

Supporting Information

Hierarchical Nickel Cobalt Phosphide @ Carbon Nanofibers Composite Microspheres: Ultrahigh Energy Densities of Electrodes for Supercapacitors

Jinqiao Zhang, Meiling Cen, Tao Wei, Qianyun Wang and Jing Xu *

College of Materials and Metallurgy, Guizhou University, Guiyang 550025,
China

* Correspondence: jxu8@gzu.edu.cn

Supporting Figures:

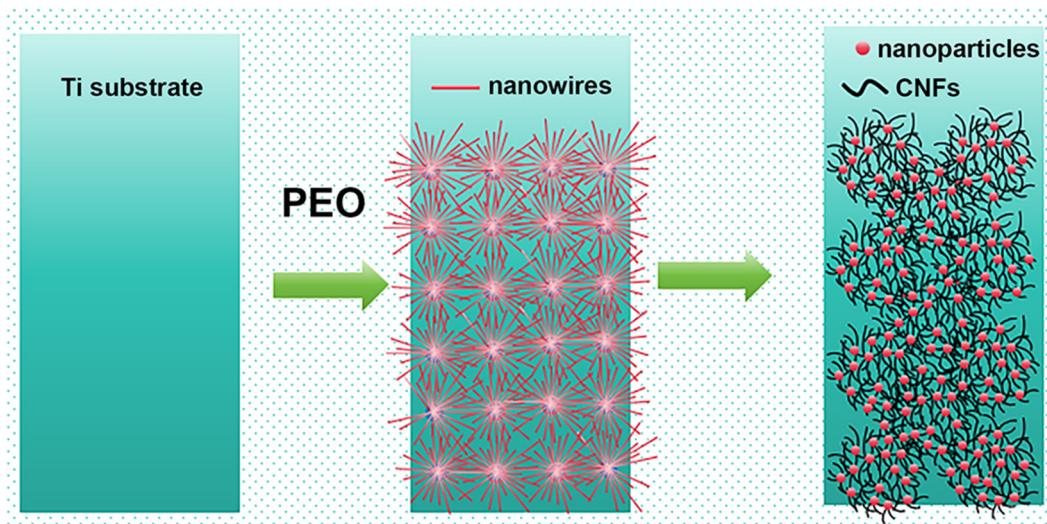


Figure S1. Schematic illustration of synthesis CoNiP@CNFs nanocomposites on a Ti substrate.

Table S1. Chemical composition of used electrolytes for PEO process.

Name	Chemical formula	Concentration (M)	Molecular weight (g/mol)
Sodium tetraborate	Na ₂ B ₄ O ₇ • 10H ₂ O	0.034	381.38
Trisodium phosphate	Na ₃ PO ₄ • 12H ₂ O	0.066	380.12
Sodium tungstate	Na ₂ WO ₄ • 2H ₂ O	0.006	329.85
Calcium acetate	C ₄ H ₆ CaO ₄ • H ₂ O	0.025	176.18
Nickel acetate	C ₄ H ₆ NiO ₄ • 4H ₂ O	0.06	248.84

Cobalt acetate

$\text{C}_4\text{H}_6\text{CoO}_4 \cdot 4\text{H}_2\text{O}$

0.04

249.08

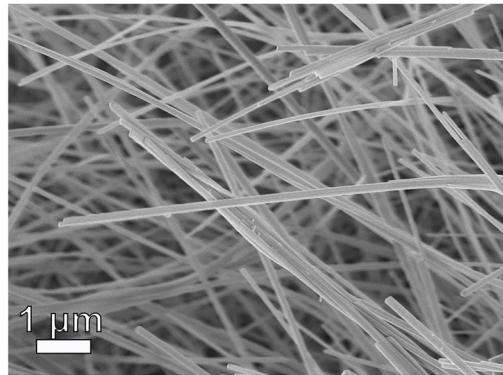


Figure S2. SEM image of Ni_3TiO_7 nanowires.

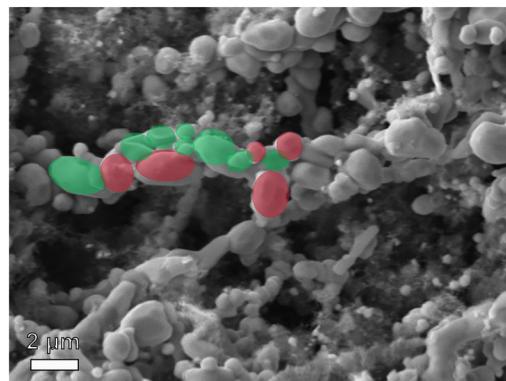


Figure S3. SEM image of $(\text{Ni}_{1-x}\text{Co}_x)\text{TiO}_7$ nanowires subjected to CVD for 10 minutes in C_2H_2 atmosphere.

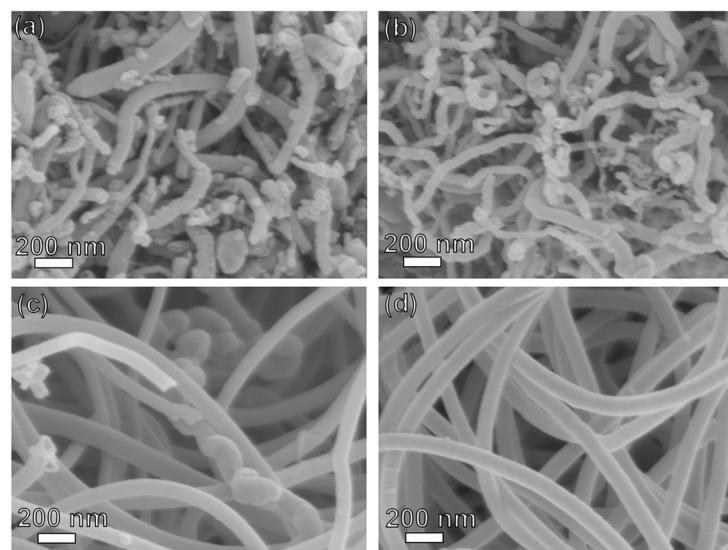


Figure S4. SEM images of a $(\text{Ni}_{1-x}\text{Co}_x)\text{TiO}_7$ film after the TCVD treatment at 700 °C in C_2H_2 atmosphere of (a) 3.75 torr, (b) 7.5 torr, (c) 37.5 torr and (d) 75 torr for 90 min.

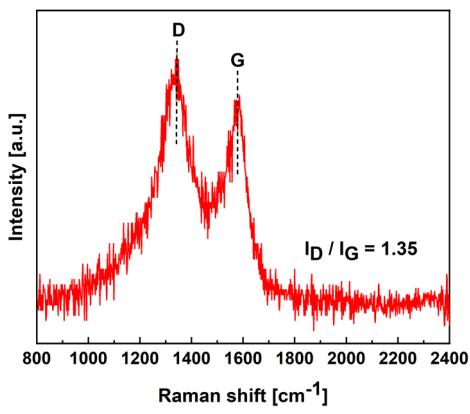


Figure S5. Raman spectroscopy of NiP@CNFs nanocomposites.

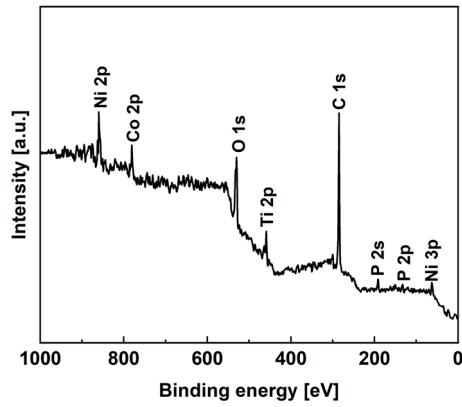


Figure S6. XPS survey spectrum of the CoNiP@CNFs nanocomposites.

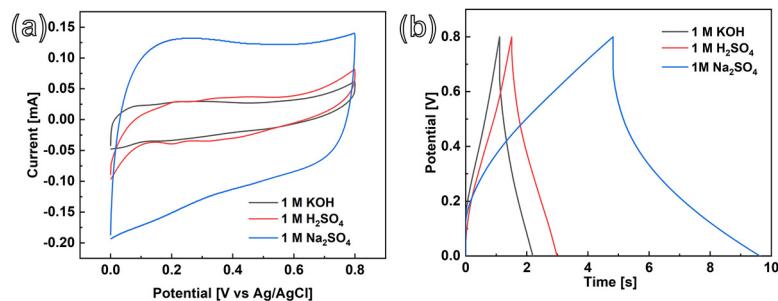


Figure S7. (a) CV and (b) GCD curves of CoNiP@CNFs nanocomposites in different electrolytes.

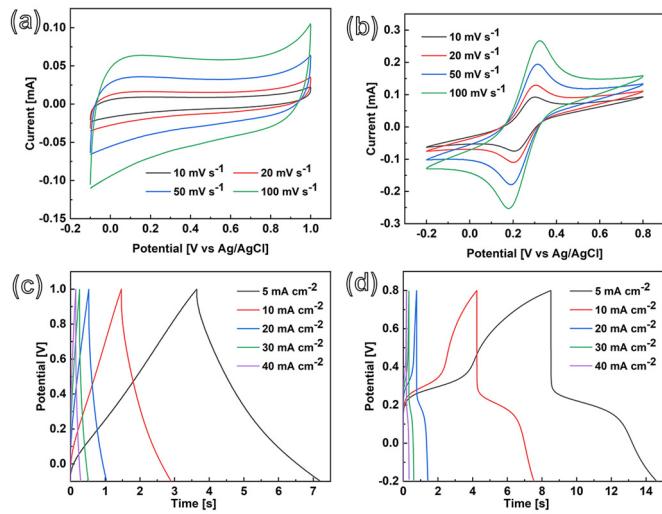


Figure S8. (a, b) CV and (c, d) GCD curves in (a, c) 1.0 M Na₂SO₄ and (b, d) 0.05 M Fe(CN)₆^{3-/4-} + 1.0 M Na₂SO₄ electrolyte solution of NiP@CNFs nanocomposites.

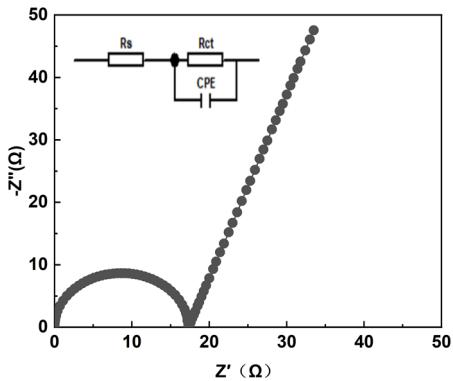


Figure S9. The EIS curve of NiP@CNFs nanocomposites.

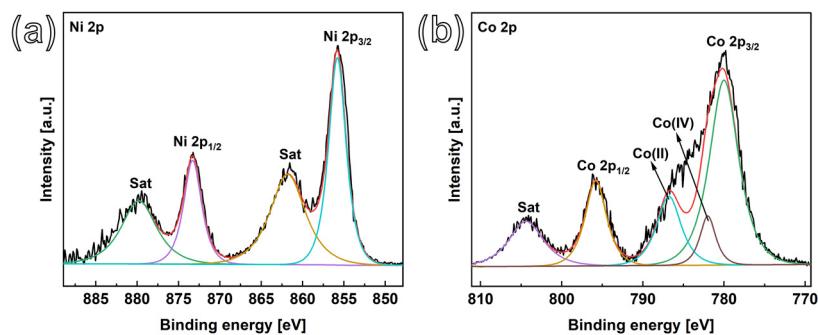


Figure S10. XPS spectra of (a) Ni 2p, (b) Co 2p core levels of CoNiP@CNFs composite after cycle life test.

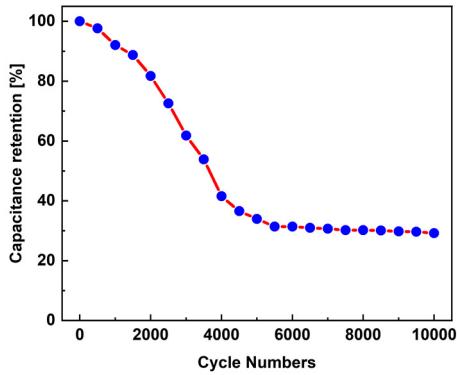


Figure S11. Capacitance retention of a CoNiP based (without growth of CNFs) electrode in 1.0 M Na₂SO₄ solution.

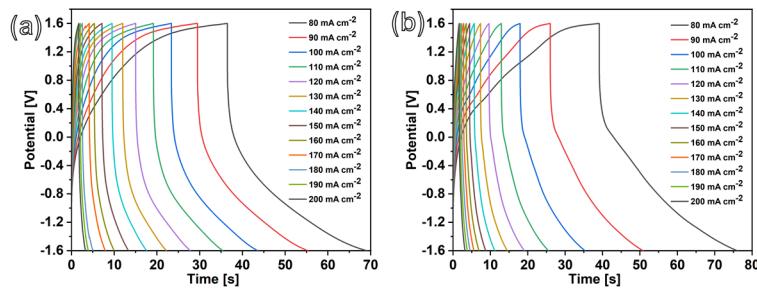


Figure S12. GCD curves of CoNiP@CNFs in (a) 1.0 M Na₂SO₄ and (b) 1.0 M Na₂SO₄ + 0.05 M Fe(CN)₆³⁻/⁴⁻ solution at various current densities from 80 to 200 mA cm⁻².

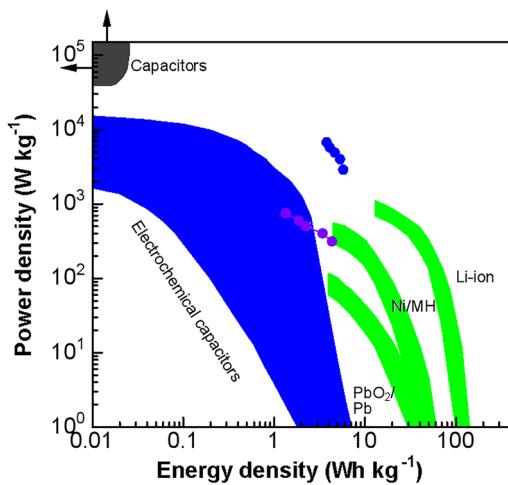


Figure S13. Comparison of Ragone plots of the NiP@CNFs based EDLCs in 1.0 M Na₂SO₄ (purple dotted line) and based PCs in 0.05 M Fe (CN)₆^{3-/4-} + 1.0 Na₂SO₄ (blue dotted line) with those of traditional capacitors, ECs and batteries [S1]

References

- S1. Simon, P.; Gogotsi, Y. Materials for electrochemical capacitors. *Nat. Mater.* **2008**, *7*, 845-854.