The antimycobacterial derivatives against potential pathogenic strains: 2-Hydroxy-3-(4-phenylpiperazin-1-yl)-propylphenylcarbamates

Karel Waisser^{*1}, Rafael Doležal¹, Jozef Čižmárik², Ivan Malík², Jarmila Kaustová³

¹Department of Inorganic and Organic Chemistry, Charles University in Prague, Faculty of Pharmacy in Hradec Králové, Heyrovského 1203, CZ-500 05 Hradec Králové, Czech Republic

²Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Comenius University, Odbojárov 10, SK 832 32 Bratislava. Slovak Republic,

³Reference Laboratory for Mycobacterium knasasii, Regional Institute of Public Health in Ostrava, Partyzánské nám. 7, CZ 702 00 Ostrava, Czech Republic

Abstract

According to our previous study 29 derivatives of 2-hydroxy-3-(4phenylpiperazin-1-yl)-propylphenylcarbamates were tested for *in vitro* antimycobacterial activity against potential pathogenic strains *Mycobacterium kansasii* and *Mycobacterium avium*. The variations in group of compounds were by the substitution on phenyl rings. The Free-Wilson method was used to evaluate structure-antimycobacterial activity relationships. The advantage of compounds under study is in the activity against *M. kansasii*.

Keywords

Carbamates, propylphenylcarbamates, mycobacterium, potential pathogenic strains

Introduction

The return of tuberculosis to Europe and North America is one of the features of the period dating from 1985 In the developing countries, due to insufficient medical care, hygienic standards and compliance of the population with the treatment, a number of mycobacterial strains became resistant to modern chemical drugs. Due to the contemporary migration of population, infection was often transferred to Europe and North America. In addition, this unfavorable state is also being influenced by an increase in AIDS, which is often accompanied by mycobacterial diseasescaused by potential pathogenic strains. New mycobacterial diseases have occurred which were until recently considered intransferable to humans (mycobacterioses produced bv potentially pathogenic strains). Mycobacterial diseases due to multiresistant strains of the complex Mycobacterium avium and Mycobacterium intracellulare are not frequent, but mostly fatal in the end. We have recently studied the derivatives of alkoxyphenylcarbamic acids [1-5]. The advantage of the derivatives of phenylcarbanic acids is the low toxicity. Goal of this study is determining of derivatives of 2-hydroxy-3-(4-phenylpiperazin-1-yl)propyl-phenylcarbamates against M. kansasii and M.avium to complete the results of previous study.

Results and Discussion

The values of antimycobacterial activity of derivatives of 2-hydroxy-3-(4phenylpiperazin-1-yl)-propylphenylcarbamates are shown in Table 1. For the sake of comparison, we also included the values of MICs of the standard isoniazide (INH). The results revealed that the compounds exhibited in vitro activity against all tested mycobacterial strains. The values of MICs are generally within the range 8-1000 μ mol/l. The compounds possessed a better activity against *M. kansasii* 235/80 and *M. avium* 330/88 than isoniazide. The influence of structural moiety on activity is analyzed by Free-Wilson approach (see Table 2). We did not study the influence on antimycobacterial activity of substitution on phenyl ring of 4phenylpiperazine in the series of phenylcarbamic acid derivatives, yet. It seems that the substitution R^1 in position 3 by trifluoromethyl increases the antimycobacterial activity. Trifluoromethyl is the strong electonaceptor and is more lipophilic than methyl or fluorine. The best substitution R^2 is by 4-propoxy- and 4-butoxy- group. The compounds of our study form the new promising group of antimycobacterials against *M.kansasii*. The activity against *M. avium* is not significant.

Experimental

Chemistry:

All compounds were prepared by coworkers of Čižmárik. Synthesis of compounds substituted by fluorine (R^1) was published [6]. Preparation of other compounds will be printed in other journals of chemistry. The structure of compounds is illustrated in Fig. 1.

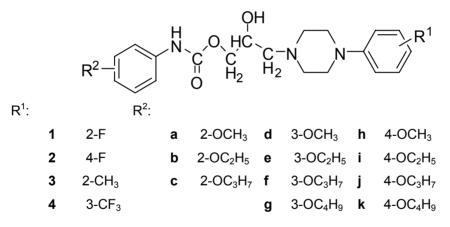


Fig. 1 Structure of derivatives of 2-hydroxy-3-(4-phenylpiperazin-1-yl)-propylphenylcarbamates

Microbiology:

For the evaluation of the antimycobacterial activity of the substances *in vitro*, the following strains were used: *Mycobacterium kansasii* CNCTC My 235/ 80, *Mycobacterium avium* CNCTC My 330/ 88, obtained from the Czech National

Collection of Type Cultures (CNCTC), National Institute of Public Health, Prague, and a clinical isolate of *Mycobacterium kansasii* 6 509/ 96. The antimycobacterial activities of the compounds was determined in the Šula semisynthetic medium (SEVAC, Prague). The compounds were added to the medium in Me₂SO at concentrations of 125, 64, 32, 16, 8, and 4 μ mol/l. The minimum inhibitory concentrations (MIC, the lowest concentration of a substance, at which the inhibition of the growth occurred) were determined after incubation at 37 °C for 14 and 21 days. The results are summarized in Table 1.

 Tab. 1:
 Minimum inhibitory concentration (µmol/l) of derivatives of phenylcarbamic acid

Compounds			MIC(µmol/l)				
				Incubation time 14 d/21 d			
	R^1	R ²	M. kansasii	M. avium	M. kansasii		
			My 235/ 80	My 330/ 88	6 509/ 96		
1a	2-F	2-0CH ₃	250/500	500/1000	250/500		
1b	2-F	2-OC ₂ H ₅	250/500	n/1000	250/500		
1c	2-F	2-OC ₃ H ₇	250/250	250/500	125/125		
1d	2-F	3-OCH ₃	125/250	250/500	125/250		
1e	2-F	3-OC ₂ H ₅	125/125	500/500	125/125		
2a	4-F	2-0CH ₃	250/250	500/1000	500/500		
2b	4-F	2-OC ₂ H ₅	250/n	250/n	125/n		
2c	4-F	2-OC ₃ H ₇	62.5/n	125/n	32/62.5		
2d	4-F	3-OCH ₃	62.5/62.5	250/500	250/n		
2e	4-F	3-OC ₂ H ₅	n/n	n/1000	n/n		
2f	4-F	3-OC ₃ H ₇	62.5/125	125/n	125/125		
2g	4-F	3-OC ₄ H ₉	250/500	500/1000	500/500		
2h	4-F	4-OCH ₃	125/250	500/n	n/500		
2j	4-F	4-OC ₃ H ₇	32/62.6	n/n	62.5/n		

2k	4-F	4-OC ₄ H ₉	n/n	n/n	125/125
3d	2-CH ₃	3-OCH ₃	125/125	250/250	125/125
3e	2-CH ₃	3-OC ₂ H ₅	125/125	125/125	125/250
i	2-CH ₃	4-OCH ₃	125/250	250/250	125/250
Зј	2-CH ₃	4-OC ₂ H ₅	250/250	250/250	125/125
4a	3-CF ₃	2-OCH ₃	125/n	n/n	125/125
4b	3-CF ₃	2-OC ₂ H ₅	125/125	125/n	62.5/62.5
4d	3-CF ₃	3-OCH ₃	32/62.5	n/n	32/62.5
4e	3-CF ₃	3-OC ₂ H ₅	32/32	n/n	32/32
4f	3-CF ₃	3-OC ₃ H ₇	16/32	n/n	32/32
4h	3-CF ₃	4-OCH ₃	16/32	32/62.5	16/32
4i	3-CF ₃	4-OC ₂ H ₅	32/62.5	32/62.5	32/62.5
4j	3-CF ₃	4-OC ₃ H ₇	32/n	62.5/n	32/n
4k	3-CF ₃	4-OC ₄ H ₉	32/62.5	n/n	32/32
INH		>250/>250	>250/>250	4/8	

n: impossible to determine (low solubility of compounds or low growth of mycobacteria).

Calculation

For the calculations, the Multireg programme (Klemera) working under Microsoft Excel was employed. The Free-Wilson approach was used for QSAR analysis. Activity contribution and statistical significant of correlations are summarized in Table 2.

	$\Delta \log$ MIC (µmol/l) for incubation time 14 d / 21 d		
	M. kansasii	M. kansasii	
	My 235/80	6 509/ 96	
R ¹ : 2-F	0.241/0.292	0.206/0.328	
4-F	0.052/0.039	0.176/0.261	
2-CH ₃	0.340/0.324	0.192/0.154	
3-CF ₃	-0.331/-0.428	-0.364/-0.413	
R ² : 2-OCH ₃	0.374/0.225	0.430/0.181	
2-OC ₂ H ₅	0.374/0.308	0.130/0.132	
2-OC ₃ H ₇	0.016/-0.052	-0.205/-0.089	
3-OCH₃	-0.139//-0.117	-0.007/-0.084	
3-OC ₂ H ₅	-0.122/-0.323	-0.096/-0.284	
3-OC ₃ H ₇	-0.299/-0.166	-0.072/-0.285	
3-OC ₄ H ₉	0.410/0.508	0.450/0.415	
4-OCH ₃	-0.159/-0.039	-0.259/0.038	
4-OC ₂ H ₅	0.007/0.141	-0.109/0.069	
4-OC ₃ H ₇	-0.299/-0.392	-0.220/-	
4-OC ₄ H ₉	-0.108/-0.132	-0.100/0.053	
μο	1.938/2.160	1.964/2.160	
r	0.944/0.964	0.958/0.953	
S	0.190/0.180	0.169/0.188	
F	7.60/8.12	9.51/7.12	
n	26/22	25/21	

 Tab. 2:
 Activity contribution of Free-Wilson analyzes and statistical significant of correlations.

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