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Fabrication and Characterization of a Self-Powered n-Bi₂Se₃/p-Si Nanowire Bulk Heterojunction Broadband Photodetector

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Abstract: In the present study, vacuum evaporation method is used to deposit Bi₂Se₃ film onto Si nanowires (NWs) to form bulk heterojunction for the first time. Its photodetector is self-powered, its detection wavelength ranges from 390 nm to 1700 nm and its responsivity reaches its highest value of 84.3 mA/W at 390 nm. In comparison to other Bi₂Se₃/Si photodetectors previously reported, its infrared detection length is the second longest and its response speed is the third fastest. Before the fabrication of the photodetector, we optimized the growth parameter of the Bi₂Se₃ film and the best Bi₂Se₃ film with atomic steps could finally be achieved. The electrical property measurement conducted by the physical property measurement system (PPMS) showed that the grown Bi₂Se₃ film was n-type conductive and had unique topological insulator properties, such as a metallic state, weak anti-localization (WAL) and linear magnetic resistance (LMR). Subsequently, we fabricated Si NWs by the metal-assisted chemical etching (MACE) method. The interspace between Si NWs and the height of Si NWs could be tuned by Ag deposition and chemical etching times, respectively. Finally, Si NWs fabricated with the Ag deposition time of 60 s and the etching time of 10 min was covered by the best Bi₂Se₃ film to be processed for the photodetector. The primary n-Bi₂Se₃/p-Si NWs photodetector that we fabricated can work in a self-powered mode and it has a broadband detection range and fast response speed, which indicates that it can serve as a promising silicon-based near- and mid-infrared photodetector.

Keywords: topological insulator; Bi₂Se₃; silicon nanowires; photodetector; bulk heterojunction



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1. Introduction

With the rapid development of the semiconductor industry, semiconductor photodetectors are widely used in the military, for scientific research and in daily life. In order to promote the performance of photodetectors, many revolutionary materials are chosen to fabricate them. Specially, two-dimensional (2D) materials, such as graphene [1] and transition metal dichalcogenides (TMDCs) [2,3], are considered as promising photodetection materials due to their ultra-high carrier mobility, superior integrability and broadband light absorption properties. However, the practical application of a graphene photodetector has been limited by its zero bandgap, which results in an extremely high dark current [1]. The TMDCs photodetectors are limited by their low detectivity of infrared light and complex fabrication techniques [2,3].

The topological insulator (TI) is another kind of novel material with internal insulation and a gapless surface [4,5]. The carriers on its surface are protected from backscattering due to the symmetry of time inversion. It also presents strange physical phenomena, such as spin-orbital coupling, quantum oscillation and weak anti-localization (WAL) effect [5–9]. Therefore, it is widely used in spintronic devices, optoelectronic devices, flexible electrodes and quantum computing [10–13]. Bi₂Se₃ is the first certified material as the second kind of three-dimensional TI, which has the advantage of simple surface state structure. In addition,

its direct bandgap is ~ 0.3 eV, which makes it less susceptible to room-temperature thermal disturbance and can absorb light with wavelengths of up to ~ 4 μm [14]. These characteristics give Bi_2Se_3 the potential to prepare high-performance infrared photoelectric detectors [15].

On the other hand, silicon material has been widely used in the fields of optoelectronic and microelectronic devices. However, it has an indirect bandgap of 1.12 eV at room temperature, which means that it cannot detect infrared light with a wavelength >1.1 μm , including the two important communication wavelengths 1.3 μm and 1.55 μm . In order to extend the silicon detection wavelength, Bi_2Se_3 has been used to deposit onto silicon to form a heterojunction infrared detector [16–19]. Hongbin Zhang et al. [16] fabricated Bi_2Se_3 by the physical vapor deposition (PVD) method inside a horizontal tube furnace, and the $\text{Bi}_2\text{Se}_3/\text{n-Si}$ photodetector with a detection range of 350 nm to 1100 nm, a responsivity of 24.28 A/W, a detectivity of 4.39×10^{12} Jones, a rise time (t_r) of 2.5 μs and a fall time (t_f) of 5.5 μs . Additionally, Xin Hong et al. [17] fabricated Bi_2Se_3 film on a micropylamidal Si substrate also by the PVD method inside a tube furnace, and the photodetector had a response wavelength range of 635 to 2700 nm, a photo-to-dark ratio of 1.01×10^4 , a dark current of 0.11 nA, a t_r of 0.52 ms and a t_f of 0.44 ms. Liu C et al. [18] fabricated Bi_2Se_3 nanowires on the n-Si substrate by an Au-catalyzed vapor–liquid–solid (VLS) method, and its photodetector had a response wavelength range of 380 to 1310 nm, a photo-to-dark ratio of 10, a responsivity of 924.2 A/W, a t_r of 45 ms and a t_f of 47 ms. Since Bi_2Se_3 is an intrinsic n-type material, we believe that n- $\text{Bi}_2\text{Se}_3/\text{p-silicon}$ heterojunction is more favorable for electron transportation [19]. Mingze Li et al. [19] fabricated a Cu-doped n- Bi_2Se_3 film on a p-Si substrate by the chemical vapor deposition (CVD) method inside a high-temperature tube furnace, and its photodetector possessed a detection range of 400 nm to 1200 nm and a responsivity of 168.9 mA/W, a t_r of 4 ms and a t_f of 4 ms. Xujie Pan et al. [20] fabricated a Bi_2Se_3 film on an Si (111) substrate by the molecular beam epitaxy (MBE) method, and its photodetector had a response of 800 nm to 1200 nm, a t_r of 2 ms and a t_f of 2 ms. Biswajit Das et al. [21] used the chemical method to synthesize the Bi_2Se_3 nanoflake (NF) on the Si NWs surface and fabricated a Bi_2Se_3 NF/Si NWs heterojunction photodetector, which had a detection wavelength range of 300 nm to 1000 nm, a responsivity of 934.1 A/W, a detectivity of 2.30×10^{13} Jones, a t_r of 41 ms and a t_f of 79 ms.

To date, however, the vacuum evaporation technique has not been used to deposit Bi_2Se_3 films for a photodetector. In comparison to MBE [20,22], its apparatus is much cheaper and easier to maintain. In comparison to PVD and CVD in a tube furnace or chemical synthesis [16,19,23], it can deposit a large area and uniform film with good reproducibility, and is easier to incorporate into an advanced Si integrated circuit (IC) technology. In order to extend the response wavelength and the responsivity of a Bi_2Se_3 photodetector at the mid-infrared (MIR) zone, a thick (>100 nm) Bi_2Se_3 film with better IR absorption is needed [24]. However, the lifetime of a minority carrier in Bi_2Se_3 is only 50 ps at 300 K [25], and thus the diffusion length of the minority carrier would also be very small. In order to facilitate the collection of a photo-generated carrier and reduce the recombination, in the present study, we plan to deposit a thick Bi_2Se_3 film of 100 nm onto Si NWs to form bulk heterojunction to shorten the diffuse distance of minority carriers to the dissociating interface. In the bulk heterostructure, two materials form interpenetrating networks minimize the distance that excitons must travel before dissociating at a heterointerface [26,27]. As a result, the photo-generated carriers are efficiently collected before they are recombined. In addition, Si NWs can be simply fabricated by the MACE method and its morphology can be easily tuned [28–30].

In this paper, Bi_2Se_3 film was grown by the vacuum thermal evaporation method and its growth parameters were optimized to obtain the best growth condition. The optical and electrical properties of the best Bi_2Se_3 film were investigated. Then, the Si NWs were grown by the MACE method and their interspace and height were tuned by the Ag deposition time and etching time. Finally, the performance of the best n- $\text{Bi}_2\text{Se}_3/\text{p-Si}$ NWs photodetector was measured and analyzed.

2. Materials and Methods

2.1. Sample Preparation

2.1.1. Preparation of Bi₂Se₃ Thin Films

The crystalline Bi₂Se₃ film was deposited on an Si (111) substrate by the vacuum thermal evaporation method. The background vacuum was lower than 3×10^{-5} Pa. Firstly, a p-type <111> single crystal silicon wafer (with a resistivity of $1\text{--}10 \Omega\text{cm}^{-1}$) was cleaned following the standard RCA process. Before it was introduced into the vacuum growth chamber, it was dipped into a hydrofluoric acid (HF) buffer solution (HF:C₂H₅OH = 1:10 (v/v)) to create a H-passivated Si (111) surface. This Si substrate was degassed at 200 °C for 30 min and then it was heated to the growth temperature and kept at this temperature for 2 h to reach the thermal equilibrium before the Bi₂Se₃ growth. High-purity Bi and Se granules (both bought from Zhongnuo Advanced Material (Beijing, China) Technology with a purity of 99.999%) were co-evaporated from two separate Knudsen effusion cells. A series of experiments were conducted at different growth temperatures (220~420 °C) and different ratios of Bi/Se evaporation rates. The evaporation rates of the Bi and Se sources were controlled by their effusion cell temperatures. Following the Bi₂Se₃ depositions, the temperature of the sample holder was cooled down naturally at an average rate of 1 °C/min.

2.1.2. Preparation of Si NWs

The MACE method was used to fabricate the Si NWs by using Ag nanoparticles (NPs) as the catalyst. The P-type double-polished Si (100) wafer with a resistivity of $1\text{--}10 \Omega\text{cm}^{-1}$ was used as the start material. Firstly, the Si (100) wafer was put into the water solution mixture of HF (5 mol/L) acid and AgNO₃ (0.01 mol/L) to deposition Ag NPs at the sample surface. Secondly, the sample was put into the water solution mixture of HF (5 mol/L) acid and H₂O₂ (0.2 mol/L) to etch the silicon surface to produce the Si NWs. Thirdly, the HNO₃ acid (6.54 mol/L) water solution was used to remove the remain Ag NPs inside the Si NWs. The height and density of Si NWs can be controlled by the etching time and diameter (deposition time) of Ag NPs, respectively. Lastly, the Si NWs samples were dried in a drying oven and were then ready for characterization.

2.1.3. Fabrication of the n-Bi₂Se₃/p-Si NWs Photodetector

The prepared Si NWs samples were used as the substrate for the Bi₂Se₃ film deposition. Then, an interdigital Ag film was deposited on the Bi₂Se₃ surface as the front-side metal electrode and a fully covered Ag film was deposited onto the Si surface as the backside metal electrode. Both of the Ag films were deposited by magnetron sputtering. The purity of the Ag piece was 99.99% and high-purity (5N) Ar gas was introduced at a rate of 50 sccm. The work pressure of the vacuum system was stabilized at around 1.4×10^{-2} mbar. The power of the radio-frequency (RF) source was 60 W. The deposition thickness of the Ag films was 60 nm.

2.2. Sample and Device Testing Method

The Bi₂Se₃ thin films were characterized by X-ray diffraction (XRD, D8 Advance, Bruker, Berlin, Germany), atomic force microscope (AFM, Multimode 8, Bruker, Berlin, Germany), Raman (inVia, Renishaw, New Mills, UK), X-ray photoelectron spectroscopy (XPS, AXIS, SUPRA+, Shimadzu, Milton Keynes, UK), scanning electron microscope (SEM, JEM-F200, JEOL, Tokyo, Japan) and PPMS (9 Ever Cool II, Quantum Design, San Diego, CA, USA). The Si NWs and the Bi₂Se₃/Si NWs were characterized by SEM (JEM-F200, JEOL, Tokyo, Japan) and UV-VIS-NIR Spectroscopy (Lambda 950, PerkinElmer, Akron, OH, USA). The performance of the device was tested by a semiconductor characteristics system (4200-SCS, Keithley, Beaverton, OR, USA) and a spectral response test system, which consisted of the following parts: light source (model 7ILT250 halogen tungsten light source with a power of 250 W, Sofn Instruments, Beijing, China), chopper (model 3501 Optical Chopper, Newport, Andover, MA, USA), monochromator (model 7ISW151 dual raster scanning spectrometer, Sofn Instruments, Beijing, China), lock-in amplifier (model SR830

DSP, Stanford Research Systems, Sunnyvale, CA, USA), DC power supply, one-dimensional lift table for sample testing, standard detector (model DET 36A, Thorlabs, Newton, NJ, USA), optical path components and related test software [31].

3. Results and Discussion

3.1. Growth and Characterization of Bi_2Se_3 Films

In order to obtain the optimal growth conditions for the Bi_2Se_3 film, we deposited the Bi_2Se_3 film on the Si (111) surface at different temperatures of the sample holder and different Bi/Se beam ratios. The Bi/Se beam ratios were realized by controlling the evaporation temperatures of Bi and Se source cells. The crystal phase, atomic vibration and surface morphology of the Bi_2Se_3 film were characterized by XRD, Raman and AFM, respectively. When the temperatures of the Se cell and sample holder were fixed at 209 °C and 320 °C, respectively, we deposited the Bi_2Se_3 film on the Si (111) substrate at different Bi cell temperatures from 650 to 850 °C with an interval temperature of 50 °C. The XRD, Raman and AFM of the obtained Bi_2Se_3 films are shown in Figure 1. From Figure 1a, we can observe that, at the Bi cell temperatures of 800 °C and 850 °C, the stoichiometric Bi/Se ratio inside the obtained BiSe alloy film is greater than that in Bi_2Se_3 . When the Bi cell temperature is at 700 °C, the diffraction peaks of the Bi_2Se_3 film are very weak. For the Bi cell temperature of 650 °C, no diffraction exists. When the Bi cell temperature is 750 °C, however, the diffraction peaks of Bi_2Se_3 are very strong and they are at 9.34°, 18.6°, 37.52°, 47.44°, 57.68° and 68.51°, corresponding to the (003), (006), (0012), (0015), (0018) and (0021) planes of the Bi_2Se_3 film. Only the diffraction peaks of the (00N) plane appear in Figure 1a, which indicates that the Bi_2Se_3 film grows strictly in the direction of the c axis [32,33].

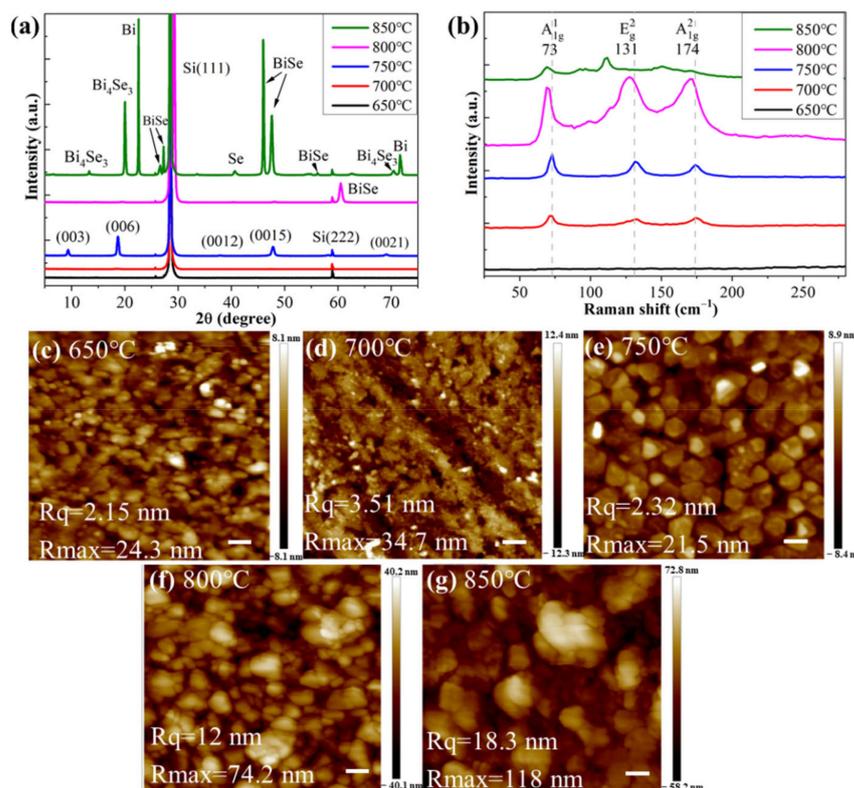


Figure 1. Characterization of Bi_2Se_3 films grown at different Bi cell temperatures (650–850 °C) and at the same growth temperature of 320 °C and Se source cell temperature of 209 °C: (a) XRD; (b) Raman; (c–g) AFM of sample surface. Scale bars in (c–g) are all 100 nm.

Figure 1b shows the Raman spectroscopy of the Bi_2Se_3 films. For the Bi_2Se_3 film grown at a Bi cell temperature of 750 °C, we can observe three distinct Raman shift

peaks at 73 cm^{-1} , 131 cm^{-1} and 174 cm^{-1} , which can be assigned to A_{1g}^1 , E_g^2 and A_{1g}^2 , respectively [34]. As is well known, the topological insulator Bi_2Se_3 is a 2D material. Each of its units is composed of five layers of atoms and can be denoted as the Se^I , Bi, Se^{II} , Bi and Se^I layers. The A_{1g} vibrational mode is caused by the out-of-plane symmetric stretching in the same direction of the outer pair of atoms (Se^I -Bi) or (Bi- Se^I). When the outer pair of atoms, (Se^I -Bi) or (Bi- Se^I), stretch symmetrically in the opposite direction, the A_{1g}^2 vibrational mode is produced. E_g^2 is caused by the in-plane symmetric stretching of the outer layer atoms (Se^I -Bi) or (Bi- Se^I). During these atomic vibration modes, the central Se^{II} atomic layer always remains stationary [35,36]. For the BiSe sample grown with Bi cell temperatures of $800\text{ }^\circ\text{C}$ or $850\text{ }^\circ\text{C}$, many vibrational modes induced from other Bi_xSe_y alloys appeared. For the Bi cell temperature of $700\text{ }^\circ\text{C}$, the three vibrational modes were weaker than those at $750\text{ }^\circ\text{C}$, indicating the thinner film of Bi_2Se_3 and thus the lower deposition rate. According to the previous literature [37], the growth rate of Bi_2Se_3 film depends on the Bi deposition rate in an Se-rich atmosphere (evaporation rate: Se:Bi = $\sim 10:1$). The Bi deposition rate depends on the temperature of the Bi source cell. When the Bi cell temperature is $700\text{ }^\circ\text{C}$, the XRD diffraction peak of the Bi_2Se_3 film cannot be observed in Figure 1a, but its weak Raman peaks can be observed in Figure 1b. These findings indicate that the thickness of the deposited Bi_2Se_3 film is very thin. For the Bi cell temperature of $650\text{ }^\circ\text{C}$, however, none of the three characteristic vibrational modes of Bi_2Se_3 exist, which demonstrates that no Bi_2Se_3 film was deposited on the Si surface, and thus no Bi was evaporated at this cell temperature. In general, the Raman results for the Bi_2Se_3 film are consistent with their XRD results presented in Figure 1a.

Figure 1c–g show the surface morphology of Bi_2Se_3 film grown with different Bi cell temperatures. Only at $750\text{ }^\circ\text{C}$ did the Bi_2Se_3 film have the obvious angularity of crystalline grain and the atomic layers, which indicates that the Bi_2Se_3 film follows a layer-by-layer growth mode in this grow condition. At the Bi cell temperatures of 800 and $850\text{ }^\circ\text{C}$, as shown in Figure 1f,g, a large size of agglomerate BiSe NPs exist. At the Bi cell temperature of $650\text{ }^\circ\text{C}$, we can observe from Figure 1a,b that there is no Bi_2Se_3 at the film surface. Therefore, the NPs should be some remaining Se. At the Bi cell temperature of $700\text{ }^\circ\text{C}$, the surface should exhibit a few Bi_2Se_3 NPs.

Following the above optimization, we discovered that the optimal cell temperature for Bi evaporation was $750\text{ }^\circ\text{C}$. Then, we fixed the temperature at $750\text{ }^\circ\text{C}$ and tuned the Se cell temperature from $194\text{ }^\circ\text{C}$ to $219\text{ }^\circ\text{C}$ at a temperature interval of $5\text{ }^\circ\text{C}$. The XRD, Raman and AFM results of BiSe films are shown in Figure S1 (see the Supplementary Materials). Figure S1a shows that the deposited BiSe films are all in composition of Bi_2Se_3 . The strongest of diffraction peak comes from the sample grown at the Bi cell temperature of $209\text{ }^\circ\text{C}$, which demonstrates that this Bi_2Se_3 film has the best crystallinity. The Raman results in Figure S1b show that the strongest Raman peak also comes from the Bi_2Se_3 film grown at an Se cell temperature of $209\text{ }^\circ\text{C}$, consistent with the results presented in Figure S1a. Figure S1c–h show that only the Bi_2Se_3 film grown in Se cell temperatures of $204\text{ }^\circ\text{C}$ or $209\text{ }^\circ\text{C}$ have the atomic step at the film surface, and thus should express a layer-by-layer growth mode. This illustrates that, when the Bi and Se cell temperatures are $750\text{ }^\circ\text{C}$ and $204\text{ }^\circ\text{C}$ (or $209\text{ }^\circ\text{C}$), respectively, they have the best Bi/Se beam ratio to deposit the 2D materials, Bi_2Se_3 film. Note that, in Figure S1e, there are many triangle Bi_2Se_3 crystalline islands, which agrees well with the triangle atomic arrangement at the Si (111) surface, and so indicates that the Bi_2Se_3 film has a good epitaxy relationship with the underneath Si substrate.

Subsequently, we fixed the Bi cell temperature at $750\text{ }^\circ\text{C}$ and Se cell temperature at $209\text{ }^\circ\text{C}$, and then began to optimize the growth temperature for the Bi_2Se_3 film. The growth temperature was changed from $220\text{ }^\circ\text{C}$ to $420\text{ }^\circ\text{C}$ in a temperature interval of $50\text{ }^\circ\text{C}$. Figure S2a,b show that the Bi_2Se_3 sample grown at $270\text{ }^\circ\text{C}$ has the highest diffraction and vibration peaks, which indicates that it has the best crystallinity. Figure S2c,g show that there are many atomic steps on the surface of the Bi_2Se_3 films grown at $220\text{ }^\circ\text{C}$, $270\text{ }^\circ\text{C}$, $320\text{ }^\circ\text{C}$ and $370\text{ }^\circ\text{C}$. For the film grown at $420\text{ }^\circ\text{C}$, there is no Bi_2Se_3 on the surface, as shown in

Figure S2a,b, possibly due to the growth temperature that is too high to absorb the arriving Se atom.

Following the growth optimization, we deposited an optimal Bi_2Se_3 film sample with a Bi cell temperature of 750°C , an Se cell temperature of 209°C and a growth temperature of 270°C . The deposition time was 15 min and the sample thickness was 30 nm. Its XRD, Raman, XPS and AFM results are shown in Figure 2. The XRD and Raman results determine that the Bi_2Se_3 film has a good purity and crystallinity. Figure 2c,d show the XPS spectra of Bi 4f and Se 3d, respectively. The Bi 4f spectrum was fitted with Bi 4f_{7/2} and Bi 4f_{5/2} peaks and present at binding energies of 158 and 163.3 eV, respectively. The weak fitting peaks centered at 159.1 and 164.25 eV showed a slight oxidation of the Bi_2Se_3 film [38]. The Se 3d spectrum can be perfectly fitted by Se 3d_{5/2} and Se 3d_{3/2} peaks present at binding energies of 53.5 and 54.4 eV, respectively [39]. The integral areas of Bi 4f and Se 3d showed an Se atomic percentage of 57.6% and a Bi atomic percentage of 42.4%, corresponding to an Se/Bi atomic ratio of 1.36. This is consistent with the EDS analysis shown in Figure S3e and again proves a lack of Se in the Bi_2Se_3 film. Figure 2e shows that, beside the triangle Bi_2Se_3 island, there are many atomic steps on the sample's surface. Then, we zoomed into a small zone (circled by a red rectangle in Figure 2e) in the AFM and measured its profile along the white line, as shown in Figure 2f. The atomic step profile is shown in Figure 2g. Two atomic steps with a height of 1 nm can be observed, of which the value agrees well with the previous literature [40,41]. That is to say, we can obtain the 2D materials, Bi_2Se_3 film, at the optimal growth condition.

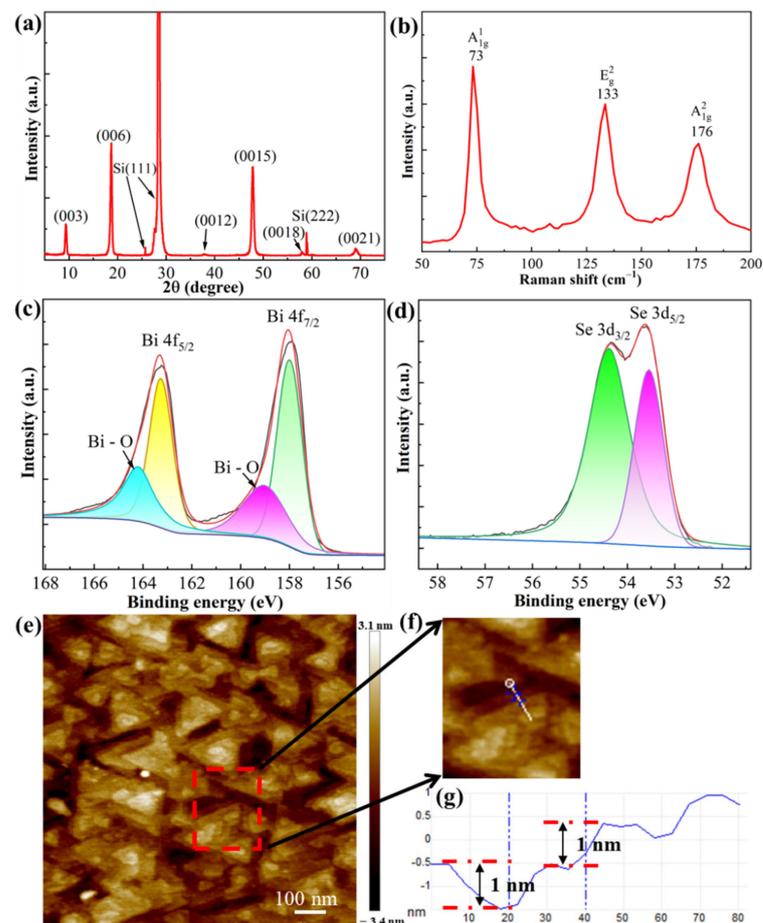


Figure 2. Bi_2Se_3 film grown at the optimal condition: Bi cell temperature of 750°C , Se cell temperature of 204°C and growth temperature of 320°C . (a) XRD; (b) Raman; (c,d) XPS of Bi 4f and Se 3d with the fitting curves; (e) AFM of Bi_2Se_3 film surface and (f) local enlargement; (g) atomic step height.

In order to further investigate the chemical composition of the optimal Bi_2Se_3 film, we performed an SEM measurement. Figure S3a shows the SEM of the sample surface. Figure S3b–d shows the elemental distribution of the Bi_2Se_3 film. We can observe that the Bi and Se elements distribute very uniformly. Figure S3e presents that the atomic percentages of Se and Bi are 59.39% and 40.61%, respectively. Their atomic ratio was 1.46, less than the stoichiometric ratio of 1.5 in Bi_2Se_3 , possibly due to the effumability of the Se atom (melting point: 217 °C) at a growth temperature of 270 °C [21].

3.2. Analysis of Electrical Properties of the Samples

In order to avoid the effect of the Si substrate on the electrical transport property of the Bi_2Se_3 film, we deposited the Bi_2Se_3 film at the optimal growth condition on the insulative SrTiO_3 (STO) substrate. The schematic of the sample and electrode is shown in Figure 3a. Figure 3b shows the curve of the longitudinal resistance (R_{xx}) of the sample as a function of temperature. The resistance of the sample increased as the temperature increased from 13 K to 300 K, indicating the metal properties of the sample, which is consistent with the results reported in the literature [42]. However, between 3 K and 13 K, the resistance increased as the temperature decreased. It may be attributed to the strong correlation between electrons at low temperatures, which causes a Mott gap, leading to the metal-insulation transition [43,44]. Figure 3c shows that the Hall resistance of the sample (R_{xy}) changes linearly with the magnetic field in the temperature range of 2~100 K, and its slope is negative, indicating that the Bi_2Se_3 is in the form of n-type conductivity, which may be due to Se vacancies.

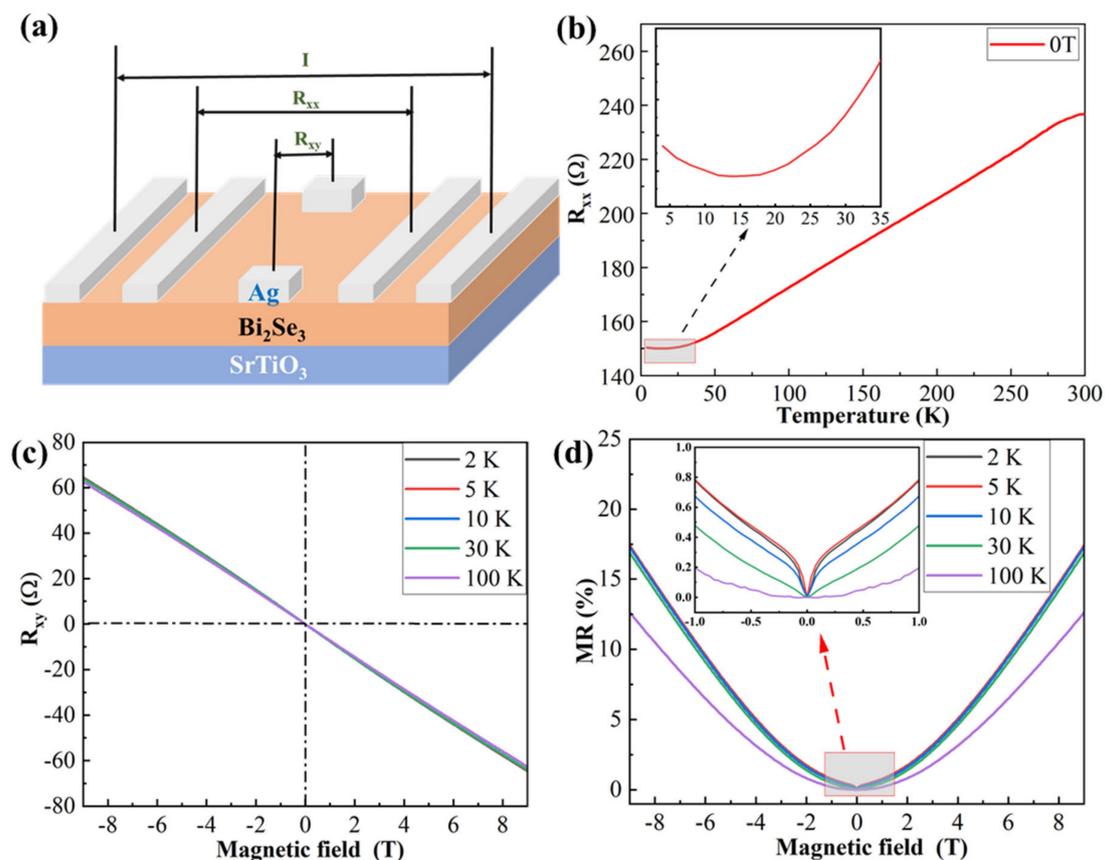


Figure 3. Electrical transport measurement of Bi_2Se_3 thin film. (a) Schematic diagram of sample and electrodes. (b) R_{xx} curves as a function of temperature under magnetic fields of 0 T and 9 T (the inset is an enlarged view at <35 K); (c) Hall resistance curves as a function of magnetic field at different temperatures; (d) magnetoresistance curve of the sample as a function of magnetic field (the inset is an enlarged view of the low magnetic field $-1\sim 1$ T).

Figure 3d shows the variation of the normalized magnetoresistance (MR) with the magnetic field of the sample in the temperature range of 2~100 K. The definition of MR is $MR = [R(B) - R(B = 0)]/R(B = 0) \times 100\%$ [45]. The MR slowly changes between 0 T and 3 T, and increases linearly between 3 T and 9 T. This LMR is related to the scattering of massless Dirac fermions on the surface and also to the gapless surface states of TI. Abrikosov proposed a model based on quantum MR, which can explain the LMR effect when the magnetic field is strong enough that all Dirac fermions are quantized to the lowest Landau level [46]. The unusual LMR suggests that our thin film may be characterized by gapless surface states and Dirac fermions. It is worth noting that, at 10 K and below, the MR of the sample sharply increases near 0 T, which is caused by the WAL effect of TIs [47]. When the temperature exceeds 10 K, the change in the trend of MR has a quadratic dependence with B; that is, a parabolic MR appears, which is mainly due to the weakening of the WAL effect and the Lorentzian deflection of the carriers [48]. The above results show that the Bi_2Se_3 films grown on STO substrates have the general properties of topological insulators, also indicating the high-quality growth of the Bi_2Se_3 films.

3.3. Results Analysis of $\text{Bi}_2\text{Se}_3/\text{Si}$ NWs Heterojunction

If depositing the Bi_2Se_3 film on the Si NWs' surface, it can enter the interspace between the NWs and form the bulk heterojunction optoelectronic device. The bulk heterojunction can usually facilitate the separation and transportation of the photon-generated carriers. Firstly, we changed the deposition time of Ag NPs to tune the interspace between Si NWs. With the different reaction times from 20 s to 100 s, the $\text{AgNO}_3 + \text{HF}$ solution deposits different sizes and densities of Ag NPs, as shown in Figure S4. It can be observed that the surface coverage of Ag NPs continuously increases with the deposition time. Then, the Ag NPs samples were put into $\text{HF} + \text{H}_2\text{O}_2$ solution to etch 10 min. The sample surface and cross-section morphology of the Si NWs are shown in Figures S5 and S6, respectively. Naturally, we can observe that the interspace between Si NWs increases with the deposition time of Ag NPs. Cross-section SEM verifies this increase in interspace with the Ag NPs deposition time. Secondly, we changed the etching time to tune the height of Si NWs. The surface and cross-section morphology of Si NWs with different etching times from 4 to 14 min and the same Ag NPs deposition time of 60 s are shown in Figures S7 and S8, respectively. We can observe that the Si NWs have the same densities, but with different heights from 1.2 μm to 4.7 μm . According to Figure S8, we plotted the height of Si NWs as a function of the etching time, as shown in Figure S9. The height of Si NWs continuously increases with the etching time.

Then, we deposited the 100 nm thick Bi_2Se_3 films on the Si NWs' surface. Figures S10 and S11 show the surface and cross-section SEM of the $\text{Bi}_2\text{Se}_3/\text{Si}$ NWs samples with the same etching time of 10 min, but the different deposition time of Ag NPs from 20 s to 100 s. From Figure S5, we can observe that the interspace between Si NWs increases with the Ag deposition time. As the interface increases, a part of the Bi_2Se_3 particles enters the interspace increase and the part on the surface of the Si NWs simultaneously decreases, as shown in Figure S11. From Figure S10, we can observe that the part of Bi_2Se_3 particles on the surface of Si NWs is large enough to form a continuous film for Ag deposition times from 20 s to 60 s, as shown in Figure S10a–d, but there are many apertures inside the surface Bi_2Se_3 film for the Ag deposition time of 100 s, as shown in Figure S10e. In the meantime, we can find from the SEM that the surface roughness increases qualitatively with the Ag deposition time. Figures S12 and S13 show the surface and cross-section SEM of the $\text{Bi}_2\text{Se}_3/\text{Si}$ NWs samples with the same deposition times (60 s) of Ag NPs, but different etching times of 4 min to 14 min. Similarly, the interspace between Si NWs also slightly increases with the etching time, as shown in Figure S7. Therefore, the Bi_2Se_3 film on top of Si NWs has many apertures for the longer etching time, such as 12 min and 14 min, as shown in Figure S12e,f. In the meantime, we can find from the SEM that the surface roughness increases qualitatively with the etching time. For the fabrication of the $\text{Bi}_2\text{Se}_3/\text{Si}$ NWs photodetector, a part of the Bi_2Se_3 film needs to enter the interface of Si

NWs to form the bulk heterojunction, and part of them should stay on the surface of the Si NWs to support the formation of the metal electrode. Finally, we chose the Bi₂Se₃/Si NWs sample with an Ag deposition time of 60 s and an etching time of 10 min to process the photodetector. For the sake of simplicity, next, we only show its morphology, optical absorption and device performance.

Figure 4a,b show the surface and cross-section SEM of Si NWs with the deposition time of 60 s for Ag NPs and the etching time of 10 min, respectively. It can be observed that the interspace between the Si NWs is uniform and also large enough for subsequent Bi₂Se₃ entering to form the bulk heterojunction. The height of Si NWs is about 2.8 μm. Following the Bi₂Se₃ deposition, the sample's surface and cross-section SEM images are shown in Figure 4c,d. The surface of Si NWs is covered with Bi₂Se₃ nanosheets deposited at the optimal growth condition. The randomly distributed Bi₂Se₃ nanosheet has a diameter of ~450 nm and a thickness of ~70 nm. Figure 4d shows that the Bi₂Se₃ enters into the Si NWs array with a depth of 1 μm. Additionally, there are still Bi₂Se₃ films on top of the Si NWs. The optical properties of Si NWs and Bi₂Se₃/Si NWs were also measured, as shown in Figure 4e,f. From Figure 4e, it can be observed that the reflectance of Si NWs is ~10%, far lower than that of ordinary Si wafers [49]. This verifies that the Si NWs has a good antireflection function in the wavelength range of 350 nm to 1000 nm. However, it increased to be >10% after the deposition of the Bi₂Se₃ film; at wavelength >1000 nm, the reflectance of Si NWs suddenly increased to ~60% due to its non-absorption, which is similar with that of the Si wafer [50]. The reflectance of the Bi₂Se₃ film was about 10~20% in the measurement rang. Figure 4f shows that Si NWs can absorb light at a wavelength <1000 nm, but Bi₂Se₃/Si NWs can absorb light up to 2500 nm due to the direct bandgap of 0.3 eV (with an absorption edge of 4.1 μm).

Following the covering of the Bi₂Se₃ film on Si NWs, we can observe that the reflectance increased because the deposited Bi₂Se₃ entered the interspace between the Si NWs, which prevented part of the light from entering the Si NWs, and thus reduced the antireflection function. In the wavelength >1100 nm, the reflectance decreased considerably from 60% for the Si NWs to 10% for the Bi₂Se₃/Si NWs, due to the much smaller bandgap (0.3 eV) of Bi₂Se₃ than that (1.1 eV) of silicon. Figure 4f shows that, when the wavelength is shorter than 1000 nm, both Si NWs and Bi₂Se₃/Si NWs have a transmittance of 0 since both of them have a bandgap smaller than 1.1 eV (corresponding to an absorption edge of 1100 nm), and thus the incident light is completely absorbed. At the wavelength >1100 nm, the transmittance in the Si NWs is ~30%, but in the Bi₂Se₃/Si NWs sample it is only less than 10%. The reason is that the Bi₂Se₃ has a bandgap of 0.3 eV and so it can absorb the light with a wavelength shorter than 4.1 μm in theory.

3.4. Performance of the n-Bi₂Se₃/p-Si NWs Photodetector

The n-Bi₂Se₃/p-Si NWs sample was covered by an Ag interdigital electrode at the front side and by the fully covered Ag film at its back side, as shown in Figure 5a. Then, a simple photodetector was completed. Figure 5 shows the performance of a typical device, and the whole fabrication process is schematically shown in Figure 5a. Figure 5b shows the responsivity of the n-Bi₂Se₃/p-Si NWs photodetector at the wavelength range of 390 nm to 1700 nm. The highest responsivity is 84.3 mA/W at the initial wavelength of 390 nm. Then, it drops linearly to 2.68 mA/W at 528 nm. Following this, it rises to 14.17 mA/W at 960 nm and drops again to 1.15 mA/W at 1200 nm. Subsequently, it increases again to 7 mA/W at 1700 nm. Figure 5c shows the detectivity of the n-Bi₂Se₃/p-Si NWs photodetector at the wavelength range of 390 nm to 1700 nm. The largest detectivity is 1.06×10^{10} Jones at the initial wavelength of 390 nm. Then, it drops linearly to 3.35×10^8 Jones at 528 nm. Subsequently, it rises to 1.79×10^9 Jones at 960 nm and drops again to 1.44×10^8 Jones at 1200 nm. Then, it increases again to 8.28×10^8 Jones at 1700 nm. In order to explain the complex change of responsivity, we drew the absorption coefficient and absorption depth of Bi₂Se₃ and Si in this wavelength range [24,25], as shown in Figure S14. In the wavelength range of 400 to 1700 nm, the absorption coefficient of Bi₂Se₃ was between

10^6 to 10^5 cm^{-1} , while that of Si rapidly dropped from 10^4 at ~ 400 nm to 10^{-8} at ~ 1500 nm. Additionally, in view of the cross-section morphology shown in Figure 4c,d, we considered that the change in responsivity can be explained as follows: (1) at ~ 400 nm, the light is mainly absorbed by the surface Bi_2Se_3 compact film. The photo-generated electrons can be transported efficiently to the Ag front electrode through the surface Bi_2Se_3 compact film, while the photo-generated holes can be transported efficiently to the Ag back electrode through Si NWs and the Si substrate, as shown by the energy-band diagrams shown in Figure S15. Therefore, the responsivity is high. (2) From ~ 400 nm to ~ 528 nm, the reflectivity of the $\text{Bi}_2\text{Se}_3/\text{Si}$ NWs sample increased, as shown in Figure 4e; the surface Bi_2Se_3 compact film cannot absorb the incident light enough and the leaky light is absorbed by the loose Bi_2Se_3 film inside the Si NWs interspace. In this situation, the transportation of the photo-generated carriers to the Ag electrode is limited due to the loose contact between the Bi_2Se_3 nanosheets and between them with Si NWs. (3) From ~ 528 nm to ~ 1200 nm, the leaky part of the incident light begins to be absorbed by the Si substrate. The photo-generated electrons diffuse through Si NWs and drift through Bi_2Se_3 to reach the Ag front electrode, while the photo-generated holes are efficiently transported to the Ag back electrode. The responsivity reaches the local largest value of 14.17 mA/W. (4) From ~ 1200 nm to 1700 nm, the light can only be absorbed by the Bi_2Se_3 film and the responsivity increases again. Therefore, we believe that part of the absorption by Si NWs does not contribute to the light current. This can be easily understood because there are many broken Si NWs (see Figure 4d) that do not make contact with the above Bi_2Se_3 film, and so the produced carrier cannot be transported to the front electrode.

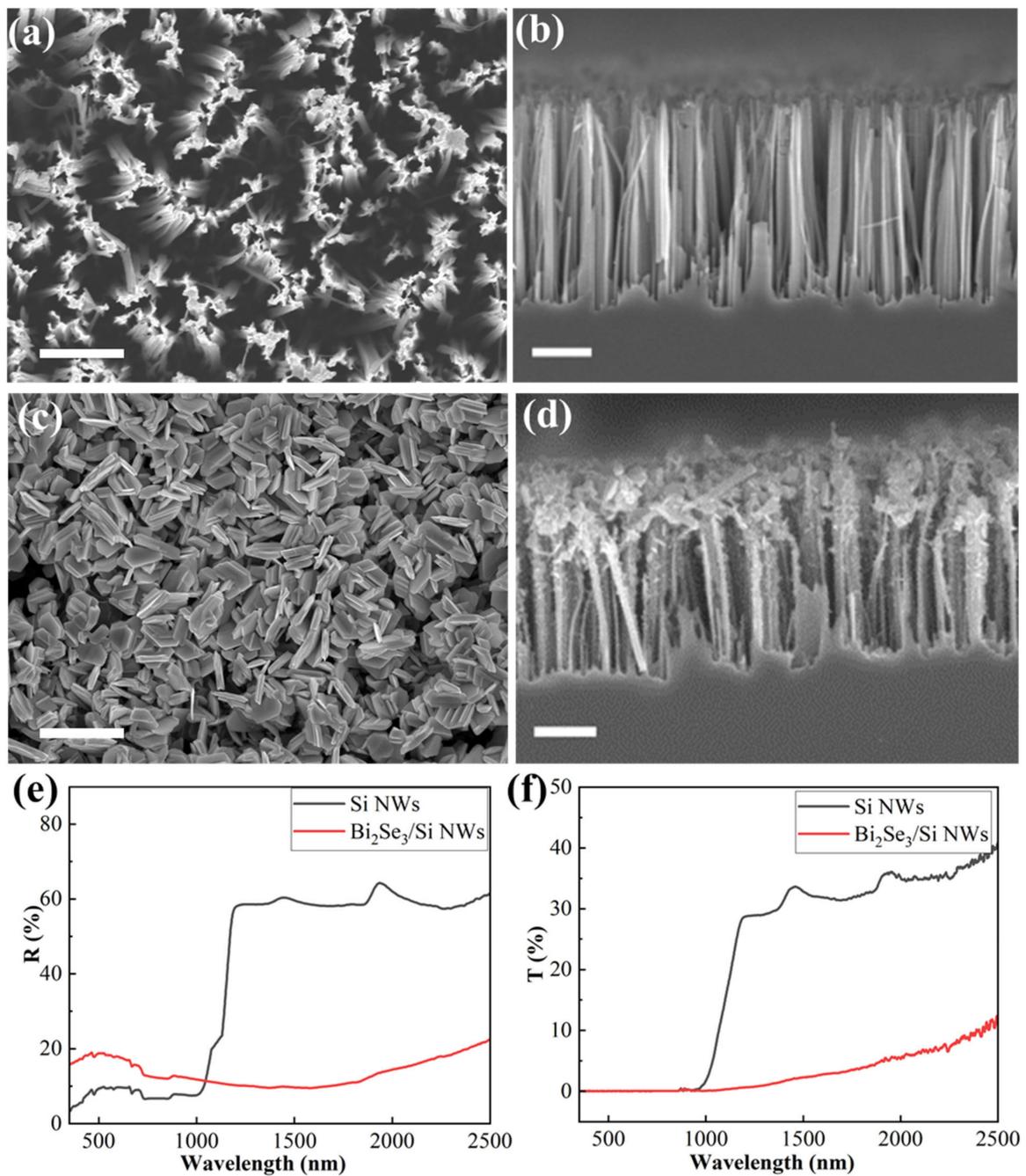


Figure 4. (a) Surface and (b) cross-section SEM images of Si NWs with the deposition time of 60 s for Ag NPs and the etching time of 10 min; (c) surface and (d) cross-section SEM images of Bi₂Se₃/Si NWs heterojunction with the deposition time of 60 s for Ag NPs and the etching time of 10 min; (e) reflectance and (f) transmittance of Si NWs and Bi₂Se₃/Si NWs samples. Scale bars in (a–d) are all 1 μm.

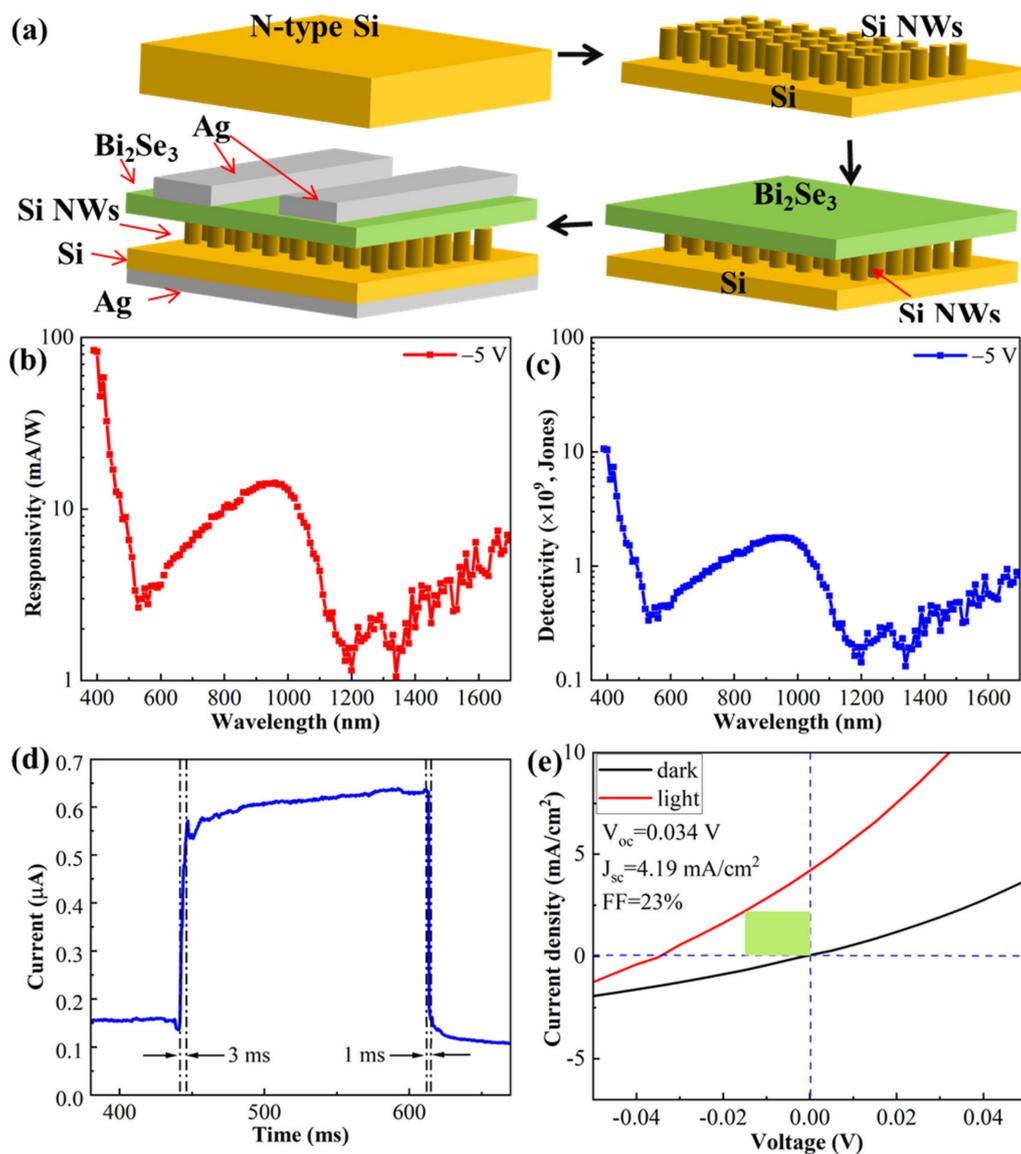


Figure 5. Characterization of n-Bi₂Se₃/p-Si photodetector: (a) its fabrication process; (b) its responsivity and (c) its detectivity at the wavelength range from 390 nm to 1700 nm; (d) its response times and (e) its photovoltaic behavior measured under a 980 nm laser illumination with a laser power of 0.6 W.

Figure 5d shows the transient photocurrent measured on the n-Bi₂Se₃/p-Si NWs photodetector under the illumination of a 980 nm laser with a power of 0.6 W. The forward voltage is 1 V. Its rise time is 3 ms and its fall time is 1 ms. In view of the long diffusion length of the carrier, this response is very rapid, which can mainly be attributed to the function of bulk heterostructure. Figure 5e shows its photovoltaic behavior under the illumination of a 980 nm laser (0.6 W). The power density of the laser was about 92.3 W/cm². As shown in Figure 5e, V_{oc} is 0.034 V, J_{sc} is 4.19 mA/cm² and FF is 23% yielding a power conversion efficiency (PCE) of $3.59 \times 10^{-5}\%$. These results imply that our n-Bi₂Se₃/p-Si photodetector can work in a self-powered mode with a zero external bias voltage, although its PCE is still very small at present.

We summarize the performances of Bi₂Se₃/Si photodetectors previously reported by others and ourselves in Table S1. For the type of Bi₂Se₃/p-Si photodetector, our detector has the longest detection wavelength and the fastest rise and fall times than those reported by and M. Li et al. [19]. We considered that the longer detection wavelength can be attributed

to the strong IR absorption (~75%, see Figure 4e,f) for our sample that came from the closely assembled Bi₂Se₃ film, taking advantage of the vacuum evaporation method. The rapid rise and fall times can be ascribed to the bulk heterojunction that can facilitate the dissociating of photo-generated carriers and their transportation to the metal electrode. In comparison to the performance of the Bi₂Se₃/n-Si photodetector, our photodetector had a longer detection wavelength than that reported by C. Liu et al. [18] and H. Zhang et al. [16], but was shorter than that reported by X. Hong et al. [17]. However, the responsivity of our photodetector was 10⁵ times better than theirs at the wavelength of ~1.5 μm, which was due to our Bi₂Se₃ film that was thicker than theirs. It is interesting that our Bi₂Se₃/p-Si photodetector had a very high responsivity at 390 nm, which has not been observed previously by others. We considered that one high-energy ultraviolet (UV) photon may pump out more than one exciton inside the Bi₂Se₃, which indicated that Bi₂Se₃ could be an excellent material for a UV photodetector. For the typical 2D materials, MoS₂ [51] and reduced graphene oxide (RGO) [52], they were also deposited onto Si or Si NWs, respectively, to fabricate the photodetectors. The detection wavelength of the former was only up to 1050 nm, much shorter than ours. Although the RGO/Si NWs photodetector can respond up to 10.6 μm, it cannot work in a self-powered mode. In general, the performance of our primitive n-Bi₂Se₃/p-Si NWs photodetector is comparative to others at present. Many studies on the morphology of Bi₂Se₃ films and photodetector structure are ongoing to further improve their performance.

4. Conclusions

We obtained high-quality topological insulator Bi₂Se₃ thin film by the vacuum thermal evaporation method, and analyzed their phase structure, chemical composition and surface morphology, and also obtained the best growth conditions after optimizing the temperatures of Bi and Se source cells, as well as the growth temperature. Then, PPMS was used to analyze the electrical properties of the Bi₂Se₃ thin film. We found that the Bi₂Se₃ film had a metallic state, WAL and LMR characteristics, typical for the TI material. However, these topological properties were measured at low temperatures (<10 K) and cannot be directly related to the performance of the detector measured at room temperature. Then, Si NWs was prepared by the MACE method and its interspace and height were tuned until its morphology was most suitable for the n-Bi₂Se₃/p-Si NWs photodetector. It was covered by a thick Bi₂Se₃ film deposited by the vacuum thermal evaporation method to form the bulk heterojunction photodetectors for the first time. Finally, we obtained a self-powered n-Bi₂Se₃/p-Si NWs photodetector with the best performance in our case: its detection wavelength ranged from 390 nm to 1700 nm, its highest responsivity was 84.3 mA/W at 390 nm and its response time reached 3 ms/1 ms, which can be attributed to the bulk heterostructure. Research efforts are underway to modify the Bi₂Se₃ filling inside Si NWs and its crystal shape as well as device structure to improve the photodetector's performance. Since the topological properties of the Bi₂Se₃ film were only observed at low temperatures, some excellent properties of the n-Bi₂Se₃/p-Si NWs photodetector obtained at room temperature should only be related to the common semiconductor properties of the Bi₂Se₃ film. In order to verify the effect of topological properties on the photodetector's performance, topological and non-topological Bi₂Se₃ films should be deposited onto Si NWs to form a photodetector and their performance should be measured at low temperatures (<10 K).

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/nano12111824/s1>, Figure S1: Characterization of Bi₂Se₃ films grown at the different Se cell temperatures (194–219 °C) and at the same Bi cell temperature of 750 °C and the growth temperature of 320 °C: (a) XRD; (b) Raman; (c–h) AFM of sample surface, scale bars in (c–h) are all 100 nm. Figure S2: Characterization of Bi₂Se₃ films grown at the different growth temperatures (220–420 °C) and at the same Bi cell temperature of 750 °C and Se cell temperature of 209 °C: (a) XRD; (b) Raman; (c–g) AFM of sample surface, scale bars in (c–g) are all 100 nm. Figure S3: (a) Surface SEM image of Bi₂Se₃ thin films; (b) EDS stratification diagram of Bi₂Se₃ thin film; (c) Se

atomic distribution map; (d) Bi atomic distribution map; (e) EDS spectrum of Bi₂Se₃ thin film. Figure S4: Surface SEM images of Ag NPs with different deposition times: (a) 20 s; (b) 30 s; (c) 40 s; (d) 60 s; (e) 100 s, scale bars in (a–e) are all 1 μm. Figure S5: Surface SEM images of the Si NWs with the same etching time of 10 min and the different deposition times of Ag NPs: (a) 20 s; (b) 30 s; (c) 40 s; (d) 60 s; (e) 100 s, scale bars in (a–e) are all 2 μm. Figure S6: Cross-sections SEM images of Si NPs with the same etching time of 10 min and the different deposition times of Ag NPs: (a) 20 s; (b) 30 s; (c) 40 s; (d) 60 s; (e) 100 s, scale bars in (a–e) are all 2 μm. Figure S7: Surface SEM images of Si NWs with the same Ag NPs deposition time of 60 s and the different etching time: (a) 4 min; (b) 6 min; (c) 8 min; (d) 10 min; (e) 12 min; (f) 14 min, scale bars in (a–f) are all 2 μm. Figure S8: Cross-section SEM images of Si NWs with the same Ag NPs deposition time of 60 s and the different etching time: (a) 4 min; (b) 6 min; (c) 8 min; (d) 10 min; (e) 12 min; (f) 14 min, scale bars in (a–f) are all 1 μm. Figure S9: The height of Si NWs as a function of the etching time. Figure S10: Surface SEM images of Bi₂Se₃/Si NWs samples with the same etching time of 10 min and the different deposition time of Ag NPs: (a) 20 s; (b) 30 s; (c) 40 s; (d) 60 s; (e) 100 s, scale bars in (a–e) are all 1 μm. Figure S11: Cross-section SEM images of Bi₂Se₃/Si NWs samples with the same etching time of 10 min and the different deposition times of Ag NPs: (a) 20 s; (b) 30 s; (c) 40 s; (d) 60 s; (e) 100 s, scale bars in (a–e) are all 1 μm. Figure S12: Surface SEM images of Bi₂Se₃/Si NWs samples with the same deposition times (60 s) of Ag NPs and the different etching times: (a) 4 min; (b) 6 min; (c) 8 min; (d) 10 min; (e) 12 min; (f) 14 min, scale bars in (a–f) are all 1 μm. Figure S13: Cross-section SEM images of Bi₂Se₃/Si NWs samples with the same deposition times (60 s) of Ag NPs and different etching times: (a) 4 min; (b) 6 min; (c) 8 min; (d) 10 min; (e) 12 min; (f) 14 min, scale bars in (a–f) are all 1 μm. Figure S14: Absorption coefficient and absorption depth of silicon and Bi₂Se₃ [24,25]. Figure S15: Energy-band diagrams for an n-Bi₂Se₃/p-Si NWs heterostructure photodetector. Table S1: Performance comparison of our Bi₂Se₃/Si NWs photodetector with previous Bi₂Se₃/Si photodetectors [16–19,21,51,52].

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