

Article

Characterization of Porous Titanium-Hydroxyapatite Composite Biological Coating on Polyetheretherketone (PEEK) by Vacuum Plasma Spraying

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Abstract: Titanium powders and hydroxyapatite powders were used to fabricate the titanium and hydroxyapatite composite coating (Ti/Ti/HA) on the Polyetheretherketone (PEEK) by vacuum plasma spraying (VPS). The phase composition and morphology of the Ti/Ti/HA coating were analyzed by XRD, SEM, industrial CT, and three-dimensional contour profiler (DEKTA XT). The results showed that the phase composition of the Ti/Ti/HA coating was dominated by the HA phase and a small amount of the Ti phase. The Ti/Ti/HA composite coating presented an obvious three-layer structure with different roughness and porosity. The bottom was Ti coating with a porosity less than 10%; the middle was porous Ti coating with a porosity of $35\% \pm 10\%$; the surface coating was HA coating with the crystallinity near 95%, a porosity of $32\% \pm 10\%$, a roughness $R_a = 30 \pm 10 \mu\text{m}$. Moreover, the TG-DSC and ATR-FTIR results showed that VPS coating has no effect on thermochemical properties of PEEK material. The average bond strength of the composite coating was achieved 32 MPa.

Keywords: PEEK; vacuum plasma spraying; composite coating; microstructure; bond strength



Citation: Hu, F.; Fan, X.; Peng, F.; Yan, X.; Song, J.; Deng, C.; Liu, M.; Zeng, D.; Ning, C. Characterization of Porous Titanium-Hydroxyapatite Composite Biological Coating on Polyetheretherketone (PEEK) by Vacuum Plasma Spraying. *Coatings* **2022**, *12*, 433. <https://doi.org/10.3390/coatings12040433>

Academic Editor: Shih-Chen Shi

Received: 26 November 2021

Accepted: 15 March 2022

Published: 24 March 2022

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1. Introduction

Polyaryletheretherketone (PEEK), a thermoplastic polymer, has an elastic modulus and density that closely matches bone. Thus, it has the potential to prevent stress-shielding and bone resorption around the implant [1–3]. In addition, the aromatic backbone and semi-crystalline nature of PEEK provide high mechanical strength, thermal and chemical stability, and biocompatibility [4–6]. Thus, PEEK is popular in a number of clinical applications.

However, the hydrophobic surface of PEEK causes a fibrous tissue formation of the bone/implant interface that prevents bone apposition, increasing micromotion and potentially loosening of the prosthesis [7]. Therefore, it is crucial to improve the osteointegration of the surface of PEEK for implant application. One feasibly successful method for attaining this goal is changing the chemical–physical property of the surface by apposition of rough titanium or hydroxyapatite coatings, which are famous for their osteoconductive properties [8]. Various methods have been developed for Ti/HA composite coatings including plasma spraying, high velocity oxy-fuel spraying, sputter deposition, pulsed laser deposition, ion-beam assisted deposition, sol-gel coating, etc. [9]. Among them, plasma spraying is the most widely used and accepted technology for surfacing implants with highly osteointegration layers [10]. In particular, several studies have demonstrated that vacuum

plasma spraying (VPS) is an appropriate method to deposit HA composite coatings in PEEK implants [11,12]. For example, Ti/HA, TiO₂/HA, etc., composite coatings prepared by VPS present higher bond strength and rough microstructure of implants osseointegration in the reported literatures [13–15]. On the other hand, the pores of coatings also increase the contact area between porous surfaces and implant bone tissue, improving the stability of the implant over time [16,17]. Some researchers suggested that the porosity percentage of the surface layer of implants should be in the range of 25–66%, which facilitates improved better osseointegration [18]. However, the existence of high porosity decreases the bond strength of coating, leading to early fatigue failure [19]. Therefore, it is essential to maintain the high mechanical properties of the coating on the implant while providing adequate porosity for the osseointegration properties.

In this study, we attempt to fabricate the Ti/Ti/HA composite coating on PEEK substrates by VPS. In detail, the bottom Ti coating to assure the bond strength, the rough porous intermediate titanium coating to improve the osseointegration, and the high crystallinity HA surface layer to assure the high bioactivity. The Ti/Ti/HA composite coatings were subsequently researched by their thickness, porosity, roughness. The bonding strength mechanism between the polymer PEEK and metal Ti coating were studied. The morphology indicates that the composite coating possesses high roughness and porosity, and the better bonding strength of the coating ensures the stability of the coating during service.

2. Experiments and Methods

2.1. Coating Deposition

PEEK substrates (melting point 334 °C, Junchen Plastic Material Company, Dongguan, China) of 25 mm diameter and 10 mm thick were used. They were degreased in ethanol and then grit-blasted prior to spraying. The Ti and HA powders (99.95%, Nanjing EPp Nanomaterials Co., LTD, Nanjing, China) were used in this work. The morphology and size of the powders are shown in Figure 1. The average size of spongy titanium powder was 30 µm (Figure 1a), which was used to fabricate the bottom titanium coating; the average size of spongy titanium was 98 µm (Figure 1b), which was used to prepare the porous and rough titanium coating; the average size of HA powder was 45 µm (Figure 1c), which was used to fabricate the surface layer. Coatings were carried out by VPS with a F4-VB plasma gun (Oerlikon Metco, Winterthur, Switzerland). Table 1 sums up the VPS operating parameters.

Table 1. Plasma spraying coating parameters.

Spraying Parameters	Ti (Bottom Coating)	Ti (Middle Coating)	HA (Surface Coating)
Spraying distance (mm)	260	280	320
Plasma gas (Ar/H ₂)	40/5	40/4	40/6
Carried gas Ar (L·min ^{−1})	2.1	2.1	2.1
Robotic arm speed (mm/s)	1000	1000	1000

2.2. Coating Characterization and Property

The thickness and porosity of the coating were measured in accordance with ASTM F1854. The porosity of the coating was analyzed and calculated by ImageJ on the polished cross-section of the coating. The pore characteristics were analyzed by industrial computed tomography (Vtomex S micron CT, General Electric Company, Boston, MA, USA). The surface roughness of all coatings was measured using the Profilometer (Mitatoya Surf test SJ-301, Bruker, Karlsruhe, Germany) to evaluate surface morphology to obtain the Ra. A FESEM (Nova-Nano-430, Hillsboro, OR, USA) was employed to observe the microstructure of the coating (morphology of cross section and surface). Additionally, EDS analysis (Oxford Incax sight 6427, Hillsboro, OR, USA) was conducted with the same equipment to study the composition of the coating. An X-ray diffractometer (D8 Advance, BRUCKER AXS, Karlsruhe, Germany) equipped with a Co-K α anti-cathode ($\lambda = 0.179$ nm) was used to determine the phase composition of the coating (scanning step: 0.02°, scanning speed

2°/min, scanning range 20° to 80°). The crystallinity of the HA coating was calculated by the peak area ratio method [20,21]. The calculation formula was as follows:

$$\text{Crystallinity}(\%) = \frac{A_P}{A_P + A_C} \quad (1)$$

where A_C was the total integrated area of the amorphous phase peaks within the range of 2θ (25°~37°) in the XRD image. A_P was the total integrated area of all HA peaks within the range of 2θ (25°~37°) in the XRD image.

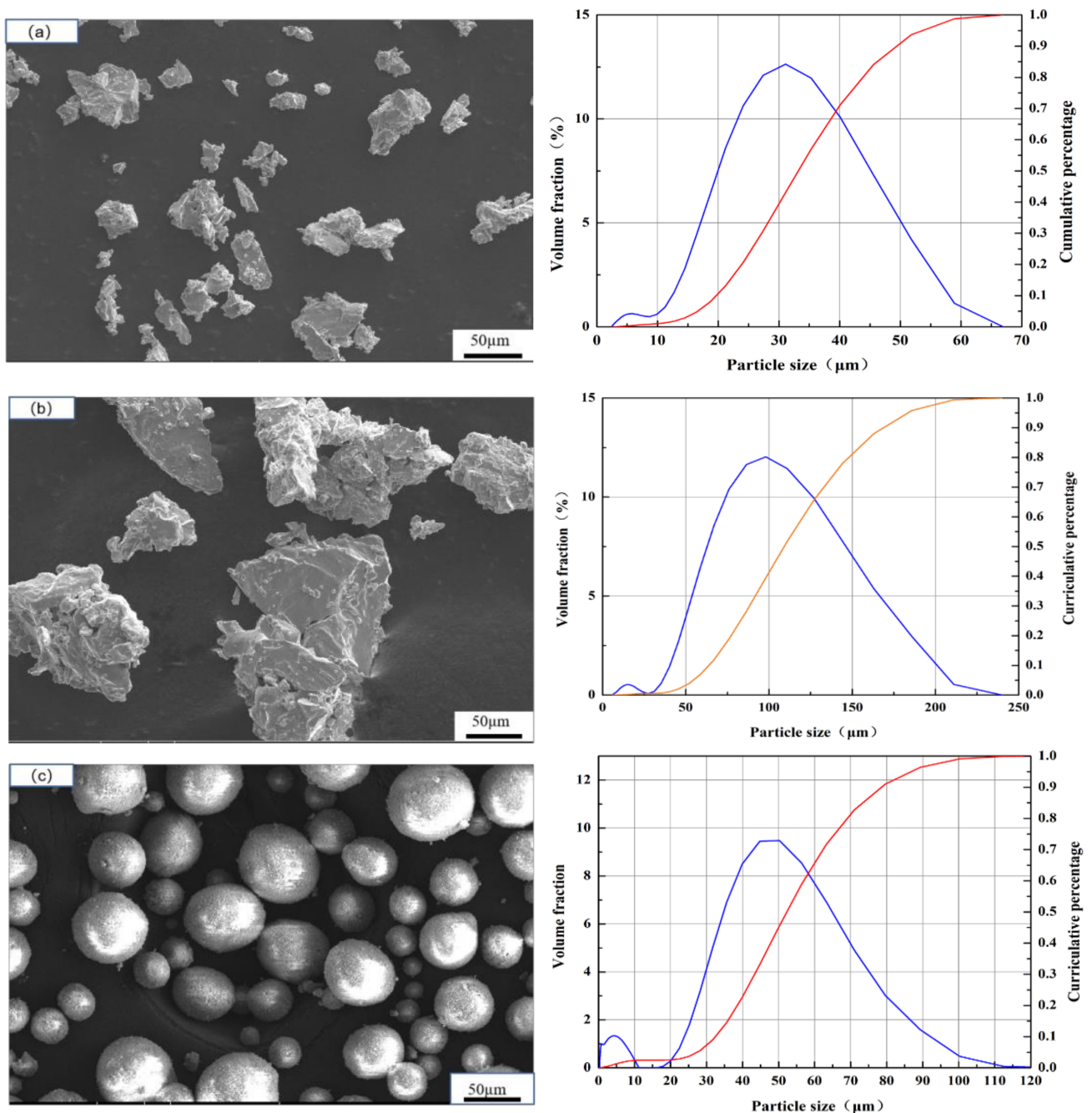


Figure 1. Size and morphology of the feedstocks: (a) spongy titanium powder with fine size; (b) spongy titanium powder with coarse size; (c) spherical hydroxyapatite powder.

Finally, the bond strength of the sprayed coatings was determined by standard adhesion strength tests in accordance with ASTM F1147-05. During the measurement, a tensile

load was applied to the sample at a crosshead velocity of 25 mm/min. The average bond strength and standard deviation were measured for every six samples.

2.3. Influence of the Coating Process on the Thermal-Chemical PEEK Properties

The thermal–chemical properties (glass transition, melting temperature) were evaluated using dynamic difference calorimetry (STA-449-F5, Semb, Germany) in accordance with ASTM D3418-12, using samples with a diameter of 3 mm and height of 1.5 mm. Coated samples as well as uncoated samples were tested. The samples were heated from 40 to 400 °C with a heating rate of 10 °C/min. Additionally, the infrared absorption spectra were determined for coated and uncoated PEEK samples using a FTIR system (Nicolet IS5, Thermo Fisher, Waltham, MA, USA) with an attenuated total reflection (ATR) unit to evaluate if the coating process led to changes of the chemical structure of functional groups of the PEEK samples. The results of the dynamic difference calorimetry and ATR FT-IR absorption spectroscopy were compared to the requirements given in ASTM F2026-14 for medical PEEK.

3. Results and Discussions

3.1. Coating Thickness, Porosity, and Roughness

The morphology of the Ti/Ti/HA coating is shown in Figure 2. The deposited coating presented an obvious three-layer structure. The bottom was a dense Ti coating with a porosity less than 10%; the middle was a porous Ti coating with the porosity of $35\% \pm 10\%$ and a roughness of $R_a = 34 \pm 15 \mu\text{m}$; the surface coating was a thin HA coating with a thickness of $30 \pm 10 \mu\text{m}$, a roughness of $R_a = 30 \pm 10 \mu\text{m}$ (Figure 3). The thickness, porosity and roughness of each layer were shown in Table 2. The total thickness of the composite Ti/Ti/HA coating was $200 \pm 30 \mu\text{m}$, which was within the thickness range (50–300 μm) reported in the relevant literature for PEEK implant coating applications [22]. Each layer had different thickness, porosity, and roughness for different purposes. The porosity and surface roughness of the coating mainly depended on the plasma spraying parameters and the sprayed powder characteristics. Compared with the middle titanium coating, the powder size of the bottom titanium coating was relatively smaller ($d_{50} = 38 \mu\text{m}$), and the plasma spraying power was greater. Therefore, the powder melting degree was high, and the deposited bottom coating was dense, thus improving the bonding strength of the coating (Figures 4 and 5). The middle titanium coating presented a porous structure in the form of interconnected pores, with a porosity $35\% \pm 10\%$ and a large pore size (partial pore sizes have been marked in Figure 2b). The pores in the coating will facilitate the transportation of liquid, the formation of capillaries, and promote the formation of new bone tissue [23,24]. The industrial computed tomography (CT) results showing the pore diameter distribution of the coating are presented in Figure 6. As it can be seen, the pore diameter of the coating was mainly between 80 and 120 μm , and the maximum pore size reached 300 μm . Due to vascularization, the pore size has been shown to influence the process of bone formation. Small pores favor hypoxic conditions that induce osteochondrogenic cells prior to osteogenesis, while large pores, which are well vascularized, lead to direct osteogenesis (without osteochondrogenic cells) [17]. In addition, a porous layer could provide an adequate conduit for bone ingrowth and avoided tissue necrosis common at the center of large fully porous implants in cases of limited vascular and nutrient supply [25].

Table 2. Thickness, porosity, and surface roughness of the coating.

Coating Properties	Bottom Coating	Middle Coating	HA Coating
Thickness	$70 \pm 10 \mu\text{m}$	$100 \pm 10 \mu\text{m}$	$30 \pm 10 \mu\text{m}$
Porosity	<10%	$35\% \pm 10\%$	$32\% \pm 10\%$
Roughness	$24 \pm 5 \mu\text{m}$	$34 \pm 15 \mu\text{m}$	$30 \pm 10 \mu\text{m}$

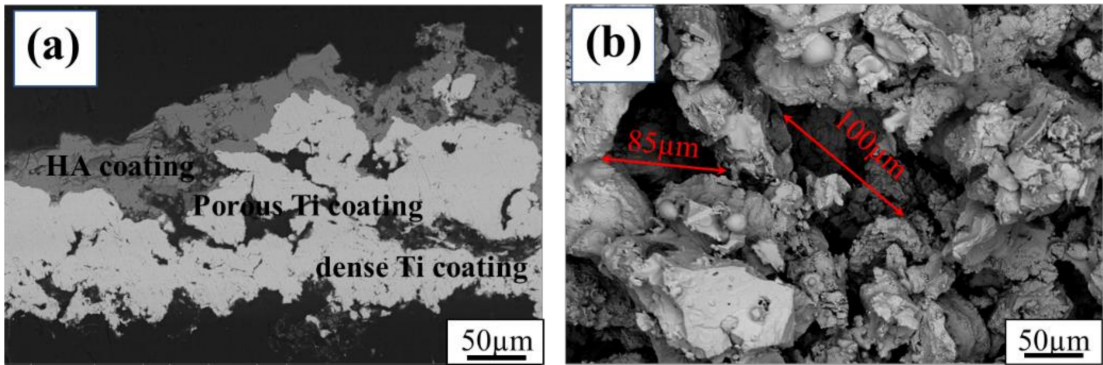


Figure 2. SEM micrographs of the Ti/Ti/HA coating: (a) cross section, (b) porous structure.

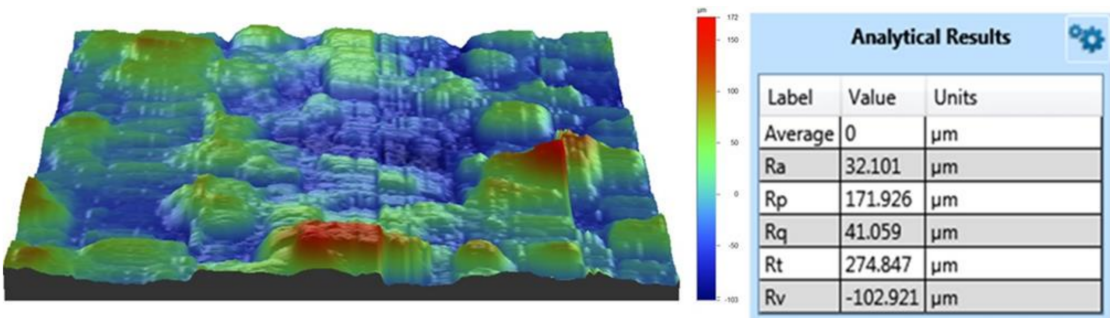


Figure 3. Roughness of the Ti/Ti/HA coating.

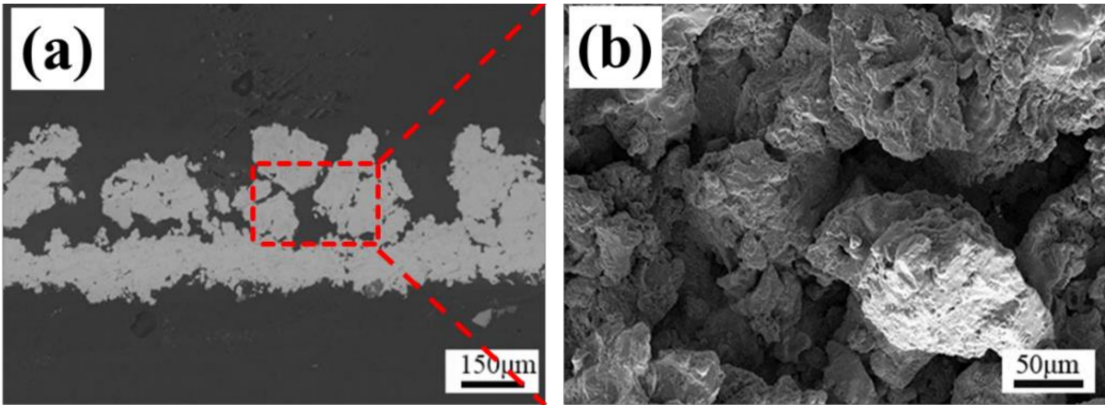


Figure 4. Morphology of the Ti/Ti coating: (a) cross section, (b) porous structure.

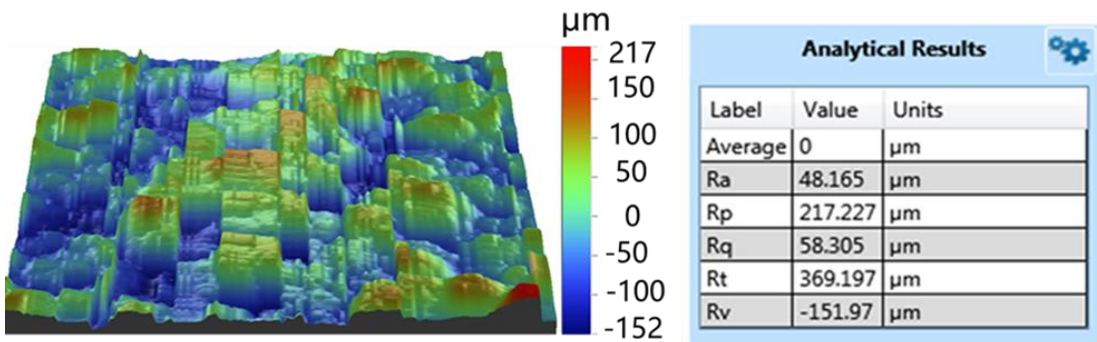


Figure 5. Roughness of the Ti coating.

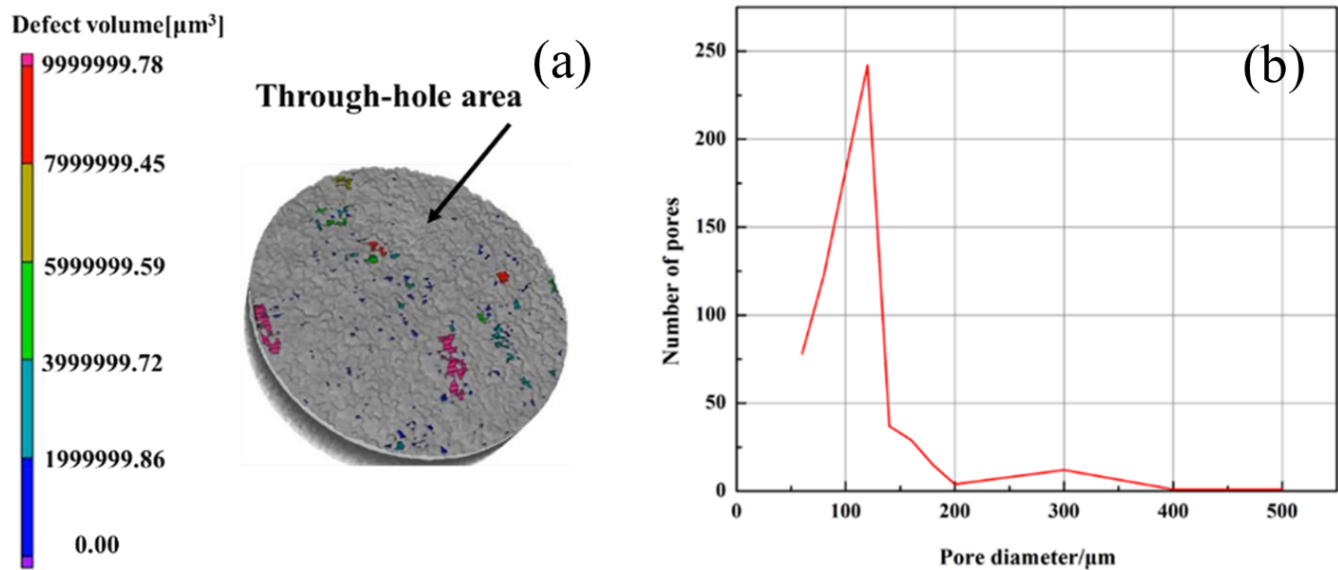
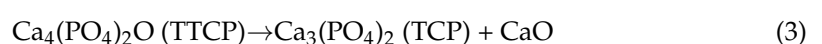


Figure 6. Industrial computed tomography of coating: (a) the pore distribution of the coating, (b) the statistics of pore diameter of coating.

3.2. Coating Surface Morphology, Phase Composition and Crystallinity

The surface morphology of the PEEK-coating at different magnification was shown in Figure 7. The flat particles of the coating surface were connected to each other to form a continuous phase. The un-melted particles were embedded in the continuous phase to form a cavity-like structure or in the form of spherical droplets, which ended up with a rough coated surface. Besides, many irregular pores of different sizes and shapes were found in the coating. Overall, the coating ensures its integrity. The EDS analysis of the coating surface showed that the main HA elements (Ca, P, O) were contained on the coating surface (seen in Figure 8). Moreover, the Ca/P of the coating was 1.9, which was slightly higher than that of original HA powder (Ca/P: 1.7). The tetra calcium phosphate phase (TTCP) was the only calcium phosphate phase with Ca/P greater than that of HA. It indicated that the amorphous phases existed on the surface of the coating. The XRD spectrums of the original powders (Ti, HA) and the coatings are shown in Figure 9. The XRD results presented that the phase of the VPS-Ti coating is the pure Ti phase, which avoided the use of traditional APS-sprayed Ti coating to generate impure phases (TiO₂, TiN, etc.), causing cracks in the coating [26]. In addition, the phase composition of the surface coating was mostly the HA phase, and the diffraction peaks of the HA coating were sharper than the original HA powder, indicating that the VPS-sprayed HA coating had higher crystallinity. At the same time, the Ti phase and the tricalcium phosphate phase (TCP) and TTCP phases with weaker diffraction intensity were also found in the spectrum of the HA coating, which confirmed the existence of the amorphous phase. The existence of the Ti phase diffraction peaks in the surface coating was the result of the diffraction of the middle Ti coating due to the thin and uneven thickness of the HA surface coating. There were a small amount of TCP and TTCP amorphous phases in the HA coating. During the plasma spraying process, some particles of the HA powder were more fully melted; the melted HA particle would be rapidly quenched as impacting onto the substrate, which caused the solidified HA particles to incompletely crystallize and form the amorphous phase. According to the reports, HA was extremely unstable and decomposes easily at high temperatures (above 1300 °C), and its chemical reaction equation was shown in the following formula [27]:



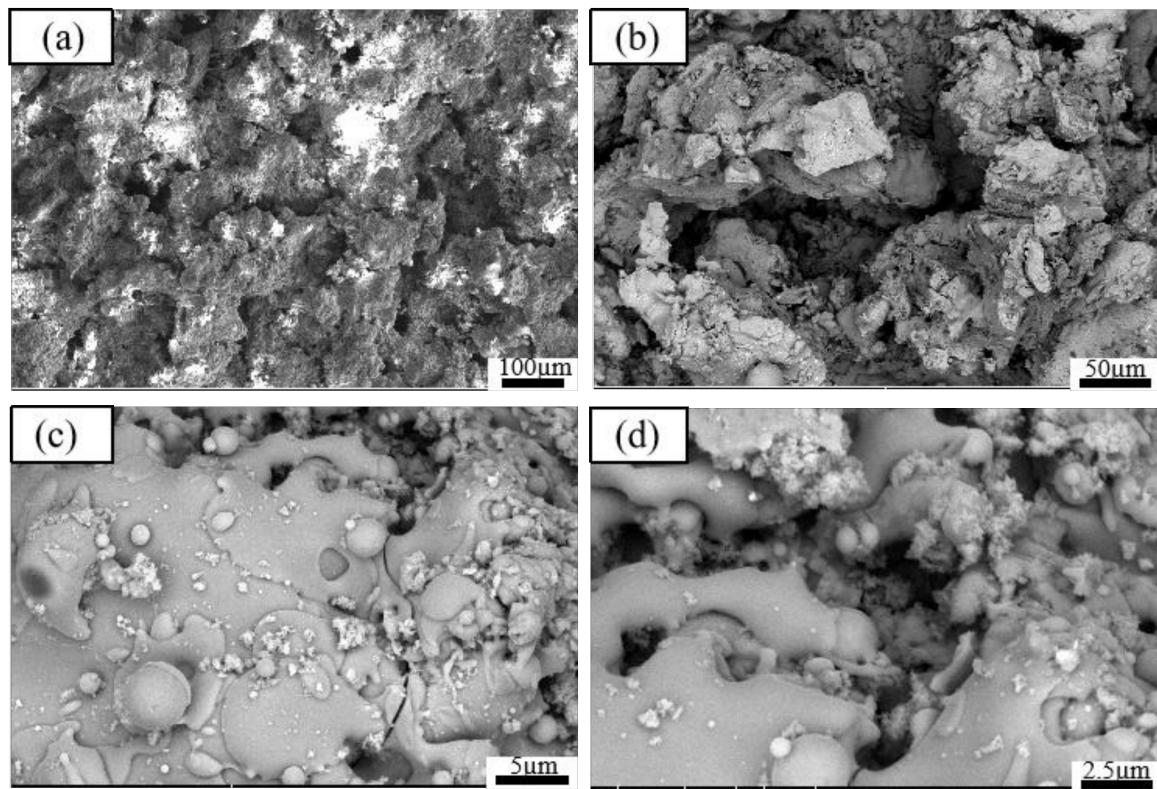


Figure 7. SEM images at different magnifications of the surface of Ti/Ti/HA coating: (a) $\times 200$, (b) $\times 500$, (c) $\times 5000$, (d) $\times 10,000$.

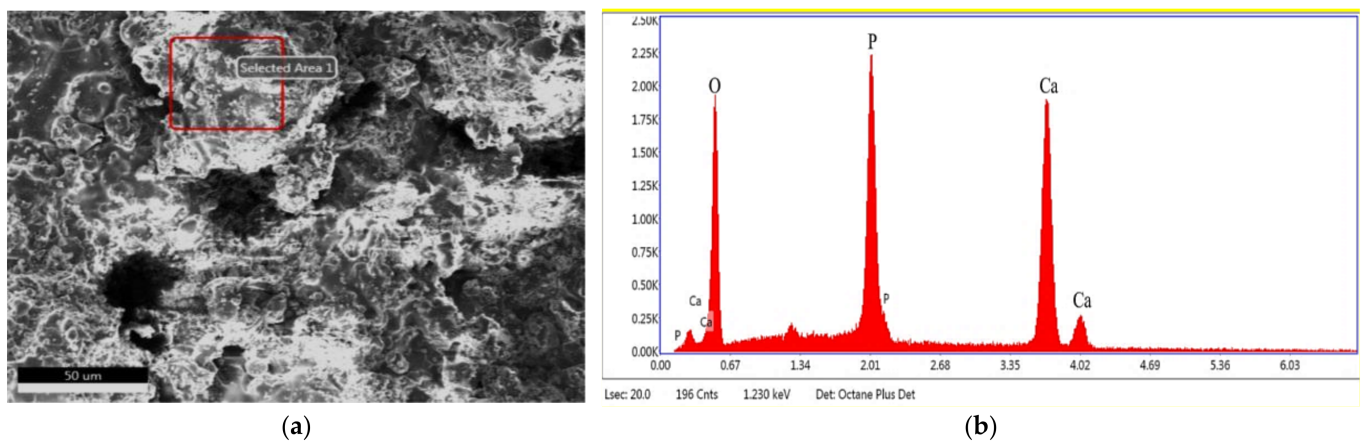


Figure 8. Element distribution of the HA coating surface: (a) EDS scanning region, (b) corresponding element distribution.

The crystallinity of HA affects the stability of the coating in human body fluids. Since the solubility of TCP, TCP, CaO and other amorphous phases in body fluids was much higher than that of HA, the too fast degradation rate could easily lead to early dissolution failure of the coating. On the other hand, the appropriate dissolution of Ca and P in the coating could help the growth of new bone tissue, allowing the implant to form a chemical and biological bond with the bone [28]. Hence, the lower spraying power (<45 KW) and hydrogen flow (<8 L/min) were used in the experiments to obtain a higher crystallinity of HA coating in this study. According to the calculation (1), the crystallinity of HA reached more than 90%. Moreover, the lower spraying power also helped to increase the roughness

of the HA coating, thereby increasing the contact area between the porous surface of the implant and the bone tissue, and improving the stability of the implant.

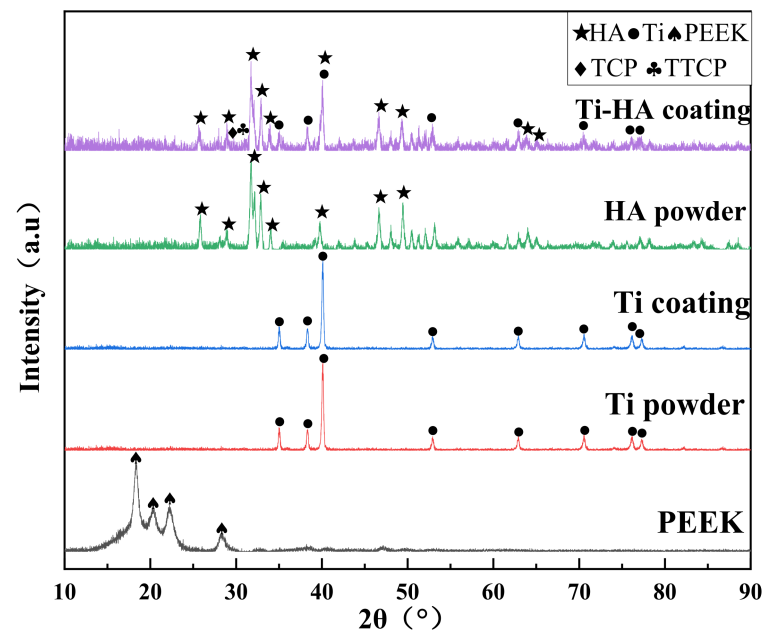


Figure 9. XRD spectrums of the original powder and coatings.

3.3. Influence of the Coating Process on Thermal-Chemical PEEK Properties

3.3.1. ATR FT-IR Spectroscopy

The infrared spectrum of the fingerprint region of coated PEEK are presented in the Figure 10a. The characteristic peaks of the PEEK polymer were clearly visible. The 1649 cm^{-1} represented the stretching vibration band ($\nu_{\text{C=O}}$) of the carbonyl group (C=O); the 1594 , 1487 , 1411 cm^{-1} represented Ar-O-Ar (Ar is an aryl group, the same below) benzene ring plane vibration band ($\nu_{\text{C-O-C}}$); the 1304 cm^{-1} represented the plane vibration band of Ar-CO-Ar benzene ring ($\nu_{\text{C-CO-C}}$); the 1279 , 1218 , and 1186 cm^{-1} represented the asymmetric stretching vibration bands of Ar-O-Ar ($\nu_{\text{asC-O-C}}$); the 1155 and 1010 cm^{-1} represented the C-H in-plane bending vibration bands ($\gamma_{\text{C-H}}$) of the benzene ring in the aryl ether or aryl ketone structure; the Ar-CO-Ar symmetric stretching vibration band ($\nu_{\text{sC-CO-C}}$) was at 925 cm^{-1} ; and the 834 and 764 cm^{-1} was the out-of-plane bending vibration bands of the benzene ring C-H ($\omega_{\text{C-H}}$). These data were obtained from Fourier transform infrared spectroscopy, which were consistent with other relevant literature [29,30]. This indicated that the polymer material used in our study was semi-crystalline polyether ether ketone. The comparison of coated and uncoated PEEK in the infrared spectrum is presented in Figure 10b. The samples showed the characteristic peaks of PEEK, but the relative strength of some peaks was slightly different, which was caused by the uneven thickness of the sample. The samples still met the requirements of ASTM F2026. This indicated that the PEEK did not undergo any chemical changes during the spraying process.

3.3.2. Dynamic Difference Scanning Calorimetry

The dynamic different scanning calorimetry (DSC) from 40 to $400\text{ }^{\circ}\text{C}$ of the uncoated PEEK and coated PEEK were determined, respectively. The results are presented in the Figure 11. The glass transition temperature was around $140\text{ }^{\circ}\text{C}$ and a melting endothermic peak was round $339\text{ }^{\circ}\text{C}$ in Figure 11a,b. These test results are both within the range required by ASTM F2026 standards, as shown in Table 3. This indicated that the plasma spraying technology had no effect on PEEK thermal performance. PEEK has a melting point of $334\text{ }^{\circ}\text{C}$, while plasma spray particles can reach temperatures of several thousand $^{\circ}\text{C}$. However, the PEEK matrix is cooled by compressed air during spraying, and the temperature of the

PEEK matrix is always kept below 200 °C. Therefore, there is almost no influence on the thermal physical properties of PEEK matrix material, and the test results are within the standard range.

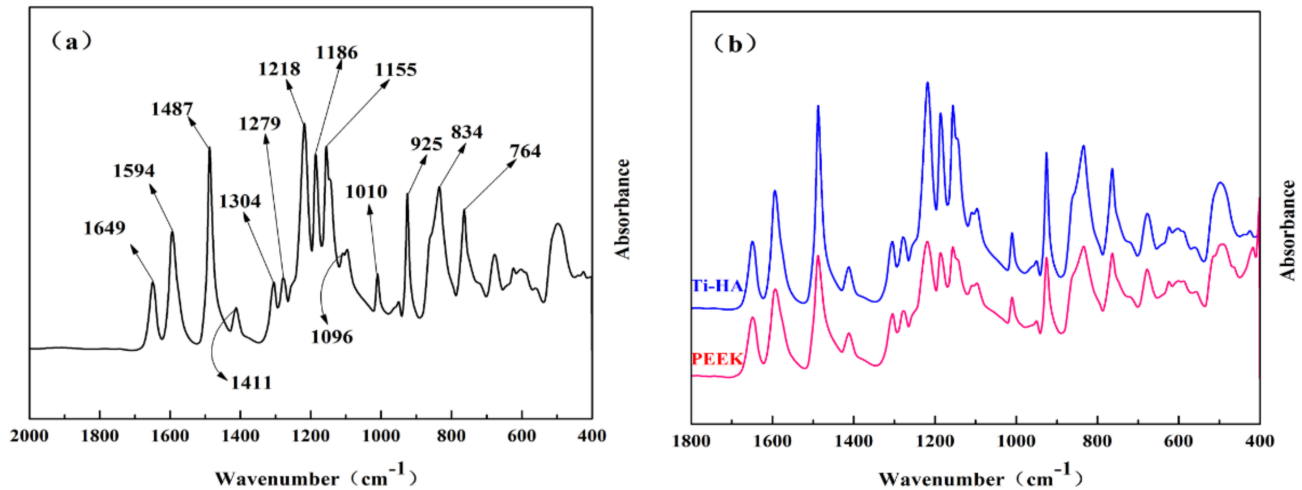


Figure 10. ATR FT-IR spectroscopy graphs of PEEK: (a) the infrared spectrum of the fingerprint region of coated PEEK; (b) the comparison of coated and uncoated PEEK in the infrared spectrum.

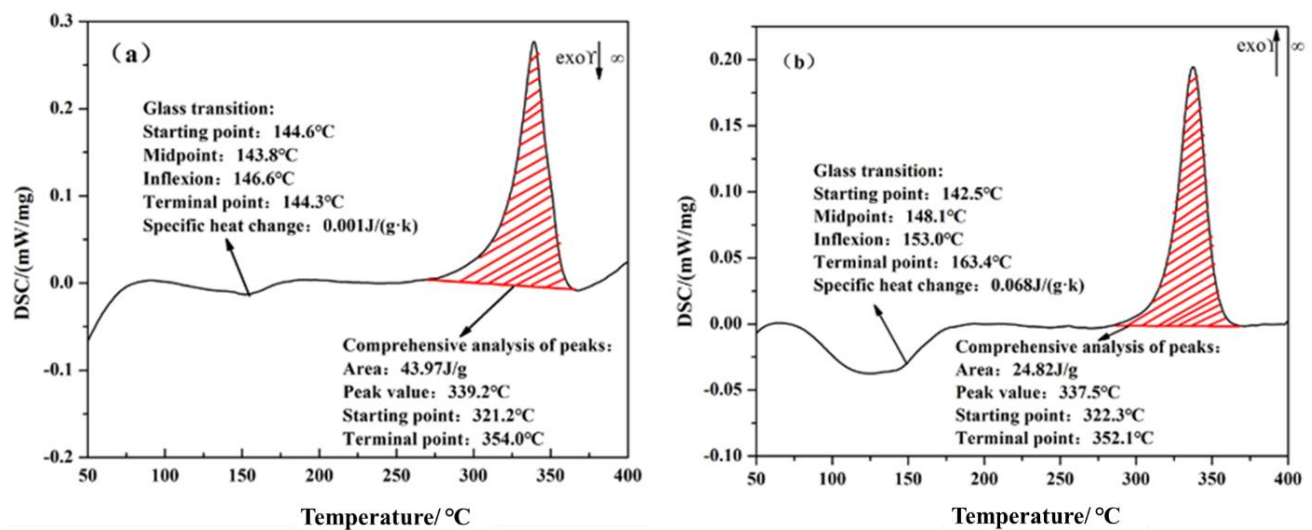


Figure 11. Dynamic difference calorimetry graphs of PEEK: (a) uncoated PEEK; (b) coated PEEK.

Table 3. Results of the dynamic difference calorimetry compared to the requirements of ASTM F2026.

Property	ASTM F2026 Requirement	PEEK (Uncoated)	PEEK (Coated)
Glass transition temperature (°C)	125–165	145	143
Melting temperature (°C)	320–360	339	338

4. Interface Bonding Mechanism of Coatings

The macro fracture morphology of the coating after the bond strength tests are shown in Figure 12a. All samples were observed to be a mixture of adhesion failure and cohesion failure at the fracture interface [31]. It can be seen from the figure that the fracture surface is incomplete, and the area fraction of cohesive failure was far less than the adhesive fracture. Therefore, the bond strength was mainly reflected in the adhesion strength between the

coating and the interface. The overall bond strength of the coating was related to the thickness of the coating, the bonding strength between layers, the porosity of the coating and the properties of the glue used [32]. For example, the excessive glue penetration into the coating would weaken the interlayer structure of the coating, thus strengthening the bond between the coating and the PEEK substrate, which had a great impact on the bond strength of the coating. Therefore, the thinner coatings tended to obtain higher bond strength. In this paper, the coating with a thickness of about 200 μm was prepared to reduce the effects of the thickness and the amount of glue on the bond strength in a certain range. The microscopic diagram of the fracture interface after the bond strength test is shown in Figure 12b. We could infer that there was a strong mechanical interlocking and diffusion connection at the interface of the coating and substrate. However, the middle Ti coating provided more fracture paths of the coating due to its porosity, which had a negative impact on the bond strength of the coating to some extent. On the other hand, its rough surface and curved profile increased the contact area between the layers, and the stronger interfacial bite force made up for the deficiency of the porosity in the bonding strength. As a result, the average bond strength of the Ti/Ti/HA reached 32 MPa, which was more than that of the FDA's clinical requirement (>22 MPa). The bonding strength of the Ti/Ti/HA composite coating is shown in Table 4.

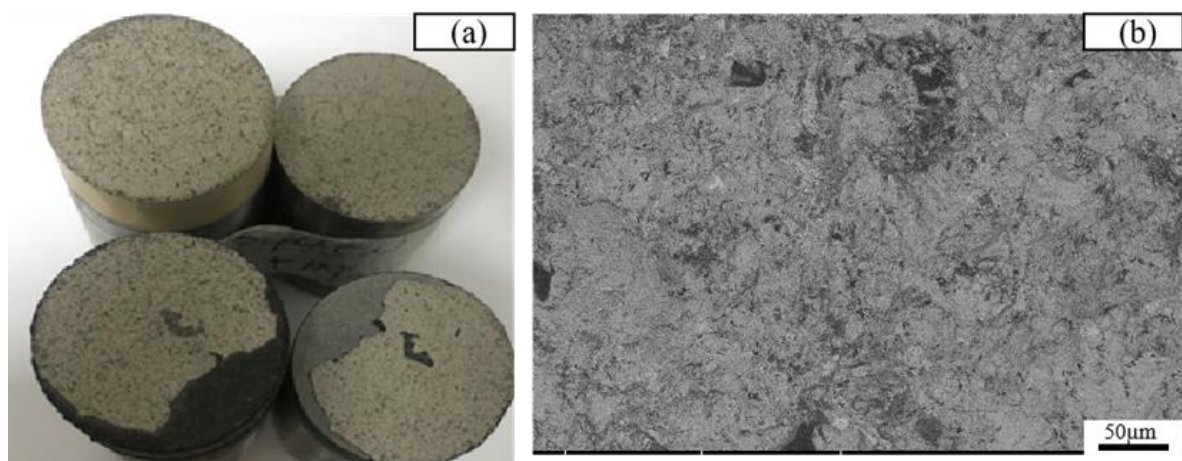


Figure 12. The fracture morphology of the coating after the bonding strength tests: (a) the macro fracture morphology; (b) the microscopic diagram of the fracture interface.

Table 4. Bond strength of the Ti/Ti/HA coating.

Number	1	2	3	4	5	6
Bond strength (MPa)	30	35	32	31	31	35

5. Conclusions

The Ti/Ti/HA composite coatings were deposited on the PEEK by VPS. The coatings presented an obvious three-layer structure: a dense bottom titanium coating with a porosity $< 10\%$; a rough porous intermediate titanium coating with a porosity of $35\% \pm 10\%$, and a roughness of $R_a = 34 \pm 15 \mu\text{m}$; and a top hydroxyapatite coating with a crystallinity greater than 90%. The average bonding strength of the coating reaches 32 MPa, which was greater than that of FDA (22 MPa). In addition, the TG-DSC and AIR-FTIR were performed on PEEK without coating and after coating. The results showed that the plasma spraying process had no effect on PEEK's thermal–chemical performance.

Author Contributions: Conceptualization, F.H. and X.F.; methodology, J.S.; software, X.Y.; validation, M.L., D.Z. and C.D.; formal analysis, C.N.; investigation, M.L.; resources, F.P.; data curation, F.H.; writing—original draft preparation, F.H.; writing—review and editing, X.F.; visualization, F.P.; supervision, M.L.; project administration, X.F.; funding acquisition, C.D. All authors have read and agreed to the published version of the manuscript.

Funding: The authors are grateful for the financial support from the Science Technology Project of Guangdong Academy (2021GDASYL-20210103062), the Guangdong Key Laboratory of Modern Surface Engineering Technology (Grant No 2020B1212060049), the Guangdong Special Support Program (No. 2019 BT02C629), and the Guangzhou Major Projects of Industry-University-Research (IUR) Collaborative Innovation “Surface Treatment and Repair for Key Components of Industrial Gas Turbine (IGT)”.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

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