



## Supplementary Material

# Synthesis of High Surface Area—Group 13—Metal Oxides via Atomic Layer Deposition on Mesoporous Silica

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## 1. Synthesis of Reference Samples

**Materials:** Silica powder (SiO<sub>2</sub> amorphous, ≥99%, high-purity grade (Davisil Grade 636), average pore size 60 Å, 35–60 mesh particle size, specific surface area 505 m<sup>2</sup>/g, Sigma-Aldrich). Gallium and indium nitrate hydrate (Ga(NO<sub>3</sub>)<sub>3</sub> · xH<sub>2</sub>O and In(NO<sub>3</sub>)<sub>3</sub> · xH<sub>2</sub>O, 99.9% trace metals basis, Merck, Germany). Water (H<sub>2</sub>O, CHROMASOLV®, for HPLC, Riedel-de Haën/ Honeywell Specialty Chemicals Seelze GmbH, Seelze, Germany).

**Protocol:** Reference samples, supported on SiO<sub>2</sub>, were synthesized via incipient wetness impregnation (IWI). For the synthesis of Ga<sub>2</sub>O<sub>3</sub>/SiO<sub>2</sub> and In<sub>2</sub>O<sub>3</sub>/SiO<sub>2</sub>, the respective nitrate hydrate was added to HPLC-grade water to obtain the required volume for IWI of one gram of silica. After impregnation, the samples were dried in air at 80 °C for 12 h. Subsequently, the samples were calcined at 500°C for 3 h under 20% O<sub>2</sub>/N<sub>2</sub> (syn-air). The three-cycle ALD samples were calcined under same conditions for comparison.

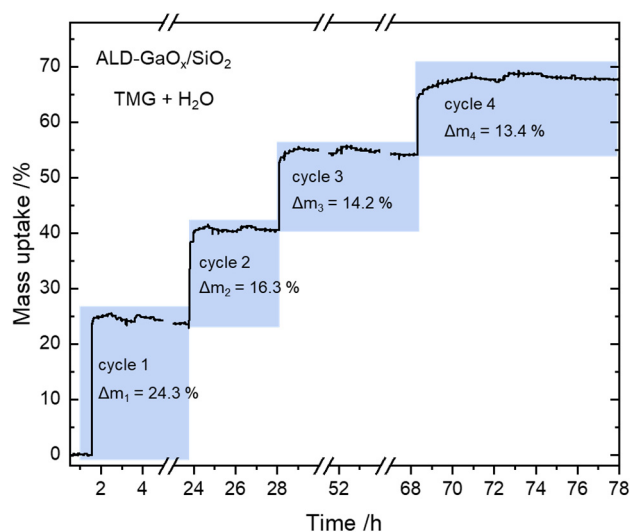
Equations

$$\text{Added mass/mol(precursor)} = \frac{\text{mass(oxide)} \times M(\text{metal})}{\text{mass(metal)}} \quad (1)$$

Consideration: Amount of deposited metal equals chemisorbed precursor

$$\text{Layer thickness(oxide)} = \frac{\text{mass(oxide)}}{\rho(\text{oxide}) \times \text{surface area(substrate)}} \quad (2)$$

Consideration: Flat substrate surface and even oxide layer



**Figure S1.** *In-situ* gravimetric monitoring of 4 ALD cycles  $\text{GaO}_x$  on  $\text{SiO}_2$  powder using the process of TMG/ $\text{H}_2\text{O}$  at  $150^\circ\text{C}$ . Mass-uptake =  $\Delta m/m_0$  ( $m_0 = \text{SiO}_2$ ).

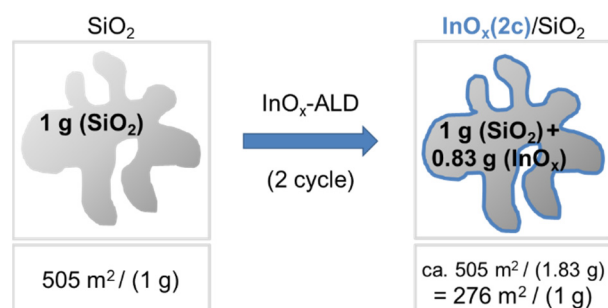
**Table S1.** Metal-uptakes of Al, Ga and In on  $\text{SiO}_2$  within 3 cycles of ALD using TMX (X = A, G, I) and water. Values are calculated from thermogravimetric data (MSB) as described in chapter 3.1. Mass-uptake =  $\Delta m/m_0$  ( $m_0 = \text{SiO}_2$ ).

Cycle Nr.	Al-Metal [1] Uptake /%	Ga-Metal Uptake /%	In-Metal Uptake /%	Uptake Ratio (Al : Ga : In)	Molar Mass Ratio (Al:Ga:In)
1 cycle	+ 8.7	+ 18.1	+ 29.9	1:2.1:3.4	1:2.6:4.3 (native)
Ø 3 cycles	+ 7.5	+ 13.6	+ 35.5	1:1.8:4.7	

## 2. Basic Consideration and Rationalization of the Changes in Surface Area and Pore Volume

### 2.1. Approximation (Density):

The specific surface areas are estimated based on the ALD-induced change of the substrate-mass, assuming no change of the exposed surface area (as discussed in chapter 3.2). The estimated (specific) surface areas and pore volumes are displayed in Table S2 (blue).



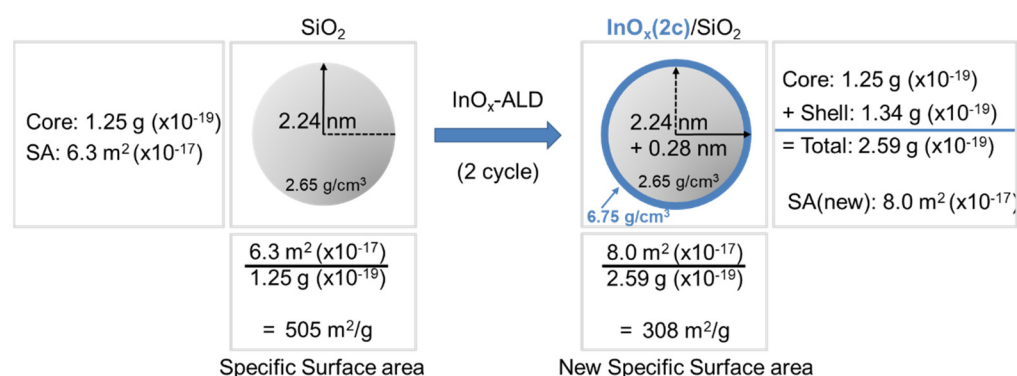
**Scheme S1.** Schematic description of model 1, based on two ALD cycles of  $\text{InO}_x$  on  $\text{SiO}_2$ .

**Table S2.** Estimated (blue) specific surface areas (ESA) and total pore volumes (EPV) of ALD-modified SiO<sub>2</sub> based on model 1. Values are compared to the measured values (green) from N<sub>2</sub> physisorption analysis (SA, PV). Mass-uptakes derive from thermogravimetric data.  $ESA = SA(SiO_2)/(1+mass-uptake)$ ,  $EPV = PV(SiO_2)/(1+mass-uptake)$ .

GaO <sub>x</sub> /SiO <sub>2</sub>					InO <sub>x</sub> /SiO <sub>2</sub>			
ALD cycle	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EPV /cm <sup>3</sup> g <sup>-1</sup>	PV /cm <sup>3</sup> g <sup>-1</sup>	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EPV /cm <sup>3</sup> g <sup>-1</sup>	PV /cm <sup>3</sup> g <sup>-1</sup>
0 (SiO <sub>2</sub> )	505		0.79		505		0.79	
1	406	336	0.64	0.57	364	277	0.57	0.55
2	359	296	0.56	0.46	276	216	0.43	0.34
3	326	259	0.51	0.39	221	142	0.35	0.23

## 2.2. Approximation (Geometric, Core-Shell):

The total surface area of silica is concentrated on (non-porous) spheres with a diameter of 4.48 nm (best fit), density of 2.65 g/cm<sup>3</sup> and no inter-particle volume. The sphere diameters are increased by a thin ALD layer of higher density (5.5 g/cm<sup>3</sup> for GaO<sub>x</sub> and 6.75 g/cm<sup>3</sup> for InO<sub>x</sub>, ref.: chapter 3.6). The layer is homogeneously distributed and its thickness is calculated using equation (2) (see chapter 3.6). The non-mass-related surface area of the sphere increases slightly while the mass-related (specific) surface area decreases drastically (Table S3).



**Scheme S2.** Schematic description of model 2 (Core-shell), based on two ALD cycles of InO<sub>x</sub> on SiO<sub>2</sub>.

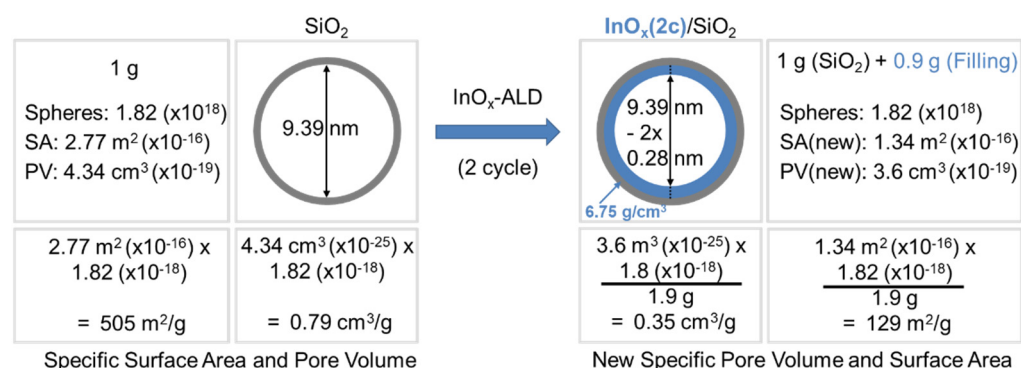
**Table S3.** Estimated (blue) specific surface area (ESA) and mass-uptake (EU<sub>p</sub>) of ALD-modified SiO<sub>2</sub> based on model 2. Values are compared to the measured values (green) from N<sub>2</sub> physisorption analysis (SA). Mass-uptakes (Up) derive from thermogravimetric data.  $ESA = SA(new)/(1 + mass-uptake)$ ,  $EU_p = \Delta m/m_0$  ( $m_0 = SiO_2$ ).

GaO <sub>x</sub> /SiO <sub>2</sub>					InO <sub>x</sub> /SiO <sub>2</sub>			
ALD cycle	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EU <sub>p</sub> /wt%	Up /wt%	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EU <sub>p</sub> /wt%	Up /wt%
0 (SiO <sub>2</sub> )	505		0		505		0	
1	434	336	26	24	399	277	39	39
2	393	296	40	48	308	216	108	83
3	364	259	67	55	252	142	191	129

## 2.3. Approximation (Volumetric):

The total pore volume and surface area of silica is concentrated inside spherical pores with a diameter of 9.39 nm (best fit) without connections between the pores. The number of pores is normalized to  $1.82 \times 10^{18}$  to reach the total pore volume and surface area of 1 g SiO<sub>2</sub> (0.79 cm<sup>3</sup> and 505 m<sup>2</sup>). The pore diameters are decreased by a homogeneously

distributed, thin ALD layers (see chapter 3.6 or equation (2)). Consequently, the inner surface area and pore volume of each pore decreases and the mass-based (specific) values as well (Table S4).

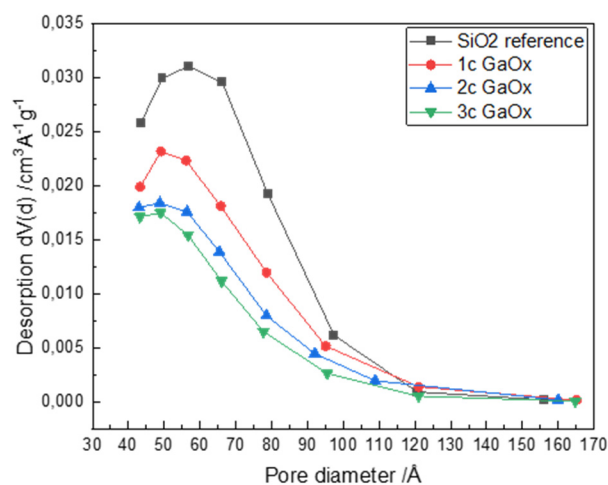


**Scheme S3.** Schematic description of model 3 (volumetric), based on two ALD cycles of InO<sub>x</sub> on SiO<sub>2</sub>.

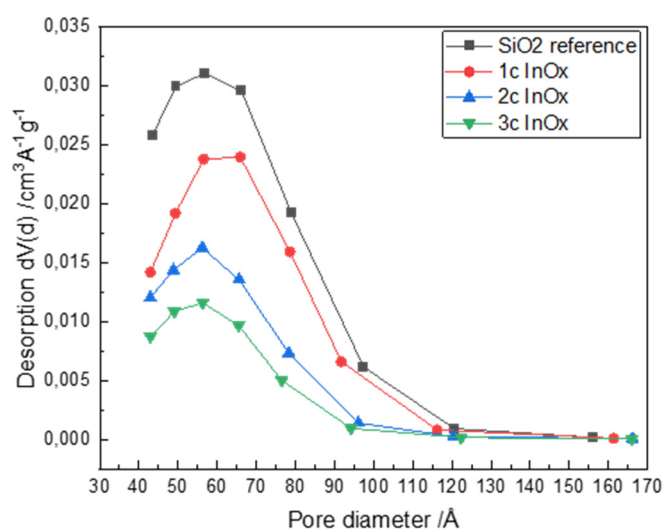
**Table S4.** Estimated (blue) surface area (ESA) and pore volume (EPV) of ALD-modified SiO<sub>2</sub> based on model 3. Values are compared to the measured values (green) from N<sub>2</sub> physisorption analysis (SA).

GaO <sub>x</sub> /SiO <sub>2</sub>							InO <sub>x</sub> /SiO <sub>2</sub>					
ALD cycle	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EPV /cm <sup>3</sup>	PV /cm <sup>3</sup>	EU <sub>p</sub> /wt%	Up /wt%	ESA /m <sup>2</sup> g <sup>-1</sup>	SA /m <sup>2</sup> g <sup>-1</sup>	EPV /cm <sup>3</sup> g <sup>-1</sup>	PV /cm <sup>3</sup> g <sup>-1</sup>	EU <sub>p</sub> /wt%	Up /wt%
0 (SiO <sub>2</sub> )	505		0.79		0	0	505		0.79		0	0
1	214	336	0.60	0.57	25	24	193	277	0.54	0.55	37	39
2	181	296	0.50	0.46	43	48	129	216	0.35	0.34	90	83
3	159	259	0.43	0.39	58	55	93	142	0.24	0.23	142	129

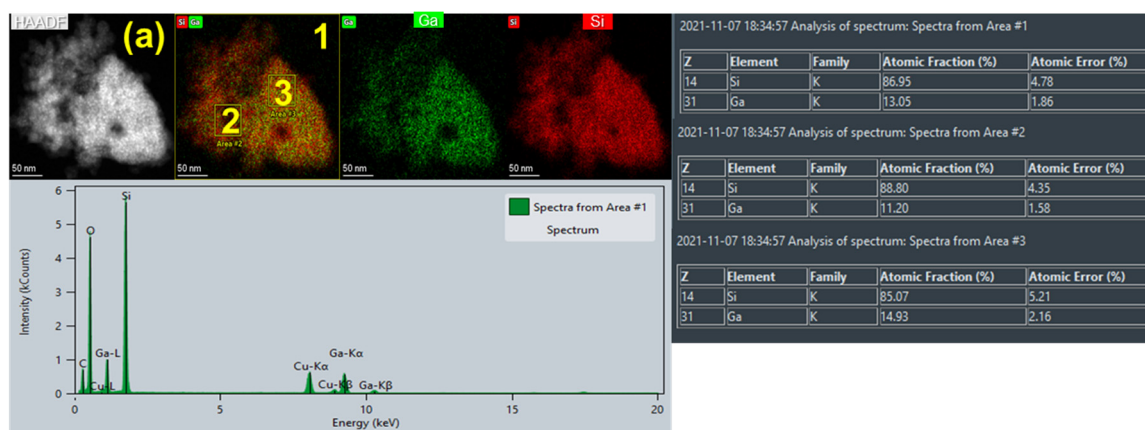
**Conclusion:** The three shown models are only rough approximations and do not fully reflect the reality. An exact model would lie somewhere in-between, yet with unknown proportions. However, the mass-uptakes, changes in specific surface areas and pore volumes are mostly in line with what was observed *via* measurements. Finally, the models justify and rationalize the drastically appearing decrease in surface areas and pore volumes by metal oxide ALD.



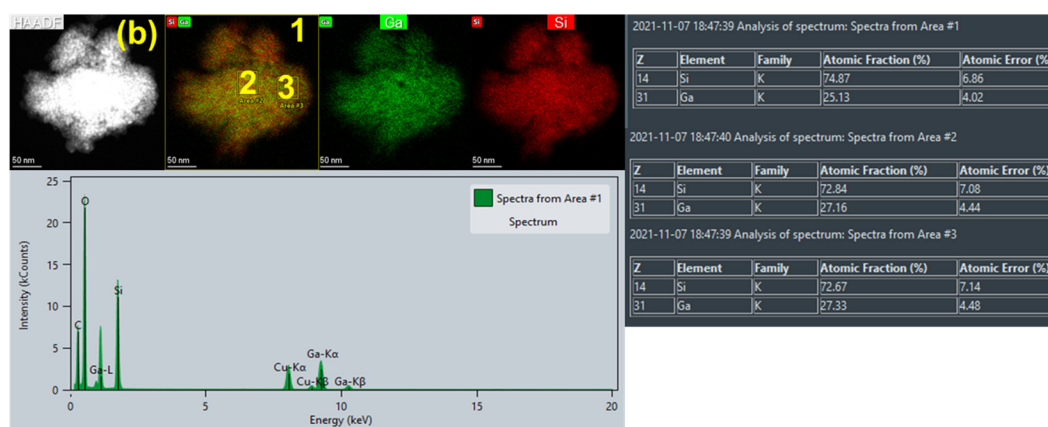
**Figure S2.** Differential pore size distributions of GaO<sub>x</sub> ALD modified SiO<sub>2</sub>, determined by the N<sub>2</sub> desorption branches and application of the BJH method. 1-3 cycles of GaO<sub>x</sub> ALD were conducted using TMG and water at 150°C.



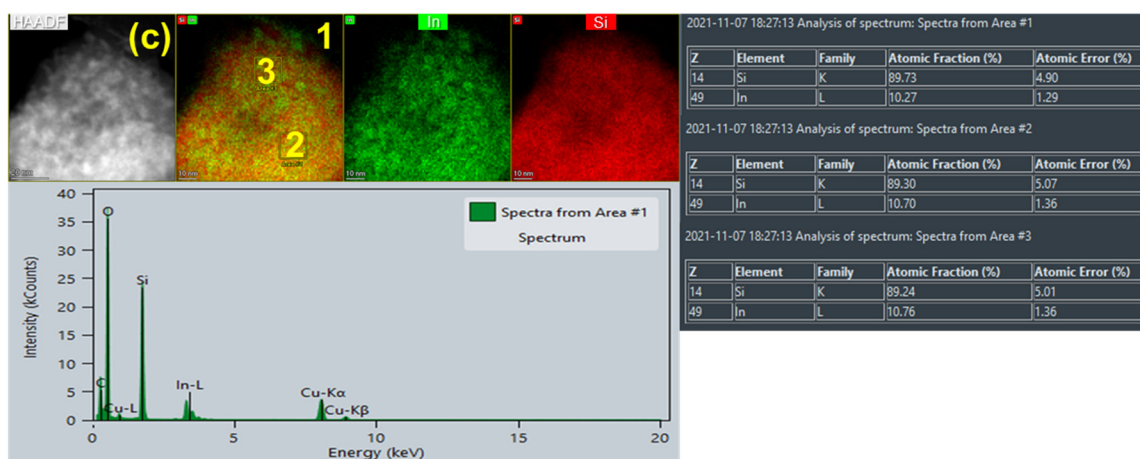
**Figure S3.** Differential pore size distributions of InO<sub>x</sub> ALD modified SiO<sub>2</sub>, determined by the N<sub>2</sub> desorption branches and application of the BJH method. 1-3 cycles of InO<sub>x</sub> ALD were conducted using TMI and water at 150°C.



**Figure S4.** STEM-HAADF image and EDX-mappings of (a) 1 ALD cycle  $\text{GaO}_x$  on mesoporous  $\text{SiO}_2$ . For analysis of Area #1 the whole image was selected.

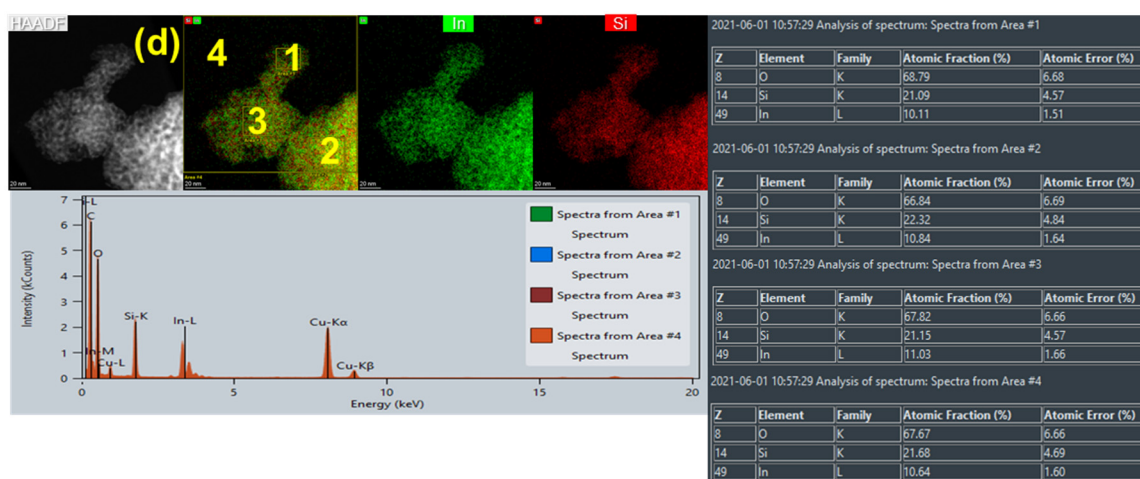


**Figure S5.** STEM-HAADF image and EDX-mappings of (b) 3 ALD cycle  $\text{GaO}_x$  on mesoporous  $\text{SiO}_2$ . For analysis of Area #1 the whole image was selected.

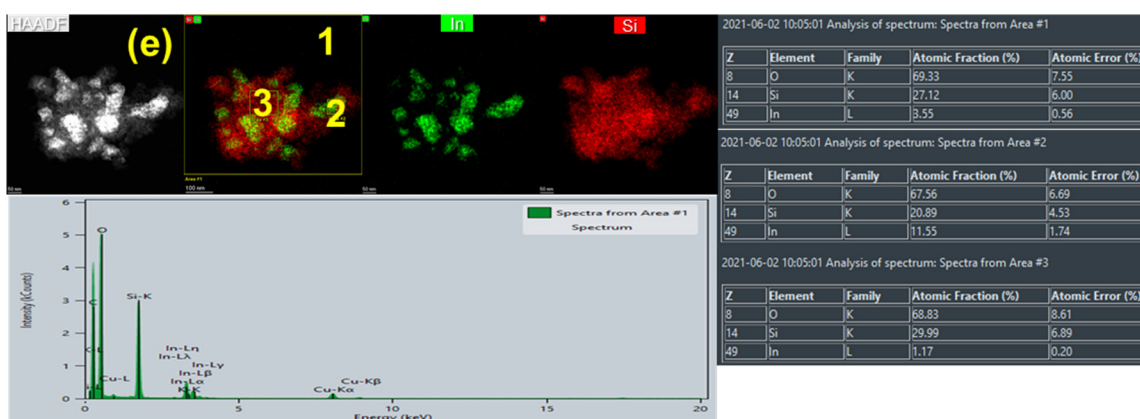


**Figure S6.** STEM-HAADF image and EDX-mappings of (c) 1 ALD cycle  $\text{InO}_x$  on mesoporous  $\text{SiO}_2$ . For analysis of Area #1 the whole image was selected.





**Figure S7.** STEM-HAADF image and EDX-mappings of (d) 3 ALD cycle InO<sub>x</sub> on mesoporous SiO<sub>2</sub>. For analysis of Area #4 the whole image was selected.



**Figure S8.** STEM-HAADF image and EDX-mappings of (e) impregnated In<sub>2</sub>O<sub>3</sub> (22 wt% In) on mesoporous SiO<sub>2</sub>. For analysis of Area #1 the whole image was selected.

**Table S5.** Fit parameters for the XPS scans of the Ga3d, In3d and O1s regions after GaO<sub>x</sub> and InO<sub>x</sub> ALD on SiO<sub>2</sub>. Shown are the peak positions, full width at half maxima (FWHM), the used L/G Mix and area ratios.

Ga3d region (after GaO <sub>x</sub> ALD)							
Sample	O2s			Ga3d (Ga2O3)			Area ratio
	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	
GaO <sub>x</sub> /SiO <sub>2</sub> (1 cycle)	24.68	3.23	30	20.66	2.78	30	0.62 to 1
GaO <sub>x</sub> /SiO <sub>2</sub> (3 cycle)	24.87	3.45	30	20.80	2.72	30	0.30 to 1
In3d region (after InO <sub>x</sub> ALD)							
Sample	In3d <sub>3/2</sub>			In3d <sub>5/2</sub>			Area ratio
	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	
InO <sub>x</sub> /SiO <sub>2</sub> (1 cycle)	452.64	2.38	30	445.10	2.38	30	0.69 to 1
InO <sub>x</sub> /SiO <sub>2</sub> (2 cycle)	452.47	1.89	30	444.93	1.89	30	0.69 to 1
InO <sub>x</sub> /SiO <sub>2</sub>	452.32	1.85	30	444.78	1.85	30	0.69 to 1

(3 cycle)

O1s region (after InO <sub>x</sub> ALD)							
Sample	O-Si (SiO <sub>2</sub> )			O-In (In <sub>2</sub> O <sub>3</sub> )			Area ratio
	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	Peak BE (eV)	FWHM (eV)	L/G Mix (%)	
InO <sub>x</sub> /SiO <sub>2</sub> (1 cycle)	532.25	2.44	30	530.00	2.44	30	1 to 0.05
InO <sub>x</sub> /SiO <sub>2</sub> (3 cycle)	532.15	1.91	30	530.22	1.90	30	1 to 0.30

## References

1. Stempel, V.E.; Knemeyer, K.; Naumann d'Alnoncourt, R.; Driess, M.; Rosowski, F. Investigating the Trimethylaluminium/Water ALD Process on Mesoporous Silica by In Situ Gravimetric Monitoring. *Nanomaterials* 2018, 8, 365. <https://doi.org/10.3390/nano8060365>.