



Supporting Information

Electroactive Ultra-Thin rGO-Enriched FeMoO₄ Nanotubes and MnO₂ Nanorods as Electrodes for High-Performance All-Solid-State Asymmetric Supercapacitors

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Abstract: A flexible asymmetric supercapacitor (ASC) with high electrochemical performance was constructed using reduced graphene oxide (rGO)-wrapped redox-active metal oxide-based negative and positive electrodes. Thin layered rGO functionality on the positive and the negative electrode surfaces has promoted the feasible surface-active sites and enhances the electrochemical response with a wide operating voltage window. Herein we report the controlled growth of rGO-wrapped tubular FeMoO4 nanofibers (NFs) via electrospinning followed by surface functionalization as a negative electrode. The tubular structure offers the ultrathin-layer decoration of rGO inside and outside of the tubular walls with uniform wrapping. The rGO-wrapped tubular FeMoO4 NF electrode exhibited a high specific capacitance of 135.2 F g⁻¹ in Na₂SO₄ neutral electrolyte with an excellent rate capability and cycling stability (96.45% in 5000 cycles) at high current density. Meanwhile, the hydrothermally synthesized binder-free rGO/MnO2 nanorods on carbon cloth (rGO-MnO2@CC) were selected as cathode materials due to their high capacitance and high conductivity. Moreover, the ASC device was fabricated using rGO-wrapped FeMoO4 on carbon cloth (rGO-FeMoO4@CC) as the negative electrode and rGO-MnO2@CC as the positive electrode (rGO-FeMoO4@CC/rGO-MnO2@CC). The rationally designed ASC device delivered an excellent energy density of 38.8 W h kg⁻¹ with a wide operating voltage window of 0.0–1.8 V. The hybrid ASC showed excellent cycling stability of 93.37% capacitance retention for 5000 cycles. Thus, the developed rGO-wrapped FeMoO4 nanotubes and MnO2 nanorods are promising hybrid electrode materials for the development of wide-potential ASCs with high energy and power density.

Keywords: electrospinning; FeMoO₄ nanotubes; rGO wrapping; MnO₂-rGO; asymmetric supercapacitors



Figure S1. SEM image of the as-spun (a, c, e) and annealed (b, d, f) nanofibers with different polymer precursors. (a, b) FeMo-PAN, (c, d) FeMo-PAN/PMMA and (e, f) FeMo-PMMA.



Figure S2. Raman analysis of the pristine and rGO wrapped FeMoO4 tubular nanofibers.



Figure S3. TEM images of the ultra-thin rGO nanoflakes prepared throught the thermal reduction of GO.



Figure S4. TEM images of the pristine FeMoO₄ tubular nanofibers, inset shows the HRTEM images of the respective tubular nanofiber.



Figure S5. TEM images of the pristine rGO wrapped FeMoO₄ tubular nanofibers, inset shows the SEAD pattern of the respective tubular nanofiber. .



Figure S6. EDAX results of the rGO wrapped FeMoO₄ nanofibers.



Figure S7. N₂ adsorption-desorption isotherm of rGO wrapped FeMoO₄ nanofibers. Inset shows the corresponding pore size distribution curve of the tubular nanofibers.



Figure S8. CV curve of the FeMoO₄, and rGO-FeMoO₄ electrodes at 10 mV s⁻¹ in a three-electrode cell.



Figure S9. CV and GCD curve of the FeMoO₄ with different loading density of rGO (20, 40, 60 % and 80 %).



Figure S10. SEM image of the rGO-FeMoO4@CC electrode after electrochemical cyclic performances.

Table S1. Comparison of the energy density of the rGO-MnO₂//rGO-FeMoO₄ ASC device with previously reported MnO₂ based and other ASC systems.

ASC device	Voltage (V)	E (W h kg ⁻¹)	P (W kg ⁻¹)	Ref.
ZnCo ₂ O ₄ -MnO ₂ //AC	1.6	69	867	[1]
ZnCo ₂ O ₄ /NG//AC	1.6	28.3	500	[2]
ZnCo ₂ O ₄ @Ni _x Co _{2x} (OH)6 _x //AC	1.7	26.2	511.8	[3]
Co ₃ O ₄ @MnO ₂ //MEGO	1.6	17.7	158	[4]
ZnCo ₂ O ₄ @MnO ₂ //AC	1.5	29.41	628.42	[5]
ZnCo ₂ O ₄ @ZnWO ₄ //AC	1.6	24	400	[6]
$MnO_2//Fe_2O_3$	1.6	0.55		[7]
CaMoO ₄ //AC	1.6	18.7	362	[8]
$MnO_2//Fe_2O_3$	1.8	53.55	1280	[9]
MnO ₂ -GNS//FeOOH-GNS-CNTs	1.7	30.4	237.6	[10]
CNT@NiO//CNT@Fe ₂ O ₃	1.6	63.3	1600	[11]
MnO2-MWCNT//VN-MWCNT	1.8	38.7	730	[12]
GNR/MnO ₂ //GNR	2.0	29.4	12.1	[13]
MnO ₂ //FeOOH	1.85	24	450	[14]
MnO ₂ nanowire//graphene	2	30.4	100	[15]
rGO-MnO ₂ //rGO-FeMnO ₄ *	2.2	31.8	1099	This work*

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