTZ is well tolerated, a fact reflected in the widespread use in pregnant women (MTZ is one of the top ten drugs used during pregnancy worldwide) [109].

Nitroimidazoles, as a group suffer from the property of being mutagenic and carcinogenic in a variety of experimental models [110]. Regarding mutagenicity, metronidazole is one of the best-investigated compounds of the nitroimidazoles. The mutagenicity of MTZ and other 5-nitroimidazoles has been studied in facultative anaerobic bacteria, particularly *S. typhimurium* and *E. coli*. The drug MTZ is mutagenic on bacteria, especially if base-pair tester strains are used and bacterial nitroreductases are present. The serum levels attained in man after intake of this drug are sufficient to cause mutations in bacteria. Furthermore, interaction with and binding to DNA occurs under anaerobic conditions and sometimes DNA breaks are observed. However, MTZ does not show mutagenic activity in mammalian cells *in vitro*; the micronucleus test is negative and chromosome aberrations are only found under anaerobic conditions. Surprisingly, no data are available concerning the mutagenicity of 5-nitroimidazoles in the anaerobic bacteria for which these drugs are the most potent antibiotics.

The nitro group reduction plays a key role in the overall activity and for the expression of mutagenicity and drug residue formation of nitroimidazoles. The prime biological target for nitro compounds is DNA, giving rise to concern regarding their mutagenic and carcinogenic toxicity. The compounds are converted to a common hydroxylamine intermediate, which is converted to an electrophilic nitrogen species, which reacts with DNA [111]. It is presumed that the separation of protozoal and genotoxic activities is not feasible, both of these properties being mediated through a common metabolic intermediate [112]. The mechanism of aerobic toxicity of 2-nitroimidazoles is due to the intracellular reactive oxygen species generated by the futile cycling of the parent compound, and cell sensitivity is modulated by the ability of the cell to detoxify these species [113]. However, experimental data supporting this mechanism of aerobic toxicity for 2-nitroimidazoles are limited [100].

Virtually every nitroimidazole that has ever been tested carefully has mutagenic potential in the Ames type assay for detecting bacterial DNA mutation by reversion. There has been a very substantial database of such activities and there are considerable structure-activity relationships that have been developed [49, 110]. It is clear that mutagenic potential in the Ames assay varies over several orders of magnitude depending upon multiple factors including position of the nitro group, substitution pattern and the presence of electron-withdrawing or donating groups. From the bacterial mutagenicity studies of nitroimidazoles it is the need to identify and study the human metabolites of such compounds as potential mutagens [49]. Since mutagenicity is an undesirable property in clinically used drugs and raises the question of their potential carcinogenicity, further research is needed in order to evaluate nitroimidazole effect on DNA, contributing to the elucidation of mechanisms involved in these processes. A nitro derivative possessing good pharmacological activity with no mutagenicity would be of great interest not only from a safety point of view, but

would also provide a basis for further investigations of the mode of action and mechanism of expression of mutagenicity.

## Author's Statement

### Competing Interests

The author declares no conflict of interest.

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