Supporting information

Nanostructured and spiky gold shell growth on magnetic particles for SERS applications

Erin E. Bedford ^{1,2}, Christophe Méthivier ¹, Claire-Marie Pradier ¹, Frank Gu ² and Souhir Boujday ^{1,*}

- ¹ Laboratoire de Réactivité de Surface (LRS), CNRS, UMR 7197, Sorbonne Université, 4 place Jussieu, F-75005 Paris, France; <u>eebedford@gmail.com (E.E.B.)</u>; <u>christophe.methivier@sorbonne-universite.fr (C.M.)</u>; <u>claire-marie.pradier@upmc.fr (C.-M.P.)</u>; <u>souhir.boujday@sorbonne-universite.fr (S.B.)</u>
- ² Department of Chemical Engineering and Waterloo Institute for Nanotechnology, University of Waterloo, 200 University Ave. W., Waterloo, Canada; <u>eebedford@gmail.com (E.E.B.)</u>; <u>frank.gu@uwaterloo.ca</u> (F.G.)
- * Correspondence: <u>souhir.boujday@sorbonne-universite.fr (S.B.</u>); Tel.: +33-1-4427-6001

Experimental conditions for SERS measurement

First set: influence of silicon-iron oxide core functionalization: In the first set of experiments, samples for surface-enhanced Raman scattering (SERS) were prepared by adding 50 μ L of particles dispersed in CTAB to 3 mL of Millipore water; this amount was chosen because it was the minimum amount required for signal saturation. Particles were initially dispersed in 1 mM CTAB (final CTAB concentration of 16.7 μ M) and varying amounts of the Raman probe, 2-mercaptopyrimidine (MPym), were added.

Second set: influence of bath conditions: In the second set of experiments, samples were also prepared by adding 50 μ L of particles dispersed in CTAB to 3 mL of Millipore water, but particles were initially dispersed in 10 mM CTAB (final CTAB concentration of 167 μ M), which resulted in a signal large enough to be compared without the use of an additional Raman probe.



UV-Visible spectra during gold seed attachment and upon shell growth

Figure S1. UV-Vis spectra of A) particles before and after gold seeds binding, B) spiky particles.



Magnetic properties

Figure S2: A- Magnetization curves of silica-coated iron oxide particles before (solid) and after (dashed) gold/silver shell coating. The inset shows the small amount of hysteresis occurring at low magnetic fields. B-Measured opacity over time for water dispersions of particles at 1 mg Fe₃O₄/SiO₂ particles/mL (greater mass upon gold coating) in a 45 T/m gradient, before (solid) and after (dashed) gold shell coating

The magnetization curve (inset in Figure S2) also shows that particles have low remanent magnetization; for the silica-coated iron-oxide particles, the remanent magnetization is 1 emu/g and for the gold-coated particles, the value is 0.25 emu/g, suggesting that the particles do not meet the formal definition of superparamagnetism, but with despite such a low value, they do exhibit superparamagnetic behavior in practical applications.

Table	S1:	Separat	ion	times	of	particl	es,	determine	ed	based	on	time	requ	ired	to	reach	5%	and
						1	%	of initial of	ppa	acity								

	Separation time (seconds)				
	5% opacity	1% opacity			
Fe ₃ O ₄ /SiO ₂	48	89			
Fe ₃ O ₄ /SiO ₂ /Au-Ag	68	108			



Figure S3: A- Change in peak height of Raman signal corresponding to the Raman reporter (MPym) at different concentrations. A fit to a Langmuir model (red line) shows saturation behavior ($R^2 = 0.98$), B- Change in peak heights corresponding to CTAB at different Raman reporter (MPym) concentrations.