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Oxide-Electrolyte Thickness Dependence Diode-Like Threshold Switching and High on/off Ratio Characteristics by Using Al₂O₃ Based CBRAM

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Abstract: Diode-like threshold switching and high on/off ratio characteristics by using an Al/Ag/Al₂O₃/TiN conductive bridge resistive random access memories (CBRAM) have been obtained. The 5 nm-thick Al₂O₃ device shows superior memory parameters such as low forming voltage and higher switching uniformity as compared to the 20 nm-thick switching layer, owing to higher electric field across the material. Capacitance-voltage (CV) characteristics are observed for the Ag/Al₂O₃/TiN devices, suggesting the unipolar/bipolar resistive switching phenomena. Negative capacitance (NC) at low frequency proves inductive behavior of the CBRAM devices due to Ag ion migration into the Al₂O₃ oxide-electrolyte. Thicker Al₂O₃ film shows diode-like threshold switching behavior with long consecutive 10,000 cycles. It has been found that a thinner Al₂O₃ device has a larger on/off ratio of >10⁸ as compared to a thicker one. Program/erase (P/E) cycles, read endurance, and data retention of the thinner Al₂O₃ oxide-electrolyte shows superior phenomena than the thicker electrolyte. The switching mechanism is also explored.

Keywords: Al₂O₃ oxide-electrolyte; resistive switching; CBRAM; diode; threshold switching; high on/off ratio

1. Introduction

The challenges faced by conventional flash memories can be overcome by introducing alternative non-volatile memories (NVM) [1]. Lots of emerging NVM devices, such as phase changeable memory (PCM) [2], magnetic random access memory (MRAM) [3], ferroelectric RAM (FeRAM) [4], resistive random access memory (RRAM) [5], conductive bridging random access memory (CBRAM) [6], and so on, are investigated since several decades to fulfill the criteria of nanoscale memory technology. These alternative NVMs have several exciting features, such as (i) physical size reduction, (ii) good endurance characteristics, (iii) longer retention, and (iv) low power consumption [7]. CBRAM is one type of promising candidate in which the operation mechanism also relies on the resistance switching effect. Cu, Ag or Co are used as active electrode material for the ion migration, while Pt, W, or TiN could be used as inert electrode [8,9]. Under an external bias on active electrode, metallic filament can be formed or dissolute inside solid-electrolyte films through redox reaction [7,8].



due to its high compatibility with the fabrication of conventional CMOS devices. Woo et al. [10] have reported the CBRAM devices using a Cu-Se/Al₂O₃/Pt structure. A thin layer (~3 nm) of Al₂O₃ film as the switching material or lanthanide metals as the interfacial layer to improve the device performance has been reported. However, the device size was large $(30 \times 30 \ \mu m^2)$. On the other hand, Kim et al. have studied the effect of two dimensional electron gas (2DEG) formation on the device performance at the interface of Al₂O₃/TiO₂ bilayer hetero-structure [11]. Sleiman et al. have investigated the Cu/AlO_x/W CBRAM devices where the thickness of AlO_x layer is 20 nm and a device size is very large $(1 \times 1 \text{ mm}^2)$ [12]. Belmonte et al. have described the restive switching characteristics using bilayer Al₂O₃ as switching material on Cu/TiN/Al₂O₃/W structure [13]. In addition, sneak path leakage is one of the major concerns in the high-density memory array. Recently, threshold switching memristor devices attracts great attention to overcome this serious issue [14–17].

In this work, a thickness dependent CBRAM characteristics of a single Al₂O₃ layer using the Ag/Al₂O₃/TiN structure has been reported. Thinner Al₂O₃ oxide-electrolyte shows superior resistive switching characteristics as compares to the thicker one. This is mainly due to the higher electric field-controlled Ag filament formation/dissolution process. Thicker Al₂O₃ film shows diode-like threshold switching behavior with long consecutive 10,000 cycles under operation current of 100 nA. Frequency dependent C-V characteristics of the devices are obtained and bipolar/unipolar switching has been explained. The Ag filament formation/dissolution under different operation currents have been explained schematically. From C-V measurement, it reveals that this CBRAM device can be used as a nano-battery, non-volatile memory, or logic gates under different operation currents.

2. Materials and Methods

2.1. Device Fabrication Process

A TiN as a bottom electrode (BE) was deposited and a patterned wafer with various via-hole sizes was fabricated. First of all, a high- κ Al₂O₃ thin film with two different thicknesses of 20 nm and 5 nm were deposited by RF sputtering. The Al₂O₃ target with a purity of 99.9% was used. During the Al₂O₃ film deposition, the chamber pressure, Ar flow rate, RF power, and deposition pressure were 1.2×10^{-5} torr, 25 sccm, 80 watt, and 20 mTorr, respectively. Then, silver (Ag) as an active electrode and aluminum as a capping layer were deposited in subsequent steps by thermal evaporation. The thicknesses of Ag and Al were 40 nm and 160 nm, respectively. Here, TiN was acting as a bottom electrode and Ag was acting as a top electrode (TE). It is true that Al is exposed to air and the surface will have 2-3 nm Al₂O₃. During probe by W on Al₂O₃/Al, the thin Al₂O₃ will be removed easily. Then, the W contact will be the underlying Al. In this study, we have investigated two types of devices of similar structure with different thicknesses of the Al_2O_3 layers as 20 nm (S1) and 5 nm (S2). Lift-off process was performed to obtain final device. Figure 1 shows the cross-sectional high-resolution transmission electron microscopy (HRTEM) image of the S1 device. The presence of individual layers is exhibited clearly in the image. The measured thickness of the amorphous Al_2O_3 switching material (SM) is approximately 18.1 nm.



Figure 1. HRTEM (high-resolution transmission electron microscopy) image of the TE/Al₂O₃/BE structure.

2.2. Measurement Procedure

Current-voltage (I-V) and program/erase (P/E) endurance characteristics were measured by using Agilent 4156C and a B1500 semiconductor parameter analyzer. Using this I-V system, a long current range from 1 fA to 100 mA could be measured. However, our probing station has minimal limited current is 100 fA. Capacitance-voltage (C-V) characteristics have been measured by using HP4284A LCR meter with frequency ranges from 20 Hz to 1 MHz. During measurement, the TiN BE was grounded and the sweeping bias was applied on Ag/Al electrode.

3. Results

3.1. Diode-Like Threshold Switching Characteristics

The Al/Ag/Al₂O₃/TiN based S1 device has shown diode-like threshold switching characteristics, as shown in Figure 2. Figure 2a shows consecutive 1000 DC cycles for the via-hole pristine device with a typical area of 4 μ m × 4 μ m at the low current compliance (CC) of 100 nA following the voltage sweeping path 1→4. Although there is a slight fluctuation of current for 1000 cycles, still an on/off ratio of 10 is being maintained in between path 1 and path 2, as shown in Figure 2b. The device shows high rectifying ratio (RR = I_F/I_R) (I_F = forward bias current = 22.35 nA at V_{read} = 1.5 V in path 1; I_R = reverse bias current = 0.88 pA at V_{read} = −1 V in path 4) of >2.5 × 10⁴. To evaluate the current conduction mechanism, a typical I-V in Figure 2a has been replotted in log-log scale and fitted linearly. It is observed that the conduction mechanism at low field under both forward (path 1) and reverse (path 4) biases has been dominated by Schottky conduction. This is confirmed from the E^{1/2} vs. In (J/T²) plot, as shown in the Figure 2c. The Schottky barrier height (φ_B) has been calculated using the Equation (1) below [18],

$$\varphi_B = \frac{kT}{q} [\ln(A^*) - I] \tag{1}$$

where *k* is the Boltzmann's constant $(1.38 \times 10^{-23} \text{ J/K})$, *T* is the absolute temperature (300 K), *q* is the electronic charge $(1.6 \times 10^{-19} \text{ C})$, *A*^{*} is the Richardson's constant (120 A/cm²-K²), and *I* is intercept. Schottky emission dominates at electric field range of 50 kV cm⁻¹ to 450 kV cm⁻¹. The obtained intercepts for the forward and reverse biases are -16.02 and -30.09, respectively. The barrier heights from the Equation (1) by using the intercept values are 0.7 eV and 1.11 eV for forward and reverse biases, respectively. However, obtained φ_{B} value is lower than the theoretical value of approximately 1.67 eV.

10⁻





(b

10

Figure 2. (a) Consecutive 1000 switching cycles of the S1 devices at low CC of 100 nA. (b) Current vs. no. of cycles plot of path 1 and path 2 at V_{read} of 0.6 V. (c) Schottky fitting for path 1 and path 4 at low field region in the forward and reverse biases. (d) RepeaTable 10,000 I-V cycles of the S1 device.

Bai et al. obtained the Φ_{SB} value of 0.58 eV at the Pt/AlO_{\delta} interface using Pt/AlO_{\delta}/Ta₂O_{5-x}/TaO_y/Pt structure [19], which is lower than the theoretical φ_B value of 2.49 eV. Yoon et al. have reported a similar φ_B value discrepancy at the Pt/Ta₂O₅ interface (0.7 vs. 2.3 eV) using Pt/Ta₂O₅/HfO₂/TiN structure [20]. They have mentioned that the barrier height lowering is due to the Fermi level pinning near the conduction band edge. According to our previous report [5], the lower φ_B value of 0.81 eV at the Pt/Al₂O₃ interface than the theoretical value of 2.57 eV using Pt/Al₂O₃/MoS₂/TaO_x/TiN structure is also obtained. This discrepancy is acceptable because of not considering the Fermi level pinning and image charge effects. The switching cycles can be maintained also up to 10,000 cycles, as shown in Figure 2d. However, it can be observed that electrical fatigue during 10,000 cycles led to creation of a non-filamentary path. This in turn has resulted in narrower window as the device is switched to long repeatable cycles. A few Ag⁺ ions were migrated during switching cycles and some of the Ag atoms were being accumulated inside the Al₂O₃ SM. Those accumulated Ag atoms were narrowing the switching window. It is expected that this switching could be improved by increasing the negative bias with number of cycles. However, further study is needed in the future. Due to this Ag⁺ ions migration, the threshold switching is observed, which is explained later.

3.2. Switching Characteristics, Uniformity and Current Conduction

The resistive switching characteristics of the S1 and S2 devices with a typical size of $0.4 \ \mu m \times 0.4 \ \mu m$ have been investigated. Figure 3a represents the typical forming cycle of pristine S1 and S2 devices. The S1 device required larger forming voltage in comparison to the S2 device due to the presence of thicker Al₂O₃. (V_{form}: 3.6 V vs. 1.75 V). After forming step, the devices exhibited bipolar resistive switching characteristics owing to a large amount of Ag⁺ atoms migrating into the switching layer for a transition to non-volatile memory from diode-like threshold switching. Figure 3b shows the bipolar resistive switching characteristics of the S1 and S2 devices following the voltage sweeping path $1\rightarrow4$. The SET voltages (V_{set}) of the S1 and S2 devices are found to be 1.5 V and 0.8 V, respectively. The RESET voltages (V_{reset}) are the same -0.4 V for both the S1 and S2 devices. It is interesting to note that the memory window (ratio of low resistance state (LRS) to high resistance state (HRS)) of the S1 and S2 devices at a read voltage of 0.2 V are approximately 10 and 10^8 , respectively. This is owing to high electric field in S2 devices and remaining Ag in the S1 devices, which is explained below.



Figure 3. (a) Typical forming cycle of the pristine S1 and S2 devices and (b) I-V switching characteristics of the S1 and S2 devices under CC of 5 mA. Weibull distribution plots of the S1 and S2 devices to study the (c) device-to-device forming voltage and (d) SET voltage variations by using linear fitting.

Figure 3c shows the Weibull distribution plot of forming voltage for the S1 and S2 devices. We measured 20 randomly chosen devices for every structure with a yield of >85%. In comparison to the S2 devices, the S1 devices require higher forming voltage owing to thicker switching material layer (20 nm vs. 5 nm). Due to the higher thickness of S1 devices, it requires larger electric field to form the initial filament. In the case of S2 devices, increment of electric field across the decreasing thickness is responsible for lower forming voltage, which may also owe to have higher leakage current in the S2 devices. The S1 devices show size dependence forming voltage, while the S2 devices do not show the size dependence forming voltage, which is also due to higher leakage as well as Ag ions are easily migrated through thinner oxide-electrolyte. Here an interesting point to be noted that the S1 device shows better uniformity of forming voltage with compare to the S2 devices. According to our previous report [21], the reason behind this betterment can be attributed to the presence of higher defect density inside the thicker switching layer. The obtained slope values from fitting curve of the Weibull distribution plot following the Equations (2) and (3) support the result [22],

$$F(Q) = 1 - e^{-\left(\frac{Q}{a_{63\%}}\right)^{\beta}}$$
(2)

$$W(Q) = \ln[-\ln(1-F)]$$
 (3)

where F(Q) is the cumulative distribution function of the failure, Q is the values of measured data, α is the scale factor value from distribution plot at approximately F = 63%, and β is the slope value or shape factor of the fitted curve in Weibull distribution plot. Higher slope values indicate the better uniformity. The obtained slope values are 4.20, 1.73, and 1.60 for the S1, S2 (0.6 µm × 0.6 µm) and S2 (4 µm × 4 µm), respectively. Weibull distribution of the SET voltages for all devices are shown in Figure 3d. In case of larger and smaller sizes, both the S2 devices show good V_{Set} uniformity as compared to the S1 devices. It may possible that the filament length of the S1 devices is longer and filament dissolution length is unstable. However, larger size S2 devices exhibit significantly lower V_{Set} values with compare to smaller size S2 devices due to more number of defects, which might help in easier filament formation/dissolution. From the fitting curve of the distribution plot, the obtained slope values are 1.25, 2.17 and 6.34 for the S1, S2 (0.6 µm × 0.6 µm) and S2 (4 µm × 4 µm) devices. Though the SET voltage uniformity of the smaller size S2 devices has been improved, but still we are lagging from our previously reported result [22]. This uniformity can be further improved by controlling the defects through the annealing process and controlling the Ag⁺ ion migration by interfacial engineering.

Further, we have explored the current conduction mechanism for the S2 devices. The current transport at HRS is dominating by trap-controlled space-charge-limited-current conduction (TC-SCLC) whereas the current transport at LRS shows Ohmic, as shown in Figure 4. The slope values for current transport fitting curve (ln (I) vs. ln (V)) for HRS are found as $sl1 \sim 1.12$ (I α V^{1.1}), $sl2 \sim 2.04$ (I α V²), and $sl3 \sim 8.3$ (I α V⁸) from low to high voltage regions, as shown in Figure 4a. Previously, the slope values for HRS are reported as 1.1, 1.3, and 8.5 by Rubi et al. [23] and 1, 2, 4, and 6 by Shang et al. [24]. In case of path 1 (Figure 2a), the slope values are 0.58 and 13.15 at voltage range of 0 to 0.8 V and 1 V to 1.2 V, respectively. This is not the SCLC mechanism. On the other hand, the slope value for ln (I) vs. ln (V) plot for LRS is $sl4\sim1.12$ (Figure 4b), which refers to Ohmic behavior. Slightly higher slope value of 1.12 with respect to 1 is attributed to the presence of some defects into the Ag conducting filaments. The switching mechanism is explained later.



Figure 4. I-V fitting characteristics (**a**) on HRS current are shown. This shows SCLC (space-charge-limited-current) conduction through the dissolution gap of the filaments. (**b**) Ohmic nature obtained from the fitting of current-voltage characteristics at LRS current.

3.3. C-V Characteristics and Understanding of Switching

In accordance with the resistive switching, capacitance-voltage (C-V) characteristics of the S2 devices have also been studied, as shown in Figure 5. Typically, 25 mV AC was superimposed on applied DC bias during measurement. Figure 5a–c concurrently exposes the bipolar as well as unipolar capacitive switching nature of the $4 \times 4 \mu m^2$ size devices. Under the positive bias sweep at a frequency of 1MHz, the concern device switches (in anticlockwise direction following the sweeping path 1–3)

from low capacitance state (LCS) to high capacitance state (HCS) at V_{set} of 4.2 V, as shown in Figure 5a, which is owing to Ag ions migration into the Al₂O₃ oxide-electrolyte. Next, we have applied the large negative bias from 0 V \rightarrow -6 V following the sweeping path 4 \rightarrow 7 under a frequency of 900 Hz, as shown in Figure 5b. For RESET observation, slightly lower frequency is applied owing to larger stress time. The device first shows a RESET at V_{reset} of -2.85 V. It is known that there is no option to limit capacitance or compliance current (CC) by using HP4284A LCR meter. Therefore, huge amount of Ag ions are migrated through Al₂O₃ oxide-electrolyte under positive sweep. It is also not easy to dissolute the Ag filament. In this way, we have observed a faint RESET phenomena at -2.85 V as well as the bipolar switching is observed. By applying more negative bias, it switches to LCS at V_{set} of -5.05 V, which is owing to regrowth of Ag filament. It is interesting to note that the C-V characteristic observed during -6 V \rightarrow 0 V bias is quite similar to the p-type metal-oxide-semiconductor (MOS) capacitor. Similarly, charge depletion for p-type MOS capacitor also occurs here when we reduce the negative bias. Consequently, as a result of oxide capacitance and depletion layer capacitance being connected in series, measured capacitance decreases with reducing negative bias.



Figure 5. C-V characteristics by applying (**a**) positive sweep voltage of 4 V and (**b**) negative sweep voltage of -6 V for the S2 devices. The device size is $4 \times 4 \mu m^2$. (**c**) Interpretation of unipolar switching by C-V characteristics. (**d**) Frequency dependent C-V response of the S1 (20 nm) and S2 (5 nm) devices with sizes of $0.4 \times 0.4 \mu m^2$ and $4 \times 4 \mu m^2$.

On the other hand, a device swept back to LCS from HCS at the V_{reset} of 3.1 V and manifests unipolar switch, as shown in Figure 5c, indicated by arrows $4\rightarrow 6$. The applied frequency is 900 kHz. By applying negative sweep, the device will be bipolar otherwise it will be unipolar if positive sweep is applied. The Ag ions will be dissolved from the Al₂O₃ film to Ag electrode. Here, significance of LCS and HCS are corresponding to HRS and LRS of the resistive switching, respectively. In Figure 5a, measured capacitance values for LCS and HCS are found to be 2.6 pF and 1.6 nF, respectively at a V_{read} of 4 V. Thus, HCS/LCS ratio is high (approximately 600). These results strongly demand that there is a possibility of Ag ion diffusion/trapping through the defects into the Al₂O₃ oxide-electrolyte. So that diffusion capacitance increases with applied positive bias after the collapse of depletion region. Kamel et al. have reported that metal atoms diffused along oxygen vacancy path in Ag/HfO₂/Pt stack and as a result C-V switching occurred [25]. In our previous report, the Cu^{z+} (z = 1, 2) diffusion in the defects of TaO_x film in a Cu/Ti/TaO_x/W CBRAM device [26]. If we calculate the geometrical capacitance (C₁) of this S2 devices as a parallel plate capacitor according to the following formula:

$$C_1 = \frac{\varepsilon_0 kA}{d} \tag{4}$$

where ε_0 (= 8.854 × 10⁻¹²) is free-space permittivity, k (= 9.1) is dielectric constant of Al₂O₃ film, 'A' is the device area (= 16 µm²), and d (= 5 nm) is the thickness of Al₂O₃ layer, then the capacitance value of via-hole region is 2.6 × 10⁻⁴ nF. The overlapping area between top and bottom electrode is 36 × 36 µm². Owing to SiO₂ with a thickness of approximately 150 nm, the capacitance is approximately 3 × 10⁻⁴ nF. Total capacitance (C_1) including overlapping and via-hole regions is 5.6 × 10⁻⁴ nF. But our measured HCS value (1.6 nF) is approximately 2.9 × 10³ times the geometrical calculated value, C_1 . This may be Ag ions near Ag electrode or Ag doped Al₂O₃ oxide-electrolyte, which help to higher k value of Al₂O₃. In Figure 2a at path 4, the left shift of minimal current is observed. The current of 400–500 fA is obtained at zero bias, which is higher than 100 fA at approximately –0.7 V. This non-zero current at zero bias indicates that this stack has capacity to hold the charges, which may be useful for future nano-battery applications [27].

Figure 5d describes the variation of HCS or ON-state capacitance with different frequencies (f) having ranges from 1 kHz to 1 MHz for both 0.4×0.4 and $4 \times 4 \ \mu m^2$ sizes of the S1 and S2 devices. Most interesting thing is that irrespective of variation HCS values of all devices are negative in the frequency zone from 1 kHz to 400 kHz. At 400 kHz, the HCS value (-0.67 nF) is almost the same. After that those stacks gain +Ve value of 0.2 nF at 500 kHz and eventually reach to 1.6 nF at 1 MHz with slow expansion. Negative (-Ve) capacitance has been reported by several groups for different structures [25,28,29]. Here, negative capacitance (NC) signifies the inductive nature of the devices. Kamel et al. have also mentioned the inductive nature of Ag/HfO₂/Pt device [25]. Misawa et al. have reported also negative capacitance for different structures [30]. Comparatively diffusion of charges is slow process through Al₂O₃ and corresponding current can't follow the change of voltages spontaneously. That is why current starts to lag behind the voltage. As a result, phase shift of current with respect to voltage takes place. In lower frequency region (1 kHz to 400 kHz), this phase shift becomes too large and capacitance falls down to negative value. Significantly conductivity increases by leaps and bounds. Thus, the device acts like an inductor in this lower frequency zone. In a true sense, negative capacitance is not like so called static/geometrical capacitance that can store charge. This is only due to phase shift of current in confined low frequency range. This negative capacitance mode can be used to amplify voltage for low power nanoscale devices. Still, it is not totally clear, and further study is needed. In conclusion, this Ag/Al₂O₃/TiN structure shows very promising for future charge storage applications.

3.4. Read Endurance and Data Retention Characteristics

Read endurance characteristics of the S1 and S2 devices are shown in Figure 6a,b, respectively. After forming, the HRS value of the S1 devices is lower than the pristine one $(5 \times 10^3 \Omega \text{ vs.} 5 \times 10^{11} \Omega)$ owing to remaining Ag into the Al₂O₃ oxide-electrolyte. It was found that the on/off ratio of the S1 device is only ~6 which is much lower than the S2 devices (>10⁸) at read voltage of 0.1 V using 1µs pulses for 1000 cycles. LRS state of S1 is constant throughout 1000 cycles while HRS state is slightly decreased. This is because as the complete filament is not ruptured, the effective filament height in consecutive cycles for a longer period is slightly decreased, causing lowering of the HRS value after a long period of cycles. However, in the S2 device, both HRS and LRS are the same throughout the measurement. This indicates that the occurrence of complete filament rupture is in the S2 devices.





Figure 6. Read endurance characteristic of the (**a**) S1 and (**b**) S2 devices at read voltage of 0.1 V using 1 µs pulse width. Data retention characteristics of the (**c**) S1 and (**d**) S2 at CC of 5 mA using read voltage of 0.1 V.

Typical data retention characteristics of the resistive memory device S1 and S2 at room temperature are shown in Figure 6c,d, respectively at CC of 5 mA where read voltage is 0.1 V. Both S1 and S2 devices show a 50-min data retention. It is clear that the S2 devices possess higher on/off ratio (>10⁸) than S1 (>5) similar as read endurance. This indicates that the S2 devices can retain more data than S1 for a particular period of time and favorable for multilevel switching, which can be studied in future. It is also observed that HRS state for the S1 devices is not uniform and it has a tendency to decrease but the S2 device has uniform LRS and HRS.

3.5. P/E Endurance Characteristics

Figure 7 shows P/E endurance characteristics for the S2 devices under different P/E endurance conditions. P/E endurance is useful as it describes the device reliability under AC pulse, which is also useful for practical implication. The S2 device could undergo few P/E endurance cycles with a huge on/off ratio of $>10^6$ with the P/E current of 1 mA/5 mA and 500 µs pulse width as shown in Figure 7a. The extremely high on/off ratio is promising for multilevel switching [31,32] and future neuromorphic computing to ensure the enhance conduction states. Though the device is unable to go for long P/E endurance keeping such a high order on/off ratio, still this result is comparable with recently published works as listed in Table 1 [10,11,33–38]. By reducing pulse width from 500 µs to 1 µs, P/E cycles are improved to 100 P/E cycles, as shown in Figure 7b. However, the P/E endurance cycles number has been increased with decreasing of pulse width, but the on/off ratio has been compromised. This is due to the lowering of stress effect by reducing the pulse width.



Figure 7. Typical P/E endurance characteristic of the S2 devices using (a) 500 µs and (b) 1 µs pulse width.

Table 1. Device performance comparison based on high on/off ratio with recently published results in literature.

Device Structure	P/E Voltage (V)	P/E Current (mA)	P/E Pulse Width (μs)	Cycle No.	On/Off Ratio	Retention (s)
Al/Ag/Al ₂ O ₃ /TiN (This work)	1.5/-1.2	1/5	500/500	10	>10 ⁶	3×10^{3}
Cu-Se/Nd/Al ₂ O ₃ /Pt [10]	1.5/-2	0.1/0.1	500/500	10,000	10^{5}	10^4 at 85 $^\circ\mathrm{C}$
Cu/Ti/Al ₂ O ₃ /TiO ₂ [11]	7/-9	0.1/10	5/5	10^{7}	10^{5}	10^{6}
Cu/TaO _x /Ta ₂ O _{5-x} /Pt [33]	-/-	0.1/0.1	-	3000	$\sim 10^{5}$	$>10^4$ at 85 $^\circ C$
Cu/HfO2:Cu/Pt [34]	-/-	1/1	0.01/100	>100	10^{7}	10^{5}
Al/CH ₃ NH ₃ Pbl ₃ :PVA m.Hl/ITO/Glass [35]	3/-1	100/100	$10^4/10^4$	500	>10 ⁵	10^{4}
Cu/Ti/PVP-PMF/Pt [36]	1/-0.5	1/5	$10^4/10^4$	$>10^{3}$	$>10^{3}$	>10 ³ at 85 °C
Ag/HfO _x :N/Pt [37]	2/-	0.1/-	50/50	10^{6}	5×10^8	-
Cu/NG/HfO2 [38]	4/-4	0.5/0.5	0.5/0.5	10^{7}	>10 ⁶	2×10^5 at 125 $^\circ \mathrm{C}$

3.6. CBRAM Mechanism

We have proposed a switching mechanism of Ag/Al₂O₃/TiN device, as shown in Figure 8. In previous reports using different structures [39–41], the metal ions are migrated under external bias and metallic filament is formed. At a CC of >1 mA, the Ag filament formation dissolution has been described. When a positive bias is applied on Ag TE, Ag atoms at Ag/Al₂O₃ oxidize into Ag ions $(Ag^{\circ} \rightarrow Ag^{+} + e^{-})$. Large number of Ag ions start to diffuse into Al₂O₃ layer. Due to high electric field at TiN BE, these mobile Ag ions migrate towards TiN BE. These Ag ions at TiN BE are deoxidize back to Ag atoms, which results in a formation of conical shaped Ag conducting filament (CF) in between Ag TE and TiN BE. Due to this CF, the device goes to LRS state (Figure 8a). When a negative voltage is applied on Ag TE, the CF start to dissolve at narrow region near TE owing to higher electric field and memory device shifts to HRS (Figure 8b). A neck of a filament can be observed near to the TE. After formation of the pristine devices, a reduction and oxidation at the neck region will be responsible for LRS and HRS of the devices. For the S2 devices, the high electric filed controls across the Al_2O_3 oxide-electrolyte and high on/off ratio is obtained owing to no remaining Ag atoms into the electrolyte. The process continues for each cycle of applied positive and negative voltage biases. At a low CC of 100 nA for the S1 devices, an Ag nanocrystal-type filament is formed under positive bias on the TE (Figure 8c). By applying negative bias on the TE, almost all Ag atoms will be returned towards the TE and the device will be treated as a pristine one (Figure 8d). This phenomena are similar to the threshold switching because of Ag ions migration through Al_2O_3 electrolyte with the same polarity [14–17]. It is possibility that Ag ions could be gathered at the Ag/Al₂O₃ interface. Therefore, a non-zero current of 400–500 fA is obtained at zero bias in path 4 (Figure 2a). Due to this small amount of Ag atoms, the filament can't be retained for longer time.



Figure 8. Schematic views of conducting filament growth and dissolution mechanism at different current compliances: (a) set at a CC of 1 mA, (b) reset after set at 1 mA, (c) set at a CC of 100 nA, and (d) reset after set at 100 nA.

4. Conclusions

In conclusion, diode-like threshold and resistive switching characteristics with current conduction mechanism have been explored using an Al/Ag/Al₂O₃/TiN structure. Before forming, the thicker Al₂O₃ layer-based device exhibits diode-like highly uniform threshold switching of >10⁴ cycles with high rectifying ratio of >10⁴, at a low CC of 100 nA. After forming, the 5 nm-thick Al₂O₃ device shows better uniformity and P/E cycle endurance with high on/off ratio. Negative capacitance at low frequency proves inductive nature and supports unipolar/bipolar resistive switching. Owing to high on/off ratio, the device is capable of multilevel resistive switching and can be potentially used for high-density data storage in near future.

Author Contributions: A.S. wrote the first manuscript and explained data under instruction of S.M. S.R. helped to measure endurance data and modified the manuscript. Y.-F.L. measured resistive switching data under instruction of M.D. and S.M. M.D. measured also the resistive switching characteristics and analyzed data under instruction of S.M. All authors have read and agreed to the published version of the manuscript.

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Conflicts of Interest: The authors hereby declare that they don't have any interest of competition.

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