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Helium Metastable Distributions and Their Effect on the Uniformity of Hydrogenated Amorphous Silicon Depositions in He/SiH₄ Capacitively Coupled Plasmas

Sanghyun Jo¹, Suik Kang¹, Kyungjun Lee^{1,*} and Ho Jun Kim^{2,*}

- ¹ Department of Mechanical Engineering, Gachon University, Seongnam 13120, Korea
- ² Department of Mechanical Engineering, Hanyang University, 55 Hanyangdaehak-ro, Sangnok-gu, Ansan 15588, Korea
- * Correspondence: leek3@gachon.ac.kr (K.L.); hojunkim9158@hanyang.ac.kr (H.J.K.)

Abstract: This study investigates, numerically, the spatial distribution of metastable helium (He*) in He/SiH₄ capacitively coupled plasma (CCP) for the purpose of optimizing plasma density distributions. As a first step, we presented the results of a two-dimensional fluid model of He discharges, followed by those of He/SiH₄ discharges to deposit hydrogenated amorphous silicon films, to investigate which factor dominates the coating uniformity. We retained our CCPs in the 300 mm wafer reactor used by the semiconductor industry in the recent past. Selected parameters, such as a sidewall gap (radial distance between the electrode edge and the sidewall), electrical condition of the sidewall, and position of the powered electrode, were considered. In addition, by increasing the gas pressure while varying the sidewall condition, we observed modification of the plasma distributions and, thus, the deposition rate profiles. According to the results, the shift in He* distributions was mainly due to the reduction in the electron mean free path under conditions of gas pressure higher than 100 Pa, as well as local perturbations in the ambipolar electric field due to the finite electrode structure. Small additions of SiH₄ largely changed the He* density profile in the midplane of the discharge due to He* quenching. Furthermore, we found that the wide sidewall gap did not improve deposition uniformity against the expectation. This was because the excitation and ionization rate profiles were enhanced and localized only near the bottom electrode edge.

Keywords: plasma-enhanced chemical vapor deposition; capacitively coupled plasmas; metastable density; process uniformity; hydrogenated amorphous silicon

1. Introduction

The rise of technologies, such as big data, the Internet of Things, and smartphones, is generating enormous amounts of data [1]. In the 4th Industrial Revolution, for example, high-end storage devices, such as recent flash memory chips, have become essential [2]. As a result, thin-film deposition technology has demonstrated substantial progress with regard to its attributes for the development of state-of-the-art memory chips. Moreover, plasma deposition technologies, such as plasma-enhanced chemical vapor deposition (PECVD) and plasma-enhanced atomic layer deposition (PEALD), offer the possibility of uniformly growing these films at high deposition rates [3–5].

In any case, plasma deposition involves an extremely complex process that is not easy to analyze, largely because the results of this process might be sensitively influenced by a large number of parameters [6,7]. Despite the physical and chemical complexity of plasma deposition not being fully resolved, the size of semiconductor devices continues to shrink, necessitating an adherence to specific requirements, including an extremely high degree of uniformity [8,9]. Additionally, industrial applications require the ability to repeat processes and the reactor size should be small to reduce cost [10]. Thus, the use of plasma-enhanced



Citation: Jo, S.; Kang, S.; Lee, K.; Kim, H.J. Helium Metastable Distributions and Their Effect on the Uniformity of Hydrogenated Amorphous Silicon Depositions in He/SiH₄ Capacitively Coupled Plasmas. *Coatings* **2022**, *12*, 1342. https://doi.org/10.3390/ coatings12091342

Academic Editors: Cheng Zhang and Qi Hua Fan

Received: 29 July 2022 Accepted: 7 September 2022 Published: 15 September 2022

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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). chemical vapor deposition (PECVD) using capacitively coupled plasma (CCP) is often chosen for coating thin films at a satisfactory level of uniformity.

In microelectronics fabrication, CCP PECVD with noble-gas-based mixtures (such as He/SiH_4 , Ar/SiH_4 , and Xe/SiH_4) has been used for thin-film deposition [11]. Even though the metastable species of the noble gases dominate the ionization processes, experiments in these devices have shown highly non-uniform distributions of excited atomic and molecular states. For example, McMillin and Zachariah presented results from their experiments in the Gaseous Electronics Conference reference cell (GECRC), demonstrating that excited state species vary with gas mixtures, gas pressures, and applied voltages [12,13]. The researchers studied the density profiles of excited state argon [Ar(4s)] in Ar, Ar/O_2 , Ar/CF_4 , and Ar/Cl_2 plasmas by using laser-induced florescence (LIF). Interestingly, McMillin and Zachariah demonstrated that a small amount of electron-attaching gases, such as O_2 and CF_4 , can significantly change the Ar* profile in Ar/O_2 and Ar/CF_4 discharges [13]. Metastable species' roles in CCPs remain extensively investigated in recent reports [14–16]. In a particle-in-cell simulation, Wen et al. investigated electron elastic scattering and plasma density profile transition by incorporating metastable atoms [16].

Various types of gas mixtures have been considered, but in many cases, it has been said that a larger electrode or a larger reactor may be helpful in improving uniformity. The feasibility resides in the fact that one can induce spatial changes of electric fields that are found away from wafer edges. However, increasing the electrode size of a reactor can result in a rising reactor price. Additionally, the enlarged reactor will take up more space on the semiconductor manufacturing line. While this is being accomplished, it is still difficult to achieve uniformity in the electrode edge deposition rate profile [17–20]. By proceeding radially outward from the electrode edge, the geometric configuration of the reactor produces significant variations in various physical aspects, including discharge structure, gas flow, and source gas distribution. Radial profiles of the radical fluxes towards an electrode's surface are affected by these features, resulting in non-uniform profiles of the edge thickness profiles. Several previous studies have observed and discussed this technical problem [20,21].

Considering this study's novelty, compared to our previous studies, we used different reactor geometries in this study. Top electrodes are directly connected to sidewalls, for example. As a result, we can more clearly observe the electrode edge effects caused by excited species because there is only one corner point of the electrode edge. Additionally, the distance between the bottom electrode and the sidewall was altered, which allows us to observe electrode edge effects under a variety of conditions. Further, one of the reactor geometries is similar to the GECRC, so our metastable species distribution results can be compared with those from previous experiments and numerical simulations. Finally, due to the different geometry, different apparatus was used for experimental measurements. In addition, despite its physical importance, prior systematic studies of the metastable species distributions on plasma uniformity used simple chemistry or applied plasma for etching (for example, Ar/CF_4 or Ar/Cl_2). Therefore, a thorough understanding of how metastable species distributions affect plasma uniformity in practical gas mixture deposition reactors is crucial. Specifically, we examine how metastable species distributions impact the plasma uniformity in practical CCP PECVD. The objective of this study was, therefore, to numerically investigate the spatial distributions of metastable helium species (He^*) in He/SiH₄ capacitively coupled plasmas (CCPs) to understand how these plasma density distributions might be optimized.

The results of a two-dimensional fluid model of He discharges are first described, followed by those of He/SiH_4 discharges to deposit hydrogenated amorphous silicon films. We examined which factor is most important to achieve uniform coatings. We retained our CCPs in the 300 mm wafer reactor used by the semiconductor industry in the recent past. We considered design parameters, such as a sidewall gap (radial distance between the electrode edge and the sidewall), electrical condition of the sidewall, and position of the powered electrode. In addition, by increasing the gas pressure while varying the sidewall

We briefly describe our computational model in Section 2, present and discuss our results in detail in Section 3, and provide our conclusions in Section 4.

2. Materials and Methods

2.1. Governing Equations

The governing equations in the self-consistent CCP model are described in our previous publications [22–24]. For brevity, we only provide a brief description of them here. Solving the following equations accounts for the electron energy balance:

$$\frac{\partial(n_e\varepsilon)}{\partial t} + \nabla \cdot \left[\frac{5}{3}n_e\varepsilon\mathbf{v}_e - \frac{5}{3}n_eD_e\nabla\varepsilon\right] = S_{\varepsilon}.$$
(1)

Here, n_e , ε , t, \mathbf{v}_e , D_e , and S_{ε} are the electron density, the electron energy, the time, the electron velocity, the electron diffusion coefficient, and the source of the electron energy, respectively. The electron energy source, S_{ε} , in Equation (1) is given by:

$$S_{\varepsilon} = -eJ_e \cdot E - n_e \sum_{i}^{m_s} \sum_{r} \varepsilon_r k_{r,i} N_i.$$
⁽²⁾

Here, -e, J_e , E, $k_{r,i}$, and N_i are the charge of an electron, the electron flux, the electric field, the rate coefficient of the *r*th reaction with the *i*th species, and the density of the *i*th species, respectively. Further, m_s is the number of species involved in this reaction and ε_r represents the threshold energy in an inelastic collision or the mean electron energy loss in an elastic collision.

2.2. Numerical Methods

Using the commercial multi-physics software CFD-ACE+ (version 2020.5), we simulated the capacitively coupled SiH₄/He plasma using a 2D fluid model [25]. A number of previous studies have successfully used CFD-ACE+ to simulate discharge cases, including both low-pressure inductively coupled plasmas and atmospheric pressure capacitively coupled plasmas [26–28].

By using the Scharfetter–Gummel exponential scheme, we discretized the electron density conservation equation and the electron energy balance equation. By using the kinetic module integrated in CFD-ACE+, the electron energy distribution function (EEDF), which calculates electron impact reaction rate coefficients, electron mobility, and electron diffusivity, was calculated a priori. To reduce computational costs, the 0D Boltzmann equation was solved for the EEDF. A lookup table was generated by solving the 0D Boltzmann equation as a preprocessor step. In Kolobov and Arslanbekov, details about solving the Boltzmann equation are provided [26].

As a result of the small integration time step that should be used for our CCP model, resolving all balances simultaneously will incur a high computational cost. This bottleneck was overcome by adopting a multi-time-scale framework that separated the numerical solution of the balances of neutral species from the numerical solution of electrons, electron energy, and the Poisson equation [29]. RF period was divided into 200 time steps to meet the numerical stability condition in the transient solution. ε_c is defined as the relative error between two estimates of plasma parameters obtained in two successive RF periods, when we take $\varepsilon_c < 10^{-5}$ after 1000 RF periods. With the fast gas velocity involved under the considered pressure condition, the neutral transport time step was set to 1.0×10^{-6} s.

2.3. SiH₄/He Plasma Chemistry

There are many numerical and experimental studies in the literature that investigate SiH_4 /He plasmas, especially those that involve the deposition of a-Si:H films [11]. In fact, the deposition of SiH_4 /He plasmas is often accomplished by diluting SiH_4 with He to

achieve the required film quality. We considered various neutral and charged species in SiH₄/He discharges in this study. The considered species include radical species of both SiH₄ and He, positive and negative ions, stable molecules, and vibrationally excited silanes (charged species: e^- , Si⁺, SiH⁺, SiH₂⁺, SiH₃⁺, H⁺, H₂⁺, H₃⁺, Si₂H₂⁺, Si₂H₄⁺, He⁺, He₂⁺, SiH₃⁻, SiH₂⁻; neutral species: SiH₄, SiH₄⁽¹⁻³⁾, SiH₄⁽²⁻⁴⁾, Si, SiH, SiH₂, SiH₃, Si₂H₂, Si₂H₄, Si₂H₅, Si₂H₆, Si₃H₇, Si₃H₈, Si₄H₉, Si₄H₁₀, Si₅H₁₁, Si₅H₁₂, H₂, H₂^(v = 1), H₂^(v = 2), H₂^(v = 3), H, He, He^{*}, and He₂^{*}). In our previous publications, we explained the chemical mechanisms used in this analysis and we reproduced the deposition rate profiles, which quantitatively matched the experimental data [7,24]. Several experimental studies performed by other researchers were used to obtain the probability coefficients of the reactive radicals in the film deposition model. A unit probability value was also assumed for silicon ions.

2.4. Reactor Geometry and Configuration

We now describe the geometric configuration and operating conditions for the showerhead CCP deposition reactor. Our reactor under consideration is shown schematically in Figure 1. The symmetrical nature of the reactor results in only half of it being depicted in the r-z plane, where r is the radial coordinate and z is the axial coordinate. Processing wafers normally takes place in a cylindrical reactor. A symmetrical axis is represented on the left border and a sidewall is represented on the right boundary. There is a top electrode (showerhead electrode) on our cylindrical CCP reactor, separated from the (dielectric or grounded) ring-shaped component. A ring-shaped component's inner surface serves as the sidewall.



Figure 1. Schematic diagram of our CCP reactor. Spatial variation in the gas velocity magnitude (V_m , m s⁻¹) is depicted for Case 1.

At 13.56 MHz, voltage is applied to the top electrode (or bottom electrode), while the bottom electrode (or top electrode) is grounded. In cases with non-isothermal boundary conditions, the bottom electrode is referred to as the heater because the wafer is sometimes heated during process optimization. A silicon (Si, lossy dielectric; dielectric constant, $k \approx 12$) wafer is placed on top of the bottom electrode. In order to resolve the electrical discontinuity at the wafer edge, a Si ring (also referred to as an edge ring) is placed around the wafer. Table 1 lists the details of the sixteen cases considered in this study.

Case No.	He Flow Rate	SiH ₄ Flow Rate	Sidewall Condition	Gas Pressure	Electrode Spacing	Sidewall Gap
1	5000 sccm	-	Perfect Dielectric	200 Pa	25 mm	52 mm
2	5000 sccm	-	Perfect Dielectric	400 Pa	25 mm	52 mm
3	5000 sccm	-	Perfect Dielectric	800 Pa	25 mm	52 mm
4	5000 sccm	-	Perfect Dielectric	1600 Pa	25 mm	52 mm
5	5000 sccm	50 sccm	Perfect Dielectric	100 Pa	25 mm	52 mm
6	5000 sccm	50 sccm	Perfect Dielectric	200 Pa	25 mm	52 mm
7	5000 sccm	50 sccm	Perfect Dielectric	300 Pa	25 mm	52 mm
8	5000 sccm	50 sccm	Perfect Dielectric	400 Pa	25 mm	52 mm
9	5000 sccm	100 sccm	Perfect Dielectric	100 Pa	25 mm	52 mm
10	5000 sccm	100 sccm	Perfect Dielectric	200 Pa	25 mm	52 mm
11	5000 sccm	100 sccm	Perfect Dielectric	300 Pa	25 mm	52 mm
12	5000 sccm	100 sccm	Perfect Dielectric	400 Pa	25 mm	52 mm
13	5000 sccm	50 sccm	Quartz	400 Pa	15 mm	7 mm
14	5000 sccm	50 sccm	Grounded	400 Pa	15 mm	7 mm
15	5000 sccm	50 sccm	Quartz	1200 Pa	15 mm	7 mm
16	5000 sccm	50 sccm	Grounded	1200 Pa	15 mm	7 mm

Table 1. List of cases considered in this study.

3. Results

3.1. Effects of the Gas Pressure on He* Density Distributions in He-Only Discharges

Based on our previous studies, we found that plasma distribution characteristics change significantly with gas pressure. As the ion-neutral collision rate increases with higher pressure, the mean free path decreases. It increases ion and electron density and, consequently, decreases the sheath width by decreasing the drift-diffusion loss rate of charged species at the electrodes. Therefore, the focus of this section is our observation that changes in gas pressure affect the spatial variation in plasma parameters as a result of the He* density distributions.

The spatial variations in the excitation (He + e⁻ \rightarrow He^{*} + e⁻) and ionization (He + e⁻ \rightarrow He⁺ + e⁻ + e⁻) rates for Case 1 (He only discharge; see Table 1), as well as the electron power density (W m⁻³) and electron density (N_e , m⁻³), respectively, are shown in Figure 2a–d. We set the gas pressure at 200 Pa, the applied RF power at 100 W, and the He flow rate at 5000 sccm.

In Figure 2a,b, excitation and ionization rates increase near the edge of the "bottom powered electrode", with an axial shift towards it due to electrons heated by the sheath. Source functions for He* and He⁺ have essentially the same shape. A radial extreme in the excitation rate and ionization rate distributions is caused by electrical field enhancement at the electrode edge, which creates additional sheath heating. A shift in the extreme towards the bottom-powered electrode is also produced as a result. The electron power density in the bulk region varies only slightly, as shown in Figure 2c. Note that near corners, converging sheaths produce a large heating rate, where electron power density is maximum. The electrons diffuse from the site of production in an axial direction and the peak electron density is seen near the maximum of the ionization rate distribution, as shown in Figure 2d.

We display the He* density distributions at 200 Pa (Case 1), 400 Pa (Case 2), 800 Pa (Case 3), and 1600 Pa (Case 4) to observe the gas pressure effects thereon. It was common for input conditions to be set at 100 W of power and 5000 sccm of flow.



Figure 2. Contour plots for the spatial profiles of the time-averaged (**a**) He^{*} production rate for Case 1 and (**b**) He⁺ production rate for Case 1 for $r \ge 0$ mm. Contour plots for the spatial profiles of the time-averaged (**c**) electron power density (W m⁻³) for Case 1 and (**d**) electron density (m⁻³) for Case 1 for $r \ge 0$ mm.

The distribution of He^{*} in the axial direction between electrodes is fairly symmetrical at $p_g = 200$ Pa. As shown in Figure 3a, He^{*} density peaks near the electrode edges in accordance with the source function. At $p_g = 200$ Pa, the electron density shows a similar profile. Similar profiles are not surprising since electrostatic excitation, the main source of He^{*}, has a distribution that is essentially identical to ionization. Both profiles should be similar so long as diffusion dominates transport.



Figure 3. The spatial profiles of the time-averaged He* density (m⁻³) (**a**) for Case 1, (**b**) for Case 2, (**c**) for Case 3, and (**d**) for Case 4.

The peak density of He* shifts slightly toward the bottom-powered electrode as the gas pressure increases to 400 Pa, resulting in a noticeable skew in the profile of Figure 3b. In Figure 3c, the skew is more evident when the gas pressure is increased further to 800 Pa and the density of He* falls closer to the bottom electrode. In Figure 3d, it can be observed at 1600 Pa that the density of He* peaks at the edge of the powered electrode at the bottom and, in addition, a secondary peak can also be observed at the center of the grounded electrode at the top.

It can be identified that He* moved towards the bottom electrode and its skew towards the corner as the pressure increased, which was principally determined by the reduction in electron mean free paths and the increase in He^{*} quenching due to the increased pressure. The electron mean free path becomes shorter as the gas pressure increases, while electron density increases as the gas pressure increases. Therefore, electrons heated by the plasma sheath and located at the powered electrode are able to dissipate their energy by excitation closer to the electrode. By increasing the electron density, the sheath thickness and velocity decrease, resulting in less energetic sheath heated electrons with a shorter dissipation length. With the collapse of the sheath (and pre-sheath), the electric field enhancement along the edge of the electrode becomes more pronounced, causing a larger skew in the He* profile. At higher pressures, the mean free path of electrons is sufficiently short for the collapsing sheath, resulting from the increase in power deposition to have an effect on the He* source function. Due to the increased plasma density, quenching of He* also increases, thus, restricting the He* nearer to the source of generation, therefore, causing it to be confined closer to its location.

3.2. Effects of SiH₄ Addition on Plasma Density Distributions in He/SiH₄ Discharges

Because He^{*} quenching has a high-reaction-rate coefficient, we numerically examined noticeable alterations to the He^{*} density by adding SiH₄ in this study. In a case at $p_g = 100$ Pa (Cases 5), a case at $p_g = 200$ Pa (Cases 6), a case at $p_g = 300$ Pa (Case 7), and a case at $p_g = 400$ Pa (Case 8), a mixture of 50 sccm SiH₄ and 5000 sccm He (so the mole fraction of SiH₄ is only about 1%) was used as the feed gas. As in the previous cases, the feed gas was set to be supplied from the showerhead and the applied RF power of 100 W was applied to the bottom electrode in the same manner.

The density profile of He* obtained by varying the gas pressure is shown in Figure 4 for Cases 5 ($p_g = 100$ Pa), 6 ($p_g = 200$ Pa), 7 ($p_g = 300$ Pa), and 8 ($p_g = 400$ Pa). In pure helium, because of reasons that were discussed earlier, the density profile of He* in pure helium is slightly skewed and the peak occurs near the edge of the electrode at $p_g \ge 400$ Pa. However, even though SiH₄ is mixed with only about 1 percent, the profile shape is substantially altered, even at a lower gas pressure of $p_g = 200$ Pa in Case 6. Furthermore, the peak density of He* decreases as a result of the quenching of He* by SiH₄. A case at $p_g = 300$ Pa (Case 7) and a case at $p_g = 400$ Pa (Case 8) demonstrate the following: their He* density profiles show more skewness (double-humped shape in the inter-electrode region) because of the higher gas pressure.





In the same way, Figure 5 shows the time-averaged Si⁺ density profiles for Cases 5, 6, 7, and 8. As a result of the inlet condition (He dilution), although the threshold energy for SiH₄ ionization is lower than that for He excitation or ionization, the peak value for Si⁺ production rate by dissociative excitation (He^{*} quenching) is higher than that for SiH₂⁺ production rate by direct-impact ionization. When Case 5 ($p_g = 100$ Pa; 50 sccm SiH₄) is considered, the peak value of Si⁺ production rate due to dissociative excitation is 8.9×10^{-7} kmol m⁻³ s⁻¹, while the peak value of SiH₂⁺ production rate due to direct-impact ionization is 7.0×10^{-7} kmol m⁻³ s⁻¹. Additionally, the peak value of the Si⁺ density is determined by the peak value of the He^{*} density. Kushner, subsequently, set the branching and rate coefficients for He^{*} quenching by SiH₄ as follows [30]:

$$He^* + SiH_4 \rightarrow Si^+ + 2H + H_2 + He + e^-, k_r = 1.3 \times 10^{-16} \text{ m}^3 \text{ s}^{-1}.$$
 (3)



In this equation, k_r represents the reaction rate coefficient.

Figure 5. The spatial profiles of the time-averaged Si⁺ density (m^{-3}) (**a**) for Case 5, (**b**) for Case 6, (**c**) for Case 7, and (**d**) for Case 8.

Expecting higher He* quenching effects, the amount of SiH₄ fed in Cases 9 (He = 5000 sccm; SiH₄ = 100 sccm; $p_g = 100$ Pa), 10 (He = 5000 sccm; SiH₄ = 100 sccm; $p_g = 100$ Pa), 11 (He = 5000 sccm; SiH₄ = 100 sccm; $p_g = 100$ Pa), and 12 (He = 5000 sccm; SiH₄ = 100 sccm; $p_g = 100$ Pa) was two-times the amount fed in Cases 5, 6, 7, and 8; we increased the SiH₄ flow rate to 100 sccm in Cases 9, 10, 11, and 12. Similar to Cases 5, 6, 7, and 8, we also varied the gas pressure in Cases 9 ($p_g = 100$ Pa), 10 ($p_g = 200$ Pa), 11 ($p_g = 300$ Pa), and 12 ($p_g = 400$ Pa). Figures 6 and 7 show results of the time-averaged density profiles of He* and Si⁺ for Cases 9, 10, 11, and 12. By comparison with Cases 5, 6, 7, and 8, the He* density profile exhibits a more severe skewness caused by the heightened effect of quenching He*. Consequently, a significantly larger skewness is observed in the Si⁺ density profile.

The time-averaged electron density profiles for Cases 9, 10, 11, and 12 are shown in Figure 8. In most of these cases, the off-axis maxima of the electron density (N_{eo}) are observed near the bottom electrode edge as a result of the excitation and ionization rate distributions. A similar comparison is shown in Figure 9, where time-averaged SiH₃ density profiles are shown for Cases 9, 10, 11, and 12. In Cases 9, 10, 11, and 12, off-axis maxima of the SiH₃ density are also observed near the bottom-electrode edge.



Figure 6. The spatial profiles of the time-averaged He^{*} density (m^{-3}) (**a**) for Case 9, (**b**) for Case 10, (**c**) for Case 11, and (**d**) for Case 12.



Figure 7. The spatial profiles of the time-averaged Si⁺ density (m^{-3}) (**a**) for Case 9, (**b**) for Case 10, (**c**) for Case 11, and (**d**) for Case 12.

For a case of 2% SiH₄ at $p_g = 100$ Pa (Case 9), a case of 2% SiH₄ at $p_g = 200$ Pa (Case 10), a case of 2% SiH₄ at $p_g = 300$ Pa (Case 11), and a case of 2% SiH₄ at $p_g = 400$ Pa (Case 12), their deposition rate profiles are compared in Figure 10. The simulated deposition rate profiles are normalized using their corresponding average values. These profiles have one important characteristic in common: at 0 mm $\leq r \leq 120$ mm, they are nearly flat, but their shapes vary greatly at r > 120 mm. Recent semiconductor processing prohibits these abrupt changes.

3.3. Effects of the Modified Reactor Geometry

Due to the fact that the He^{*} and Si⁺ density distributions are localized near the bottomelectrode edge in the previous He/SiH₄ mixture cases, the electron and SiH₃ density distributions are also localized. As a result, the profiles of deposition rates for Cases 9, 10, 11, and 12 all increase abruptly in the vicinity of the edges. To resolve the localization problem, we examined the effects of the modified reactor geometry in this section (to obtain sufficient uniformity).



Figure 8. The spatial profiles of the time-averaged electron density (N_e , m⁻³) (**a**) for Case 9, (**b**) for Case 10, (**c**) for Case 11, and (**d**) for Case 12.



Figure 9. The spatial profiles of the time-averaged SiH₃ density (m^{-3}) (**a**) for Case 9, (**b**) for Case 10, (**c**) for Case 11, and (**d**) for Case 12.

Figure 11 shows the modified reactor. Not only did we reduce the electrode spacing (d = 15 mm) but also the radial gap between the bottom electrode and the sidewall. In the design of industrial reactors, reducing the reactor geometry is preferable. As a mimic of our experimental apparatus, we set the top electrode to receive RF power and the bottom electrode to be grounded. The sidewall boundary condition was altered in cases with the modified reactor from dielectric to grounded. We used a quartz sidewall with a finite thickness (45 mm) for experimental validation for cases with a dielectric sidewall, instead of a perfect dielectric sidewall (See Figure 1). In order to maintain consistency, a small amount of 50 sccm SiH₄ was diluted in 5000 sccm He and a frequency of 13.56 MHz V_{RF} was applied.



Figure 10. Deposition rate profiles are plotted for Cases 9, 10, 11, and 12. The profiles are normalized with their corresponding averaged values.



Figure 11. Schematic diagram of the modified CCP reactor. Spatial variation in the gas temperature (T_{g}, K) is depicted for Case 13.

Furthermore, it is noteworthy that the results presented so far pertain to isothermal wall reactors, where the showerhead surface, the sidewall surface, and the wafer surface were all maintained at the same temperature ($T_{sh} = T_{sw} = T_w = 423$ K; See Figure 1). In practice, however, PECVD was performed frequently under non-isothermal wall boundary conditions to control the process results. Our reactor is set up so that the showerhead temperature (T_{sh}) is 473 K, the sidewall temperature (T_{sw}) is 423 K, and the wafer temperature (T_w) is 673 K, in order to achieve non-isothermal wall boundary conditions. In Figure 11, for the modified reactor, the effect of non-isothermal wall boundary conditions on gas temperature distributions is shown ($T_{sh} = 473$ K, $T_{sw} = 423$ K, and $T_w = 673$ K).

Modulating the electrode temperature allows the film properties to be controlled, since the deposition is taking place on the bottom electrode. Due to the previous validation of our surface model under the $T_w = 673$ K condition, we decided to use the non-isothermal wall temperature condition. We chose a shorter distance between the top and bottom electrodes to achieve better process uniformity. Additionally, the shorter distance between the bottom electrode and the sidewall facilitates easier control of plasma distribution by changing the sidewall conditions. The bottom electrode was grounded since the RF- powered electrode enhances the electrode edge effect more. Furthermore, the bottom electrode in our experimental apparatus is grounded.

Figure 12 depicts the time-averaged He* density profiles obtained by varying the sidewall condition from dielectric to grounded by increasing the gas pressure from 400 to 1200 Pa for Case 13 (p_g = 400 Pa; modified geometry; quartz sidewall), Case 14 (p_g = 400 Pa; modified geometry; grounded sidewall), Case 15 (p_g = 1200 Pa; modified geometry; quartz sidewall), and Case 16 (p_g = 1200 Pa; modified geometry; grounded sidewall). Because the electric fields are enhanced near the top- and bottom-electrode edges, no matter what the gas conditions and sidewall conditions are, the excitation occurs efficiently. These four cases exhibit radially uniform excitation rate profiles at r < 130 mm. Due to strong local electric fields and the narrow gaps between the bottom electrode and the grounded sidewall, the surface losses of electrons and ions are enhanced when the grounded sidewall is present. Therefore, between the case with the dielectric sidewall at $p_g = 400$ Pa (Cases 13) and the case with the grounded sidewall at $p_g = 400$ Pa (Case 14) and between the case with the dielectric sidewall at $p_g = 1200$ Pa (Cases 15) and the case with the grounded sidewall at $p_g = 1200$ Pa (Case 16) there is a huge difference in the distributions of plasma parameters: cases with quartz sidewalls have lower peak values of He* density than cases with grounded sidewalls. In addition, for cases with grounded sidewalls, the density profiles of He^{*} are distorted near the bottom-electrode edge, regardless of the gas pressure. In cases with the quartz sidewall, despite the fact that the peak values are located a little closer to the edge of the wafer, the He* density profiles are relatively uniform across the inter-electrode region, regardless of gas pressure.





The contours in Figure 13a–d illustrate the time-averaged electron density profiles for Cases 13, 14, 15, and 16, respectively. A similar feature can be seen in Figure 12, where radial distributions of He* density are uniform at r > 130 mm and, since the electron density profile corresponds to He* density profiles, peak electron densities are seen near the electrode edges in Figure 13. Comparing the case with the dielectric sidewall at $p_g = 400$ Pa (Cases 13) with the case with the dielectric sidewall at $p_g = 1200$ Pa (Cases 15) and the case with the grounded sidewall at $p_g = 1200$ Pa (Cases 16), although the gas pressure is higher, the electron density profiles are more uniform in the radial direction for Case 15 and Case 16, respectively. By suppressing the disturbance effect that arises not only from the shape of the electrode edge but also from the narrow electrode spacing (d = 15 mm), the distribution of electron density becomes

more uniform. The suppression of the disturbance is primarily the result of electrons moving radially inwards. Additionally, the peak electron density increases in Case 14 and Case 16, regardless of the gas pressure when the sidewall condition changes from dielectric to grounded. The bulk electron densities, however, are maintained at similar levels. The thinner sheath thickness on the dielectric sidewall is detected, regardless of the gas pressure. The result is an improvement in the plasma uniformity in the inter-electrode region.



Figure 13. The spatial profiles of the time-averaged electron density (N_e , m⁻³) (**a**) for Case 13, (**b**) for Case 14, (**c**) for Case 15, and (**d**) for Case 16.

In Figure 14a–d, we illustrate the time-averaged SiH₃ density profiles for Cases 13, 14, 15, and 16, respectively. Gas pressure affects the differences between the local maximum SiH₃ densities (on-axis maximum values detected on the axis). In spite of the common edge-high distributions among the four cases, there are still substantial differences in the details, especially near the electrode edges: radial changes in the thickness of boundary layers are occurring. The phenomenon is explained by the fact that there are non-negligible differences between the off-axis maxima of the electron densities (N_{eo}). N_{eo} 's difference in position causes the radical density field to behave differently at the edge of the electrode due to a difference in boundary layer thickness. Moreover, since the grounded sidewall induces a thicker radial sheath for Cases 14 and Case 16, their radial positions of N_{eo} and their maximum SiH₃ density are detected farther away from the sidewall.

In Figure 15a, the deposition rate profiles for Cases 13 and 14 are compared and in Figure 15b, those for Cases 15 and 16 are compared. The simulated deposition rate profiles are normalized using their corresponding average values. Deposition rate profiles for Cases 13 and 15 match well with experimental data for their respective cases. This comparison was performed in the radial direction due to the axisymmetry in the experimental data. Azimuthal non-uniformity is indicated by the error bars. It is apparent that all profiles have an important feature in common: when 0 mm $\leq r \leq 130$ mm, their shapes are nearly flat; at r > 130 mm, however, their shapes vary greatly. By using the grounded sidewall, the areas of the effective grounded surfaces are enlarged, which results in a reduction in the surface fluxes of charged species near the bottom-electrode edge. It is mostly the result of distortion of the electric potential distribution caused by the higher gas pressure. Thus, in Cases 14 and 16, the thicker bottom sheath induced at approximately r = 150 mm reduces the deposition rate profile, whereas the dielectric sidewall significantly increases the deposition rate in Cases 13 and 15.



Figure 14. The spatial profiles of the time-averaged SiH₃ density (m^{-3}) (**a**) for Case 13, (**b**) for Case 14, (**c**) for Case 15, and (**d**) for Case 16.



Figure 15. Effect of the sidewall and the gas pressure on the deposition rates. (**a**) The deposition rate profiles of Cases 13 and 14 are compared. Experimental data from Case 13 are superimposed on the profile of Case 13. (**b**) The deposition rate profiles of Cases 15 and 16 are compared. Experimental data from Case 15 are superimposed on the profile of Case 15. The profiles are normalized with their corresponding averaged values.

4. Conclusions

In this study, we numerically investigated helium metastable distributions and their effects on the uniformity of hydrogenated amorphous silicon depositions in He/SiH₄ CCPs. We presented the results of a two-dimensional fluid model of He discharges and then the results of He/SiH₄ discharges to deposit hydrogenated amorphous silicon films. Utilizing a 300 mm wafer reactor in the semiconductor industry, we investigated what factor determines the uniformity of plasma density distribution.

As observed in Cases 5–12, excitation and ionization rate profiles were enhanced and localized near the electrode edge when a reactor similar to GECRC was used. We, therefore, investigated the effects of modifying the reactor geometry in order to improve uniformity in Cases 13–16. In the modification, we reduced the electrode spacing and the radial gap between the bottom electrode and the sidewall. In addition, by toggling the sidewall condition from dielectric to grounded and by varying the gas pressure (from 400 Pa to 1200 Pa), we observed that the effect of the sidewall on the deposition rate profiles was affected by the gas pressure. Regardless of the gas pressure, for the dielectric sidewall, the thinner sheath thickness on the sidewall was detected and the plasma uniformity was improved in the inter-electrode region.

Author Contributions: Conceptualization, H.J.K.; Data curation, S.K. and H.J.K.; Formal analysis, S.J., S.K., K.L. and H.J.K.; Funding acquisition, K.L. and H.J.K.; Investigation, S.K., K.L. and H.J.K.; Methodology, S.J., S.K. and H.J.K.; Project administration, H.J.K.; Resources, H.J.K.; Software, H.J.K.; Supervision, H.J.K.; Validation, S.J. and H.J.K.; Visualization, S.J. and H.J.K.; Writing—original draft, S.J. and H.J.K.; Writing—review and editing, K.L. and H.J.K. All authors have read and agreed to the published version of the manuscript.

Funding: This work was supported by the Gachon University research fund of 2020 (GCU-202008470001). This work was supported by Korea Institute for Advancement of Technology (KIAT) grant funded by Korea Government (MOTIE) (P0020612, 2022 The Competency Development Program for Industry Specialist).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Data sharing is not applicable to this article.

Conflicts of Interest: The authors declare no conflict of interest.

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