

## Supplementary Material

# Hydroxycinnamyl derived BODIPY as a Lipophilic Fluorescence Probe for Peroxyl Radicals

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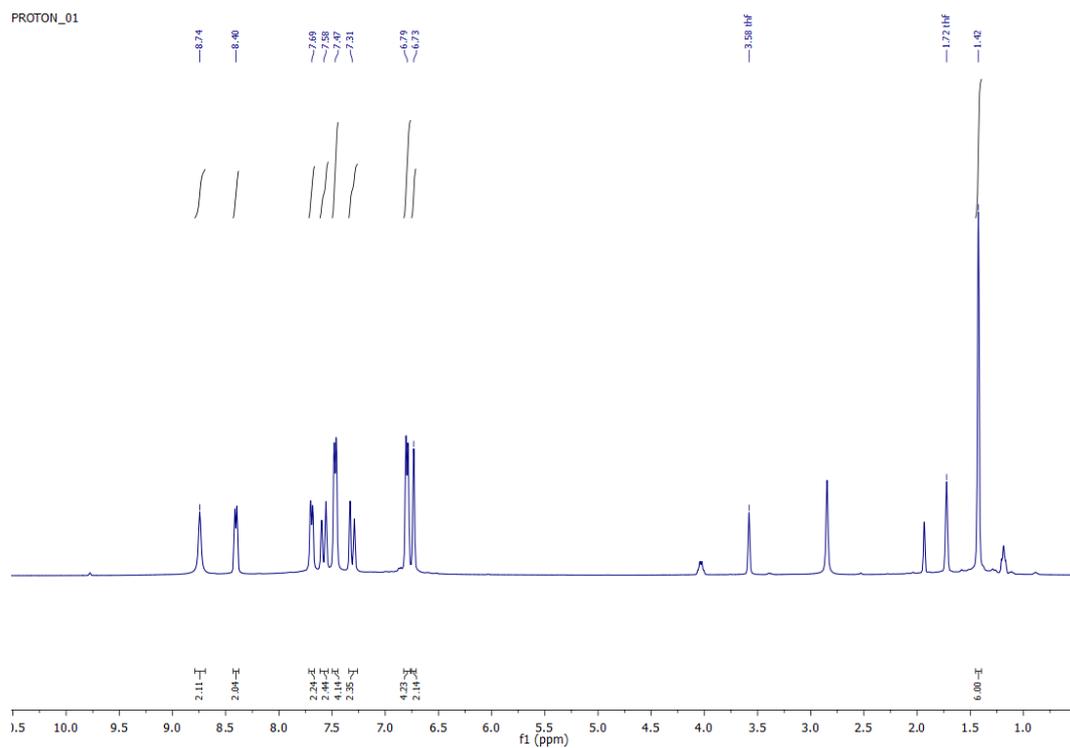
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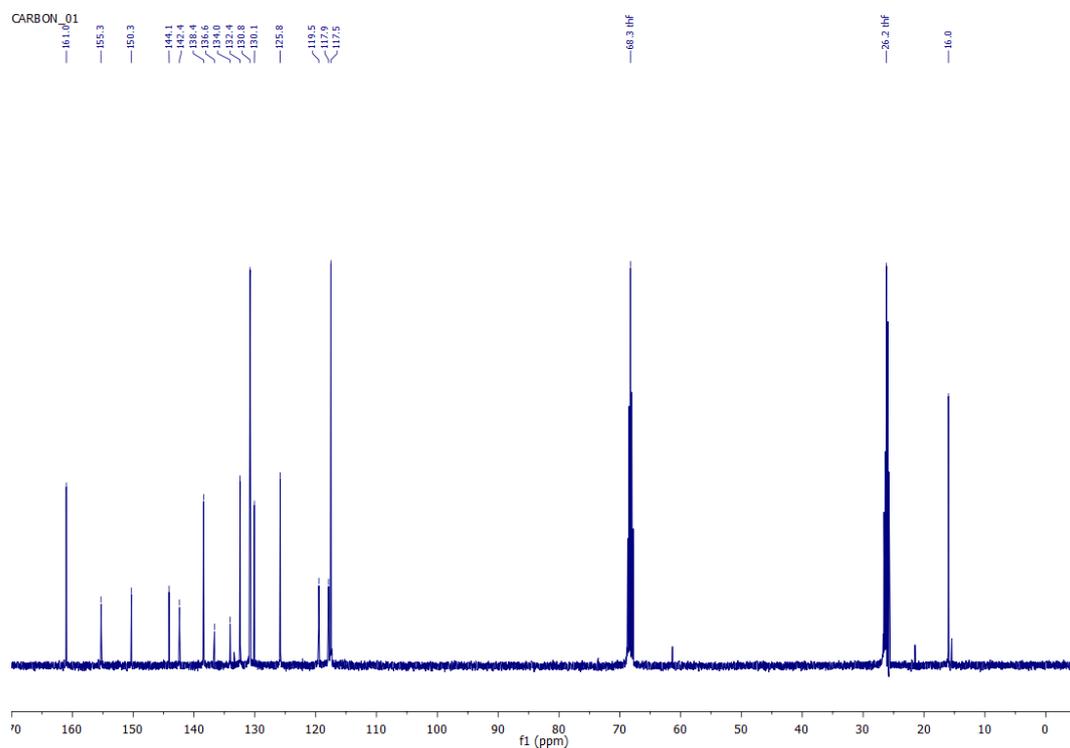
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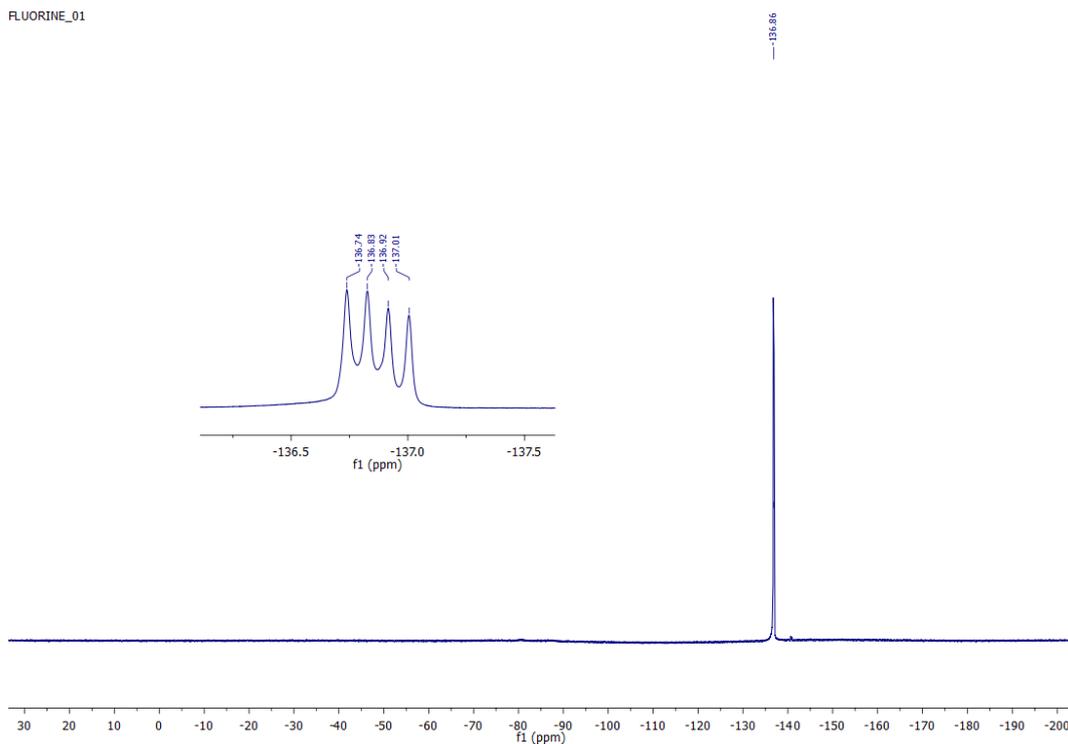
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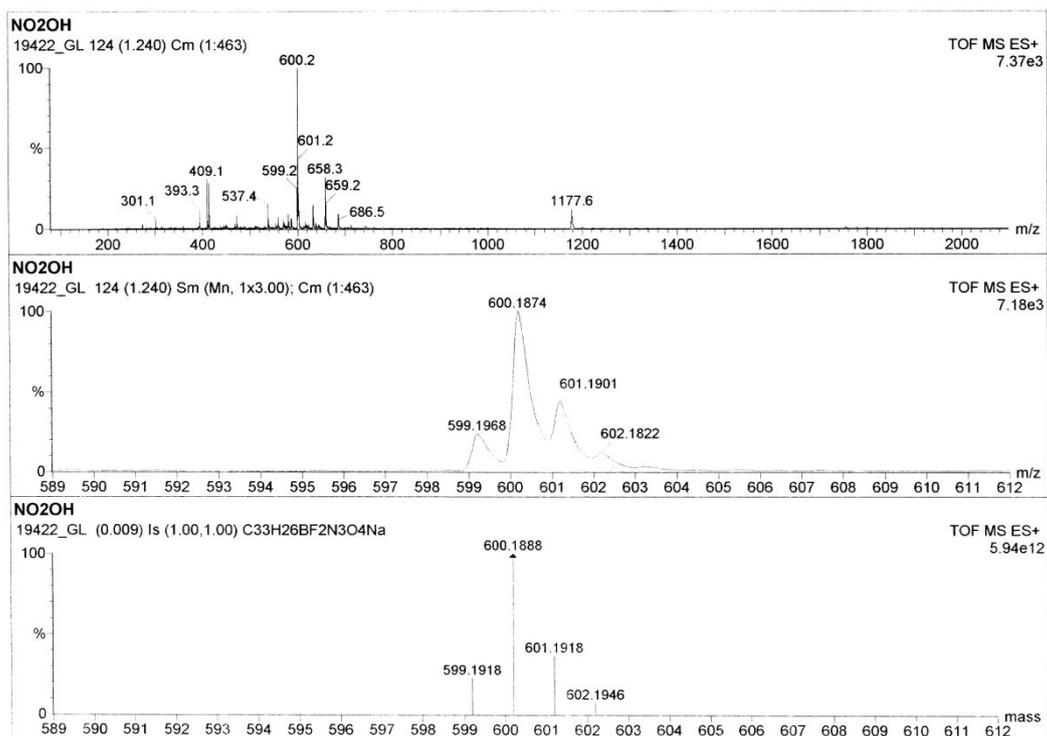
**Figure S1.**  $^1\text{H}$  NMR spectrum of NB-2 (400 MHz, THF- $d_8$ )  $\delta$  1.42 (s, 6H), 6.73 (s, 2H), 6.79 (dq,  $J = 8.8, 2.1$  Hz, 4H), 7.31 (dd,  $J = 16.3, 3.0$  Hz, 2H), 7.43 – 7.50 (m, 4H), 7.57 (d,  $J = 16.3$  Hz, 2H), 7.63 – 7.74 (m, 2H), 8.40 (dq,  $J = 8.8, 2.1$  Hz, 2H), 8.74 (brs, 2H).



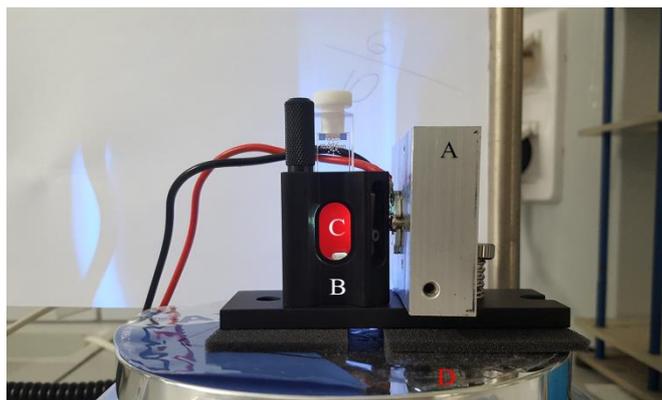
**Figure S2.**  $^{13}\text{C}$  NMR spectrum of NB-2 (101 MHz, THF- $d_8$ )  $\delta$  16.0, 117.5, 117.9, 119.5, 125.8, 130.1, 130.8, 132.4, 134.1, 136.7, 138.4, 142.4, 144.1, 150.3, 155.3, 161.0.



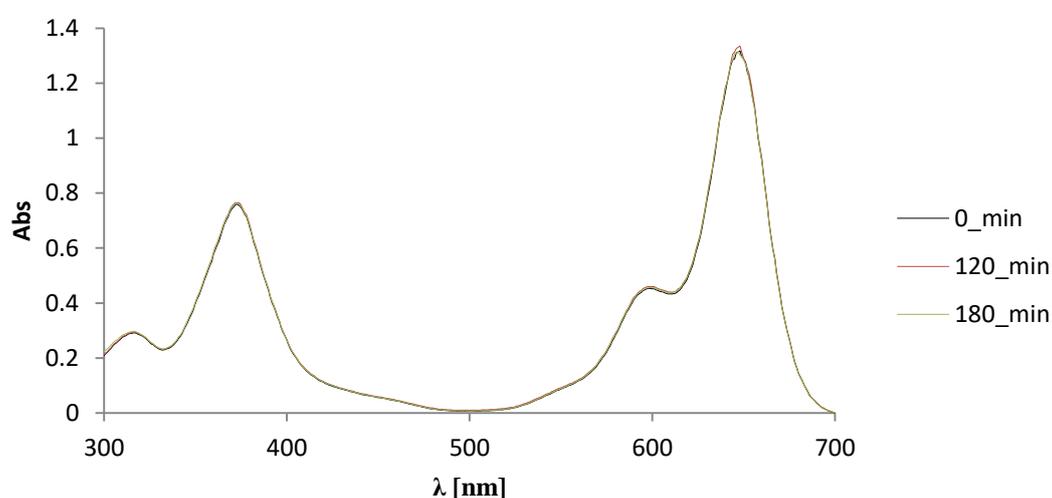
**Figure S3.**  $^{19}\text{F}$  NMR spectrum of NB-2 (376 MHz,  $\text{THF-}d_8$ )  $\delta -136.86$  (dd,  $J = 67.0, 33.1$  Hz).



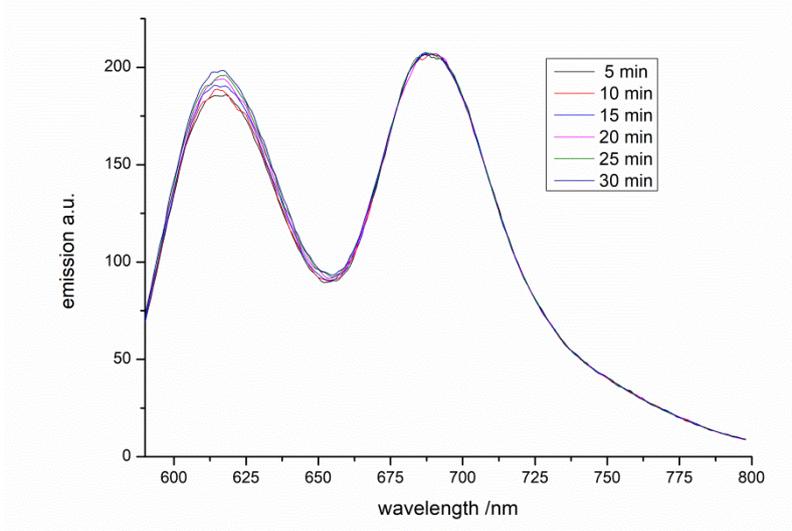
**Figure S4.** HRMS (ESI+) spectrum of NB-2 calc. for  $[\text{M}+\text{Na}]^+$  ( $\text{C}_{33}\text{H}_{26}\text{BF}_2\text{N}_3\text{O}_4\text{Na}$ ): 600.1888, found: 600.1874.



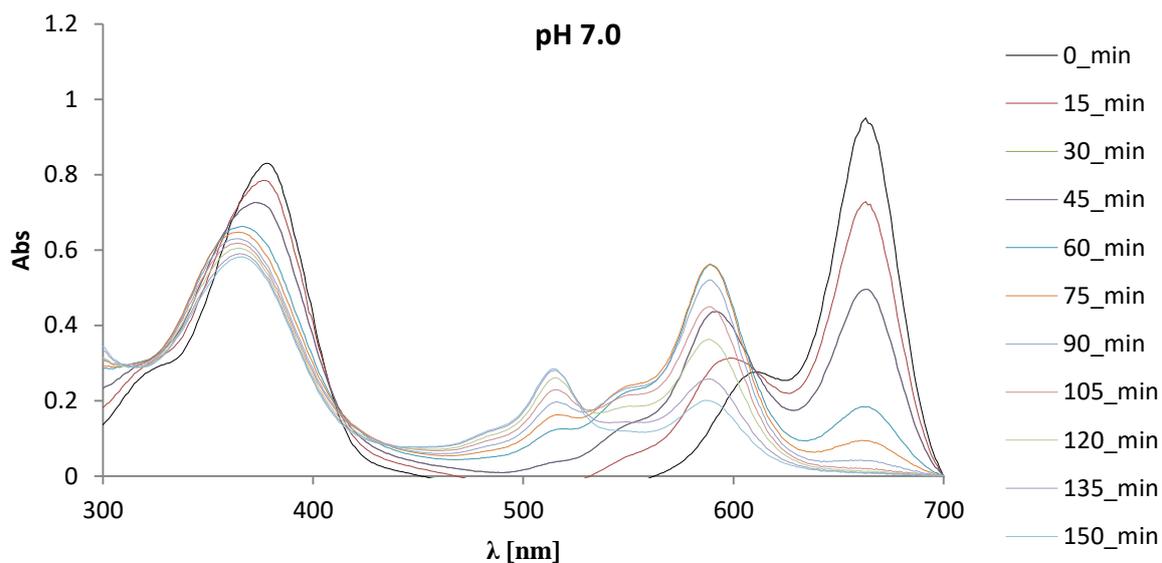
**Figure S5.** UV Irradiation Chamber: UV LED 365 nm with aluminium radiator (A), cuvette holder with in/out windows (B), quartz fluorescence cuvette filled with solution of compound (C), all elements standing on the magnetic stirrer plate (D).



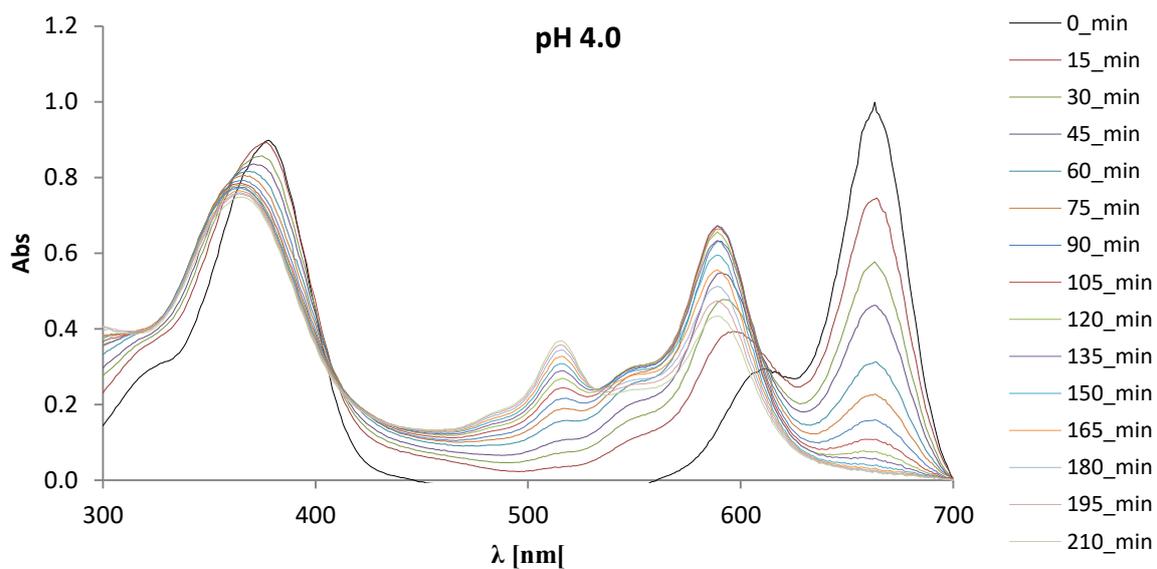
**Figure S6.** Absorption spectra of NB-2 in (~20  $\mu\text{M}$ ) methanol at 37°C during 180 minutes.



**Figure S7.** Stability of NB-2 (10  $\mu\text{M}$ ) in micellar system (8 mM Triton X-100 micelles containing 2.74 mM methyl linoleate) at 37°C and pH 7.4 (phosphate buffer). Emission spectra recorded at 585-800 nm,  $\lambda_{\text{ex}} = 575$  nm.



**Figure S8.** UV-Vis spectra recorded every 15 min. during peroxidation of 2.74 mM methyl linoleate in 8 mM Triton X-100 micelles containing 9.0  $\mu\text{M}$  NB-2 at 37°C and pH 7.0 (Tris buffer). Peroxidation was initiated with 25 mM ABAP.



**Figure S9.** UV-vis spectra recorded every 15 minutes during peroxidation of 2.74 mM methyl linoleate in 8 mM Triton X-100 micelles containing 9.0  $\mu\text{M}$  NB-2 at 37°C and pH 4.0 (acetate buffer). Peroxidation was initiated with 25 mM ABAP.

**Table S1.** The lengths of induction periods,  $\tau_{\text{ind}}$ , the rates of initiation,  $R_i$ , kinetic chain length,  $\nu_{\text{ox}}$ ,  $\nu_{\text{inh}}$ ,  $\nu_{\text{ox1}}$  and the inhibition rate constants,  $k_{\text{inh}}$ , determined for peroxidation of MeLin/Triton X-100 micelles inhibited by 1  $\mu\text{M}$  PMHC /or **NB-1**/ or **NB-2**. Experiments were performed in 8 mM Triton X-100 micelles with 2.73 mM MeLin at 37°C, pH 7.0. Peroxidation was initiated by 10 mM BAP. All experiments were repeated 3-6 times. Values are expressed as the mean  $\pm$  standard deviation (SD).

Compound	$\tau$ /min	$R_i$ /nMs <sup>-1</sup>	$R_{\text{inh}}$ /nM <sup>-1</sup>	$k_{\text{inh}} \times 10^{-3}$ /M <sup>-1</sup> s <sup>-1</sup>	$R_{\text{ox}} \times 10^7$ /M <sup>-1</sup>	$R_{\text{ox1}} \times 10^7$ /M <sup>-1</sup>	$\nu_{\text{ox}}^a$	$\nu_{\text{inh}}^a$	$\nu_{\text{ox1}}^a$
PMHC	6.0 $\pm$ 0.6	4.3	37 $\pm$ 13	12.1 $\pm$ 3.0	4.3 $\pm$ 0.3	2.9 $\pm$ 0.2	100	9	67
NB-1	- <sup>b</sup>	4.3	220 $\pm$ 15 <sup>b</sup>	-	4.3 $\pm$ 0.3	-	100	51	-
NB-2	20.2 $\pm$ 0.8	4.3	90 $\pm$ 9	1.0 $\pm$ 0.1	4.3 $\pm$ 0.3	1.8 $\pm$ 0.1	100	21	42

<sup>a</sup> The kinetic chain length  $\nu$  is the number of peroxidation cycles triggered by one initiating radical. Here, for non-inhibited peroxidation,  $\nu_{\text{ox1}}=R_{\text{ox1}}/R_i$ . <sup>b</sup> For this system, the inhibition period was not detected (see curve 3 in Figure 5) and the rate of the retarded process is listed.