



## Supplementary Material An All-Glass Microfluidic Network with Integrated Amorphous Silicon Photosensors for on-Chip Monitoring of Enzymatic Biochemical Assay

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## 1. Materials

All reagents were purchased from Aldrich Chemicals. 2-Hydroxyethyl methacrylate (HEMA) was distilled prior to use, whereas the other chemicals were used without further purification. 2-Bromo-2-methyl-propionic acid 3-trichlorosylanyl-propyl ester was synthesized following a reported procedure [1,2]. Ethanol (analytical reagent grade) was used without further purification, while toluene was distilled over sodium. Water was purified with a Milli-Q pulse (MILLIPORE, R 1/4 18.2 MU cm) ultra-pure water system.

## 2. Functionalization of the Microfluidic Channel

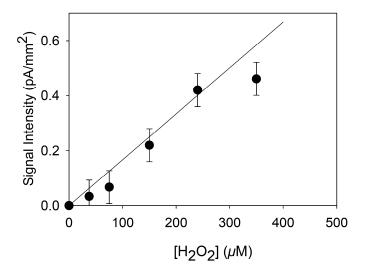
Afterwards, a solution of 0.2% 2-bromo-2-methyl-propionic acid 3-trichlorosylanyl-propyl ester in toluene was flowed through the channel (one night; 0.1 mL/min) to form a self assembled monolayer [3]. Subsequently, the channel was rinsed with toluene and dried with a stream of nitrogen. A solution of 10 mL 2-hydroxyethyl methacrylate (HEMA) and 10 mL of MilliQ water was degassed by bubbling with dry nitrogen (N<sub>2</sub>) for 30 min and transferred in a schlenk tube in which it was stored under an argon environment. Copper(I) chloride (0.055 g), copper(II) bromide (0.036 g) and 2.2'-dipyridyl (0.244 g) were added. To dissolve the solid, the mixture was stirred for 10 min (while degassing), which yielded a dark brown solution. The solution was sonicated till complete dissolution of the solid, then sucked into a Hamilton syringe and inserted in the microfluidic chip in which it was kept inside under no flow conditions (for 60 min) in order to form the PHEMA brush film. The channel was rinsed with ethanol and water (10 min; 1  $\mu$ L/min) to completely remove the copper residues, and dried with a nitrogen flux. A tetrahydrofurane (THF) solution (2 mL) of succinic anhydride (100 mg) and triethylamine (100 mL) was flowed through the chip (24 h; 0.1  $\mu$ L /min), followed by rinsing with ethanol (10 min; 1  $\mu$ L /min). After this step, a water solution

(1 mL) of N-hydroxysuccinimide (NHS) (15 mg) and 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC) (75 mg) was flowed in the channel (0.1  $\mu$ L/min; 1 h), followed by rinsing with purified water. After drying the channel with a stream of nitrogen, a phosphate buffer solution of 10 mM and pH=7.5 (this buffer was prepared by diluting a 25 mM phosphate buffer solution pH 7.5, NaH<sub>2</sub>PO<sub>4</sub> 5 mM; Na<sub>2</sub>HPO<sub>4</sub> 20 mM) containing HRP (0.5 mg/mL) was flushed through the chip and kept inside for one night at 4 °C in order to immobilize HRP onto the activated PHEMA-brushes formed onto the channel. Finally, the channel was rinsed with Blocking buffer (Tris 10 mM pH=8) for 30 min (1  $\mu$ L/min), in order to remove all enzyme molecules that did not link covalently to the polymer film and to block the non-reacted NHS-ester moieties.

[H2O2] μΜ	Flow Rate (µL/min)	Sensors	Time (s)	Intensity (pA/mm <sup>2</sup> )
37.5	70	1-2	0.1357	0.035
	50	1-2	0.23	0.025
	25	1-2	0.56	0.028
	70	5-6	0.6071	0.0248
	50	5-6	0.89	8.33E-03
	25	5-6	1.88	9.78E-03
75	70	1-2	0.1357	0.06
	50	1-2	0.23	0.0875
	70	3-4	0.3714	0.06
	50	3-4	0.56	0.075
	50	5-6	0.89	0.044
	50	7-8	1.22	0.033
	25	5-6	1.88	0.037
	25	7-8	2.54	0.0311
150	70	1-2	0.1357	0.15
	50	1-2	0.23	0.2
	50	3-4	0.56	0.2
	50	5-6	0.89	0.176
	50	7-8	1.22	0.194
	25	5-6	1.88	0.1417
	25	7-8	2.54	0.0933
225	70	1-2	0.1357	0.225
	50	1-2	0.23	0.3
	70	3-4	0.3714	0.426
	70	5-6	0.6071	0.432
	50	5-6	0.89	0.4
	50	7-8	1.22	0.32
	25	5-6	1.88	0.2833
	25	7-8	2.54	0.2194

**Table S1.** Data of the kinetic plot of Figure 7. In the table are reported also the flow rates and the sensors used for the measurements. The intensity values reported come from the average of the intensities of the indicated sensors normalized for the sensor area (error based on standard deviation of three measurements using different devices is ca. 10% for all the values).

350	70	1-2	0.1357	0.2315	
	50	1-2	0.23	0.3512	
	70	3-4	0.3714	0.4418	
	70	5-6	0.6071	0.4832	
	50	5-6	0.89	0.4356	
	50	7-8	1.22	0.3941	
	25	5-6	1.88	0.2946	
	25	7-8	2.54	0.2153	



**Figure S1.** Calibration curve extended to higher  $H_2O_2$  concentrations showing a saturation profile at concentrations higher than 250  $\mu$ M. This trend is probably due to the limited amount of enzyme immobilized in the surface of the microchannel network.

## References

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