

Supplementary Materials



Synthesis, Structures, Electrochemistry, and Catalytic Activity towards Cyclohexanol Oxidation of Mono-, Di-, and Polynuclear Iron(III) Complexes with 3-Amino-2-Pyrazinecarboxylate

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Figure S1 2D fingerprint plots of **1-4**, exhibiting different intermolecular interactions.



Figure 2. FT-IR spectra of **3** before (in blue), after (in red) the microwave-assisted catalytic peroxidative oxidation of cyclohexanol.



Figure S3 Cyclic voltammograms (current intensity in mA *vs.* potential in V), initiated by the anodic sweep, of compounds **1** - **4** in a 0.2 M [*n*Bu₄N][BF₄]/DMSO solution, at a Pt disc working electrode. Internal standard used: $[Fe(\eta^5-C_5H_5)_2]^{0/+}$ redox couple ($E_{1/2}^{ox} = 0.42$ V vs. SCE)

Table S1: Crystal data and structure refinement details for compounds 1-4					
Identification name	1	2	3	4	
Formulae	C15H12FeN9O6	C16H24Cl2FeN7O4	C22H22Fe2N12O10	C10H9FeN9NaO5	
Mol. wt.	470.19	505.17	726.21	414.10	
Crystal system	Monoclinic	Triclinic	Monoclinic	Monoclinic	
Space group	C2/c	P-1	C2/c	C2/c	
Temperature /K	296	296	296	296	
Wavelength /Å	0.71073	0.71073	0.71073	0.71073	
a /Å	30.021(4)	7.9489(6)	15.014(9)	23.0847(10)	
b /Å	8.3527(11)	16.4168(13)	17.162(10)	9.0844(4)	
c /Å	14.4755(19)	17.8085(13)	15.390(11)	15.1865(6)	
<i>α</i> /°	90	86.504(3)	90	90	
β/°	92.979(5)	79.786(3)	115.92(2)	99.570(2)	
γ/°	90	79.198(3)	90	90	
V/ Å ³	3624.9(8)	2245.6(3)	3567(4)	3140.4(2)	
Ζ	8	4	4	8	
Density/Mgm ⁻³	1.723	1.494	1.352	1.752	
Abs. Coeff. /mm ⁻¹	0.892	0.946	0.876	1.036	
F(000)	1912	1044	1480	1672	
Refl. collected	23983	26499	11850	23080	
Refl. unique	3332	8221	3348	3223	
Max. 20/°	25.556	25.395	26.055	26.422	
	-36<= h <=36	-9<= h <=9	-16<= h <=18	-28<= h <=28	
Ranges (h, k, l)	-10<= k <=10	-19<= k <=19	-20 <= k <=20	-11<= k <=11	
	-17<=l<=17	-21<=1<=21	-18<=l<=16	-19<=1<=18	
Complete to 2θ (%)	98.3	99.3	99.3	99.5	
Refl. with $I > 2\sigma(I)$	3016	5130	1651	2741	
Data/Restraints/Parameters	3332/0/281	8221/31/545	3348/30/208	3223/0/237	
Goof (F ²)	1.113	1.014	1.008	1.067	
R1 [I > 2s(I)]	0.1603	0.0598	0.0918	0.0610	
wR2 [I > 2s(I)]	0.4508	0.1413	0.2345	0.1698	
R1 [all data]	0.1658	0.1123	0.1648	0.0707	
wR2 [all data]	0.4533	0.1645	0.2887	0.1789	

Tabl	le S2: Selected bond distances (Å) and angles (°) for compounds 1-4
1	Fe01-O5 1.882(12); Fe01-O3 1.962(10); Fe01-O1 1.979(11); Fe01-N1 2.047(14); Fe01-N4
	2.134(12); Fe01-N7 2.137(12.
	<pre><o5-fe01-o3, 166.0(5);="" 96.3(5);="" 96.6(5);="" <o3-fe01-o1,="" <o5-fe01-n1,<="" <o5-fe01-o1,="" pre=""></o5-fe01-o3,></pre>
	170.5(5); <o3-fe01-n1, 77.2(4);="" 90.7(5);="" 93.6(5);="" <o1-fe01-n1,="" <o3-fe01-<="" <o5-fe01-n4,="" th=""></o3-fe01-n1,>
	N4, 78.6(5); <o1-fe01-n4, 81.1(4);="" 93.8(4);="" 95.1(5);="" <n1-fe01-n4,="" <o3-<="" <o5-fe01-n7,="" th=""></o1-fe01-n4,>
	Fe01-N7, 93.6(4); <o1-fe01-n7, 170.2(5).<="" 92.3(4);="" 93.7(5);="" <n1-fe01-n7,="" <n4-fe01-n7,="" th=""></o1-fe01-n7,>
2	Fe1-O3 1.968(3); Fe1-O1 2.047(3); Fe1-N1 2.200(4); Fe1-N4 2.236(4); Fe1-Cl1 2.2691(16);
	Fe1-Cl2 2.2725(14); Fe2-O7 1.980(3); Fe2-O5 1.987(3); Fe2-N7 2.186(4); Fe2-N10 2.258(3).
	<o3-fe1-o1, 158.31(15);="" 75.37(14);="" 86.83(15);="" <o1-fe1-n1,="" <o3-fe1-n1,="" <o3-fe1-n4,<="" th=""></o3-fe1-o1,>
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	Fe1-Cl1, 164.54(11); <n1-fe1-cl1, 86.16(11);="" 92.35(12);="" <n4-fe1-cl1,="" <o3-fe1-cl2,<="" th=""></n1-fe1-cl1,>
	93.13(10); <o1-fe1-cl2, 100.18(11);="" 169.00(11);<="" 93.67(11);="" <n1-fe1-cl2,="" <n4-fe1-cl2,="" th=""></o1-fe1-cl2,>
	<cl1-fe1-cl2, 160.85(13);="" 92.79(14);="" 97.78(6);="" <o5-fe2-n7,<="" <o7-fe2-n7,="" <o7-fe2-o5,="" th=""></cl1-fe1-cl2,>
	76.52(13); <o7-fe2-n10, 75.45(13);="" 80.71(13);<="" 87.04(13)="" ;="" <n7-fe2-n10,="" <o5-fe2-n10,="" th=""></o7-fe2-n10,>
	<07-Fe2-Cl3, 96.62(12); <05-Fe2-Cl3, 91.36(11); <n7-fe2-cl3, 165.69(12);="" <n10-fe2-cl3,<="" th=""></n7-fe2-cl3,>
	91.23(11); <o7-fe2-cl4, 103.70(10);="" 89.71(10);="" 91.93(10);="" <n10-fe2-<="" <n7-cl4,="" <o5-fe2-cl4,="" th=""></o7-fe2-cl4,>
	Cl4, 163.63(11); <cl3-fe2-cl4, 100.69(6).<="" th=""></cl3-fe2-cl4,>
3	Fe01-O5 1.976(5); Fe01-O1 1.972(6) ; Fe01-O3 1.991(5); Fe01-N4 2.140(6); Fe01-N1
	2.167(8).
	<pre><05-Fe01-05, 79.6(2); <05-Fe01-01, 92.5(2); <05-Fe01-01, 99.6(2); <05-Fe01-03,</pre>
	102.4(3); <o5-fe01-o3, 162.2(2);="" 92.7(2);="" 94.4(2);="" <o1-fe01-o3,="" <o5-fe01-<="" <o5-fe01-n4,="" th=""></o5-fe01-o3,>
	N4, $168.0(2)$; <o1-fe01-n4, 166.6(2);="" 78.4(2);="" 91.0(2);="" <o3-fe01-n4,="" <o5-<="" <o5-fe01-n1,="" th=""></o1-fe01-n4,>
	FeU1-N1, 93.4(2); $<$ O1-FeU1-N1, 77.3(3); $<$ O3-FeU1-N1, 89.2(3); $<$ N4-FeU1-N1, 94.4(3).
4	FeI-O5 1.7720(5); FeI-O3 2.036(3); FeI-N7 2.042(4); FeI-O1 2.052(3); FeI-N1 2.216(3);
	FeI-N4 2.262(4); NaI-O2 2.292(5); NaI-O6 2.372(7); NaI-O3 2.416(4); NaI-O4 2.469(5); NaI-O4 2.469(5);
	Na1-O1 2.962(4).
	$- OE = E_{0} 1 O2 = 102 04(0)$, $- OE = E_{0} 1 Ni7 = 0(04(14))$, $- O2 = E_{0} 1 Ni7 = 02 82(17)$, $- OE = E_{0} 1 O1$
	(05-ref-05, 102.04(9); < 05-ref-107, 90.94(14); < 05-ref-107, 92.02(17); < 05-ref-01, 06.18(8); < 02. Eq. (17); < 05-ref-01, 06.18(8); < 05. Eq. (17); < 05-ref-01, 06.18(8); < 05-ref-01, 06-ref-01, 06-ref-0
	50.10(0), $50.761-01$, $150.15(12)$, $50.761-01$, $100.20(10)$, $50.761-101$, $50.52(0)$, $50.761-01$, $100.20(10)$, $50.761-101$, $50.52(0)$, $50.761-101$, $50.72(0)$, $50.761-101$, $50.72(0)$, $50.761-101$, $50.72(0)$, $50.761-101$, $50.72(0)$, $50.761-101$, $50.761-101$, $50.72(0)$, $50.761-101$, $50.761-100$, $50.761-1000$
	$170 21(10) < O2 E_01 N/ 86 57(12) < N/7 E_01 N/ 86 05(17) < O1 E_01 N/ 74 25(12) < N/7$
	170.51(10), <05-ref-1N4, 00.57(15), <107-ref-1N4, 00.55(17), <01-ref-1N4, 74.55(15), <107-ref-1N4, 75.55(15), <107-ref-1N4, 75.55(15), <107-ref-1N4, 75.55(15), <107-ref-1N4, 75.55(15), <107-ref-1N4, 75.55(15), <107-ref-1N4, 75.55(15), <107-ref-
	$\begin{array}{c} 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 $
	$N_{21} \cap A = 130.5(3) \cdot \langle \cap A = 103.07(17), \langle \cap A = 108.10(18), \langle \cap A = 104.0(2), $
	$[10a_{1}-04, 150.5(5), -00-10a_{1}-04, 100.10(10), -00-10a_{1}-04, 55.74(15); -102-10a_{1}-04, 88.30(16), -25a_{1}-01.1a_{2}-160.67(16)$
	00.30(10), \re1-\1-1\11, 100.0/(10).

Table S3: Hydrogen bond geometry (Å, °) in compounds 1-4				nds 1-4	
Compoun	D-H…A	D…H			<d-h…a(°)< td=""></d-h…a(°)<>
d		(Å)	п…А (А)	D…A (A)	
1	N9-H9A…O1	0.86	2.51	3.169(17)	134.2
	N9-H9B…O6	0.86	2.14	2.77(2)	129.4
	N3-H3A…O1	0.86	2.59	3.345(18)	146.5
	N3-H3B…O2	0.86	2.14	2.764(19)	128.6
	N3-H3B····N8	0.86	2.59	3.24(2)	132.9
	N6-H6B…O4	0.86	2.08	2.70(3)	128.7
	C5-H5…O4	0.93	2.40	3.106(19)	132.1
	C15-H15…O6	0.93	2.29	3.096(19)	144.4
	C14-H14…O2	0.93	2.59	3.36(2)	141.0
2	N12-H12A…N11	0.86	2.27	3.076(5)	156.3
	N12-H12B…O8	0.86	2.13	2.761(5)	130.1
	N12-H12B…O8	0.86	2.26	2.876(5)	128.5
	N9-H9A…Cl4	0.86	2.64	3.445(5)	157.4
	N9-H9B…O6	0.86	2.13	2.755(6)	129.6
	N6-H6A…N5	0.86	2.26	3.114(6)	170.6
	N6-H6B…O4	0.86	2.12	2.746(6)	129.7
	N6-H6B…O4	0.86	2.21	2.834(6)	129.4
	N3-H3A…Cl4	0.86	2.83	3.663(5)	164.5
	N3-H3B…O2	0.86	2.10	2.734(7)	130.6
	N14-H14A…O5	0.98	1.93	2.886(6)	163.0
	N13-H13N…O1	0.93	1.93	2.844(6)	167.0
	C19-H19-O7	0.93	2.45	3.359(6)	164.4
	С9-Н9…О3	0.93	2.31	3.238(6)	177.1
	C14-H14…Cl2	0.93	2.75	3.658(6)	165.3
	C4-H4…O6	0.93	2.39	3.167(7)	141.0
	C28-H28A…O6	0.97	2.33	3.179(16)	146.0
3	N3-H3A…N6	0.88	2.71	3.567(17)	164.2
	N3-H3B…O2	0.88	2.10	2.720(14)	126.4
	N6-H6A…N5	0.88	2.30	3.063(10)	144.8
	N6-H6B…O4	0.88	2.10	2.757(10)	130.4
	N6-H6B…O4	0.88	2.43	3.026(10)	125.7
	C4-H4…O2	0.93	2.39	3.166(14)	141.0
	С9-Н9…О3	0.93	2.56	3.475(10)	170.1
	С10-Н10…О1	0.93	2.49	3.312(9)	148.2
4	N3-H3A…O1	0.86	2.18	2.963(5)	151.3
_	N3-H3BO4	0.86	2.09	2.727(6)	130.6
	N6-H6B…O2	0.86	2.10	2.739(8)	130.1
	С9-Н9…О4	0.93	2.48	3.224(6)	137.1
	C4-H4…N9	0.93	2.54	3.396(9)	152.7

Table S4 – MW-assisted oxidation ^a of cyclohexanol to cyclohexanone with							
TBHP an	TBHP and catalyzed by $1 - 4$.						
Entry	Catalyst	t/h	T /ºC	Yield/% ^b	TOF/h ⁻¹ ^c		
1	1	0.5	60	47.6	952		
2		1.0	60	87.3	873		
3		1.5	60	86.8	579		
4		2.0	60	81.3	407		
5		2.5	60	79.7	319		
6		3.0	60	77.1	257		
7		0.5	70	52.1	$1.04 \ge 10^3$		
8		1.0	70	91.0	910		
9		1.5	70	91.1	607		
10		2.0	70	84.2	421		
11		2.5	70	80.6	322		
12		3.0	70	73.8	246		
13		0.5	80	66.2	$1.32 \ge 10^3$		
14		1.0	80	89.1	891		
15		1.5	80	81.4	543		
16		2.0	80	71.8	359		
17		2.5	80	60.4	242		
18		3.0	80	46.5	155		
19	2	0.5	60	53.7	$1.07 \ge 10^3$		
20		1.0	60	90.8	908		
21		1.5	60	89.6	597		
22		2.0	60	87.3	437		
23		2.5	60	81.1	324		
24		3.0	60	74.2	247		
25		0.5	70	57.6	1.15 x 10 ³		
26		1.0	70	92.5	925		
27		1.5	70	91.1	607		
28		2.0	70	87.3	437		
29		2.5	70	86.2	345		
30		3.0	70	82.8	276		
31		0.5	80	63.0	1.26 x 10 ³		
32		1.0	80	86.1	861		
33		1.5	80	80.4	536		
34		2.0	80	70.1	351		
35		2.5	80	61.4	246		
36		3.0	80	49.3	164		
37	3	0.5	60	51.1	1.02 x 10 ³		
38		1.0	60	89.2	892		

	1.5	60	88.8	592
	2.0	60	85.3	427
	2.5	60	82.7	331
	3.0	60	82.1	274
	0.5	70	66.6	1.33 x 10 ³
	1.0	70	93.1	931
	1.5	70	93.0	620
	2.0	70	91.3	457
	2.5	70	87.2	349
	3.0	70	83.4	278
	0.5	80	67.0	$1.34 \ge 10^3$
	1.0	80	88.2	882
	1.5	80	82.5	550
	2.0	80	74.6	373
	2.5	80	66.9	268
	3.0	80	51.7	172
4	0.5	60	71.6	$1.43 \ge 10^3$
	1.0	60	90.4	904
	1.5	60	90.3	602
	2.0	60	88.3	442
		(0)	0 (F	
	2.5	60	86.7	347
	2.5 3.0	60 60	86.7	347 273
	2.5 3.0 0.5	60 60 70	86.7 82.1 75.2	347 273 1.50 x 10 ³
	2.5 3.0 0.5 1.0	60 60 70 70	86.7 82.1 75.2 92.7	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \end{array} $
	2.5 3.0 0.5 1.0 1.5	60 60 70 70 70 70	86.7 82.1 75.2 92.7 92.5	347 273 1.50 x 10 ³ 927 617
	2.5 3.0 0.5 1.0 1.5 2.0	60 60 70 70 70 70 70	86.7 82.1 75.2 92.7 92.5 89.0	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ \end{array} $
	2.5 3.0 0.5 1.0 1.5 2.0 2.5	60 60 70 70 70 70 70 70	86.7 82.1 75.2 92.7 92.5 89.0 82.2	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ \end{array} $
	2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0	60 60 70 70 70 70 70 70 70 70	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ \end{array} $
	2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5	60 60 70 70 70 70 70 70 70 80	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1 67.2	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ 1.34 \times 10^3 \\ \end{array} $
	$ \begin{array}{r} 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ 1.5 \\ 2.0 \\ 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ \end{array} $	60 60 70 70 70 70 70 70 70 80 80 80	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1 67.2 89.0	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ 1.34 \times 10^3 \\ 890 \\ \end{array} $
	$ \begin{array}{r} 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ 1.5 \\ 2.0 \\ 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ 1.5 \\ \end{array} $	60 60 70 70 70 70 70 70 80 80 80 80	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1 67.2 89.0 82.5	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ 1.34 \times 10^3 \\ 890 \\ 550 \\ \end{array} $
	$\begin{array}{r} 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ 1.5 \\ 2.0 \\ 2.5 \\ 3.0 \\ 0.5 \\ 1.0 \\ 1.5 \\ 2.0 \end{array}$	60 60 60 70 70 70 70 70 70 80 80 80 80 80 80 80 80	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1 67.2 89.0 82.5 76.1	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ 1.34 \times 10^3 \\ 890 \\ 550 \\ 381 \\ \end{array} $
	$\begin{array}{r} 2.5 \\ \hline 3.0 \\ 0.5 \\ \hline 1.0 \\ 1.5 \\ 2.0 \\ 2.5 \\ \hline 3.0 \\ 0.5 \\ \hline 1.0 \\ 1.5 \\ 2.0 \\ \hline 2.5 \\ \end{array}$	60 60 60 70 70 70 70 70 70 80 80 80 80 80 80 80 80 80 80	86.7 82.1 75.2 92.7 92.5 89.0 82.2 74.1 67.2 89.0 82.5 76.1 67.3	$ \begin{array}{r} 347 \\ 273 \\ 1.50 \times 10^3 \\ 927 \\ 617 \\ 445 \\ 328 \\ 247 \\ 1.34 \times 10^3 \\ 890 \\ 550 \\ 381 \\ 269 \\ \end{array} $
		$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	1.5 0.0 $0.0.0$ 2.0 60 85.3 2.5 60 82.7 3.0 60 82.1 0.5 70 66.6 1.0 70 93.1 1.5 70 93.0 2.0 70 91.3 2.5 70 87.2 3.0 70 83.4 0.5 80 67.0 1.0 80 88.2 1.5 80 82.5 2.0 80 74.6 2.5 80 66.9 3.0 80 51.7 4 0.5 60 71.6 1.0 60 90.4 1.5 60 90.3 2.0 60

^{*a*}Reaction conditions: Cyclohexanol (5 mmol), TBHP (70 % aqueous solution, 10 mmol) and **1** - **4** (5 µmol, 0.1 mol% *vs.* substrate), 10 W of MW irradiation, 60-80°C. ^{*b*}Product yield = (moles of cyclohexanone / initial moles of cyclohexanol)*100. ^{*c*}Turnover frequency = moles of cyclohexanone per mol of catalyst per hour.