## **Supplementary Information**

## The Fluorescent Quenching Mechanism of N and S Co-Doped Graphene Quantum Dots with Fe<sup>3+</sup> and Hg<sup>2+</sup> Ions and Their Application as a Novel Fluorescent Sensor

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	Mate	rials	Atomic Populations (Mulliken)				Bond Population		d (Å)
		S	р	d	Total	Charge (e)			
Fe <sup>3+</sup>	C4	1.20	2.98		4.18	-0.18	C1-0	0.48	1.43
	С9	1.09	3.02		4.12	-0.12	C1-S	0.53	1.76
	C10	1.26	2.92		4.18	-0.18	C2-S	0.64	1.70
	Ν	1.40	3.84		5.23	-0.24	C2-N	0.88	1.39
	0	1.81	4.60		6.42	-0.42	C3-N	0.83	141
	S	1.63	3.60		5.22	0.78	C3-O	-0.05	2.63
	Fe	0.28	-0.10	7.16	7.34	0.66	Fe-O	-0.12	2.64
							Fe-S	0.17	2.033
							Fe-N	-0.17	2.80
Hg <sup>2+</sup>	C4	1.36	2.83		4.19	-0.19	C1-0	0.32	1.57
	C9	1.07	3.07		4.14	-0.14	C1-S	0.43	1.76
	C10	1.03	2.75		3.77	0.23	C2-S	0.59	1.71
	Ν	1.39	3.87		5.25	-0.25	C2-N	0.86	1.39
	0	1.73	4.55		6.28	-0.28	C3-N	0.91	1.41
	S	1.70	3.50		5.20	0.80	C3-O	0.50	1.51
	Hg	1.87	0.01	10.0	11.87	0.13	Hg-O	-0.07	3.37
	-						Hg-S	-0.06	3.85
							Hg-N	-0.13	3.24

**Table 1.** The atomic populations of  $Fe^{3+}@N$ , S-GQDs and  $Hg^{2+}@N$ ,S-GQDs.

Sample	Spiked concentration	<b>Detected concentration</b>	Recovery ± RSD (%)
Fe <sup>3+</sup> in Drinking water	700 nM	654 nM	93 ±4.1
-	1 μM	1.27 μM	$127 \pm 5.3$
	3 μM	3.63 µM	121 ±6.6
Hg <sup>2+</sup> in Drinking water	50 nM	62 nM	$124 \pm 3.4$
2 0	100 nM	114 nM	$114 \pm 4.6$
	300 nM	284 nM	95 ±7.6

**Table S2** Recovery of  $Fe^{3+}$  and  $Hg^{2+}$  detection in drinking water samples.



Figure S1. The schematic diagram of detection device geometry and testing process.



Figure S2. The stability of fluorescence intensity of as-synthesized N, S-GQDs solutions.



Figure S3. The fluorescence intensity of N, S-GQDs in real sample detection.