

Supplementary Information

Nano Hard Carbon Anodes for Sodium-Ion Batteries

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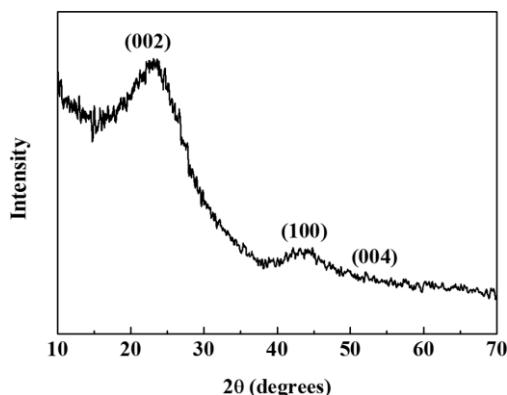


Figure S1. X-ray diffraction patterns of SCB.

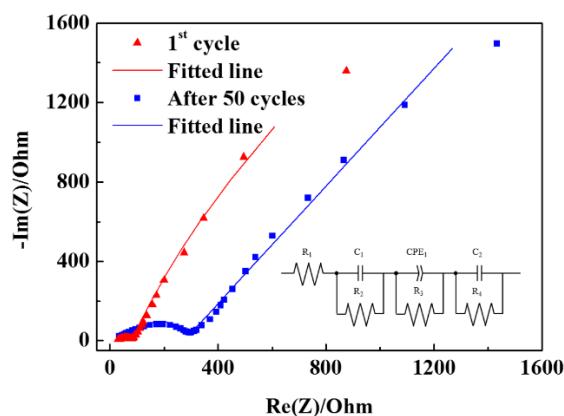


Figure S2. Nyquist plots of SCB (in a 1:1 (v/v) mixture of ethylene carbonate and dimethyl carbonate) after 1st and 50th cycles.

Table S1. The fitting values for the elements of 1st cycle and after 50 cycles.

	$R_1 (\Omega \text{ cm}^{-2})$	$C_1 (\text{F cm}^{-2})$	$R_2 (\Omega \text{ cm}^{-2})$	$\text{CPE}_1 (\text{S}^{1/2} \text{ cm}^{-2})$	$R_3 (\Omega \text{ cm}^{-2})$	$C_2 (\text{F cm}^{-2})$	$R_4 (\Omega \text{ cm}^{-2})$
1st cycle	24.7	1.193×10^{-1}	55.45	3.804×10^{-4}	66.45	1.109×10^{-2}	2,672
After 50 cycles	16.05	5.161×10^{-3}	241	7.593×10^{-7}	332.7	8.012×10^{-3}	3,285

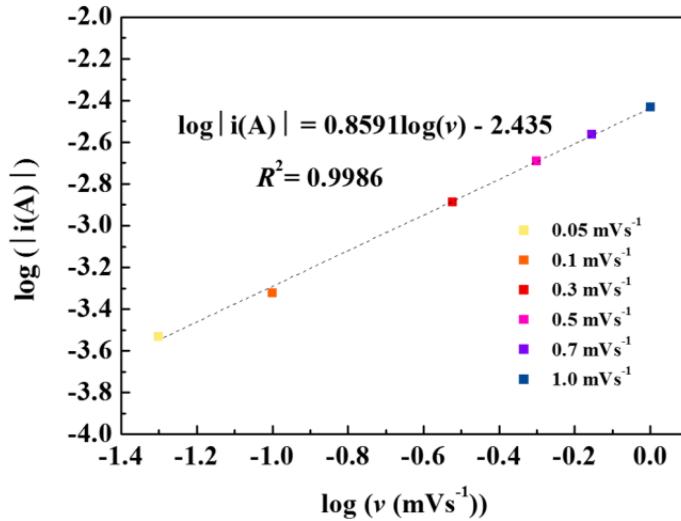


Figure S3. Determination of cathodic peak current values.

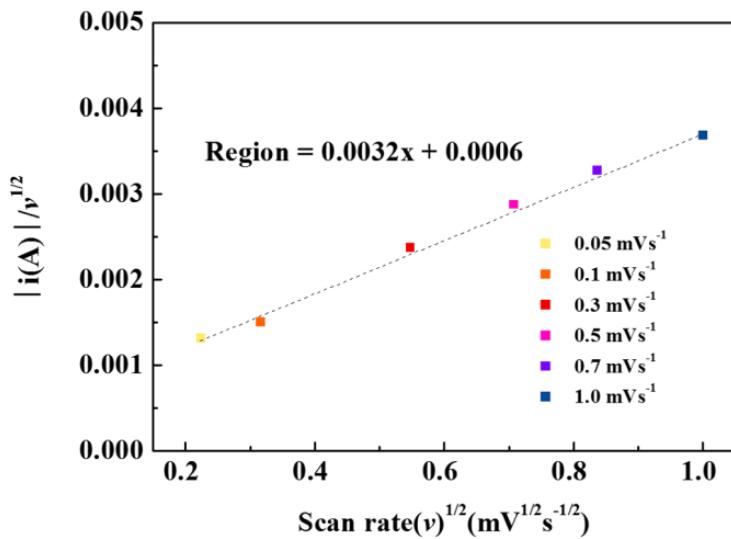


Figure S4. Cathodic peak current dependence on scanning rate, which is used to determine capacitive and intercalation contributions to energy storage.