

Supplementary Materials

Preparation and Electrochemical Characterization of Si@C Nanoparticles as an Anode Material for Lithium-Ion Batteries via Solvent-Assisted Wet Coating Process

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Citation: Hwang, J.; Jung, M.; Park, J.-J.; Kim, E.-K.; Lee, G.; Lee, K.J.; Choi, J.-H.; Song, W.-J. Preparation and Electrochemical Characterization of Si@C Nanoparticles as an Anode Material for Lithium-Ion Batteries via Solvent-Assisted Wet Coating Process. *Nanomaterials* **2022**, *12*, 1627. <https://doi.org/10.3390/nano12101627>

Academic Editors: Zhipeng Sun, Bin Dong and Dong Xie

Received: 2 April 2022

Accepted: 5 May 2022

Published: 12 May 2022

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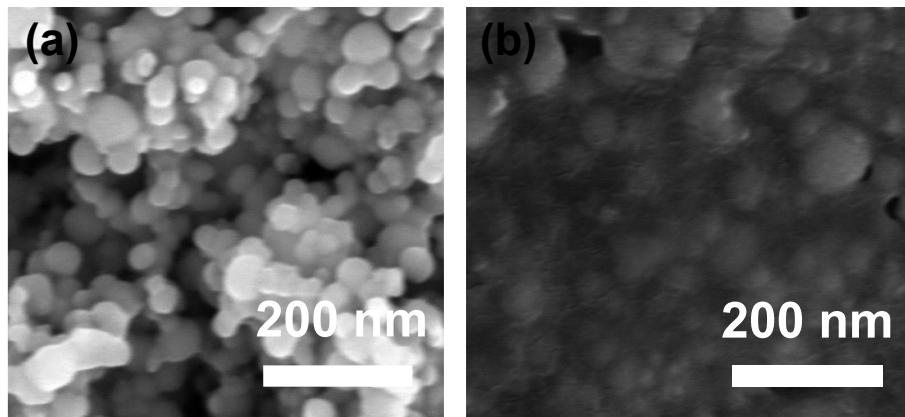


Figure S1. The SEM image of (a) Si NPs and (b) phenolic resin-coated Si NPs.

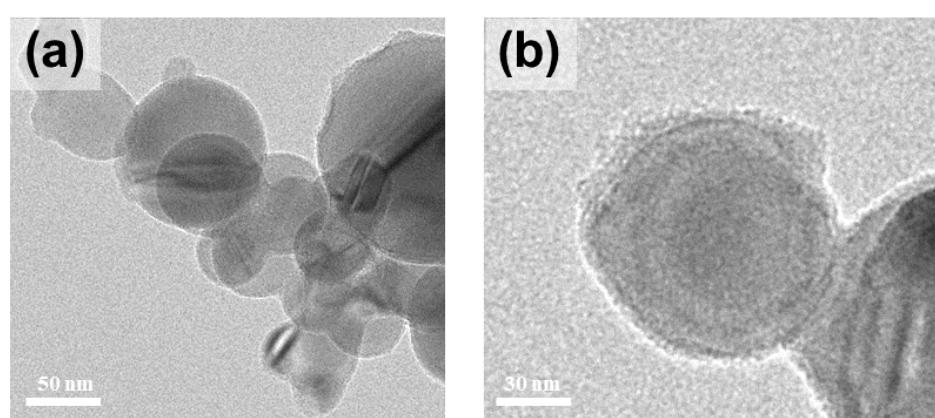


Figure S2. High resolution TEM images of (a,b) Si@C.

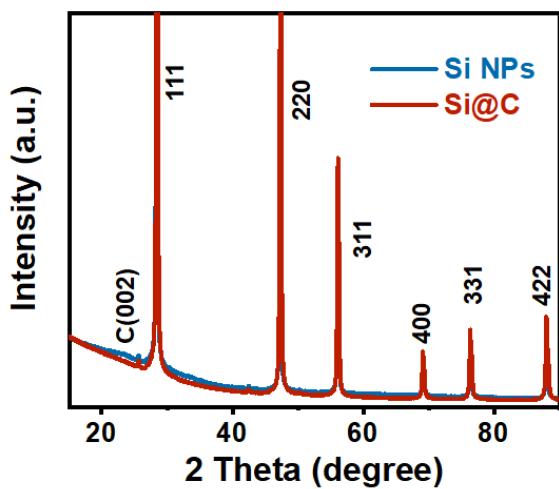


Figure S3. XRD data of Si NPs and Si@C.

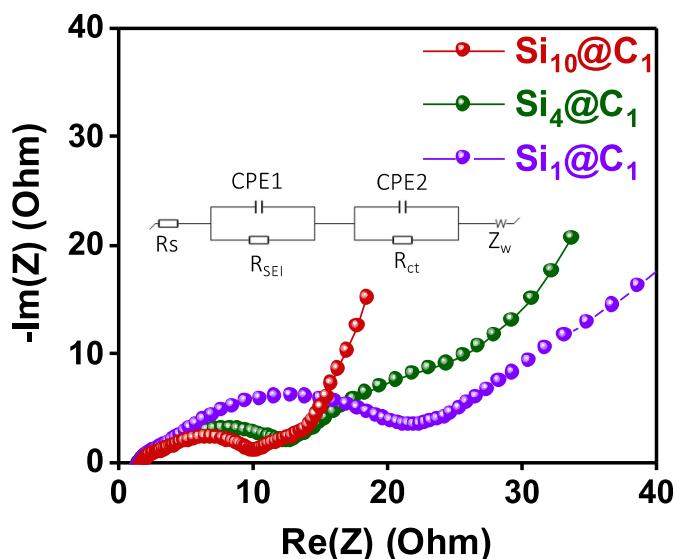


Figure S4. EIS data of $\text{Si}_{10}\text{@C}_1$, $\text{Si}_4\text{@C}_1$, and $\text{Si}_1\text{@C}_1$.

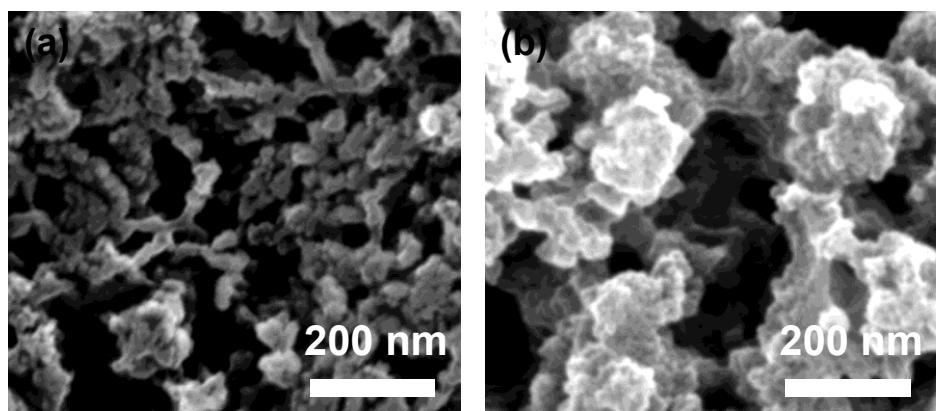


Figure S5. The SEM image of (a) Si NPs electrode and (b) $\text{Si}_{10}\text{@C}_1$ electrode after 100 cycles.

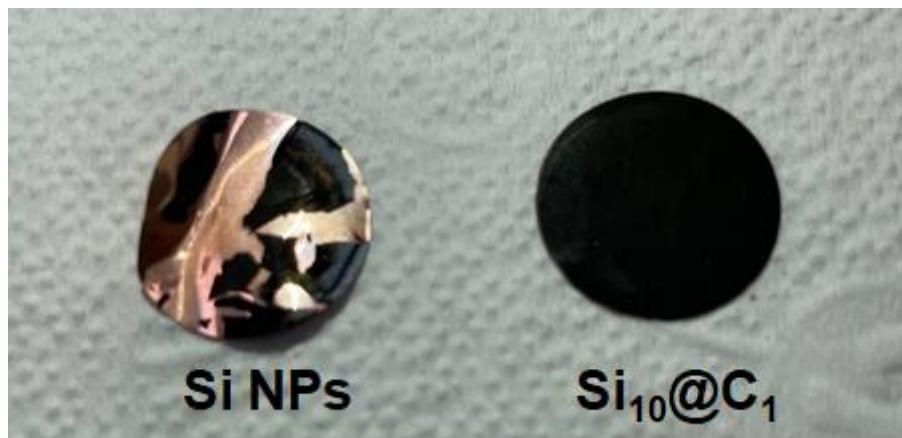


Figure S6. The photo image of Si and Si₁₀@C₁ electrodes after 100 cycles.

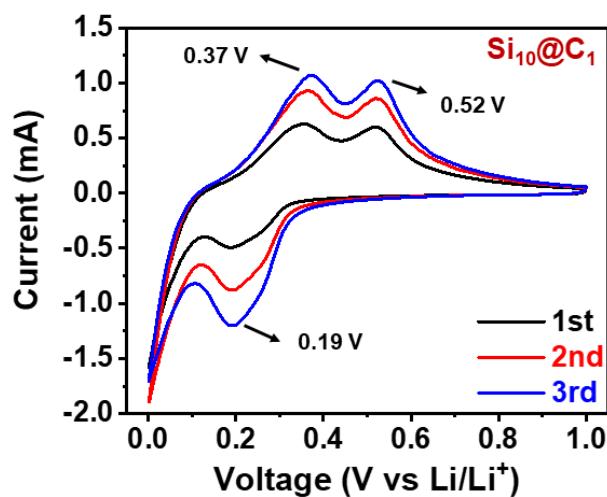


Figure S7. The cyclic voltammetry curves of Si@C anode.

Table S1. EIS data of Si NPs and Si₁₀@C₁.

Electrodes.	R _s	R _{SEI}	R _{ct}
Si NPs	1.82	19.46	7.68
Si ₁₀ @C ₁ NPs	1.84	10.04	3.51

Table S2. Comparison of electrochemical performance of various carbon-coated silicon anodes.

Active Material	Carbon Precursors	Discharge Capacity (mA h g ⁻¹)	ICE (%)	Retention (%) / Cycles	Capacity after Cycles (mA h g ⁻¹)	Ref
Si NPs (~50 nm)	Phenolic resin	2283	66.7	86.3/50	1107	[1]
Si NPs (~100 nm)	Dopamine	2258	72.2	83/50	804	[2]
Si NPs	Graphene/egg white	2169	86.2	56/200	1045	[3]
Si NPs (50~100 nm)	rGO	2200	61	63/200	968	[4]
Si NPs (~50 nm)	Reusable phenolic resin	3092	83.5	100/50	1140	Our work

Reference

1. Lu, Z.; Li, B.; Yang, D.; Lv, H.; Xue, M.; Zhang, C., A self-assembled silicon/phenolic resin-based carbon core–shell nanocomposite as an anode material for lithium-ion batteries. *RSC Adv.* **2018**, *8*, 3477–3482.
2. Pan, L.; Wang, H.; Gao, D.; Chen, S.; Tan, L.; Li, L.; Facile synthesis of yolk–shell structured Si–C nanocomposites as anodes for lithium-ion batteries. *Chem. Commun.* **2014**, *50*, 5878–5880.
3. Ou, J.; Jin, F.; Wang, H.; Wu, S.; Zhang, H.; Carbon coated Si nanoparticles anchored to graphene sheets with excellent cycle performance and rate capability for Lithium-ion battery anodes, *Surf. Coat. Technol.* **2021**, *418*, 127262.
4. Wang, Q.; Meng, T.; Li, Y.; Yang, J.; Huang, B.; Ou, S.; Meng, C.; Zhang, S.; Tong, Y. Consecutive chemical bonds reconstructing surface structure of silicon anode for high-performance lithium-ion battery. *Energy Storage Mater.* **2021**, *39*, 354–364.