

Review

Advancing Photodynamic Therapy for Endodontic Disinfection with Nanoparticles: Present Evidence and Upcoming Approaches

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Abstract: The persistence of microorganisms in the root canal system is one of the leading causes of root canal treatment failure. Root canal anatomy is complex, and it is often a challenge to obtain optimal disinfection. Biofilms of putative pathogens hidden inside dentin tubules and other root canal ramifications may limit current disinfection protocols. The search for additional disinfection of the root canal has been intensely carried out over the last twenty years. Antimicrobial photodynamic therapy (aPDT) is an adjunctive, conservative, non-selective bacterial kill approach. aPDT has been used to improve root canals disinfection without inducing bacterial resistance. This review focuses on the up-to-date aPDT performance and upcoming promising strategies for disinfection of the root canal system. First, we summarized the barriers encountered by photosensitizer (PS) and light delivery applied to root canal disinfection. Second, we compile the most updated clinical literature. A systematic search for scientific articles was conducted in PubMed, MEDLINE, SCOPUS, and EMBASE to screen the related in vivo studies about this theme. Third, we summarized and critically analyzed the current developments to overcome the aPDT limitations, and we revealed upcoming perspectives in this scoping literature review. We present a timely and opportune review article focusing on the significant potential of aPDT in endodontic disinfection. aPDT offers multiple capabilities that may be considered toward the root canal system's disinfection with future outlooks in nanosized-platforms' design and performance.

Keywords: infection control; dental; dental pulp diseases; photosensitizing agents; dental care; microbiology; biofilms

1. Introduction

Root canal infections are initially treated by removing the infected tissues and disinfecting the root canal system with instrumentation and irrigation protocols. After that, clinicians perform an adequate root canal and apical sealing with the obturation and place an appropriate coronal sealing with a permanent restoration [1]. The root canal treatment usually saves the tooth and clears the infection; however, endodontic treatment can still fail [2]. Different factors have been identified as contributors to endodontic treatment failure, such as untreated or

inadequate filling of the canals. However, pathogenic biofilms' persistence after the standard disinfection has been strongly attributed to endodontic failure and is known as one of the prime causes of unsuccess [3]. The presence of biofilms of putative pathogens trapped inside dentinal tubules, its ramifications and its relation with the recurrence of clinical symptoms and presence of periapical radiolucency is well reported in the literature [4]. The geometry of dentinal tubules with a narrow lumen (1–2 μm) and considerable length (2–3 mm) challenge the disinfection. Previous reports have shown bacterial migration into dentinal tubules of the root at a depth of approximately 360–420 μm [5].

In contrast, the irrigating solution, most commonly, sodium hypochlorite (NaOCl), has penetrability into dentinal tubules of approximately 130 μm . Over and above, the quantification of penetration of the photosensitizer (PS) Toluidine Ortho Blue (TBO) into the dentin substrates via confocal Raman micro-spectroscopy noted the maximum depth of penetration of 45–60 μm [6]. As illustrated in Figure 1, using confocal laser scanning microscopy, viable bacterial cells of *E. faecalis* strain, labeled to show fluorescence in blue, penetrate deep into the dentinal tubules.

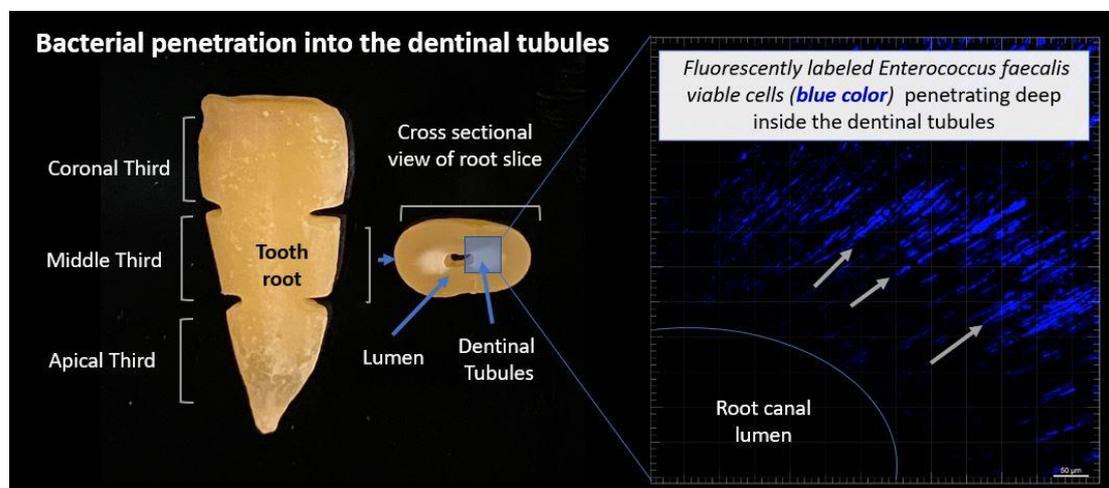


Figure 1. Representative confocal microscopic image of *Enterococcus faecalis* colonizing inside dentinal tubules (20 \times magnification) may complicate the chemo-mechanical disinfection of root canal therapy.

The root canal system is very complex with an isthmus, lateral canals, accessory canals, and root canal ramifications [7]. Such complexity offers a challenge for optimal disinfection of the root canal system. These areas of isthmus and ramifications can harbor putative pathogens that may evade the antimicrobial activity of conventional disinfection protocols, including instrumentation and irrigation of the canals with auxiliary chemical substances. Previous investigations have shown that conventional root canal therapy is limited to 44% to 70% of disinfection [6,8]. Moreover, studies, using culture-dependent approaches, have indicated bacterial growth after root canal disinfection completion [9].

Limited putative pathogens can survive in the root canal system with the lack of nutrients after the endodontic treatment [10]. *E. faecalis* is a Gram-positive bacterium that is frequently recovered from persistent endodontic infections [11]. *E. faecalis* is detected in 33% of persistent endodontic infections [12]. This pathogen can be found as a monoculture in root canals with failed treatment [13]. *E. faecalis* has unique characteristics to survive, even after disinfection. This pathogen can penetrate the dentinal tubules, persist in a low-nutrient environment and pH, resist high salinity and temperatures, survive even with the presence of intercanal medications and irrigants, develop antibiotic resistance, and form biofilms in medicated canals [14].

Overall, the complexity of the root canal system and the current infection protocols offer a challenge for the success of root canal therapy. Therefore, dentist–scientists are constantly searching for additional disinfection approaches to improve root canal disinfection.

tion. Antimicrobial photodynamic therapy (aPDT) has been proposed as a supplemental approach to optimize root canal disinfection. In this manuscript, we conducted a scoping review with a systematic search of the literature, using PubMed, MEDLINE, SCOPUS, and EMBASE to screen the related in vivo studies. We summarized and critically analyzed the current developments to overcome the aPDT limitations, and we reveal upcoming perspectives along with this review.

2. Overview of Chemo-Mechanical Disinfection and Current Intracanal Medications

Endodontic treatment involves a chemo-mechanical approach to eliminate the infected tissues and disinfect the root dentin [15]. In a single-visit treatment appointment, the infected tissues are removed using rotary file systems called “intracanal instrumentation.” Around 35% of the intracanal walls may remain intact after the instrumentation, reflecting in remaining infected tissues [16]. This scenario mandates combining the mechanical instrumentation and irrigation of the canals, using a sodium hypochlorite (NaOCl) solution [17]. NaOCl irrigation is not only helpful to disinfect the canals, but also to eliminate endotoxins produced by the putative endodontic pathogens [1,18]. Free chlorine in NaOCl irrigation can dissolve necrotic tissues by breaking down proteins into amino acids [19]. Consistent exchange and the use of a large volume of irrigation are recommended to maintain the antibacterial effectiveness of NaOCl [20].

Although robust bacterial reduction was gained, using instrumentation and irrigation, efficient disinfection of the root canal system is still not reached. As mentioned earlier, specific anatomic landmarks and the complexity of the formed biofilms may limit this chemo-mechanical approach’s effectiveness [3,21,22]. *E. faecalis* can penetrate dentinal tubules to a deep extent, escaping from instrumentation and irrigation [23]. Another approach used to increase bacteria reduction is the use of antibacterial intracanal medications. These medications are applied inside the root canal and left to act between the two visits to increase the endodontic treatment’s success rate [24,25].

Calcium hydroxide ($\text{Ca}(\text{OH})_2$) is the most commonly used intracanal medication in endodontic treatment. $\text{Ca}(\text{OH})_2$ reduces the bacterial content via the cell layer’s damage by direct contact and increasing the pH [26]. Even though it has a wide range of antibacterial activity, calcium hydroxide is not effective against *E. faecalis* [7,8]. Its proton pump enables *E. faecalis* to resist high pH [8] and grow in the presence of calcium hydroxide [7]. Moreover, *E. faecalis* can invade tubules and bind to collagen [6], depending on environmental signals to regulate its genetic expression [9]. Therefore, two-visit appointments with calcium hydroxide as intracanal medication does not show better periapical healing results, as *E. faecalis* can be recovered after the treatment [10,11].

Chlorhexidine (CHX) at 2% was found more effective against *E. faecalis* than $\text{Ca}(\text{OH})_2$ [12]. However, there is no further bacterial reduction when compared to chemo-mechanical instrumentation. In this context, the use of antibiotics as intracanal medications is recommended. However, antibiotics’ diffusion through the root canal may not be sufficient to inactivate bacteria [27]. Several investigations found that antibiotics could not fully eradicate pathogens inside the root canal system [13,14]. In addition, bacterial resistance development following the use of antibiotics is a rising concern among dentists. In the United States alone, around 2 million individuals are affected by infectious disease owing to bacterial resistance, resulting in 20,000 deaths annually [28]. Therefore, more efforts should be directed to explore and design other approaches that replace antibiotic use in targeting dental pathogens.

3. Antimicrobial Photodynamic Therapy (aPDT)

Antimicrobial Photodynamic Therapy (aPDT) is an adjuvant technique used to improve root canal disinfection without inducing bacterial resistance [29]. In this technique, a light at a specific wavelength is used to activate PSs, generating reactive oxygen species (ROS) [30]. The aPDT mechanisms are mainly divided into Type I and Type II, as shown in Figure 2.

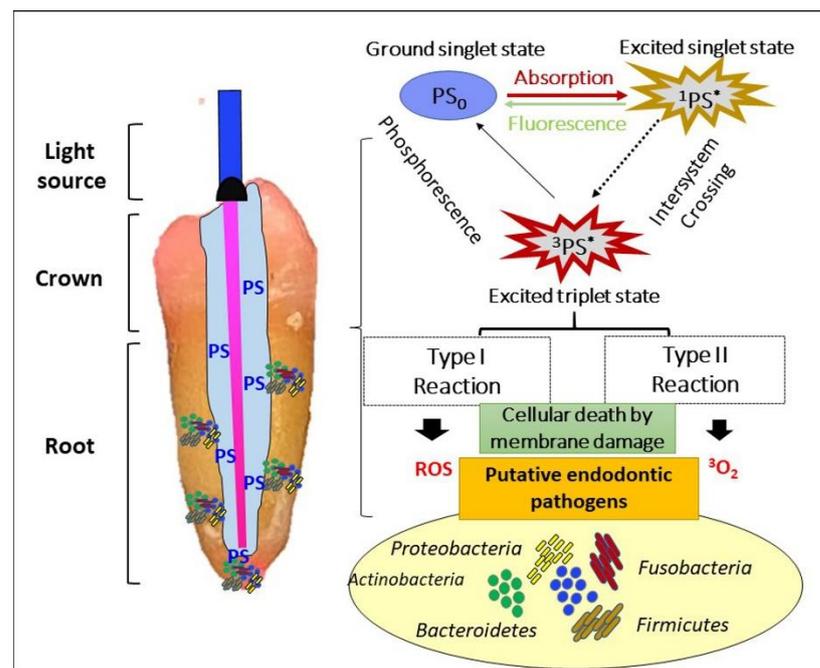


Figure 2. A light source with a specific wavelength can activate a photosensitizer to an excited singlet state and then to a triplet singlet state in antimicrobial photodynamic therapy. The triplet singlet state can induce antimicrobial killing via the generation of the reactive oxygen species (Type I), singlet oxygen (Type II), or both.

Type I: The light-excited PS interacts with the surrounding molecules by an electron or a hydrogen atom exchange. This process triggers specific environmental changes that lead to hydroxyl radicals, one of ROS's responsive forms [31]. After light activation of the PS, a series of released oxygen ions and free radicals kill the targeted cells [32,33]. One of the significant advantages of aPDT is the reliance on the enhanced penetration and accessibility of the reactive oxygen species [34,35].

Type II: Similarly, the PS is activated by light and reacts with the ground state molecular oxygen, generating excited singlet-state oxygen ($^1\text{O}_2$) that directly targets biofilm-triggered diseases [36]. Both types of reactions could happen simultaneously. The ratio between these mechanisms depends on the type of PS used and the PS molecules' microenvironment [31,37].

The photosensitizers that were studied for the elimination of microorganisms belong to different classes of compounds: phenothiazinium derivatives, such as toluidine blue and methylene blue; porphyrin [36] as described in Table 1; phthalocyanine derivatives (disulphonated aluminum phthalocyanine and cationic Zn(II)-phthalocyanine) [37]; halogenated xanthenes derivatives, such as Rose Bengal; triarylmethane dyes, such as Malachite green, acridines; some conjugates of chlorins; and perylenequinones, such as hypericin [37]. However, the most used class for aPDT in endodontics is phenothiazinium derivatives, such as toluidine blue and methylene blue.

Table 1. General features of main photosensitizers used in aPDT against oral bacteria.

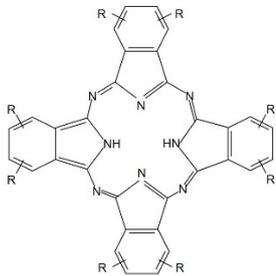
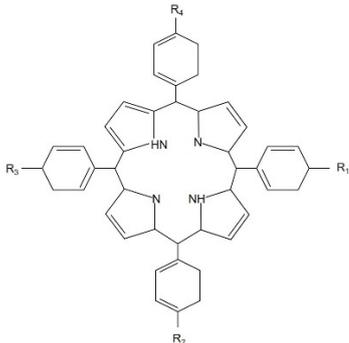
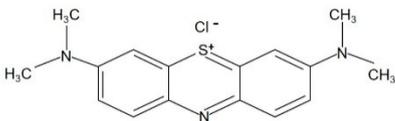
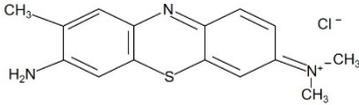
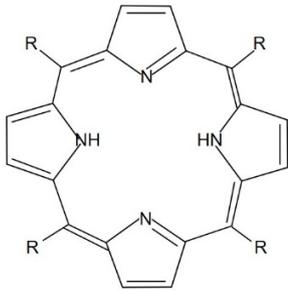
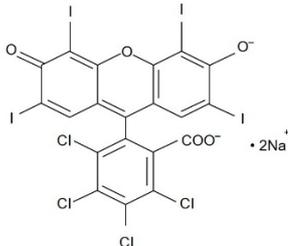
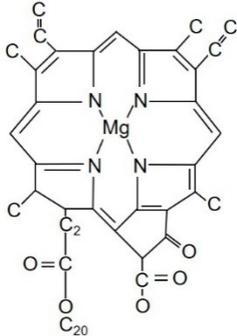
Chemical Class	Photosensitizer	Spectra Absorption	Basic Chemical Structure
Phthalocyanines	Zinc phthalocyanine	600–700 nm	
	Aluminum disulphonated phthalocyanine (AlPcS2)	675 nm	
Phenothiazines	Methylene blue	600–650 nm	
	Toluidine blue ortho	632–638 nm	
Porphyrin platform	Porphyrin	632 nm	
	Rose Bengal	500 nm	

Table 1. Cont.

Chlorophyll platform Chloryns	Chloryn e6	645–675 nm	
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Most of them possess intense absorption bands in the red end of the visible spectrum ($\lambda > 600$ nm). The absorption in red provides a relatively high penetration power in most human tissues and is not substantially absorbed by endogenous cell/tissue constituents, thus minimizing the risk of undesired side effects [38]. Differences in antibacterial action concerning the phenothiazinium photosensitizers are known for toluidine blue O (TBO) and MBO. Against *E. coli*, TBO is known to be membrane active since it interacts more easily with the bacterial membrane than the methylene blue because this dye presents great solubility in the hydrophobic region of the membrane. This dye's efficient antibacterial photoactivity causes increased permeability, whereas MB causes strand breaks in the organism's nucleic acid. These two compounds present physico-chemical structures and properties that are similar, but the photodynamic efficiency varies between different microorganisms.

In the early 2000 s, aPDT has become more exciting and promising in the dental field as a non-to-minimally invasive approach [38]. The reasonable easy access to the oral cavity by the light source/PS solution and the effective killing of putative pathogens in vitro have granted clinical investigations in many dental specialties, such as periodontics, cariology, and endodontics [39] over the last twenty years. By 2013, the field of endodontics effectually started to experience rapid growth in the aPDT investigations, as described in Figure 3A. However, aPDT still has a long path to reach the status of standard care. In our up-to-date evidence synthesis, aPDT lacks much high-level evidence, as shown in Figure 3B, where only approximately 10% of the studies reported are randomized clinical trials.

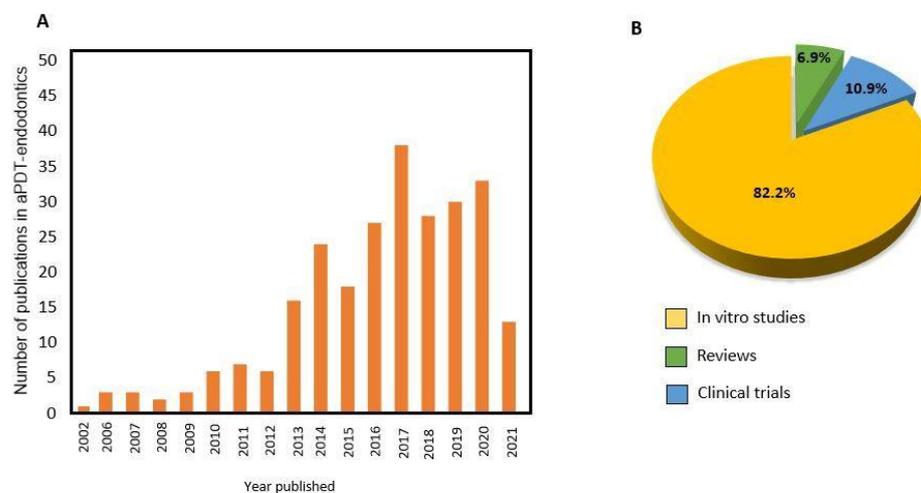


Figure 3. (A) Over time, the number of publications containing the term “antimicrobial photodynamic therapy AND endodontics” via the searched database and (B) these data represent a synopsis of aPDT investigations in the endodontic literature and reviews in the subject.

In a complementary therapeutic context, aPDT applied to oral biofilm-related diseases brings to the dental field three inherent relevant advantages. First, dual site-specificity in which only the target cells that uptake the PS and are irradiated are compromised. Consequently, non-irradiated tissues will not show toxicity issues. Furthermore, PSs characteristically do not exhibit toxicity in the dark, which does not trigger the bacteria to participate in adaptive survival mechanisms against the PSs [40]. Second, the repeated protocol for aPDT with multiple applications per session does not promote bacterial resistance against the PSs. The timeframe for the photo process to occur is too short. In addition, it is difficult for the bacteria to note the upcoming oxidative stress provided by the photo process and create a defensive mechanism, such as the antioxidant defense [41]. Third, after the photo-oxidative attack, the bacteria are too weak to activate their cross-generation adaptivity [41].

4. aPDT Performance Based on In Vitro Studies

Laboratory outcomes are promising for aPDT against several oral microorganisms, such as *Streptococcus mutans*, *Fusobacterium nucleatum*, *Porphyromonas gingivalis*, and *Prevotella intermedia* [42,43]. This approach opens the doors to investigate the effectiveness of aPDT against putative endodontic pathogens. Figure 4 shows the clinical protocol of aPDT applied to complement the disinfection of the root canal. The reported literature covers either in vitro studies where the therapy was applied against planktonic cultures of endodontic pathogens strains or monospecies biofilms, or studies where the aPDT effectiveness was assessed against endodontic bacterial strains grown inside root canals of extracted teeth. These in vitro studies explored the potential effect of aPDT against specific putative endodontic pathogens (mainly *E. faecalis*) and provide optimal parameters as a variable controlled in this type of designed study. The in vitro outcomes can pave the way for well-designed clinical studies and help support any discussion on the biological plausibility of aPDT, but they offer a low level of scientific evidence.



Figure 4. Schematic sequence of the antimicrobial photodynamic therapy clinical protocol applies for root canal disinfection. (A) Pre-treatment tooth diagnosis with infected canals. (B) Access to root canal system to remove the infective pulpal tissues. (C) Chemical debridement and irrigation with 2.5% of sodium hypochlorite (NaOCl). (D) Mechanical instrumentation using rotary files to remove the infective radicular dentine. (E) A final rinse with 0.9% of saline. (F) A photosensitizer applied inside the canal for 1 min. (G,H) A red light-emitting diode (LED; 660 nm; 200 mW), used to activate the photosensitizer.

Consequently, one of the pioneers in in vitro studies was conducted in 2002, where *Streptococcus intermedius* biofilms were grown inside the root canal of extracted teeth for 48 h [44]. Toluidine Blue Ortho (TBO) at concentrations of 12.5, 25, 50, and 100 µg/mL was used to disinfect the root canal, using different energy doses ranged between 2.1 and 21 J/cm². It was found that the concentration of the PS and light dosimetry are determinants concerning the antibacterial reduction. aPDT using TBO was found effective in reducing the *S. intermedius* biofilm by around 5-log [44].

Azulene-based paste (25 µg/mL) as a PS and at an energy dose of 1.8 J/cm² was found to inhibit the *E. faecalis* biofilm [45]. aPDT using chlorine and polyethyleneimine along with the root canal treatment was more effective than each approach alone in inhibiting the growth of *Proteus mirabilis* and *Pseudomonas aeruginosa* [46]. The importance of the PS's concentration can be observed in another study where the effectiveness of 6.25 µg/mL of methylene blue ortho (MB), combined with an energy density of 30 J/cm², against *E. faecalis* biofilm was limited to < 1-log reduction [47]. However, MB applied at a high concentration of 100 µg/mL combined with an energy dose of 16.2 J/cm² promoted an around 2.5-log reduction [48]. It worth noting that the applied energy density is approximately half of the value in the previous study. This outcome highlights the relevance of PS concentration selection.

Often, in vitro studies evaluate aPDT against early-grown endodontic biofilms. Early-grown biofilms may not represent the complexity and maturity of biofilms found inside the oral cavity. The lack of laboratory models that mimic the clinical scenarios may not result in reliable outcomes [49]. Therefore, the use of mature biofilms grown for an extended time (proximally 21 days) is recommended [50]. In an earlier investigation, 30-day *E. faecalis* biofilms grown inside extracted teeth were treated with aPDT [50]. The outcome of the aPDT (MB; 25 µg/mL) was similar to conventional irrigation [50].

5. aPDT Performance in Clinical Studies

Evidence-based dentistry guides dentists to integrate clinical judgment and the patient's values with the best available evidence. Overall, the existing literature on aPDT performance can be categorized as studies that have a randomized experimental design. For this review, we searched for keywords and subject terms related to randomized clinical studies that evaluated the antibacterial effects of aPDT against endodontic pathogens. Two independent reviewers (R.A.A. and A.A.B) performed the searches, using PubMed, MEDLINE, SCOPUS, and EMBASE databases. Table 2 and Box 1 demonstrate a summary of the randomized clinical trials investigating the role of aPDT in root canal disinfection. The following search strategy was used: “((photodynamic therapies) OR photodynamic OR Photochemotherapy) AND (endodontic* OR (root canal)) AND ((case control) OR (clinical trial) OR (in vivo))”. Previously published reviews about this theme were also assessed to look for cross-references about in vivo studies. Grey literature was also reviewed. A total of 205 studies were extracted. After removing the duplicates and reading the abstracts, 27 articles were eligible for full-text evaluation. The inclusion criteria include the randomized clinical trials that used aPDT for endodontic disinfection. Articles with no clear randomization approach or those that used only a light therapy without PSs were excluded. After the full-text evaluation, only seven randomized clinical trials were included (Table 1).

Table 2. Summary of the randomized clinical trials reporting aPDT outcomes against endodontic infections.

Author	Target Tooth	PS	Light Parameters	Protocol	Main Outcome
Ahangari Z et al. (2017) [51]	Root canal treated molars with periapical lesion	MB (50 mg/mL)	Diode laser (808 nm; 0.2 W)	The PS intracanal application; 5 min + 10 s irradiation	Both aPDT and calcium hydroxide therapies significantly reduced the CFUs counts of <i>E. faecalis</i> and <i>C. albicans</i> , with no significant difference between the two approaches.
Asnaashari M et al. (2017) [52]	Root canal treated molars with periapical lesion but with no existing pain, swelling, or any systematic diseases	TBO (0.1 mg/mL)	Red LED (630 nm; 2–4 mW; 1.2–4.4 J/cm ²)	The PS intracanal application; 5 min + 60 s irradiation	The microbiological sampling revealed that aPDT could disinfect the canals in a single visit. aPDT was associated with a lower number of colonies compared to the calcium hydroxide group.
Rabello DGD et al. (2017) [53]	Root canal treated teeth (single root) with apical periodontitis	MB (0.1 mg/mL)	Diode Laser (660 nm; 60 mW; 129 J/cm ²)	The PS intracanal application; 1 min + 2 min irradiation	In the single-visit treatment, aPDT significantly reduced the bacterial load inside the root canals. In the two-visit treatment, aPDT was used following calcium hydroxide, and no additional benefits from using the aPDT were observed. Using the aPDT did not complement the reduction of endotoxins inside the canals, while calcium hydroxide therapy was significantly reduced.
da Silva C.C. et al. (2018) [54]	Non-treated single-rooted teeth diagnosed with necrotic pulp and apical periodontitis	MB (100 µg/mL)	Indium-gallium-aluminum-phosphide laser (660 nm; 100 mW; 7 J/cm ²)	The PS intracanal application; 5 min + 2 × 40 s irradiation at the apical level + 1 × 30 s irradiation By light tip movement	aPDT was associated with significant <i>E. faecalis</i> inhibition at the second visit.
de Miranda and Colombo (2018) [55]	Non-treated molars diagnosed with pulp necrosis and radiographic apical periodontitis	MB (25 µg/mL)	Diode laser (660 nm; 100 mW)	The PS intracanal application; 5 min + 5 min irradiation	Both aPDT and conventional therapies promoted an increase in periapical healing over time, but aPDT resulted in better healing at 6-month follow-up compared to conventional endodontic treatment alone.
Barciela B. et al. (2019) [56]	Non-treated single-rooted teeth diagnosed with necrotic pulp and apical periodontitis	MB (0.5 mg/mL)	Diode laser (660 nm; 320 J/cm ²)	The PS intracanal application; 5 min + 90 s irradiation	The post-operative pain between aPDT and conventional endodontic treatment was similar.
Coelho M.S. et al. (2019) [57]	Non-treated single-rooted teeth diagnosed with necrotic pulp	MB (1.56 µM/mL)	CO ₂ or ND:Yag (660 nm; 100 mW; 600 J/cm ²)	The PS intracanal application; 2 min + 3 min irradiation	aPDT was efficient in reducing post-operative pain in single-visit root canal treatment of teeth with necrotic pulps.

Box 1. Summary of aPDT PS and light sources found in the in vitro studies.

- Phenothiazine photosensitizers, mostly methylene blue (MBO) is the chosen PS.
- The concentrations of MB range from 25 to 100 µg/mL.
- Toluidine blue ortho (TBO) at 0.1 mg/mL concentration was used for one study.
- The most frequently used light source was a diode laser at 660 nm.
- Energy doses range from 1.4 to 200 J/cm².

In a clinical trial by Ahangari et al., a significant inhibition against *E. faecalis* and *Candida albicans* was reached when aPDT was performed using MB at 50 mg/mL and diode laser (810 nm; 0.2 W power) delivered by a 200 µm-diameter end tip [51]. After root canal instrumentation and irrigation, bacterial biofilm samples were taken. Then, in the aPDT group, 0.5 mL of methylene blue was placed inside the canals for 5 min, followed by

10 s irradiation. The canals were irrigated again, and another bacterial sample was taken. In the other group, Ca(OH)_2 was used as an intracanal medicament for seven days, and then another bacterial sample was taken. Both approaches revealed significant inhibition against *E. faecalis* and *C. albicans* with no significant difference between them. This report was among the first studies to investigate the effect of aPDT in endodontics in vivo.

Similarly, the use of TBO (0.1 mg/mL) to treat molars with periapical infections was also attempted by Asnaashari et al. [52]. The design was similar to the previous study, where aPDT was compared to Ca(OH)_2 . However, in this study, the TBO was incubated for 5 min and irradiated for another 1 min with an energy density of 1.2–4.4 J/cm² and light intensity of 2.4 mW, using LED light. At the same time, the Ca(OH)_2 was placed for two weeks. The microbiological results revealed that both interventions were effective in eliminating the bacterial biofilms [52]. The results suggested the benefit of aPDT in minimizing the number of endodontic treatment visits. However, long-term evaluation to monitor the periapical infections around the apex may provide more critical information.

The combinatory effect of aPDT and Ca(OH)_2 was investigated to illustrate if a synergistic effect was evident [53]. Following the chemo-mechanical debridement, teeth were either treated in a single visit using aPDT or in two visits, using Ca(OH)_2 medication for 14 days followed by aPDT treatment. The aPDT protocol involved the use of 0.1 mg/mL of MB incubated inside the canal for 1 min and then irradiated for 2 min, using a light intensity of 60 mW and an energy density of 129 J/cm². In the one-visit group, bacterial samples were isolated before and after the chemo-mechanical preparation and after the aPDT treatment. In the two-visit group, bacterial samples were isolated at four time points: (i) before the treatment, (ii) after the chemo-mechanical preparation, (iii) after the Ca(OH)_2 medication, and (v) after the aPDT treatment.

Both approaches significantly reduced the bacterial load following the chemo-mechanical debridement. It worth noting that the bacterial reduction obtained with aPDT treatment in the one-visit group was equivalent to that found after Ca(OH)_2 application. The supplementary aPDT following the Ca(OH)_2 application did not increase the bacterial reduction or endotoxins' content. These results may emphasize the ability of aPDT to minimize the number of endodontic visits.

The adjunctive use of aPDT following the Ca(OH)_2 medication was found significant in reducing the counts of *E. faecalis*, compared to the use of Ca(OH)_2 alone [54]. In this study, MB 100 µg/mL was incubated for 5 min and then activated via an indium-, gallium-, and aluminum-phosphide laser (InGaAlP) for a total of 70 s with a total energy density of 7 J/cm². In another investigation, endodontic treated teeth were observed six months after the intervention [55]. Teeth diagnosed with pulp necrosis and periapical periodontitis were treated with either Ca(OH)_2 medications, kept for 7–10 days with or without supplemental aPDT. The aPDT protocol involved the use of 0.5 mL of MB 25 µg/mL, incubated inside the canal for 5 min followed by 5 min of irradiation time via diode laser. Using radiographic evaluation by observing the tissues around the treated teeth apex, aPDT was significantly associated with improved healing than the conventional endodontic therapy using the Ca(OH)_2 medication. No significant difference in the microbial sampling was found between the two groups. This study was unique for being the first study to perform long-term evaluation following aPDT [55].

Our search also reveals contradicting outcomes. To date, only two investigations have focused on post-operative pain as an outcome after aPDT endodontic disinfection. While one clinical report concluded that no additional benefit is associated with using aPDT to reduce post-operative pain [56], the other showed that using aPDT was effectively significant [57]. Both studies used MB as a PS; however, variation in the parameters may suggest a divergent outcome.

From the perspective of bacterial reduction, five of the seven clinical studies, found in Table 1, reported a positive outcome for aPDT against endodontic pathogens. Four of the five used MB as a PS with different concentrations ranged between 0.025 and 50 mg/mL [51,53–55], and one study used TBO at a concentration of 0.1 mg/mL [52]. The

study that used the highest methylene blue concentration (50 mg/mL) reported that no additional effect could be obtained from using aPDT for disinfection [51]. It is critical to observe here that the irradiation time was low (10 s), which may affect the dosimetry of the aPDT. Oppositely, the other three studies using methylene blue used low concentrations, but the time of irradiation was increased, resulting in some positive outcomes because of using aPDT intervention. Some of the studies did not report the amount of energy density, explaining the results more difficult as the aPDT effect depends on the dosimetry of aPDT [33].

Considering aPDT dosimetry, it was critical to observe how these studies reported energy density delivered to the target area. In the reported studies, fiber optics were used to irradiate the canal's working length level. Logically, the light distribution among the fiber is not similar, resulting in different energy density values at various locations inside the root canal system. Only one study reported the energy density dose as a range [52].

Some of the reported studies applied aPDT as an independent intervention [51,52]. It is also essential to consider aPDT as an adjunctive treatment. More importantly, future studies may attempt to explore the standardization of aPDT dosimetry and new strategies to enhance the disinfection outcomes.

6. Nanostructures-Based Photosensitizers (PSs) to Overcome the Drawback of Conventional Endodontic Therapy: Present and Future Approaches

Bio-nanotechnology has opened new avenues for aPDT. Two significant shortcomings of the current phototherapeutic interventions for the disinfection of root canals are the restricted penetration of photosensitizer (PS) and light propagation inside the dentinal tubules [36]. Extensive laboratory studies have shown that an essential aspect of this therapy is that the two components, when used independently, produce no effect on bacteria or healthy tissue. It is only the combination of PS and light that affects the bacteria [33].

Facing the paucity of delivery strategies to target sites, advances in photonanomedicine strategies have conveyed optimization and feasibility of the clinical outcomes of photodynamic therapy in medicine and biotechnology [58]. Nanoparticles associated with aPDT have greatly enhanced bacterial disinfection outcomes. Nanoparticles (1–100 nm in size) represent emergent PS carriers that show great promise for aPDT. They present the three dimensions in nanoscale and can be organic, inorganic, or combined [59]. Among these structures, polymeric-based nanocapsules and nanospheres and metallic and oxide nanoparticles can be found.

In endodontics, the use of nanoparticles for aPDT has also gained interest. Conventional aPDT, as aforementioned, is based on the use of biocompatible PS applied in a specific site and photoactivated to generate ROS, decreasing microorganisms' viability. MB and TBO are well-recognized PS due to their lack of cytotoxicity and targeting ability against Gram-positive and Gram-negative bacteria [60]. However, the incomplete destruction of biofilms along the root canal using MB as the main PS has boosted the search for enhanced strategies. In this context, nanoparticles can be used as active PS because of their diverse design and increased penetration capacity within biofilms. Different nanoparticle-based PS strategies may be classified, as shown in Figure 5 [61].

Nanoparticle-based PS present many advantages over free photosensitizing molecules. As pointed out by the growing literature [62], the conjugation of nanoparticles with PS significantly improves aPDT effects; this theme is addressed in this review. The most remarkable strategies where nanotechnology is applied to PS are discussed here. Mainly, the use of PS loaded in polymeric nanoparticles, nanoparticles as an active PS, PS in nanoemulsions, quantum dots and nanodiamond roles, and magnetic nanoparticles conjugated with PS are summarized in Box 2.

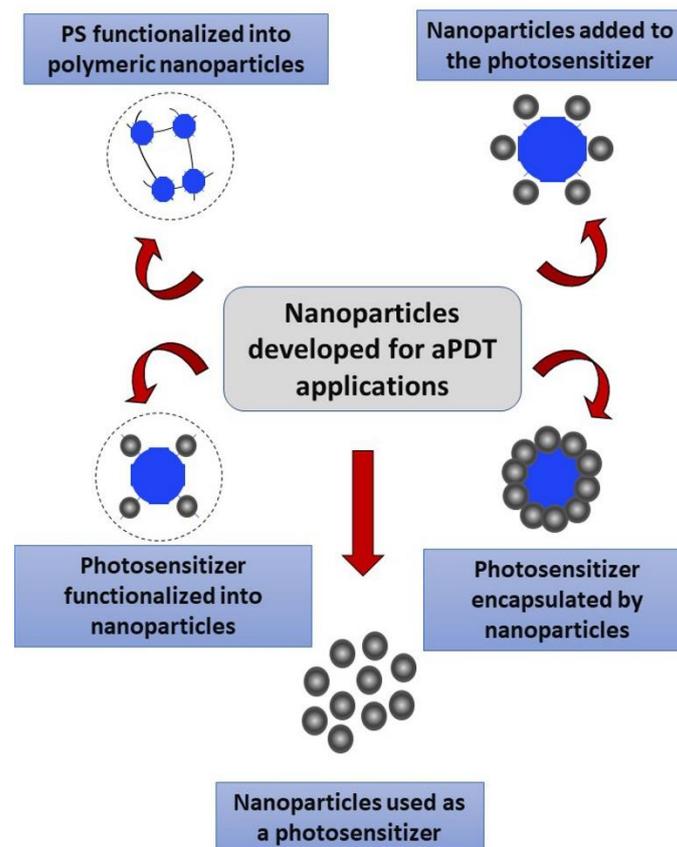


Figure 5. Most remarkable strategies where nanotechnology is applied to PS to enhance the aPDT disinfection outcomes.

6.1. PS Loaded in Polymeric Nanoparticles

The use of polymeric materials as therapeutic carriers is well-established in the literature. Because of their unique properties and designs, such as low cytotoxicity and excellent permeability [63,64], polymeric nanoparticles have been used as carriers for antimicrobial and anti-cancer therapeutic agents [65,66]. In dentistry, polymeric compounds construct dental appliances, resin-based materials, and tissue engineering carriers [67,68]. Several polymeric nanoparticles have been conjugated with PS to improve the performance of aPDT. The most used polymers are chitosan and polylactic-co-glycolic (PLGA). These polymers' cationic nanoparticles can interact effectively with the negatively charged bacterial membrane, inducing nanoholes on the membrane, resulting in bacterial death [36].

In one study, TBO and chitosan polymeric nanoparticles were functionalized to inhibit *E. faecalis* biofilm inside the root canal system. aPDT treatment resulted in a biofilm inhibition similar to the NaOCl irrigation and the use of chitosan alone. However, conjugating the TBO and chitosan resulted in significant and increased inhibition, compared to each treatment alone [69]. Chitosan was also functionalized with rose bengal to target *E. faecalis* and *Pseudomonas aeruginosa* biofilms. At the energy density of 60 J/cm², the chitosan-rose Bengal conjugation achieved significant inhibition with a 7-log reduction in *E. faecalis* biofilm compared to 5-log when rose bengal was used alone. In *P. aeruginosa* biofilms, the conjugation resulted in complete eradication of the biofilms [70]. The conjugation of PLGA and methylene blue resulted in a 2-log reduction in *E. faecalis* biofilm [71].

More investigations are needed to optimize and validate the use of polymeric nanoparticles in aPDT against endodontic pathogens. Future investigations may focus on using ex vivo models where the biofilm must be initiated inside the root canal system for several days. Conducting such experiments may provide more information about the benefits of polymeric nanocarriers in endodontic treatment.

Box 2. The benefits of nanoparticle-based photosensitizers.

- Higher PS per mass content can be achieved when PS are conjugated with nanoparticles, leading to a higher ROS production.
- Reduced ability of the target microorganism to pump molecules out of the cell, which leads to reduced resistance against agents.
- Prospect of targeting the microorganisms due to the improved relationship between nanoparticles and bacteria because of the electronic charge of nanoparticles surfaces.
- The PS achieve higher stability when combined with nanoparticles.
- Lower physical quenching due to PS aggregation. Most PS form aggregates in the aqueous medium when they are in their free form, leading to self-quenching when they are excited and reduced ROS generation.
- Possibility of controlled release of ROS after photoactivation.

6.2. Nanoparticles as an Active PS

Nanoparticles themselves can act as PS via the generation of ROS when photoexcited. Titanium dioxide (TiO₂), zinc oxide (ZnO), and fullerenes nanoparticles produce reactive species, such as singlet oxygen, depending on the wavelength irradiated over them [61]. The crystalline structure of the materials influences the ROS generation. For instance, when rutile and anatase (TiO₂ crystalline structures) were compared, the following order and quantity of generated ROS were observed: anatase, at 365 nm > rutile, at 405 nm > rutile, at 365 nm > anatase, at 405 nm [72]. Despite the possibility of using these particles alone to generate ROS with photoactivation, most of the studies conjugate organic PS with the nanoparticles, such as ZnO with crystal violet [73] or silver with TBO [74]. There is a lack of literature concerning the use of such particles as active PS. Further investigations could be performed to investigate the cytotoxic and antibacterial effects of nanoparticles as an active PS, with or without another PS conjugation.

6.3. PS in Nanoemulsions

A method to carry the PS in photodynamic therapy is in nanoemulsions. Nanoemulsions are “thermodynamically stable colloidal dispersions composed of two immiscible liquids” [75,76]. In nanoemulsions, one of the liquids is presented as a small droplet with a range size inferior to 100 nm dispersed within the other liquid [76]. This approach is currently explored to carry hydrophobic drugs and PS in an aqueous biological environment [77]. Moreover, the oil-based nanoemulsions present antimicrobial activity because of the hydrophobic character and interaction with the phospholipid bilayer of bacterial membranes [75].

Positively and negatively charged nanoemulsions composed of chloro-aluminum phthalocyanine were tested against *Staphylococcus aureus* in biofilm and planktonic forms. Promising antibacterial activity outcomes were achieved when the positively charged nanoemulsion composed of chloro-aluminum phthalocyanine was used at 31.8 µM and a wavelength at 660 nm was applied for 26 min [78].

The effect of nanoemulsions composed of clove oil and zinc phthalocyanines against *E. faecalis* and methicillin-resistant *S. aureus* (MRSA) light-irradiation was attempted to improve aPDT effectiveness [79]. Zinc phthalocyanines present great photobiological activity to be applied in photodynamic therapy. However, they are not miscible with aqueous solutions, which make their handling difficult. The maximum zinc phthalocyanines loaded was 5% and clove oil, 5%, resulting in a nanoemulsion of 30 nm and maintaining the drop size as lower than 50 nm. The minimum inhibitory concentration against *E. faecalis* was 1.09 µg/mL and 0.065 µg/mL against MRSA. When zinc phthalocyanine was within the nanoemulsion, there was higher antibacterial activity than free zinc phthalocyanine. Moreover, as aforementioned, nanoemulsions can present antibacterial activity due to their hydrophobicity. In this study, blank nanoemulsions also presented antibacterial activity [79].

6.4. Quantum Dots in Antimicrobial Photodynamic Therapy

Quantum dots are nanocrystals created from semiconductors and are readily synthesized by processes of self-organization of particles, such as sol-gel [80,81]. Their nano dimension of up to 10 nm confer to them unique optical features and influence their fluorescence wavelength. Moreover, these particles have a high photostability [82]. The fluorescence of quantum dots is greatly attributed to their smaller size more so than the exciton Bohr radius. The exciton Bohr radius is the distance between the electron-hole pair. When the particle's radius is smaller than the electron-hole pair, quantum confinement occurs [83]. The random switching of bright and dark, a usual behavior of quantum dots, was observed by the fluorescence microscopy due to illumination-induced charging (on off). Simultaneously, a non-radiative Auger recombination predominates, and the re-neutralization period (off, on) occurs while radiative recombination prevails [84,85]. This switching effect is observed as a photoemission blinking, which was also indicated when zinc oxide quantum dots were incorporated into experimental adhesives for therapeutic purposes [80].

Quantum dots of tantalum oxide [86], zinc oxide [87], and titanium dioxide [88] (oxides that may show bioactivity) were already synthesized and incorporated into dental materials to provide antibacterial activity. CdSe/ZnS core-shell quantum dots were also incorporated in a resin to tailor dental composites' fluorescence [89]. In recent years, quantum dots have emerged as novel PS in photodynamic therapies [82] due to the aforementioned optical properties. An interesting approach is based on the quantum dots' ability to carry antibiotics, proteins, drugs to combat tumors, and other biomolecules beyond the fluorescence property [82].

Despite the increased bioengineering application, these particles are poorly investigated in photodynamic therapy for disinfection purposes. In 2019, graphene quantum dots doped with curcumin, a natural PS extracted from turmeric roots, were tested against periodontal pathogens mixed biofilms [90]. The particles were irradiated (435 nm for 1 min), and the quantity of ROS formed, as well as the biofilm-formation ability and the changes in gene expressions implicated in the biofilm formation, were evaluated. As an outcome, the photoexcited particles showed antimicrobial activity against planktonic and biofilm forms and regulated the gene expression, evidencing a great alternative PS against perio-pathogens [90].

Moreover, photoactivable polymers with cadmium quantum dots conjugated with crystal violet were tested against important clinical multidrug-resistant bacteria: methicillin-resistant *S. aureus* and a carbapenemase-producing strain of *Escherichia coli* [91]. Crystal violet was added in this study to improve ROS generation. The analyzed photoluminescence lifetime and the ROS generation showed a chemical interaction between the crystal violet and the quantum dots. Furthermore, the particles evidenced great antimicrobial activity against these strains [91]. So far, none of the quantum dots strategies have been investigated in the endodontics field for proof of concept. However, quantum dots seem to be a promising approach to treat biofilms along the root canal due to their ability to be functionalized, conjugate with other drugs, penetrate through narrow canals, emit electromagnetic energy, and be identified via microscopic images.

6.5. The Conjugates of PS and Nanodiamonds

Nanodiamonds, also called diamond nanoparticles, are carbon-based nanomaterials. These particles have been highlighted in developing platforms for delivery agents, as shown in Figure 6. Their fluorescence favors their application for photodynamic purposes [92,93]. Defective sites are generated on nanodiamond surfaces when treated with acids, conferring them high photostability during fluorescence. Nanodiamonds were initially investigated to improve the mechanical properties of polymethyl methacrylate resin [94].

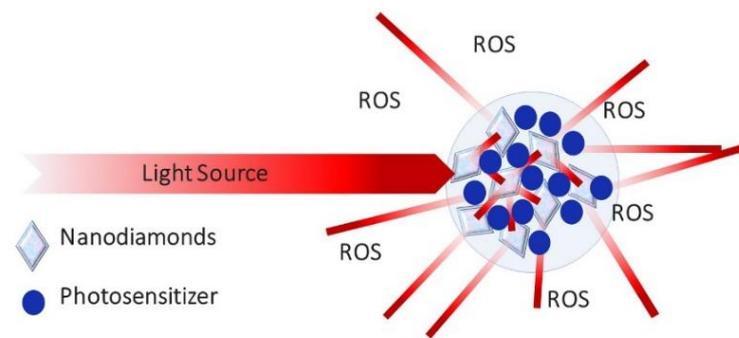


Figure 6. Schematic drawing of the promising conjugation of nanodiamonds with a photosensitizer to enhance aPDT disinfection. Nanodiamonds can act as photosensitizers by themselves, increasing the generation of reactive oxygen species and killing efficiency against biofilms.

Moreover, nanodiamonds have reported inherent antibacterial activity against key pathogen for dental caries, such as *S. mutans* [95]. In endodontics, nanodiamonds were incorporated into gutta-percha and showed activity against *S. aureus* when functionalized with amoxicillin [96]. This approach can be an exciting avenue to increase aPDT disinfection outcomes in constricted areas, such as the root canal, because of light scattering and ROS generation.

6.6. The Conjugates of PS and Magnetic Nanoparticles

Several investigations reported using magnetic nanoparticles and magnetic fields to improve the penetration capabilities of PS [97,98]. The conjugation of curcumin and iron oxide nanoparticles effectively enhanced the killing efficiency against cancer cells in vivo [99]. The same conjugation was investigated by Sun and colleagues for the elimination of periodontal pathogens [100]. The magnetic nanoparticles act as carriers for the PS, leading the PS to the core of biofilms when the magnetic field is applied. This strategy demonstrates the potential to improve PS penetrability through thick biofilms that could be difficult to be removed via conventional approaches [36]. No studies were reported concerning the use of magnetic fields and aPDT in endodontics. Exploring this field in the future may improve the clinical performance of aPDT in clinical use.

6.7. The Conjugates of PS and Liposomes

Liposomes are potential PSs carriers due to their biocompatibility [101]. They are lipid-based systems that can encapsulate hydrophobic and hydrophilic therapeutic agents within their hydrophobic bilayers to control their release and protect them from aggregation or degradation [101]. As most of the PSs are hydrophobic, using liposomes as carriers is advantageous [102]. Several investigations were conducted to implement liposomes in aPDT in dermatology and oncology [103,104]. Only in vitro investigations were conducted in dentistry to validate liposome integration with aPDT to target oral pathogens [105]. Liposomes functionalized into zinc phthalocyanine significantly reduced the growth of *P. gingivalis* [106]. Irradiating the *E. faecalis* biofilms, using 10 and 30 μM of 5,10,15,20-tetra(m-hydroxyphenyl)chlorin functionalized into liposomes resulted in around 5 to 7-log reduction, while using 50 μM of the same conjugation completely eradicated the *E. faecalis* biofilms [107]. 5,10,15,20-tetra(m-hydroxyphenyl)chlorin functionalized into liposomes was found effective in killing *E. faecalis* biofilms up to 300 μm inside the dentinal tubules of root canal systems [108,109]. Future investigations may translate such a conjugation to a clinical model to evaluate its effectiveness, compared to conventional aPDT.

7. Concluding Remarks

The effect of aPDT to disinfect root canal systems is well-described in the endodontic literature. aPDT is a promising approach to prevent reinfection without inducing bacterial resistance. However, the lack of specific protocols to use aPDT is a substantial barrier that may jeopardize the success of aPDT in endodontic disinfection. Further investigations may establish a reliable and effective protocol with excellent capabilities to induce clinical benefits in disinfecting the root canal system. Future studies may explore the use of different nano-platforms to improve the efficiency of aPDT as an adjunctive treatment in root canal therapy. Most of the PSs are hydrophobic with high susceptibility for aggregation and degradation in aqueous solutions. The use of nanotechnology to assist in overcoming limitations and enhance the stability, biocompatibility, and killing capabilities of PSs is a new era for aPDT, targeting oral biofilms.

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