

Short-Chain Modified SiO₂ with High Absorption of Organic PCM for Thermal Protection

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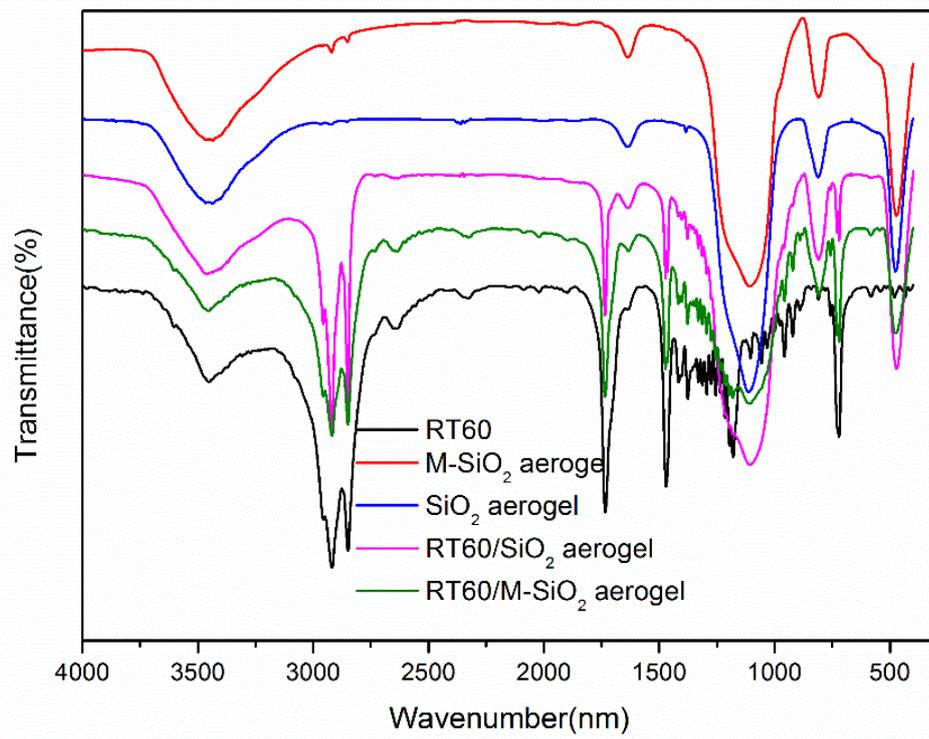


Figure S1. FT-IR spectrum of all samples.

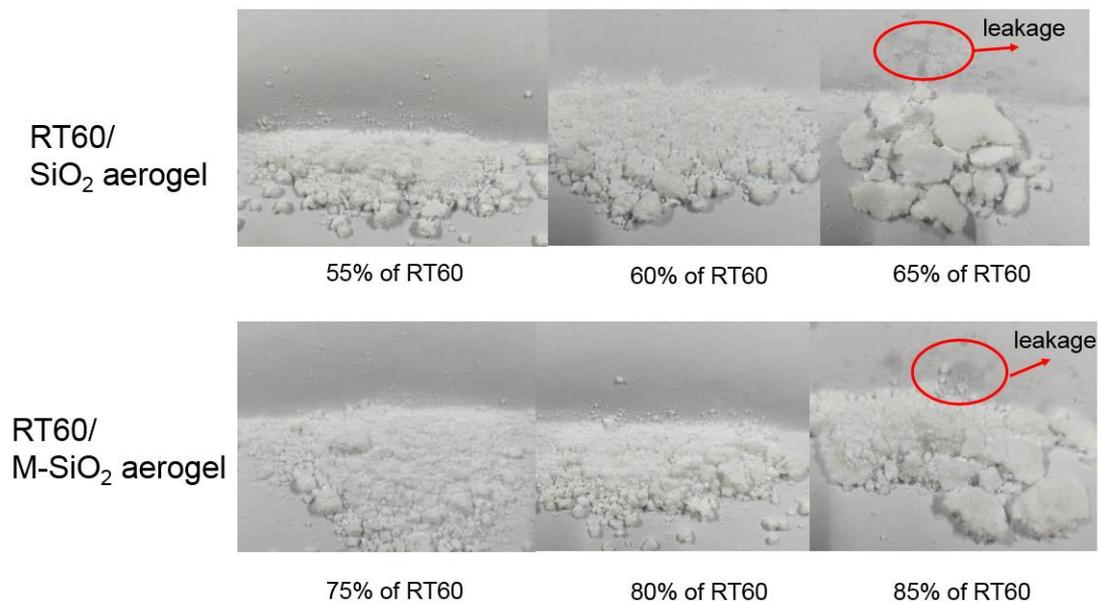


Figure S2. leakage test of RT60/SiO₂ aerogel and RT60/M-SiO₂ aerogel.

Table S1. The absorption capacity of different supporting materials.

Supporting materials	Modified agent	Organic PCMs	T _m / °C	$\Delta H_m/J \cdot g^{-1}$	Mass fraction %
Montmorillonite[1]	--	RT 20	20.8	53.6	39.9%
Modified montmorillonite[2]	Hexadecyltrimethyl ammonium bromide	RT 20	23.0	79.3	58.1%
Diatomite[3]	--	Paraffin	47.8	70.5	47.4%
Calcined diatomite[4]	--	Paraffin	57.3	125.9	61.0%
Expanded perlite[5]	--	Paraffin	17.2	35.5	26.6%
Modified expanded perlite[5]	--	Paraffin	16.3	60.9	45.7%
SiO ₂ aerogel[6]	--	Paraffin	17.4	78.1	54.8%
Modified SiO ₂ aerogel[6]	Dimethyldichlorosilance	Paraffin	17.7	98.9	69.5%
Our work	Hexamethyl disilazane	RT60	60	180.2	80.0%

The XRD results of SiO₂ aerogel and M-SiO₂ aerogel based PCM were shown in Figure S3 and S4. In the pattern of SiO₂ aerogel and M-SiO₂ aerogel, both samples has only one broad peak around 20°. The pattern of RT60 exhibits peaks at 10°, 14.2°, 21.5°, 23.9°, 38.9° and 42.1°. The sharp diffraction peaks at 21.5° and 23.9° are attributed to the diffractions of (110) and (200) crystal planes of paraffin[7]. Figure S4 shows the pattern of RT60/SiO₂ aerogel and RT60/M-SiO₂ aerogel, both RT60/SiO₂ aerogel and RT60/M-SiO₂ aerogel keep two sharp peaks at 21.5° and 23.9° of RT60, while other peaks strongly decrease the intensity, because the RT60 is encapsulated by SiO₂ aerogel and M-SiO₂ aerogel. In general, both of RT60/SiO₂ aerogel and RT60/M-SiO₂ aerogel do not exhibits new peaks in the XRD pattern, verifying the physical combination of RT60 and SiO₂ aerogel/M-SiO₂ aerogel.

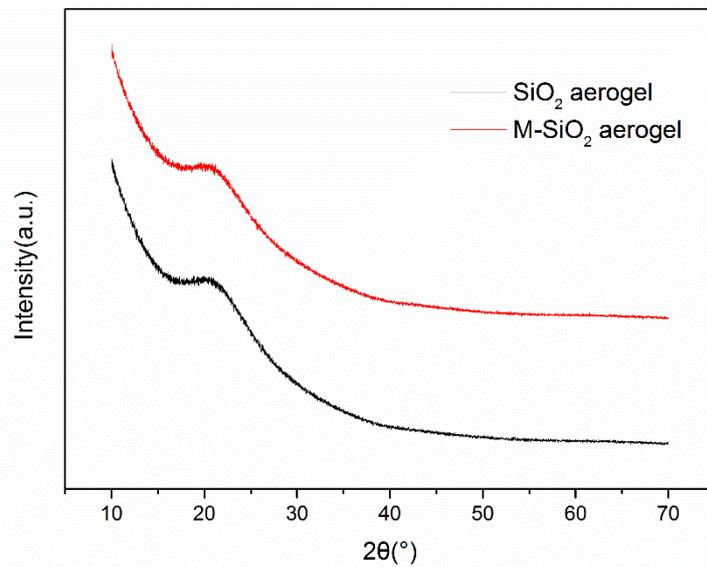


Figure S3. XRD patterns of the SiO₂ aerogel and M-SiO₂ aerogel.

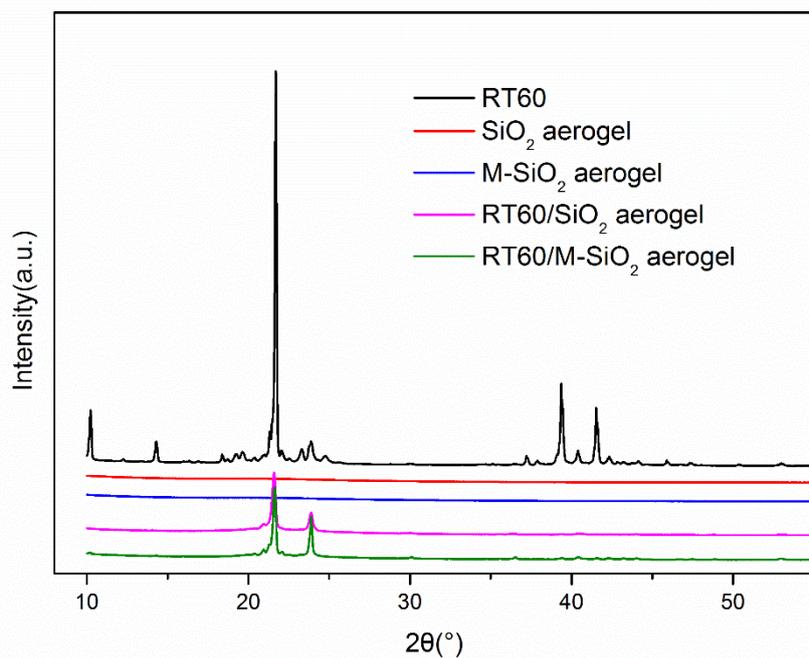


Figure S4. XRD patterns of all samples.

Table S2. The melting and freezing behavior of SiO₂ aerogel and M-SiO₂ aerogel based PCM.

	$T_m/^\circ\text{C}$	$\Delta H_m/\text{J}\cdot\text{g}^{-1}$	$T_f/^\circ\text{C}$	$\Delta H_f/\text{J}\cdot\text{g}^{-1}$	$\eta/\%$
RT60	57.98	225.3	56.61	223.6	-
RT60/SiO ₂ aerogel	57.78	130.5	57.56	129.4	57.9
RT60/M-SiO ₂ aerogel	57.32	180.2	57.16	178.9	80.0

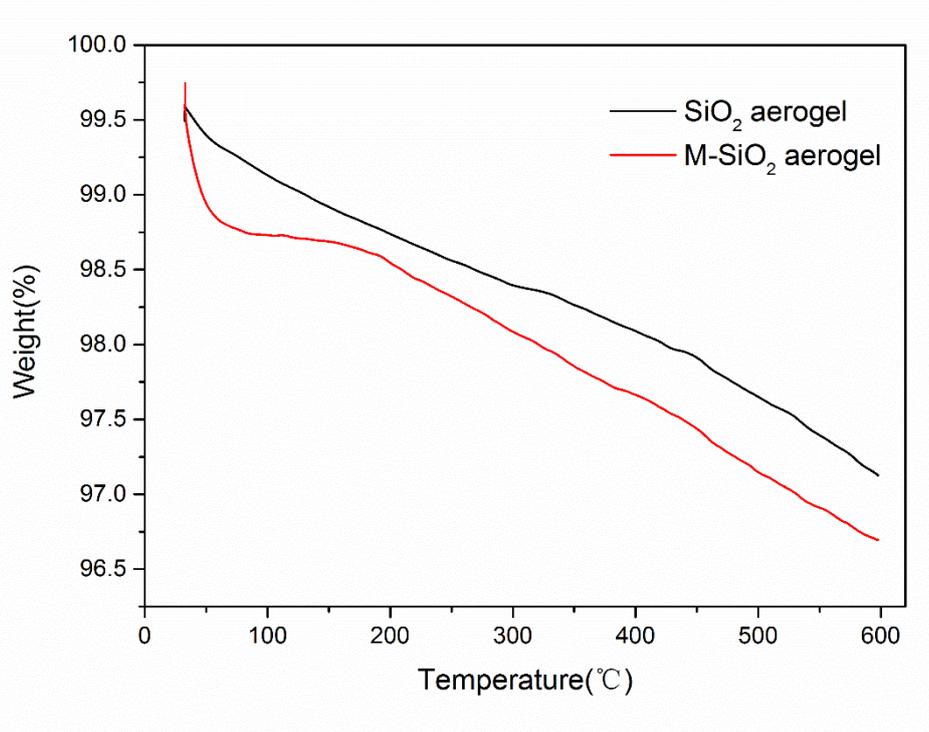


Figure S5. Weight loss of SiO₂ aerogel and M-SiO₂ aerogel.

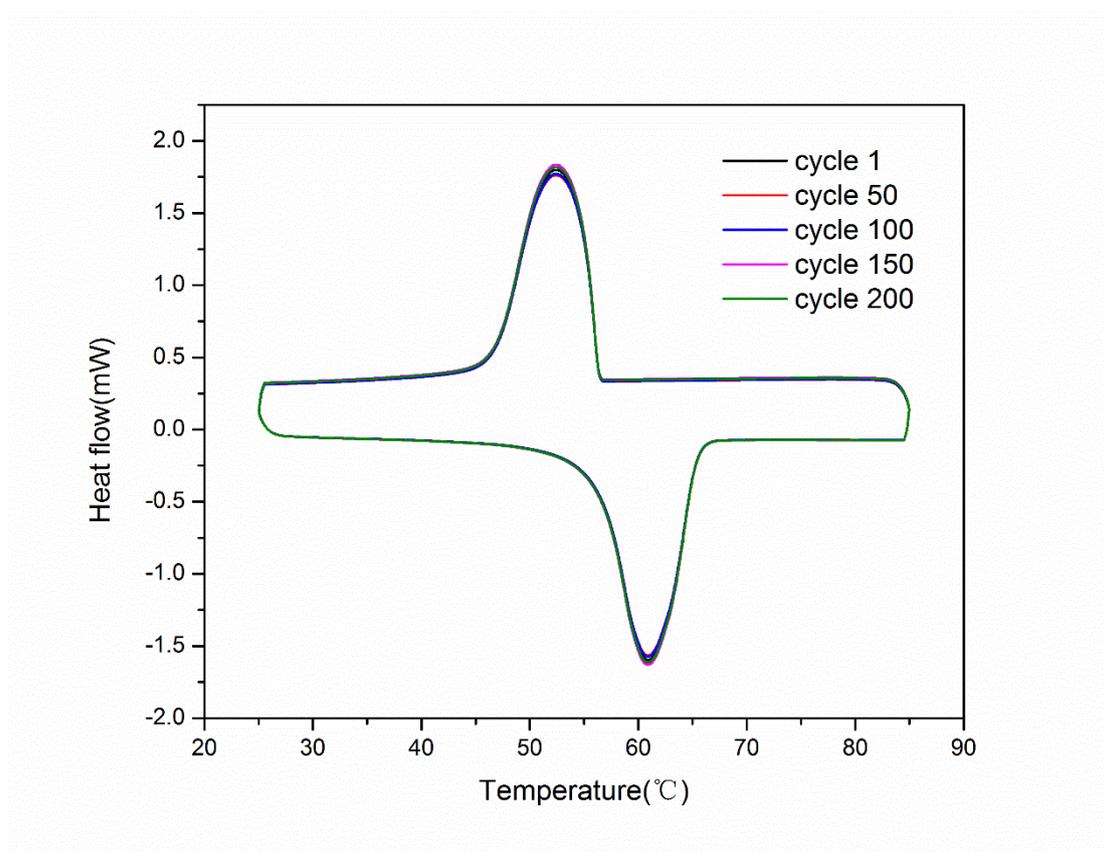


Figure S6. Melting and freezing behavior of RT60/ SiO₂ aerogel with different heating/cooling.

cycle.

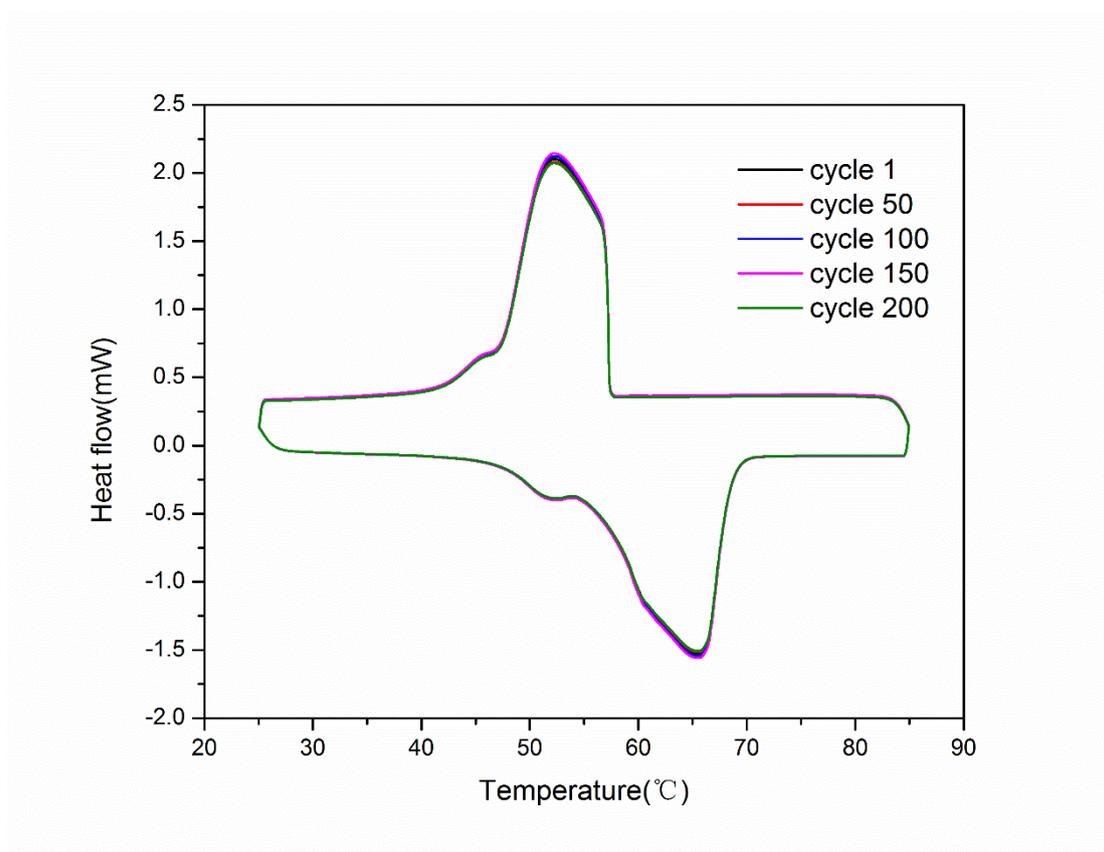


Figure S7. Melting and freezing behavior of RT60/M-SiO₂ aerogel with different heating/cooling cycle.

REFERENCE

1. Fang, X.; Zhang, Z. A novel montmorillonite-based composite phase change material and its applications in thermal storage building materials. *Energy and Buildings* 2006, 38, 377-380.
2. Fang, X.; Zhang, Z.; Chen, Z. Study on preparation of montmorillonite-based composite phase change materials and their applications in thermal storage building materials. *Energy Conversion and Management* 2008, 49, 718-723.
3. Xu, B.; Li, Z. Paraffin/diatomite composite phase change material incorporated cement-based composite for thermal energy storage. *Applied Energy* 2013, 105, 229-237.
4. Sun, Z.; Zhang, Y.; Zheng, S.; Park, Y.; Frost, R.L. Preparation and thermal energy storage properties of paraffin/calcined diatomite composites as form-stable phase change materials. *Thermochimica Acta* 2013, 558, 16-21.
5. Ramakrishnan, S.; Sanjayan, J.; Wang, X.; Alam, M.; Wilson, J. A novel paraffin/expanded perlite composite phase change material for prevention of pcm leakage in cementitious composites. *Applied Energy* 2015, 157, 85-94.
6. Li, H.; Chen, H.; Li, X.; Sanjayan, J.G. Development of thermal energy storage composites and prevention of pcm leakage. *Applied Energy* 2014, 135, 225-233.

7. Zhang, Z.; Zhang, N.; Peng, J.; Fang, X.; Gao, X.; Fang, Y. Preparation and thermal energy storage properties of paraffin/expanded graphite composite phase change material. *Applied Energy* 2012, 91, 426-431.