

Article

Ti₃C₂ MXene Membranes for Gas Separation: Influence of Heat Treatment Conditions on D-Spacing and Surface Functionalization

Aline Alencar Emerenciano ^{1,*}, Rubens Maribondo do Nascimento ¹, Ana Paula Cysne Barbosa ¹, Ke Ran ^{2,3}, Wilhelm Albert Meulenberg ^{4,5,6} and Jesus Gonzalez-Julian ^{4,7}

¹ Materials Science and Engineering Postgraduate Program, UFRN, Natal 59078-570, Brazil

² Central Facility for Electron Microscopy GFE, RWTH Aachen University, 52074 Aachen, Germany

³ Ernst Ruska-Centre for Microscopy and Spectroscopy with Electrons ER-C, Forschungszentrum Jülich GmbH, 52425 Jülich, Germany

⁴ Forschungszentrum Jülich GmbH, Institute of Energy and Climate Research, Materials Synthesis and Processing (IEK-1), 52428 Jülich, Germany

⁵ Jülich Aachen Research Alliance: JARA-Energy, D-52425 Jülich, Germany

⁶ Inorganic Membranes, Faculty of Science and Technology, University of Twente, 7500 AE Enschede, The Netherlands

⁷ Department of Ceramics, Institute of Mineral Engineering, RWTH Aachen University, D-52074 Aachen, Germany

* Correspondence: aline.alencar.071@ufrn.edu.br

Citation: Emerenciano, A.A.; do Nascimento, R.M.; Barbosa, A.P.C.; Ran, K.; Meulenberg, W.A.; Gonzalez-Julian, J. Ti₃C₂ MXene Membranes for Gas Separation: Influence of Heat Treatment Conditions on D-Spacing and Surface Functionalization. *Membranes* **2022**, *12*, 1025. <https://doi.org/10.3390/membranes12101025>

Academic Editors: Chenxiao Jiang, Zhe Yang and Ying Mei

Received: 27 September 2022

Accepted: 17 October 2022

Published: 21 October 2022

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

Abstract: Two-dimensional (2D) MXene materials have recently been the focus of membrane research due to their unique properties, such as their single-atomic-layer thickness, flexibility, molecular filtration abilities and microstructural similarities with graphene, which is currently the most efficient precursor material for gas separation applications. In addition, the potential to process nanoscale channels has motivated investigations of parameters which can improve membrane permeability and selectivity. Interlayer spacing and defects, which are still challenging to control, are among the most crucial parameters for membrane performance. Herein, the effect of heat treatment on the d-spacing of MXene nanosheets and the surface functionalization of nanolayers was shown regarding its impact on the gas diffusion mechanism. The distance of the layers was reduced by a factor of over 10 from 0.345 nm to 0.024 nm, the defects were reduced, and the surface functionalization was maintained upon treatment of the Ti₃C₂ membrane at 500 °C under an Ar/H₂ atmosphere as compared to 80 °C under vacuum. This led to a change from Knudsen diffusion to molecular sieving, as demonstrated by single-gas permeation tests at room temperature. Overall, this work shows a simple and promising way to improve H₂/CO₂ selectivity via temperature treatment under a controlled atmosphere.

Keywords: Ti₃C₂ MXenes; gas separation membrane; d-spacing control; sieving membrane; H₂/CO₂ selectivity

1. Introduction

Regarding the high demand for technologies in the context of sustainability and resource recovery, the development of gas separation membranes has been of major interest in the past few decades [1]. A range of different membranes have been presented to date; however, the necessity for precise and rapid gas separation leads to a trade-off relationship between permeability and selectivity, which stills comprises a major weakness in this field [2]. Therefore, there is considerable demand for membranes with both high permeability and selectivity [3]. The recent use of two-dimensional (2D) materials has been proposed as a promising alternative thanks to simple preparation and

the possibility for large scale-fabrication [4]. High-performance two-dimensional (2D) materials, such as graphene oxide (GO) [5], metal-organic framework (MOF) nanosheets [6], covalent organic frameworks (COFs) [7], transition-metal dichalcogenides (TMDs) [2] and zeolite nanosheets [6,8], have recently been developed for several applications, including in the field of molecular sieving. These 2D materials facilitate selective gas transport by size sieving and/or electrostatic repulsion [2]. The selectivity factor defines the efficiency of a gas separation membrane and is considered ideal when larger molecules are retained while smaller molecules move through the free membrane path.

In this context, the recently discovered 2D material MXene is an excellent candidate for use in a high-performance sieving membrane (e.g., for H₂ purification or CO₂ capture) due to its layered structure, ideal thickness and low-mass barrier [7]. MXenes were first reported by Gogotsi et al. [9]. MXenes consist of transition-metal carbides, nitrides and carbonitrides obtained by selectively etching out the A group (mainly IIIA or IVA), which occupies the interspace in the MAX phase, the precursor material. They are represented by the formula $M_{n+1}X_nT_x$ ($n = 1, 2$ or 3), where M is a transition-metal and X is carbon or nitrogen. T represents the nanosheet's surface functionalization, which appears after centrifugation cycles (O, -OH and -F) and changes depending on the synthesis parameters [10]. The diversity of functional groups leads to the ordered deposition of nanosheets, which can create specific spaces for highly efficient sieving membranes [11]. Membrane thickness is also crucial for the selectivity and high permeability of smaller gases while it maintains the blocking effect of larger molecules [2]. MXene free-standing membranes with different thicknesses have been produced via vacuum-assisted filtration (VAF) [12–15]. Ding et al. produced a free-standing membrane via VAF with free spacing of 0.35 nm and a thickness of 2 μ m, which attained a high H₂/CO₂ selectivity (166.6) greater than the Knudsen coefficient (4.7) [14]. The selectivity of H₂/CO₂ in an ultrathin continuous 2D CTf-1 membrane (COF, 100 nm) was improved by increasing the membrane thickness while the gas permeance was reduced [11].

Nanochannel size plays an important role regarding the parameters which influence the separation performance in terms of permeability and selectivity [16,17]. A few studies have focused on the manipulation of the d-space for improvement in gas separation; for example, Arshadi et al. tailored the free interlayer spacing of the MXene nanosheets through impregnating different types of ions in an electrical separation system [16]. Additionally, chemical tuning of the MXene nanochannel for H₂ preferential transport was designed to regulate the interlayer space of MXene nanosheets [15]. Cheng et al. post-treated MXenes with a H₂ atmosphere at higher temperatures to verify if there was a shift in cycling performance; however, the effect of the thermal treatment on interlayer spacing or stacking was not investigated [18].

The sintering of MXene films has been reported in a few works, which opened a window to investigate interlayer spacing adjustment. The temperature tolerance of 2D MXene was studied under higher temperatures for hydrogen sieving [19]. The narrowing of interlayer spacing was adjusted via sintering-temperature regulation of MXene films at temperatures up to 500 °C under an air atmosphere [20]. Moreover, previous works have reported the change in free d-spacing to improve properties to be applied in other fields. The intercalation of Ti₃C₂, Ti₃CN and TiNbC MXenes resulted in the increase in c-lattice parameters, resulting in an increase in Li-ion capacity [21]. The study of MXene membranes for gas selectivity is mainly based on the construction of interstitial space and has not been focused on the interaction effect between the termination groups or on the control of interlayer space [2].

Although many reports have focused on the size exclusion effect by adjusting the interlayer spacing, some efforts have been made to evaluate the effect of functional groups on gas behavior during diffusion [22]. Studies have reported that the adsorption capacity of CO₂ in the MXene interlayer is superior due to its higher quadrupole moment compared to other gases (i.e., N₂), which increases the separation factor for the H₂/CO₂ pair by the suppression of CO₂ into the MXene nanochannels in parallel to the faster

diffusion of smaller molecules such as H_2 [23]. Other works have presented the effect of different functional groups on simulated MD gas permeation for single and mixed gases, and the results show no significant difference between both the permeation behavior of a surface presenting $-F$ and with a variety of functional groups such as $-OH$, $-O$ and $-F$ [14].

In view of the above information, this work aimed to study the effect of heat treatment under a controlled H_2/Ar atmosphere and vacuum on the interlayer spacing of MXene nanosheets, as well as the nature of functional groups to improve the membrane's gas permeation behavior. Thus, MXene nanosheets were produced by the top-down conventional method and Al_2O_3 substrates were coated with MXenes via dip coating to effectively produce thin films. The intended thickness was achieved by controlling the Ti_3C_2 suspension concentration, and the d-spacing was narrowed through heat treatment at $500\text{ }^\circ\text{C}$ under a H_2 atmosphere to improve the H_2/CO_2 selectivity. Single-gas permeation tests were conducted under different pressures at room temperature, and the gas diffusion mechanism was associated with the free path between the nanosheets achieved using different heat treatment conditions of the coated substrate.

2. Materials and Methods

2.1. Ti_3C_2 MXene Synthesis

The precursor powder for the synthesis of Ti_3C_2 MXenes was a MAX phase (Ti_3AlC_2), which was produced in-house with a high purity of 98% by mixing Ti, Al and C powders in stoichiometric quantities with KBr powder, followed by pressing the compacted powder and a heat treatment at $1200\text{ }^\circ\text{C}$ through the molten salt method as described in [24]. Using the top-down method [25], the $Ti_3C_2T_x$ MXene was synthesized using HF in situ from the reaction of 1 g of lithium fluoride powder, -300 mesh (Sigma Aldrich, Germany), LiF and 10 mL of 9 M HCl, which was obtained by diluting 10.2 M hydrochloric acid 32% (Sigma Aldrich, Germany) in deionized water to achieve the desired concentration. This reaction was performed in a 50 mL PTFE reaction flask under stirring using a magnetic bar with a rotation speed of 500 rpm for 30 min at room temperature.

Next, 0.5 g of Ti_3AlC_2 (the precursor MAX phase) was introduced stepwise into the acid solution for 5 min. After 24 h at $35\text{ }^\circ\text{C}$ of etching aluminum from the precursor, the acid solution was carefully removed and washed with deionized water (DI) by centrifugation. Before each centrifugation cycle, 250 mL of DI water was mixed with the remaining slurry followed by gently shaking the 50 mL falcon tubes and subsequent centrifugation at 3500 rpm for 5 min. A pH of 6–7 was achieved after 3 centrifugation cycles. The obtained slurry and supernatant were then separated with a pipette. The slurry, which contained MXene and non-etched MAX phase, was diluted in 100 mL of DI water and sonicated for 1 h in order to enhance delamination of $Ti_3C_2T_x$ particles and to obtain a high amount of single MXene flakes in the suspension. Surfactants were not needed for the delamination, as MXenes can be well-dispersed in water [21,26,27]. The final suspension after the sonication step was purified through decantation, and so it was left to rest overnight prior to decantation, and the etched and non-etched MAX phases were mechanically separated.

2.2. Membrane Processing

$Ti_3C_2T_x$ membranes with different thicknesses were produced by dip coating using an $\alpha-Al_2O_3$ substrate. Thus, 2 concentrations were studied: C_1 , is the concentration of the final suspension, which was obtained using the method described in the previous section. The second and lower concentration ($C_2 = 0.5 C_1$) was obtained by diluting 25 mL of the C_1 suspension in 25 mL of deionized water. Freestanding membranes were produced for each suspension by vacuum-assisted filtration of an amount of 3 mL using Whatman Anodisc inorganic filter membrane substrates (AAO, diameter 47 mm, pore size $0.2\text{ }\mu\text{m}$). The free-standing membranes were used for posterior characterizations of the termination

groups and the thermal stability of the material. Thin membranes were obtained by dip coating using α -Al₂O₃ Pervatec substrates (diameter 38 mm, thickness 2.2 mm). Next, 10 membranes were produced from each suspension (C₁ and C₂) and heat-treated under two conditions: A) at 80 °C, under vacuum, for 24 h (C_{1_80} and C_{2_80}); and B) at 500 °C, under a H₂/Ar atmosphere, for 6 h (C_{1_500} and C_{2_500}). Finally, five similar repetitions were produced for each of these four samples.

2.3. Permeation Tests

All coated α -Al₂O₃ substrates were tested through single-gas permeation under different conditions using He, H₂, CO₂, N₂ and CH₄. The behavior of the membranes was analyzed under 1, 2, 3, 4 and 5 bar at room temperature. Permeance values were obtained, and the H₂/CO₂ selectivity factor was analyzed.

2.4. Characterization

The crystal structure from the Ti₃AlC₂ MAX phase precursor powder to Ti₃C₂ MXene thin film was analyzed using X-ray diffraction, which was performed on a D4 Endeavor X-ray powder diffractometer (Bruker, Germany) using a Cu-K alpha wavelength. The data were collected in the 2Theta range of 5–80° with a 0.02° step size and 0.75 s/step. The d-spacing between the MXene layers of the membranes produced by dip coating was estimated through Bragg's law [14]. Then, Raman spectra were collected using a InVia Raman Microscope (Renishaw, Hoffman Estates, IL) equipped with a solid-state excitation laser (532 nm) and 2400 lines per mm grating in order to characterize the surface functionalization of MXene layers. Regions of interest were then characterized by Scanning electron microscopy (SEM) and imaged using a Zeiss Ultra 55 microscope (Zeiss, Oberkochen, Germany). TGA was also performed to investigate the thermal stability of the membrane under different atmospheres and temperatures to analyze the effect of temperature and a controlled atmosphere on the nature of surface functionalization and its implication on H₂/CO₂ selectivity. The TGA was performed in a NETZSCH STA 449F3 high DTA (NETZSCH-Gerätebau GmbH, Seligenstadt, Germany) furnace at a temperature range from 25 °C to 1200 °C with a heating rate of 5 °C/min under vacuum and a NETZSCH, STA 449F1, Jupiter (NETZSCH-Gerätebau GmbH, Seligenstadt, Germany), at a temperature range from 30 °C to 1100 °C with a rate of 5 °C/min under an Ar and H₂ atmosphere.

Transmission electron microscopy was subsequently performed to investigate the order, quality and nanosheet spacing of deposited nanofilms. TEM specimens were cut from the MXenes on the substrate via focused ion beam (FIB) milling using an FEI Strata400 system (FEI Company, Hillsboro, USA) with a Ga ion beam. Further thinning and cleaning were performed with an Ar ion beam in a Fischione Nanomill 1040 (E.A. Fischione Instruments, Inc., Export, USA) at 900 eV and 500 eV beam energy, respectively. TEM imaging was performed by an FEI Tecnai F20 microscope (FEI Company, Hillsboro, USA) at 200 kV. High-resolution high-angle annular dark-field (HAADF) imaging and energy-dispersive X-ray spectroscopy (EDXS) were conducted with an FEI Titan G2 80–200 ChemiSTEM microscope (FEI Company, Hillsboro, USA) equipped with an XFEG, a probe Cs corrector and a super-X EDXS system. The gas permeation performance was studied by analyzing single-gas permeation by flowing the following gases: He, H₂, CO₂, N₂ and CH₄. A custom massive metal module was employed for this purpose, in which internal space dip-coated Al₂O₃ substrates were adjusted with thin metal rings and sealed with rubber rings. The measurements were conducted using a single-gas test rig for porous membranes, GMP1. The flow rate and permeance were obtained in real time through the LabVIEW software, 2016 (National Instruments, Austin/US), for further calculations of permeability and selectivity.

3. Results

3.1. MXenes Synthesis and Membrane Processing

Figure 1 shows the X-ray diffraction measurement results. The spectrum of the slurry contained several peaks, which correspond to non-etched MAX phases and MXenes. A high amount of $\text{Ti}_3\text{C}_2\text{T}_x$ phase emerged in the slurry and suspension as a result of Al etching followed by washing cycles, with the main peak (002) appearing between 6° and 8° [25,28]. Aluminum, which was present between the Ti-C layers, was efficiently removed and was not present in the membranes, as observed by the absence of the corresponding MAX phase peaks. The formation of the MXene phase was also confirmed by the dark-greenish color of the supernatant, which came out after the third centrifugation cycle instead of a purple/magenta color, which indicates the presence of secondary phases such as Ti_2CT_x [25] (Figure S1 in Supplementary Materials). The small fraction of the precursor MAX phase, which can be observed in the slurry and suspension spectra, disappeared in those representing the Ti_3C_2 due the decantation step introduced after centrifugation, which led to the suspension being purified before the membrane processing.

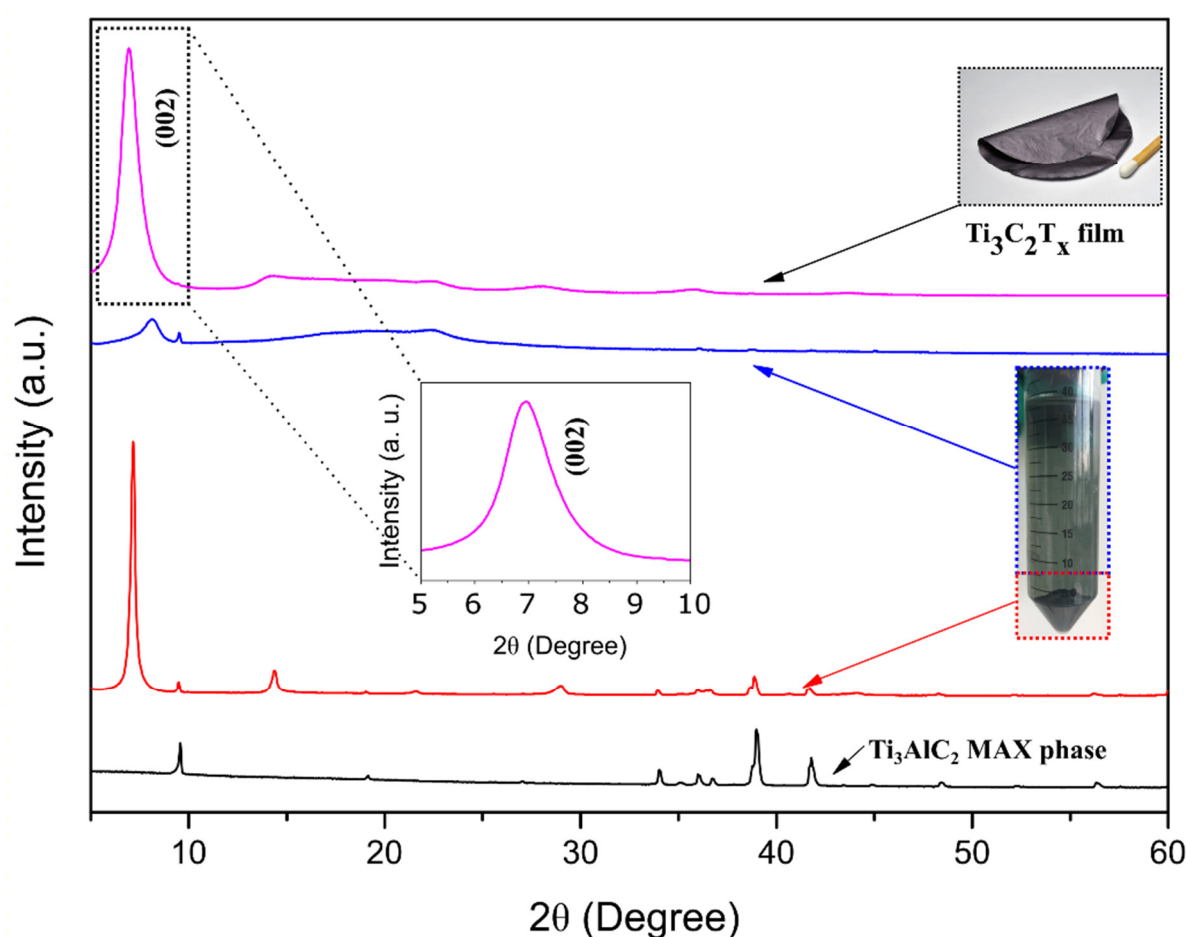


Figure 1. XRD patterns of the MAX (Ti_3AlC_2) powder (black), the slurry (red) and suspension (blue) after the third centrifugation cycle and the Ti_3C_2 MXene free-standing membrane (pink).

The slurry and the free-standing membrane produced from the vacuum-filtrated supernatant after the third centrifugation cycle are shown in Figure 2. Panels (A) and (B) show the presence of MXene particles in the slurry and indicate the presence of MXene flakes

instead of non-etched Ti_3AlC_2 MAX phases, which agrees with the XRD results and indicates the efficiency of the synthesis method. It is important to consider that MXene flakes presented a barely distinguishable compact structure from that of MAX-phase powder thanks to the narrow space between the two-dimensional layers, as was also found in Ti_3AlC_2 particles. The expanded accordion-like structure usually occurs due the exothermic reaction between HF with a high concentration and Al between Ti-C bonds [9], which was not similar to the MXene microstructure obtained in this work. The observed structure is similar to those of MXenes synthesized at low HF concentrations [25]. It was possible to not only observe MXene flakes, but also the presence of transparent monolayers of Ti_3C_2 , which can be explained by the easy delamination process which occurs along the centrifugation. The remaining solvated Li ions stayed between the MXene layers after synthesis during this phase [29]. Panels (C) and (D) show the microstructure of the free-standing membrane resulting from vacuum-assisted filtration of 30 mL of the supernatant after the third centrifugation cycle (0.22 mg/mL). The film surface containing deposited Ti_3C_2 nanosheets with a thickness of around 280 nm is shown in panel (D). It can be noted that the MXene layers are not fully aligned in a few regions, and there are also some wrinkles. Wrinkles lead to imperfections in the membrane microstructure and lower efficiency in permeation [30]. This factor can be explained by the heterogeneous distribution of terminations [31,32] and the presence of water [33]. Thus, the piled free-standing membrane, which was produced for posterior characterization, showed purity, well-deposited MXene layers and mechanical stability, as observed in previous reports [34], which can reproduce the same quality of those nanosheets deposited on $\alpha\text{-Al}_2\text{O}_3$ substrates.

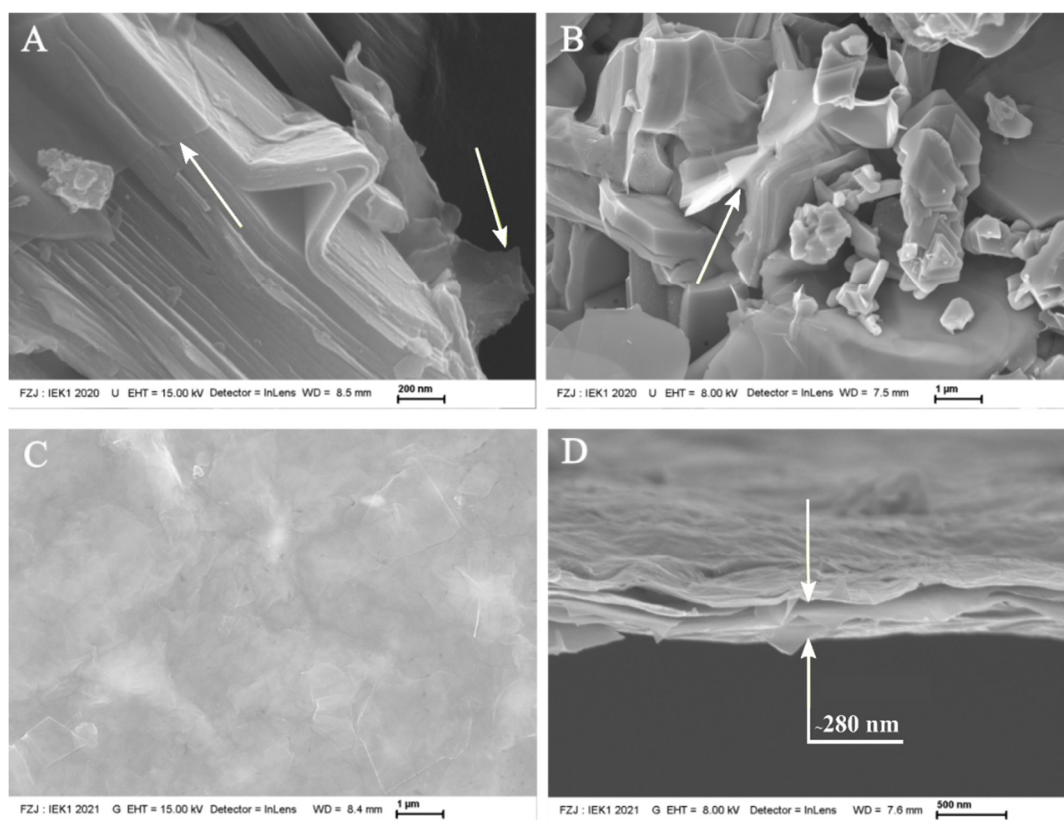


Figure 2. (A,B): SEM images of the slurry after the third centrifugation cycle showing the presence of transparent nanosheets, as indicated by the yellow arrows. The free-standing membrane: (C) surface morphology and (D) fracture surface.

Figures 3 and 4 show the spectrum of dip-coated Al_2O_3 substrates heat-treated at 80 °C under vacuum and at 500 °C under a H_2/Ar atmosphere. The respective graphs

comprise the superposition of peaks from samples treated under the same conditions. The graphs indicate the formation of pure MXene phase, where the highest peak (002) showed a sharpness which is characteristic of the stacking of Ti_3C_2 nanosheets due to ordered deposition (i.e., by horizontal stacking) [16]. The Al_2O_3 phase of the substrates is represented by the peaks marked with red squares and could be observed in all spectra. In this work, the heat treatment at higher temperatures for both samples dip-coated with the two different concentrated suspensions (C_1 and C_2) resulted in a shift in the peak from lower to higher angles. A small change in the second and third Ti_3C_2 peaks representing samples thermally treated at 80°C can be noticed when compared to those treated at 500°C . The spectrum of the sample treated at a higher temperature showed a small new peak at 17° [16], which suggests stacking of nanolayers, as also reported by Arshadi et al. When comparing the figures, it is important to note that the main phase was kept even after heat treatment B, which evidently suggests the absence of secondary and unlike phases. Additionally, the superposition of the peaks indicates reproducibility regarding the presence of the Ti_3C_2 phase.

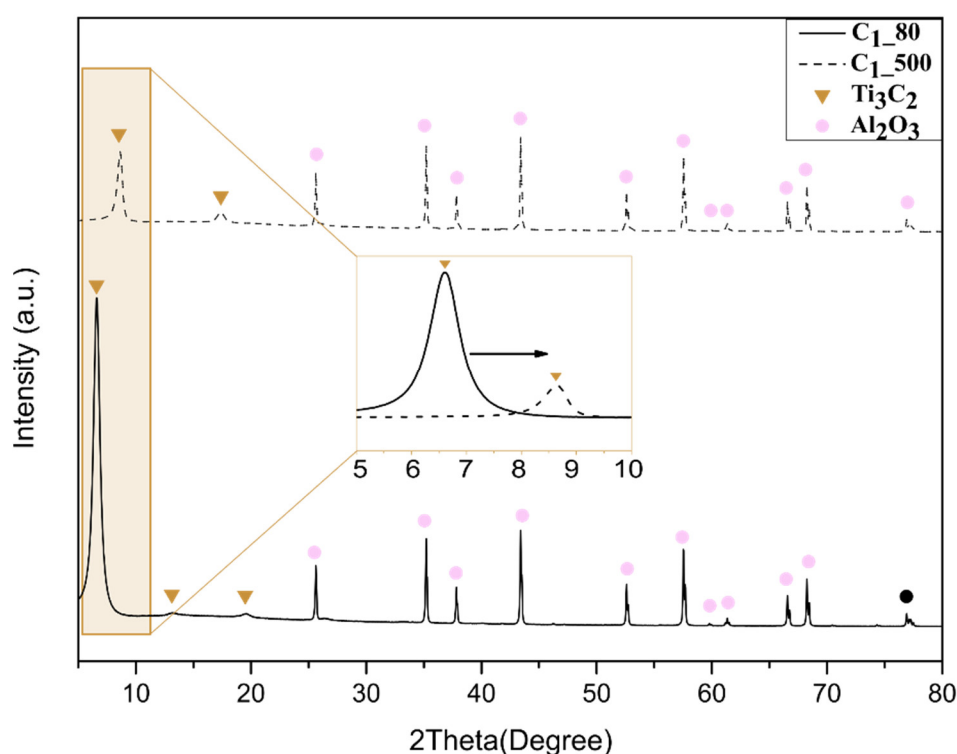


Figure 3. XRD patterns of Al_2O_3 substrates coated with Ti_3C_2 by dip coating using suspension C_1 treated at 80°C under vacuum and 500°C under H_2/Ar atmosphere.

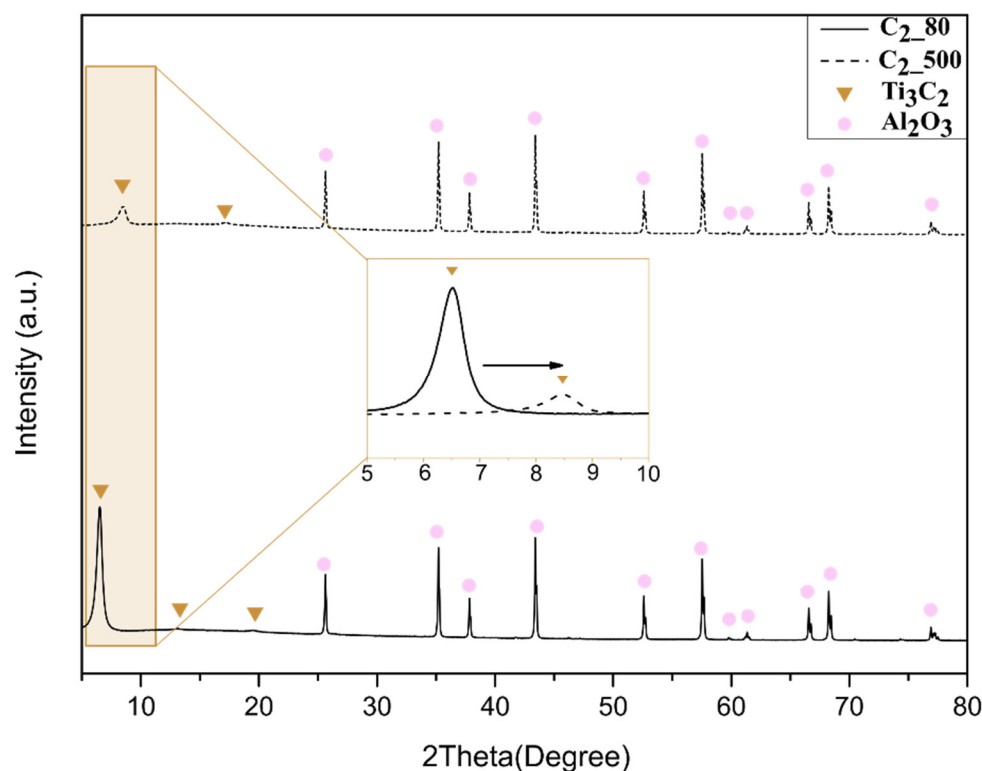


Figure 4. XRD patterns of Al_2O_3 substrates coated with Ti_3C_2 by dip coating using suspension C_2 treated at 80 °C under vacuum and 500 °C under H_2/Ar atmosphere.

The surface morphology and fracture surface of Ti_3C_2 supported on $\alpha\text{-Al}_2\text{O}_3$ substrates are shown in Figures 5 and 6, respectively. The first sample produced by dip coating (C_1_{80}) presented a surface microstructure without impurities or defects (Figure 5A), and although Ti_3C_2 nanosheets are transparent, the film was thick enough that the microstructure of the substrate could not be seen when compared to thinner membranes (i.e., C_2_{80} , Figure 6C). Due to the ease of peeling off the substrate, all the samples produced with C_1 and C_2 concentrations could provide the fracture surface of the free-standing membrane to be analyzed by SEM.

The heat treatment had a considerable impact on the membrane thickness produced by different concentrations of suspensions C_1 and C_2 . A decrease in thickness could be observed with the increase in temperature concerning the heat treatment followed by the membrane processing (i.e., dip coating). MXenes supported on α -alumina substrates treated at a higher temperature (condition B, Figures 5D and 6D) exhibited an almost 20% reduced thickness compared to those treated at 80 °C under vacuum (condition A, Figures 5B and 6B). Confirming the narrowing of interlayer spacing demonstrated by XRD in the previous section, the decrease in thickness occurred as expected for those samples treated at 500 °C under a H_2/Ar atmosphere. Additionally, the absence of defects on the surface of those samples heat-treated at lower temperatures was also seen in those sintered at a higher temperature, which shows that the chosen free-oxygen atmosphere is better than the air atmosphere used in previous works for reducing d-spacing due to the higher temperature that led to the presence of secondary phases and defects [20].

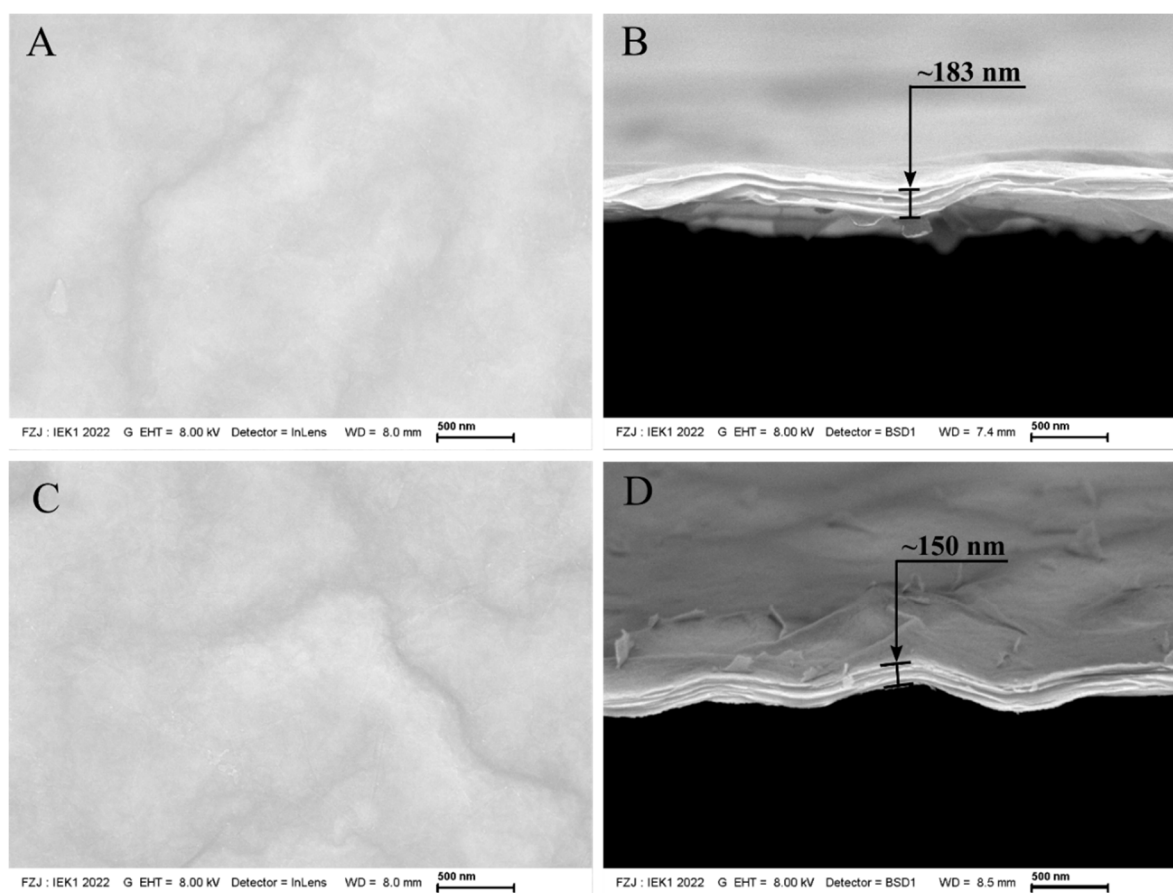


Figure 5. (A–D): SEM images of Ti_3C_2 MXene membranes produced via dip coating on an $\alpha\text{-Al}_2\text{O}_3$ substrate from suspensions C_1 treated at 80 °C and 500 °C under H_2/Ar atmosphere, respectively.

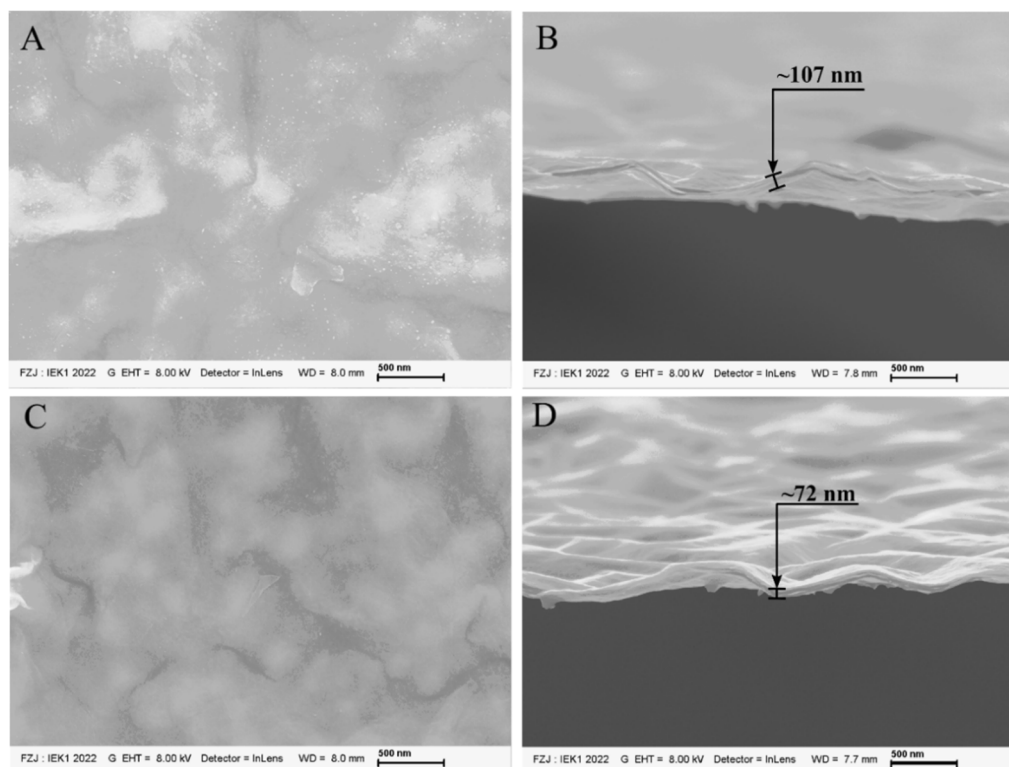


Figure 6. (A–D): SEM images of Ti_3C_2 MXene membranes produced via dip coating on an $\alpha\text{-Al}_2\text{O}_3$ substrate from suspensions C_2 treated at 80 °C and 500 °C under H_2/Ar atmosphere, respectively.

Figure 7 shows an enlarged view of the region which corresponds to the main peak of MXenes (002), highlighting the shift in the peak (002) towards a higher angle in samples treated at higher temperatures (condition B). Peaks at lower angles, approximately 6.5° , correspond to C_1 and C_2 suspensions treated at 80°C (condition A). The peaks which appeared around 8.65° represent the main Ti_3C_2 peaks of those treated at a higher temperature (500°C). These observations indicate smaller MXene d-spacing for samples dried under condition B according to Bragg's law.

Previous studies have shown a shift in the mean peak (002) through different techniques for controlling interlayer spacing [16,35,36]. Cheng et al. treated MXene with a H_2 atmosphere at higher temperatures, up to 400°C , to check if there was a shift in cycling performance; however, they did not investigate the effect of the thermal treatment on interlayer spacing or stacking [37]. The magnification of the main peak showed the effect of the temperature. A considerable shift in the peak was achieved for the two different heat treatments, and even the samples presenting different thicknesses (i.e., C_{1_80} and C_{2_80} or C_{1_500} or C_{2_500}) showed similar (002) peak values when heat-treated at the same conditions. As the diffraction peaks give rise to the d-spacing according to Bragg's law, the free path can be assumed considering MXene nanolayers are $\sim 1\text{ nm}$ [25]. Such similarities in peak values suggest that temperature under an oxygen-free atmosphere is the main parameter regulating the free path, and does not depend on the membrane thickness. The narrowing of the interlayer spacing can be attributed to water loss and de-functionalization ($-\text{OH}$), and the low free energy at higher temperatures favored the decrease in interlayer spacing [19]. The interaction between the adjacent nanosheet favored changing the $-\text{OH}$ part to $-\text{O}$ functional groups, which helps to enhance the chemical bonding and consequently to decrease the interlayer spacing, as also observed in [37].

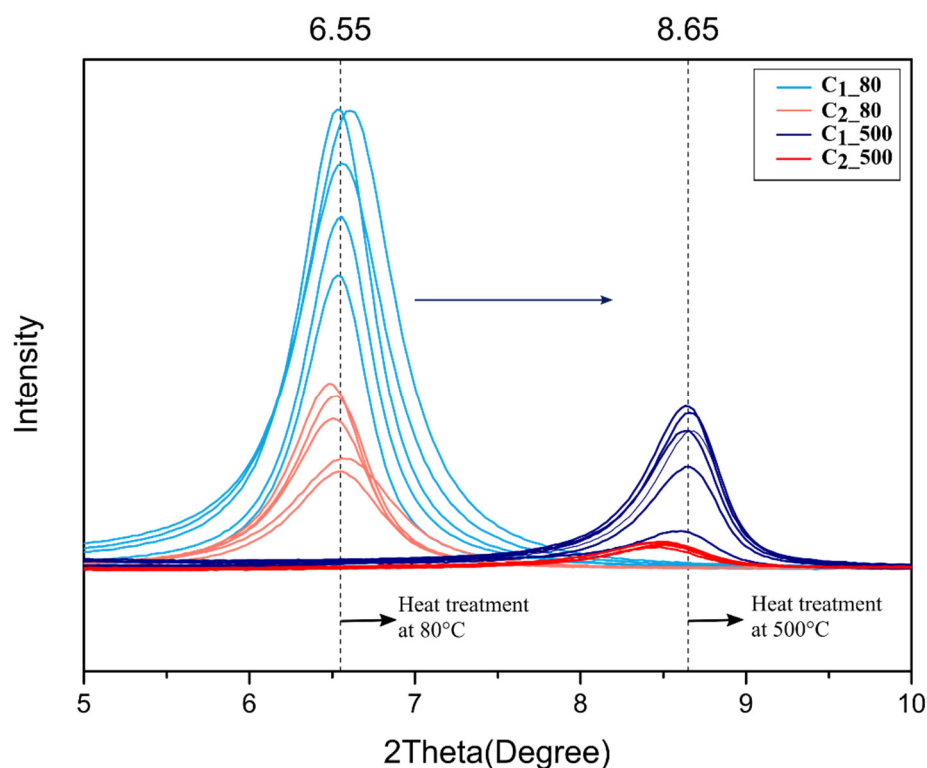


Figure 7. XRD patterns of samples heat-treated at different conditions. An interval from 5° to 10° shows the main MXene peak (002).

Table 1 shows the obtained c-lattice values for the C_{1_80} , C_{1_500} , C_{2_80} and C_{2_500} samples, and their respective free d-spacings are exposed, which suggests similarity

between the interlayer space of samples thermally treated at the same temperature. Similarly, the d-spacing in previous reports was obtained through Bragg's Law by the use of constant c-lattice values. The average (002) reflection at $2\theta = 6.55$ previously shown in Figure 4 corresponds to an interlayer distance of 0.347 and 0.352 nm for C_{1_80} and C_{2_80} samples, respectively. The average (002) reflection at $2\theta = 8.65$ corresponds to an interlayer distance of 0.024 and 0.040 nm for C_{1_500} and C_{2_500}, respectively.

Table 1. C-lattice constants and respective d-spacing obtained through Bragg's law for samples produced with the most concentrated suspension (C₁) and the least concentrated suspension (C₂).

Sample ID	c-Lattice (Å)	d(002) (Å)	free d-Spacing (nm)	Average Free D-Spacing (nm)
Heat treatment temperature 80 °C				
C1_1	26.623	13.311	0.331	0.347 ± 0.003
C1_2	27.036	13.518	0.352	
C1_3	26.940	13.470	0.347	
C1_4	26.782	13.391	0.339	
C1_5	26.952	13.476	0.348	
C2_1	27.079	13.540	0.354	0.352 ± 0.003
C2_2	27.195	13.597	0.360	
C2_3	26.755	13.378	0.338	
C2_4	26.980	13.490	0.349	
C2_5	27.092	13.546	0.355	
Heat treatment temperature 500 °C				
C1_1	20.467	10.234	0.023	0.024 ± 0.001
C1_2	20.413	10.206	0.021	
C1_3	20.491	10.246	0.025	
C1_4	20.580	10.290	0.029	
C1_5	20.475	10.237	0.024	
C2_1	20.773	10.387	0.039	0.040 ± 0.001
C2_2	20.891	10.445	0.045	
C2_3	20.808	10.404	0.040	
C2_4	20.766	10.383	0.038	
C2_5	20.903	10.452	0.045	

The Raman spectrum was collected from two 1-micron MXene films, each of which was treated at different conditions. The technique was able to provide information about the surface chemistry, stacking and quality of Ti₃C₂T_x. Figure 8 shows the comparison between both the Raman shift, which exhibited characteristic wave numbers correlated to vibrations of Ti₃C₂T_x, and their respective surface functionalization. Similarly, as shown in [38], these vibrations of Ti₃C₂T_x consisted of E_g (in-plane) and A_{1g} (out-of-plane) peaks, both of Ti and C atoms. Thus, the spectrum was divided into three regions: the flake region, which corresponds to a group vibration of carbon, two titanium layers and functional groups; the Tx region, which represents vibrations of surface groups; and the carbon region (C region), in which both in-plane and out-of-plane vibrations can be found. Both studied free-standing membranes presented vibrations which correspond to the peaks at 205, 255, 286, 404, 580, 622 and 725 cm⁻¹, but small differences in peak intensities can be observed. As the laser utilized for this measurement was 532 nm, the first resonant peak coupled with the plasmonic peak was not detected [32].

Next, the A_{1g} peak at 205 cm⁻¹ and the E_g peak at 255, 286 and 404 cm⁻¹ indicate vibrations due to surface groups of titanium. The E_g and A_{g1} peaks at 580/622 cm⁻¹ and 725

cm^{-1} , respectively, correspond to the carbon vibrations [31]. The line broadening and merging in the spectra is indicative of exfoliation and delamination of the Ti_3AlC_2 precursor. Annealing at a higher temperature (condition B) led to the partial removal of functional groups [39]. In agreement with this, a small difference in the general peaks was observed, but the nature of the functional group was maintained. In addition, heat treatment can lead to changes in the surface functionalization of MXenes. There is a tendency for the $-\text{OH}$ functional groups to change to $-\text{O}$, which leads to increased membrane mechanical stability and a decrease in d-spacing [32]. Therefore, sintering of the MXene membrane at a higher temperature, such as 500°C under an oxygen-free atmosphere, led to conservation of the surface functionalization and showed that MXene flakes were not affected by this increase in temperature (i.e., no important changes were noted), which is in contrast to the considerable removal of functional groups after treatment around 500°C in air [40]. Membranes after heat treatment and consequently after removing water between the nanolayers are supposed to remain chemically stable, leading to longer times when stored in an oxygen-free atmosphere [13,41]; however, they have limitations regarding their use in humid environments, especially at high temperatures [22,42].

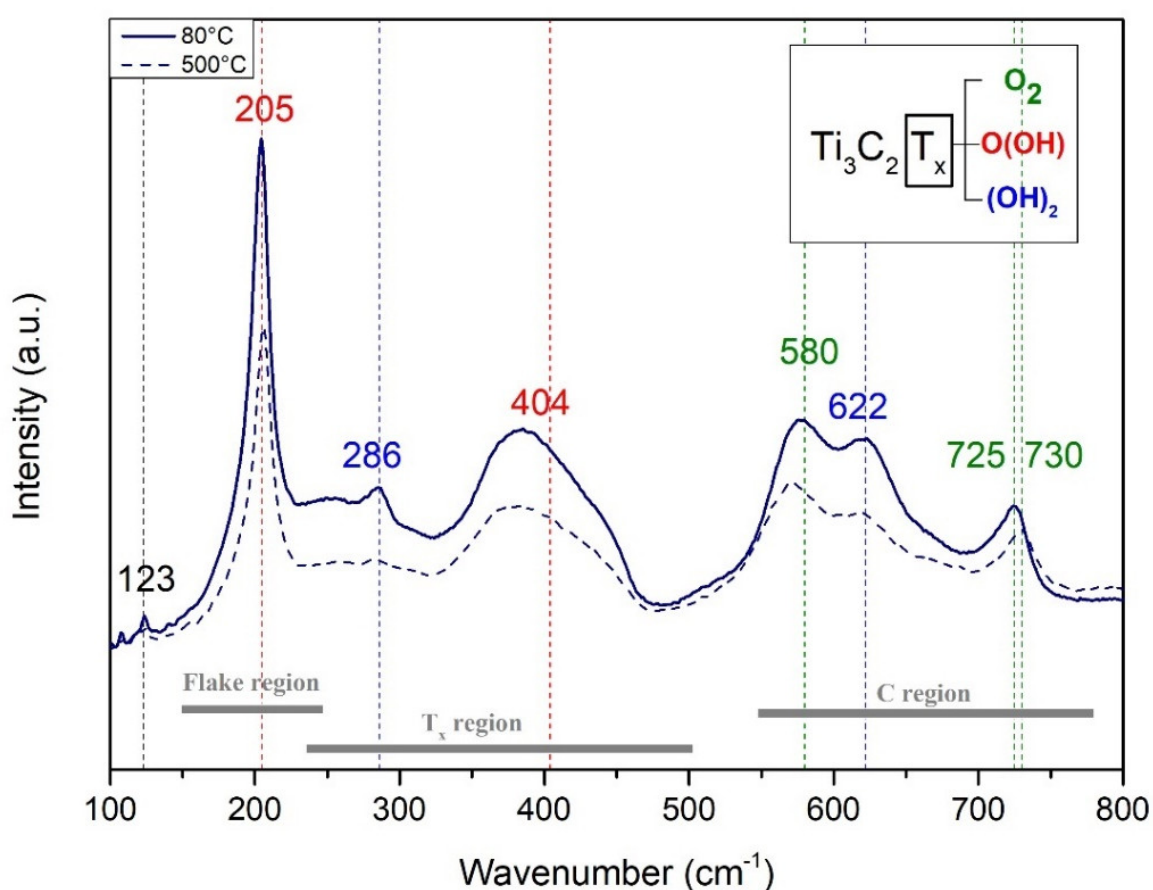


Figure 8. Raman spectra of Ti_3C_2 (laser 532 nm). Tx functional terminations were compared to the literature.

The TEM results suggest a distinct deposition of Ti_3C_2 nanosheets on the surface of Al_2O_3 substrates, which were processed by dip coating using the most concentrated suspension, which was identified as C_1 . Although both of them presented a continued thin

film on the surface of a porous Al_2O_3 substrate, a slight difference in thickness can be observed between the sample heat-treated in condition A (Figure 9A) and the one heat-treated in condition B (Figure 9C); this agrees with what was seen in SEM images (Figures 5 and 6) resulting from the decrease in the average d-spacing confirmed by XRD analysis (Table 2). However, some defects remained after heat treatment.

Table 2. H_2 and CO_2 permeabilities and permeances and the respective H_2/CO_2 selectivity.

Sample ID	H_2 Permeability (Barrer)	H_2 Permeance ($10^3 \text{ mol/m}^2 \text{ s}^1 \text{ Pa}^1$)	CO_2 Permeability (Barrer)	CO_2 Permeance ($10^3 \text{ mol/m}^2 \text{ s}^1 \text{ Pa}^1$)	H_2/CO_2 Selectivity
<i>Heat treatment temperature 80 °C</i>					
C1_1	6.4	33.7	2.0	10.2	3.3
C1_2	2.1	10.9	0.5	2.5	4.4
C1_3	2.0	10.5	0.7	3.6	3.0
C2_1	16.4	254.7	4.9	71.7	3.3
C2_2	23.9	85.4	4.9	25.8	4.8
C2_3	48.8	124.9	13.7	25.8	3.6
<i>Heat treatment temperature 500 °C</i>					
C1_1	0.86	4.5	0.12	0.6	7.0
C1_2	0.54	2.8	0.09	0.5	5.8
C1_3	0.83	4.4	0.14	0.7	6.0
C2_1	35.6	186.2	9.1	47.6	3.9
C2_2	9.7	50.8	2.5	12.9	3.9
C2_3	28.5	149.1	7.3	37.9	3.9

For this work, heat treatment at higher temperatures led to stacking of MXene nanosheets (Figure 9D), while the heat treatment at lower temperatures showed the considerable presence of defects such as an increase in d-spacing between the layers [38,43] due the non-well deposition of MXene nanosheets or the presence of water (Figure 9B). Similar to a study by Zheng, Z. et al. that investigated the surface morphology of MXenes supported on alumina substrates after hydrogen annealing, the nanolayers were well-dispersed in the film; therefore, the boundaries of MXene sheets were more indistinct after hydrogen annealing, which can be explained by the formation of a tightly packed network via the partial sintering of sheets at higher temperatures [43].

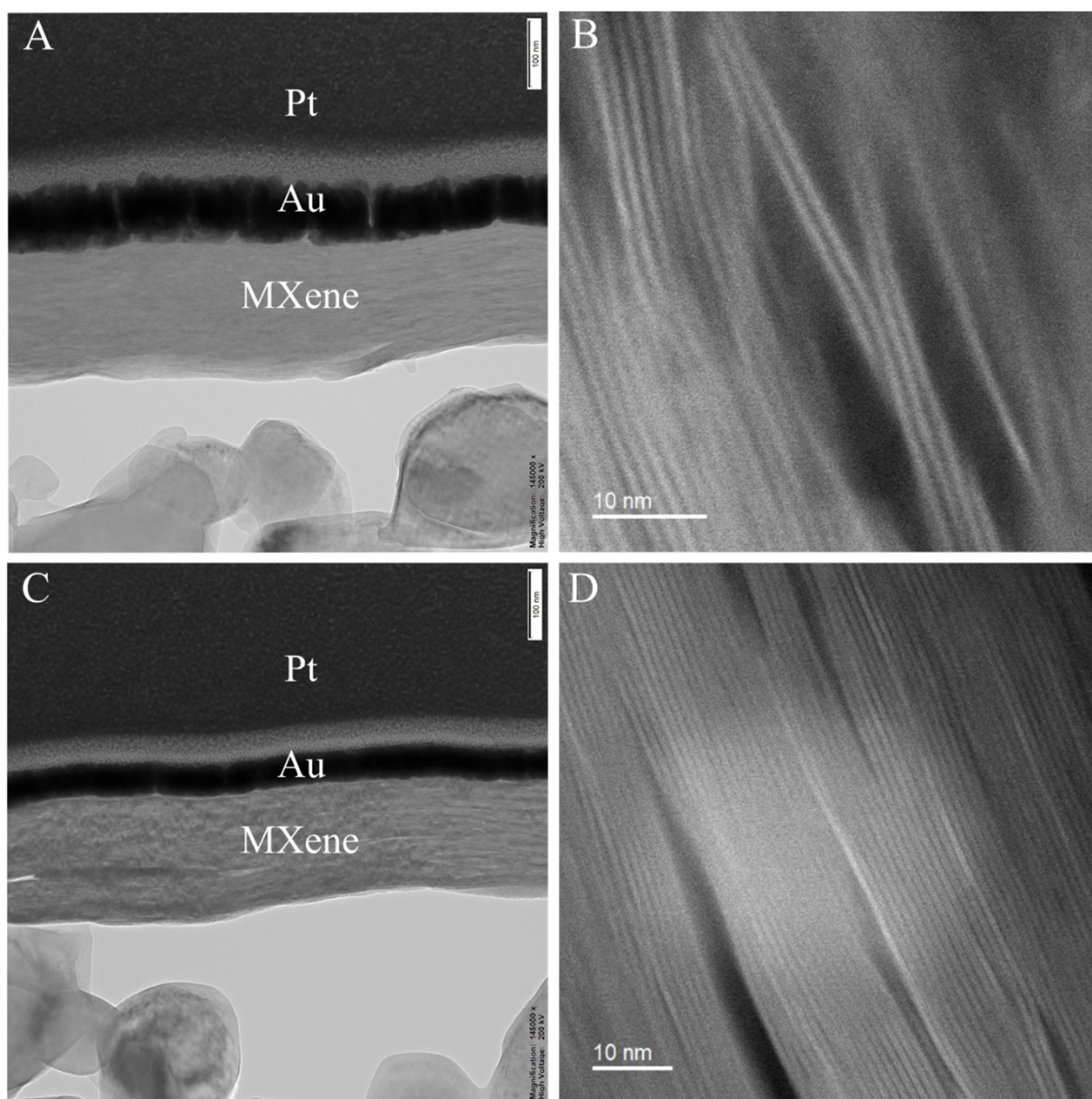


Figure 9. Morphology and structure of exfoliated MXene ($\text{Ti}_3\text{C}_2\text{T}_x$) nanosheets and stacked MXene membranes. (A) SEM image showing the cross-section of the sample C_{1_80}; (B) TEM images of the sample C_{1_80} showing regions containing non-ordered and spaced nanosheets; (C) SEM image showing the cross-section of the sample C_{1_500}; (D) TEM images of the sample C_{1_500} indicating regions with stacked MXene nanosheets.

In order to evaluate the chemical stability of the membrane for further investigations on the heat treatment to narrow the d-spacing between the 2D nanolayers, a TGA investigation was carried out under two atmosphere conditions, namely vacuum and an Ar/H₂ atmosphere (Figure 10a,b), respectively. The mass loss curve of free-standing MXenes measured under vacuum showed that the slight initial water loss of 4.96% was followed by a mass gain path, indicating oxidation (i.e., conversion of titanium carbide into titanium dioxide), as indicated by the XRD pattern of the final powder. The functional groups between the MXene layers, generally $-\text{OH}$, $-\text{F}$ and $-\text{O}$, favor the interaction of the oxygen

present in the air or water and promote its oxidation [13], which indicates the presence of remaining oxygen during the measurement under vacuum.

The chemical stability of MXenes depends on its exposure to an oxidative environment and limits their potential in applications where the longevity of the material is required [44]. Herein, the total loss of water occurred until the temperature achieved a value around 450 °C. Other reports have also shown that water was removed at temperatures higher than 300 °C [45]. Figure 10a indicates an improvement in the chemical stability of the free-standing membrane concerning the thermal evaluation under a protected atmosphere. Similarly, as previously indicated, the first mass loss of 1.58% occurred at temperatures below 200 °C. The second and largest loss in mass, 9.64%, started at a temperature near 835 °C and showed phase transition from $\text{Ti}_3\text{C}_2\text{T}_x$ to TiC, which took place at higher temperatures when compared to those heat-treated under vacuum. In previous studies, it was observed that the structural phase transition from $\text{Ti}_3\text{C}_2\text{T}_x$ to cubic TiC may occur at annealing temperatures higher than 900 °C [40], as also observed in this work. The absence of titanium oxide indicates a free-oxygen atmosphere as a proper condition to sinter MXene films in order to achieve smaller free interlayer spacings.

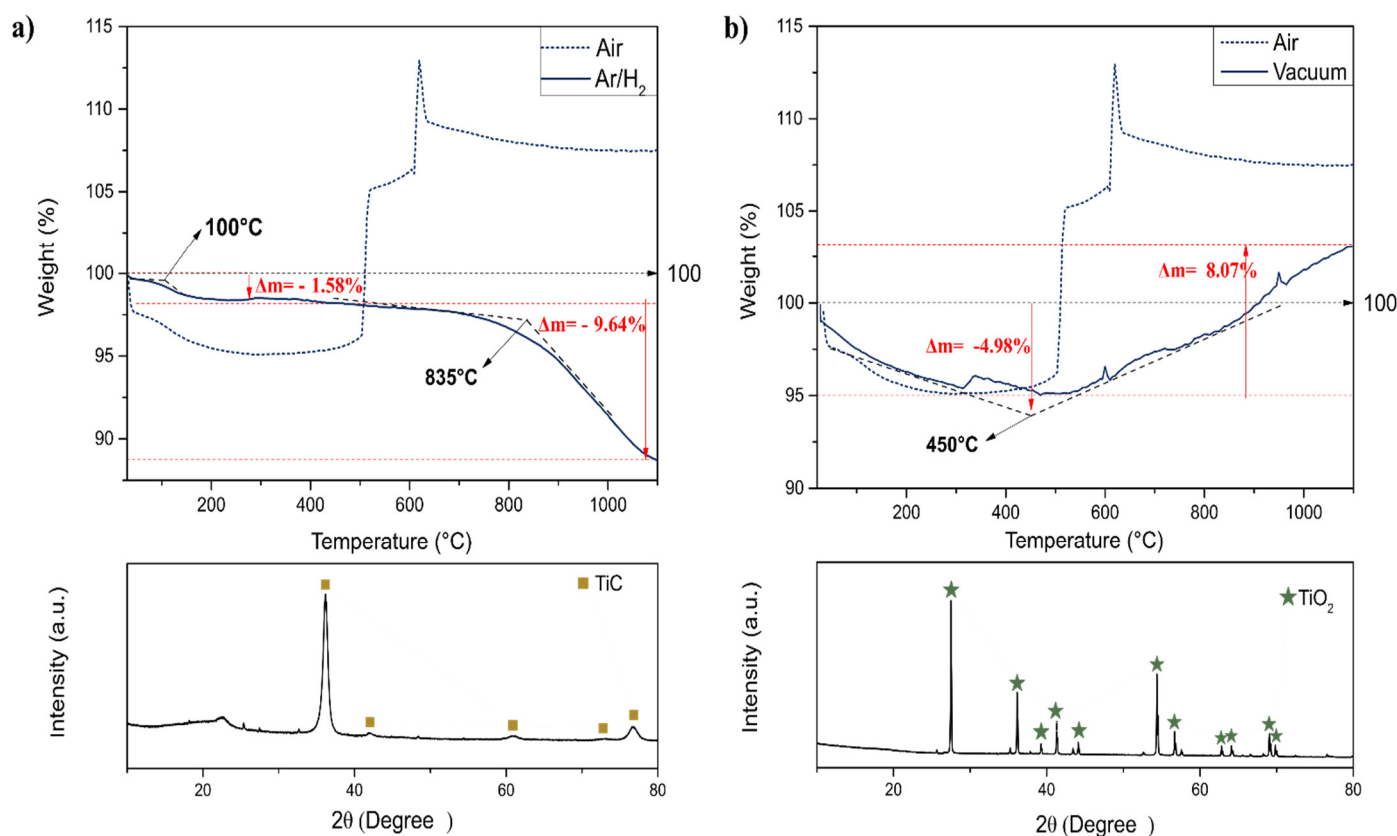


Figure 10. TG curves of $\text{Ti}_3\text{C}_2\text{T}_x$ MXene (a) under vacuum and (b) in H_2/Ar atmosphere and their respective XRD patterns representing the phases which were formed after TG measurement.

3.2. Gas Permeation Tests

Single-gas permeation tests were carried out at room temperature and under different pressures for all samples, respecting the ascending order of the gas ratio from the smallest to the largest: He, H_2 , CO_2 , N_2 and CH_4 . Three samples for each condition were reproduced and tested by single-gas permeation. The dip-coated substrates were put into a massive permeation module and sealed with a rubber ring (Figure S2 in Supplementary Materials) to measure the gas permeation performance. This work showed a higher permeability of small molecules such as He and H_2 when compared with larger molecules

(e.g., CO₂, N₂ and CH₄). Stacked membranes with a free interlayer spacing of 0.024 nm that were obtained after heat treatment presented low permeability values for smaller and larger molecules due the presence of defects, as previously analyzed by TEM. Only membranes produced by dip coating could be tested due to their mechanical stability when deposited on alpha-Al₂O₃ substrates, which held the membrane when a negative pressure was applied (common procedure during the gas switching).

In order to analyze the influence of the thickness of MXene films on the membrane behavior, samples produced using different concentrations (C₁ and C₂) followed by the same drying condition (A or B) were compared, as shown in Figures 11–14. Figures 11 and 13 show the permeance curves and H₂/CO₂ selectivity of the samples C₁ and C₂ treated at 80 °C (condition A), indicated by curves and bars, respectively. It is possible to note that a decrease in thickness led to higher permeances, as also seen in a previous work [14]. However, the C_{1_80}, ~180 nm thickness showed H₂ permeance higher than 10 [10³ mol/m²sPa], and the C_{2_80}, ~107 nm thickness increased it to values higher than 250. Similarly, for samples C₁ and C₂ annealed at 500 °C (condition B, Figures 12 and 14, respectively), the decrease in thickness was accompanied by higher permeances from approximately >4.6 to >180 [10³ mol/m²sPa]. Additionally, a loose microstructure, which can be seen in Figure 9B, is related to the flux increase and consequently the permeability [46].

Jeng et al. showed that the disorder of GO layers induced a short pathway in the membrane, implying the permeation increase. Regarding H₂/CO₂ selectivity, slight differences were observed at 1 bar between C_{1_80} and C_{2_80}. Although both presented Knudsen diffusion, which occurs when the mean free path of a gas molecule is larger than the channel through which it travels [47], the thicker membrane with the H₂/CO₂ selectivity factor ~4.69 as a dominant gas separation mechanism showed a small increment in selectivity, as selectivity is increased by the decrease in the membrane thickness, as also seen in [48]. In the same way, higher H₂/CO₂ selectivity was observed for thicker samples heat-treated at 500 °C, such as C_{1_500}.

Interlayer spacing and a decrease in defects must be considered as parameters to improve the selectivity among the factors influencing a gas-separating membrane's behavior [36]. Thus, pure MXene membranes produced using the same concentrated suspension, C₁, and treated at different temperatures (Figures 11 and 13, respectively), were compared to analyze this important factor. Substrates treated at 80 °C presented an average free-interlayer spacing between ordered nanolayers, approximately 0.35 nm, which enabled molecules such as H₂ and CO₂ with different molecular sizes to move due to the large free path. Additionally, the larger spacing previously observed in this work (i.e., defects generated by the non-ordered deposition of nanosheets as well as the presence of water, seen in Figure 9B) strongly influenced the diffusion of molecules with no size selection. In turn, molecular weight defined the selectivity in this diffusion mechanism, which was lower than 4.69 (the substrate in this work was measured and showed a H₂/CO₂ selectivity of 4.5, Figure S4 in Supporting Information).

Then, a considerable decrease in average spacing was observed after heat treatment at a higher temperature, and as expected, a decrease in flux of all gases and consequently reduced permeance was observed. Additionally, the selectivity increased to 7 (1 bar), which is considered molecular sieving. A free path in the interval between hydrogen and carbon dioxide molecular size should be expected for this diffusion mechanism, which is not in agreement with the average free interlayer spacing obtained in this work, 0.024 nm, constituting well-stacked MXene nanolayers. Thus, the results suggest that larger spacings (defects), which were maintained even after the heat treatment, were capable of selecting larger molecules from smaller molecules, improving the diffusion mechanism. Even though there is a partial removal of functional groups, it is supposed that the decrease in defects and in interlayer spacing was mainly associated with the water removal and functional group modification at high temperature (i.e., –OH tends to change to –O), which contributed to a greater interaction between adjacent nanosheets. In addition, the fact that the adsorption capacity of CO₂ in the MXene interlayer is superior due to its

higher quadrupole moment compared to other gases (i.e., N_2) explained an increase in the separation factor due to the decrease in CO_2 diffusion into the MXene nanochannels in parallel with the faster diffusion of smaller molecules such as H_2 [23].

Therefore, this shift in the diffusion mechanism from Knudson to molecular sieving was only observed for thicker membranes treated at higher temperatures due to the instability of the thinner membranes and greater possibility of leakage or mechanical damages (i.e., abrasion) during the module assembly and operation or with the increase in pressure (Figure S3, in Supplementary Materials).

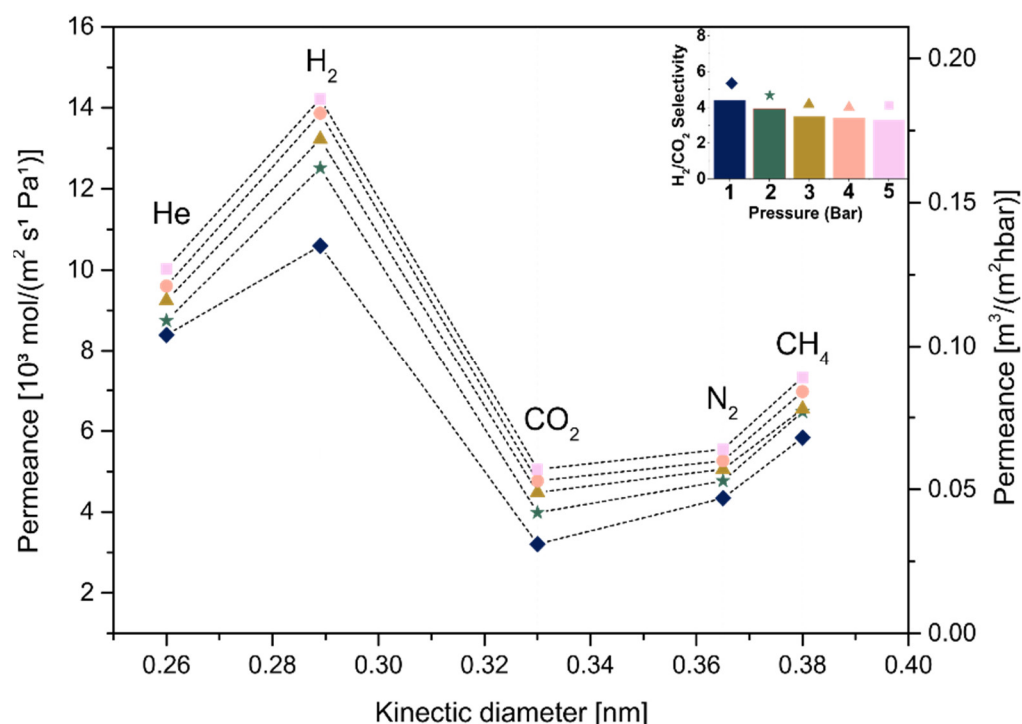


Figure 11. Single-gas permeabilities, permeances and H_2/CO_2 selectivity under different pressures for C1_80.

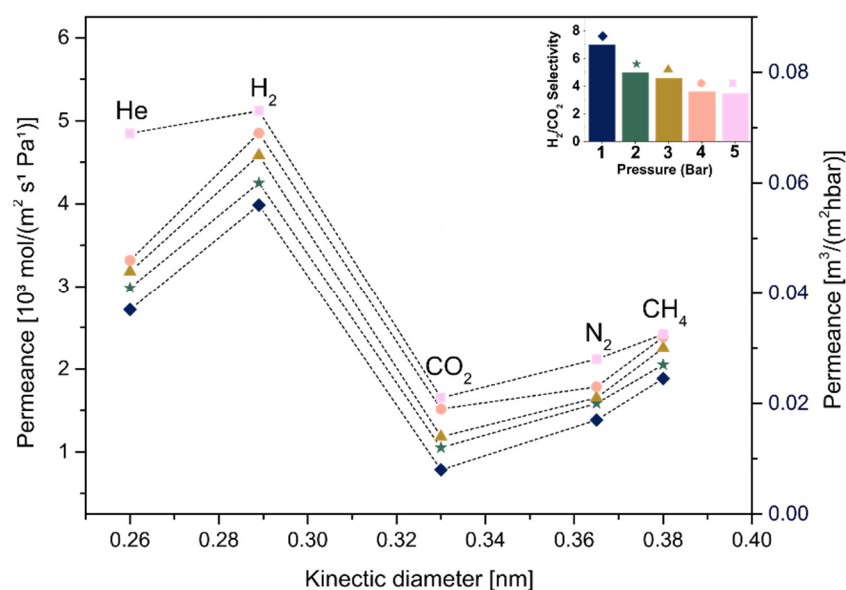


Figure 12. Single-gas permeabilities, permeances and H_2/CO_2 selectivity under different pressures for C1_500.

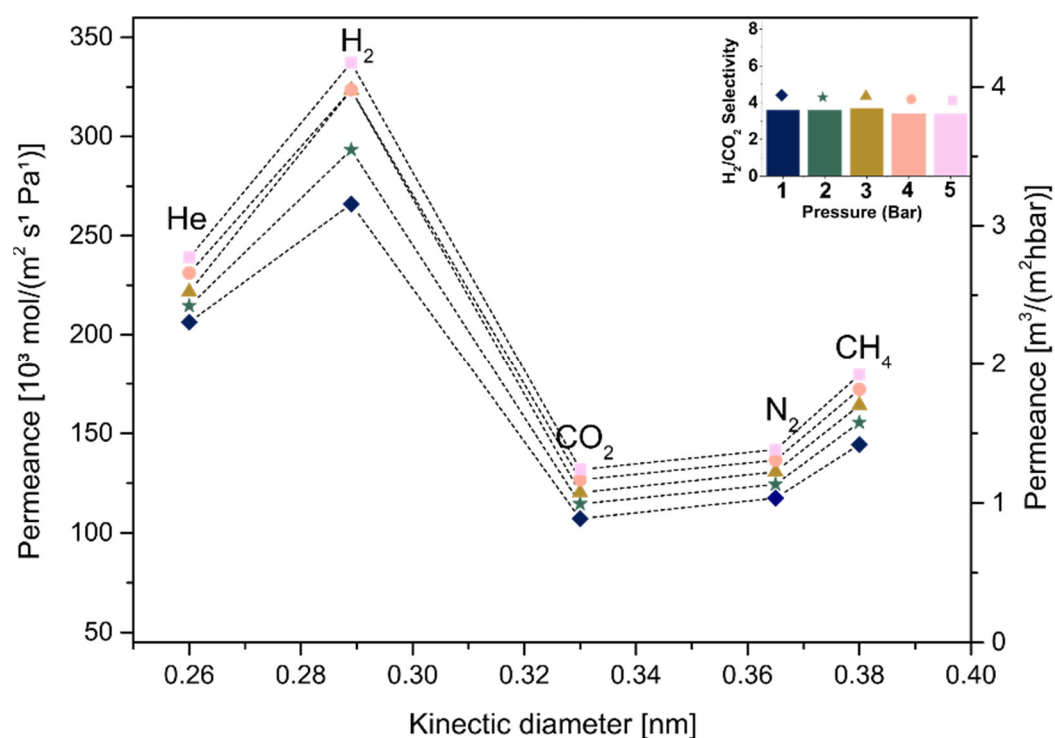


Figure 13. Single-gas permeabilities, permances and H_2/CO_2 selectivity under different pressures for C2_80.

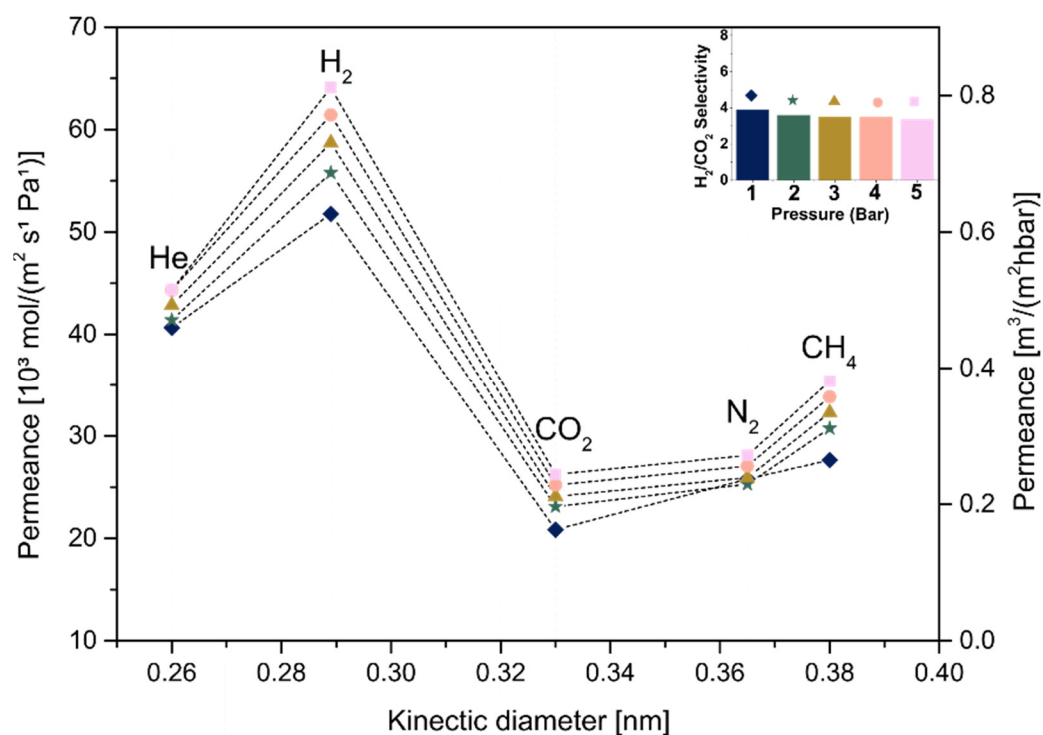


Figure 14. Single-gas permeabilities, permances and H_2/CO_2 selectivity under different pressures for C2_500.

Permeabilities and the respective selectivity values for the membrane batches used in this study are summarized in Table 2. As already reported, thinner membranes are required to improve the increase in gas flux and well-defined pore size [49] or interlayer

spacing [35] to maximize selectivity. In this work, thinner membranes showed higher permeability values and lower H_2/CO_2 selectivity. Among the reasons which explain this behavior, the mechanical instability of the membrane and leakage [49] (Figure S1 in Support Information), which may occur during the single-gas permeation tests and/or preparation of the sample inside the module and/or mechanical instability of the membrane, contributed to the decrease in gas separation efficiency. The small change concerning the slight removal of terminating groups, as observed through the previously discussed Raman spectrometry, might lead to weakening of the chemical interaction between the MXene layers. However, since the removal happened in a smaller proportion, a considerable part of the functional groups as well as their variety was maintained, which contributed to maintaining a strong interaction between the MXene nanosheets. Moreover, the interaction between adjacent nanosheets favored a change of functional groups at higher temperatures (i.e., from $-OH$ to $-O$ terminations), as also observed in [37]. For thinner membranes (i.e., C2_80), the partial removal of the functional groups did not considerably interfere with the selectivity. However, it can be assumed that a lower mechanical instability due to its thinner thickness was the crucial reason for the lower separation performance. Considering long-term stability, the decrease in d-spacing after heat treatment at higher temperatures contributed to stacking MXene nanosheets, which favors chemical stability when compared to individual flakes with a high surface area.

4. Conclusions

In this work, pure and flexible MXenes were produced by controlling synthesis parameters, and laminar thin films were processed by dip coating with different concentrations of MXene suspensions. Moreover, the decrease in defects as a consequence of the d-spacing narrowing through heat treatment at 500 °C under an Ar/H_2 atmosphere promoted a change in the diffusion mechanism, which led to molecular sieving. A heat treatment under an oxygen-free atmosphere was required for layer stacking due to the chemical instability of MXenes. The absence of important changes in the nature of the pristine material as well the conservation of functional groups is an important factor which favored controlling the interlayer spacing through the heat treatment under higher temperatures. In addition, membranes with different thicknesses showed approximately the same interlayer narrowing factor, which led to the assumption that the membrane thickness did not limit the stacking efficiency by heat treatment under a controlled atmosphere. This work indicated a potential method for the efficient manipulation of MXene interlayer spacing by controlling the heat treatment conditions. The good chemical stability of Ti_3C_2 flakes after heat treatment at higher temperatures in the presence of H_2 favors the use of this method to break the permeability–selectivity trade-off of MXene-based membranes for gas separation.

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/membranes12101025/s1>. Figure S1: Dark-greenish supernatant after the third cycle of centrifugation; Figure S2: $\alpha-Al_2O_3$ porous substrates coated with nano-dispersion of $Ti_3C_2T_x$ flakes. Pore size of substrate presented an approximate value exceeding 200 nm. Surface morphology showing a few sizes of MXene flakes, which indicates different sizes, from 640 nm to 2 μm , as can be seen in panel (a), which shows a membrane 20 nm-thick and panel (b), showing a 40 nm-thick membrane; Figure S3: Free-standing membrane after the assembling inside the metal massive module (a) and thin film dip-coated on alumina oxide substrate after assembling into the module (b). In the first case, it was not possible to conduct the single-gas permeation test through the free-standing membrane. In the second case, the thinner membrane attached to the hard substrate was capable of dealing with the assembling and permeation tests; Figure S4: Single-gas permeation through the substrate at 1 bar. The H_2/CO_2 selectivity of the substrate, which supported the thin MXene membranes, was 4.6, which is comparable with previous work.

Author Contributions: A.A.E.: methodology, investigation, writing—original draft preparation; R.M.d.N.: project administration, funding acquisition, supervision; A.P.C.B.: writing—review and editing, supervision; K.R.: investigation and formal analysis; W.A.M.: conceptualization, writing—

review and editing, resources; J.G.-J.: conceptualization, writing—review and editing, supervision, resources. All authors have read and agreed to the published version of the manuscript.

Funding: This study was financed in part by the Coordenação de Aperfeiçoamento de Pessoal de Nível Superior—Brasil (CAPES/PrInt)—Finance Code 001 and Deutscher Akademischer Austauschdienst (DAAD)—Co-financed Short-Term Research Grant Brazil, 2021 (91808202).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Acknowledgments: This work acknowledges the following colleagues: Yoo Jung Sohn and Tim Van Gestel. This work acknowledges the following institutions: Coordenação de Aperfeiçoamento de Pessoal de Nível Superior—Brasil (PrInt/CAPES), Deutscher Akademischer Austauschdienst (DAAD) and the Institute of Energy and Climate Research (IEK-1)—Forschungszentrum Jülich GmbH.

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Yuan, S.; Li, X.; Zhu, J.; Zhang, G.; Van Puyvelde, P.; Van Der Bruggen, B. Covalent Organic Frameworks for Membrane Separation. *Chem. Soc. Rev.* **2019**, *48*, 2665–2681.
2. Li, J.; Li, X.; Van der Bruggen, B. An MXene-Based Membrane for Molecular Separation. *Environ. Sci. Nano* **2020**, *7*, 1289–1304.
3. Park, H.B.; Kamcev, J.; Robeson, L.M.; Elimelech, M.; Freeman, B.D. Maximizing the Right Stuff: The Trade-off between Membrane Permeability and Selectivity. *Science* **2017**, *356*, 1138–1148. <https://doi.org/10.1126/science.aab0530>.
4. Liu, G.; Jin, W.; Xu, N. Two-Dimensional-Material Membranes: A New Family of High-Performance Separation Membranes. *Angew. Chem.-Int. Ed.* **2016**, *55*, 13384–13397. <https://doi.org/10.1002/anie.201600438>.
5. Mi, B. Graphene Oxide Membranes for Ionic and Molecular Sieving. *Science* **2014**, *343*, 740–742. <https://doi.org/10.1126/science.1250247>.
6. Zornoza, B.; Seoane, B.; Zamaro, J.M.; Téllez, C.; Coronas, J. Combination of MOFs and Zeolites for Mixed-Matrix Membranes. *ChemPhysChem* **2011**, *12*, 2781–2785. <https://doi.org/10.1002/cphc.201100583>.
7. Ahmed, Z.; Rehman, F.; Ali, U.; Ali, A.; Iqbal, M.; Thebo, K.H. Recent Advances in MXene-Based Separation Membranes. *Chem-BioEng Rev.* **2021**, *8*, 110–120.
8. Varoon, K.; Zhang, X.; Elyassi, B.; Brewer, D.D.; Gettel, M.; Kumar, S.; Lee, J.A.; Maheshwari, S.; Mittal, A.; Sung, C.Y.; et al. Dispersible Exfoliated Zeolite Nanosheets and Their Application as a Selective Membrane. *Science* **2011**, *334*, 72–75. <https://doi.org/10.1126/science.1208891>.
9. Naguib, M.; Kurtoglu, M.; Presser, V.; Lu, J.; Niu, J.; Heon, M.; Hultman, L.; Gogotsi, Y.; Barsoum, M.W. Two-Dimensional Nanocrystals Produced by Exfoliation of Ti₃AlC₂. *Adv. Mater.* **2011**, *23*, 4248–4253. <https://doi.org/10.1002/adma.201102306>.
10. Lakhe, P.; Prehn, E.M.; Habib, T.; Lutkenhaus, J.L.; Radovic, M.; Mannan, M.S.; Green, M.J. Process Safety Analysis for Ti₃C₂T_x MXene Synthesis and Processing. *Ind. Eng. Chem. Res.* **2019**, *58*, 1570–1579. <https://doi.org/10.1021/acs.iecr.8b05416>.
11. Ying, Y.; Liu, D.; Ma, J.; Tong, M.; Zhang, W.; Huang, H.; Yang, Q.; Zhong, C. A GO-Assisted Method for the Preparation of Ultrathin Covalent Organic Framework Membranes for Gas Separation. *J. Mater. Chem. A* **2016**, *4*, 13444–13449. <https://doi.org/10.1039/c6ta04579k>.
12. Hantanasirisakul, K.; Gogotsi, Y. Electronic and Optical Properties of 2D Transition Metal Carbides and Nitrides (MXenes). *Adv. Mater.* **2018**, *30*, 1804779P. <https://doi.org/10.1002/adma.201804779>.
13. Chae, Y.; Kim, S.J.; Cho, S.Y.; Choi, J.; Maleski, K.; Lee, B.J.; Jung, H.T.; Gogotsi, Y.; Lee, Y.; Ahn, C.W. An Investigation into the Factors Governing the Oxidation of Two-Dimensional Ti₃C₂ MXene. *Nanoscale* **2019**, *11*, 8387–8393. <https://doi.org/10.1039/c9nr00084d>.
14. Ding, L.X.; Wei, Y.; Li, L.; Zhang, T.; Wang, H.; Xue, J.; Ding, L.X.; Wang, S.; Caro, J.; Gogotsi, Y. MXene Molecular Sieving Membranes for Highly Efficient Gas Separation. *Nat. Commun.* **2018**, *9*, 155. <https://doi.org/10.1038/s41467-017-02529-6>.
15. Shen, J.; Liu, G.; Ji, Y.; Liu, Q.; Cheng, L.; Guan, K.; Zhang, M.; Liu, G.; Xiong, J.; Yang, J.; et al. 2D MXene Nanofilms with Tunable Gas Transport Channels. *Adv. Funct. Mater.* **2018**, *28*, 1801511. <https://doi.org/10.1002/adfm.201801511>.
16. Arshadi, F.; Mohammad, M.; Hosseini, E.; Ahmadi, H.; Asadnia, M.; Orooji, Y.; Korayem, A.H.; Noorbakhsh, A.; Razmjou, A. The Effect of D-Spacing on the Ion Selectivity Performance of MXene Membrane. *J. Memb. Sci.* **2021**, *639*, 119752. <https://doi.org/10.1016/j.memsci.2021.119752>.
17. Dikin, D.A.; Stankovich, S.; Zimney, E.J.; Piner, R.D.; Dommett, G.H.B.; Evmenenko, G.; Nguyen, S.T.; Ruoff, R.S. Preparation and Characterization of Graphene Oxide Paper. *Nature* **2007**, *448*, 457–460. <https://doi.org/10.1038/nature06016>.
18. Hadler-Jacobsen, J.; Fagerli, F.H.; Kaland, H.; Schnell, S.K. Stacking Sequence, Interlayer Bonding, Termination Group Stability and Li/Na/Mg Diffusion in MXenes. *ACS Mater. Lett.* **2021**, *3*, 1369–1376. <https://doi.org/10.1021/acsmaterialslett.1c00316>.

19. Fan, Y.; Wei, L.; Meng, X.; Zhang, W.; Yang, N.; Jin, Y.; Wang, X.; Zhao, M.; Liu, S. An Unprecedented High-Temperature-Tolerance 2D Laminar MXene Membrane for Ultrafast Hydrogen Sieving. *J. Memb. Sci.* **2019**, *569*, 117–123. <https://doi.org/10.1016/j.memsci.2018.10.017>.
20. Sun, Y.; Li, S.; Zhuang, Y.; Liu, G.; Xing, W.; Jing, W. Adjustable Interlayer Spacing of Ultrathin MXene-Derived Membranes for Ion Rejection. *J. Memb. Sci.* **2019**, *591*, 117350. <https://doi.org/10.1016/j.memsci.2019.117350>.
21. Mashtalir, O.; Naguib, M.; Mochalin, V.N.; Dall'Agnese, Y.; Heon, M.; Barsoum, M.W.; Gogotsi, Y. Intercalation and Delamination of Layered Carbides and Carbonitrides. *Nat. Commun.* **2013**, *4*, 1716. <https://doi.org/10.1038/ncomms2664>.
22. Dixit, F.; Zimmermann, K.; Alamoudi, M.; Abkar, L.; Barbeau, B.; Mohseni, M.; Kandasubramanian, B.; Smith, K. Application of MXenes for Air Purification, Gas Separation and Storage: A Review. *Renew. Sustain. Energy Rev.* **2022**, *164*, 112527. <https://doi.org/10.1016/j.rser.2022.112527>.
23. Petukhov, D.I.; Kan, A.S.; Chumakov, A.P.; Konovalov, O.V.; Valeev, R.G.; Eliseev, A.A. MXene-Based Gas Separation Membranes with Sorption Type Selectivity. *J. Memb. Sci.* **2021**, *621*, 118994. <https://doi.org/10.1016/j.memsci.2020.118994>.
24. Dash, A.; Vaßen, R.; Guillon, O.; Gonzalez-Julian, J. Molten Salt Shielded Synthesis of Oxidation Prone Materials in Air. *Nat. Mater.* **2019**, *18*, 465–470. <https://doi.org/10.1038/s41563-019-0328-1>.
25. Alhabeib, M.; Maleski, K.; Anasori, B.; Lelyukh, P.; Clark, L.; Sin, S.; Gogotsi, Y. Guidelines for Synthesis and Processing of Two-Dimensional Titanium Carbide (Ti₃C₂T_x MXene). *Chem. Mater.* **2017**, *29*, 7633–7644. <https://doi.org/10.1021/acs.chemmater.7b02847>.
26. Naguib, M.; Unocic, R.R.; Armstrong, B.L.; Nanda, J. Large-Scale Delamination of Multi-Layers Transition Metal Carbides and Carbonitrides “MXenes”. *Dalt. Trans.* **2015**, *44*, 9353–9358. <https://doi.org/10.1039/c5dt01247c>.
27. Ghidui, M.; Lukatskaya, M.R.; Zhao, M.Q.; Gogotsi, Y.; Barsoum, M.W. Conductive Two-Dimensional Titanium Carbide “clay” with High Volumetric Capacitance. *Nature* **2015**, *516*, 78–81. <https://doi.org/10.1038/nature13970>.
28. Shekhirev, M.; Shuck, C.E.; Sarycheva, A.; Gogotsi, Y. Characterization of MXenes at Every Step, from Their Precursors to Single Flakes and Assembled Films. *Prog. Mater. Sci.* **2021**, *120*, 100757. <https://doi.org/10.1016/j.pmatsci.2020.100757>.
29. Shahzad, F.; Alhabeib, M.; Hatter, C.B.; Anasori, B.; Hong, S.M.; Koo, C.M.; Gogotsi, Y. Electromagnetic Interference Shielding with 2D Transition Metal Carbides (MXenes). *Science* **2016**, *353*, 1137–1140. <https://doi.org/10.1126/science.aag2421>.
30. Wei, Y.; Zhang, Y.; Gao, X.; Yuan, Y.; Su, B.; Gao, C. Declining Flux and Narrowing Nanochannels under Wrinkles of Compacted Graphene Oxide Nanofiltration Membranes. *Carbon* **2016**, *108*, 568–575. <https://doi.org/10.1016/j.carbon.2016.07.056>.
31. Anasori, B.; Lukatskaya, M.R.; Gogotsi, Y. 2D Metal Carbides and Nitrides (MXenes) for Energy Storage. *Nat. Rev. Mater.* **2017**, *2*, 16098. <https://doi.org/10.1038/natrevmats.2016.98>.
32. Hu, T.; Hu, M.; Gao, B.; Li, W.; Wang, X. Screening Surface Structure of MXenes by High-Throughput Computation and Vibrational Spectroscopic Confirmation. *J. Phys. Chem. C* **2018**, *122*, 18501–18509. <https://doi.org/10.1021/acs.jpcc.8b04427>.
33. Yazdanparast, S.; Soltanmohammad, S.; Fash-White, A.; Tucker, G.J.; Brennecke, G.L. Synthesis and Surface Chemistry of 2D TiVC Solid-Solution MXenes. *ACS Appl. Mater. Interfaces* **2020**, *12*, 20129–20137. <https://doi.org/10.1021/acsami.0c03181>.
34. Qian, A.; Hyeon, S.E.; Seo, J.Y.; Chung, C.H. Capacitance Changes Associated with Cation-Transport in Free-Standing Flexible Ti₃C₂T_x (T[Dbnd]O, F, OH) MXene Film Electrodes. *Electrochim. Acta* **2018**, *266*, 86–93. <https://doi.org/10.1016/j.electacta.2018.02.019>.
35. Chen, L.; Shi, G.; Shen, J.; Peng, B.; Zhang, B.; Wang, Y.; Bian, F.; Wang, J.; Li, D.; Qian, Z.; et al. Ion Sieving in Graphene Oxide Membranes via Cationic Control of Interlayer Spacing. *Nat. Publ. Gr.* **2017**, *550*, 380–383. <https://doi.org/10.1038/nature24044>.
36. Wang, J.; Zhang, Z.; Zhu, J.; Tian, M.; Zheng, S.; Wang, F.; Wang, X.; Wang, L. Ion Sieving by a Two-Dimensional Ti₃C₂T_x Alginate Lamellar Membrane with Stable Interlayer Spacing. *Nat. Commun.* **2020**, *11*, 3540. <https://doi.org/10.1038/s41467-020-17373-4>.
37. Cheng, R.; Hu, T.; Zhang, H.; Wang, C.; Hu, M.; Yang, J.; Cui, C.; Guang, T.; Li, C.; Shi, C.; et al. Understanding the Lithium Storage Mechanism of Ti₃C₂T_x MXene. *J. Phys. Chem. C* **2019**, *123*, 1099–1109. <https://doi.org/10.1021/acs.jpcc.8b10790>.
38. Sarycheva, A.; Gogotsi, Y. Raman Spectroscopy Analysis of the Structure and Surface Chemistry of Ti₃C₂T_x MXene. *Chem. Mater.* **2020**, *32*, 3480–3488. <https://doi.org/10.1021/acs.chemmater.0c00359>.
39. Seredych, M.; Shuck, C.E.; Pinto, D.; Alhabeib, M.; Precetti, E.; Deysher, G.; Anasori, B.; Kurra, N.; Gogotsi, Y. High-Temperature Behavior and Surface Chemistry of Carbide MXenes Studied by Thermal Analysis. *Chem. Mater.* **2019**, *31*, 3324–3332. <https://doi.org/10.1021/acs.chemmater.9b00397>.
40. Sang, X.; Xie, Y.; Yilmaz, D.E.; Lotfi, R.; Alhabeib, M.; Ostadhossein, A.; Anasori, B.; Sun, W.; Li, X.; Xiao, K.; et al. In Situ Atomistic Insight into the Growth Mechanisms of Single Layer 2D Transition Metal Carbides. *Nat. Commun.* **2018**, *9*, 2266. <https://doi.org/10.1038/s41467-018-04610-0>.
41. Zhang, C.J.; Pinilla, S.; McEvoy, N.; Cullen, C.P.; Anasori, B.; Long, E.; Park, S.H.; Seral-Ascaso, A.; Shmeliov, A.; Krishnan, D.; et al. Oxidation Stability of Colloidal Two-Dimensional Titanium Carbides (MXenes). *Chem. Mater.* **2017**, *29*, 4848–4856. <https://doi.org/10.1021/acs.chemmater.7b00745>.
42. Wan, Y.J.; Rajavel, K.; Li, X.M.; Wang, X.Y.; Liao, S.Y.; Lin, Z.Q.; Zhu, P.L.; Sun, R.; Wong, C.P. Electromagnetic Interference Shielding of Ti₃C₂T_x MXene Modified by Ionic Liquid for High Chemical Stability and Excellent Mechanical Strength. *Chem. Eng. J.* **2021**, *408*, 127303. <https://doi.org/10.1016/j.cej.2020.127303>.

-
43. Zheng, Z.; Guo, C.; Wang, E.; He, Z.; Liang, T.; Yang, T.; Hou, X. The Oxidation and Thermal Stability of Two-Dimensional Transition Metal Carbides and/or Carbonitrides (MXenes) and the Improvement Based on Their Surface State. *Inorg. Chem. Front.* **2021**, *8*, 2164–2182. <https://doi.org/10.1039/d1qi00041a>.
 44. Cao, F.; Zhang, Y.; Wang, H.; Khan, K.; Tareen, A.K.; Qian, W.; Zhang, H.; Ågren, H. Recent Advances in Oxidation Stable Chemistry of 2D MXenes. *Adv. Mater.* **2022**, *34*, 2107554. <https://doi.org/10.1002/ADMA.202107554>.
 45. Feng, A.; Yu, Y.; Jiang, F.; Wang, Y.; Mi, L.; Yu, Y.; Song, L. Fabrication and Thermal Stability of NH₄HF₂-Etched Ti₃C₂ MXene. *Ceram. Int.* **2017**, *43*, 6322–6328. <https://doi.org/10.1016/j.ceramint.2017.02.039>.
 46. Chong, J.Y.; Wang, B.; Mattevi, C.; Li, K. Dynamic Microstructure of Graphene Oxide Membranes and the Permeation Flux. *J. Memb. Sci.* **2018**, *549*, 385–392. <https://doi.org/10.1016/j.memsci.2017.12.018>.
 47. Mulder, M. *Basic Principles of Membrane Technology*; Springer: Dordrecht, The Netherlands, 1996; ISBN 9780792342489.
 48. Xu, W.L.; Fang, C.; Zhou, F.; Song, Z.; Liu, Q.; Qiao, R.; Yu, M. Self-Assembly: A Facile Way of Forming Ultrathin, High-Performance Graphene Oxide Membranes for Water Purification. *Nano Lett.* **2017**, *17*, 2928–2933. <https://doi.org/10.1021/acs.nanolett.7b00148>.
 49. Koenig, S.P.; Wang, L.; Pellegrino, J.; Bunch, J.S. Selective Molecular Sieving through Porous Graphene. *Nat. Nanotechnol.* **2012**, *7*, 728–732. <https://doi.org/10.1038/NNANO.2012.162>.