

Letter

# Growth of Si<sub>3</sub>N<sub>4</sub> Thin Films on Si(111) Surface by RF-N<sub>2</sub> Plasma Nitriding

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**Abstract:** Ultra-thin Si<sub>3</sub>N<sub>4</sub> films were grown on Si(111) surface by radio frequency (RF)-N<sub>2</sub> plasma exposure at 900 °C with 1–1.2 sccm of a flux of atomic nitrogen. We discuss the effect of various conditions such as N<sub>2</sub> flow rate, nitriding time and RF power on the optical, chemical, and structural properties of a nitrided Si<sub>3</sub>N<sub>4</sub> layer. The optical properties, surface morphology and chemical composition are investigated by using ellipsometry, atomic force microscopy (AFM), X-ray photoelectron spectroscopy (XPS), and transmission electron microscopy (TEM). Cross-sectional TEM images show that an RF power of 350 W induced some damage to the Si(111) surface. The thickness of nitrided Si<sub>3</sub>N<sub>4</sub> was measured to be about 5–7 nm. XPS results shown that the binding energy of Si 2p<sub>3/2</sub> located at 101.9 ± 0.1 eV is attributed to the Si–N bonds in the Si<sub>3</sub>N<sub>4</sub> compound. Smooth Si<sub>3</sub>N<sub>4</sub> ultra-thin films were obtained at a nitridation time close to 1 h with an RF power of 300 W, with a measured refractive index (*n*) nearly to 1.88 at 632 nm. The increase in refractive index with decreased RF-plasma power and nitrogen flow rate is probably attributed to the change in the stoichiometry of the film and less surface damage.

**Keywords:** RF plasma; nitridation; Si<sub>3</sub>N<sub>4</sub>



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## 1. Introduction

Silicon nitride (Si<sub>x</sub>N<sub>y</sub>, normally Si<sub>3</sub>N<sub>4</sub>) is a ceramic material that has drawn widespread attention in studies related to microelectronic applications. It can also be used as a coating layer in high-temperature and corrosive environments [1]. Its high-temperature properties are better than most metals, combining high mechanical strength and creep resistance and good oxidation resistance. In addition, Si<sub>3</sub>N<sub>4</sub> has good thermal shock resistance compared with most ceramic materials, attributed to its low thermal expansion coefficient. There are two well-known structures of silicon nitride: α-Si<sub>3</sub>N<sub>4</sub> and β-Si<sub>3</sub>N<sub>4</sub>; both have a hexagonal structure. Among them, α-Si<sub>3</sub>N<sub>4</sub> and β-Si<sub>3</sub>N<sub>4</sub> can be formed under normal pressure, while γ-Si<sub>3</sub>N<sub>4</sub> (cubic spinel lattice) can be formed under a high temperature and pressure [2–4]. In addition, it is used as a wide-band-gap semiconductor (e.g., 4.7 eV) [5,6]. Growth of high-quality silicon nitride is very important due to its wide applications in microelectronics such as etch masks [7], and a buffer layer for III-nitride growth [8]. Recently, the growth of group III–Nitride on Si substrate has been very important for the mass production of high electron mobility transistors (HEMTs) due to the low cost, large surface area, and high thermal conductivity of Si substrates. Therefore, crystalline Si<sub>3</sub>N<sub>4</sub> films are used as a buffer layer for III–nitride growth on silicon substrate [9]. Nakada et al. report that the dislocation density in heteroepitaxial GaN layers can also be reduced by using

a silicon nitride buffer layer [10], which leads to the prospect of silicon-based integrated optoelectronic circuits (Si-OEICs) with silicon-based devices. Epitaxial growth of  $\text{Si}_3\text{N}_4$  film on Si(111) is also theoretically possible because of the good match in lattice parameters between Si(111) and  $\text{Si}_3\text{N}_4(0001)$ . However, the growth of high-quality crystalline  $\text{Si}_3\text{N}_4$  is very difficult due to the low self-diffusion coefficient of nitrogen in  $\text{Si}_3\text{N}_4$  caused by its density. [11]. On the other hand,  $\text{Si}_3\text{N}_4$  films deposited on Si are usually amorphous. However, due to a nearly perfect lattice match, the  $2 \times 2$  unit cell of the Si(111) surface is only  $\sim 1.1\%$  larger than the unit cell of  $\beta\text{-Si}_3\text{N}_4(0001)$  [12]. Recently, various methods have been used to fabricate  $\text{Si}_3\text{N}_4$  films, including thermal chemical vapor deposition (CVD) [13], plasma-enhanced/low-pressure CVD [14], microwave plasma nitriding [15], radio frequency (RF) sputtering [16], and various gas plasma nitridation such as NO [17],  $\text{N}_2$  [18], N [19], or N [20] ion beam at high substrate temperatures. Although CVD and sputtering have often been used in the growth of  $\text{Si}_3\text{N}_4$  thin films, few studies have reported nitrided  $\text{Si}_3\text{N}_4$  grown directly on a Si(111) substrate using RF-metal-organic molecular-beam epitaxy (MOMBE).

Nitridation occurs when a silicon substrate is exposed to gases such as  $\text{NH}_3$  [21], NO [17], and  $\text{N}_2\text{H}_4$  [22], or to N atoms and ions [1]. For nitridation of Si by plasma-assisted molecular-beam epitaxy (MBE)-related systems which can lower the nitridation temperature below  $\text{NH}_3$ -MBE (such as  $700\text{--}900\text{ }^\circ\text{C}$  for RF-MBE [23],  $800\text{--}1100\text{ }^\circ\text{C}$  for  $\text{NH}_3$ -MBE [24]), the control of nitrogen flux is an important issue because the nitridation area and rate are affected by the amount of nitrogen atom flux [25].

Previous work on nitriding Si for the formation of  $\text{Si}_3\text{N}_4$  has been performed with thermal and radio frequency plasma processing using  $\text{NH}_3$ ,  $\text{N}_2$ , and  $\text{H}_2$  [26–28]. Zhai et al. reported that a coherent  $\text{Si}_3\text{N}_4(0001)/\text{Si}(111)$  interface can be formed using a thermal nitridation process, where the Si substrate is exposed to nitrogen-containing gases such as  $\text{NH}_3$  at high substrate temperatures [29]. Due to its inertness, growth of  $\text{Si}_3\text{N}_4$  by nitriding to  $\text{N}_2$  requires relatively high pressures, and it is highly sensitive to contamination in the supplied gas. In previous reports, Ermolieff et al. and Bolmont et al. reported that the nitriding process of Si(111) [30] and Si(100) [31] used a flux in atomic N at low temperature [32].

Optoelectronic devices of III-nitride on Si substrate are very attractive due to the cost-effective device fabrication on the platform of Si-based integrated circuits [33]. However, the difference in the thermal expansion coefficients and lattice mismatch between nitride and Si can lead to cracking of III-N epitaxial layers [34].

Silicon-based materials have the advantages of compatibility with current silicon technology and lower cost compared to compound materials. On the other hand, the growth mechanism of silicon nitride films on silicon and their morphology, including the surface atomic structures, are important to these applications. Our current knowledge of these fundamental properties, however, is quite limited. Therefore, in the present work, we explore the effect of RF-plasma nitriding on crystalline  $\text{Si}_3\text{N}_4$  formation on nitrided Si(111) substrate. The substrate was nitrided at  $900\text{ }^\circ\text{C}$  for 60 min at various nitridation conditions. The effects of nitridation conditions on the structural and optical properties and surface morphology of the  $\text{Si}_3\text{N}_4$  films were then investigated. The effectiveness of this method is evaluated using ellipsometry, atomic force microscopy (AFM), transmission electron microscopy (TEM), and XPS.

## 2. Materials and Methods

Crystalline  $\text{Si}_3\text{N}_4$  ultra-thin films were grown on a Si(111) surface using RF- $\text{N}_2$  plasma exposure at various RF powers and  $\text{N}_2$  flow rates and a fixed nitridation time. The nitridation process was performed in the growth chamber of self-designed UHV-RF-MBE (home-made). The active nitrogen ( $\text{N} + \text{N}^*$ ) radicals were generated by a 13.56 MHz RF-plasma source with a high pure  $\text{N}_2$  (99.9999%) gas. The growth chamber was vacuumed by a turbo molecular pump to reach a base pressure of  $1 \times 10^{-9}$  Torr. The substrate were  $20 \times 20\text{ mm}^2$  slabs cut from Si(111) wafer. The silicon wafers were cleaned by Radio Corporation of America (RCA) clean and then dipped in Buffered Oxide Etch (BOE) solvent

for 10 min to remove the native oxide of silicon surface. Prior to the nitridation process, the cleaned silicon substrate was placed in an ultra-high-vacuum chamber (base pressure  $\sim 1 \times 10^{-9}$  Torr) and thermally annealed at 900 °C for more than 120 min to remove the remaining thin native oxide. After this, the substrate was exposed to a flux of atomic nitrogen generated by a RF-plasma source for 60 min with RF  $\sim 300, 350$  W and  $N_2 \sim 1, 1.2$  sccm. During the silicon nitrided growth, the  $N_2$  gas flow and the reactor pressure were maintained at a constant. The work pressure corresponded to  $1 \times 10^{-5}$  Torr and  $1.2 \times 10^{-5}$  Torr with  $N_2$  flow rate at 1 and 1.2 sccm, resistively. The detailed experimental conditions of the nitrided  $Si_3N_4$  layer were listed in Table 1. Various properties of nitrided  $Si_3N_4$  ultra-thin films were measured by ellipsometry (SENTECH Instruments GmbH, Berlin, Germany), AFM, and TEM analysis. The chemical evolution of the nitridation layer has been studied by x-ray photoelectron spectroscopy (XPS; VG ESCA/XPS Theta Probe; Thermo Fisher Scientific Inc., Waltham, MA, USA). The XPS experiments were performed in a VG ESCA Scientific Theta Probe using Al  $K\alpha$  ( $h\nu = 1486.6$  eV) radiation to characterize the bonding characteristics of elements in the films. The surface morphologies and cross-section microstructure of the films were analyzed using an atomic force microscope (AFM, Dimension 3100 SPM, Digital Instruments, Tonawanda, NY, USA) with the tapping mode and a transmission electron microscope (TEM, Philips Tecnai 20 Amsterdam, The Netherlands).

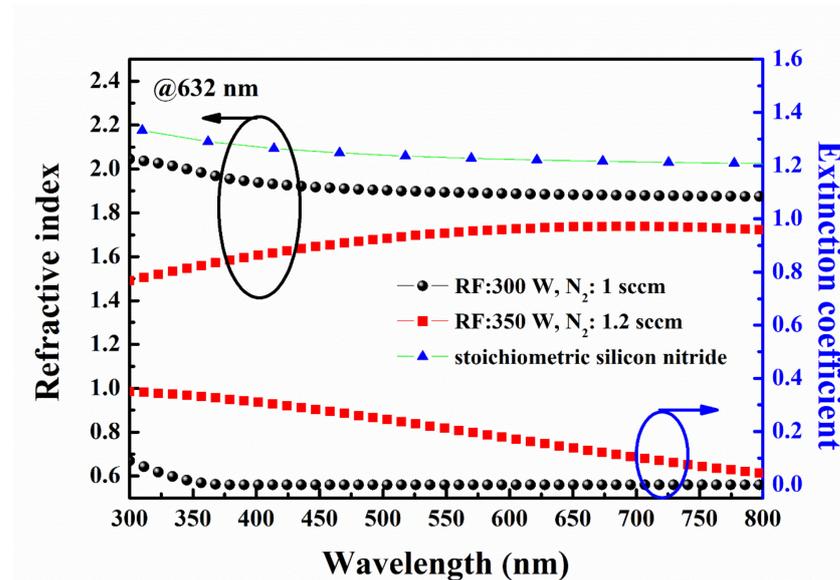
**Table 1.** Experimental parameters for nitrided  $Si_3N_4$  layer.

Sample Name	Sample A	Sample B
RF power	300 W	350 W
$N_2$ flow rate	1 sccm	1.2 sccm
Nitridation time	60 min	60 min

### 3. Results and Discussion

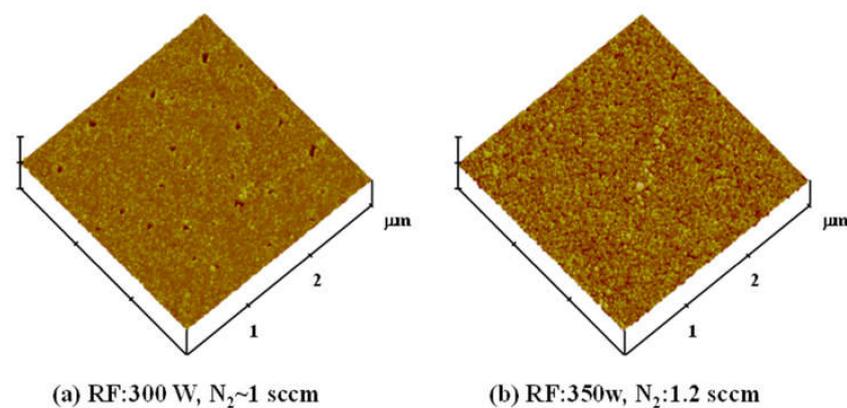
The thickness and refractive index of all the nitrided  $Si_3N_4$  layers were characterized by spectroscopic ellipsometry. The refractive index of  $Si_3N_4$  layer is fitted by the Tauc-Lorentz models at wavelengths ranging from 300 to 800 nm, using an angle of incidence of 75° and an illumination spot diameter of 1.0 mm. The measured dispersion of the refractive index and extinction coefficient of silicon nitride at 632 nm are shown in Figure 1. The data for stoichiometric silicon nitride [35] are also shown for comparison. In the visible wavelength range, silicon nitride film thickness and refractive index were determined by ellipsometry measurements of  $5 (\pm 0.1)$  nm and  $1.88 (\pm 0.10)$  for sample A, and  $7 (\pm 0.1)$  nm and  $1.73 (\pm 0.10)$  for sample B, respectively. The surface roughness of the samples was obtained using the AFM, and the value of surface roughness is 15%–20% of the thickness. Therefore, the surface roughness may impact thickness analysis in fitting result. The refractive-index values are smaller than that the corresponding average values for stoichiometric silicon nitride film (refractive index  $\sim 2.02$  at 300 K) [36]. The value of the fitted film thickness closely matches the data obtained from the cross-section TEM image. For sample A (RF: 300 W with  $N_2$ : 1 sccm), it can be seen that the silicon nitride is characterized by slightly weaker variation with wavelength and smaller refractive index. This is due to the fact that the surface morphology shows some pits on the  $Si_xN_y$  thin film surface, possibly attributable to plasma damage. For RF power at 300 W with  $N_2 \sim 1$  sccm, the value of the refractive index at 632 nm of nitrided  $Si_xN_y$  is  $\sim 7\%$  smaller than that of stoichiometric silicon nitride. The result indicated that the decrease may be caused by the physical and chemical structural modifications induced in the  $Si_xN_y$  by the nitridation process. However, the measured shrinkage was small, which clearly indicates that the physical modifications (such as pits) within the structure of the nitrided  $Si_xN_y$  are the main cause for the refractive-index variation. In addition, for RF power at 350 W with  $N_2 \sim 1.2$  sccm, the  $Si_xN_y$  thin film shows large variation, probably due to nonstoichiometric Si–N–O compounds ( $n = 1.457\text{--}1.909$  at 632.8 nm) and/or high-density pits. In addition, extinction coefficients of sample A are smaller than  $1 \times 10^{-4}$ . However, sample B shows

that the extinction coefficients were nearly 0.15, which is probably due to nonstoichiometric Si–N–O compounds.



**Figure 1.** Ellipsometer spectra of  $\text{Si}_x\text{N}_y$  ultra-thin films. For a nitridation time of 60 min, the refractive index ( $n$ ) of  $\text{Si}_x\text{N}_y$  film is close to 1.88 at 632 nm with RF: 300 W and  $\text{N}_2$ : ~1 sccm.

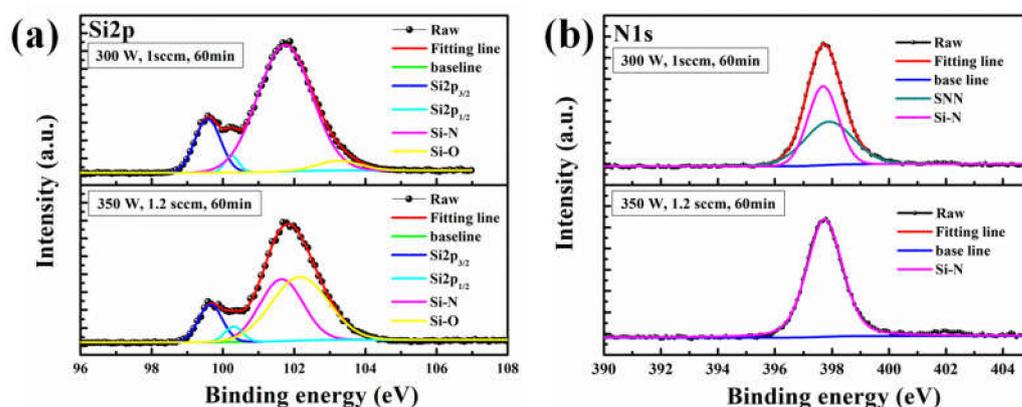
Figure 2 shows AFM images of the surfaces of two samples. The root mean square (RMS) surface roughness values were measured from  $3 \times 3 \mu\text{m}^2$  AFM scans. The RMS roughness of nitrided  $\text{Si}_x\text{N}_y$  was found to increase with increasing RF power and nitrogen flow from 0.891 nm for RF: 300 W with 1 sccm of  $\text{N}_2$  flow rate to 1.089 nm for RF: 350 W with 1.2 sccm of  $\text{N}_2$  flow rate. The AFM image of sample A (RF: 300 W with  $\text{N}_2$ : 1 sccm) is shown in Figure 2a. The image shows irregular shape surface pits with diameters in the range 20–30 nm, which suggests that the plasma nitriding process induced damage to the surface of the samples. AFM images of the surface of sample B (RF: 350 W with  $\text{N}_2$ : 1.2 sccm) are also shown in Figure 2b. The image exhibits rough surface morphology with the pits of high density. Surface pit density was measured by AFM image analysis with ImageJ software. Its analysis resulted in a count of 45 and 835 pits, leading to a density of  $3 \times 10^{10} \text{ cm}^{-2}$  and  $7.6 \times 10^{12} \text{ cm}^{-2}$  for sample A and sample B, respectively.



**Figure 2.** (a) Surface morphologies, refractive index, and extinction coefficient of nitrided  $\text{Si}_x\text{N}_y$  grown on Si(111) surface for 60 min with (a) RF: 300 W,  $\text{N}_2$ : ~1 sccm, and (b) RF: 350 W,  $\text{N}_2$ : ~1.2 sccm.

Two elemental species, Si, and N, were identified in the films by their corresponding signals observed in the XPS spectra. Figure 3 shows two samples of the Si 2p and N 1s

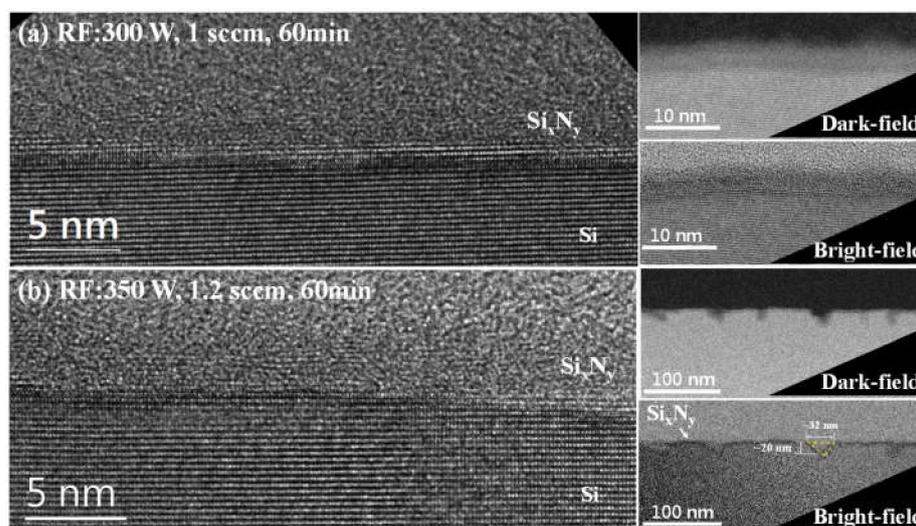
spectra for the Si–N films prepared at various conditions. A Gaussian–Lorentzian sum function was used in the fitting, and the results are shown as red solid lines. The spectrum shows two main peaks at 99.5 and 101.78 eV. In order to investigate the Si-related peak in more detail, measurement and fitting were analyzed in a high-resolution spectrum. The peak in Si 2*p* and N 1*s* exhibit a slightly asymmetric shape, suggesting that they have several bonds other than Si–N in the Si<sub>*x*</sub>N<sub>*y*</sub> layer. The asymmetric peaks were fitted by using a linear combination of Gaussian and Lorentzian line shapes commonly referred to as a pseudo-Voigt function [37]. The peak can be deconvoluted to three components (Figure 3a), which correspond to silicon nitride, silicon oxide, and silicon [38]. XPS spectra results indicated that the binding energy of Si 2*p*<sub>3/2</sub> and Si 2*p*<sub>1/2</sub> spin-orbit doublets located at around at 99.48 and 100.03 eV ± 0.1 eV and equal to 0.63 eV are characteristic of the Si–Si bonds. For the sample A (RF: 300 W with N<sub>2</sub>: 1 sccm), the spectra corresponding to the surface of Si show a strong peak at 101.7 eV, which is assigned to the binding energy of Si–N in the nitride compounds. Additionally, the weak broad peak centered at 103.18 ± 0.1 eV was attributed to Si–O–C and/or the silicon oxide band [39]. In previous reports, the N 1*s* peak for bulk Si<sub>3</sub>N<sub>4</sub> was observed at 397.88 eV where the second nearest neighbor (SNN) was nitrogen [40], and N [Si(Si)<sub>3</sub>]<sub>3</sub> was observed at 397.5 ± 0.2 eV with the Si atoms being SNNs to nitrogen [41]. On the other hand, thick Si<sub>*x*</sub>N<sub>*y*</sub> (~7 nm) obtained for 60 min of nitridation with 350 W of RF power suggests that the diffusion length for the nitridant species across the nitrided Si and/or an amorphous SiO<sub>2</sub> region was greater than the Si<sub>*x*</sub>N<sub>*y*</sub> obtained. For the binding energies of Si and N in sample B, as shown in Figure 3a,b, the nitrided film can be clearly identified as SiNO. These results indicate that a higher process temperature offers more reactions between N radicals of the plasma and the Si atoms on the surface. The result is similar to that reported by Wu et al. [42].



**Figure 3.** (a) Si 2*p* spectra and (b) N 1*s* spectra of binding energy of nitrided Si<sub>*x*</sub>N<sub>*y*</sub> grown on Si(111) surface for 60 min with RF: 300 W, N<sub>2</sub>: ~1 sccm, and RF: 350 W, N<sub>2</sub>: ~1.2 sccm.

The microstructures of the nitrided Si<sub>*x*</sub>N<sub>*y*</sub> were characterized by TEM. The cross-sectional TEM (XTEM) specimens were prepared from the grown layers. Figure 4 shows bright-field TEM images of nitrided Si<sub>*x*</sub>N<sub>*y*</sub> grown on the Si(111) substrate under various conditions. As estimated from cross-sectional TEM images, the thickness of the all nitrided Si<sub>*x*</sub>N<sub>*y*</sub> thin films were approximately ~5 nm for sample A (RF: 300 W with N<sub>2</sub>: 1 sccm), and ~7 nm for sample B (RF: 350 W with N<sub>2</sub>: 1.2 sccm), respectively. The results are in agreement with the ellipsometry measured values. For sample A, the TEM images showed that the nitrided Si<sub>*x*</sub>N<sub>*y*</sub> films were grown continuously by plasma nitriding. In addition, when RF power was increased to 350 W, the surface morphology exhibited rough features, in agreement with the AFM results. Further, bright- and dark-field images revealed that non-continuous V-like pits formed in both film and substrate regions, indicating that the result is probably caused by plasma damage to the nonstoichiometric silicon nitride layer [43], as shown in Figure 4b. The typical depth and width of the V-like pits was measured to be approximately 20 and 24 nm in average, respectively. Nevertheless, the structural and

morphological characterizations imply that the substrate nitriding with a lower RF power and  $N_2$  flow rate is always beneficial for the crystal quality of the epitaxial  $Si_3N_4$  layer on silicon.



**Figure 4.** Bright field XTEM images of nitrided  $Si_xN_y/Si(111)$  surface for 60 min with (a) RF: 300 W,  $N_2$ : ~1 sccm, and (b) RF: 350 W,  $N_2$ : ~1.2 sccm.

#### 4. Conclusions

We fabricated crystalline Si–N ultra-thin layer nitrided Si(111) substrates via RF- $N_2$  plasma and studied the effect of nitridation conditions on the surface, structural, morphological, and compositional properties of the Si–N ultra-thin layer. RF-plasma exposure of Si(111) substrate to a flux of atomic N is suitable for growing ultra-thin layers of silicon nitride. XPS analysis indicated that silicon nitride was formed in the nitrided Si and/or an amorphous  $SiO_2$ . In order to investigate this in more detail, nitriding of the Si  $2p$  peak from sample B exhibits a slightly larger broadening peak than that sample A, suggesting that there may be two different bonding in the nitrided film, where the major one is Si–N (101.7 eV) and the others are minor Si–O–C bonds (103.18 eV). AFM images show that surface of the nitrided Si(111) substrate was rough, likely as a result of damage from the RF-plasma beam. TEM results show that the thickness of silicon nitride formed via nitridation of the Si(111) substrate can be tuned from the ultra-thin (~5 nm) to a thin regime (~7 nm) by varying the nitrogen flow and RF power from 1 to 1.2 sccm and 300 to 350 W, respectively. The XTEM image shows that silicon nitride was nano-crystalline on the surface of Si(111) due to plasma nitriding. These results indicate that the nitrided Si(111) substrate is essential for engineering the growth of nitride on Si wafers.

**Author Contributions:** W.-C.C. carried out most of the experimental work, including the preparation and characterization of  $Si_3N_4$  layer and the drafting of the manuscript. T.-Y.Y. carried out the high-resolution XPS measurements. S.C., J.S., Y.-W.L., H.-P.C. and C.-P.C. supported the analysis of the Si–N samples. All authors have read and agreed to the published version of the manuscript.

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**Conflicts of Interest:** The authors declare no conflict of interest.

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