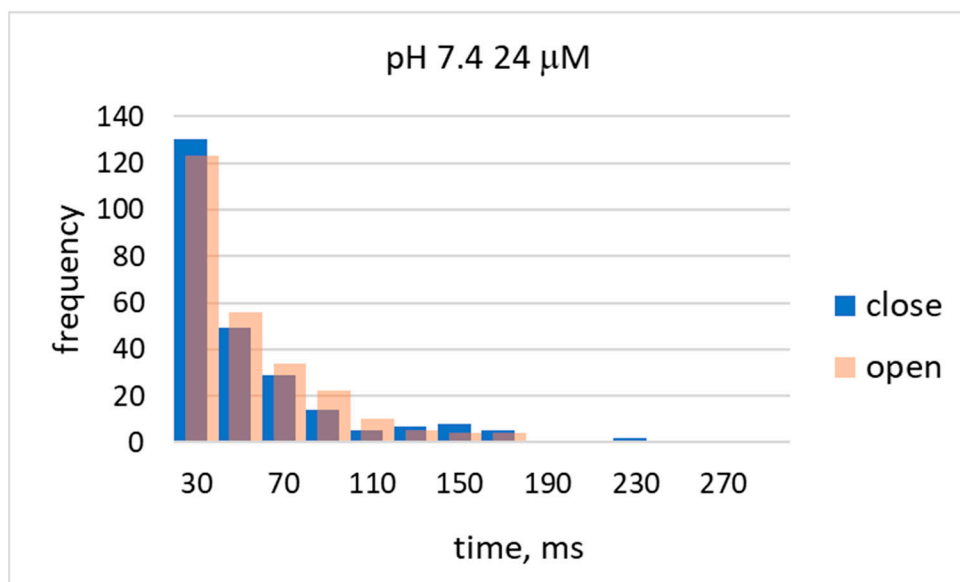
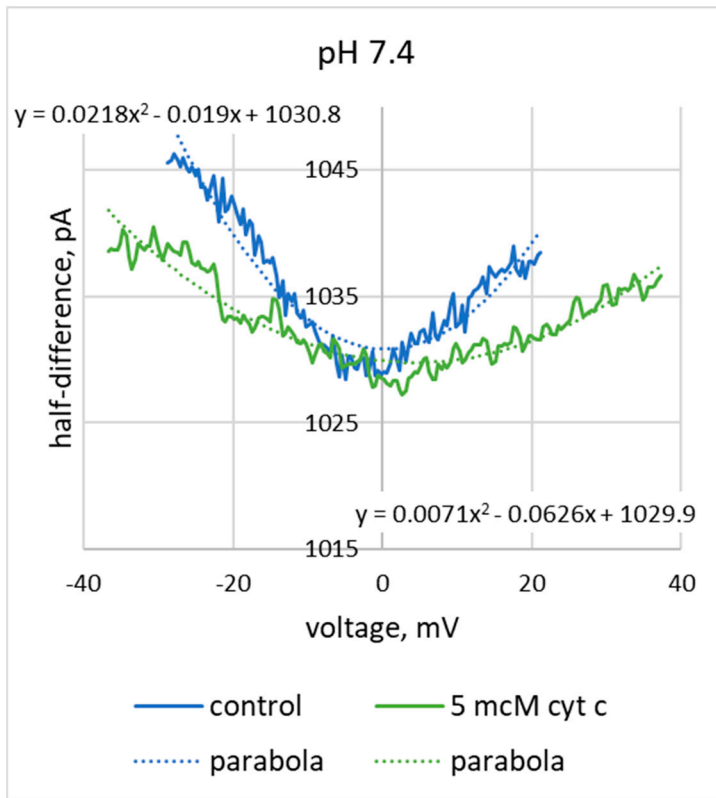


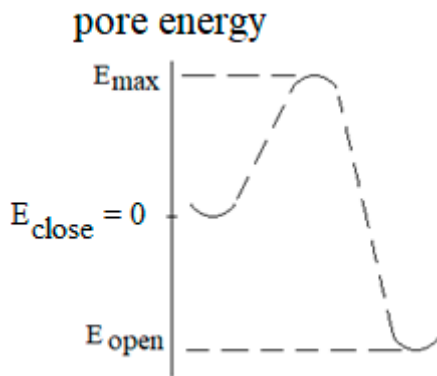
**Figure S1.** Experimental (markers) and approximating (dotted lines) dependences of conductivity increment on cyt c concentration at different pH.



**Figure S2.** Distributions of the durations of the open and closed states of pore presented in Figure 3c. Voltage +50 mV, 0.1 M KCl, pH=7.4, 24  $\mu\text{M}$  cyt C.



**Figure S3.** The half-difference of membrane currents in control and after one-side addition of 5  $\mu$ M cyt c. Parameters of the triangular voltage are 100 mV and 1 Hz. 0.1 M KCl, pH=7.4. The calculated minimum is shifted from 0.4 (control) to 4.4 mV.



**Figure S4.** Energy diagram for the long-lived pore  $E_{open} < E_{close} = 0$

To estimate  $E_{max}$  we can use the literature on value  $\nu$ . The scatter of  $\nu$  value is significant – from  $5 \times 10^{32} \text{ s}^{-1} \text{ m}^{-3}$  [1] to  $2 \times 10^{42} \text{ s}^{-1} \text{ m}^{-3}$  [2]. In [3] we have got for  $\nu$   $5.6 \times 10^{33} \text{ s}^{-1} \text{ m}^{-3}$ .

According to [2],  $V_{pore} = V \approx 2\pi r^2 h$  is the volume of the membrane portion immediately connected with a pore,  $r = 0.4 \text{ nm}$  is the pore radius,  $h = 5 \text{ nm}$  is the membrane thickness.

$E_{max}$  values at different  $\nu$  are presented in the table:

**Table S1.** Energy barrier between the open and close states of a pore

$\nu, \text{s}^{-1} \text{ m}^{-3}$	$\Theta=0.4 \text{ s}, E_{max}, kT$	$\Theta=0.06 \text{ s}, E_{max}, kT$	reference

$5 \times 10^{32}$	13.8	11.9	[1]
$5.6 \times 10^{33}$	16.2	14.3	[3]
$2 \times 10^{42}$	35.9	34.0	[2]

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