## **Supplementary Materials**

## Fabrication of hybrid silver microstructures from vermiculite templates as SERS substrates

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**Figure S1.** Energy-dispersive X-ray (EDX) spectra of vermiculite (Ve) and silver-coated vermiculites (Ve@Ag).

Sample	Ag/Si	Mg/Si	Al/Si	Fe/Si	O/Si	Na/Si
Vermiculite (Ve)	0	0.75	0.16	0.20	4.18	0.16
Ag-coated vermiculite (Ve@Ag)	0.31	0.78	0.22	0.19	4.19	0
Activated vermiculite (AVe)	0	0.68	0.24	0.19	4.17	0.17
Ag-coated activated vermiculite (AVe@Ag)	0.28	0.71	0.25	0.22	4.05	0

**Table S1.** Atomic ratios obtained from the ESEM-EDX characterization illustrated in Fig. 1 and S1.



**Figure S2.** SERS detection of benzenethiol. Detail of the SERS spectra of hybrid Ag microstructures from activated vermiculite in the presence of decreasing concentration of BT, from the top to the bottom: 10 nM, 1 nM, 10 pM, 1 pM. SERS experiments were performed as follows. 5  $\mu$ L of the Ag microstructures suspension were dispersed with 995  $\mu$ L of BT aqueous solutions at decreasing concentrations. The samples were left under shaking for 2 hours, then the Ag microstructures were left to sediment. 980  $\mu$ L of the supernatant were removed, and the microstructures were redispersed in suspension prior to the SERS analysis. SERS spectra were acquired with 20 seconds exposure time, 5 accumulations and by illuminating with a 633 nm laser.