

# Supplementary materials: One-step Methods to Fabricate Durable Superhydrophobic Coatings for Flexible Electronic Sensors

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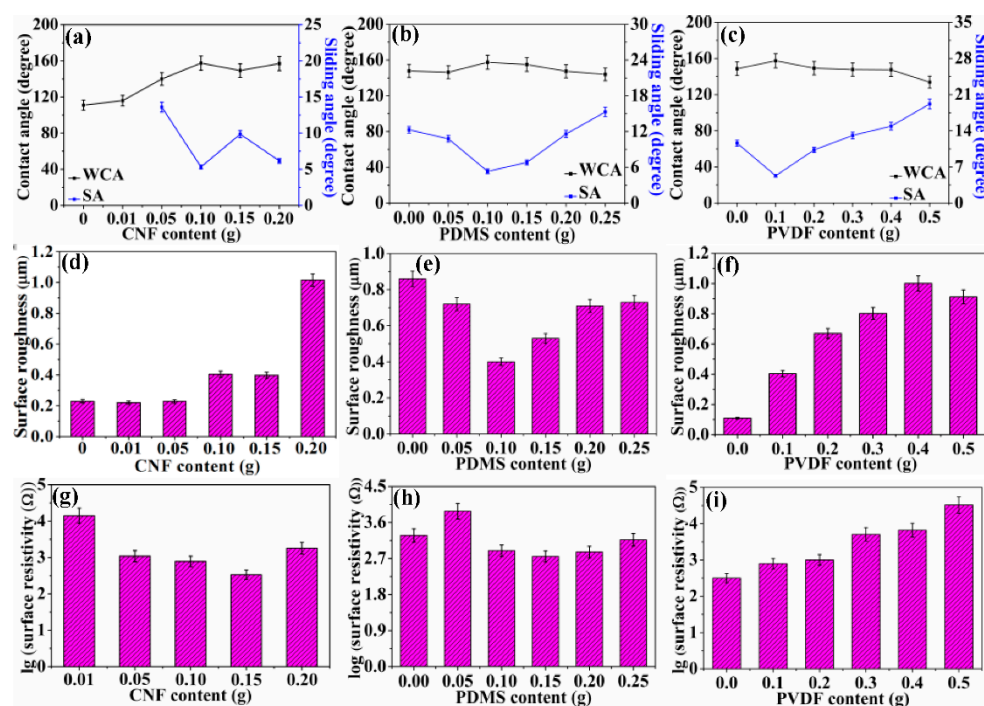
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**Figure S1.** Line charts of the contact angle and sliding angle of the FCS coating at different (a) CNF, (b) PDMS and (c) PVDF contents. Variations in the surface roughness with (d) CNF, (e) PDMS and (f) PVDF contents. Variations in the surface resistivity with (g) CNF, (h) PDMS and (i) PVDF contents.

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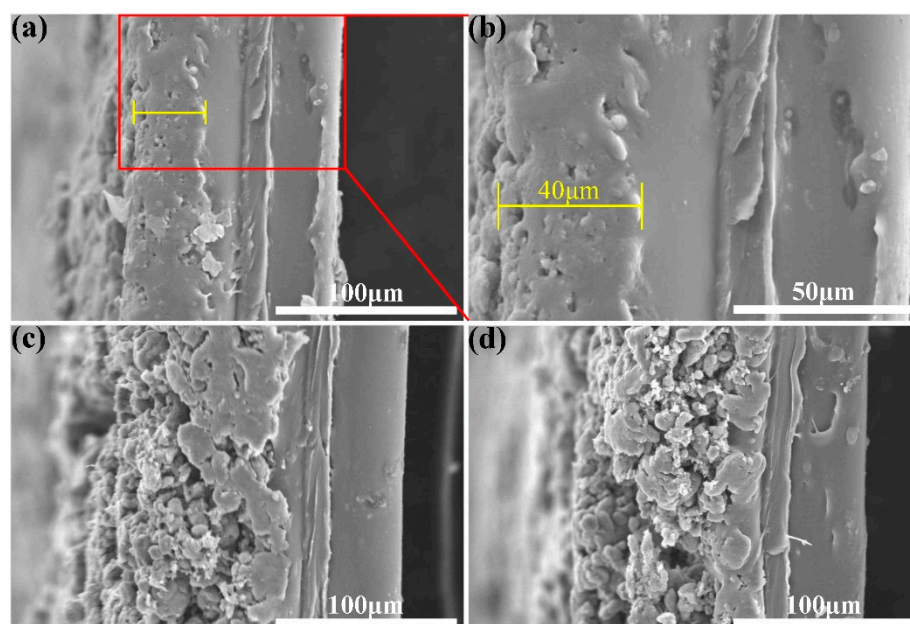
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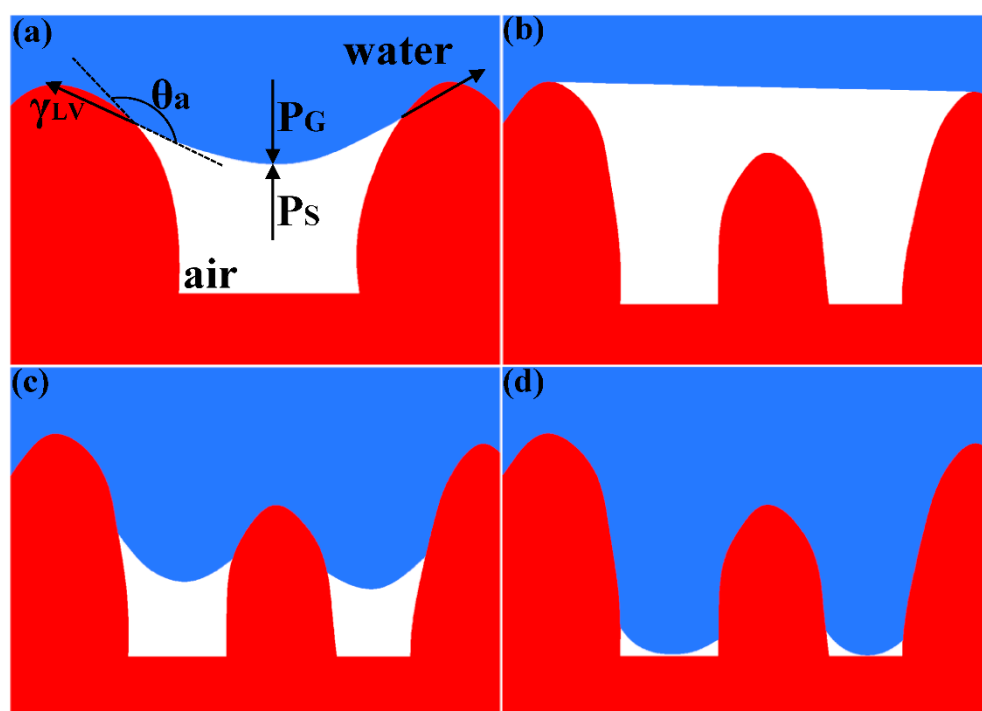
**Table S1.** The effect of CNF content on the surface resistivity.

CNF Content (g)	wt %	Surface Resistivity ( $\Omega$ )	lg (Surface Resistivity)
0.01	4.76	$1.41 \times 10^4$	4.15
0.05	20.00	$1.09 \times 10^3$	3.04
0.10	33.33	$7.84 \times 10^2$	2.89
0.15	42.86	$3.39 \times 10^2$	2.53
0.20	50.00	$1.81 \times 10^3$	3.26

**Table S2.** The contact angle of water and diiodomethane on the polished FCS coating surface. The dispersion force and polarity force of water and Diiodomethane.

Liquid Drop	Contact Angle( $^\circ$ )	Dispersive Force (mN/m)	Polarity Force (mN/m)
Deionized water	106.3	21.8	51.0
Diiodomethane	82.8	49.5	1.3

**Figure S2.** Cross-section morphologies of the FCS coating prepared by brush coating under different magnifications ((a) 500 $\times$  SEM image and (b) 1000 $\times$  SEM image). Cross section morphologies of the FCS coatings obtained by spraying ((c) 500 $\times$  SEM image) and dip-coating ((d) 500 $\times$  SEM image).



**Figure S3.** (a) Force analysis of water droplet in the FCS coating microstructure. Schematic diagram of gas-liquid interface changing with immersion time (b,c).