



Supplementary Materials

Plasmon Effect of Ag Nanoparticles on TiO₂/rGO Nanostructures for Enhanced Energy Harvesting and Environmental Remediation

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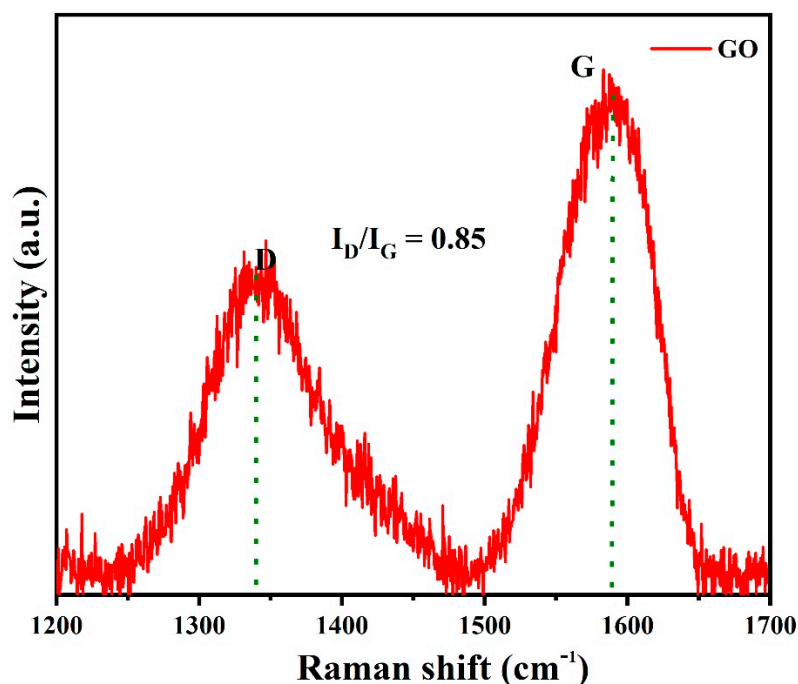


Figure S1. Raman spectra of prepared GO nanosheets.

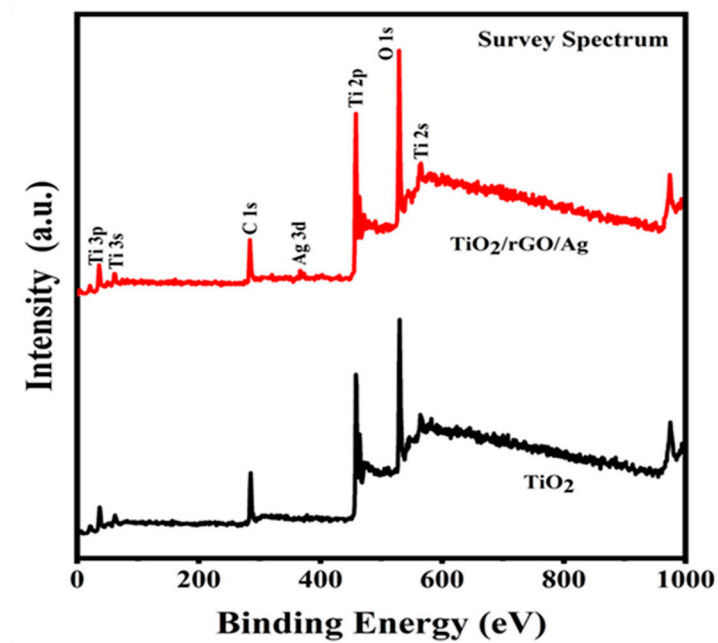


Figure S2. XPS survey spectrum of TiO_2 , TiO_2/rGO and $\text{TiO}_2/\text{rGO}/\text{Ag}$.

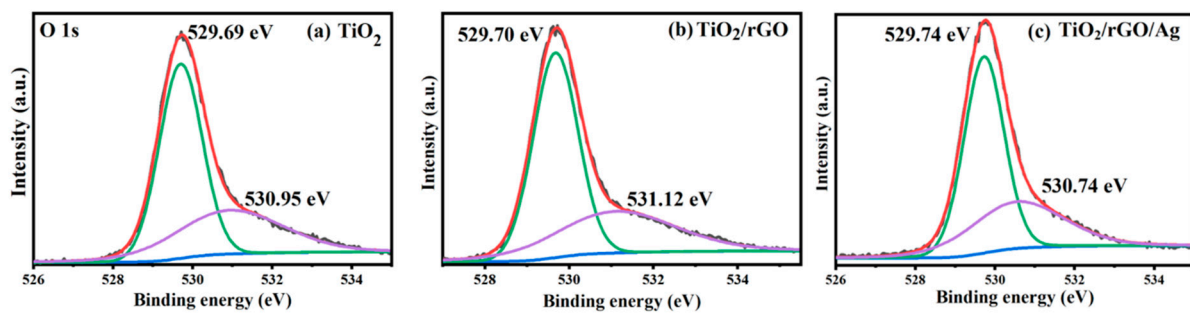


Figure S3. XPS O 1s spectra of (a) TiO_2 (b) TiO_2/rGO and (c) $\text{TiO}_2/\text{rGO}/\text{Ag}$ (green line (Ti-O-Ti bond) and violet line (Ti-O-C bond)).

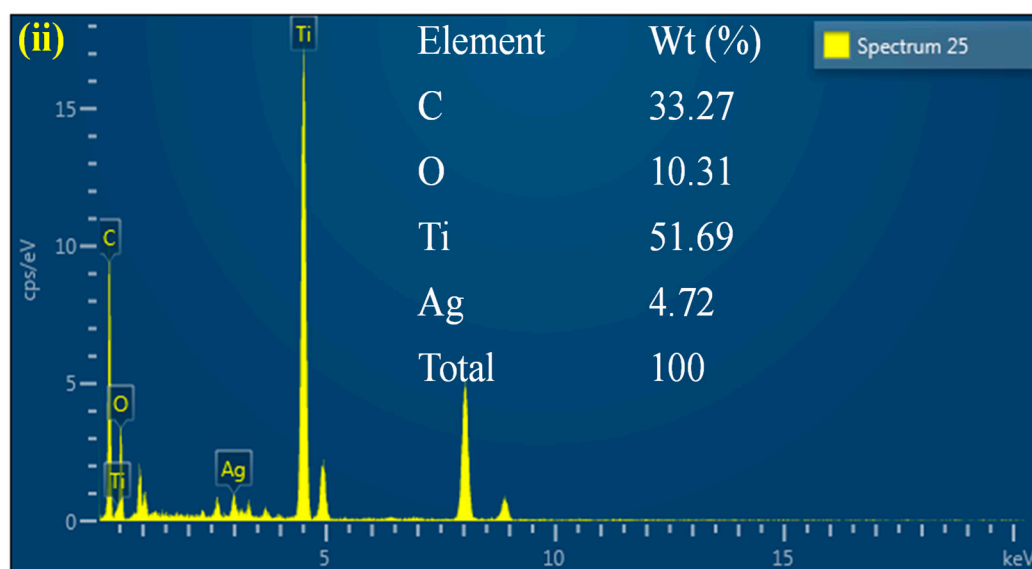
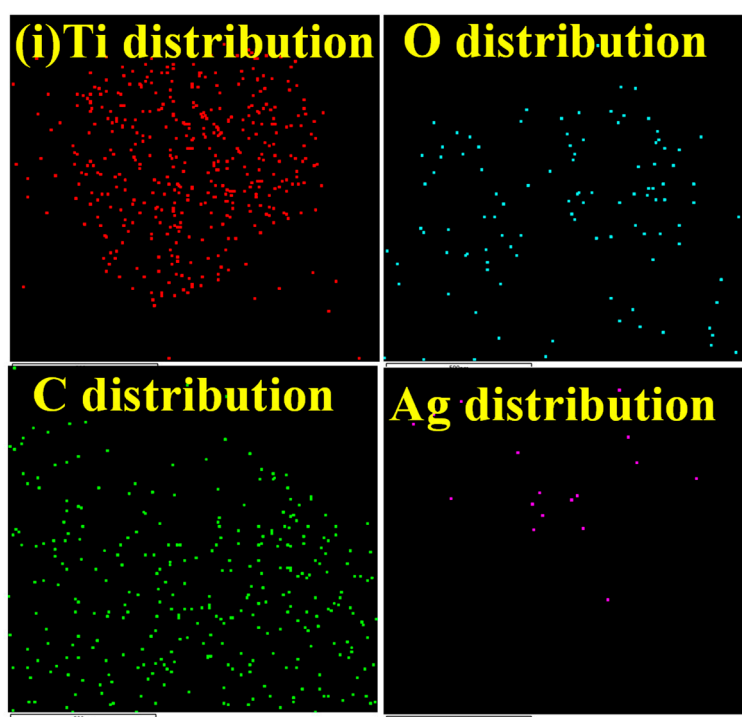


Figure S4. (i) STEM elemental mapping and (ii) EDX spectrum of TiO₂/rGO/Ag hybrid ternary nanostructures.

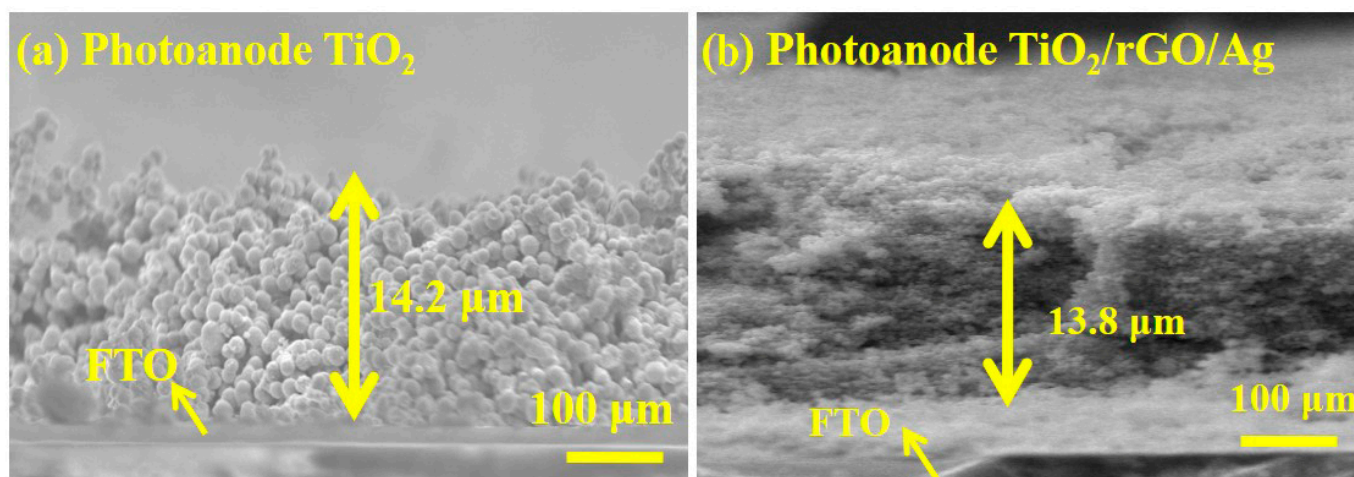


Figure S5. The HR-SEM cross section of (a) TiO_2 (b) $\text{TiO}_2/\text{rGO}/\text{Ag}$ photoanode.

The functional group of the synthesized samples were performed by FTIR spectrum as shown in Figure S6. The strong and broad peaks are observed in range of 2800 to 3400 cm^{-1} , and it's attributed to the stretching vibration of the O-H bond (the hydroxyl groups in carboxylic acid). Besides, the other peak at 1623 cm^{-1} represents the C=O group which direct confirms the presence of GO sheet. On other hand, the hydroxyl peak has observed at 3430 cm^{-1} and 1224 cm^{-1} , respectively [1]. Then, major peak intensity of GO is reduced significantly which reveals that removal of oxygen-containing groups namely reduced graphene oxide after the solvothermal treatment of TiO_2/rGO . Besides, the weak peak is observed at 1224 cm^{-1} due to the epoxy C-O-C stretching vibration [2]. All the peaks in FTIR study are originated from the oxygenated functional group and C=C indicating the existence of rGO in the hybrid nanostructure.

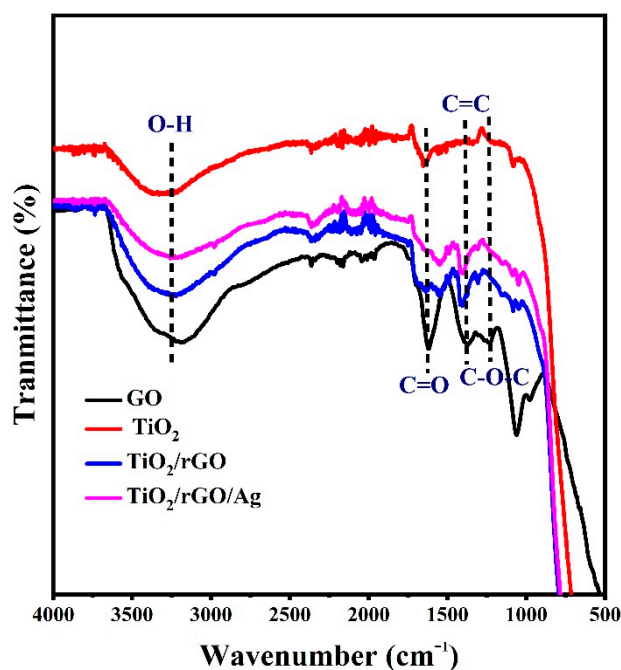


Figure S6. FTIR spectra of GO, TiO_2 , TiO_2/rGO and $\text{TiO}_2/\text{rGO}/\text{Ag}$.

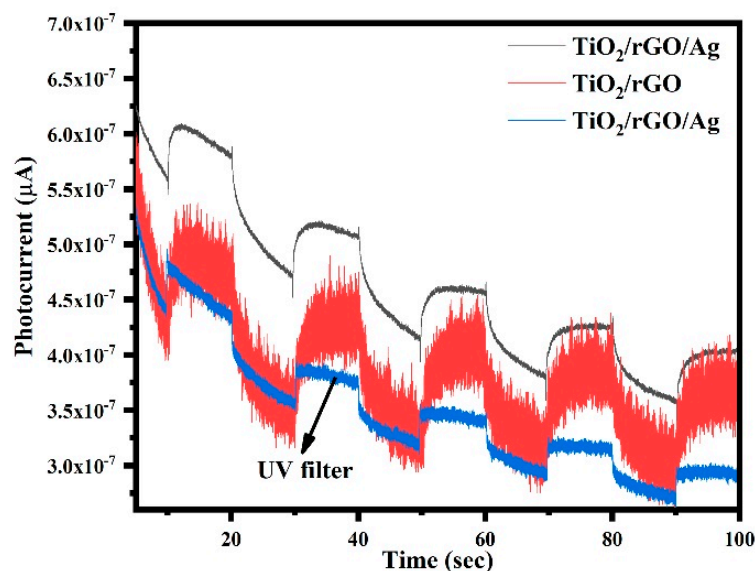


Figure S7. Transient photocurrent spectra of TiO_2/rGO and $\text{TiO}_2/\text{rGO}/\text{Ag}$ (under with and without UV filter).

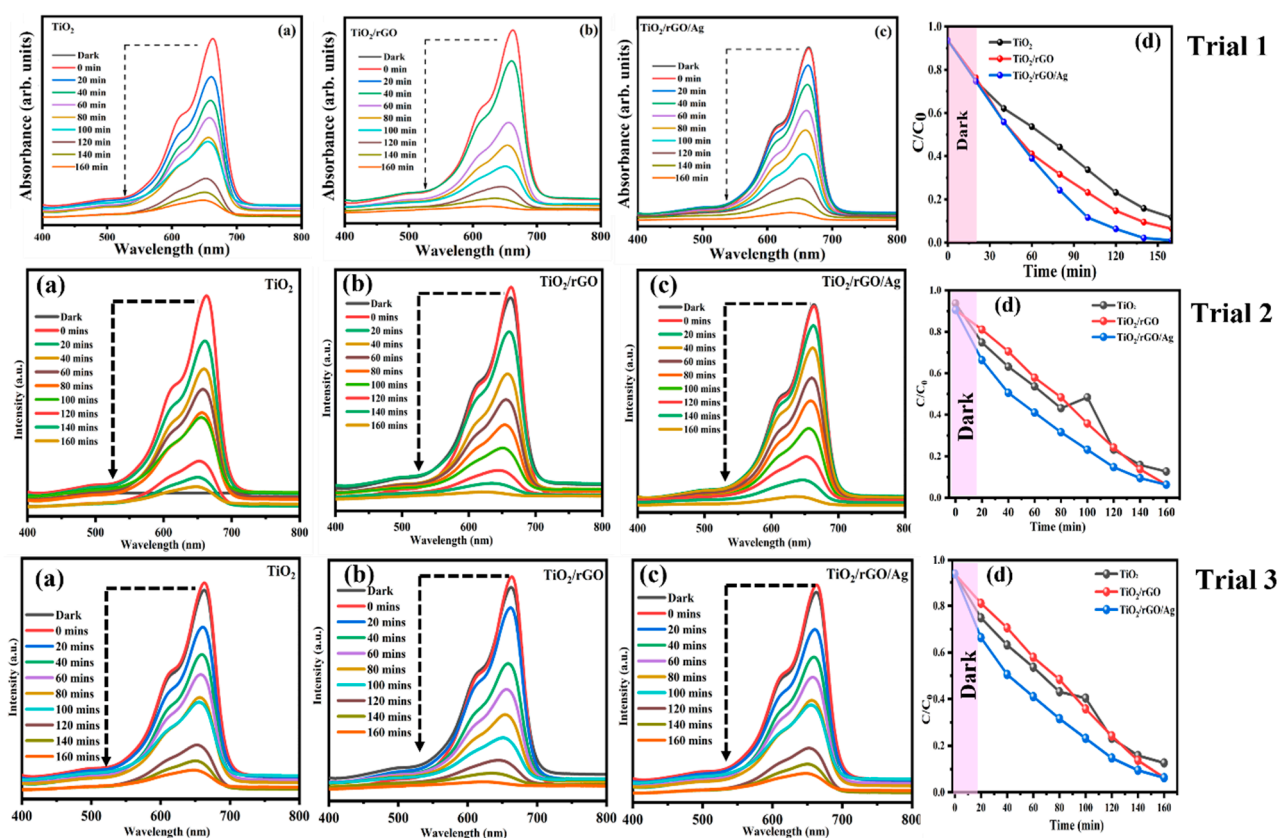


Figure S8. (a-c) Three different trials of UV absorption spectra of MB degradation with different interval time under natural sunlight. (d) Plots of $\ln(C/C_0)$ as a function of time (min) towards the photo degradation of MB.

References

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