



Article TiO₂/SnO₂ Bilayer Electron Transport Layer for High Efficiency Perovskite Solar Cells

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Abstract: The electron transport layer (ETL) has been extensively investigated as one of the important components to construct high-performance perovskite solar cells (PSCs). Among them, inorganic semiconducting metal oxides such as titanium dioxide (TiO₂), and tin oxide (SnO₂) present great advantages in both fabrication and efficiency. However, the surface defects and uniformity are still concerns for high performance devices. Here, we demonstrated a bilayer ETL architecture PSC in which the ETL is composed of a chemical-bath-deposition-based TiO₂ thin layer and a spin-coating-based SnO₂ thin layer. Such a bilayer-structure ETL can not only produce a larger grain size of PSCs, but also provide a higher current density and a reduced hysteresis. Compared to the mono-ETL PCSs with a low efficiency of 16.16%, the bilayer ETL device features a higher efficiency of 17.64%, accomplished with an open-circuit voltage of 1.041 V, short-circuit current density of 22.58 mA/cm², and a filling factor of 75.0%, respectively. These results highlight the unique potential of TiO₂/SnO₂ combined bilayer ETL architecture, paving a new way to fabricate high-performance and low-hysteresis PSCs.

Keywords: perovskite solar cells; electron transport layer; low hysteresis; SnO₂; TiO₂

1. Introduction

The high-efficiency, low-cost and facile fabrication process of halide perovskite solar cells (PSCs) have attracted tremendous attention in the field of photovoltaics in the past decade [1–5] and been regarded as the most promising substitute for traditional silicon (Si) and copper indium gallium selenide (CIGS) solar cells [6–8]. The sandwich structure of hybrid organic-inorganic based PSCs includes the electron transport layer (ETL), perovskite absorber layer, hole transport layer (HTL) and electrodes. Among them, ETL and HTL are used for the electron and hole extraction, respectively. However, the Spiro-OMeTAD are widely used as HTL in PSCs because of the simple synthesis, high carrier mobility and suitable valance band. The HTL are always fabricated by spin-coating on the top of a perovskite absorber layer with a dense and uniform film. In contrast, the ETL in PSCs is usually fabricated in a planar and/or mesoporous structure under the perovskite absorber layer [9–11]. The surface quality of ETL can substantially influence the deposition of perovskite film. Therefore, the electron transport layer and the corresponding interface of ETL/perovskite are significantly important parts to fabricate high-quality PSCs. Titanium dioxide (TiO₂) and/or tin oxide (SnO₂) thin films have been extensively investigated as an effective ETL in the PSCs, which can be fabricated by several different methods such as spin-coating, sputtering and chemical bath deposition (CBD) [12–16], to pursue a higher performance device.

Due to the facile planar configuration of PSCs, fabricating uniform, and compact ETL thin layer, it is imperative to pursue high performance. The conventional spin-coating



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). method shows a facile and efficient way to fabricate the TiO_2 -ETL. However, the uneven distribution of TiO_2 nanoparticles result in the carrier accumulation between perovskite (PVSK) and the ETL interface and an insufficient carrier extraction, leading to a low efficiency of resultant device [17,18]. Moreover, the large hysteresis of TiO₂-ETL also impedes the further application of TiO₂ in the PSCs [19]. Alternatively, SnO₂ presents a reduced hysteresis, high carrier mobility and good energy level towards perovskite, which can greatly improve the performance of PSCs [20-22]. For example, You et al. proposed SnO_2 as a planar ETL in the PSCs, which not only reduces the energy barrier between ETL/PVSK, but also reduces the hysteresis of devices, resulting in a high performance PSC with a champion PCE of 20.5% [21]. However, uniformity of SnO₂ nanoparticles is still a concern for the device fabrication because of its uneven distribution by spin-coating technique. Therefore, high-quality ETL plays a crucial role in the fabrication of devices, which paves a promising way for high-efficiency PSCs. To address this issue, Xu et al. introduced a bilayer ETL of TiO_2/ZnO thin layers into PSCs, which produces a compact interfacial layer to avoid direct contact between the FTO substrate and PVSK, leading to a reduced carrier accumulation at ETL/PVSK interface [23].

In this work, we propose a bilayer of ETLs that is composed of a CBD TiO₂ layer and a spin-coated SnO₂ layer. The presence of the SnO₂ thin layer on the top surface of CBD TiO₂ film can provide a higher current density and reduce the hysteresis of PSCs simultaneously. In addition, the diffusion of the K ion from SnO₂ can significantly improve the crystallinity of grains in the perovskite films. On the basis of this bilayer strategy, a higher power conversion efficiency (PCE) of 17.64% was achieved in comparison with the mono-TiO₂ ETL based PSCs with a PCE of 16.16%.

2. Materials and Methods

Materials: All reagents were used as received without further purification. Methylammonium iodide (MAI), methylammonium bromide (MABr), methylammonium chloride (MACl), formamidinium iodide (FAI), lead(II) iodide (PbI₂) and 2,2',7,7'-tetrakis(N,N-di-pmethoxyphenylamine)9,9'-spirobifluorene (Spiro-OMeTAD) (99.5%) were purchased from Xi'an Polymer Light Technology (Xi'an, China). Dimethylformamide (DMF), dimethyl sulfoxide (DMSO), isopropanol (IPA), chlorobenzene (CB), and titanium tetrachloride (TiCl₄) were purchased from Sigma-Aldrich (Milwaukee, Germany).

Device Fabrication: The cleaned fluorine-doped tin oxide (FTO) substrates are treated using UV-ozone for 60 min. Then, the TiO_2 thin layer was prepared by using the CBD method and the SnO₂ thin layer was fabricated with spin-coating technologies, as shown in Figure 1. First, 2M aqueous TiCl₄ mother solution was prepared by dropping TiCl₄ into distilled water. During the preparation, the mother solution was continuously stirred at a low temperature of around 0 °C. The as-prepared TiCl₄ mother solution was stored in the refrigerator (<10 $^{\circ}$ C). Second, the as-prepared TiCl₄ mother solution was diluted to a 0.2 M TiCl₄ solution. The cleaned FTO substrates were placed vertically in the glassware. Then, 300 mL of 0.2 M TiCl₄ solution was poured into the glassware. The glassware was put into an oven with a temperature of 75 °C. After 1 h heating, the glassware was taken out followed by rinsing the FTO substrates several times using distilled water. Finally, the FTO substrates were annealed at a high temperature of 450 °C for 30 min. The FTO substrates were washed by the acetone, distilled water, and ethyl alcohol for 20 min, respectively. Before the deposition of TiO_2 thin films, the FTO substrates are treated by using UV-ozone for 60 min. SnO_2 films were prepared by spin-coating Alfa Aesar SnO_2 (diluted by H_2O to 3%) at a speed of 3500 rpm for 30 s. The perovskite films were deposited by a two-step spin-coating method. Specifically, 1.35 M PbI₂ and 0.0675 M CsI were dissolved in organic solvent (DMF/DMSO = 19:1). The PbI₂ precursor solution was stirred at a temperature of 70 °C for 60 min. The mixed MAFA based organic cation precursor solution was prepared by dissolving 200 mg FAI, 100 mg MAI, 25 mg MABr and 25 mg MACl dissolved in 5 mL isopropanol. The PbI_2 precursor solution was first spin-coated at a speed of 3000 rpm for 30 s. The MA/FA cation solution was spin-coated at 3000 rpm for 30s. After annealing at

150 °C for 10 min, the perovskite film of $Cs_{0.05}FA_{0.54}MA_{0.41}Pb(I_{0.98}Br_{0.02})_3$ was obtained. The hole transport layer of the spiro-OMeTAD film was deposited by spin-coating the spiro-OMeTAD solution at a speed of 3500 rpm for 25 s. Finally, 80 nm Au film was deposited as a counter electrode by thermal evaporation.



Figure 1. Schematic illustration of the bilayer of ETLs (TiO₂ and SnO₂ films) and perovskite films fabricated by chemical bath deposition and spin-coating.

Device Characterization: The diffraction data of perovskites are collected by using a Bruker D8 Discover diffractometer (Bruker AXS) from 10° to 60°. Surface and crosssection morphology images are recorded by a scanning electron microscope (SEM) (Helios NanoLab G3). The TRPL results were collected by using the Hamamatsu equipment which can provide an excitation wavelength of 450 nm. The photoluminescence (PL) spectra were acquired by a JASCO FP-8500 spectrometer with an excitation wavelength of 450 nm. The current-voltage (J-V) measurements were performed under one sun illumination (AM1.5G, 100 mW/cm²) by using a Keithley 2420. The devices were test by using a metal shadow mask with a dimension of 0.3×0.3 cm². The EQE spectra of the devices were characterized by using Oriel IQE 200 equipment.

3. Results and Discussion

Figure 2a–d shows the top-view SEM images of the perovskite films fabricated on the TiO_2 and TiO_2/SnO_2 substrates, which clearly shows a larger grain size of perovskite thin film based on the TiO_2/SnO_2 substrates, compared with that on the TiO_2 substrates, with an average value changing from ~380 nm to ~540 nm, which can be verified by the

statistics of perovskite grain size based on the TiO_2 and TiO_2/SnO_2 substrates, as presented in Figure 2e,f. As is well-known, the commercial SnO_2 colloid precursor is stabilized by incorporating potassium hydroxide (KOH) [24]. The presence of K ion in the SnO_2 will diffuse into the perovskite thin film during the annealing process, which greatly enhances the crystallinity of perovskite grains, and reduces the hysteresis of resultant devices [25–27].



Figure 2. (a) Top-view SEM images of monolayer TiO₂-ETL PSCs and bilayer TiO₂/SnO₂-ETL PSCs (**a**–**d**), and their corresponding statistic of grain size (**e**,**f**).

Furthermore, the phase structure of perovskite thin film deposited on the TiO_2 and TiO_2/SnO_2 substrates was investigated by X-ray diffraction (XRD), as presented in Figure 3a. The increase of XRD intensity (on the TiO_2/SnO_2 substrate) verifies that the improved crystallinity of perovskites is accomplished with high absorption in a short-wavelength region (as shown in Figure 3b).



Figure 3. (a) XRD patterns, (b) absorption spectra, (c) PL spectra and (d) TRPL curves of the perovskite films deposited on FTO/TiO₂ and FTO/TiO₂/SnO₂ substrates.

In addition, the steady-state photoluminescence (PL) and time-resolved photoluminescence (TRPL) experiments were carried out to investigate carrier transport behavior. As seen in Figure 3c,d, the faster PL quenching of the perovskite thin film on the TiO_2/SnO_2 substrate indicates an enhanced electron extraction capability [28]. Moreover, the lifetimes of the corresponding perovskite thin films were fitted by a biexponential decay function [29,30]. The lifetime of the TiO_2/SnO_2 -based sample is 15.4 ns, which is shorter than that of the TiO_2 -based sample (22.2 ns), indicating a faster carrier extraction from the perovskite thin film to TiO_2/SnO_2 electron transport layer [31].

Figure 4a,b shows the cross-section SEM images of devices fabricated on TiO_2 and TiO_2/SnO_2 substrates. The uniform and dense perovskite absorber layers not only ensure the light harvest, but also effectively impede the carrier recombination in the devices. The current density-voltage (J-V) curves of the devices were measured under standard AM 1.5 G illumination and are shown in Figure 5a and Table 1, while the key performance parameters of open-circuit voltage (V_{OC}), short-circuit current (J_{SC}), fill factor (FF), power conversion efficiency (PCE) and their statistical analyses are displayed in Figure 6a–d

and Table 2, respectively. The PCE of 16.16% (V_{OC} = 1.012 V, J_{SC} = 22.06 mA/cm² and FF = 72.4%) and 10.37% (V_{OC} = 0.905 V, J_{SC} = 22.06 mA/cm² and FF = 51.9%) under reverse scan (RS) and forward scan (FS) indicate large hysteresis in the TiO₂-based devices. In contrast, the high PCE of 17.64% ($V_{OC} = 1.041 \text{ V}$, $J_{SC} = 22.58 \text{ mA/cm}^2$ and FF = 75.0%) and 15.29% (V_{OC} = 1.001 V, J_{SC} = 22.73 mA/cm² and FF = 67.2%) under RS and FS were obtained for TiO_2/SnO_2 -based solar cells. The improved efficiency of TiO_2/SnO_2 -based solar cells can be attributed to a higher crystallinity of perovskite grains, which enhances light capture and reduces the defects at grain boundaries [14,25]. The EQE spectra of the corresponding devices were presented in Figure 5b. The improved EQE in the short wavelength in terms of TiO2/SnO2-based device indicates faster carrier extraction and reduced recombination at the TiO₂/SnO₂/PVSK interface [32]. Similarly, the enhanced EQE at the long wavelength region also suggests that reduced defects and carrier recombination in the perovskite bulk film, which can be explained by the enlarged grain size and improved crystallinity of the perovskite grains [32]. As a result, the integrated J_{SC} from EQE of the TiO₂/SnO₂-based device is 21.59 mA/cm², which is higher than that of the TiO₂ based device (21.17 mA/cm²). Furthermore, the TiO_2/SnO_2 -based device exhibited a stable output (under initial maximum power point (MPP) voltage) with a PCE of 17.65%. In contrast, the TiO₂ based solar cell shows a poor output under MPP, yielding a low PCE of 15.74% (Figure 5c). More importantly, the hysteresis (hysteresis index (HI) = PCE_{RS}/PCE_{FS}) of the TiO_2/SnO_2 -based devices is also reduced, compared to TiO_2 -based devices [32–34]. The HI of the TiO₂-based PSC is 1.56, which is decreased to 1.15 by incorporating SnO_2 into devices to construct the TiO₂/SnO₂ bilayer ETL. Compared to TiO₂ based devices with a large hysteresis of 1.51, the improved efficiency and reduced HI of 1.18 for TiO_2/SnO_2 based PSCs indicates the bilayer ETL can improve the reproducible fabrication and the device performance.



Figure 4. Cross-section images of PSCs in (a) TiO₂-PSCs and (b) TiO₂/SnO₂ PSCs.



Figure 5. (a) Measured current density-voltage curves of champion devices. (b) Corresponding EQE spectra and their integrated current density. (c) Stable output of perovskite solar cells based on based on TiO_2 and TiO_2/SnO_2 ETLs.

Sample	Scan Direction	V _{oc} (V)	J _{sc} (mA/cm ²)	FF	PCE (%)	HI
TiO ₂ /SnO ₂ PSC	FS. RS.	1.001 1.041	22.73 22.58	0.672 0.750	15.29 17.64	1.18
TiO ₂ -PSCs	FS. RS.	0.905 1.012	22.06 22.06	0.519 0.724	10.37 16.16	1.51

Table 1. Photovoltaics parameters of PSCs based on TiO_2 and TiO_2/SnO_2 ETLs.



Figure 6. Statistical distribution of TiO_2 and TiO_2/SnO_2 based PSCs (10 devices). (a) V_{OC} , (b) J_{SC} , (c) FF and (d) PCE.

Table 2. Average photovoltaics parameters of PSCs based on TiO₂ and TiO₂/SnO₂ ETLs.

Sample	Scan Direction	V _{oc} (V)	J _{sc} (mA/cm ²)	FF	PCE (%)
TiO ₂ /SnO ₂ PSC	FS. RS.	$\begin{array}{c} 0.985 \pm 0.010 \\ 1.029 \pm 0.007 \end{array}$	$\begin{array}{c} 22.17 \pm 0.34 \\ 22.16 \pm 0.27 \end{array}$	$\begin{array}{c} 0.626 \pm 0.039 \\ 0.709 \pm 0.028 \end{array}$	$\begin{array}{c} 13.68 \pm 1.04 \\ 16.18 \pm 0.78 \end{array}$
TiO ₂ -PSCs	FS. RS.	$\begin{array}{c} 0.917 \pm 0.018 \\ 1.003 \pm 0.016 \end{array}$	$\begin{array}{c} 22.11 \pm 0.40 \\ 22.01 \pm 0.43 \end{array}$	$\begin{array}{c} 0.5090 \pm 0.040 \\ 0.707 \pm 0.014 \end{array}$	$\begin{array}{c} 10.33 \pm 0.88 \\ 15.61 \pm 0.40 \end{array}$

4. Conclusions

In summary, we developed a bilayer electron transport layer by combining CBD-TiO₂ and spin-coated SnO_2 in the perovskite solar cells. The TiO₂/SnO₂ bilayer ETLs provide not only a compact electron transport layer, but also accelerate the carrier transport in the solar

cells. Furthermore, the presence of K ion from SnO_2 can greatly improve the crystallinity of perovskite thin film and significantly reduce the hysteresis of resultant devices. Compared with the TiO₂-based solar cells, the TiO₂/SnO₂-based solar cells demonstrate a higher PCE of 17.64% and a lower hysteresis index. These results highlight the potential fabrication of the TiO₂/SnO₂ bilayer electron transport layers and will be a beneficial strategy to fabricate a high-quality perovskite thin film solar cell.

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