

Article

Suppression of electric field-induced segregation in sky-blue perovskite light-emitting electrochemical cells

Tatiana G. Liashenko ^{1,*,‡}, Anatoly P. Pushkarev ^{1,*,‡}, Arnas Naujokaitis ², Vidas Pakštas ², Marius Franckevičius², Anvar A. Zakhidov^{1,3}, Sergey V. Makarov¹

- 1 Department of Physics and Engineering, ITMO University, 197101 St. Petersburg, Russia;
- 2 Center for Physical Sciences and Technology, LT-10257 Vilnius, Lithuania;
- 3 University of Texas at Dallas, Richardson TX 75080, USA;
- * Correspondence: tatiana.liashenko@metalab.ifmo.ru (T.G.L.); anatoly.pushkarev@metalab.ifmo.ru (A.P.P.)
- These authors contributed equally to this work. ŧ

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Chemical equations: 1

- 1) CsCl + PbBr₂ \longrightarrow CsPbClBr₂ 2
- 2) $\frac{5}{4}$ CsCl + PbBr₂ $\longrightarrow \frac{1}{12}$ Cs₄PbCl_{2.46}Br_{3.54} + $\frac{11}{12}$ CsPbCl_{1.14}Br_{1.86} 3) $\frac{4}{3}$ CsCl + PbBr₂ $\longrightarrow \frac{1}{9}$ Cs₄PbCl_{2.24}Br_{3.76} + $\frac{8}{9}$ CsPbCl_{1.22}Br_{1.78}

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Figure S 1. (a) Fluorescent microimages of the films **1-3**. (b) Large-scale SEM images of the samples. (c) High-resolution SEM images of the samples. (d) Grains size distribution derived from the high-resolution SEM images.



Figure S 2. (a) CIE 1931 coordinates displaying a color of photoluminescence from 1-3. (b) A photograph of the sample 1 under intense UV light of $1 \text{ W} \cdot \text{cm}^{-2}$.



Figure S 3. (a) Partial passivation of surface halide vacancies in a perovskite grain by oxygen atoms of PEO. (b) More complete passivation of surface halide vacancies in a perovskite grain by both PEO and hexahalide crystallites.



Figure S 4. Electroluminescence external quantum efficiency vs bias relations for the PeLECs 1-3.