

Supporting Information

MOFs derived hetero-ZnO/Fe₂O₃ nanoflowers with enhanced photocatalytic performance towards efficient degradation of organic dyes

Fakhr uz Zaman, Bing Xie, Jinyang Zhang, Tianyu Gong, Kai Cui, Linrui Hou,* Jiali Xu, Zhirou Zhai, Changzhou Yuan*

School of Materials Science & Engineering, University of Jinan, Jinan 250022, P. R. China

*Corresponding authors: mse_houlr@ujn.edu.cn; houlr629@163.com (*Prof. Hou*). mse_yuancz@ujn.edu.cn (*Prof. Yuan*).

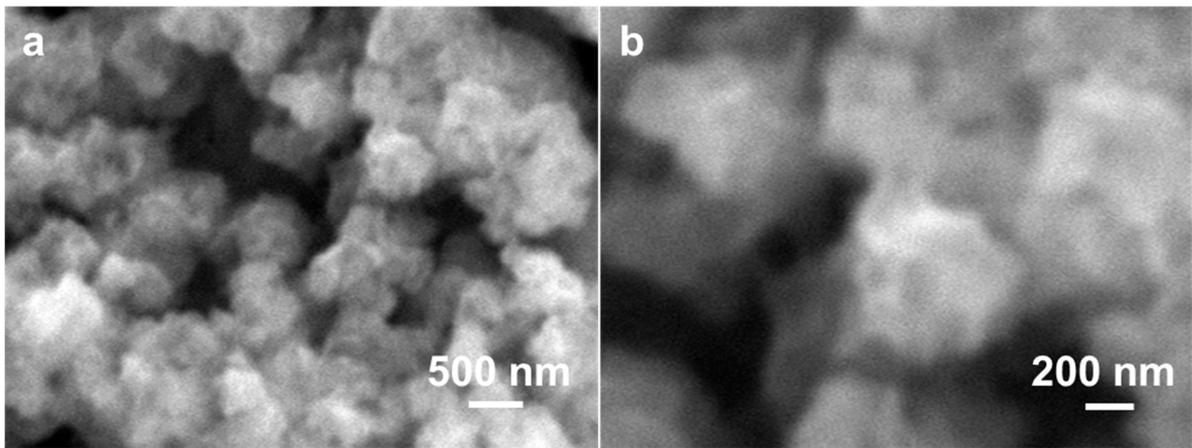


Figure S1. SEM of the sample before annealing for synthesizing ZnO /Fe₂O₃ NFs.

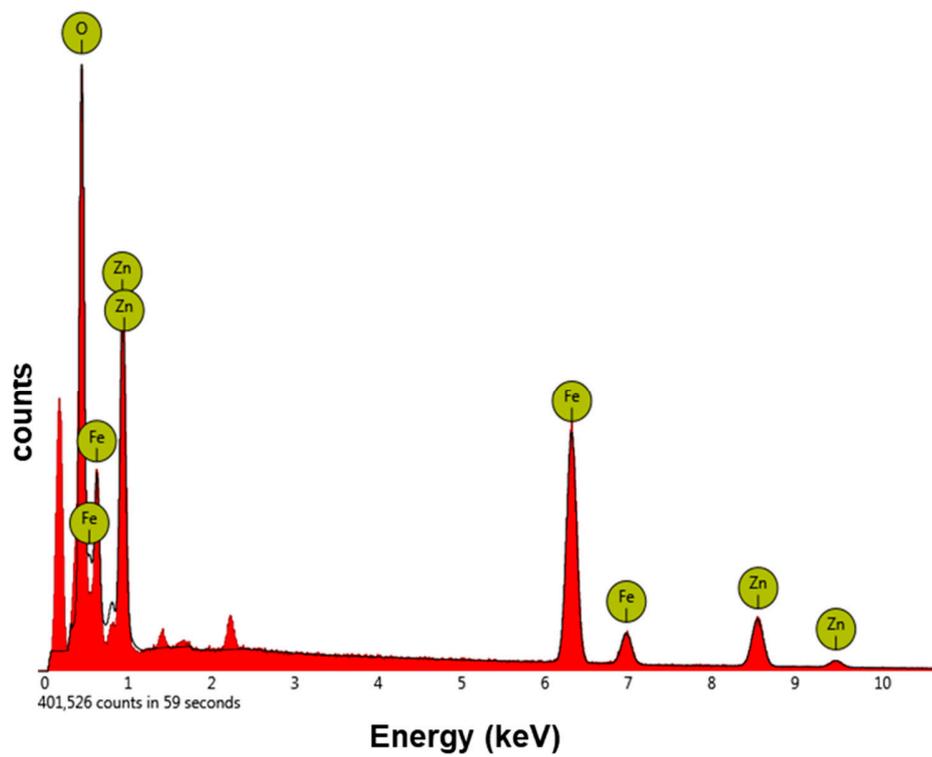


Figure S2. EDX spectrum of ZnO/Fe₂O₃ NFs.

Table S1 The elemental composition of Zn, O, and Fe in the ZnO/ Fe₂O₃ NFs.

Element Symbol	Element Name	Weight Conc.
O	Oxygen	30.88
Fe	Iron	42.01
Zn	Zinc	27.11

Table S2. Comparison in degradation efficiency of MB and MO dyes using the ZnO/Fe₂O₃ NFs as electrocatalysts with other catalysts.

composite	Light	Dyes		Irradiation time(min)	Percent Degradation	Ref
		MB	MO			
ZnO/Fe ₂ O ₃	UV	/	MO	140 min	70%	1
ZnO/Fe ₂ O ₃	Visible	MB	MO	50 and 80 min	95 and 82.5%	2
ZnO-modified g-C ₃ N ₄	200W	MB	/	120 min	90%	3
ZnO/MOS ₂	UV-vis	MB	/	80 min	81%	4
Cu-ZnO/S-gC ₃ N ₄	UV	MB	/	90 min	93%	5
Ag/ZnO	UV	MB	/	60 min	60%	6
ZnO/Cu ₂ O	UV	/	MO	180 min	73%	7
ZnO/SnO ₂	UV	/	MO	100 min	56%	8
ZnO/Eu	Sun	/	MO	150 min	62%	9
Graphene/ZnO	Sun	/	MO	360 min	96%	10
ZnO/Fe ₂ O ₃ NFs	UV	MB	MO	90 and 150 min	100 and 96%	This work

Table S3 Values of k for MB and MO dyes using different photocatalysts

Dyes	Sample	ZnO	Fe ₂ O ₃	ZnO/Fe ₂ O ₃
MB	k (min) ⁻¹	0.0109	0.015	0.034
MO	k (min) ⁻¹	0.006	0.0095	0.024

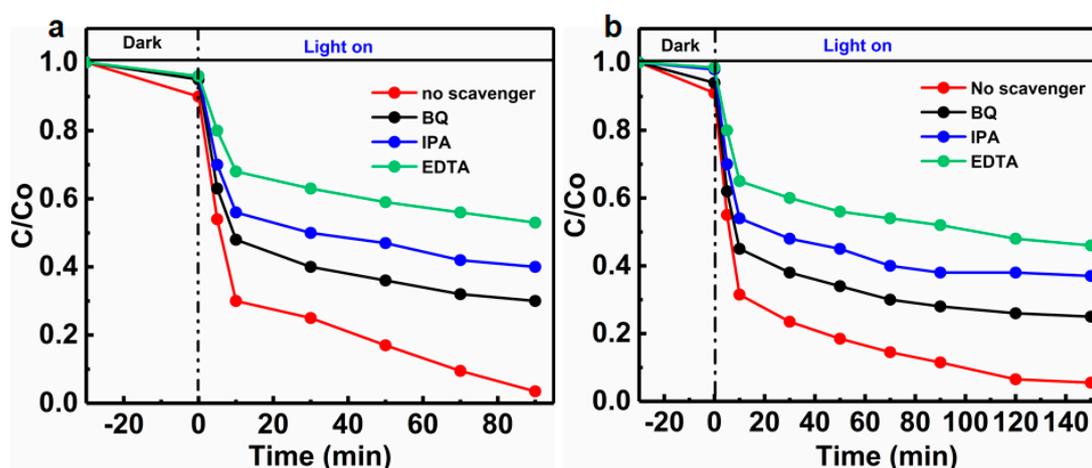


Figure S3. Active species trapping experiments using the ZnO/Fe₂O₃ NFs as the photocatalyst under UV light irradiation: (a) MB (1 mg L⁻¹) and (b) MO (1 mg L⁻¹).

References

1. Yan, W.; Fan, H.; Yang, C. Ultra-fast synthesis and enhanced photocatalytic properties of alpha Fe₂O₃/ZnO core-shell structure, *Mater. Lett.* 2011, 65, 11, 1595–1597
2. Liu, Y.; Yu, L.; Hu, Y.; Guo, C.; Zhang, F.; Lou, X. W. D. A magnetically separable photocatalyst based on nest-like γ -Fe₂O₃/ZnO double-shelled hollow structures with enhanced photocatalytic activity, *Nanoscale* 2012, 4, 1, 183-187.
3. Paul, DR.; Gautam, S.; Panchal, P.; Nehra, SP.; Choudhary, P.; Sharma, A. ZnO-Modified g-C₃N₄: A Potential Photocatalyst for Environmental Application. *ACS Omega*. 2020, 5, 8, 3828-3838.
4. Krishnan, U.; Kaur, M.; Kaur, G.; Singh, K.; Dogra, AR.; Kumar, M.; Kumar, A. MoS₂/ZnO nanocomposites for efficient photocatalytic degradation of industrial pollutants, *Mater. Res. Bull.* 2019, 111, 212-221.
5. Javed, M.; Qamar, MA.; Shahid, S.; Alsaab, H.; Asif, S. Highly efficient visible light active Cu ZnO/S-gC₃N₄ nanocomposites for efficient photocatalytic degradation of organic pollutants, *RSC Adv.*, 2021, 11, 59, 37254–37267.
6. Height, M.J.; Pratsinis, S.E.; Mekasuwandumrong, O.; Praserttham, P. Ag-ZnO catalysts for UV photodegradation of methylene blue, *Appl. Catal. B*, 2006, 63, 3-4, 305-312.
7. Xu, C.; Cao, L.; Su, G.; Liu, W.; Liu, H.; Yu, Y.; Qu, X. Preparation of ZnO/Cu₂O compound photocatalyst and application in treating organic dyes. *J. Hazard. Mater.* 2010, 176, 1-3, 807-813.

8. Ali, W.; Ullah, H.; Zada, A.; Alamgir, M.K.; Muhammad, W.; Ahmad, M.J.; Nadjman, A. Effect of calcination temperature on the photoactivities of ZnO/SnO₂ nanocomposites for the degradation of methyl orange. *Mater. Chem. Phys.* 2018, 213, 259-266.
9. Trandafilović, LV.; Jovanović, DJ.; Zhang, X.; Ptasińska, S.; Dramićanin, M. Enhanced photocatalytic degradation of methylene blue and methyl orange by ZnO: Eu nanoparticles. *Appl. Catal. B*, 2017, 203, 740-752.
10. Hao, C.; Yang, Y.; Shen, Y.; Feng, F.; Wang, X.; Zhao, Y.; Ge, C. Liquid phase-based ultrasonic-assisted synthesis of G-ZnO nanocomposites and its sunlight photocatalytic activity. *Mater. Des.* 2016, 89, 864-871.